Aliphatic Amidase from *Rhodococcus rhodochrous* **M8 Is Related to the Nitrilase/Cyanide Hydratase Family**

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Abstract—A comparative study of amino acid sequence and physicochemical properties indicates the affiliation of an amidase from *Rhodococcus rhodochrous* M8 (EC 3.5.1.4) to the nitrilase/cyanide hydratase family. Cluster analysis and multiple align ments show that Cys166 is an active site nucleophile. The enzyme has been shown to be a typical aliphatic amidase, being the most active toward short-chain linear amides. Small polar molecules such as hydroxylamine and O-methyl hydroxylamine can serve as effective external nucleophiles in acyl transfer reactions. The kinetics of the industrially important amidase-catalyzed acrylamide hydrolysis has been studied over a wide range of substrate concentrations; inhibition during enzymatic hydrolysis by the substrate and product (acrylic acid) has been observed; an adequate kinetic scheme has been evaluated and the corresponding kinetic parameters have been determined.

Key words: *Rhodococcus rhodochrous* amidase, sequence alignment, amidase classification, substrate specificity, acrylamide hydrolysis

Amidases belong to the hydrolase family, a subclass of acylamide amidohydrolases (EC 3.5.1, 3.5.2). These enzymes catalyze the hydrolysis of carboxylic acid amides to free carboxylic acids and free ammonium [1]. Certain amidases in conjunction with nitrile hydratases are involved in nitrogen exchange of pro- and eukaryotes utilizing nitriles. The reactions, which are catalyzed by these complex enzymes, are of primary interest for large scale production of acrylamide and acrylic acid in industry [2 4]. As amidases are studied, it becomes obvious that they exhibit fairly wide substrate specificity [5, 6], demonstrate stereoselectivity towards amides of α -substituted carboxylic acids $[5-7]$, and catalyze acyl transfer reactions [6, 8, 9]. Enzymes in this group attract attention from sci entists in various fields, for instance, in neurobiochem istry [10], industrial microbiology, and chirotechnology [5, 7, 11].

Amidases are still not sufficiently investigated and their classification is not definitely formulated. The clas sification based on substrate specificity [1] occasionally integrates enzymes with different structural organization, mechanism, and catalytic properties; for instance, the amidases from *R. rhodochrous* J1 [12] and *P. aeruginosa* [13] classified in the EC 3.5.1.4 group. Investigation of bonds distinguishes four families: enzymes containing GGSS motif in the primary structure (family index PF01425, IPR000120), nitrilase/cyanide hydratase (fam ily index PF00795, IPR000132), acyl transferases, and ureases [13, 14]. Along with other enzymes, each of these families contains amidases. For instance, the first family includes bacterial amidases from *R. rhodochrous* J1 [12], *Rhodococcus* sp. R312 [6, 14], *Sulfolobus solfataricus* [15], etc. Their amino acid sequences contain the invariant GGSS motif named amidase consensus (index in Prosite Dictionary PDOC00494; PS00571) [16]. These enzymes have quaternary structure and usually are organized as homodimers and homooctamers [12-14, 16]. The main role in the active site of its typical representative, amidase from *R. rhodochrous* J1, belongs to the invariant amino acid residues Asp191 and Ser195 [12]. These amidases exhibit wide specificity, they are active towards aliphatic and aromatic amides and amides of α -substituted carboxylic acids, and they exhibit stereospecificity [5, 12, 15].

amino acid sequences of the enzymes cleaving the $C-N$

In contrast, the bacterial amidases from the nitri lase/cyanide hydratase family are supposedly sulfhydryl enzymes [13, 17, 18]. These include amidases from *P. aeruginosa* [13], *B. stearothermophilis* [19], *R. erythropolis* [20], *Rhodococcus* sp. R312 [9, 20, 21], and others. The

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characteristic feature of their primary structure is the two conservative amino acid sequences named nitrilase/ cyanide hydratase consensus (index in Prosite Dictionary PDOC00712; PS00920/1). Unlike amidases from the first family, the amidases from the nitrilase/cyanide hydratase family form homotetramers and homohexamers. The data available in the literature show that aliphatic amides are the preferred substrates for this enzyme group.

An amidase from *R. rhodochrous* M8 [22] was recent ly isolated and characterized, and its gene was cloned [23]. The first results of the study demonstrate significant differences between this enzyme and amidase from *R. rhodochrous* J1. The detailed comparison of these two amidases is of great interest: it is necessary to understand the differences in mechanisms of protein induction and production, similarities and differences in their catalytic mechanisms; establish the correlation between structure and function, as well as the patterns of quaternary struc ture formation and its role in catalysis. In this work a number of potential substrates for the new enzyme are considered, the kinetic principles of hydrolysis and acyl transfer reactions are investigated, and bioinformatic analysis of amino acid sequence of *R. rhodochrous* M8 amidase is performed.

MATERIALS AND METHODS

Analysis of protein amino acid sequences. *A search for homologous sequences* was performed using protein data bases and the following software: BLAST and PSI BLAST [24] (NCBI, www.ncbi.nlm.nih.gov) or FASTA [25] (EMBL, www.ebi.ac.uk).

Alignment of amino acid sequences was performed using CLUSTAL W v. 1.82 [26] (EMBL, www.ebi.ac.uk) software.

A search for conservative regions in the BLOCK data base was performed via BLOCK Search software on SWISSPROT server (http://www.ebi.ac.uk/swissprot) and using MACAW software [27].

Profile construction for the nitrilase/cyanide hydratase family was carried out using LINEUP, PILE UP, and PROFILEMAKE software. Profile specificity was verified by SWISS-PROT database using PROFILE-SEARCH software. All software is a part of the GCG v. 7.3 package [28].

A dendrogram of related proteins was constructed based on results of multiple alignment of amino acid sequences using CLUSTAL W v. 1.82 software at the EMBL site [26], and further refinement was done with Philip software package and the PromL program.

Protein secondary structure was predicted by hydrophobic cluster analysis [29] and the PredictProtein Server (www.embl-heidelberg.de/predictprotein) [30].

Data about enzymes and proteins were extracted from SWISS-PROT, PIR, and TrEMBL databases (Table 1).

Chemicals. The following reagents were used in this work: acrylamide (Reanal, Hungary); mercaptoethanol, mandelic acid, semicarbazide, O-methyl hydroxylamine hydrochloride, sodium azide (Fluka, Switzerland); phenylglycine (Sigma, USA); methanol, sodium dodecyl sulfate (Merck, Germany); acrylic acid, phenylacetic acid (Aldrich Chemie, Germany); dithiothreitol (Serva, Germany); phenylacetamide, phenoxyacetic acid (Reakhim, Russia); *o*-phthalic aldehyde (Koch Light, UK); amide of phenoxyacetic acid (Acros, Belgium); acetonitrile (Cryochrom, Russia); other reagents and buffer solution components were of domestic origin and chemically pure.

Homogeneous amidase preparation was obtained from E. L. Kotlova (Institute of Genetics and Selection of Industrial Microorganisms, Russia). The protein concen tration in the solution was 1 mg/ml by the Bradford method [31]. Sample purity was characterized by using denaturing SDS electrophoresis in polyacrylamide gel by the Laemmli method [32]. Acyl transferase activity of amidase measured using acetyl hydroxamic acid was 2.9 µmol/min per mg protein at 37°C.

HPLC-analysis. The chromatographic system consisted of a Waters M6000 eluent supply module, Luna 5u $C-18$ column for the reverse-phase chromatography $(250 \times 4.6 \text{ mm}, 5 \text{ \mu m})$ (Phenomenex, USA), Marathon autosampler (Spark Holland, The Netherlands), and LC 235 Diode Array detector (Perkin Elmer, USA). The chromatograms were recorded at 210 nm on a Multichrome software–hardware complex for acquisition and processing of chromatographic data (Ampersand, Russia). Flow rate was 0.5 ml/min. The mobile phase contained (v/v) 40% CH₃CN, 0.9 g/liter SDS, and 0.78 g/liter KH_2PO_4 (pH 3.0). Retention times (min): acrylamide (7.1) , acrylic acid (9.0) , N-acryloyl hydroxylamine (6.5) , N-acryloyl- $(O$ -methyl) hydroxylamine (8.0) , phenylglycine (6.5) , phenylglycine amide (15) , N a cryloyl-phenylglycine amide (13.5) , mandelic acid amide (8.0), mandelic acid (10), phenylacetamide (8.1), phenylacetic acid (14.5), phenylacetic acid amide (13), and phenoxyacetic acid (16).

Potentiometry. Potentiometric detection of ammoni um ions was performed using an Expert-00-3 ionometer (NPO Ekonix-Expert, Russia) and ammonium selective electrode. The measurements were performed at fixed ionic strength of 0.1 M versus buffer solution containing magnesium sulfate.

Spectrophotometry. The concentration of primary amino groups was determined using modification by *o* phthalic aldehyde and mercaptoethanol [33] on a UV 1601 spectrophotometer (Shimadzu, Japan).

Protocol for enzymatic reactions. Enzymatic hydrol ysis and acyl transfer reactions were performed at 37°C according to the Michaelis–Menten scheme at $[S]_0$ >> $[E]_0$ in 0.2 M Tris-HCl buffer containing 0.1 M KCl, 1 mM dithiothreitol, and 0.02% NaN₃ (pH 7.5). The reactions were carried out in a thermostatted cell of 719S Titrino pH-stat (Metrohm, Switzerland); the total volume of the reaction mixture was 3-5 ml. The concentration of the original substrates in the reaction mixture was varied in the range of $0.1-1000$ mM; amidase concentration in the range of 0.01 -0.3 mg/ml. Aliquots of the reaction mixture were collected after certain periods of time and analyzed by the HPLC-technique, spectrophotometry, and potentiometry.

RESULTS AND DISCUSSION

Study of primary structure. The search for homolo gous proteins in the databases using the primary structure of *R. rhodochrous* M8 resulted in a number of amino acid sequences (about 100) with identity level $82-28\%$. The first ten sequences with the highest identity level (82 75%) were amidases from the nitrilase/cyanide hydratase family. Thus, it was already in the first step of our investi gation when the similarity between the primary structures of *R. rhodochrous* M8 amidase and proteins of the nitri lase family was found.

Based on a global multiple alignment of protein amino acid sequences listed in Table 1, a dendrogram of amidases and related proteins was constructed (Fig. 1). Nitrilase/cyanide hydratase and GGSS motif-containing amidase families include several groups. Together with other proteins, amidase from *R. rhodochrous* M8 forms the amidase group of the nitrilase family. According to primary structure, the studied enzyme is most similar to amidases from *Brevibacterium* sp. and *R. erythropolis*, with a common ancestor for these enzymes.

Five highly conservative regions were identified as a result of global and local multiple alignment of amino acid sequences of amidases from the nitrilase/cyanide hydratase family (Fig. 2). These include conservative amino acid residues Cys, Glu, Lys involved in a catalytic triad of *P. aeruginosa* active site, a representative of this family [17]. One should note common elements of the secondary structure (Fig. 2) and characteristic structure of nucleophilic region of the active site: C(DY)(DG) for amidases, and C(WA)E for nitrilases. These sulfhydryl elements supposedly act according to the classical "ping pong" mechanism [8]. An inhibitory effect of specific sulfhydryl group reagents was shown for the studied ami dase [22] and related enzymes [8].

Similarity between the primary structure of the stud ied amidase and amidases from nitrilase family, the exis tence of characteristic catalytic triad Cys-Glu-Lys, as well as common elements of the secondary structure show that members of this group of enzymes have similar prop erties.

Substrate specificity. Acrylamide hydrolysis. Large scale production of acrylic acid by enzymatic hydrolysis of acrylamide is important in polymer industry. *Rhodococcus*

strains are presently used for biocatalytic production of acrylamide and acrylic acid [3, 11, 22]. Special interest is aroused by enzymatic conversion at high substrate con centrations. In this connection, enzyme hydrolysis of acrylamide was investigated over a wide range of substrate concentrations $(1-1000 \text{ mM})$ (Fig. 3). The maximum reaction rate was achieved already at initial substrate con centrations of 20-30 mM, however further increase in acrylamide concentration resulted in a decrease in hydrolysis rate. Kinetic analysis demonstrates that inhibi tion of enzyme activity is due to the binding of additional substrate molecules and formation of catalytically inactive ES_3 enzyme–substrate complex (Fig. 4). Such inhibition of enzyme activity was observed earlier in the case of ami dohydrolases (EC 3.5.1.11) [34, 35]. Reversible inhibition of amidase activity by acrylamide [19] was recently found for the enzyme from *B. stearothermophilus* (nitrilase fami ly, Table 1). Analysis of integral kinetic curves indicates the inhibition of enzyme hydrolysis by reaction product acrylic acid (the inhibition constant for the product is \sim 10 mM). Therefore, optimal conditions for acrylamide hydrolysis are determined by the kinetic parameters tak ing into account the influence of the system components. In general, the enzyme displays relatively high catalytic activity, but it has medium affinity towards acrylamide (15 mM), which is typical for related amidases as well (Table 2), and, obviously, is determined by the structure of the acyl moiety of the substrate. The GGSS motif-containing amidases catalyze acrylamide hydrolysis with lower efficiency.

Hydrolytic and transferase activity*.* Along with hydrolysis, there is considerable interest in enzymatic transfer of an acyl group on external nucleophile (for instance, hydroxylamine). Production of N-acryloylderivatives of amino compounds is one of the steps in pro duction of polyacrylates with modified properties. Considering this, different acyl group donors were inves tigated as potential substrates for amidase in the reaction of acyl transfer, including aromatic and chiral α -substituted amides: acrylamide, semicarbazide, amides of phenylacetic, phenoxyacetic, (R) - and (S) -mandelic acid, and phenylglycine. The following compounds were investigated as acceptors: hydroxylamine, semicarbazide, methoxyamine, (\pm) -phenylglycine amide.

Enzymatic reaction of acyl group transfer from acrylamide to hydroxylamine is highly efficient and occurs practically without formation of hydrolysis prod uct. The acyl transfer reaction to methoxyamine is less efficient, whereas phenylglycine amide does not act as an acceptor of acyl group at all. It should be noted that at high concentrations of amine components (more that 100 mM) non-enzymatic side processes were observed (especially in the case of hydroxylamine). Semicarbazide (a urea analog) is not hydrolyzed by amidase, but it inhibits acrylamide hydrolysis as well as that of urea itself [36, 37]. The fact that none of the investigated aromatic

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Table 1. Characterization of amidases and related proteins by protein databases

Note: A, amidase; CH, nitrile hydratase; N, nitrilase; IAAH, indole acetamido hydrolase; AH, 6-aminohexanoate cyclic dimer hydrolase; U, urease; enantios., enantioselective.

Fig. 1. Dendrogram of amidases and related proteins (see Table 1). Enzyme family names are presented according to the commonly accept ed nomenclature [14]. Abbreviations for related enzyme groups: BA, bacterial amidases; IAAH, indole acetamide hydrolase; EA, eukary otic amidase; AH, nylon oligomer cleaving hydrolases; AA, acetamidases and related eukaryotic amidases; ANC, amidases from the nitri lase/cyanide hydratase family; N, nitrilases; CH, nitrile hydratases.

	EEEEE	EEEEEEEE	EEEEEE						HHHH	
AMID RHOR8	51 GMDLVVFPEYSTMG 64	116 YNTLVLINDQGEIVQKYRKILP 137				156	PKGLKISLIICDDGNYPEIWR 176			
AMIE BACSP	51 GMDLVVFPEYSTMG 64	116 YNTLVLINNKGEIVOKYRKIIP 137					156 PKGLKISLIVCDDGNYPEIWR 176			
ALAM RHOER	51 GMDLVVFPEYSTOG 64	116 YNTLILINDKGEIVORYRKILP 137					156 PKGLKISLIICDDGNYPEIWR 176			
ALAM PSEAE	51 GMDLVVFPEYSLQG 64	116 YNTLVLIDNNGEIVOKYRKIIP 137				156	PKGMKISLIICDDGNYPEIWR 176			
AMIE RHOER	51 GMDLVVFPEYSTOG 64	116 YNTLVLIDNNGEIVOKYRKILP 137					156 PKGLKISLIICDDGNYPEIWR 176			
AMIE BACST	51 GMDLVIFPEYSTMG 64	116 YNTLVLINNKGEIVQKYRKIIP 137					156 PKGLKISLIICDDGNYPEIWR 176			
AMIE HELPY	51 GLDLIIFPEYSTHG 64	115 YNTLILINDKGEIVOKYRKILP 136				155	PKGLKVSLIICDDGNYPEIWR 175			
NRL1 ARATH	57 GAELVLFPEGFIGG 70	134 YCTVLFFSPOGOFLGKHRKLMP 155				176	TPIGKLGAAICWENRMPLYRT 196			
NRL2 ARATH	50 GSELVVFPEAFIGG 63	127 YCTALFFSPOGOFLGKHRKLMP 148				169	TPIGKLGAAICWENRMPLYRT 189			
NRL3 ARATH	57 GAKLVLFPEAFIGG 70	134 YCTALFFSPOGOFLGKHRKVMP 155				176	TPIGKIGAAICWENRMPLYRT 186			
NRL4 ARATH	68 GSQLVVFPEAFIGG 82	144 YCTVLFFDSQGLFLGKHRKLMP 165				181	TPIGKLGAAICWENRMPSYRT 191			
NRL1 RHORH	45 GAEFLAFPEVWIPG 58	118 YLSQVFIDQNGDIVANRRKLKP 139				160	FGFGRVGGLNCWEHFOPLSKY 180			
NRL2 RHORH	39 GCELVAFPEVFIPG 52	112 YMTOLVIDADGOLVARRRKLKP 133				154	MPFARLGALNCWEHFOTLTKY 174			
NRLA ALCFA	39 GCDLIVFGETWIPG 52	111 YLGOCLIDDKGOMLWSRRKLKP 132				153	TELGRVGALCCWEHLSPLSKY 173			
CYHY GLOSO	38 GCKLIAFPEVWIPG 51	110 YLTQVLISPLGDVINHRRKLKP 131				153	-EIGRLGQLNCWENMRPFLKS 173			
CYHY GIBBA	38 GCKFVAFPEVWIPG 51	208 IOPLTWLLLSMLSRLARGLLLP 229				154	LRLAASGQLNCWENMRPFLKS 174			
	$^{\circ}$			$_{\odot}$				O		
	GhtLVVFPEh G	Y t hhItttG		RKILP			hhKI	CDDG P		
	FII G	I F	М	G LI			RV.	WEN	T	
	LЬ	L		VK.			L	Η		
	EEEEE		HHHHHHHHH		HH					
AMID RHOR8	231 GHSAIIGFDG 240	249 EDYGVOYAQLSLSTIRDARANDOSONHLFK 278								
AMIE BACSP	231 GHSAIIGFDG 240	249 EENGIQYAEVSLSQIRDFRKNAQSQNHLFK 278								
ALAM RHOER	231 GHSAIIGFDG 240	249 EEYGIQYAQLSVSTIRDARENDQSQNHIFK 278								
ALAM PSEAE	231 GHSAIIGFDG 240	249 EEMGIQYAQLSLSQIRDARANDQSQNHLFK 278								
AMIE RHOER	231 GHSAIIGFDG 240	249 EEYGIOYAOLSVSAIRDARENDOSONHIFK 278								
AMIE BACST	231 GHSAIIGFDG 240	249 EENGIQYAEISLSQIRDFRQNAQSQNHLFK 278								
AMIE HELPY	230 GHSSIIGFDG 239	248 EENGLQYAQLSVQQIRDARKYDQSQNQLFK 277								
NRL1 ARATH	268 GGSVIISPLG 277	286 ESEGLVTADLDLGDIARAKLYFDSVGYYSR 315								
NRL2 ARATH	261 GGSVIISPLG 270	279 ESEGLITADLDLGDVARAKLYFDSVGHYSR 308								
NRL3 ARATH	268 GGSVIISPLG 277	286 ESEGLVTADLDLGDIARAKLYFDVVGHYSK 315								
NRL4 ARATH	278 GGSSIISPLG 288	297 RGEALITADLDLGDIARAKFDFDVVGHYSR 306								
NRL1 RHORH	258 GVARIYGPDG 267	277 DAEGLLYAELDLEQIILAKAAADPAGHYSR 306								
NRL2 RHORH	261 GFARIIGPDG 270	270 DEEGILYADIDLSAITLAKOAADPVGHYSR 299								
NRLA ALCFA	250 GSSMIFAPDG 259	269 DAEGLIIADLNMEEIAFAKAINDPVGHYSK 398								
CYHY GLOSO	256 GHARIFRPDG 259	274 DFDGLMYVDIDLNESHLTKALADFAGHYMR 303								
CYHY GIBBA	257 GHARIYRPDG 259	274 DFDGLLFVDIDLNECHLTKVLADFAGHYMR 304								
	G shIIGFDG	Et GIhhAtLtLt h		R	H hK					
	FSP	GL VII D		K	Y R					
	YR	GV.	V V		\circ					

Fig. 2. Multiple alignment of amino acid sequences for amidases from the nitrilase/cyanide hydratase family. Enzyme abbreviations are pre sented in Table 1. Invariant amino acid residues are marked with bold. © sign indicates the supposed amino acid residues of the active site [17]. Upper line shows the secondary elements: E (β-structures), H (α -helices). Three lower lines show alignment consensuses: one main and three the most frequently encountered amino acid residues (h, hydrophobic residues; t, polar residues; s, small residues). The most con servative residues are marked with the bar above sequences.

amines are donors of an amine group in the reactions cat alyzed by amidase should be pointed out.

Summarizing the data accumulated to date, it can be concluded that in contrast to amidases containing the GGSS motif [5, 6, 15], amidases from the nitrilase/ cyanide hydratase family [9, 19, 21, 22] efficiently cat alyze reactions with aliphatic amides and are not specific towards the substrates with large aromatic acyl moiety.

* Nitrilase/cyanide hydratase family.

** GGSS motif containing amidase family.

Fig. 3. Dependence of the initial rate of acrylamide hydrolysis catalyzed by amidase from *R. rhodochrous* M8 on initial concentration of the substrate (a). The experimental data are presented with symbols: the line describing the experimental points was calculated in accordance with minimal kinetic scheme (Fig. 4) by the equation: $v = V_{\text{max}}[S]_0 / (K_S + [S]_0 + [S]_0^2 / K_S')$. The dashed line corresponds to the formation of inactive ES_2 enzyme–substrate complex. The values of $K_m = 15$ mM and $V_{max} = 2.9$ µmol/sec per mg were determined from experimental data at low substrate concentrations (b). The value of $K_S' = 1.4$ M was determined at high substrate concentrations according to the equation: $1/[S]_0 + [S]_0^2/(K_m \cdot K_S') = V_{\text{max}}/K_m \cdot v^{-1} - 1/K_m$ (c). Agreement between experimental data and the model is achieved only in the case of formation of ES_3 enzyme–substrate complex. Experimental conditions: pH 7.5, 37°C, ionic strength 0.1 M.

Thereby, specificity towards a nucleophile in acyl transfer reactions catalyzed by both amidase families is obviously limited by small polar molecules such as hydroxylamine and hydrazine. It is surprising that even ethanolamine and methyl hydrazine, having similar structures, do not act as acyl group acceptors [6, 9]. However, the question about the substrate specificity of amidases from the two families towards the acyl group donor as well as acceptor should not be closed. Further investigations of structure-function relationships including three-dimensional enzyme organization are needed.

The performed analysis allows classifying the ami dase from *R. rhodochrous* M8 as a member of the nitrilase/cyanide hydratase family and assuming that

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Fig. 4. "Minimal" kinetic scheme for amidase catalyzed acryl amide hydrolysis. E is amidase, S is acrylamide, ES is enzyme–substrate complex, P_1 and P_2 are hydrolysis products (ammonium and acrylic acid, respectively), EP_2 is enzyme complex with acrylic acid, ES_{n+1} is catalytically inactive enzyme complex with additional substrate molecules $(ES_3$ for amidase from *R. rhodochrous* M8).

Cys166 is the active site nucleophile. Inhibition of activi ty by specific sulfhydryl-group reagents, specificity towards aliphatic amines, and quaternary structure char acteristic for proteins from this family all support this assumption. Effective catalysis of acrylamide hydrolysis as well as acyl transfer is obviously a special feature of aliphatic amidases from the nitrilase/cyanide hydratase family, but the question of the boundaries of substrate specificity of enzymes from this family requires further investigations.

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