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Morphology and properties of cellulose/silk fibroin blend fiber prepared with 1-butyl-3-methylimidazolium chloride as solvent

Yongbo Yao • Enjie Zhang • Xiaolin Xia • Jinchao Yu • Kaijian Wu • Yumei Zhang • Huaping Wang

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Abstract 1-Butyl-3-methylimidazolium chloride ([BMIM]Cl) was selected as co-solvent to dissolve cellulose and silk fibroin and the cellulose/silk fibroin blend fibers were fabricated with dry-jet wet spinning technology. The phase morphology of cellulose and silk fibroin in the blend fibers was studied by scanning electron microcopy and laser scanning confocal microscope. It is shown that the cellulose is in the continuous phase and silk fibroin exists as ''fibril-like'' in cellulose, in which the radial dimension of silk fibroin phase is $0.5-1.0 \mu$ m. The phase size of silk fibroin along the fiber axis increased with the increase of silk fibroin content and draw ratio. From the wideangle X-ray scattering, it is found that the total crystallinity of the blend fibers decreased with increasing silk fibroin content. The hydrogen bond between cellulose and silk fibroin was observed from Fourier transform infrared spectra. Although the tensile strength and initial modulus of blend fibers decreased with increasing silk fibroin content, the tensile strength of blend fibers contain 35 wt% silk fibroin was up to 191 MPa.

State Key Laboratory for Modification of Chemical Fibers and Polymer Materials, Donghua University, Shanghai 201620, China e-mail: zhangym@dhu.edu.cn

Keywords Cellulose - Silk fibroin - Ionic liquid - Blend fiber - Phase morphology

Introduction

Both cellulose and protein are abundant renewable resources, so the utilization of cellulose and silk fibroin has gained widely attention. The regenerated cellulose fiber such as viscose fiber and Lyocell fiber has been comprehensively developed (Fink et al. [2001\)](#page-9-0). Protein material is biocompatible and nonirritation of human skin (Koller et al. [2007](#page-9-0)). However, the utilization of protein resource is still under various restrictions (Heslot [1998\)](#page-9-0). The regenerated protein fibers have limitation as low strength (Um et al. [2001;](#page-10-0) Ki et al. [2007\)](#page-9-0) or inefficiency (Zhou et al. [2009](#page-10-0)), which is caused by the severe degradation of protein during dissolution process (Vollrath et al. [2007](#page-10-0)). In order to improve the mechanical property of the regenerated protein fiber, the commercial regenerated protein fibers are prepared by grafting to synthetic polymer or dispersing in spinning dope. One of the commercial regenerated protein fiber is polyacrylonitrile-g-milk protein fiber which has excellent feeling and luster like silk. But the price of such fiber is very high (Morimoto [1970](#page-9-0)). Another protein grafting fiber is poly(vinyl alcohol)-g-soybean fiber which has disadvantages as pilling, not resistance to laundering and poor dimensional stability (Zhang et al. [2003\)](#page-10-0). Liu

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et al. [\(2009](#page-9-0)) dispersed silk protein powders into polyurethane/dimethyl formamide dope and prepared blend fibers by wet spinning. However, the particle sizes of silk protein powders are too large for fiber forming and the dispersion of silk protein in the blend fiber is not good which makes it easy to fall off.

To make full use of the natural renewable resources, cellulose/protein blend fibers which have excellent miscibility and mechanical property can be expected to be prepared with solution blending method. Hirano et al. [\(2002](#page-9-0)) mixed the 10 % aqueous silk fibroin solution and 9 % viscose solution together to fabricate cellulose/silk fibroin blend fiber by wet spinning. The strength of the fiber is low (less than 1.5 cN/dtex) which is caused by the instability of silk fibroin aqueous LiBr/viscose solution mixture. As we know, the performance of polymer blends is strongly related to the miscibility of polymer components (Macosko [2000](#page-9-0)). Co-solvent, which can dissolve protein and other components under the same condition, is an effective method to improve the miscibility of protein blends. N,N-dimethylacetamide/LiCl was used as co-solvent for cellulose/silk fibroin blend fiber, the miscibility between cellulose and silk fibroin was enhanced (Marsano et al. [2007](#page-9-0)). Chen and Zhang [\(2004](#page-9-0)) dissolved cellulose and soybean protein with NaOH/thiourea aqueous solution as co-solvent and fabricated cellulose/soy bean blend membranes. The phase morphology of soybean protein and cellulose was studied by transmission electron microscope (TEM). The size distribution of soybean protein is about 50–100 nm. Marsano et al. [\(2008](#page-9-0)) made cellulose/silk fibroin blend membranes and fibers with Nmethylmorpholine N -oxide hydrates (NMMO/H₂O) as co-solvent. The phase size of silk fibroin is $2-10 \mu m$ in the blend membranes. Ionic liquid has also been found to be able to dissolve cellulose and silk fibroin under the same condition (Phillips et al. [2004](#page-9-0); Swatloski et al. [2002\)](#page-10-0). Shang et al. ([2011\)](#page-10-0) prepared cellulose/silk fibroin blend membranes using [BMIM]Cl as solvent. The surface morphology of the blend membranes was studied by atomic force microscopy. The surface morphology became smoother and the strength of the blend membrane was improved with the increase of cellulose content. The interaction between silk fibroin and cellulose in the blend has also been comprehensively discussed by Fourier transform infrared spectra (FTIR) (Shang et al. [2013\)](#page-10-0). Otherwise, the mechanical strength of cellulose/silk fibroin blend membrane can be enhanced by improved coagulation technology as vapored methanol and cold pressing during vacuum dry (Zhou et al. [2013\)](#page-10-0).

In our past work, the rheological behaviors of cellulose/silk fibroin solution using 1-butyl-3-methylimidazolium chloride ([BMIM]Cl) as co-solvent were studied. The change of microstructure in the solution was imaged according to the derivations of viscosity, dynamic modulus and shear rate. The phase morphology of cellulose and silk fibroin deduced from the solution property was agreed with the result from morphology of regenerated cellulose/silk fibroin film (Yao et al. [2014\)](#page-10-0). As the continuous work, the regenerated cellulose/silk fibroin blend fibers were fabricated with [BMIM]Cl as co-solvent. The effect of cellulose/silk fibroin blend ratio, draw ratio on the phase morphology, structure and property of blend fibers were discussed.

Experiments

Materials

Cellulose pulp [degree of polymerization $(DP) = 500$] was supplied by Shandong Silver Eagle Chemical Fiber Co. Ltd, China. Silk fibroin was degummed from silk cocoon in boiling 0.5% (w/w) Na₂CO₃ solution with a bath ratio of 1:50 and then washed with deionized water to remove silk sericin (Cao et al. [2009a\)](#page-9-0).

1-Butyl-3-methylimidazolium chloride ([BMIM]Cl) was synthesized and purified in our laboratory as previously described (Csihony et al. [2002](#page-9-0)).

Solution preparation

The 8 wt% concentration of cellulose/silk fibroin/ [BMIM]Cl solution were prepared by changing the weight ratio of cellulose to silk fibroin as 10/0 (Cell), 8/2 (CS82), 6/4 (CS64), 0/10 (SF). Cellulose and silk fibroin were dried under vacuum at 80 $^{\circ}$ C to remove water. Spinning solutions were obtained by kneading at 90 \degree C.

Spinning trials

The cellulose/silk fibroin blend fiber was spun through a laboratory-scale dry-jet wet spinning equipment. The cellulose/silk fibroin/[BMIM]Cl solution with 8 wt% was degassed and extruded through the spinneret orifices of 0.15 mm in diameter at 85 $^{\circ}$ C. The air gap distance is 50 mm and the air temperature is 25 \degree C. The coagulation bath was ethanol at the temperature of 10 $^{\circ}$ C. The blend fibers were stretched in the coagulation bath. The possible ranges of drawing ratios for this blend fibers is 1–5 according to the results of our experiments. The as-spun fiber was further washed in ethanol bath, and then dried and taken up. The spinning condition of the blend fibers are listed in Table 1. The viscosity of pure silk fibroin/ [BMIM]Cl solution is too low to be used for dry-jet wet spinning.

Measurements

The zero shear rate viscosity of the solutions was measured by rotational rheometer (Physica MCR 301, Anton Paar). The 25-mm-diameter concentric parallel plate geometry was utilized. The chosen gap was 1 mm and the chosen temperature was 85° C for all solutions.

The protein content in the blend fibers was obtained by testing nitrogen element (N) content with element analyzer (Elementar Vario EL III, German). The protein content was calculated according to the Eq. (1) (Yang et al. [2002](#page-10-0)):

$$
P_{SF}\% = 5.97N\% \tag{1}
$$

where P_{SF} is protein content, N is the nitrogen content in blend fibers.

The cross-section morphology of the fibers was measured on environment scanning electron microscope (SEM) (Quanta-250, FEI, USA). The fibers were

Table 1 Spinning condition of cellulose/silk fibroin/[BMIM] Cl solutions

Sample name	Cellulose/silk fibroin blend ratio	Zero shear rate viscosity at 85 \degree C (Pa s)	Draw ratio
Cell-3	100/0	147	3.3
CS82-3	80/20	135	3.3
$CS64-3$	60/40	109	3.3
Cell-4	100/0	147	4.3
CS82-4	80/20	135	4.3
$CS64-4$	60/40	109	4.3
Protein	0/100	14	

cryogenically broken under liquid nitrogen and sputtercoated with a thin layer of gold.

The intrinsic fluorescence of silk fibroin in the blend fibers was captured by a laser scanning confocal microscope (LSCM) (TCS SP5, Leica, German), the excitation wavelength is 488 nm and the emission wavelength range is 500–550 nm.

Wide-angle X-ray scattering (WAXS) was measured at Beamline (BL16B) in Shanghai Synchrotron Radiation Facility with the wavelength of 0.124 nm. A bundle of blend fibers were put in a holder with the fiber direction perpendicular to the X-ray beam. The distance between the sample and Mar-CCD (165) was 107.8 mm. The 2D WAXS was processed with the software of Xpolar and Peakfit. The total crystallinity of the blend fibers was calculated according to Eq. (2):

$$
w_{c,x} = \frac{I_c}{I_c + I_a} \times 100\% \tag{2}
$$

where $W_{c,x}$ is the total crystallinity, I_c and I_a are respectively the crystal peak area and amorphous phase area. The crystal orientation f_c of cellulose along the longitudinal axis was calculated as previously described (Jiang et al. [2012b\)](#page-9-0).

FTIR was measured by a Nicolet spectrometer (Nexus 670, Thermo Fisher, USA) at wavenumber of 700–4,000 cm⁻¹ with a resolution of 0.09 cm⁻¹. A bundle of blend fibers were pinched under total internal reflection accessory.

The mechanical property of the fibers was tested on a fiber mechanical strength tester (XQ-2, China). The stretching rate is 20 mm/min with a 20 mm gauge length. The diameter of the fiber was measured using microscope (XSZ–360AP, China). The tensile strength, initial modulus and elongation were calculated as the average of at least 20 measurements from stress–strain curves.

The moisture sorption curves of the blend fibers were determined by the following method. The blend fibers were dried at 105 \degree C to constant weight. The same fibers were then kept under standard condition (20 \degree C, 65 % RH). The mass change of each sample with time was measured to draw sorption curves. The moisture desorption curves were obtained as follows, the blend fibers were conditioned under 20 \degree C, 100 % RH. The same fibers were then kept under standard condition to get the weight change and draw desorption curves.

Results and discussion

Protein content of cellulose/silk fibroin blend fibers

The structure and property of regenerated silk fibroin fiber are greatly affected by coagulation solvents. Water, methanol, acetonitrile were chosen by Phillips et al. [\(2005\)](#page-9-0) as coagulation solvents to prepare regenerated silk fiber. Only the fiber which is coagulated by methanol could be handled. So it is necessary to show that ethanol is the effective coagulation solvent for cellulose/silk fibroin blend fiber from the result of protein content measurement. The nitrogen content was measured by elemental analysis, the protein content was calculated according to Eq. [\(1\)](#page-2-0). As shown in Table 2, there is only a slightly reduction of silk fibroin. Both of cellulose and silk fibroin were coagulated, which proved that ethanol is the effective coagulant for cellulose and silk fibroin. The viscosity of silk fibroin phase is low which might came to the coagulation bath with the solvent [BMIM]Cl together. With the increase of draw ratio, the diameters of the blend fibers decreased, the specific surface area between blend fibers and coagulation bath also decreased, more silk fibroin entered the coagulant with [BMIM]Cl, the weight loss of silk fibroin increased.

Morphology of cellulose/silk fibroin blend fibers

The cross-section morphology of the blend fibers was observed by SEM, as shown in Fig. [1](#page-4-0). The typical fibrous and microporous structure exist in the blend fibers, which are similar as that of regenerated cellulose fiber (Fink et al. [2001](#page-9-0); Cao et al. [2009b](#page-9-0)). These features appeared in all samples and were independent on the blend ratio and draw ratio.

Table 2 Protein content in the cellulose/silk fibroin blend fibers

Protein content $(\%)$	
None	
17.0	
37.2	
None	
15.6	
35.0	

In order to study the phase morphology of cellulose and silk fibroin in the blend fibers, the blend fiber was observed by LSCM. Due to the existence of amino acids, the intrinsic fluorescence spectroscopy of silk fibroin has been found. Georgakoudi et al. ([2007\)](#page-9-0) characterized the intrinsic fluorescence spectra of silk fibroin from the excited range 250–335 nm and emission range 265–600 nm. Rice et al. [\(2008](#page-9-0)) used two photon excited fluorescence and second harmonic generation as non-invasive characterization to study the morphology of silk fibroin in solution, film, hydrogel and scaffold. The obvious silk fibroin morphology was obtained from emission range of 380–700 nm. The detection scale of LSCM varies from micro-scale to millimeter-scale which can be used as a new method to analyze the protein phase morphology in the blend material. As shown in Fig. [2,](#page-5-0) the figures of regenerated cellulose fibers are black, which means the absence of cellulose fluorescence. The green regions show the dispersion of silk fibroin, what exist as "fibril-like" in cellulose phase, in which the radial dimension of silk fibroin phase is $0.5-1.0$ µm. Compare with sample CS64-3 and sample CS82-3, the radial dimension is a little larger with the increase of silk fibroin content. Along the fiber axis, the dispersion of silk fibroin in sample CS82-3 is not continuous. With the increase of draw ratio, the size of silk fibroin phase along the fiber axes become longer, as shown in sample CS82-4.

Crystalline structure of cellulose/silk fibroin blend fibers

As shown in Fig. [3a](#page-6-0), the crystalline structure of regenerated cellulose fibers is cellulose II (Langan et al. [2001](#page-9-0)) with the characteristic reflections (110) , (110), (012), (020), (103) (The corresponding peaks of 2θ at the wavelength of 0.124 nm are respectively 9.7°, 16.1°, 16.4°, 17.8°, 23.0°) (Jiang et al. [2012a](#page-9-0)). The crystalline structure of silk fibroin which is coagulated by ethanol is silk II, the characteristic reflections are (002) and (201/021) (the corresponding peaks are respectively 16.5° , 20.5° at the wavelength of 0.124 nm) (Sun et al. 2012). These peaks were chosen to calculate the total crystallinity of the blend fibers. As shown in Fig. [3b](#page-6-0), c, the total crystallinity of blend fibers and the crystal orientation of cellulose decreased with increasing silk fibroin content.

Fig. 1 SEM of the crosssection of cellulose/silk fibroin blend fibers

Because the ordering of cellulose and silk fibroin deduced owing to the interface of molecular chains (Zhou et al. [2013](#page-10-0)). During drawing process, the crystal orientation of cellulose can be enhanced with drawing. The separated silk fibroin domains are stretched. But the total crystallinity of the cellulose/silk fibroin blend fiber is hard to be enhanced with drawing owing to the low crystalline of silk fibroin (Um et al. [2001\)](#page-10-0).

Molecular structure of cellulose/silk fibroin blend fiber

FTIR has been comprehensively used to study the interaction of polymer blends (Shang et al. [2009](#page-9-0); Lü et al. [2010](#page-9-0), [2011;](#page-9-0) Huang et al. [2013](#page-9-0)). The FTIR of cellulose/silk fibroin blend fiber is shown in Fig. [4.](#page-7-0) The conformation of silk fibroin includes random coil, α -Helix and β -Sheet structure (Drummy et al. [2005](#page-9-0)). The characteristic absorption peaks of β -Sheet silk fibroin are at $1,630 \text{ cm}^{-1}$ (amide I), $1,530 \text{ cm}^{-1}$ (amide II) and $1,260^{-1}$ (amide II) (Freddi et al. [1995;](#page-9-0) Hyoung-Joon et al. [2005\)](#page-9-0). The adsorption peaks at 1,626 and 1,530 cm^{-1} appeared in the blend, which recommends that β -Sheet structure was formed through fiber forming process. The characteristic peak of OH stretching vibration is around $3,400 \text{ cm}^{-1}$. There are three different hydroxyl groups of cellulose chain. The peak of OH stretching vibration is broader in sample Cell-3 and Cell-4, which is contributed to the intramolecular hydrogen bonding and intermolecular hydrogen bonding. The stretching vibration of NH in silk fibroin chain is around $3,288$ cm⁻¹ (Yao et al. [2014\)](#page-10-0). As shown in Fig. [4](#page-7-0), the peak around $3,288$ cm⁻¹ became narrow and the peak at

 $3,400 \text{ cm}^{-1}$ became weaker with the increase of silk fibroin, which recommends that the new intermolecular hydrogen bonds exist between OH at the C2, C3 position of cellulose and NH in the amide groups of silk fibroin (Kondo et al. [1994](#page-9-0); Yang et al. [2000\)](#page-10-0).

Mechanical properties of cellulose/silk fibroin blend fibers

The mechanical properties of cellulose/silk fibroin blend fibers is shown in Table [3.](#page-7-0) The diameter of the fibers are between 20 and $30 \mu m$, which decreased with drawing. The tensile strength (σ_b) and initial modulus (E) increased with increasing cellulose content. Cellulose is the continuous phase in the blend which contributes much to the mechanical strength. The dispersion of silk fibroin phase in the cellulose matrix component blend fibers is similar with increasing silk fibroin content. Hence, the tensile strength of the blend fiber decreased little from sample CS82 to CS64. The tensile strength increased with the drawing ratio, owing to the increase of crystal orientation of cellulose. The tensile strength of blend fiber which contains 35 wt% silk fibroin was still up to 191 MPa.

Fig. 3 2D WAXS (a), total crystallinity and crystal orientation of cellulose (b, c) of cellulose/silk fibroin blend fibers (silk fibroin crystal planes are distinguished in *italics*)

Moisture absorption and desorption of cellulose/ silk fibroin blend fibers

The moisture absorption and desorption are related to the wearing comfort to human skins. As shown in Fig. [5,](#page-8-0) the moisture absorption and desorption process occur mainly within 120 min, the equilibrium moisture absorption regain of blend fiber is 11.8–13.1 % and the equilibrium desorption regain is 15.0–17.0 %. The equilibrium moisture of the blend fibers is not only related to the hydrophilic of cellulose and silk fibroin chains, but also the amorphous regions, crystallites and voids of the blend fibers (Okubayashi et al. [2004\)](#page-9-0). The standard moisture absorption of degummed silk is 8.9 % (Charles [1938](#page-9-0)). The equilibrium moisture regain of Lyocell fiber at 60 % relative humidity and 20 \degree C is 9.2 % (Siroka et al. [2008](#page-10-0)). The regenerated cellulose fiber and silk have similar moisture ability. Hence, the moisture absorption and desorption regains of cellulose/silk fibroin blend fibers and regenerated cellulose fiber are almost same in this research.

Conclusions

The cellulose/silk fibroin blend fibers were fabricated with [BMIM]Cl as solvent by co-dissolution method. Even though the tensile strength and initial modulus decreased with increasing silk fibroin, the tensile strength of blend fibers containing 35 wt% of silk fibroin was up to 191 MPa. This is related to ''fibrillike'' morphology of silk fibroin existed in the continuous phase of cellulose. The phase size of silk fibroin along the fiber axis increased with silk fibroin Fig. 4 FTIR of cellulose/silk fibroin blend fibers content and draw ratio, and the hydrogen bond

Table 3 Mechanical properties of cellulose/silk fibroin blend fibers

Fig. 5 Moisture absorption and desorption of cellulose/silk fibroin blend fibers

interaction between cellulose and silk fibroin in the blend fibers was detected.

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