

BRIEF COMMUNICATION

Changes of leaf antioxidant system, photosynthesis and ultrastructure in tea plant under the stress of fluorine

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Abstract

Seedlings of *Camellia sinensis* were grown hydroponically for 30 d in order to study the effect of fluorine (F) on growth parameters, antioxidant defence system, photosynthesis and leaf ultrastructure. Fresh and dry mass, chlorophyll (Chl) content and net photosynthetic rate (P_N) decreased with increasing F concentration. Superoxide dismutase (SOD) activity decreased significantly, catalase (CAT) and guaiacol peroxidase (GPX) activities reached maximum under 0.21 and 0.32 mM F, respectively. Proline, malondialdehyde (MDA) and hydrogen peroxide (H_2O_2) contents increased significantly. These results suggested, that antioxidant defence system of leaves did not sufficiently scavenge excessive reactive oxygen species. The cell ultrastructure was not changed under 0.11 - 0.21 mM F, however, it was destroyed at 0.32 - 0.53 mM F. So tea plants tolerated F in concentration less than 0.32 mM.

Additional key words: catalase, *Camellia sinensis*, chloroplasts, guaiacol peroxidase, H_2O_2 , malondialdehyde, net photosynthetic rate, proline, superoxide dismutase.

Fluorine, a phytotoxin in air, water, soil, and vegetation, is released into the environment from a number of industrial sources (Mackowiak *et al.* 2003), application of phosphate fertilizers in agriculture (Loganathan *et al.* 2001), and weathering of volcanic ashes (Cronin *et al.* 2003). F is transferred from soil to roots, and then to above ground parts, or absorbed by leaves from the air. The F content was reported to reach 871 - 1337 mg kg⁻¹(f.m.) in mature tea leaves, and even more than 2000 mg kg⁻¹(f.m.) in leaves of old tea plants (Ruan and Wong 2001, Shu *et al.* 2003). To our knowledge, little information is available regarding the effects of excessive F on physiological functions in tea plant. In the present study, the changes of antioxidant defence system, photosynthesis and cell ultrastructure of tea leaves under F stress were investigated, in order to study the mechanisms of tea plant F tolerance.

Camellia sinensis (L.) O. Kuntze cv. Fu Ding da Bai 1-year-old cuttings, provided by the Fruit and Tea Research Institute of Hubei Agricultural Academy, China, were planted in plastic pots containing 1.5 dm³ of 1/2 strength Hoagland nutrient solution (Hoagland and Arnon 1950). F (as NH_4F) was supplied at five concentrations: 0 (control), 0.11, 0.21, 0.32, 0.53 mM. For each treatment, 5 pots (with 5 seedlings each) were used, and the pots were arranged in the glasshouse in random design. The liquid solutions (pH 5.5) were ventilated with air pumps and replaced completely every 5 d. The seedlings were cultivated for 30 d in a glasshouse under day/night temperature of $25 \pm 3/15 \pm 2$ °C and irradiance, of 250 - 280 $\mu\text{mol m}^{-2} \text{s}^{-1}$ during 16-h photoperiod.

Fresh mass (f.m.) of the whole plant was determined immediately after harvesting. Dry mass (d.m.) of the whole plant was determined after drying at 80°C till

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Abbreviations: CAT - catalase; Chl - chlorophyll; F - fluorine; GPX - guaiacol peroxidase; MDA - malondialdehyde; P_N - net photosynthetic rate; SOD - superoxide dismutase; TEM - transmission electron microscope.

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constant mass. Chlorophyll (Chl) content was measured in an 80 % acetone extract spectrophotometrically at 663 and 645 nm as described by Jiang *et al.* (2007). Net photosynthetic rate (P_N) was measured by *TPS-1* photosynthesis system (*PP Systems*, Amesbury, MA, USA). The seedlings were put outdoors in the morning from 9:00 to 11:00. The photosynthesis of the 3rd or 4th leaf from the top was detected at irradiance of 800 to 900 $\mu\text{mol m}^{-2} \text{s}^{-1}$, CO_2 concentration of 370 to 380 $\mu\text{mol mol}^{-1}$, air temperature of 28 ± 0.5 °C and relative humidity of 60 ± 0.8 %. Three seedlings were measured in each treatment and repeated three times. Total F was tested with an ion selective electrode (*Orion 9609BNWP* with *Orion pH/ISE meter 710*, both *Thermo Scientific*, Waltham, MA, USA). The leaves were oven-dried at 80 °C for 24 h, and then ground to pass through a 2-mm sieve. The grounded sample (0.2 g) was extracted with 10 cm^3 of 0.2 M HCl at room temperature for 1 h. After filtration, 25 cm^3 total ionic strength adjustment buffer (TISAB) was added. Finally, the volume was fixed to 100 cm^3 with super-pure water produced by a purification system (*Millipore*, Bedford, MA, USA).

For enzyme extraction, fresh leaves (0.5 g) were ground on ice with 0.5 g quartz sands and 5 cm^3 of 50 mM precooled phosphate buffer (pH 7.8) containing 1 mM ethylenediaminetetraacetic acid (EDTA) and 5 % (m/v) polyvinylpyrrolidone (PVP). The homogenate was centrifugated (16 000 g) at 4 °C for 15 min. Then the supernatant was used for superoxide dismutase (SOD; E.C.1.15.1.1), catalase (CAT; E.C.1.11.1.6) and guaiacol peroxidase (GPX; E.C.1.11.1.7) analysis (Pereira *et al.* 2002). The activity of total SOD, Cu/Zn-SOD, Mn-SOD and CAT were determined by using the reagent kit (Nanjing Jiancheng Bioengineering institute, Nanjing, China). The GPX was determined by modified guaiacol method. 0.1 cm^3 of enzyme solution, 0.9 cm^3 of 2 %

guaiacol and 1.0 cm^3 of 0.3 % H_2O_2 were added into 1.0 cm^3 of potassium phosphate buffer (pH 5.7), following absorbance variation at 470 nm within 5 min (Ramiro *et al.* 2006).

Proline content in the leaves was measured by acidic ninhydrin method according to Khedr *et al.* (2003). The content of malondialdehyde (MDA) in the leaves was measured by thiobarbituric acid (TBA) method according to Dhindsa *et al.* (1981). The H_2O_2 content in the leaves was measured according to Patterson *et al.* (1984).

For transmission electron microscope (TEM) analysis, the middle sections of the third leaves were cut (1 × 1 mm), and then put into the fixation solution composed of 2.5 % glutaraldehyde at pH 7.4 immediately to make the samples sunk. The samples were fixed in 1 % OsO_4 for 2 h, dehydrated by acetone, embedded with epoxy resin (*SPI-812*) for 24 h, and sliced up by ultramicrotome (*UC6*, *Leica Microsystems*, Wetzlar, Germany). The sections were stained with a mixture of lead citrate and uranyl acetate. Finally the configuration of cells was checked using TEM (*H-7650*, *Hitachi*, Tokyo, Japan) and photographed with *Gatan832* digital imaging system (*Gatan*, Pleasanton, USA).

All statistical analyses were done using the statistical package of the SAS software computer program. ANOVA followed by LSD test were carried out to test the significance.

SODs are ubiquitous metalloenzymes that play a role in defense against toxic reduced oxygen species. Cu/Zn-SODs are found throughout the plant cell, existing in both chloroplastic and cytosolic forms; Mn-SODs are located in mitochondria and also in peroxisomes (Attia *et al.* 2008). Our data showed that the activity of total SOD, Cu/Zn-SOD and Mn-SOD all decreased significantly ($P < 0.01$) with increasing F concentration (Table 1). The result is consistent with the findings of Wilde and Yu

Table 1. The whole plant fresh and dry masses, F, Chl, proline, H_2O_2 and MDA contents, P_N , and activities of antioxidative enzymes in leaves of tea cultivated in a half strength Hoagland nutrient solution with 0, 0.11, 0.21, 0.32 or 0.53 mM F for 30 d. Means \pm SD ($n = 3$). Different letter in each row represents significant differences at $P < 0.05$, based on LSD's multiple range test.

Parameters	0	0.11	0.21	0.32	0.53
Fresh mass [g plant ⁻¹]	5.51 \pm 0.08a	5.49 \pm 0.04a	5.28 \pm 0.1b	4.90 \pm 0.09c	4.58 \pm 0.09d
Dry mass [g plant ⁻¹]	1.72 \pm 0.03a	1.67 \pm 0.03a	1.57 \pm 0.04b	1.45 \pm 0.02c	1.33 \pm 0.04d
F content [mg kg ⁻¹ (d.m.)]	1283 \pm 16a	1525 \pm 6b	1702 \pm 15c	1763 \pm 20d	1876 \pm 14e
Chl content [mg g ⁻¹ (f.m.)]	2.21 \pm 0.02a	2.26 \pm 0.01a	2.15 \pm 0.03b	1.99 \pm 0.04c	1.86 \pm 0.03d
P_N [$\mu\text{mol m}^{-2} \text{s}^{-1}$]	12.70 \pm 0.16a	13.30 \pm 0.26a	10.70 \pm 0.36b	8.60 \pm 0.70c	3.90 \pm 0.16d
Total SOD [U g ⁻¹ (f.m.)]	3191 \pm 46a	3084 \pm 22b	2899 \pm 25c	2818 \pm 52c	2639 \pm 64d
Cu/Zn-SOD [U g ⁻¹ (f.m.)]	2314 \pm 34a	2239 \pm 113a	2109 \pm 67b	2091 \pm 46b	1934 \pm 45c
Mn-SOD [U g ⁻¹ (f.m.)]	918 \pm 66a	921 \pm 32a	806 \pm 47b	711 \pm 12c	705 \pm 58c
CAT [U g ⁻¹ (f.m.)]	380 \pm 13c	430 \pm 11b	447 \pm 7a	395 \pm 11c	358 \pm 22d
GPX [U g ⁻¹ .min ⁻¹]	1143 \pm 28c	1149 \pm 70c	1274 \pm 34b	1514 \pm 5a	1206 \pm 50c
Proline [$\mu\text{g g}^{-1}$ (f.m.)]	22.66 \pm 1.54c	30.68 \pm 1.72b	32.91 \pm 1.37b	48.39 \pm 1.03a	49.73 \pm 0.74a
H_2O_2 [$\mu\text{mol g}^{-1}$ (f.m.)]	29.08 \pm 1.60c	30.18 \pm 0.63bc	33.20 \pm 0.81ab	36.08 \pm 2.39a	36.65 \pm 2.22a
MDA [nmol g ⁻¹ (f.m.)]	6.56 \pm 0.51c	6.69 \pm 0.57c	8.15 \pm 0.02b	10.35 \pm 0.49a	10.30 \pm 0.26a

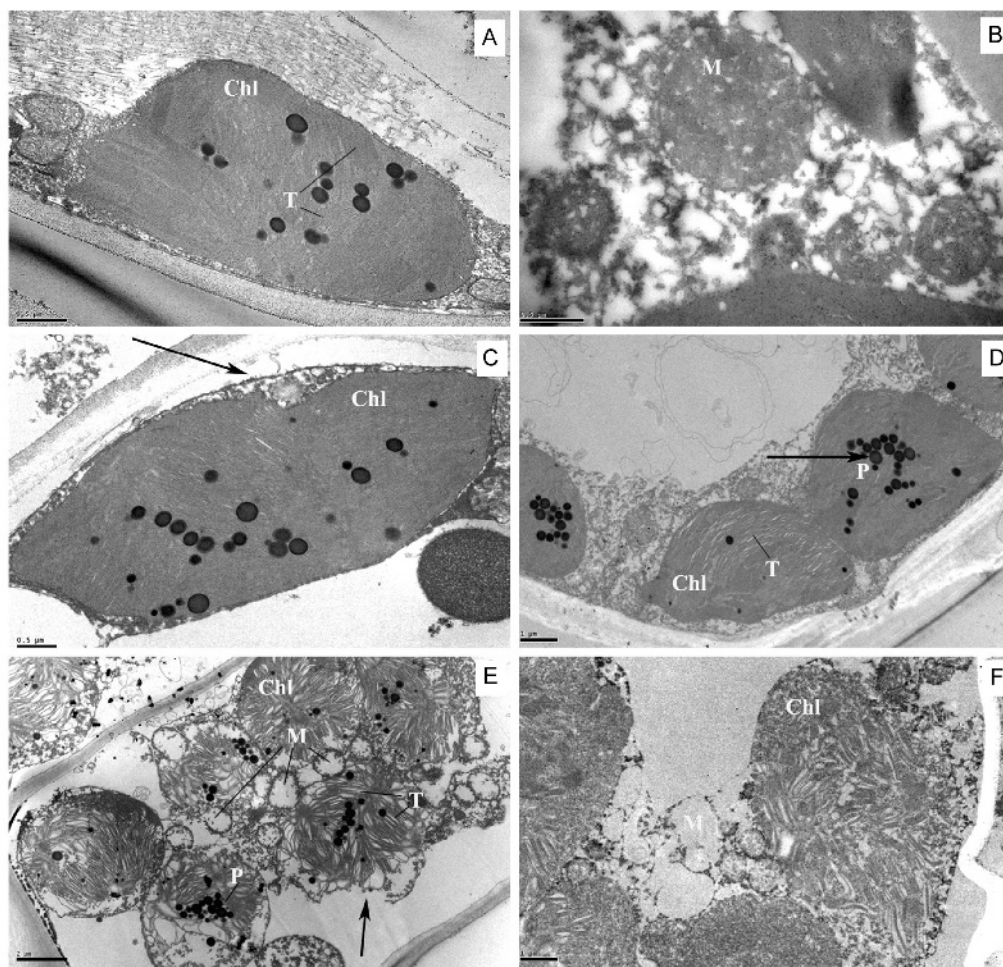


Fig. 1. Leaf ultrastructure of tea seedlings grown under 0 mM (control, *A,B*), 0.11 mM (*C*), 0.21 mM (*D*), 0.32 mM (*E*) and 0.53 mM (*F*) fluorine for 30 d, respectively. Bars = 0.5 μm (*A,B,C*), 1 μm (*D,F*) and 2 μm (*E*), respectively. *A* - chloroplast was elliptical and thylakoids were arranged closely and aligned in an orderly manner; *B* - mitochondria were intact, with complete membrane structure and clear cristae; *C* - the plasmolysis observed at 0.11 mM F treatment; *D* - the plastoglobulus increased (arrow) and thylakoids expanded slightly at 0.21 mM F treatment; *E* - the shape of chloroplast changed, membrane was ruptured (arrow), the system of the membranes was obliterated, mitochondria were cavitated completely; *F* - chloroplasts and thylakoids were disintegrated, and mitochondria degraded. Chl - chloroplast, T - thylakoid, M - mitochondria, P - plastoglobulus.

(1998). CAT and GPX activities increased at low F concentrations, reached their peaks when the F concentrations were 0.21 and 0.32 mM, respectively, and decreased at higher concentrations (Table 1). Generally, antioxidant enzymes and non-enzymatic antioxidants including proline have synergistic effects on free radical scavenging, thus the generation and removal of free radical is balanced (Mascher *et al.* 2002, Tasg n *et al.* 2003). However, the reduced activity of antioxidant enzymes under stressed condition can lead to lipid peroxidation and membrane damage (Scandalios 2005). The content of MDA, a product of peroxidation of membrane lipids, is often used as an indicator of oxidative damage (Gunes *et al.* 2007). In the present study, MDA and H_2O_2 contents increased significantly (Table 1). Accumulation of H_2O_2 and lipid peroxidation resulted in a significant decrease in cell membrane stability. TEM also

proved it, *e.g.* membranes of chloroplasts were ruptured (Fig. 1*E*). These results suggested that antioxidants did not sufficiently scavenge excessive reactive oxygen species to protect the tissue from free radical injury under the F stress.

The related research had proved that the increase of proline was a common response of most plants to environmental stress (Almansouri *et al.* 1999, Meloni *et al.* 2001). In addition to osmotic adjustment (Tripathi and Gaur 2004), proline also stabilizes cellular structures and acts as a free radical scavenger (Alia and Matysik 2001, Siripornadulsil *et al.* 2002). In this trial, proline content increased remarkably ($P < 0.01$). Significant increase in proline content has been reported earlier in tea under Cu^{2+} and Al^{3+} stress (Yadav and Mohanpuria 2009).

Based on the results of TEM, it was discovered that cell ultrastructure changed little under 0.11 - 0.21 mM F, but an irreversible destruction under the concentration of F

higher than 0.32 mM was observed. Under high F stress, chloroplast damage, such as membrane rupture, thylakoid expansion (Fig 1E) and even disintegration (Fig 1F) was found, which would strongly influence the photosynthesis. In present study, both Chl content and P_N increased slightly at low F concentration (0.11 mM) but not significantly. However, these two parameters decreased significantly ($P < 0.05$) with increasing F concentration. The high electronegativity of F destroyed the Chl molecule and accelerated disintegration of chloroplasts

(Fornasiero 2001, Aboal *et al.* 2008).

In conclusion, exposure of tea seedlings to F decreased fresh mass, dry mass, Chl content and P_N, increased MDA, proline and H₂O₂ contents and changed SOD, POX, CAT activities. It is suggested that antioxidant defence system did not sufficiently protect the tissue under severe F stress. This was confirmed by the damage of cell ultrastructure. Tea plants are able tolerate F in concentration less than 0.32 mM.

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