

Mechanism of structure formation of microbial cellulose during nascent stage

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Abstract The structure of microbial cellulose (MC) produced by *Acetobacter xylinum* was studied in presence of Fluorescent Brightener, Direct Blue 1, 14, 15, 53, Direct Red 28, 75 and 79, as probe. X-ray diffraction pattern of the product showed that it was a crystalline complex of dye and cellulose. The product has the structure in which the monomolecular layer of the dye molecule is included between the cellulose sheets corresponding to the (1 $\bar{1}$ 0) planes of microbial cellulose. As a result of dye inclusion, d-spacing of lower angle plane (100) of products becomes 8.0–8.8 Å instead of 6.1 Å of MC. The d-spacing for the higher angle plane must be (010) plane due to stronger van der Waals forces between the pyranose rings which reduced 5.3 Å space of (110) plane of MC to 3.9–4.5 Å in the product. However, cellulose regenerated from FB, DR28 products was cellulose I and IV, respectively, and that from each DB1, 14, 15, 53, DR75 and 79 products was cellulose II. Solid state ^{13}C NMR and deuteration-IR showed the product was non-crystalline which was contrasted to X-ray results. The regenerated celluloses were cellulose I $_{\beta}$, IV $_1$ and II, respectively. Thus the structure of the product depends on the characteristics of dye which affects the conformation of cellulose at the nascent stage by the

direct interaction with cellulose chains. The different regenerated celluloses as well as different fine structure in the same cellulose allomorph were produced depending mainly on number and position of the sulfonate groups in the dye.

Keywords Microbial cellulose · *Acetobacter xylinum* · Direct dye · Nascent cellulose · Dye-cellulose complex · Regenerated cellulose

Introduction

Cellulose is the most abundant macromolecules produced on earth (Zhao et al. 2007) and most cellulose is produced by vascular plants. Hence, the forest which is the main source of cellulose supply is now on jeopardy. Cellulose produced by an acetic acid bacteria, that is, microbial cellulose (MC) or bacterial cellulose (BC) is expected for an alternative source of cellulose biomass in the near future. MC is an exopolysaccharide produced by various species of bacteria, such as those of the genera *Acetobacter*, *Agrobacterium*, *Aerobacter*, *Achromobacter*, *Azotobacter*, *Rhizobium*, *Sarcina*, and *Salmonella* (Chawla et al. 2009). *Acetobacter xylinum* (or reclassified as *Gluconacetobacter xylinus*) is one of the best bacterial species for the production of bacterial cellulose in large-scale and utilizes a wide variety of substrate (Moosavi-Nasab and Yousefi

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2011). However, four major types of cellulose allomorphs have been reported so far, cellulose I, II, III and IV on the basis of X-ray diffraction patterns (Isogai 1994; Sarko 1978; Marchessault and Sarko 1967; Walton and Blackwell 1973; Marchessault and Sundararajan 1985). Each polymorph yields a recognizable diffraction patterns. Crystal structures of cellulose allomorphs, such as unit cell sizes, packing modes of cellulose chains, conformations at C1, C4 and C6 (rotational angles ϕ , φ and χ , respectively) have been studied from as early as 1920s by X-ray diffractometry, electron microscopy, IR spectroscopy and Raman spectroscopy, and other techniques.

The mechanisms of formation of cellulose I microfibril in nature, and the control of cellulose I microfibril formation by living organisms have been of great interest to the scientists. The microfibril produced in *A. xylinum* is crystallize cellularly (Brown et al. 1976; Zaar 1977, 1979). Arguments for and against various relationships between polymerization and crystallization in microfibril formation in plants and bacteria were based largely on theoretical considerations and indirect evidence until 1979, when it was discovered that some chemicals could be used to alter cellulose microfibril assembly in vivo (Haigler et al. 1980; Haigler and Benziman 1982; Brown et al. 1983). These alternating chemicals, such as fluorescent brightening agents, direct dyes and cellulose derivatives, allowed direct testing of the temporal relationship between polymerization and crystallization in cellulose biogenesis, and provided inside into how organisms regulate the size of microfibrils. The present article will discuss elaborately into how direct dyes have been used to investigate the control of cellulose biogenesis in *A. xylinum* which was recognized as a model organism to date. These alternating agents have therefore allowed elucidation of basic principles that govern the biogenesis of bacterial cellulose as well as its formation mechanisms.

Only a few studies relating to the structure of nascent fibril have been reported to date. Among these investigators, two working groups Kai et al. and Brown et al. are mentionable. Brown and his groups (Haigler et al. 1980; Brown et al. 1983; Haigler and Chanzy 1988) judging from the electron microscopy of products from *Acetacter*-culture with direct dyes reported that the dyes adhered to the surface of cellulose protofibril in a stact state formed non-crystalline product thus, this product formed cellulose

I upon dye-extraction. On the other hand, Kai and his groups (Kai et al. 1990, 1994; Kai 1984a) found that the products from the incubated medium with Fluorescent Brightener, Congo Red and Direct Red 80 showed the characteristic X-ray diffraction diagrams indicating the formation of crystalline complexes, and these complexes reproduced celluloses I_β, IV and II, respectively, after dye-extraction. However, direct dyes have different affinities for cellulose under the experimental conditions, which would be affected by the kind, number and position of substituents on the planar molecular backbone (Kai and Mondal 1996, 1997; Mondal and Kai 1998, 2001).

In this paper, we specially investigate the structure of the nascent fibril produced by *A. xylinum* bacteria using direct dyes as alternating agents. As these alternating agents allow direct testing of the temporal relationship between polymerization and crystallization in cellulose biogenesis, this is very important for the clarification of the mechanism of formation of cellulose fibril. The direct dyes considered in this paper are Fluorescent Brightener (FB), Direct Blue 1, 14, 15, 53 (DB1, DB14, DB15 and DB53) Direct Red 28, 75 and 79 (DR28, DR75, DR79). DR28, DB 1, 14, 15 and 53 have same biphenylene bis(azo) skeletal backbone, and DR75 and 79 have same ureylene-bis(naphthylazo) benzene skeletal backbone. Although each group contains same skeletal backbone, their substituent groups are different. Hence, it is possible to encounter their effect on the product structure as well as on the regenerated cellulose. The structure of the product from *Acetobacter*-culture in the presence of direct dyes and its regenerated cellulose was characterized by X-ray diffractometer, solid state ¹³C NMR and infrared spectroscopies.

Experimental

Materials

Preparation of product sample

Acetobacter xylinum (IFO 13693), the most studied bacterial cellulose (BC) producer, was cultured in a 15 cm diameter Petridish which contains a 100 ml Hestrin–Schramm medium (Hestrin and Schramm 1954) (pH 6.8) at 28 °C for 3 days in an incubator. Cellulose-free bacterial cell suspension was prepared

from a matured Petridish in an ice cooled 120 ml phosphate buffer solution, and 60 ml of the cell suspension was added to 140 ml HS medium (pH 7.0) containing 0.1 wt% direct dye. The mixture was then incubated at 28 °C for 24 h in a static condition or for 12 h in a stirred condition (Kai et al. 1990; Kai and Mondal 1996; Mondal and Kai 1999). The product sample (product) was collected either by centrifuging (6,000 rpm, 5.15 g, 20 min) or by a Tea strainer, and was washed well with 0.1 wt% aqueous sodium hydroxide solution to remove the dye which was not related to dyeing. This was then rinsed well with distilled water and was preserved at about 3 °C in the wet for the next experiments. The name and structure of the direct dyes used in present article are listed in Fig. 1.

Regenerated cellulose from product sample

Extraction of dye from the above product samples was performed by boiling in 70 vol% aqueous EtOH solution for 18 h, and the fresh aqueous ethanol solution was exchanged for every 3 h. This dye extracted sample was again boiled in 1.0 wt% aqueous sodium hydroxide solution for 10 h under N₂ atmosphere to remove proteinous matters. This regenerated cellulose sample was neutralized with 1.0 vol% aqueous acetic acid solution and, subsequently, rinsed with distilled water (Kai et al. 1990; Kai and Mondal 1996; Mondal and Kai 1999). The samples were preserved as above.

Preparation of control cellulose sample

The control cellulose I sample was prepared by washing the MC with distilled water to remove the medium components, and then was boiled in 1.0 wt% aqueous sodium hydroxide solution for 10 h under N₂ atmosphere. The final sample (cellulose I or MC) was neutralized and rinsed as above. Mercerization of MC was carried out at 20 °C for 25 h to prepare control cellulose II sample. The control cellulose IV_{II} was obtained by which cellulose II was annealed in glycerol at 260 °C for 30 min.

Methods for the measurements

In order to understand the structure of the product from *Acetobacter*-culture in the presence of each of FB, DB1, 14, 15, 53, DR28, 75 and 79, and the cellulose regenerated from them, the following measurements were performed.

X-ray diffractometry

The X-ray diffraction diagram of an isotropic product sample and its regenerated cellulose was performed by transmission method. In this method, the dried sample was compressed as a pellet into the pore of a sample holder of 1 mm diameter and 1 mm thickness.

Again, an uniplanar oriented membrane of product sample was prepared on a Teflon plate and was dried at room temperature. Its dry size was 3.0 cm × 2.5 cm and dry thickness was not less than 100 μm. The X-ray diffraction diagram of this uniplanar oriented membrane was measured with the reflection and transmission methods using MXP¹⁸ diffractometer (MAC Science Co, Ltd., Tokyo) with Ni-filtered CuKα radiation. The measurement conditions were as follows: divergence slit, 1.0; receiving slit, 0.15; scanning speed, 4.0° min⁻¹; X-ray radiation, 40 kV, 200 mA.

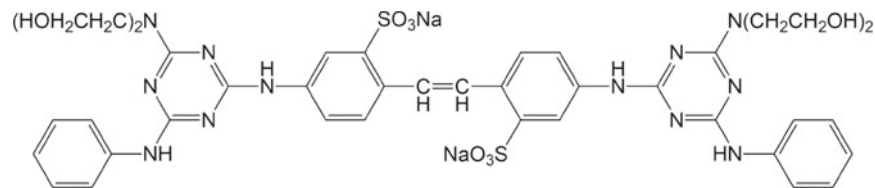
Solid state ¹³C NMR spectroscopy

Solid state ¹³C NMR experiments of the wet samples were performed at room temperature by a JEOL JNM EX 270 spectrometer operating at 6.8 MHz under a static magnetic field of 6.35 T. The mass rates were 5.0–5.5 kHz, and the contact time for the CP process was 1,000 ms throughout this work. The chemical shifts relative to tetramethyl silane were determined by using the crystalline peak at 17.3 ppm of hexamethyl benzene as an internal standard. The non-dried samples were packed in a cylinder type MAS rotor with an O-ring sealed to avoid the loss of water during NMR measurements. The spectrum of the crystalline component was measured selectively by taking NMR signals after a relaxation of the non-crystalline component. Therefore, to determine the crystalline component of the control and regenerated celluloses, signals obtained after τ = 50 s were measured by using T1CP pulse sequence with CP process as the spectrum of the crystalline component (Torchia 1978).

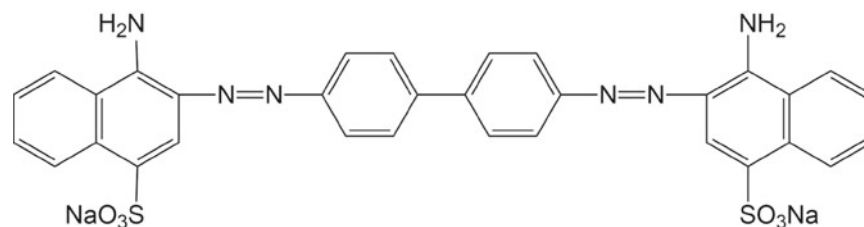
Deuteration-IR spectroscopy

In order to obtain IR spectra of the sample, a membrane suitable for IR spectra (about 10 μm in dry thickness) was prepared on a Teflon plate from non-dried samples and was dried at room temperature. The deuteration of the sample was performed by the method described in previous article (Kai and Mondal 1997). The sample

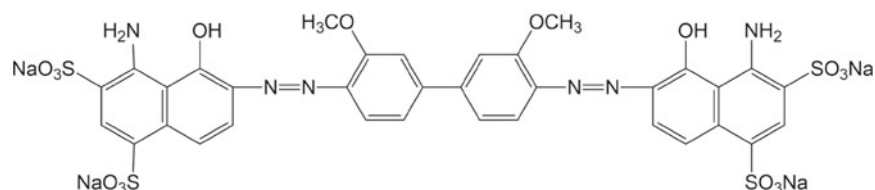
Fig. 1 Chemical structure and name of direct dyes



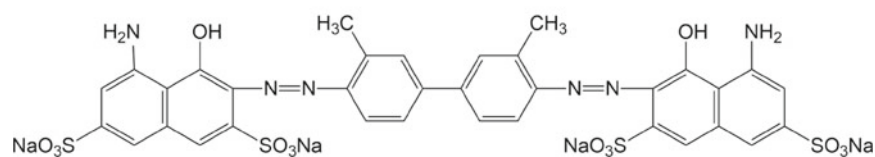
Fluorescent Brightener: Disodium 4,4'-bis[4-bis(2-hydroxyethyl) amino-6-(phenylamino)-1,3,5-triazin-2-ylamino] stilbene-2,2'-disulfonate.



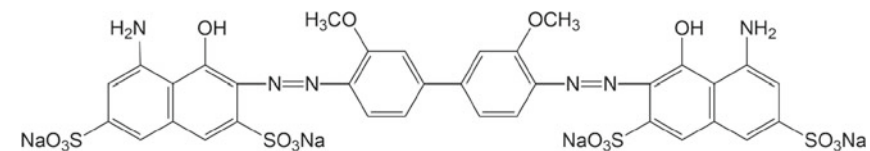
Direct Red 28: 2,2'-[4,4'-biphenylenebis(azo)]-bis[1-aminonaphthalene-4-sulfonic acid] disodium salt.



Direct Blue 1: 2,2'-[4,4'-[3,3'-bis(methoxy)]biphenylenebis(azo)]bis[8-amino-1-hydroxy-naphthalene-5,7-disulfonic acid] tetrasodium salt.

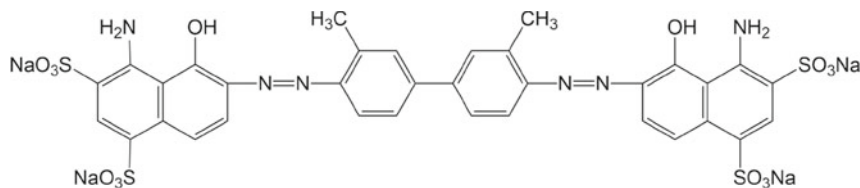


Direct Blue 14: 2,2'-[4,4'-(3,3'-dimethyl)biphenylenebis(azo)]bis(8-amino-1-hydroxy-naphthalene-3,6-disulfonic acid) tetrasodium salt.

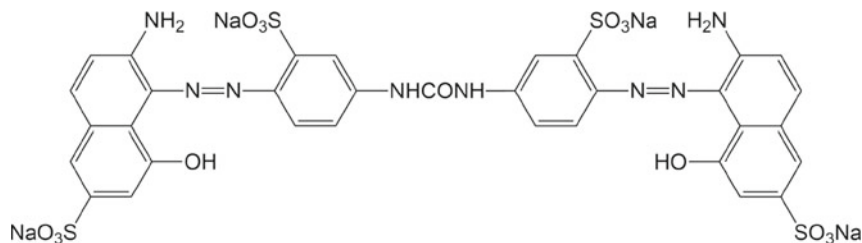


Direct Blue 15: 2,2'-[4,4'-[3,3'-bis(methoxy)]biphenylenebis(azo)]bis[8-amino-1-hydroxy-naphthalene-3,6-disulfonic acid] tetrasodium salt.

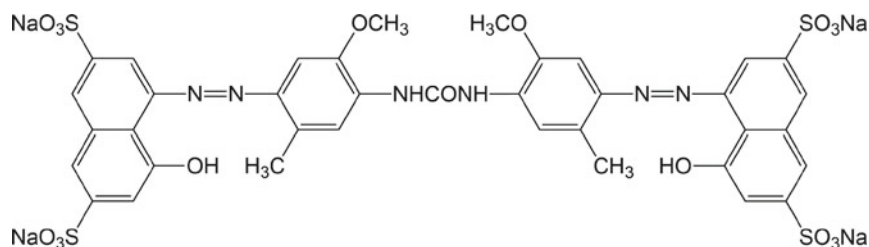
Fig. 1 continued



Direct Blue 53: 2,2'-[4,4'-(3,3'-dimethyl)] biphenylenebis(azo) bis[8-amino-1-hydroxy-naphthalene-5,7- disulfonic acid] tetrasodium salt.



Direct Red 75: 1,1'-ureylenebis[3-sulfo-4-(2-amino-8-hydroxy-6-sulfo-1-naphthylazo) benzene] tetra sodium salt.



Direct Red 79: 1,1'-ureylene-bis[2-methoxy-5-methyl-4-(8-hydroxy-3,6-disulfo-1-naphthyl-azo)benzene] tetrasodium salt.

membrane was set inside a glass cell so that the IR beam was perpendicular to the membrane surface. Before deuteration, the glass cell with the sample was dried under reduced pressure (10^{-3} torr) for 3 h, and then vapour phase deuteration of the sample was performed by D_2O (99.8 %, Aldrich Chemical Co. USA) for a given time at room temperature. In this experiment, the deuteration times used in all the samples were 0, 1, 3, 7 20, 80 and 1,000 min. The deuteration sample was again dried under reduced pressure for the same time of deuteration up to 20 min, and over 20 min deuteration the drying time was fixed to 20 min, and then IR spectrum was measured. FT-IR systems 800 (Nicolet) was used for IR spectroscopy. The amount of OH and OD groups is calculated

according to the method described in the previous paper (Kai and Mondal 1997; Mondal and Kai 1998, 1999).

Results and discussion

Overview of the mechanism of cellulose biosynthesis

Model system for the study of cellulose biosynthesis

No single system has emerged as ideal for the study of cellulose biosynthesis. Very few genera of bacteria synthesize cellulose, but the gram-negative bacterium

A. xylinum secretes large quantities of cellulose as microfibrils for a row of synthetic sites along the longitudinal axis of the cell (Ross et al. 1991). The microfibrils formed each synthetic sites merge to form a large ribbon of cellulose in the growth medium. These ribbons and associated cells tangle, and form a floating pellicle that allows the non-motile strictly aerobic bacteria to grow in the higher oxygen tension at the surface. *A. xylinum* is easily maintained in culture, and conditions of synthesis are readily manipulated; therefore, this bacterium is an excellent model system for studying the molecular mechanisms of cellulose polymerization.

Biosynthesis of cellulose I microfibrils

The biogenesis of native cellulose is a complex process requiring the biosynthesis of high molecular weight 1, 4-glucans and the formation of highly crystalline metastable cellulose I microfibrils; and in many cases, requiring the organization and orientation of these microfibrils into cell walls (Colvin 1964). Proposed mechanisms of cellulose biosynthesis have historically been intimately related to the contemporary understanding of the physical structure of cellulose (Ranby 1952). It is evident that any proposed mechanism of cellulose biosynthesis must be consistent with the ultimate physical properties of macromolecules. Since almost all native cellulose is fibrillar cellulose I, it is probable that there is a common biosynthetic mechanism among all organisms that produce crystalline cellulose I microfibrils (Sarko and Muggli 1974).

Native cellulose I is generally thought to exist in a crystalline lattice of extended parallel glucan chains. It has been proposed that cellulose II, which is infrequently found in nature, is composed of antiparallel glucan chains, and is more thermodynamically stable than cellulose I (Sarko and Muggli 1974; Gardner and Blackwell 1974). For example, cellulose II polymorph mutants of *Acetobacter* lacked organised particles in the lipopolysaccharide (LPS) envelope. They have a functional cellulose synthase, which affect “export” component (Brown 1989). Hence, loss of this control means loss of native cellulose I polymorph synthesis. An understanding of the molecular mechanisms by which organisms synthesize and assemble crystalline metastable cellulose I fibrils is not yet complete. No hypothesis has heretofore been advanced that adequately unifies knowledge of the crystal structure of

cellulose, the substructure of microfibrils, and the biosynthetic mechanism.

Fluorescent brightening agents disrupts ribbon assembly by associating with the subunits of the ribbon called elementary fibrils, and preventing their normal aggregation, when it is added to active cultures of *A. xylinum*. The planar calcofluor molecule binds along the glucan chains through hydrogen bonding (Maeda and Ishida 1967) and dipolar interactions (Rattee and Breuer 1974). In the presence of fluorescent brightening agents and direct dyes, the bacteria synthesize broad bands that are extruded perpendicular to the longitudinal cell axis instead of normal twisting ribbon (Haigler et al. 1980; Haigler and Chanzy 1988; Kai et al. 1994; Kai 1984a; Kai and Kitamura 1985; Kai and Koseki 1985; Benziman et al. 1980). Hence, this is the only direct method for the discussion of mechanism and crystallization of cellulose.

Regulation by a unique mechanism

Acetobacter xylinum has long been considered the most successful prokaryotic model for studies on the mechanism of cellulose biogenesis in intact cell (Zaar 1979; Schramm and Hestrin 1954; Brown Jr 1979; Delmer et al. 1982; Aloni and Benziman 1982). The largest enzyme in the regulation of cellulose synthesis is the membrane bound cellulose synthase. Regulation of the enzyme involves an unusual system of “fine control” based on a novel nucleotide, cyclic diguanylic acid (c-di-GMP) and the regulatory enzymes maintaining its intracellular turnover (Ross et al. 1985, 1986, 1987). A conceptual scheme for the mechanism of regulation of cellulose synthesis, which accounts for the results obtained from the study of the cell-free preparations from *A. xylinum*, is outline by Ross et al. (Ross et al. 1987). Diguanylate cyclase catalyzes the synthesis of c-di-GMP from two molecules of guanosine triphosphate (GTP), via linear dinucleotide triphosphate pppGpG, in two distinct PPI-releasing steps. Within the cell, PPI is rapidly cleaved to yield Pi (Benziman and Palgi 1970). The pathway of c-di-GMP degradation is initiated by a c-di-GMP-specific Ca^{2+} sensitive phosphodiesterase PDE-A, which cleaves a single phosphodiester in the cyclic structure yielding linear dimer pGpG and simultaneously inactivates the molecule. The mononucleotide residues of the inactive dimer structure are then recovered as free 5'-GMP units through the action of a second phosphodiesterase, PDE-B. The Ca^{2+}

inhibition of PDE-A represents an additional locus of regulatory control. According to this feature of the system fluctuations in Ca^{2+} levels may modulate the rate of cellulose synthesis as they influence the persistence of activator in the system.

Mechanism of dye-cellulose complex formation and its model structure

It is well known that fluorescent brightener and direct dyes alter normal ribbon of cellulose fibrils produced when *A. xylinum* cultures in HS medium (Hestrin and Schramm 1954). The X-ray diffraction diagrams of the product in the presence of direct dyes, such as FB, DB1, DB14, DB15, DB53, DR28, DR75 and DR79 indicate good agreement with the alteration concept of cellulose assembly. The 2θ values of native MC observed at 14.5, 16.8 and 22.7° correspond to the (1 $\bar{1}$ 0), (110) and (020) planes respectively (Table 1). These data indicated that the BC samples were the typical crystalline forms of cellulose I (Oh et al. 2005; Yan et al. 2008). These three diffraction peaks of MC appear as two peaks in the products: one of which occurs at lower angle side in the range $2\theta = 10.1\text{--}10.9^\circ$ and the other in the range $2\theta = 19.9\text{--}22.9^\circ$ (higher angle side) (Table 1). However, 2θ values at lower and higher angle sides differ somewhat from each other depending on the dye used. These new diffraction patterns of product do not overlap either with the dye powder or with MC means the product is a complex of cellulose and a dye (dye-cellulose complex). The 2θ values at lower angle side mostly remain unchanged in all cases, whereas those at higher angle side vary a wide range depending on dye characteristics. Although the

experimental results were reproducible, it is assumed that the dye molecule is included in the lower angle plane of X-ray diffraction as its space become broader. From the electron microscopy, Brown et al. (1983) reported that fluorescent brightener was included between the (020) planes of MC. On the other hand, Kai (1984a, b), from their X-ray measurements results, suggested that fluorescent brightener was included corresponding to the (1 $\bar{1}$ 0) plane of MC and thereby formed complex. In this section we will discussed a model structure which may clear the diffraction plane where the dye is actually included.

The diffraction diagram of the uniplanar oriented membrane of MC obtained by reflection method is strong at low angle side (near 14.4°), whereas those at high angle side (near 16.8°) are very weak. The opposite diffraction pattern is found in the case of transmission method (Table 2) (Takai et al. 1975). However, the diffraction diagram of the FB and DR28 products measured by reflection method shows that the diffraction at lower angle side becomes strong whereas that at higher angle side becomes very weak like MC. In contrast, in the diffraction diagram by the transmission method the diffraction peak at lower angle side becomes very weak, and that at high angle side becomes strong. The result of FB and DR28 is a good agreement with the result of Mondal and Kai (2001) and Kai and Mondal (1997). Although the diffraction peak at low angle side of each DB1, DB14, DB15, DB53, DR75 and DR79 products obtained by reflection method is somewhat stronger than that obtained by transmission method, the strength of the diffraction peak at high angle side does not change remarkably. Furthermore, the diffraction diagrams of each DB1,

Table 1 X-ray diffraction angles of the isotropic MC and isotropic product samples obtained from *Acetobacter*-culture in the presence of direct dyes (0.1 wt%)

Sample	2θ (°) (d-spacing in Å)				
MC	...	14.5 (6.12)	16.8 (5.28)	...	22.7 (3.92)
FB	10.6 (8.34)	22.1 (4.02)
DR28	10.4 (8.51)	22.9 (3.89)
DB1	10.9 (8.09)	21.3 (4.17)	...
DB14	10.1 (8.76)	20.6 (4.32)	...
DB15	10.4 (8.53)	20.3 (4.36)	...
DB53	10.3 (8.55)	20.2 (4.40)	...
DR75	10.1 (8.76)	20.6 (4.31)	...
DR79	10.1 (8.78)	19.9 (4.45)	...

Table 2 X-ray diffraction angles of oriented membrane of MC and product samples measured by reflection and transmission methods

Sample	Reflection		Transmission	
	2θ ($^{\circ}$)			
	Lower angle ($1\bar{1}0$)	Higher angle (110)	Lower angle ($1\bar{1}0$)	Higher angle (110)
MC	14.4 (vs)	16.8 (vw)	14.6 (vw)	16.9 (vs)
	Low angle	High angle	Low angle	High angle
FB	9.2 (vs)	22.5 (vw)	11.0 (w)	22.6 (vs)
DR28	9.6 (vs)	23.0 (vw)	10.8 (w)	22.9 (vs)
DB1	12.3 (w)	21.2 (s)	9.5 (w)	20.5 (s)
DB14	10.1 (s)	21.9 (s)	9.5 (w)	19.8 (s)
DB15	11.2 (s)	21.2 (s)	9.5 (w)	20.5 (s)
DB53	10.8 (w)	21.1 (s)	9.6 (w)	20.3 (s)
DR75	10.9 (w)	19.9 (s)	10.1 (w)	20.0 (s)
DR79	11.3 (w)	20.0 (s)	9.8 (w)	20.0 (s)

Strength of the peak intensity of the corresponding angles in the parenthesis is indicated by *vs* very strong, *s* strong, *w* weak, *vw* very weak

DB14, DB15, DB53, DR75 and DR79 products differ somewhat from each other.

From comparison of X-ray diagrams of the products with MC, it is clear that the structure of the products in the presence of a direct dye differs from that of MC. According to Takai et al. (1975) the ($1\bar{1}0$) plane at low angle side of oriented membrane of MC is preferentially parallel to the membrane surface, and the (110) plane is perpendicular to ($1\bar{1}0$) plane. As the membrane of FB and DR28 products were prepared on a Teflon plate and the diffraction nature are closely similar to MC, the diffraction plane at low angle side is parallel to its surface, whereas that at high angle side is perpendicular to it. In contrast, although the sample preparation method is similar in all cases for the membranes of DB1, DB14, DB15, DB53, DR75 and DR79 products, the diffraction planes do not attain such orientation like MC, FB or DR28 product. However, the stronger diffraction by reflection at low angle side of these membranes indicates that it has a trend to become parallel to its surface (Takai et al. 1975). The diffraction plane at high angle side is not necessarily perpendicular to the diffraction plane at low angle side. The aspect of the plane orientation of these product membranes is similar to that of cellulose II membrane (Kai and Mondal 1997). The model of the reflecting planes of MC with inter-molecular

hydrogen bonding, complex loss of the inter-molecular hydrogen bonding due to the inclusion of dye layer between the cellulose sheets, and generalized model structure of MC (cellulose I) and dye-cellulose complex are shown in the Figs. 2, 3 and 4 respectively. As shown in the Fig. 3, if a dye molecule is included in the form of monomolecular layer between the cellulose chains corresponding to the ($1\bar{1}0$) plane of MC, the X-ray measurement result can be explained satisfactorily. The diffraction plane of the FB or DR28 product near $2\theta = 10.4^{\circ}$ is parallel to the surface of the membrane similar to the ($1\bar{1}0$) plane of MC. As a result of inclusion of dye molecule in the form of a monolayer between the (100) planes of product (Fig. 4), the d-spacing of product at low angle side must be become broader by at least 2.0 Å than that of 6.1 Å of MC. However, the diffraction plane of the product near 22.9° must be the (010) plane of the product, because van der Waals forces between pyranose rings of the cellulose chains in the sheets become more stronger due to which 5.3 Å space between the (110) planes of MC become narrowed to 1.4 Å. Kai (1984b) reported from lattice space calculation with X-ray diffraction of the cellulose produced in the presence of fluorescent brightener that the spacing become ($1\bar{1}0$) planes of MC was available to the dye inclusion rather than (020) planes which

Brown et al. (1983) assumed. As the $(1\bar{1}0)$ plane of MC shifted to the broader d-spacing in the product, and $(1\bar{1}0)$ and (110) planes are perpendicular to one another, there are no observed effect of dye inclusion between the (020) plane, and this is a good agreement with that in Kai et al. reports. The structure of cellulose regenerated from the product differs from that of MC means a dye is able to affect the conformation of cellulose chain by the direct interaction of cellulose chain with the dye molecule through the formation of complex product. If a dye is stacked only on the surface of a microfibril as Haigler et al. (Haigler and Chanzy 1988) suggested, it becomes very difficult for it to affect the conformation of all cellulose chains directly, therefore, it is not possible to explain the fact that structures of the product and cellulose regenerated from it differ from each other by the dye used.

MC is extruded from the cell in the shape of the cellulose sheet (or microfibril) corresponding to the $(1\bar{1}0)$ plane (Kai 1984a, b) and the hydrogen bonding is not recognized in the nascent cellulose sheet. Thus, the structure has the cellulose I formative ability which is maintained by the hydrogen bonding or van der Waals forces (Kai and Koseki 1985; Kai et al. 1989). Due to the inclusion of dye between nascent cellulose sheets, the inter-molecular hydrogen bonding between the sheets is prevented, and it is, therefore, suggested that the cellulose chain distance in the sheet will be

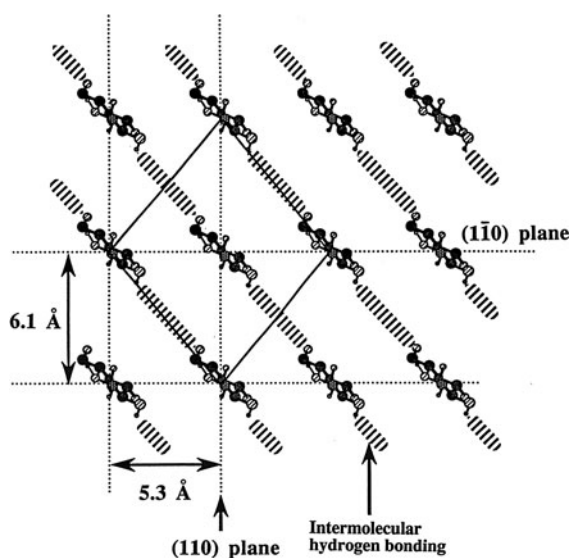


Fig. 2 Schematic model of the reflecting planes of MC with inter-molecular hydrogen bonding

changed from 5.3 Å to lower value at 3.9 Å as shown in Fig. 4. On the other hand, as the dye is included between the $(1\bar{1}0)$ planes, the d-spacing of low angle plane of product become 8.3 Å instead of 6.1 Å of MC. This variation of d-spacing in the complex depends on the size and structure of dye, and their interactions with cellulose in the nascent stage.

If the structure of the product has the unit cell shown in the Fig. 4, infrared and solid state ^{13}C NMR results can also be explained easily. Solid state ^{13}C NMR spectra of cellulose component of product samples are broader than those of the MC. Moreover, the upfield C4 and C6 resonance lines for all the products appear at 84.7 and 62.5 ppm, respectively (Table 3), which are the non-crystalline regions of the cellulose component of MC. No remarkable resonance lines of dye powder overlap onto lines of the product sample. This study suggest that the whole spectra of cellulose components of each dye product are very similar to that of the non-crystalline component of MC, and hence, the hydrogen bonding are not found as in MC. Although the product shows the crystalline X-ray diffraction pattern, these NMR results indicate that all products are in the non-crystalline state. This non-crystalline structure is due to

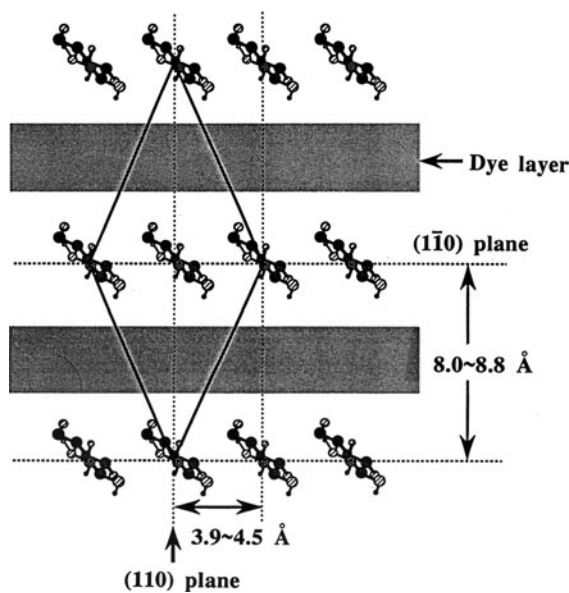


Fig. 3 Schematic model of reflecting planes of dye-cellulose complex. Dye molecule is included between the $(1\bar{1}0)$ planes in the form of monomolecular layer and intermolecular hydrogen bonding is not found like MC

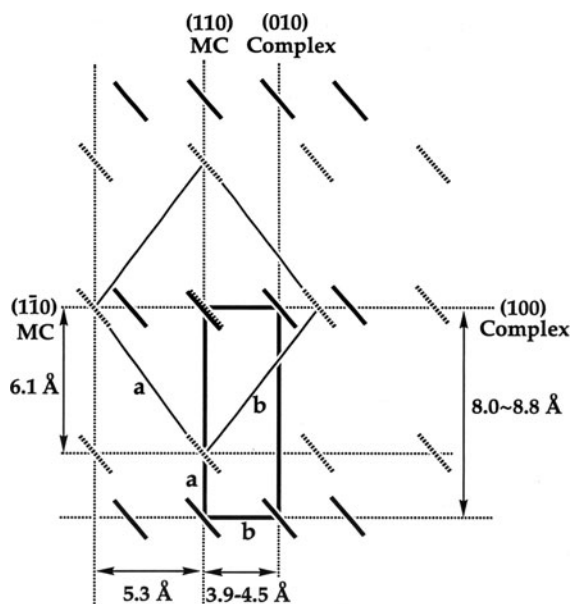


Fig. 4 Generalized model structure of MC (cellulose I) and dye-cellulose complex. Resulting from X-ray diagrams of the product samples, the dye molecule seems to be included between the $(1\bar{1}0)$ planes of MC or (100) planes in the complex, respectively *dotted lines* and *solid lines*: Configuration of cellulose chains in MC and complex, respectively

inclusion of a dye layer between the (100) planes of the product where hydrogen bonding of MC is hindered.

Infrared spectra of the product membrane in the presence of direct dyes show that the absorptions due to intra- and inter-molecular hydrogen bondings, which are observed in the MC (Moosavi-Nasab and Yousefi 2011; Oh et al. 2005), are not appeared. This means that due to the effect of a direct dye, the intra- and intermolecular hydrogen bond formation is prevented in the product sample. Thus existence of hydrogen bonding or not will be in the deuteration-IR. When MC is deuterated for about 80 min, all accessible OH groups (non-crystalline OH groups) changes to OD groups with the peak value near to $2,530\text{ cm}^{-1}$ (Kai and Mondal 1997). Even if the degree of deuteration is different from product to product, this 80 min time is sufficient to deuterate most of the accessible OH groups. Hence, after 80 min deuteration, the absorption spectrum of the product shows the spectrum of only crystalline structure. In 80 min deuteration, around 77–80 % of OH groups of FB and DR28 products transform to OD, respectively, and the spectrum of the deuterated sample is too broad with the peak value at about $3,405\text{ cm}^{-1}$ (inter-molecular

Table 3 Solid state ^{13}C NMR chemical shifts of MC and product samples obtained from *Acetobacter*-culture in the presence of direct dyes (0.1 wt%)

Sample	Chemical shifts (ppm)							
	C1	C4	C2,3,5			C6		
MC	106.3	90.2	84.7	75.9	73.8	72.3	66.5	62.5
FB	105.6	...	84.8	75.9	62.6
DR28	106.0	...	85.2	76.1	63.4
DB1	105.2	...	84.7	75.2	62.4
DB14	105.2	...	85.0	75.2	62.6
DB15	105.2	...	84.9	75.3	62.9
DB53	105.6	...	85.0	75.5	62.9
DR75	105.4	...	85.0	75.3	62.8
DR79	105.6	...	85.2	75.5	63.1

hydrogen bonding) of the MC (Table 4). In this case, if experimental error is taken into consideration, all OH groups in the complex may be deuterated in 80 min. In contrast to these, the absorption spectra of OH groups of deuterated DB1, DB14, DB15, DB53, DR75 and DR79 products indicate that in these dye products, there exist some intra- and inter-molecular hydrogen bonding. Although the degree of deuteration is lower than that of FB and DR28 products and characteristically depends on the dye used, the observed band near to $3,478\text{--}3,480$, $3,438\text{--}3,442$ and $3,339\text{--}3,342\text{ cm}^{-1}$ indicate that the structure of these products has conceived cellulose II crystallinity. In this case we may consider two causes: (a) dye-cellulose interaction does not occur in all region of microfibril; and (b) during washing loosely bound dye molecules are washed off, as a result some of the product may be regenerated.

The region where a dye molecule is dropped out seem to occur cellulose II which crystallize during rinsing or drying process, as a result the diffusion of D_2O is very difficult in that region. Judging deuteration results of the product, it can suggested that the direct binding of the cellulose sheets as well as intra- and inter-molecular hydrogen bonding of the cellulose sheet in the product do not exist as it does in the MC. Increased diffusion of D_2O in the product structure due to lack of hydrogen bonding between the (100) planes supports that the model structure (Fig. 4) is ideal. The easy deuteration of accessible OH groups in the product reveals that the amount of accessible OH groups is always higher than that of MC or cellulose II

Table 4 Relationship between the degree of deuteration and deuteration time of MC and the product samples obtained from *Acetobacter*-culture in the presence of direct dyes (0.1 wt%)

Sample Deuteration time (min)	MC Amount of OD groups (%)	FB	DR28	DB1	DB14	DB15	DB53	DR75	DR79
0	0	0	0	0	0	0	0	0	0
1	9.1	34.2	31.4	39.9	28.9	42.8	43.6	36.6	38.9
3	10.7	47.9	52.6	49.4	39.0	57.9	53.3	51.4	49.7
7	11.5	59.7	63.6	54.7	51.1	65.6	58.9	58.9	57.9
20	12.3	65.7	70.9	62.1	62.9	75.4	62.6	63.1	65.1
80	13.9	85.8	77.9	64.2	68.7	81.9	66.3	65.5	68.8
1,000	16.3	88.6	79.7	65.2	73.5	83.8	68.0	70.5	71.3

as well as of regenerated cellulose from the product. The prevention of hydrogen bond formation by the dye molecule by the inclusion of the dye between cellulose sheets of the product corresponding to the (100) plane in the X-ray diffraction, also support the X-ray model structure of complex formation.

Mechanism of crystalline structure formation in the regenerated cellulose

It is clear from the X-ray diffraction, solid state ^{13}C NMR and IR spectroscopies results that the products obtained in the presence of direct dyes, such as FB, DB1, DB14, DB15, DB53, DR28, DR75 and DR79, are crystalline, and the dye molecule is included between the monolayer of cellulose sheets corresponding to the (1 $\bar{1}$ 0) plane of X-ray diffraction. If the dye is extracted from the product by EtOH and NaOH procedure, the structure of the regenerated cellulose is dependent on the dye used in the culture medium. Tables 5 and 6 show that celluloses I_β and IV_I are regenerated from the FB and DR28 products respectively, and cellulose II is regenerated from each of the DB1, DB14, DB15, DB53, DR75 and DR79 products, but the fine structure of the latter products is different from one another. Furthermore, the crystallinity of these regenerated celluloses are lower than that of the corresponding control celluloses (Table 7).

X-ray diffraction pattern of FB and DR28 celluloses gives the characteristic crystals of celluloses I and IV, respectively. Although X-ray diffraction of FB and DR28 celluloses are two typical celluloses I and IV, it is very difficult to characterize their family without IR and ^{13}C NMR. In our previous article, we have reported that FB cellulose is cellulose I_β (Mondal

and Kai 2001). Hayashi et al. (1987) suggested that chemical shifts of C4 and C6 at about 66.3–67.0 and 63.0–64.0 ppm are due to cellulose IV_I and IV_II , respectively. Thus, the ^{13}C NMR spectra of DR28 cellulose and its chemical shifts confirms that it is cellulose IV_I crystal (Table 6) which is also supported by the IR results. Assigned IR spectra of the hydrogen bonding near 3,425 and 3,341 cm^{-1} support the ^{13}C NMR results that the DR28 cellulose is cellulose IV_I (Kai and Koseki 1985). The development of intra- and inter-molecular hydrogen bondings assigned by IR, and the splitting of C1 and C4 lines, and the chemical shift of C6 line at 63.0–64.0 ppm characterized by ^{13}C NMR (Table 6) in each DB1, DB14, DB15, DB53, DR75 and DR79 celluloses are in a good agreement with the X-ray results of cellulose II. Although the regenerated cellulose from these products are

Table 5 X-ray diffraction angles of MC, cellulose II, cellulose IV_II and celluloses regenerated from the product samples

Polymorphs	2θ ($^\circ$) (d-spacing in Å)		
	(1 $\bar{1}$ 0)	(110)	(020)
MC	14.5 (6.12)	16.8 (5.28)	22.7 (3.92)
FB	14.3 (6.19)	16.0 (5.24)	22.1 (3.99)
Cell IV_II	15.4 (5.75)	15.4 (5.75)	22.0 (4.04)
DR28	15.3 (5.77)	15.3 (5.77)	22.0 (4.04)
Cell II	12.0 (7.39)	19.8 (4.49)	21.1 (4.21)
DB1	12.3 (7.2 1)	20.2 (4.38)	21.4 (4.15)
DB14	12.0 (7.36)	20.0 (4.43)	21.0 (4.22)
DB15	12.3 (7.19)	20.6 (4.30)	21.4 (4.15)
DB53	11.8 (7.47)	19.9 (4.46)	21.6 (4.11)
DR75	12.1 (7.34)	20.0 (4.43)	21.8 (4.07)
DR79	12.2 (7.27)	20.0 (4.44)	21.7 (4.10)

Table 6 Solid state ^{13}C NMR chemical shifts of crystalline component ($\tau = 50$ s) of MC, cellulose II, cellulose IV_{II} and celluloses regenerated from product samples

Sample	Chemical shifts (ppm)									
	C1		C4		C2, 3, 5			C6		
MC	106.3	–	89.7	–	–	75.0	73.5	71.9	66.5	
FB	107.8	105.2	89.8	–	–	75.9	73.4	72.5	66.2	
Cell IV _{II}	106.1	–	89.9	88.7	83.4	77.1	75.3	73.7	64.3	
DR28	106.2	–	89.9	–	85.4	76.1	75.4	73.5	66.1	
Cell II	107.7	105.5	89.4	88.1	–	77.2	75.4	73.4	63.6	
DB1	107.9	105.7	89.6	88.3	–	77.4	75.7	73.4	63.2	
DB14	107.9	105.6	89.5	88.2	–	77.3	75.6	73.4	63.6	
DB15	107.6	105.3	89.2	87.9	–	77.0	75.3	73.1	63.4	
DB53	108.3	106.2	90.1	88.7	–	77.8	75.9	73.8	64.2	
DR75	105.8	103.7	87.5	86.3	–	75.4	73.6	71.7	61.7	
DR79	107.9	105.6	89.5	88.2	–	77.3	75.6	73.4	63.7	

Table 7 Crystallinity of MC, cellulose II, cellulose IV_{II} and celluloses regenerated from product samples

Sample	Crystallinity (%)
MC	76.3
Cell IV _{II}	67.6
FB	51.3
DR28	50.7
Cell II	72.1
DB1	53.8
DB14	50.4
DB15	48.8
DB53	55.2
DR75	71.9
DR79	69.8

cellulose II, the weaker downfield peak of C1 carbon indicates that the hydrogen bonding pattern in DB1, DB14, DB15 and DB53 celluloses are different to that of DR75 and DR79 celluloses. According to Horii et al. (1982), the downfield peak of the corresponding carbon indicate the condition of hydrogen bonding. The weaker downfield peak of C1 carbon of DB1, DB14, DB15 and DB53 celluloses than that of DR75 and DR79 celluloses suggests that the hydrogen bonding pattern in the former celluloses is weaker than that of latter celluloses, which have similar pattern to cellulose II. Higher crystallinity in X-ray results and lower accessible OH groups (i.e. higher non-accessible OH groups) in deuteration-IR results strongly support the stronger hydrogen bond formation in DR75 and DR79 celluloses (Table 8). On the

other hand, Hayashi et al. (1987) reported from their solid state ^{13}C NMR and IR results that bands or signals related to the chain conformation were singular (little or no splitted) in cellulose I family and doublets (splitted into two peaks) in the cellulose II family. This is consistent with two chain conformations, i.e. “bend” for I family and “bend and twisted” for II family, that have one and two types of glucosidic linkage. Thus, the single or little splitted peaks of FB and DR28 celluloses suggest that the chain conformation of FB and DR28 celluloses is “bend” and more symmetric. Similarly, the doublets or two peaks of DB1, DB14, DB15, DB53, DR75 and DR79 celluloses suggest that the chain conformation of these celluloses is “bend and twisted” and asymmetric. This discussion confirms that structures of FB and DR28 celluloses are cellulose I _{β} and IV_I, respectively, and that of other celluloses are cellulose II.

Influence of direct dyes on the nascent structure

A number of direct dye are selected to clarify the effect on the nascent structure of MC. These are FB, DB1, DB14, DB15, DB53, DR28, DR75 and DR79. When *A. xylinum* was cultured in the presence of each of them, the normal ribbon formation become hindered (Haigler and Chanzy 1988) and produced an altered product in which dye molecule was included between the monolayer of cellulose sheet corresponding to the (1 $\bar{1}$ 0) plane (Haigler and Chanzy 1988; Kai and Mondal 1996, 1997; Kai 1984b). However, cellulose regenerated from the altered product was a cellulose

Table 8 Relationship between the degree of deuteration and deuteration time of MC, cellulose II and the celluloses regenerated from the product samples

Sample Deuteration time (min)	MC Amount of OD groups (%)	FB	DR28	Cell II	DB1	DB14	DB15	DB53	DR75	DR79
0	0	0	0	0	0	0	0	0	0	0
1	9.1	24.8	39.7	32.3	37.7	22.6	35.6	39.9	35.9	29.9
3	10.7	29.5	45.0	41.2	44.5	27.5	45.2	45.3	43.9	40.9
7	11.5	32.1	47.9	46.5	7.4	33.1	51.7	48.2	45.5	46.5
20	12.3	35.2	50.5	50.0	1.2	43.0	57.6	49.7	48.4	51.3
80	13.7	44.1	53.2	53.8	3.0	54.7	60.7	52.6	50.7	51.8
1,000	16.3	45.9	54.4	55.7	54.8	57.8	62.9	54.2	52.3	53.6

allomorph other than native MC whose unit cell type was $I\alpha$. Celluloses $I\beta$ and IV_1 were regenerated from FB and DR28 products, respectively, and cellulose II was regenerated from each DB1, DB14, DB15, DB53, DR75, DR79 product depending on the characteristic of direct dye used. These results are due to the easy of dye-cellulose interactions in the nascent stage and rearrangement of intra- and inter-molecular hydrogen bondings produced during dye extraction. Which factors of dye is mainly responsible for the transformation of $I\alpha$ rich MC to other cellulose allomorphs will be discussed in latter.

The structure of FB is different among the direct dyes used. It has two sulfonate groups in its stilbene skeletal backbone and two hydroxyethyl amino groups attached to triazinyl group in each side of stilbene (Table 9). The molecular weight of FB is 960.9 and the cellulose regenerated from FB product is cellulose $I\beta$. Again, all direct dyes DR28, DB1, DB14, DB15 and DB53 have same biphenylenebis (azo) skeletal backbone structure of DR28, but contain different substituent groups and have different position of the same substituent groups in the backbone. The smallest direct dye among the dyes used is DR28 having two sulfonate ($-\text{SO}_3\text{Na}$) and two amine ($-\text{HN}_2$) groups, and its molecular weight is 696.67. On the other hand, DB1 has two sulfonate, two hydroxyl ($-\text{OH}$) and two methoxy ($-\text{OCH}_3$) groups more than that of DR28 and its molecular weight is 992.82. The regenerated cellulose from each DR28 and DB1 product are cellulose IV_1 and II, respectively. If the interaction of sulfonate and amine groups with cellulose microfibril produces cellulose IV_1 from DR28 product, then interaction of additional groups in DB1 produces

cellulose II from DB1 product. There are no differences between the structure of DB1 and DB53 except the position of methoxy groups in DB1. Instead of methoxy groups, there are methyl groups in the same position in DB53. However, the structure of DB1 and DB53 products is similar, and cellulose regenerated from each of them is cellulose II. As the regenerated celluloses from both DB1 and DB53 are cellulose II, it is assumed that methoxy and methyl groups have no remarkable effect on the product as well as on regenerated celluloses from them. Similar effect is observed in the case of DB14 and DB15. Now we may compare the effect of DB1 and DB15 on the nascent MC. Between these dyes, there are no other differences except only the position of sulfonate groups. That is, positions of sulfonate groups are, at 5 and 7 in the naphthalene nucleus in DB1, and at 3 and 6 in the naphthalene nucleus in DB15. As positions of sulfonate groups in these dyes are different, the uniplanar orientation of DB1 product membrane measured by X-ray diffraction is different to that of DB15 product membrane. Although the cellulose regenerated from these dyes is cellulose II, the different absorptions due to the hydrogen bonding observed by deuteration-IR suggest that their fine structures are not same. Similar effect is observed in the case of DB14 and DB53. On the other hand, as the sulfonate groups position of DB1 and DB53 are same, but their molecular weights are different (992.8 for DB1 and 960.82 for DB53), there are no observed effect of these dyes on the product structure as well as their regenerated celluloses. The molecular weight of DR79 is 1,050.89 and is maximum among the dyes used. The regenerated cellulose from DR79 product is cellulose II. From these facts,

Table 9 Molecular weight and substituent groups present in the dye structure

Dye	Mol. wt.	–OH	–NH ₂	–SO ₃ Na	–CH ₃	–OCH ₃	–N = N–	–NHCONH–
FB*	960.96	0	0	2	0	0	0	0
DR28	696.67	0	2	2	0	0	2	0
DB1	992.80	2	2	4	0	2	2	0
DB14	960.82	2	2	4	2	0	2	0
DB15	992.82	2	2	4	0	2	2	0
DB53	960.82	2	2	4	2	0	2	0
DR75	990.80	2	2	4	0	0	2	1
DR79	1,050.89	2	2	4	2	2	2	1

* FB contains two hydroxyethyl amino group, and stilbene skeletal backbone structure

it has been suggested that only molecular weight has no effect on the nascent MC.

From the Table 9, it can be seen that DR75 and DR79 have common four sulfonate, two hydroxyl and –NHCONH– groups, whereas DR75 has two uncommon amine groups and DR79 has two uncommon methoxy and two methyl groups. The skeletal backbone structure of both DR75 and DR79 is ureylenebis(naphthylazo)benzene, but its effect on the nascent MC is closely similar to each other. So, it can be taken into account that uncommon groups, such as methyl, methoxy and amine groups have no extra effect on the nascent MC. But these dyes have different effect on the uniplanar orientation of the product membrane as well as on the fine structure of their regenerated celluloses compared to dyes having four sulfonate groups in biphenylenebis(azo) skeletal structure. It is assumed that these different effects of DR75 and DR79 are due to the interaction of ureylene group of dyes with nascent MC. However, among FB, DB1, DB14, DB15, DB53, DR28, DR75 and DR79, two sulfonate and two amino groups in FB and DR28 are responsible for producing celluloses I_β and IV_I, respectively, while four sulfonate and two hydroxyl and two amino groups in DB1, DB14, DB15 and DB53 are responsible for cellulose II, although the effect of amino groups is not clear.

From the above discussion, it is clear that the reactivity of sulfonate groups in a dye is the most important for the formation of form dye–cellulose complex (product) as well as for the structure of cellulose regenerated from it. In our experiment, cellulose II was obtained from the product of dyes having four sulfonate groups, while celluloses I_β and IV_I were obtained from product of dyes (FB and DR28) having two sulfonate groups. In the former case, higher sulfonate groups of dye increased mainly

the solubility toward water, while hydroxyl, amine, etc. groups increased the affinity toward the cellulose chains. During dye extraction from the products, higher hydrophilicity due to four sulfonate groups may affect the polarity as well as the conformation of the cellulose chains and, as a results, cellulose II is regenerated through the conformational change from DB1, DB14, DB15, DB53, DR75 and DR79 products. Similarly, lower hydrophilicity due to two sulfonate groups in FB and DR28 affects the conformation which produces celluloses I_β and IV_I (cellulose I family) from FB and DR28 products. Thus the results suggest that the interaction of nascent microfibril with two sulfonate groups containing direct dyes have cellulose I family formative ability and that four sulfonate groups containing dyes have cellulose II family formative ability. The other substituent groups in dye have little effect to alter cellulose structure. The order of the reactivity of substituent groups as observed in a dye are: sulfonate > hydroxyl > amine > methoxy ≥ methyl.

Conclusions

The product from *Acetobacter*-culture in the presence of a direct dye is in the crystalline complex composed of a dye and cellulose. The product has the structure in which the monomolecular layer of the dye molecule is incorporated between cellulose sheets corresponding to the (110) plane of microbial cellulose. Cellulose regenerated from the product can have any one cellulose allomorph depending on the dye structure. Cellulose regenerated from FB and DR28 products is Cellulose I_β and IV_I, and that from each DB1, DB14, DB15, DB53, DR75 and DR79 product is Cellulose II.

The dye molecules may bind along the glucan chains of nascent microbial cellulose through hydrogen bonding, dipolar interactions or van der Waals forces. X-ray diffraction, solid state ^{13}C NMR and deuteration-IR study reveal that the dye molecule hinders to form hydrogen bonding in or between nascent cellulose chains or sheets in the product. During dye extraction, the conformation of cellulose chain and the alignment of cellulose chains in the sheets are rearranged with the development of hydrogen bonding corresponding to the characteristics of dye. It appears that the different interactions between direct dye and nascent cellulose depend mainly on the number as well as position of sulfonate groups in the dye. The nascent cellulose produced by *A. xylinum* cell is in the form of a microfibril in which hydrogen bonding is absent. But through the conformational change, this nascent cellulose forms the crystalline regenerated cellulose.

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