


Healable superhydrophobicity of novel cotton fabrics modified via one-pot mist copolymerization

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Abstract We describe a simple one-pot mist copolymerization process to fabricate superhydrophobic cotton fabrics. A mixture solution consisting of a free radical initiator, tert-butyl peroxybenzoate (TBPB), and three monomers, lauryl methacrylate (LMA), 2-isocyanatoethyl methacrylate (IEM), and ethylene glycol dimethacrylate (EGD), is atomized to one side of a cotton fabric and polymerized on the surface. SEM images indicate that the copolymer layer on the cotton fiber surface has a randomly wrinkled morphology exhibiting nanoscale roughness. Wetting tests demonstrate that the modified surface possesses a remarkable superhydrophobicity with multiple healing functionalities. A simple ironing treatment at about 200 °C can recover the degraded superhydrophobicity of the modified cotton fabric suffered from 60 cycles of laundries or 2000 cycles of

Martindale abrasion. Notably, the mist copolymerization process has no significant impact on the cotton advantages, such as flexibility, water absorptivity, and vapor permeability.

Keywords Superhydrophobicity · Cotton fabric · Ironing · Healing ability · Mist polymerization

Introduction

Superhydrophobic surfaces have drawn considerable attention because of their potential applications such as self-cleaning, water–oil separation, anti-fogging, anti-corrosion, drag reduction, anti-bioadhesion, and highly protective clothes (Chu et al. 2015; Wang et al. 2015a; Wen et al. 2015; Wolfs et al. 2013; Dyett et al. 2014). According to the basic principle that the combination of low surface energy material and appropriate hierarchically nanostructures produces superhydrophobicity (Wenzel 1936), various superhydrophobic surfaces have been successfully prepared (Zhang et al. 2012; Wang et al. 2015b; Shirtcliffe et al. 2011). Fluorinated chemicals and polysiloxane polymers were the materials most frequently utilized to lower surface energy (Zhou et al. 2012; Zou et al. 2013; Hua et al. 2013; Wang et al. 2013; Shateri-Khalilabad and Yazdanshenas 2013). On the other hand, various methodologies, for example, dip coating (Nguyen et al. 2012), sol-gel coating (Periolatto et al. 2012), soft-lithography (Kim et al. 2015; Liu et al.

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2006; Yao et al. 2009), electrochemical oxidation (Lee et al. 2010; Darmanin et al. 2013; Xu et al. 2011; Ottone et al. 2014), chemical vapor deposition (Liu et al. 2004; Crick et al. 2012), spray deposition (Hwang et al. 2011; Sparks et al. 2013), chemical etching (Qian and Shen 2005; Dong et al. 2011), surface polymerization (Nystrom et al. 2006; Miao et al. 2010; Yu et al. 2013; Xue et al. 2015), and laser microfabrication (Chen et al. 2013), have been tried to produce such rough surfaces with well-designed nanostructuring. In recent years, beyond the fabrication of superhydrophobic surfaces, more attention has been paid to the functional integration of superhydrophobic surfaces towards practical applications (Zhang et al. 2012; Liu et al. 2014).

Cotton fabrics are soft, comfortable, breathable, skin-friendly, and have an excellent water and moisture absorption ability. Recently, many attempts have been made to increase the functionalities of conventional cotton textiles, such as water repellence (Zou et al. 2013; Abbas et al. 2014; Shin et al. 2012), antimicrobial activity (Dastjerdi and Montazer 2010; Ali et al. 2014; Xi et al. 2015), and flame retardance (Abou-Okeil et al. 2013; Chen et al. 2015). Despite the successful advances in the modification of cotton fabrics reported to date, there are several shortcomings from the perspective of textile engineering. For example, (1) most of the fluorine-containing chemicals and organosilicon compounds are toxic and problematic for the environment (Conder et al. 2008; Isquith et al. 1988); (2) the coating films often affect the original cotton properties such as softness and wearing comfort; (3) the abrasion stability and laundering durability of the superhydrophobic coatings are hardly acceptable in practical applications.

In our previous works (Wang et al. 2014; Wan et al. 2014), we reported a “mist polymerization” technique developed from the gas-phase-assisted surface polymerization methodology (Yang et al. 2012; Andou et al. 2009a, b) to modify cotton fabric surfaces. The “mist polymerization” process can apply to a wider range of monomers and can build nanostructures simply on various substrates. Furthermore, the damage to the original cotton properties can be reduced to a very low level. However, the cumbersome introduction of the radical initiator to the fabric surface, and the unsatisfactory abrasion resistance and laundering durability of the modified cotton fabrics are the obvious disadvantages.

In this work, we report an improved one-pot mist copolymerization technique to generate robust superhydrophobic surfaces having abrasion-durable and healable behaviors. The mixture solution of lauryl methacrylate (LMA), 2-isocyanatoethyl methacrylate (IEM), ethylene glycol dimethacrylate (EGD), and a free radical initiator in cyclohexane are atomized to and heated on the cotton surface. The resulting layer possesses superhydrophobicity with remarkable healing functionality by heating. Notably, the original cotton properties such as water absorbency and moisture transmissibility are still found in the modified cotton fabrics.

Experimental section

Materials

The cotton fabrics used in this work were purchased from a local fabric store (60 ends cm^{-1} , 30 picks cm^{-1} , 0.42 mm thickness, 120 g m^{-2} weight, 35.2 $\text{m}^2 \text{g}^{-1}$ specific surface area). Before chemical modification, the cotton samples (15 mm \times 15 mm) were cleaned by ultrasonic washing in ethanol (2 h) and deionized water (30 min \times 3 times). LMA, IEM, ethylene glycol dimethacrylate (EGD), tert-butyl peroxybenzoate (TBPB), and cyclohexane were purchased from Shanghai Crystal Pure Industrial Co., Ltd. (China). All the chemicals were used as received without further purification. Deionized water with a resistivity of 18.2 $\text{M}\Omega \text{ cm}$ was used in all experiments.

Typical procedures of the one-pot mist copolymerization

A mixture solution of LMA (monomer 1, 3.562 g, 14.000 mmol), IEM (monomer 2, 0.261 g, 1.680 mmol), EGD (monomer 3, 0.056 g, 0.280 mmol), and TBPB (initiator, 0.068 g, 0.350 mmol) in cyclohexane (16 ml) was atomized using an air compression-type atomizer (DongHan DH-M01, China), fed (0.360 ml min^{-1}) to a cotton fabric sample (15 mm \times 15 mm) for 1 min, heated at 80 $^{\circ}\text{C}$ for 1 h and at 110 $^{\circ}\text{C}$ for 4 h, washed by cyclohexane (50 ml \times 3 times) to remove the unreacted monomer, and dried under vacuum to obtain the asymmetrically superhydrophobic cotton fabric. Other modified cotton fabric samples were obtained by adjusting the monomer concentrations and the mist stream

feeding time as listed in every Figure or Table. Fully modified cotton fabrics were prepared via a similar process in which the mist feeding is replaced by a dipping treatment in a mixed solution of LMA (0.700 mol l^{-1}), IEM (0.084 mol l^{-1}), EGD (0.014 mol l^{-1}), and TBPB (0.018 mol l^{-1}).

Characterizations

Static contact angles were measured using deionized water droplets ($4 \mu\text{l}$) with a Kruss contact angle instrument (DSA 100, Germany) at $25 \text{ }^\circ\text{C}$. The water contact angle (WCA) value was recorded when the exposure time of the water droplet on the cotton fabric surface was 5 s. Average contact angle values were obtained by measuring three different positions on the same sample. Surface morphology was investigated by a JSM-6700F field emission scanning electron microscope (FE-SEM, JEOL, Japan) after gold coating (approximately 10 nm thickness). X-ray photoelectron spectroscopy (XPS) analysis was performed by an AXIS multifunctional X-ray photoelectron spectrometer (ULTRA DLD, Shimadzu Ltd., Japan) at a power of 450 W. FTIR spectra were collected from a Nicolet Fourier Transform spectrophotometer (AVATAR 5700, US) with an attenuated total reflection (ATR) accessory.

Laundering durability was evaluated by monitoring WCA on the cotton surface periodically after every stringent washing process. The cotton fabrics ($15 \text{ mm} \times 15 \text{ mm}$) were washed by 50 ml of an aqueous solution of sodium dodecane sulfonate (2.0 %, w/w) in a beaker (diameter, 50 mm) with stirring (300 rpm, magnetic stirrer, $9 \text{ mm} \times 25 \text{ mm}$) at $25 \text{ }^\circ\text{C}$ for 10 min, rinsed with deionized water ($35 \text{ }^\circ\text{C}$, $10 \text{ ml} \times 4$ times), and dried at $60 \text{ }^\circ\text{C}$. The wearing durability tests were carried out on a Martindale-type fabric abrader (HZ-8029A, China). The abrasive material was pristine cotton fabric, and the loading pressure was 12 kPa. The abrasion rotation rate of the top pristine cotton fabric ($D = 120 \text{ mm}$) against the modified fabric surface ($D = 20 \text{ mm}$) was 50 rpm. The number of rotation cycles required for exposing the cellular structure of the fabric was noted as the index of abrasion durability. After every 200 (or more) abrasive cycles, the modified cotton surface was ironed by an electric iron (iron surface temperature was about $200 \text{ }^\circ\text{C}$) for 1 min.

Water absorptivity was examined by weighing the weight increment of the cotton sample after soaking

in plenty of deionized water for 10 min and hanging it out for another 10 min. It is expressed as the weight ratio of adsorbed water to the cotton sample. Water vapor permeability was evaluated using the ASTM E-96 (open cup test) method. The test fabric sample was placed tightly over a shallow dish containing distilled water. The weight loss of the test assembly over 24 h was measured, and the vapor transmission rate ($\text{g m}^{-2} \text{ d}^{-1}$) was calculated as water vapor permeability. Flexibility was determined by the flat loop method (IS 7016 Part 11). Fabric samples were cut from warp and weft directions ($40 \text{ mm} \times 160 \text{ mm}$). A loop was made and placed on a horizontal plane. The height of the loop was measured as an idea of the flexibility of the fabric. The lower the height of the loop, the greater is the flexibility.

Results and discussion

Fabrication of the superhydrophobic cotton surfaces

As shown in Scheme 1, the superhydrophobic surface was obtained through a one-pot mist copolymerization procedure. The mixture solution, consisting of a free radical initiator (TBPB) and three monomers (LMA, IEM, and EGD), was atomized and fed to the cotton surface. The liquid droplets in the mist stream are small (diameter, 150–500 nm), and only a few percent of them can be fixed on the cotton surface (Wang et al. 2014; Wan et al. 2014). Distinguished from other spray-coating processes (Hwang et al. 2011; Sparks et al. 2013) by these features, mist polymerization is beneficial to fabricating a very thin layer with complex morphology on the substrate surface. The three monomers have their respective roles in the surface modification. LMA, a methacrylate monomer having a long aliphatic chain, is chosen to lower the surface energy. EGD, a difunctional monomer, is used as a crosslinker to improve porous structures. IEM, a methacrylate monomer having an isocyanate group, is designed to react with the hydroxyl groups of cellulose. The copolymerization on the cotton fabric surface was monitored using ATR-FTIR. Figure 1 shows the ATR-FTIR spectra of the pristine cotton fabric sample and the modified samples with different monomer proportions. The three characteristic peaks

Scheme 1 Mist copolymerization on the cotton fiber surface. In the resulting random copolymer chain, the $-N=C=O$ group (IEM) reacts with a hydroxyl group of cellulose, the dodecyl chain (LMA) lowers the surface energy, and the double bond (EGD) results in a crosslinkage with another copolymer chain

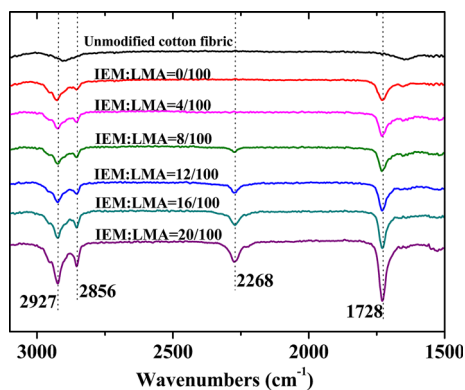
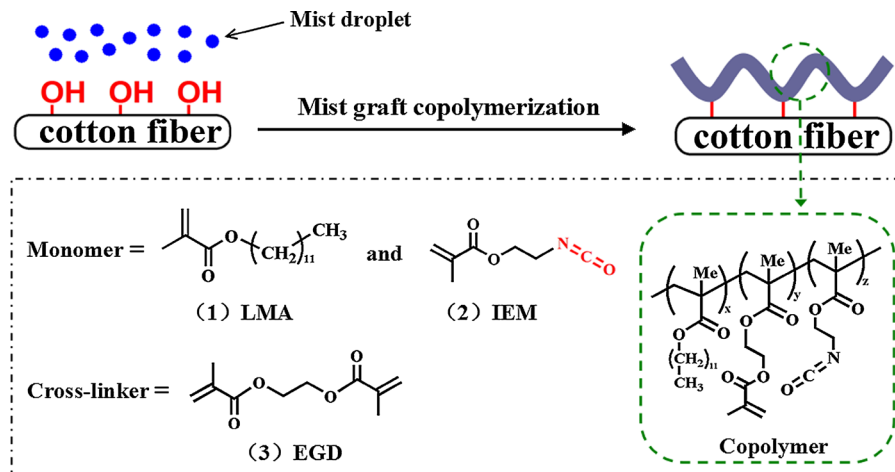


Fig. 1 IR-ATR spectra of pristine cotton fabric and the modified cotton fabrics with different monomer ratios. Modification conditions: LMA, 0.700 mol l⁻¹; EGD, 0.014 mol l⁻¹; mist feeding, 1 min

at 2927, 2856, and 1728 cm⁻¹, which are attributable to C–H₂ and C=O bonds, respectively, are found in all the spectra of the modified cotton fabrics. The peak at 2268 cm⁻¹, which is assigned to the isocyanate group, is not observed in the spectrum of the unmodified fabric. The intensities of the peaks in the spectra of the modified cotton fabrics decrease with decreasing IEM concentration, indicating that there are excessive isocyanate groups remaining in the coating layer after the copolymerization process.

XPS spectroscopy was employed to better understand the interfacial chemical composition of the modified samples. As shown in Fig. 2, the modified fabric sample exhibits C1s, N1s, and O1s peaks (Fig. 2a), but the pristine cotton fabric sample only displays two typical binding energy peaks for C1s and

O1s at 284 and 532 eV (Fig. 2c). Furthermore, the deconvoluted C1s spectrum (Fig. 2b) of the modified fabric sample is different from that of the pristine cotton (Fig. 2d). The peaks at 285.9 and 288.7 eV, which are assigned to C=O and C–N bonds, respectively, arise only in the modified cotton sample. Compared to Fig. 2d, the decrease of the C–OH peak and appearance of the C–N peak in Fig. 2b suggest that the isocyanate reacts with the hydroxyl groups at the interface of the cotton fiber.

The surface morphologies of the modified cotton surfaces were investigated using FE-SEM. Figure 3 shows the low- and high-magnification SEM images for the modified fabrics with a variety of IEM dosages. Low magnification SEM images of the original and modified cotton fibers, shown respectively in Fig. 3a, b, exhibit no significant morphological difference on their fabric surface. The statistic diameters calculated from the SEM images are 14.7 ± 1.6 and 15.0 ± 1.0 μm for the pristine and modified cotton fibers, respectively, meaning that the copolymer layer is very thin. High-magnification SEM images of the modified fabrics (Fig. 3c–f) display the presence of wrinkled structures (300–600 nm) on the fiber surface. The surface roughness of the modified fibers increase with increasing IEM dosage, suggesting that the wrinkle structures may be caused by the combination of the precipitation of the networked polymer and reaction between the isocyanate and hydroxyl groups of cotton. Undoubtedly, the fine structures are helpful for improving the surface superhydrophobicity.

The surface wetting behavior of the modified fabrics was evaluated by measuring the static water

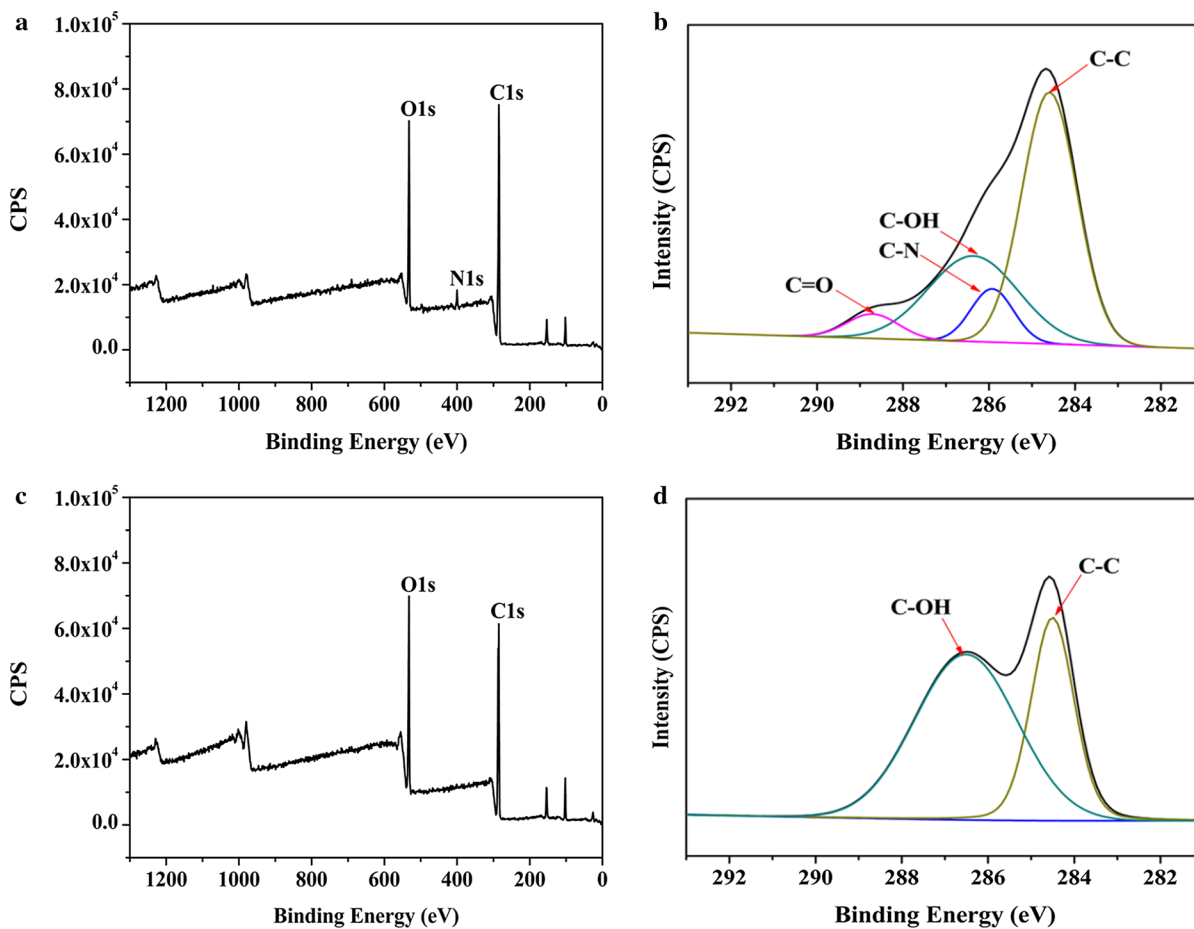


Fig. 2 XPS survey spectra and deconvoluted C1s XPS spectra of modified cotton fabric (a, b) and the pristine cotton fabric (c, d), respectively. Modification conditions: LMA, 0.700 mol l^{-1} ; IEM, 0.084 mol l^{-1} ; EGD, 0.014 mol l^{-1} ; mist feeding, 1 min

contact angle. Figure 4a shows the relationship between the water contact angle and mist feeding time for each of the IEM loading levels. For the mist solution containing 0.028 mol l^{-1} IEM (IEM/LMA = 4 mol%), the static WCA increased steadily upon increasing the mist feeding time, trending to a constant value of 151° . At other IEM loadings, the WCA increased first and then slowed down to being almost constant, and their maximum WCA values were obtained within 30–60 s. The best superhydrophobicity, WCA of 157° , was obtained at the optimized condition, IEM/LMA = 12 mol%, 1 min of the mist feeding time. For original cotton fabrics, the capillary action generally drives the water droplets to move through the fabric from one face to the other side. In contrast, the modified fabrics show

stable superhydrophobicity as shown in Fig. 4b. The WCA values decreased by $<7\%$ over 30 min. This result indicates that the inherent capillary action of the original cotton fabrics is weakened by the mist copolymerization.

The optical observations in Fig. 5 also show the asymmetrical superhydrophobicity of the modified fabrics. The water droplet (stained by blue ink) quickly disappeared on the unmodified face, whereas another water droplet placed on the modified side kept the spherical shape and exhibited a high contact angle of over 150° (Fig. 5a). A similar superhydrophobic behavior was observed on the modified surface with a variety of water droplets stained by other inks or coffee (Fig. 5b). Figure 5c, d shows the modified cotton fabrics immersed in water. The modified

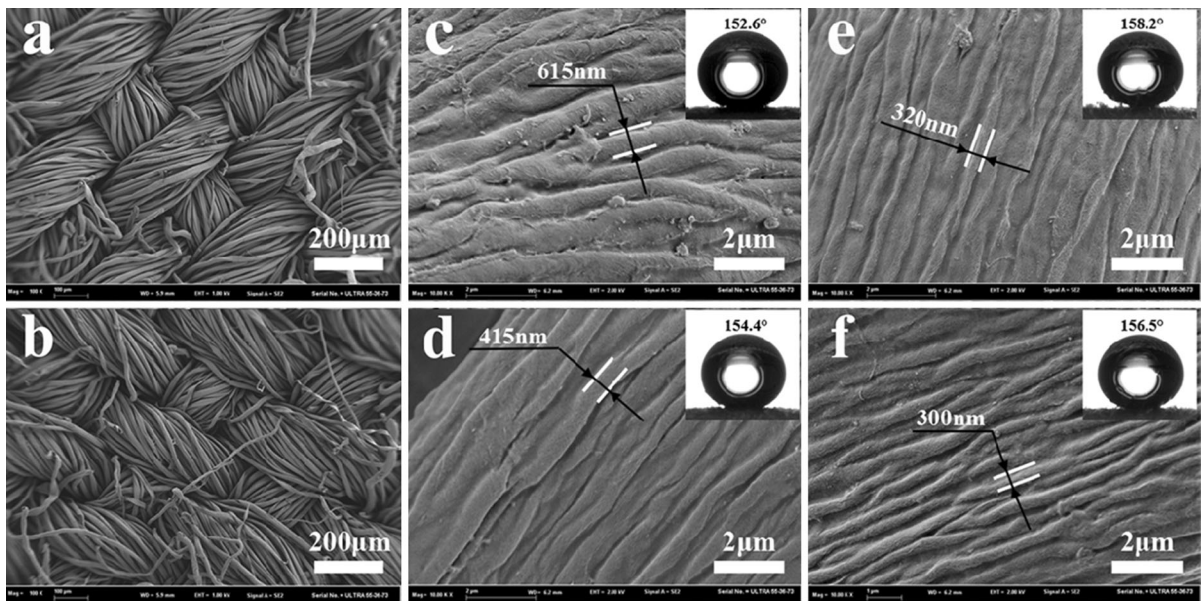
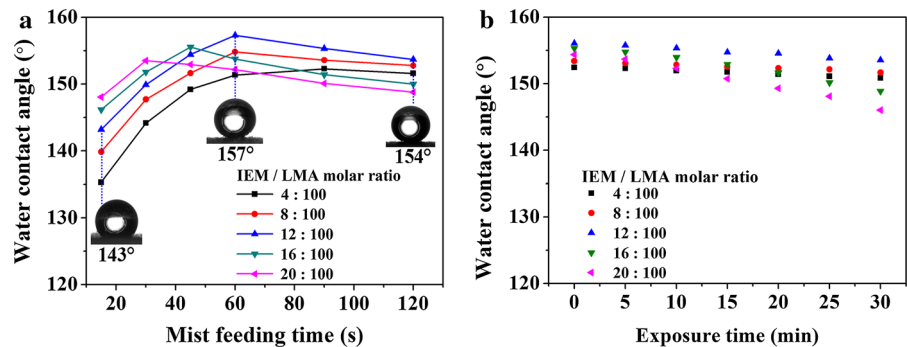


Fig. 3 Low magnification SEM images of **a** pristine and **b** modified cotton fabrics (IEM/LMA = 12/100); high magnification SEM images of the modified cotton fabrics with

different IEM/LMA ratios: **c** 4/100, **d** 8/100, **e** 12/100, and **f** 16/100. Modification conditions: LMA, 0.700 mol l^{-1} ; EGD, 0.014 mol l^{-1} ; mist feeding, 1 min

Fig. 4 WCA values on the modified surfaces against the mist feeding times of the mist stream (**a**); weak dependency of the WCA on the exposure time (**b**). Modification conditions: LMA, 0.700 mol l^{-1} ; EGD, 0.014 mol l^{-1} ; mist feeding, 1 min



surface remained dry with many air bubbles, but the non-modified fabric was fully wetted. Figure 5e, f compares the wetting results of the modified and unmodified surfaces after a dynamic water repellency test (video is provided in Supporting Information Video S1 and 2). The fabric sample was placed 1.5 cm below the nozzle and at 45° from the horizontal, and liquid coffee was continuously dropped on the fabric for 30 s. After the coffee droplets rolled down to the fabric, almost no dirt was stuck on the modified side (Fig. 5e), but an obvious dirt trace was left on the unmodified side (Fig. 5f).

Wearability, laundering durability, and healability of the superhydrophobic surfaces

A Martindale-type abrasion test was employed to evaluate the durability and the healing ability of the modified fabrics. As shown in Fig. 6a, the WCA of the modified fabric decreases by several degrees every 200 abrasive cycles, but the loss can be regained mostly by the ironing treatment at 200°C for 1 min. The decreased amplitude of the WCA for each 200 abrasion cycles increases with increasing abrasion cycle numbers; however, the repeated ironing

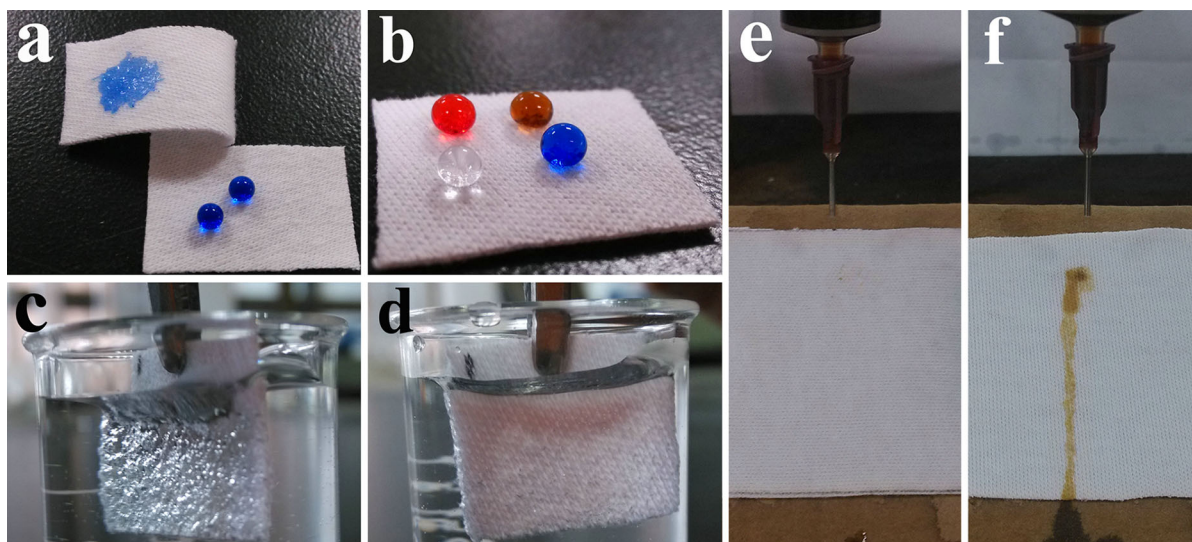


Fig. 5 Optical images of the asymmetrically superhydrophobic cotton fabrics, including stained water droplets on the cotton fabric surfaces with or without modification (**a**), different droplets (water, coffee, and stained water) on the modified cotton surface (**b**), the modified (**c**) and unmodified (**d**) cotton

surfaces in water, and the wetting results after a dynamic water repellency test of the cotton surfaces with (**e**) or without (**f**) modification. Modification conditions: LMA, 0.700 mol l^{-1} ; IEM, 0.084 mol l^{-1} ; EGD, 0.014 mol l^{-1} ; mist feeding, 1 min

processes made the recovered WCA surpass 150° even up to 1800 cycles. In contrast, the copolymer coating without IEM loading was not durable toward abrasion, as shown in Fig. 6c. The WCA rapidly declined from 154° to 122° at 1000 abrasion cycles, and the WCA loss was not recoverable. Additional results are provided in Fig. 6b to show the effect of the healing frequency on the abrasion durability of the modified cotton fabrics. After 2000 abrasion cycles, the modified fabric healed three times (Fig. 6b), showing a higher WCA compared with the fabric healed ten times (Fig. 6a). This result suggests that the frequent healing treatments may reduce the abrasion durability and even the healing ability of the modified cotton fabrics.

The laundering durability was assessed by a stringent washing process based on AATCC Test Method 61-2003. Figure 7 shows the WCA of the modified cotton fabrics after a number of laundering cycles. Sample 1 was the cotton fabric modified using LMA, EGD, and IEM monomers, while sample 2 was the control without IEM loading. As shown in Fig. 7a, sample 1 kept the WCA almost constant over 30 washing cycles, but the WCA of sample 2 linearly decreased with increasing washing cycle numbers

(Fig. 7c). Similar to the previous work (Wang et al. 2014), the improved laundering durability of sample 1 is contributed to the addition of the IEM monomer, which can increase linkages between the grafting polymer and cotton fibers by reactions between the isocyanate groups and the hydroxyl groups in cellulose chains. Although the ironing treatment recovered the WCA of both sample 1 and 2 by several degrees (Fig. 7b, d), it could not make sample 2 superhydrophobic ($\text{WCA} > 150^\circ$).

To investigate the healing mechanism of the superhydrophobicity, FE-SEM and ATR-FTIR were employed to explore the morphological changes and chemical reactions caused by the ironing treatment, respectively.

SEM images show that the pristine cotton fiber has a smooth surface (Fig. 8a), whereas the modified fiber exhibits a nanoscale wrinkled surface (Fig. 8b), which produces a high water repellency ($\text{WCA} = 156.3^\circ$). After 1600 abrasion cycles, the wrinkles were almost removed from the fiber surface as shown in Fig. 8c, but several scale-like fragments, which are guessed to be the disintegrated parts of the copolymer layer, remained. This relatively smooth fiber surface corresponds to the lowered WCA of 143.6° . Remarkably,

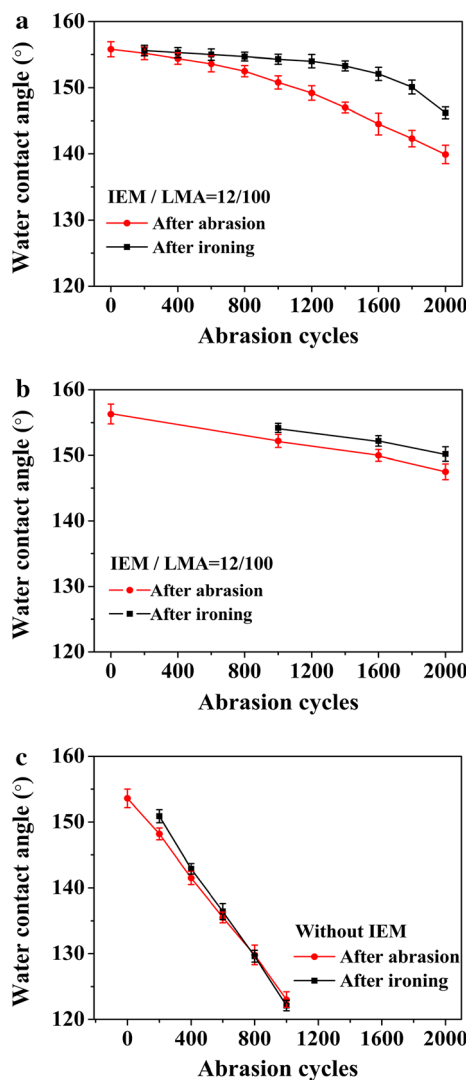


Fig. 6 Healing behaviors of the modified cotton fabrics with (a, b) or without (c) IEM loading. The total ironing treatments for samples a and b were ten and three times, respectively. Modification conditions: LMA, 0.700 mol l^{-1} ; EGD, 0.014 mol l^{-1} ; mist feeding, 1 min

the smooth fiber surface wrinkles again after the ironing treatment (Fig. 8d), and the WCA was recovered to 152.5° .

Figure 9 shows the ATR-FTIR spectra of the four samples for each step in the abrasion-healing process, revealing the interesting result that the peak at 2268 cm^{-1} disappears after the ironing treatment (Fig. 9d). This suggests that a considerable number of residual isocyanate groups existed in the polymer

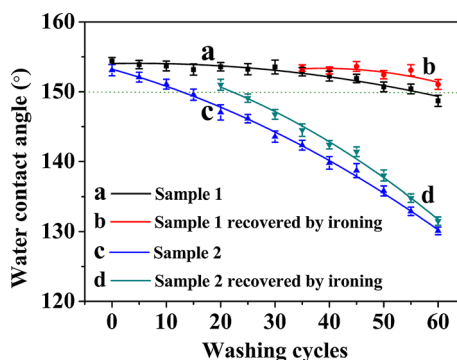


Fig. 7 Laundering durability of the modified cotton fabrics. Modification conditions: LMA, 0.700 mol l^{-1} ; IEM, 0.084 mol l^{-1} ; EGD, 0.014 mol l^{-1} ; mist feeding, 1 min

coating after the abrasion tests, but finally converted to other bonds via some heating reactions. As shown in Scheme 2, isocyanate may reversibly react with the urethane group to form an allophanate (Scheme 2a) or irreversibly react with other isocyanate groups to produce heterocyclic isocyanurate compounds (Scheme 2b) (Delebecq et al. 2013) at an elevated temperature. The first reaction may cause the damaged polymer coatings to wrinkle again, while the last one is likely to consume the reversible groups such as allophanate. In summary, the isocyanate group plays important roles for improving the abrasion durability of the modified cotton fabrics. It not only increases the linkages between the copolymer layer and cotton surface by the reactions with the hydroxyl groups of cellulose, but also endows the modified fabric with healing ability by reactions like in Scheme 2. Besides the chemical reactions, physical realignment of the copolymer chains is another factor involved in the healing behaviors of the modified coating. When the coating was heated (PLMA, $T_g < 0^\circ \text{C}$) (Chatterjee and Mandal 2006), the long side chains of PLMA could rapidly turn to cover the loss of the aliphatic chains.

Influences on the intrinsic properties of cotton

The balance between new functionalities and the desired cotton natures is important for finished cotton fabric products. Therefore, the influence on flexibility, water absorptivity, and vapor permeability of the modified fabrics was examined. Water vapor

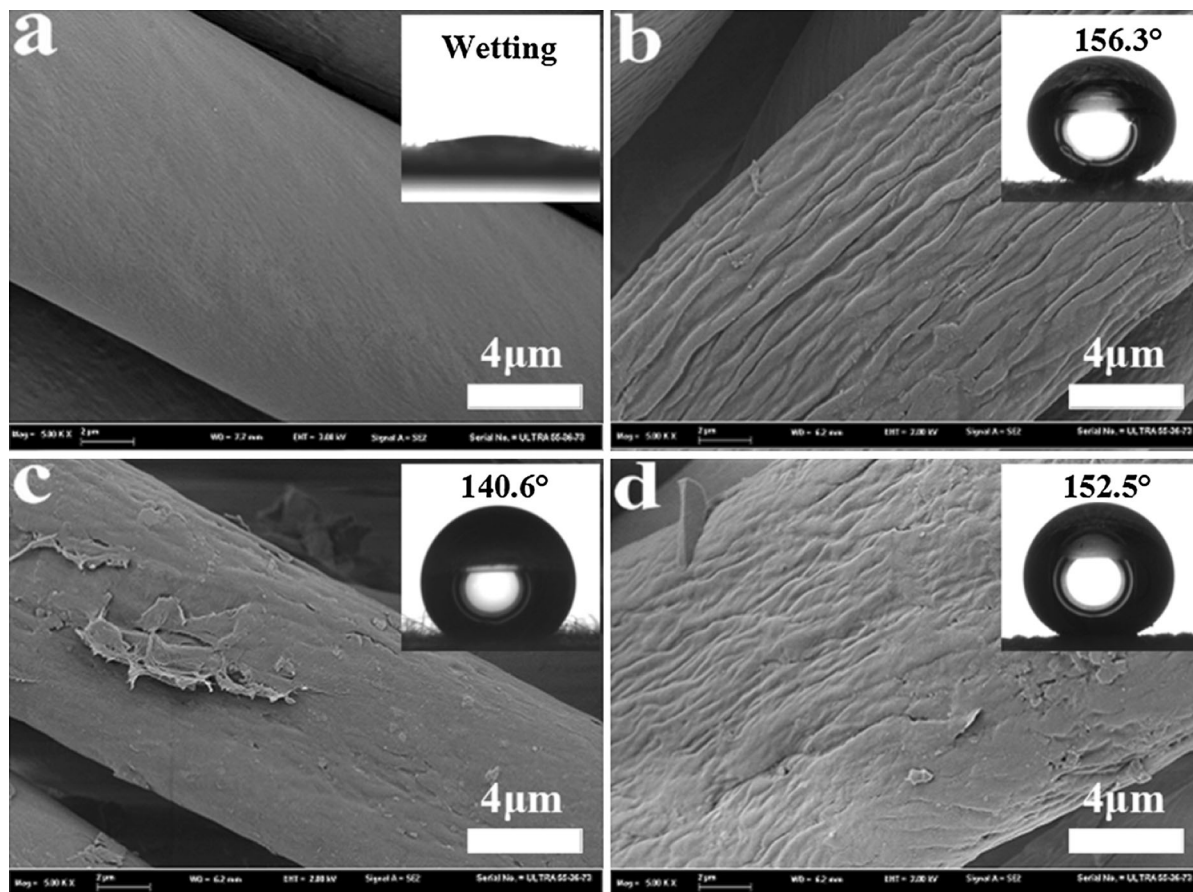


Fig. 8 SEM images of **a** pristine cotton fabric, **b** the modified cotton fabric before the abrasion test, **c** the modified cotton fabric after 1600 abrasion cycles, and **d** the modified cotton

fabric after 1600 abrasion cycles and an ironing treatment. Modification conditions: LMA, 0.700 mol l^{-1} ; IEM, 0.084 mol l^{-1} ; EGD, 0.014 mol l^{-1} ; mist feeding, 1 min

transmission rates were tested at $25 \text{ }^\circ\text{C}$ and 50 \% RH for 1 day. As shown in Fig. 10a, the vapor transmission rate of the fully modified (dipping method) cotton fabrics is $708 \pm 30 \text{ g m}^{-2} \text{ d}^{-1}$, only about 62 % of that of pristine cotton fabric, $1145 \pm 35 \text{ g m}^{-2} \text{ d}^{-1}$. Comparatively, the cotton fabrics modified by the mist copolymerization process show acceptable vapor transmission rates above $1000 \text{ g m}^{-2} \text{ d}^{-1}$, which is still satisfactory for practical use.

Figure 10b shows the water absorptivity of the modified cotton fabrics. The pristine cotton fabrics are of good water absorptivity above 270 %, but the fully modified cotton fabrics are poor at 75 %. In contrast, the cotton fabrics modified by mist polymerization show a medium-level water absorptivity ranging from 260 to 175 %, which is slightly lower than that of the original cotton fabric. These results suggest that the cotton fabrics with a single-sided modified surface

keep a large part of the excellent water absorptivity of cotton. For most clothing products, the desired water absorption can lower wetting by sweat drops, thereby being pleasant for the wearer.

The mechanical properties of the cotton fabrics were also studied by measuring the breaking tensile strength. As shown in Fig. 10c, the pristine cotton fabric has a general breaking strength of 17.7 MPa, whereas the modified cotton fabrics are lightly strengthened by the copolymerization process. The test results for the cotton fabrics modified via mist polymerization and the solution dipping method are 18.9 and 22.3 MPa, respectively.

Figure 10d compares the flexibilities of the modified cotton fabrics and pristine cotton fabric. The original cotton fabric exhibited a good flexibility, as the height of the loop less than 10.2 mm, whereas the fully modified cotton fabric revealed a large loop

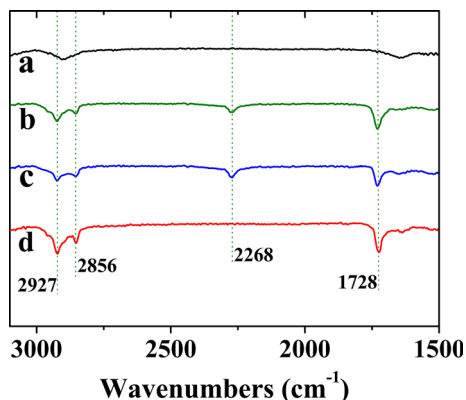


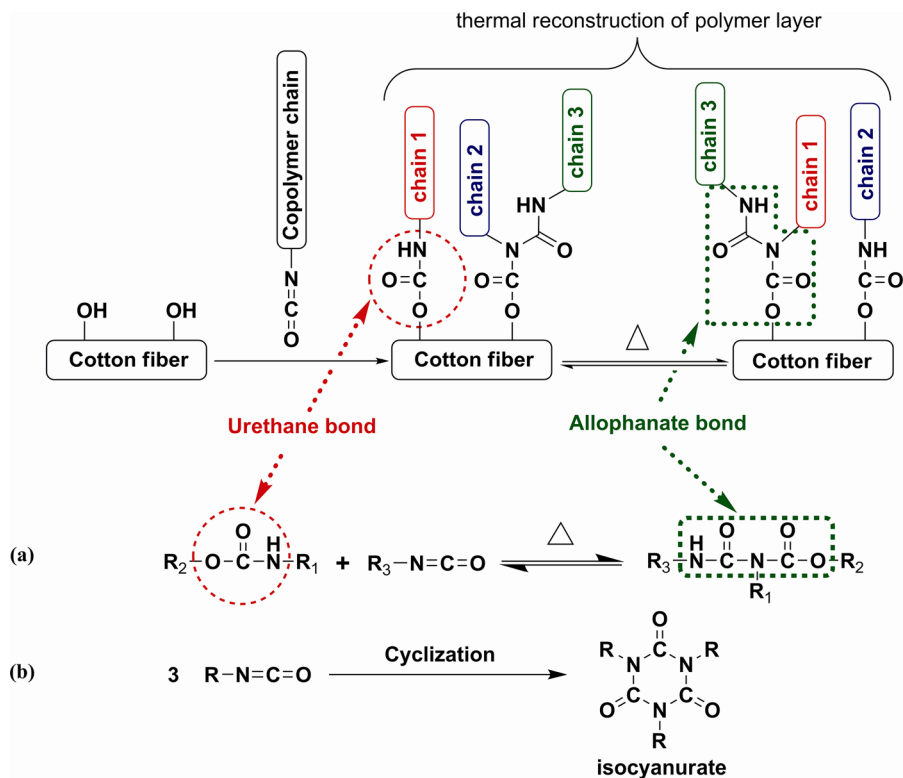
Fig. 9 IR-ATR spectra of **a** pristine cotton fabric, **b** the modified cotton fabric before the abrasion test, **c** the modified cotton fabric after 1600 abrasion cycles, and **d** the modified cotton fabric after 1600 abrasion cycles and an ironing treatment. Modification conditions: LMA, 0.700 mol l⁻¹; IEM, 0.084 mol l⁻¹; EGD, 0.014 mol l⁻¹; mist feeding, 1 min

height of more than 17.0 mm. The fabric modified by the mist polymerization showed a small loop height of 10.8 mm, which is similar to that of the original cotton fabric.

Conclusion

We have demonstrated a one-pot mist copolymerization technique that enables fabrication of healable and asymmetrically superhydrophobic cotton fabrics. With sufficient IEM loading, a randomly wrinkled morphology exhibiting nanoscale roughness is formed on the cotton fiber surface as shown by SEM. The combination of the hydrophobic composition of PLMA and the fine wrinkles endows the modified surface with superhydrophobicity as the water contact angles are greater than 150°. The superhydrophobic copolymer layers are abrasion resistant, laundering-durable, and healable by a simple ironing treatment. Notably, the one-pot mist copolymerization process does not significantly affect the original cotton characteristics such as flexibility, water absorptivity, and vapor permeability. Considering the excellent balance of the new functionalities with the intrinsic natures of cotton fabrics, we believe the process has potential for fabricating superhydrophobic surfaces for textiles and other industrial fabrics.

Scheme 2 Possible chemical mechanisms of the thermal reconstruction of the copolymer layer during the ironing treatment



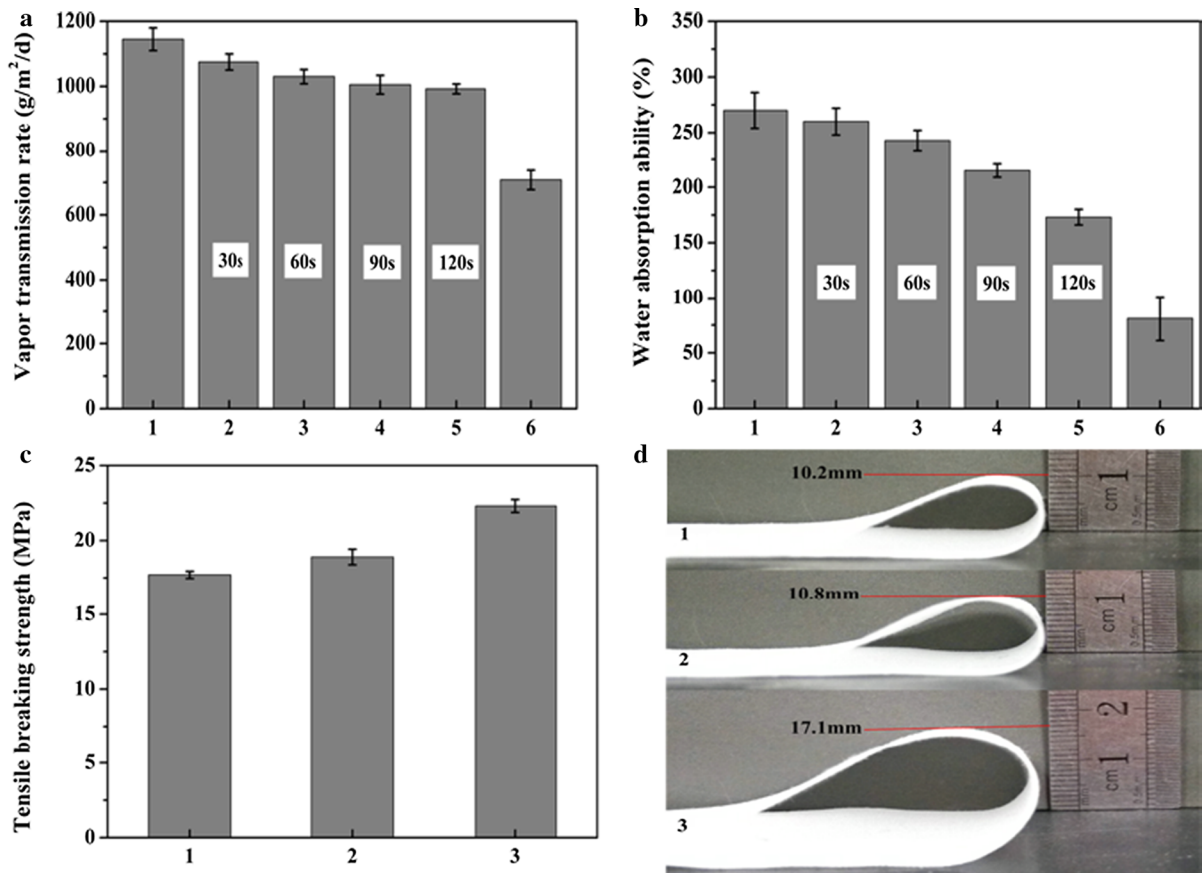


Fig. 10 Influence of the surface modification on several cotton natures, including **a** water vapor transmissibility, **b** water absorbability, **c** tensile strength, and **d** flexibility. In graphs (a) and (b), the bars refer to the pristine cotton fabric (1), the modified cotton fabrics with different mist feeding time (2–5), and the fully modified cotton fabric (6), respectively; in graph

(c) and photo (d), the numbers refer to the pristine cotton fabric (1), the modified cotton fabric with mist feeding time of 1 min (2), and the fully modified cotton fabric (3). Modification conditions: LMA, 0.700 mol l⁻¹; IEM, 0.084 mol l⁻¹; EGD, 0.014 mol l⁻¹

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