

Tellurite reduction by *Escherichia coli* NDH-II dehydrogenase results in superoxide production in membranes of toxicant-exposed cells

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Abstract Tellurite, the most soluble tellurium oxyanion, is extremely harmful for most microorganisms. Part of this toxicity is due to the generation of reactive oxygen species that in turn cause oxidative stress. However, the way in which tellurite interferes with cellular processes is not well understood to date. Looking for new cellular tellurite targets, we decided to evaluate the functioning of the electron transport chain in tellurite-exposed cells. In this communication we show that the *E. coli ndh* gene, encoding NDH-II dehydrogenase, is significantly induced in toxicant-exposed cells and that the enzyme displays tellurite-

reducing activity that results in increased superoxide levels in vitro.

Keywords Tellurite · Tellurite reductase · NDH-II · Superoxide · Electron transport chain · *Escherichia coli*

Introduction

Although scarce in the earth's crust, tellurium is commonly found in alloys with other metals like calaverite (AuTe_2) and silvanite (AgAuTe_4) and nagyagita $\text{AuPb}(\text{Sb},\text{Bi})\text{Te}_{2-3}\text{S}_6$, among others (Cairnes 1911). Te belongs to the chalcogens (Group 16), along with other biologically important elements such as oxygen, sulfur and selenium (Fischer 2001) and is generally recovered as a waste of the mining activity. In its elemental form (Te^0), tellurium does not display toxicity to living organisms, mostly because of poor solubility. Conversely, the tellurium oxyanions tellurite (TeO_3^{2-} , very soluble) and tellurate (TeO_4^{2-} , poorly soluble) are highly toxic to most microorganisms, especially Gram-negative bacteria (Taylor 1999).

Tellurite-mediated damage includes—among other effects—altered ΔpH , decreased ATP levels and increased generation of reactive oxygen species (ROS) in *Escherichia coli* (Tantaleán et al. 2003; Lohmeier-Vogel et al. 2004; Pérez et al. 2008), *Pseudomonas pseudoalcaligenes* KF707 (Borsetti et al. 2005) and *Rhodobacter capsulatus* (Tremaroli et al. 2007, 2009).

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Once inside the cell, tellurite is reduced in a chemical or enzymatic manner and superoxide (O_2^-) is generated (Pérez et al. 2007, 2008; Tremaroli et al. 2007, 2009). In turn, this oxygen radical may cause membrane lipid peroxidation either by increasing the production of toxic aldehydes or changing the monounsaturated fatty acids ratio (Pérez et al. 2007; Pradenas et al. 2012, 2013). Tellurite also affects the activity of key metabolic enzymes such as phosphofructo kinase, pyruvate kinase, pyruvate dehydrogenase and α -ketoglutarate dehydrogenase while a number of metabolites including pyruvate, α -ketoglutarate and some phosphorylated sugars are accumulated in the presence of the toxicant (Castro et al. 2008; Reinoso et al. 2012; Valdivia-González et al. 2012). In addition, tellurite-mediated damage is reflected in protein carbonylation (Pérez et al. 2007, 2008; Contreras and Vásquez 2010), dismantling of [4Fe–4S] clusters from enzymes such as fumarase A and aconitase B (Imlay 2006; Calderón et al. 2009) and in a rapid decrease of the cell's reduced thiol content, especially that of glutathione (Turner et al. 1999, 2001). Induction of oxidative stress response-regulator genes such as *oxyR* and *soxRS* along with those encoding catalase (*katA*, *katG*), superoxide dismutase (*sodA*, *sodB*, *sodC*), fumarase C (*fumC*), aconitase A (*acnA*), aldehyde reductase YqhD (*yqhD*) and glutathione peroxidase (*btuE*) has also been observed upon tellurite exposure (Imlay 2003; Pérez et al. 2007, 2008; Arenas et al. 2010, 2011).

Tellurite (Te^{4+}) can be enzymatically reduced in vivo to elemental tellurium (Te^0) in a NAD(P)H-dependent manner by branch activities exhibited by some enzymes that include catalase, dihydrolipoil dehydrogenase from the pyruvate dehydrogenase (PDH) and α -ketoglutarate dehydrogenase complexes and isocitrate dehydrogenase (Calderón et al. 2006; Castro et al. 2008, 2009; Reinoso et al. 2012), among others. Interestingly, enzymes involved in electron transport, such as nitrate reductases of *E. coli*, *Paracoccus denitrificans*, *P. pantotrophus* and *R. sphaeroides* (Avazeri et al. 1997; Sabaty et al. 2001), as well as the *E. coli* terminal oxidases have been also involved in tellurite reduction (Trutko et al. 1998, 2000), which is concomitant with superoxide generation and with the accumulation of intracellular black deposits whose cell location would probably be associated with redox sites such as the cell membrane (Taylor et al. 1988; Lloyd-Jones et al. 1994; Calderón et al. 2006).

The electron transport chain (ETC) is responsible for generating a proton motive force that is used to transport molecules through the membrane, allowing processes such as flagellar rotation and ATP synthesis during oxidative phosphorylation. In aerobic conditions, the *E. coli* ETC is formed by the NADH dehydrogenase complex I (NDH-I), NADH dehydrogenase II (NDH-II), succinate quinone oxide reductase complex and the terminal oxidases *bo*, *bd-I* and *bd-II* (Matsushita et al. 1987; Hayashi et al. 1989; Puustinen et al. 1991; Calhoun et al. 1993; Nakamura et al. 1996).

Our work has focused on NDH-II, a 47 kDa enzyme encoded by the *ndh* gene. Similar to NDH-I, NDH-II catalyzes NADH oxidation (Young et al. 1978) but unlike it, NDH-II does not participate in translocating protons from the cytoplasm to the periplasmic space (Matsushita et al. 1987). Recently it was shown that *E. coli* NDH-II displays the ability to reduce Cu^{2+} to Cu^{1+} when incubated in the presence of FAD and NADH (Rapisarda et al. 1999, 2002; Volentini et al. 2011). Increased copper sensitivity has been observed in *ndh* mutants, suggesting that NDH-II can alleviate the oxidative damage generated by this metal (Rodríguez-Montelongo et al. 2006).

In this communication we show that the *ndh* gene is induced in tellurite-exposed *E. coli* and that NDH-II catalyzes the NADH-dependent tellurite reduction while generating superoxide in vitro.

Materials and methods

Bacterial strains and growth conditions

Bacterial strains used in this work are indicated in Table 1. Cells were grown at 37 °C with vigorous shaking in non-fermentative minimal medium (NFMM) that contained M9 1× salts (Na_2HPO_4 34 g, KH_2PO_4 15 g, NH_4Cl 5 g and $NaCl$ 2.5 g in 1 l of a 5× stock solution), casaminoacids (0.1 %) and glycerol (1 %) (Lin 1976). Solid media contained 2 % agar. When required, the medium was amended with the appropriate antibiotic. Growth was initiated with 1:10,000 dilutions of saturated cultures. For tellurite treatments, cultures in early exponential growth phase ($OD_{600} \sim 0.3$) were exposed for 15 or 30 min to sublethal (1.5 μM) tellurite concentrations (corresponding to one half of the minimal inhibitory concentration, MIC, as determined in NFMM medium).

Table 1 *E. coli* strains used in this study

Strain	Relevant genotype	Source or reference
BW25113	BW25113 $\Delta(\text{araD-araB})567$, $\Delta\text{lacZ4787}(\text{:rrmB-3})$, λ^- , rph-1 , $\Delta(\text{rhaD-rhaB})568$, hsdR514 (KC ^a)	Baba et al. 2006
NDH-I ⁻	BW25113, <i>nuoF</i> NADH:ubiquinone oxidoreductase, F chain (KC ^a)	Baba et al. 2006
NDH-II ⁻	BW25113, <i>ndh</i> NADH:ubiquinone oxidoreductase II (KC ^a)	Baba et al. 2006
BWpBAD	BW25113 carrying Pbad	This work
pB <i>nuoF</i>	NDH-I ⁻ carrying pBAD <i>nuoF</i>	This work
pB <i>ndh</i>	NDH-II ⁻ carrying pBAD <i>ndh</i>	This work
AG1	<i>recA1</i> , <i>endA1</i> , <i>gyrA96</i> , <i>thi-1</i> , <i>hsdR17</i> ($\text{r}_{\text{k}}^- \text{m}_{\text{k}}^+$) <i>supE44</i> , <i>relA1</i> , pCA24 N	Kitagawa et al. 2005
pCNDH-II	AG1 carrying pCA24 <i>ndh</i>	Kitagawa et al. 2005
AN387	F ⁻ , <i>rpsL</i> , <i>gal</i>	J. Imlay
AS454	AN387 <i>sodC::spc</i>	J. Imlay
<i>nuo</i>	AS454 <i>nuo zej-223::Tn10</i>	J. Imlay
<i>nuo/ndh</i>	AS454 <i>nuo zej-223::Tn10 ndh::cam</i>	J. Imlay
<i>ndh</i>	AS454 <i>ndh::cam</i>	J. Imlay
pEE3	BL21(DE3) carrying pET-TOPO101 <i>lpdA</i>	Castro et al. 2008

^a Keio Collection

Gene cloning

Using *E. coli* genome sequences deposited in databases, specific primers were designed to amplify the genes of interest (Table S1). PCR products were ligated to the pBAD/TOPO (Invitrogen) vector and transformed into competent *E. coli* JW2279 (*nuoF*) and JW1095 (*ndh*) strains. Plasmid purification was accomplished by using the Miniprep Kit QIAGEN (Darmstadt, Germany). Analysis of recombinant plasmids was carried out by PCR and *NcoI* restriction.

RNA purification

After exposing *E. coli* to tellurite for 15 or 30 min, cells were harvested at 8,000×g for 5 min and suspended in 20 mM sodium acetate pH 5.5 buffer, 1 mM EDTA, 5 % SDS (lysis buffer) and heated at 70 °C for 10 min. Then 500 μl of acid phenol

(Trisure) were added and the mix was gently stirred once per min for 10 min at 70 °C. Chilling was at room temperature and after adding 200 μl of chloroform the solution was mixed as before. After centrifuging for 15 min at 15,000×g at 4 °C, the aqueous phase was rescued and nucleic acids were precipitated with two volumes of isopropanol. The tube was kept at -80 °C for 1 h and centrifuged as above. The pellet was air dried, suspended in 20 μl of a buffer (Qiagen) that contained RNase-free DNase I and incubated for 1 h at 37 °C. DNase was inactivated at 75 °C for 10 min and RNA was precipitated as described above. The pellet was suspended in nuclease-free water that contained diethylpirocarbonate (DEPC) and 0.01 % RNAsin (Invitrogen). RNA quantification was assessed using the Quanti-iT RNA kit (Invitrogen). Total RNA was finally diluted to a concentration of 250 ng/μl for qPCR reactions. Specific primers used are shown in Table S1.

Real-time PCR

Primers for qPCR assays (~300 bp) (Table S1) were designed utilizing the Vector NTI[®] 11 software (Invitrogen). Two hundred-fifty ng of RNA were used for qPCR with the LightCycler RNA Amplification SYBR Green I Kit (Roche Applied Science). The reaction mix contained 2 μl of each primer (10 pmol/ml), 0.4 μl of LightCycler RT-PCR Enzyme Mix, 4 μl of resolution buffer, 2 μl of 10 mM nucleotide mix (dATP, dUTP, dCTP, dGTP), 4 μl SYBR Green I and 2.4 μl of 25 mM MgCl₂. PCR-grade water was added to 20 μl final volume. The RT reaction was carried out at 30 °C using an amplification program that consisted of 45 cycles (30 s denaturation at 95 °C, 10 s annealing at 55 °C, 30 s at 72 °C); *rpoD* gene was used as the housekeeping gene. The LightCycler 2.0 software was used to record the fluorescence increase in each cycle. Finally the results were analyzed as relative expression as described earlier (Pfaffl 2001).

Isolation of *E. coli* membrane fractions

NFMM-grown *E. coli* cultures were harvested at 8,000×g for 5 min at 4 °C and washed with 50 mM Tris-HCl pH 7.4 buffer (buffer A) or, when indicated with 50 mM MES-NaOH pH 7.4 buffer (buffer B). Cell extracts were prepared according to the method previously described (Messner and Imlay 1999) with

minor modifications. Membrane protein concentration was assessed as described (Bradford 1976) using 1:1 membrane fraction:guanidine hydrochloride (GndCl, 6 M) to improve both membrane solubility and sensitivity. BSA was used as protein standard.

NADH dehydrogenase activity in *E. coli* membrane fractions

NADH dehydrogenase activity was assessed at 25 °C in membrane fractions by monitoring NADH oxidation at 340 nm. The reaction mix (500 µl) contained buffer A and 60 µM NADH. Reactions were initiated by adding 25 µg of membrane proteins as previously described (Matsushita et al. 1987). A unit was defined as the amount of enzyme that oxidized 1 µmol of NADH per min at 25 °C.

Tellurite reductase (TR) activity

Escherichia coli membrane fractions were incubated for 10 min at 37 °C in a mix that contained 50 mM Tris-HCl pH 7.4 buffer, 1 mM K₂TeO₃, 1 mM 2-mercaptoethanol and 0.5 mM NADH (TR) buffer as previously described (Calderón et al. 2006). A unit was defined as the amount of enzyme that increased the OD₅₀₀ by 0.001 units per min at 37 °C.

Protein carbonylation in *E. coli* membrane proteins

Carbonyl group content was assessed in membrane fractions as described earlier (Semchishyn et al. 2005) with minor modifications (Contreras and Vásquez 2010).

Protein purification

NDH-II, FumA and dihydrolipoil dehydrogenase were purified after inducing *ndh*, *fumA* and *lpd* genes with 1 mM IPTG for 4 h in appropriate *E. coli* strains (Table 1). Briefly, cell extracts were prepared by sonication and after eliminating the cell debris by low speed centrifugation, cleared supernatants were loaded onto HisTrap affinity columns under the vendor's instructions (GE Healthcare Life Sciences).

In vitro superoxide generation

Affinity-purified NDH-II, FumA or Lpd (E3) were mixed with TR buffer (TRB) that contained the water

soluble probe 2-(4-Iodophenyl)-3-(4-nitrophenyl)-5-(2,4-disulfophenyl)-2H-tetrazolium monosodium salt (WST-1, Sigma) and incubated for 5 min at 25 °C. Superoxide formation was monitored at 439 nm as described earlier (Calderón et al. 2006).

Te quantification by inductively coupled plasma-optical emission spectrometry (ICP-OES)

Membrane fractions from cells previously exposed to tellurite were suspended in 1 ml of buffer A and gently sonicated. Samples were diluted tenfold with 10 % HNO₃ and the whole volume was used for Te determination using a Spectro CIROS Vision ICP-OES instrument using the 214 nm Te line. A calibration curves were constructed with commercially-available tellurium standards (Sigma-Aldrich) also containing 10 % HNO₃.

Scanning electron microscopy (SEM)

Purified membrane fractions were incubated with sodium desoxycholate (2 %) for 2 h to form membrane vesicles. After centrifuging at 120,000×g for 1 h, the sediment—consisting mainly of membrane vesicles—was washed with buffer A and fixed with glutaraldehyde (1.25 %) for later viewing in a SEM EVO MA 10 (Karl Zeiss) microscope.

Results and discussion

Tellurite reductase activity in *E. coli* membranes

It is well known that the Te oxyanion tellurite is toxic for most living organisms. Once inside the cell this toxicant causes a number of alterations including [FeS] cluster dismantling, decreased glutathione pool, increased ROS levels and accumulation of key metabolites, among other effects (Tremaroli et al. 2007; Chasteen et al. 2009). Nevertheless, the complete description of tellurite-mediated damage is currently unknown.

Our laboratory has been interested for a long time in unveiling tellurite-mediated damage to the cell's general metabolism. In these investigations it has been observed that important metabolic pathways are also affected by TeO₃²⁻ including glycolysis (Valdivia-González et al. 2012), Krebs cycle (Calderón et al.

2009; Reinoso et al. 2012) and the pentose phosphate shunt (Sandoval et al. 2011). Given that these paths provide the real ETC substrates, namely NADH and FADH₂, it was very interesting to focus on tellurite effects on the *E. coli* ETC. This goal finds relevance in the fact that tellurite damage occurs mainly in the presence of oxygen (Tantaleán et al. 2003) and in recent results showing that the bacterial membrane is likewise affected by the toxicant (Pérez et al. 2007; Pradenas et al. 2012, 2013).

On the other hand, it has been shown that some NAD(P)H-dependent dehydrogenases like dihydrolipid dehydrogenase (E3) (Castro et al. 2008, 2009) and isocitrate dehydrogenase display tellurite-reducing ability (Reinoso et al. 2013). In this context, preliminary results from our laboratory indicate that while NADH dehydrogenase activity from NDH-I is completely abolished by tellurite in toxicant-exposed cells, that of NDH-II remains almost unaltered (to be published elsewhere). The obvious question to answer therefore was if NDH-II exhibits or not TR activity.

Purified membrane fractions from tellurite-exposed *E. coli* displayed a characteristic black color, indicative of metallic tellurium deposits (Fig. S1A). Similar results using whole cells were communicated earlier (Taylor et al. 1988; Trutko et al. 2000). Membrane vesicles were clearly identifiable in SEM images (Fig. S1B). It is not clear if elemental Te has an impact or not on membrane structure or fluidity. If so, transport and movement of lipid-soluble molecules could be altered, thus affecting the whole functioning of the membrane. For instance, altered quinone diffusion could be reflected in decreased electron flux (and hence proton motive force), terminal oxidase activity and/or ATP synthesis.

To determine if tellurite reduction occurs at the level of the ETC's dehydrogenases in vitro, membranes from the parental strain were incubated with TR buffer that contained cyanide to eliminate the potential contribution of terminal oxidases. This experiment was undertaken because it had been previously communicated that ETC terminal oxidases can use tellurite as the final electron acceptor (Trutko et al. 1998, 2000; Baesman et al. 2007). Opposite to the findings of Trutko et al. (2000), who observed that tellurite reduction as well as the accumulation of elemental Te crystals decreased in cyanide-exposed cells, our results clearly show that tellurite reduction actually increases in the presence of KCN (Fig. 1a). This occurs probably because instead of being conducted from the

dehydrogenases to the terminal oxidases, electrons are being directly transferred to tellurite. These data allow us to infer that the cyanide target in the terminal oxidases is not related to that where tellurite reduction is taking place. Alternatively, this could also mean that the terminal oxidases would not be the only ETC macromolecules involved in tellurite reduction.

With this in mind, tellurite reduction was assessed in membrane fractions from wild type and mutant *E. coli* strains, exposed or not to the toxicant. After the treatment, no change was observed in tellurite reduction by wild type and NDH-II⁻ strains. However, TeO₃²⁻-exposed membranes from NDH-I⁻ cells showed an important augment of tellurite reduction (Fig. 1b). These results suggest that exposing cells to the toxicant causes important changes in membrane composition and/or structure that could promote Te⁴⁺→Te⁰ reduction. In this context, NDH-II could represent a NADH-dependent tellurite-reducing enzyme whose expression could be increased in the tellurite-provoked absence of the NDH-I complex. Further evidence stating that NDH-II but not NDH-I is able of tellurite reduction came from experiments in which dNADH, an exclusive substrate for NDH-I, was used instead of NADH. No tellurite reduction was observed at all under these conditions (not shown). On the other hand and with the exception of lawsome, quinone participation in tellurite reduction has been recently ruled out (Wang et al. 2011).

NDH-II displays TR activity

Given that NDH-II is the ETC's most important dehydrogenase during the *E. coli* aerobic growth (Unden and Bongaerts 1997), it would not be unexpected that its structural gene, *ndh*, would be induced in the presence of oxygen (see below). This could also represent a cell response against tellurite damage to the ETC, which is in agreement with the observation that, as mentioned before, NDH-I and not NDH-II is affected by tellurite.

To determine if effectively NDH-II is the membrane component that is responsible for TR activity, the enzyme was purified and used for assessing tellurite-reducing activity. The E3 component of the *E. coli* pyruvate dehydrogenase complex was also purified and used as positive control for NADH-dependent TR activity (Castro et al. 2008, 2009)(Fig. S2a). NAD-II's TR activity was determined in the

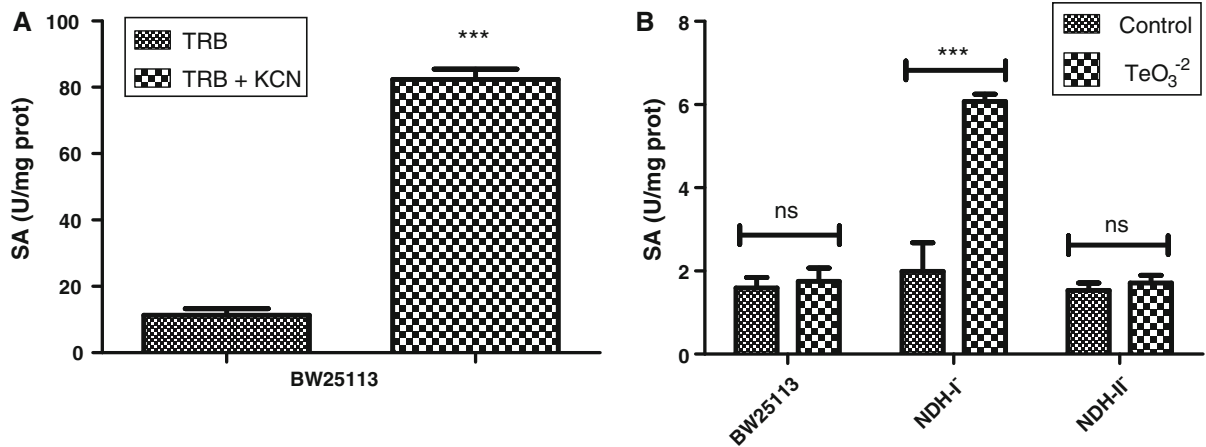


Fig. 1 Tellurite reductase activity in *E. coli* membrane fractions. TR activity was assayed in membrane fractions from the indicated bacteria unexposed (a) or exposed (b). KCN was

used to inhibit cytochrome oxidase activity. SA specific activity. Bars represent the average of three independent trials \pm SE. $P < 0.001$ (***), NS not significant

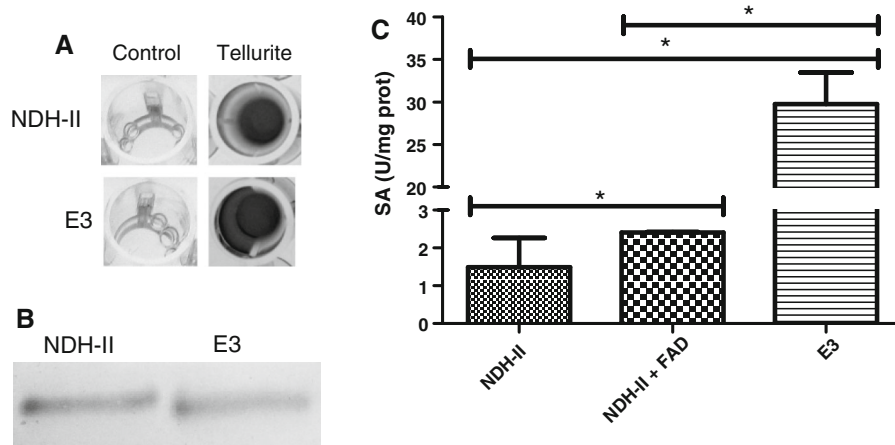


Fig. 2 TR activity of purified *E. coli* NDH-II. TR activity was assayed visually (a) and in situ after fractionation by native PAGE (b), as described in “Materials and methods” section. c TR activity of NDH-II previously incubated with 20 μ M FAD.

Purified *E. coli* dihydrolipoil dehydrogenase (E3) was used as positive control for TR activity. Bars represent the average of three independent trials \pm SE. $P < 0.05$ (*)

presence of NADH-containing TR buffer (Fig. 2a) and also in situ after being fractionated by native PAGE (Fig. 2b).

Since some flavoproteins lose activity upon FAD withdrawing (Hefti et al. 2003), it was of interest to determine if this situation affects DH and TR activities of the NDH-II flavoprotein. Thus, purified NDH-II was preincubated with FAD and assayed for DH and TR activity. The results showed that NDH-II's DH and TR activities increased 56 and 62 %, respectively, in regard to the control without FAD (Fig. S2b and 2c).

NDH-II-mediated tellurite reduction results in superoxide generation

Since (i) enzymes exhibiting TR activity such as catalase generate superoxide (Calderón et al. 2006), (ii) NDH-II generates high ROS amounts under normal conditions (Messner and Imlay 1999) and (iii) that preliminary experiments from our laboratory show that the *E. coli* ETC's terminal oxidases would not participate in ROS generation, these facts support the idea that the tellurite-mediated membrane oxidative

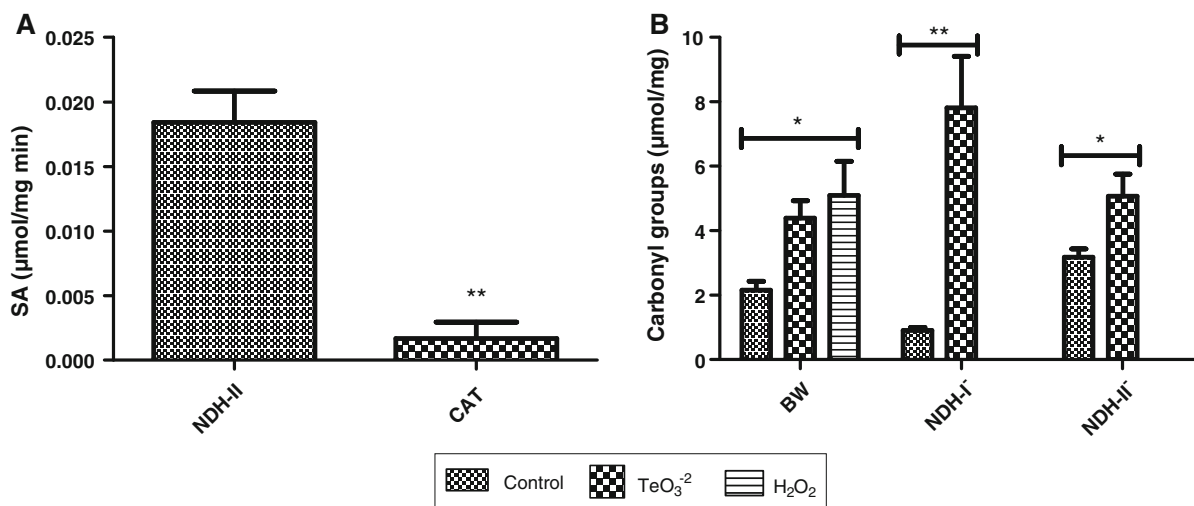


Fig. 3 In vitro superoxide generation by NDH-II and oxidative damage to *E. coli* membrane proteins. **a** superoxide generation by purified NDH-II was determined using the probe WST-1 as described in “Materials and methods” section. Catalase (bovine liver) was used as positive control for O_2^- production.

unbalance would be significantly related to the ETC’s dehydrogenases.

Next, we determined if the in vitro, NDH-II-mediated tellurite reduction results in superoxide generation which in turn, and among other effects, could provoke the loss of activity of some ETC components. The results showed that NDH-II displays higher superoxide levels than catalase (Calderón et al. 2006) (Fig. 3a). Thus, NDH-II can generate ROS at the membrane and as a consequence, an oxidative unbalance that would result in membrane lipid peroxidation (Pradenas et al. 2012). The carbonyl group content was found to increase in membrane proteins from tellurite-exposed cells as compared with the respective controls. This observation was particularly evident for the NDH-I⁻ strain, suggesting the generation of an oxidative unbalance provoked either by the absence of a functional NDH-I complex and/or by an increased amount of NDH-II in the membrane (Fig. 3b). However, regarding the control situation, the NDH-I⁻ strain showed a lower basal carbonylation than that observed for the NDH-II⁻ mutant.

Tellurium content in membranes from tellurite-exposed *E. coli*

To assess if *nuoF* or *ndh* over expression influences the observed increase of in vitro tellurite reduction

b carbonyl group content was assessed in membranes of the indicated tellurite-exposed *E. coli* strains. Hydrogen peroxide (2.5 mM) was used as positive control for oxidative damage. Bars represent the average of three independent trials \pm SE. $P < 0.05$ (*), $P < 0.01$ (**)

(Fig. 1b), membranes from *E. coli* BW25113, p*BnuoF* and p*Bndh* previously grown in the presence of arabinose and exposed to tellurite were analyzed by ICP-OES as described in “Materials and methods” section. While the parental strain showed no difference in the intracellular Te, genetically-complemented mutant strains p*BnuoF* and p*Bndh* showed increased tellurium content. Interestingly p*Bndh* cells showed higher amounts of Te in the presence of arabinose (Fig. 4a). Similar results were obtained with *nuo*, *ndh* and the double mutant *nuo/ndh* strains, where *nuo* cells showed the highest level of elemental tellurium (Fig. 4b). This result rules out the participation of NDH-I-mediated tellurite reduction in vivo and defines the *E. coli* ETC’s NDH-II dehydrogenase as a new membrane enzyme exhibiting TR activity.

Relative gene expression studies in tellurite-exposed *E. coli*

To determine if the enhanced level of tellurite reduction by NDH-II is the result of increased *ndh* expression, assays for relative gene expression were carried out by qPCR after 15 or 30 min of toxicant exposure. At 15 min, while *nuoF* and *ndh* expression was repressed and slightly induced, respectively, that of *sdhB* (which encodes the succinate dehydrogenase

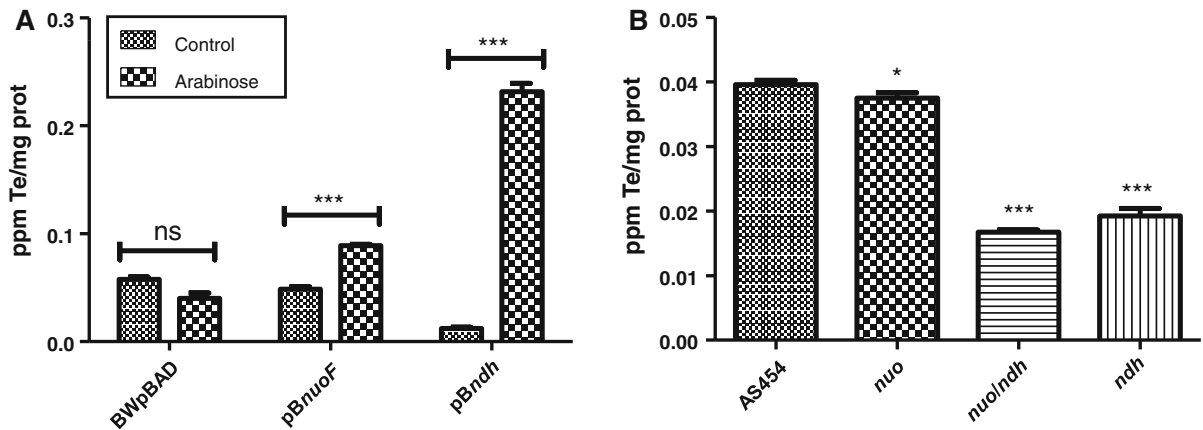


Fig. 4 Intracellular Te content in *E. coli* as determined by ICP-OES. Te content in membranes from genetically-complemented strains previously grown to early exponential phase, induced with 0.02 % arabinose and then exposed to tellurite (a) and in

the indicated mutant strains previously exposed to tellurite (b). Bars represent the average of 3 independent trials \pm SE. $P < 0.05$ (*), $P < 0.001$ (***), NS not significant

B subunit) showed no apparent changes (Table 2, Fig. S3a). In addition, while *fnr* and *arcA* genes (related to aerobic/anaerobic respiration regulation) showed a faint repression, *napA* (encodes the major subunit of cytoplasmic nitrate reductase) and *narZ* (encoding the α -subunit of nitrate reductase Z) showed no expression changes. The only exception was *narG*, encoding a nitrate reductase involved in tellurite reduction at the membrane (Avazeri et al. 1997), whose expression was enhanced over sixfold. These results could be explained since, in addition to the important role of NDH-II in aerobic respiration, the enzyme continues functioning during nitrate respiration (Jackson et al. 2004). The fact that NDH-II and NarG are capable of tellurite reduction is indicative that the toxicant is being continuously reduced at the membrane during aerobic respiration. On the other hand and as expected, *soxS* expression was also found enhanced (fivefold), which most probably reflects the activation of the antioxidant machinery as consequence of tellurite exposure.

After 30 min of toxicant exposure, *ndh* showed an important over expression which may explain the observed increase of tellurite reduction (Figs. 1b, 4b). While the relative expression of *fnr*, *arcA*, *narG* and *narZ* did not show almost any change that of *napA* was highly induced (6.4-fold) (Table 2, Fig. S3b). To date, it is not known if the periplasmic nitrate reductase NapABC is able of tellurite reduction. Most probably pyruvate accumulation occurring upon tellurite

Table 2 Relative expression of the indicated genes in *E. coli* exposed for 15 or 30 min to sub-lethal tellurite concentrations

Gene	Relative expression	
	Tellurite exposure (min)	
	15	30
<i>nuoF</i>	0.4	1.7
<i>ndh</i>	2.6	95.0
<i>sdhB</i>	1.2	0.7
<i>soxS</i>	5.0	2.3
<i>fnr</i>	0.4	1.1
<i>arcA</i>	0.2	1.0
<i>napA</i>	1.0	6.4
<i>narG</i>	6.2	1.3
<i>narZ</i>	1.1	1.4

exposure (Valdivia-González et al. 2012), makes the PDH repressor no longer blocking *ndh* and pyruvate dehydrogenase complex genes (*aceE*, *aceF* and *lpd*) transcription (Ogasawara et al. 2007), thus allowing to maintain the required oxidized NAD^+ pool for pyruvate oxidative metabolism. It is worth to notice that the Lpd dihydrolipoamide dehydrogenase is also a tellurite reductase (Castro et al. 2008). In turn, *soxS* showed a lower induction than that observed at 15 min, suggesting that genes related to the oxidative stress response reach a maximal expression at short times of toxicant exposure.

Finally, in this communication we showed that NADH dehydrogenase and tellurite reductase activities of *E. coli* NDH-II are not affected by tellurite. However, the enzyme can still be harmful because of its ability to generate superoxide while reducing the toxicant. Additional experiments to address this issue are being carried out in our laboratory.

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