

# **Green Fabrication of Polyimide Fibers with Phosphate Monoester Network to Reinforce Carbon Nanotube Polymer Composites: Simultaneously improving the Mechanical Properties and Flame Retardancy**

Hongyang Dang<sup>1</sup> · Yunfeng Cao<sup>2</sup> · Ruqiang Zhang<sup>1</sup> · Zhu Long<sup>1,3</sup> · Shuai Guo<sup>3</sup> · **Shihua Wang3,4 · Zhiqiang Li3,4**

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## **Abstract**

To further enhance the reinforcing efect of polyimide (PI) fbers on carbon nanotube polymer composites, in this work, networks of long chains of phosphate monoester (PMOEs) were grown on the surface of PI fbers, and silanized multiwalled carbon nanotubes (Si-MWCNTs) were dispersed into the phenolic resin (PR) matrix. Subsequently, these two interpenetrating networks were entangled tightly at the PI-PMOEs fber paper/ PR matrix by vacuum-assisted wet papermaking technology and impregnation and vacuum curing technology. A novel type of PI-PMOEs fber-reinforced carbon nanotube polymer composite was prepared. Structural observation and analysis revealed that the fber surface activity and roughness were improved, resulting in enhanced interfacial adhesion. The Si-MWCNTs at the interface play an anchoring role, which leads to a strong mechanical interlock between the PI-PMOEs fbers and the matrix. Therefore, the generation and propagation of cracks encountered greater resistance, and the mechanical properties of the composites were improved. Compared with the Si-MWCNTs/PR composite with tensile strength of 20.29 MPa and elongation at break of 0.1%, the tensile strength and elongation at break of 80PI-PMOEs/Si-MWCNTs/PR (the basis weight of paper is 40 wt.%) composite are increased to 91.22 MPa and 2.0%, respectively. In addition, a tightly connected fber "large skeleton" was provided by the PI-PMOEs fber paper with a uniform pore size distribution. The Si-MWCNTs network and the thermally degraded carbides of the matrix formed a "small skeleton" based on the "large skeleton", resulting in a complete and dense protective barrier layer. The volatilization of pyrolysis products (CO, smallmolecule hydrocarbons, aromatic compounds, etc.) was restricted, and heat and mass transfer were suppressed. Finally, the composites exhibited excellent fame retardancy.

**Keywords** Polyimide fber · Enzyme catalysis · Carbon nanotube polymer composites · Mechanical properties · Flame retardancy

 $\boxtimes$  Zhu Long longzhu@jiangnan.edu.cn

Extended author information available on the last page of the article

### **1 Introduction**

In recent years, carbon nanotube (CNT) polymer composites have been the focus of attention of researchers [\[1](#page-21-0)–[5\]](#page-21-1). Compared with traditional reinforcements, the addition of nano-level materials such as CNTs can efectively improve the internal stress transmission of the composites. Moreover, well-dispersed CNTs will form bridges, which will hinder heat transfer. Even a small amount of additives can signifcantly improve the properties of the polymer matrix  $[6–8]$  $[6–8]$  $[6–8]$  $[6–8]$ . In addition, the interphase regions are commonly formed in polymer nanocomposites, due to the outstanding surface area of nanoparticles and strong interfacial adhesion between matrix and fller phases [\[9](#page-21-4)–[14\]](#page-21-5). And the interphase region can signifcantly enhance the properties of nanocomposites.

However, CNTs have large aspect ratios and specifc surface areas, resulting in the formation of many aggregates in the polymer matrix due to van der Waals forces [\[15](#page-21-6), [16\]](#page-21-7). When the composite is loaded, the aggregated CNT particles cause a concentration of stress on the crack propagation surface [[17\]](#page-21-8). In the process of composite combustion, uniformly dispersed CNTs can act as bridges between the fame retardant fllers in the polymer matrix and help to inhibit thermal degradation [\[18](#page-21-9), [19\]](#page-21-10). Therefore, it is necessary to improve the dispersibility of CNTs and the interfacial compatibility between CNTs and the polymer matrix, which helps to enhance the properties of carbon nanotube polymer composites. Gkikas et al. processed CNTs and an epoxy matrix by ultrasonic treatment, and they found that the fracture toughness of the composite increased approximately 50% when the CNT treatment time was 1 h [[20\]](#page-22-0). Zhang et al. prepared MWCNTs with amino and acyl chloride groups on the surface by chemical functionalization, and the mechanical properties of the MWCNT-reinforced epoxy composite were improved [[21\]](#page-22-1). Wang et al. enhanced the properties of a two-dimensional molybdenum disulfide  $(MoS<sub>2</sub>)/epoxy$  composite with the help of an MWCNT nano bridge. It was found that the MWCNTs were well dispersed in the matrix under the synergistic efect of mixed acid and ultrasonic treatment, and the tensile strength of the composite was improved. Furthermore, the dense barrier network composed of MWCNTs and  $MoS<sub>2</sub>$ nanosheets inhibited the pyrolysis reaction and further hindered the combustion process [[22\]](#page-22-2). It can be found from the above studies that a fexible combination of diferent dispersion methods can play a better role in synergy and further enhance the properties of composites.

The increase in CNT surface activity can improve its dispersibility and compatibility between two phases, which is helpful for preparing high-performance composites. It is undeniable that the efective length of CNT is limited, and the network formed in the matrix is unstable. Consequently, high-strength and high-modulus organic synthetic fbers are considered better candidates for reinforced CNT polymer composites. Micheli et al. investigated a multifunctional composite composed of an aramid fber fabric and epoxy matrix containing CNTs and found that an aramid fabric layer with a thickness of 3.5 mm can efectively resist and absorb energy at a speed of 400 m/s and a strength of 600 J, and at least provide a shielding efect with a frequency of 80 dB [[23\]](#page-22-3). Ahmadi et al. used ultra-high molecular weight polyethylene (UHMWPE) fbers grafted with unidirectional glycidyl methacrylate (GMA) to reinforce an epoxy resin dispersed with carbon nanofbers (CNFs). They found that the combined use of CNFs and GMA-grafted UHMWPE fbers leads to signifcant synergy in the mechanical properties of nanocomposites. In particular, the UHMWPE fbers play a leading role in improving the bending strength of composites  $[24]$  $[24]$ . In conclusion, the introduction

of high-performance organic synthetic fbers results in composite materials with higher mechanical strength and other properties, thereby improving their competitiveness in application felds.

Polyimide (PI) fber is a new member of the family of synthetic fbers, and it has been widely used in the aerospace, military, automotive and construction industries in recent years  $[25-27]$  $[25-27]$  $[25-27]$ . PI fiber is an ordered linear polymer formed by a benzene ring, fivemembered heterocyclic ring, imide ring, carboxyl group, and other structural units [[28](#page-22-7), [29](#page-22-8)]. PI fbers exhibit outstanding thermal stability, prominent mechanical properties, excellent electrical insulation and low dielectric performance due to their unique aramid molecular structure [[30](#page-22-9)[–34\]](#page-22-10). However, PI fber has more inert groups and smaller specifc surface areas, leading to weak interface affinity and difficulty in uniform dispersion in the polymer matrix. The enhancement of the properties of CNT polymer composites by PI fbers is greatly hindered [\[35–](#page-22-11)[37\]](#page-22-12). Therefore, it is necessary to simultaneously improve the interfacial compatibility between the fbers and the polymer matrix and the dispersibility of the fbers in the matrix.

In the present work, we chose green horseradish peroxidase (HRP) as a high-efficiency catalyst, and phosphate monoester (PMOE) with polyhydroxy groups was grown on the surface of PI fbers in an inverse microemulsion environment by free-radical polymerization. The surface activity and hydrophilicity of the fber were improved, and the loss of its excellent properties decreased. Then, a PI-PMOEs fber paper precursor with a uniform pore size distribution was prepared by means of vacuum-assisted wet papermaking technology. In addition, Si-MWCNTs were dispersed in a PR matrix/ethanol solution by high-frequency ultrasound, and the mixture was used for the impregnation of fber paper precursors. Finally, PI-PMOEs fber-reinforced Si-MWCNT/PR composites were prepared by vacuum curing technology. Additionally, the reinforcing results and factors infuencing the mechanical properties and fame retardancy of the composites upon introduction of PI-PMOEs fbers were investigated. Finally, the electrical property of composites was determined. The purpose of this study was to prepare PI fbers with better hydrophilicity by an eco-friendly and less destructive processing method. Moreover, a novel technology was developed to prepare organic fber-reinforced carbon nanotube polymer composites with good mechanical properties and fame retardancy. Broad application prospects in fberreinforced carbon nanotube polymer composites can be expected in future engineering, especially in the feld of fre-resistant composites.

### **2 Experimental**

#### **2.1 Materials**

PI fiber (length = 3 mm, diameter = 11.28  $\mu$ m, density = 1.41 g/cm<sup>3</sup>, strength = 4.38 GPa, elongation at break  $= 5.0\%$ ) was kindly supplied by Aoshen New Material Co., Ltd. (Lianyungang, China). Aramid pulp (XGJP1201, density  $= 1.47$  g/cm<sup>3</sup>, length  $= 0.3$ -0.6 mm, breaking strength = 1.94 GPa) was purchased from Dupont China Holding Co., Ltd. (Shanghai, China). Graphitized multiwalled carbon nanotubes (MWCNTs, HQNANO-CNTs-008-4H) were purchased Tanfeng Graphene Technology Co., Ltd. (Suzhou, China), and the carboxyl group content of the MWCNTs was  $0.36$  wt.% (length =  $10-30$  µm, inner diameter  $= 5{\text -}15$  nm, outer diameter  $> 50$  nm). Thermosetting phenolic resin (PR, 2123-A, free phenol content  $\leq 1.5$  wt.%) was purchased from Chengfeng Resin Industry Co.,

Ltd. (Xingtai, China). Phosphate-buffered saline (PBS,  $pH = 6.9$ ) was purchased from Haibiao Technology Co., Ltd. (Xiamen, China). Horseradish peroxidase (HRP, activity  $\geq$ 250 U/mg) was purchased from Dulai Biotechnology Co., Ltd. (Nanjing, China). N-Dodecyltrimethylsilane (DBS,  $\geq$  96%) was purchased from Hongming Chemical Reagent Co. Ltd. (Jining, China). All other reagents—including hydroxyethyl methacrylate (HEMA,  $\geq$ 96%), phosphorus pentoxide ( $P_2O_5$ , AR), hydroquinone (HQ, AR), n-butanol (NBA, AR), n-octane ( $C_8H_{18}$ , AR), sodium dodecylbenzene sulfonate (LAS, AR), 1,4-dioxane (Diox, AR), sodium percarbonate (SPC,  $\geq$  98%), acetone (CH<sub>3</sub>COCH<sub>3</sub>, AR), ethanol (C<sub>2</sub>H<sub>6</sub>O, AR), sulfuric acid (H<sub>2</sub>SO<sub>4</sub>, AR) and nitric acid (HNO<sub>3</sub>, AR)—were provided by Sinopharm Chemical Reagent Co., Ltd. (Shanghai, China). Deionized water was prepared in the laboratory. All chemicals and materials were used as received without further purifcation. Ethanol ( $\gamma^d = 23.00 \text{ mJ/m}^2$ ,  $\gamma^p = 19.00 \text{ mJ/m}^2$ ) and deionized water ( $\gamma^d = 21.80 \text{ mJ/m}^2$ ,  $\gamma^p =$  $51.00 \text{ mJ/m}^2$ ) were used as fiber contact angle experimental liquids.

#### **2.2 Preparation of PI Fiber with Surface‑Grown PMOEs Network**

To remove the organic impurities from the surface of pristine PI fbers, the pristine PI fbers were treated ultrasonically (200 W, DS-1000Y, Dusi Instrument Co., Ltd., China) in acetone solution for 1 h and then dried at  $60^{\circ}$ C. The resulting fibers were denoted as PI.

Construction of the inverse (W/O) microemulsion system: Approximately 9 mL n-octane was used as the oil phase, and 6 g LAS and 3 mL n-butanol were added as the emulsifer and co-emulsifer, respectively. Finally, 23.4 mL PBS (0.1 M) was added as the aqueous phase. After slight shaking and stirring with a glass rod, a transparent and clear inverse microemulsion was prepared.

Approximately 0.3 g PI fber bundles, 64.8 mL Diox, 8 mL PBS containing 2 mg HRP, and 12.8 g PMOE (10 mL, see the synthetic process in Section I of the Supporting Information) were successively added to a three-necked fask equipped with a thermometer and condensation refux device. After the mixture was degassed for 1 h, approximately 0.45 g SPC was added to the fask in batches for more than 1 h. Next, the temperature was held at 70℃ for 14 h under continuous stirring. Consequently, the fbers were refuxed in a mixture of deionized water and acetone for 24 h and fnally dried at 80℃. The resultant fber was denoted as PI-PMOEs-1. Furthermore, PI-PMOEs-2 and PI-PMOEs-3 were prepared under the above conditions, and the amounts of PMOE were 19.2 g  $(15 \text{ mL})$  and 25.6 g  $(20 \text{ m})$ mL), respectively.

#### **2.3 Fabrication of PI‑PMOEs Fiber‑Reinforced Si‑MWCNTs/PR Composites**

The PI-PMOEs-3 fbers were selected as composite reinforcements and assigned as PI-PMOEs. Scheme [1](#page-4-0) describes the fabrication process of fber-reinforced Si-MWCNTs/PR composites. The PI-PMOEs fbers and aramid pulp were dispersed in a vertical standard fber disintegrator (3000 rpm, PL28-2, Xianyang Taisite Test Equipment Co., Ltd., China) at a mass ratio of 7:3 for 15 min. Next, the fber suspension was poured into a paper former (ZQJ1-B-II; Shaanxi University of Science and Technology Machinery Factory, China), and the PI-PMOEs fiber paper precursor with a basis weight of 50  $g/m^2$  was prepared by vacuum fltration and squeezing.

The procedure for preparing PI-PMOEs fber paper-reinforced Si-MWCNTs/PR composites is as follows: i) Approximately 60 g PR fragments were dissolved in ethanol solution by mechanical agitation. ii) Then, approximately 0.4 g Si-MWCNTs (see



<span id="page-4-0"></span>**Scheme 1.** Schematic illustration for the preparation of PI-PMOEs fber-reinforced Si-MWCNTs/PR composites: (**a**) vacuum-assisted wet papermaking, (**b**) impregnation and vacuum curing.

the silanization process in Section II of the Supporting Information) were uniformly dispersed in the above PR/ethanol solution by ultrasonic vibration for 30 min. iii) Afterwards, the PI-PMOEs fber paper was immersed in a stainless steel container (25  $\text{cm} \times 25 \text{ cm} \times 5 \text{ cm}$ ) containing the above mixture for 5 min. iv) Finally, resin curing and composite moulding processes of the impregnated fber paper were carried out in a temperature-controlled vacuum environment. The curing scheme was maintained for 1 h at 80℃, 4 h at 120℃ and 2 h at 150℃. The fber content in the polymer composites was controlled by changing the concentration of the impregnation mixture.

Furthermore, neat Si-MWCNTs/PR polymer composites without fbers were prepared in a polytetrafuoroethylene mould by using the above raw material ratio and curing scheme. All the details of the formulation are listed in Table [1.](#page-4-1)

Specimen	Components	
	Si-MWCNTs/ $PR(wt.\%)$	Fiber paper $(wt,\%)$
Si-MWCNTs/PR	100	$\Omega$
50g/m <sup>2</sup> PI fiber paper/Si-MWCNTs/PR (50PI/Si-MWCNTs/PR)	75	25
60g/m <sup>2</sup> PI fiber paper/Si-MWCNTs/PR (50PI/Si-MWCNTs/PR)	70	30
70g/m <sup>2</sup> PI fiber paper/Si-MWCNTs/PR (50PI/Si-MWCNTs/PR)	65	35
80g/m <sup>2</sup> PI fiber paper/Si-MWCNTs/PR (50PI/Si-MWCNTs/PR)	60	40
50g/m <sup>2</sup> PI-PMOEs fiber paper/Si-MWCNTs/PR (50PI-PMOEs/Si-MWCNTs/PR)	75	25
60g/m <sup>2</sup> PI-PMOEs fiber paper/Si-MWCNTs/PR (50PI-PMOEs/Si-MWCNTs/PR)	70	30
70g/m <sup>2</sup> PI-PMOEs fiber paper/Si-MWCNTs/PR (50PI-PMOEs/Si-MWCNTs/PR) 65		35
80g/m <sup>2</sup> PI-PMOEs fiber paper/Si-MWCNTs/PR (50PI-PMOEs/Si-MWCNTs/PR)	-60	40

<span id="page-4-1"></span>**Table 1.** Formulation of the Si-MWCNTs/PR and its composites

#### **2.4 Characterization**

The chemical structures of the synthesized PMOE and MWCNT before and after silanization were analyzed by Fourier transform infrared spectroscopy (FTIR, Nicolet is10, Thermo Fisher Scientific, USA) with a resolution of  $4 \text{ cm}^{-1}$  and sixteen scans. It is noted that MWCNT specimens were powdered using an agate mortar under a drying lamp. The surface elemental composition and the functional groups of various fbers were detected by X-ray photoelectron spectroscopy (XPS, Thermo ESCALAB 250XI, Thermo Fisher Scientifc, USA) with a monochromatic Al K source (1486.6 eV). A curve ftting program (XPS Peak Fit software) was used to ft the XPS peaks with one or more Gaussian-Lorentz functions. The various fbers, fractured surfaces of the composites, carbonaceous residue on diferent surfaces after fame combustion and high-temperature treatment of the composites were characterized using scanning electron microscopy (SEM, SU151, Hitachi, Japan). Before imaging, the specimens were coated with gold-platinum to make the specimen conductive to prevent the accumulation of static electric charge during scanning electron microscopy. The thermal stabilities of various fbers and composites were determined by using a thermogravimetric analyzer (TG 209F1, Netzsch, Germany) from 400 to 800°C and 900°C, respectively, with a heating rate of 10℃/min in a nitrogen environment. The particle size of MWCNTs before and after silanization in an ethanol dispersion was estimated by a particle size analyzer (CC-100, Brookhaven, USA). The two dispersions were diluted to 0.01 mg/mL with ethanol before measurement. At least 5 values of particle size for each dispersion were tested, and the average result was taken. The X-ray difraction (XRD, D2 Phaser, Bruker, USA) patterns of the various fbers were recorded using a PANalytical X'Pert PRO diffractometer. The Cu K $\alpha$  ( $\lambda = 0.154$  nm) radiation source was operated at 45 kV and 40 mA with a scan range (2θ) between 5º and 90º. A curve ftting program (Jade software) was used to fit the XRD patterns, and the relative crystallinity  $(X_c)$ was calculated via the following equation:

$$
X_c = \frac{\sum I_c}{\sum I_c + \sum I_a} \times 100\%
$$
 (1)

where  $\sum I_c$  is the diffraction integral intensity of the crystalline region and  $\sum I_a$  is the scattering integral intensity of the amorphous region.

The three-dimensional morphology of the fber surface was analyzed by atomic force microscopy (AFM, Dimension ICON, Bruker, Germany) performed on a multimode instrument in tapping mode using a monolithic Si tip with a resonance frequency between 250 and 300 kHz. In addition, the fber surface roughness, including root mean square roughness  $(R<sub>o</sub>)$  and arithmetic average roughness  $(R<sub>a</sub>)$ , was calculated by the following equations:

$$
R_q = \sqrt{\frac{1}{N^2} \sum_{i=1}^{N} \sum_{j=1}^{N} (z_{ij} - z_{av})^2}
$$
 (2)

$$
R_a = \frac{1}{N} \sum_{i=1}^{N} \sum_{j=1}^{N} |z_{ij} - zcp|
$$
 (3)

where *N* is the number of pixels in the image; *i* and *j* are the pixel locations on the AFM image, respectively;  $z_{ij}$  is the peak height at locations i and j;  $z_{av}$  is the average height within the given area; and  $z_{cp}$  is the height from the center plane.

The contact angles of various fbers were evaluated with a dynamic contact angle tensiometer (DCAT-21, Dataphysics, Germany). After the values of the contact angle to the experimental liquids were determined, the surface free energy  $(\gamma_f)$  of the fiber was calculated in terms of Owens and Wendt's method using the following equations [\[38](#page-22-13), [39\]](#page-22-14):

$$
\gamma_l (1 + \cos \theta) = 2 \sqrt{\gamma_l^d \gamma_l^d} + 2 \sqrt{\gamma_l^p \gamma_l^p}
$$
(4)

$$
\gamma_{\rm f} = \gamma_{\rm f}^{\rm d} + \gamma_{\rm f}^{\rm p} \tag{5}
$$

where  $\theta$  (°) is the contact angle at the fiber/liquid interface;  $\gamma^d$  (mJ/m<sup>2</sup>) and  $\gamma^p$  (mJ/m<sup>2</sup>) represent the dispersive and polar components of the surface free energy, respectively; the subscript *l* indicates the test liquid; and the subscript f indicates the various fbers.

The fber dispersion degree was used to evaluate the dispersion stability of the fber suspension. The fiber/deionized water suspension at a concentration of 3 wt. $\%$  was well prepared. The settlement state of the suspension after standing for 24 h was recorded, and the dispersion degree (f) of the fbers was calculated using the following equation:

$$
f = \frac{v_0 - v_1}{v_0} \times 100\%
$$
 (6)

where  $v_0$  (mL) is the total volume of the suspension initially and  $v_1$  (mL) is the volume of the supernatant in the suspension after standing for 24 h.

A specimen ( $3 \times 3$  cm<sup>2</sup>) was cut from the fiber paper, and the pore size distribution of the fber paper was analyzed by a porous-material pore-size analyzer (CFP-1100A; PMI, USA). The tensile tests of composites were conducted on a universal tensile testing machine (UTM6502, Sans, China) at a tensile rate of 2 mm/min according to the standard ASTM D638, and the specimens were dumbbell-shaped with sizes of 75 mm  $\times$  10 mm  $\times$  2 mm. Five parallel specimens for each composite were employed in the tensile tests. The limiting oxygen index (LOI) values of the composites with splines of 130 mm  $\times$  6.5 mm  $\times$  3.2 mm were determined using an oxygen index analyzer (OI-3, Deatak, USA) according to ASTM D2863. According to ASTM D3801, a vertical combustion tester (CZF-2, Jiangning Analytical Instrument Co., Ltd., China) was used to measure the UL-94 values of the composites, and the specimen dimensions were 120  $mm \times 13$  mm  $\times 3.2$  mm. The heat release rate of the composites was evaluated using a microscale combustion calorimeter (MCC-2, GOVMARK, USA) based on ASTM D7309-07. The specimens were approximately 4 mg and were heated at a constant heating rate of 1℃/s from room temperature to 800°C under nitrogen at 80 cm<sup>3</sup> /min. Then, the volatile, anaerobic thermal degradation products in the nitrogen gas stream were mixed with a 20 cm<sup>3</sup>/min stream of 20% oxygen and 80% nitrogen prior to entering the combustion furnace at 900°C. Two composite specimens of  $2 \times 2$  cm<sup>2</sup> were exposed to an ethanol fame for 120 s, and the combustion process was recorded by a digital camera. The electrical conductivity of the specimens was measured using a semiconductor parameter analyzer (4155C, Agilent Technologies Inc., China). Five parallel specimens for each composite were employed in this test.

## **3 Results and Discussion**

#### **3.1 Surface Growth and Characterization of the PI Fiber**

Horseradish peroxidase (HRP) is mainly produced by the horseradish plant and is a glycoprotein complex composed of a colourless enzyme protein and dark brown iron porphyrin. As a kind of green biological enzyme, HRP is widely used in chemical catalysis as a result of its good stability and the catalytic modifcation of aromatic compounds in the presence of  $H_2O_2$  [[40](#page-23-0)–[42](#page-23-1)]. The growth mechanisms of the irregular PMOEs network on the PI fber surface under HRP catalysis are as follows: i) Chain initiation: Sodium percarbonate (SPC), which is called "solid hydrogen peroxide", is an adduct of  $H_2O_2$  and sodium carbonate and can release  $H_2O_2$  in the solvent [\[43\]](#page-23-2). HRP acts as a catalyst to catalyse the oxidation of PI fibers in the presence of  $H_2O_2$ . Therefore, HRP was oxidized by  $H_2O_2$  to highly active HRP I. HRP I captured a hydrogen atom at the meta position of the amide group of the benzene ring to generate a free electron (represented by PI·). ii) Chain propagation: PI· captured a hydrogen atom on the  $C = C$  group at the end of the PMOE chain to generate the PMOE radical (represented by PMOE·), resulting in homopolymerization of PMOE on the PI fber surface. iii) Polymerization of the monomer PMOE and PI· led the PMOE branched chain to be grafted onto the PI fber main chain. Ultimately, the long chains of phosphate monoester (PMOEs) network grew on the PI fber surface by cross-linking and entanglement of multiple branched chains [[44](#page-23-3), [45\]](#page-23-4) Scheme [2](#page-8-0).

The surface micromorphologies of various fbers are shown in Fig. [1](#page-9-0)a-d. The pristine PI fber presented a smooth and neat surface, which greatly reduced the interface perfor-mance of the fiber (Fig. [1a](#page-9-0)). In contrast, all kinds of PI-PMOEs fibers showed surface morphologies with diferent roughness values. It was found that the lower concentration of the monomer resulted in a smaller molecular weight of the resulting polymer, which could only form intermittent "fsh-scale" structures on the fber surface (Fig. [1](#page-9-0)b). With increasing monomer addition, free radical reactions occurred at more sites on the fber surface, leading to more intense polymerization and increased molecular weight of PMOE. The molecular chain was extended, and "petal" aggregates were grown to the outside of the fber surface (Fig. [1](#page-9-0)c). The appearance of "flament" structures was attributed to the physical entanglement and chemical crosslinking between long chains of the polymer. Most of the long PMOEs chains were closely connected on the fber surface to form a dense network, which increased the coating density of the polymer. The remaining chains extended out of the fber surface (Fig. [1](#page-9-0)d).

To further confrm that the PMOEs network grew on the fber surface, the three-dimensional morphology (Fig. [1e](#page-9-0)-h) and surface roughness (Fig. [1](#page-9-0)i) of the fibers were analyzed by AFM. The pristine PI fber surface was smooth (Fig. [1](#page-9-0)e). With increasing PMOE concentration, the surfaces of all kinds of PI-PMOEs fbers became rougher, and many bumps appeared. Moreover, the fber surface roughness was evaluated by the arithmetic average roughness  $(R_a)$ , root mean square roughness  $(R_a)$  and average maximum height difference  $(R_{tm})$ . It is worth noting that the surface roughness of the fiber was greatly improved only with less monomer addition (Fig. [2f](#page-10-0)). The values of  $R_a$  and  $R_q$  increased from 36 nm and 53 nm to 180 nm and 223 nm, respectively. The aggregation of entangled long chains on the fiber surface increased the  $R_{tm}$  by 197 nm. Up to a PMOE dose of 20 mL (Fig. [1](#page-9-0)h), the  $R_a$  and  $R_q$  values increased to 326 nm and 411 nm, respectively, and  $R_{tm}$  also reached 634 nm. Higher  $R_{tm}$  values indicate that there are more PMOEs networks on the fber surface. The extended polymer molecular chains provided more contact sites



<span id="page-8-0"></span>**Scheme 2.** Schematic illustration of PMOE growth on the surface of polyimide (PI) fber initiated by enzyme (HRP) catalyst.

for the interface between the fber and matrix, which is helpful to improve the mechanical interlock. In addition, the observed micromorphologies proved that the network density and average height of PMOEs can be adjusted by the PMOE concentration during the preparation process.

The changes in the polymer crystal structure of the fbers can be analyzed by the X-ray difraction technique. Figure [1j](#page-9-0) displays the XRD patterns of pristine PI fbers and all kinds of PI-PMOEs fibers, and the relative crystallinity  $(X<sub>c</sub>)$  is shown in Fig. [1](#page-9-0)k [[46](#page-23-5), [47](#page-23-6)]. There are many amorphous regions in the pristine PI fber structure, indicating that the fber is a semicrystalline polymer with a certain intensity diffraction peak and an  $X_c$  of 18.63%. After HRP catalytic treatment, the difraction peak intensity of the fber was weakened, and the crystallinity was slightly decreased. This is mainly due to the fact that PBS in the reaction mixture was alkalescent, while PI fibers have poor alkali resistance. It is difficult to remove the small molecules by Soxhlet extraction because they penetrated into the fber after the surface structure of the fber changed. Therefore, the stereoregular structure of the fber polymer chain was slightly damaged, and the tight packing between the macromolecular chains was weakened, which further reduced the crystallization performance.

The specifc contents of chemical elements and functional groups on the surface of the fber can be analyzed by XPS [\[48\]](#page-23-7). Figure [2](#page-10-0) shows the XPS spectra of pristine PI fber



<span id="page-9-0"></span>**Fig. 1.** SEM micrographs and AFM micrographs of (**a** and **e**) pristine PI fber, (**b** and **f**) PI-PMOEs-1 fber, (**c** and **g**) PI-PMOEs-2 fber and (**d** and **h**) PI-PMOEs-3 fber; (**i**) results of roughness of various fbers; (**j**) XRD patterns of various fbers; (**k**) The relative crystallinity of various fbers

and various PI-PMOEs fibers. Each element  $(C, O, and P)$  peak was fitted into multiple constituent peaks to determine the relative contents of diferent functional groups. Three markedly intense peaks (O1s, N1s, and C1s) were displayed in the spectrum of pristine PI fber. It is noted that the new P2p and P2s peaks contained in the spectra of all kinds of PI-PMOEs fbers, and we chose the P2p peak as the quantitative analysis standard for P. The PMOEs on the surface of various PI-PMOEs fbers contain a large number of oxygencontaining groups (O-C,  $O = C$ ,  $O = P$ , and O-P), leading to an increase in the intensity of the O1s peak. The PMOEs network growing on the fber surface limited the detection of N only in the fber structures since the detection depth of XPS technology is approximately 10 nm, which explained the weakening of the N1s peak intensity. The intensity of the element peak is determined by the proportion of the element, so an increase in the other elemental peak intensities led to a decrease in the C1s peak intensity.

Furthermore, the C1s spectrum of the pristine PI fber exhibited four obvious peaks (Fig. [2](#page-10-0)b), which are located at  $284.7$  eV (C-C/C-H),  $285.8$  eV (C-N),  $286.4$  eV (C-O), and [2](#page-10-0)88.2 eV (N-C = O). An additional peak in the C1s spectrum (Fig. 2e, h, and k) was observed at 288.6 eV ( $O-C = O$ ). Simultaneously, the contents of the C-O and  $O-C = O$ groups increased gradually with increasing PMOE concentration and increased to 29.9% and 7.4%, respectively (Fig. [2m](#page-10-0)). Moreover, the O1s spectrum of pristine the PI fber (Fig. [2](#page-10-0)c) exhibited two peaks at 531.9 eV (O-C) and 533.5 eV (O = C). The O1s spec-trum of the PI-PMOEs-1 fiber (Fig. [2](#page-10-0)f) showed two additional peaks at 532.3 eV ( $O = P$ ) and 532.8 eV (O-P). The contents of  $O = P$  and O-P groups in the PI-PMOEs-3 fiber also reached 10.8% and 19.1%, respectively (Fig. [2n](#page-10-0)), indicating that the PMOEs network successfully grew on the fiber surface. The P2p splitting peaks at 133.8 eV ( $2p_{1/2}$ ) and 134.7



<span id="page-10-0"></span>**Fig. 2.** (**a**) Wide scan XPS spectra and calculated elemental concentrations of various fbers; (**b**) C1s and (**c**) O1s XPS spectra of pristine PI fber; (**d**) P2p, (**e**) C1s and (**f**) O1s XPS spectra of PI-PMOEs-1 fber; (**g**) P2p, (**h**) C1s and (**i**) O1s XPS spectra of PI-PMOEs-2 fber; (**j**) P2p, (**k**) C1s and (**l**) O1s XPS spectra of PI-PMOEs-3 fber; (**m**) C1s and (**n**) O1s curve ftting results of various fbers

eV ( $2p_{3/2}$ ) were found in the P2p peak spectra of the three kinds of PI-PMOEs fibers, which further confrmed the existence of PMOEs [[49](#page-23-8), [50\]](#page-23-9).

Thermogravimetric analysis (TGA) under a  $N_2$  atmosphere was used to evaluate the thermal stability of the fbers [\[51](#page-23-10)]. The TGA and DTG curves of various fbers are shown in Fig. [3a](#page-11-0)-b, and the typical thermodegradation data, including the initial degradation temperature ( $T_{di}$ ), the maximum degradation rate temperature ( $T_{max}$ ) and the char yield ratio  $(Y_c)$ , are summarized in Table [2](#page-12-0). The pristine PI fiber showed a maximum degradation



<span id="page-11-0"></span>**Fig. 3.** (a) TGA and (b) DTG thermograms of various fibers under  $N_2$  atmosphere; (c) contact angles and (**d**) surface free energy of various fbers; (**e**) dispersion degree and dispersion state of fbers in water; (**f**) Pore size distribution of fber paper

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Specimen	Phase 1			Phase 2		
	$T_{di,1}$ (°C)	$T_{\text{max},1}$ (°C)	$Y_{c,1}$ at 500°C $(wt,\%)$	$T_{di,2}$ (°C)	$T_{\text{max},2}$ (°C)	$Y_{c,2}$ at $900^{\circ}$ C $(wt.\%)$
PI				490.1	605.8	52.70
PI-PMOEs-1				472.5	603.2	51.81
PI-PMOEs-2	٠		٠	475.1	604.6	51.92
PI-PMOEs-3	279.6	334.5	97.81	480.3	605.2	49.80

<span id="page-12-0"></span>**Table 2.** Typical thermodegradation data of various fibers under N<sub>2</sub> atmosphere

rate at 605.8℃ and one mass loss stage at 490.1-800.2℃, which indicated that the PI fber exhibited excellent thermal stability. Comparatively, another weight loss of approximately 2.19 wt.% was observed at 279.6℃-503.5℃ only in the TGA thermogram of the PI-PMOEs-3 fber, which is likely attributed to the thermal decomposition of the PMOEs network growing on the fber surface. This phenomenon is due to the small molecular weight of the polymers on the surface of PI-PMOEs-1 and PI-PMOEs-2 fbers, and a more complete PMOEs network was not formed. As mentioned above, the molecular chain regularity of the fber was destroyed after the HRP catalytic treatment. Therefore, the high-temperature resistance of the fibers was affected to some extent, and the  $Y_c$  value at 900°C of all kinds of PI-PMOEs fbers was slightly decreased.

The purpose of this study is to improve the dispersibility of PI fbers in the medium, and the increase in surface free energy is conducive to promoting the interfacial properties and hydrophilicity of the fber. Therefore, the contact angles between the experimental liquids and the surface free energy were measured and calculated in Fig. [3c](#page-11-0)-d to evaluate the wettability of the various fbers. It was observed that the contact angles of the pristine PI fber with strongly polar deionized water and weakly polar ethanol were 89.3º and 78.2º, respectively. In contrast, the contact angles of all kinds of PI-PMOEs fbers with two kinds of wetting liquids decreased significantly. The surface free energy (γ<sub>f</sub>) of PI-PMOEs-3 fiber was  $42.70 \text{ mJ/m}^2$ , which was an increase of  $34.36\%$  compared to that of the PI fiber. The increase in various PI-PMOEs fber surface free energies was attributed to the fber's higher surface roughness and larger specifc surface area, so the fber was more likely to overcome the surface tension of the liquids and the liquids were easier to spread on the fber surface. It is interesting to note that the increase in surface free energy of various PI-PMOEs fbers was mainly due to the contribution of the polar component ( $γ<sup>p</sup>$ ), which resulted in enhancement of the surface polarity, inductive efect and the infuence of hydrogen bonds. Furthermore, an increase in the number and types of polar functional groups led to higher affinity of three kinds of PI-PMOEs fbers with deionized water than of ethanol. Therefore, it was found that the contact angle of PI-PMOEs-3 fber with deionized water decreased by 13.6°, while that with ethanol decreased by only 9.9°.

To evaluate the dispersion stability of various fbers more intuitively, the dispersion degree of a fber suspension and the pore size distribution of fber paper were measured and are shown in Fig. [3](#page-11-0)e-f. The dispersion degree of pristine PI fbers was 10.0%, while the dispersion degree of all kinds of PI-PMOEs fbers increased. The dispersion degree of the PI-PMOEs-3 fbers reached 43.6%, a relative increase of 33.6%. The pore size of the pristine PI fber paper prepared by vacuum-assisted wet papermaking technology varied. By comparison, the pore size of the PI-PMOEs-3 fber paper was smaller, and the pore size distribution

was relatively uniform. The proportion of pores with a diameter of 20-50 μm was 57.58%, and that of pores with a diameter of 20-30 μm was 27.53%. This phenomenon is attributed to the high inertness of the pristine PI fber surface, which will form compact "cotton ball" or "focculent" fber clusters under van der Waals forces. Once this structure formed, the fbers became more hydrophobic and hindered the dispersion of other fbers. The poor dispersion stability of the fber suspension also resulted in more fber bundles and an uneven number of fber layers in the fber paper. However, there were more polar groups on the rough surface of the PI-PMOEs-3 fber, which led to a higher afnity of the fber for the aqueous media and stronger repulsion between the fbers. The fber bundles were more easily dispersed into single fbers and formed a large three-dimensional network with a uniform pore size in the process of vacuum-assisted wet papermaking. In conclusion, the hydrophilicity of PI-PMOEs fbers improved with increasing PMOE concentration, which is benefcial for enhancing the interfacial properties of the fbers. Moreover, the paper formation of PI-PMOEs-3 fbers fabricated by vacuum-assisted wet papermaking technology also improved.

### **3.2 Mechanical Properties of PI‑PMOEs Fiber‑Reinforced Si‑MWCNTs/PR Composites**

The mechanical properties of Si-MWCNTs/PR composites containing diferent amounts of PI fbers and PI-PMOEs fbers were evaluated. Figure [4a](#page-14-0)-b shows the resulting tensile strength and elongation at break of each of the composites. Obviously, Si-MWCNTs/PR composites with better tensile properties can be obtained by introducing longer fexible PI fbers compared with those of PR composites with only Si-MWC-NTs as reinforcement fllers. It was found that the tensile strength improved linearly with increasing fber content from 20.29 MPa (Si-MWCNTs/PR) to 75.02 MPa (80PI/ Si-MWCNTs/PR). This is due to the fact that PI fbers with high strength and a high modulus improved the energy consumption of the 80PI/Si-MWCNTs/PR composites during break, resulting in enhanced mechanical properties of the composites [[52](#page-23-11)]. It is worth noting that the tensile strength of the composite with PI-PMOEs fibers as a reinforcing fller further increased by 21.59% to 91.22 MPa (80PI-PMOEs/Si-MWCNTs/ PR). This phenomenon is most likely due to the anchoring of the PMOEs network and Si-MWCNTs network constructed at the interface between the PI-PMOEs fbers and the matrix. The interface adhesion was enhanced, and the stress transmission efficiency was improved. On the other hand, the pore size distribution of PI-PMOEs fber paper was more uniform (Fig. [3f](#page-11-0)), and the contact area between the matrix and PI-PMOEs fbers increased. The combination of the two phases was stronger, which is benefcial to the transfer of stress from the matrix to the fber via the shear transfer mechanism [[53](#page-23-12)]. In addition, it was found that the elongation at break of the composites showed an increasing trend similar to that of the tensile strength. This results from the fact that the failure mode of the composites gradually transitioned from brittle breakage to ductile fracture, and the ductility of the composites was improved. Compared with the straight and smooth PI fber surface, PI-PMOEs fbers with higher roughness can interact with the matrix to retard fber pullout and promote fber rupture, so the elongation at break was further improved.

After that, the fracture mechanism of the fiber-reinforced Si-MWCNTs/PR composites was qualitatively analyzed by observing the micromorphologies of the fracture surfaces in Fig. [4c](#page-14-0)-h. It can be seen from Fig. [4](#page-14-0)c-e that the PI fibers drawn from the composite have smooth and neat surfaces, and the debonding and pullout of the



<span id="page-14-0"></span>**Fig. 4.** Tensile properties of Si-MWCNTs/PR and its composites: (**a**) tensile strength, (**b**) elongation at break; SEM micrographs of fracture surfaces: (**c**-**e**) 80PI/Si-MWCNTs/PR composite, (**f**-**h**) 80PI-PMOEs/ Si-MWCNTs/PR composite

fibers were the main means of energy consumption during fracture. The propagation of cracks during fracture of the sample was hindered, which increased the tensile strength of the composite. However, it is undeniable that the interface between the matrix and the fibers was still poor, and the mechanical properties of the composite were less than expected. It is obvious that the fracture surface of the 80PI-PMOEs/ Si-MWCNTs/PR composite showed a highly rough PI-PMOEs fiber surface, and the pulled fiber surface adhered to a large amount of matrix (Fig. [4f](#page-14-0)-h). The improvement of fiber surface activity enhanced the interfacial adhesion. On the other hand, the two networks interpenetrate at the resin-fiber interface, which further increases the interfacial bonding strength. In the end, a stronger mechanical interlock formed between the PI-PMOEs fibers and the resin matrix. Furthermore, the uniformly dispersed Si-MWCNTs in the matrix bore part of the external load, and the load transferred to the fiber was reduced, which plays a toughening role in the composite [\[54](#page-23-13)]. In general, the relatively concentrated stress in the composite was released, and the resistance of crack generation and propagation was increased, so the mechanical properties were further improved.

## **3.3 Thermal Stability and Flame Retardancy of PI‑PMOEs Fiber‑Reinforced Si‑MWCNTs/PR Composites**

Thermal degradation is primarily important for assessing the fre resistance of composites. The TGA and DTG curves of Si-MWCNTs/PR and all 80PI-PMOEs/Si-MWCNTs/ PR composites are shown in Fig. [5a](#page-15-0), b and summarized in Table [3.](#page-16-0) It was found that the thermal degradation of the Si-MWCNTs/PR composite exhibited three weight loss stages. The first weight loss, observed in the range of  $100-220^{\circ}$ C, was due to the residual moisture after PR curing and small molecules such as formaldehyde volatilized by the PR after further condensation. The weight loss at 360-550℃ corresponded to the main stage of the thermal degradation of PR. The process is that PR further crosslinks and cyclizes to form carbides and releases hydrocarbon and alcohol pyrolysis products such as methane [[55](#page-23-14)]. The fnal stage of pyrolysis observed at 570-660℃ was attributed to PR dehydrogenation and carbonization. At the same time, a certain amount of small molecules, such as CO,  $CO<sub>2</sub>$ , and  $CH<sub>4</sub>$ , were released. The data in Table [3](#page-16-0) also show that the residual char mass of the Si-MWCNTs/PR composite was the lowest at 600℃ and 800℃. In contrast, when PI-PMOEs fbers were introduced into the Si-MWCNTs/PR composite, the composites showed a greater increase in the residue after thermal degradation. The increase in the residual amount results in further enhancement of the thermal stability of the composite. Interestingly, the PI-PMOEs/Si-MWCNTs/PR composites exhibited fve stages of thermal



<span id="page-15-0"></span>**Fig. 5.** (a) TGA and (b) DTG thermograms of Si-MWCNTs/PR and its composites under  $N_2$  atmosphere; (**c**) heat release rate and (**d**) total heat release curves of Si-MWCNTs/PR and its composites

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<span id="page-16-0"></span>

a T<sub>10wt.</sub>% is defined as the temperature of 10 wt.% weight loss in the TGA curve.

decomposition. The moisture evaporation and thermal decomposition of the volatilization of small molecules were advanced to 40-90℃ and 90-180℃, respectively. This may have resulted from the large three-dimensional interconnected structure formed between PI-PMOEs fbers, which provided bridges for Si-MWCNTs, while the Si-MWCNTs with high thermal conductivity accelerated heat transfer and promoted thermal decomposition of composites. However, the thermal conductivity of the PI-PMOEs fber itself is poor, and it cannot provide enough heat to support the next stage of thermal decomposition in time. Therefore, the decomposition temperature range of the remaining small molecule groups in PR was delayed to 200-300℃, and the maximum decomposition rate was relatively low. This stage was followed by further heat transfer within the composite, and the main crosslinking and carbonization stages of PR thermal degradation also occurred at 300- 450℃. This is due to PI-PMOEs fbers have good high-temperature resistance and form a strong network of "large skeleton" after uniform dispersion, and the Si-MWCNTs and carbonized PR matrix also formed "small skeleton" network connection. The two networks overlap to establish an efective physical barrier layer, which greatly reduces the maximum thermal degradation rate. The last temperature range of 500-700℃ was attributed to the thermal decomposition of PI-PMOEs fbers, which also corresponded to the results of the thermal decomposition of fbers in Fig. [4a](#page-14-0). In general, using PI-PMOEs fbers as the reinforcing fllers of Si-MWCNTs/PR composites helps to improve the thermal stability and enhance the fame retardancy.

Microcombustion calorimetry (MCC) is a new, convenient and fast detection method. As a small-scale fammability test technology, MCC requires only a few milligrams of sample to measure the heat release when chemicals are decomposed during the combustion process and to eliminate physical factors such as swelling, dropping and shielding unrelated to the test results [\[56\]](#page-23-15). The peak heat release rate (PHRR) and the total heat release rate (THR) refect the potential combustion characteristics of the composite and evaluate its fre resistance. A lower value indicates that the composite is less fammable. As shown in Fig. [5](#page-15-0)c-d, the PHRR of the Si-MWCNTs/PR composite reached 611.3 W/g at 360- 550℃, and the THR was as high as 154.6 KJ/g. This result indicated that the composite released a large amount of heat during the combustion process and had high fammability. With the increase in PI-PMOEs fber content in the composites, the PHRR at 300-450℃ decreased gradually to 262.3 W/g. The increase in PHHR at 500-700°C was due to the thermal decomposition of PI-PMOEs fbers. In addition, the THR of PI-PMOEs/Si-MWC-NTs/PR composites also showed a similar downward trend as that of PHRR, with a maximum decrease to 55.7 KJ/g and a relative decrease of 64.0%. This further illustrated that a stable char layer supported by fbers was formed after carbonization of the composite,

<span id="page-17-0"></span>

which hindered the further transfer of heat and mass. The above results are consistent with the TGA results of the composites, which proved that PI-PMOEs fber-reinforced Si-MWCNTs/PR composites have potential freproof application value.

The combustibility of Si-MWCNTs/PR and PI-PMOEs/Si-MWCNTs/PR composites was investigated by LOI and UL-94 testing, as listed in Table [4](#page-17-0). It was observed that the LOI value of the Si-MWCNTs/PR composite was only 23.0%. The composite failed the UL-94 test, combustion decreased, and self-extinguishing occurred during dropping. When the mass ratio of PI-PMOEs fiber paper was 25 wt.%, the LOI value of the composite reached 29.8% and passed the UL-94 V-1 level rating. Upon increasing the amount of PI-PMOEs fbers, the LOI values of the 60PI-PMOEs/Si-MWCNTs/PR, 70PI-PMOEs/Si-MWCNTs/PR and 80PI-PMOEs/Si-MWCNTs/PR composites increased to 32.1%, 33.4%, and 35.0%, respectively, and the 80PI-PMOEs/Si-MWCNTs/PR composite passed the UL-94 V-0 rating.

To analyze the fre resistance of fber-reinforced composites more intuitively, Si-MWC-NTs/PR and 80PI-PMOEs/Si-MWCNTs/PR composites were exposed to a 500-700℃ ethanol fame and a 500℃ high-temperature environment, and the relevant combustion process and sample appearance changes are shown in Fig. [6a](#page-18-0)-b for evaluation. As shown in Movie S1 in the Supporting Information, the Si-MWCNTs/PR and 80PI-PMOEs/Si-MWCNTs/ PR composites were simultaneously moved into the ethanol fame, while they exhibited opposite fame-retardant behaviors. It can be seen that the Si-MWCNTs/PR composite was completely burned in approximately 30 s, during which black smoke escaped and dropped obviously. In comparison, the 80PI-PMOEs/Si-MWCNT/PR composite material exhibited better fre resistance, and there was no drop or black smoke during the combustion process. Even if the material was continuously exposed to high-temperature fame for 120 s, it was still inert to the fame. At the end of combustion, the composite maintained the contracted state of the original physical form without any change in shape. Moreover, after heat treatment at 500℃ for 10 min, the Si-MWCNTs/PR composite exhibited expansion and structural damage, while the 80PI-PMOEs/Si-MWCNTs/PR composite still showed good structural integrity without shrinkage. The above phenomena indicated that PI-PMOEs fbers as fllers efectively improved the resistance of Si-MWCNTs/PR composites to high temperature and fame.

The carbonaceous residual plays a crucial role in the fame retardant and smoke suppression performances, so increasing the char layer is an efective method to judge the improvement of fame retardancy of the composite. By studying the appearances and micromorphologies of the composites after combustion, the char residues can also be observed, and the fame-retardant mechanism can be analyzed. As mentioned above,



<span id="page-18-0"></span>**Fig. 6.** (**a**) Combustion processes of Si-MWCNTs/PR and 80PI-PMOEs/Si-MWCNTs/PR composites; (**b**) digital images of Si-MWCNTs/PR and 80PI-PMOEs/Si-MWCNTs/PR composites before and after treatment at 500℃ for 10 min;(**c**) outer surface and (**e**) inner surface SEM micrographs of Si-MWCNTs/PR composite after treatment at 500°C for 10 min; (**d**) surface and (**f**) cross-sectional surface SEM micrographs of 80PI-PMOEs/Si-MWCNTs/PR composite after exposing it to the ethanol fame

Si-MWCNTs/PR composite has poor flame retardancy, and it is difficult to preserve a complete physical form after fame burning. Therefore, the micromorphologies of the composite surface after treatment at 500℃ for 10 min were observed and are shown in Fig. [6c](#page-18-0) and e. It is obvious that many globular particles with diferent sizes appeared on the outer surface char layer and were independent of each other. This is due to agglomeration of the resin-coated Si-MWCNTs upon carbonization and shrinkage of the matrix. However, this only occurred in a small region of the outer surface. As shown in Fig. [6](#page-18-0)b, the Si-MWCNTs/PR composite expanded, and a large number of holes appeared after high-temperature treatment. This phenomenon was attributed to the crosslinking and carbonization of the PR matrix and the release of gaseous molecules when the composite was exposed to the environment with a rapid temperature increase. Then, with the rapid increase in internal pressure, sufficient external thrust was generated, leading to the volume of the composite increasing sharply after heating. Many large-scale holes appeared in the composite concurrently (Fig. [6e](#page-18-0)), which also provided a channel for the release of combustible volatiles and the transfer of fame heat. In contrast, as shown in Fig. [6d](#page-18-0) and f, the surface of 80PI-PMOEs/Si-MWCNTs/PR exhibited a complete carbon layer and almost no pores after fame combustion. A tightly connected carbon layer appeared between the fber layers, and the hole and crack size were relatively small. In general, the fame retardant mechanism of PI-PMOEs/Si-MWCNTs/PR composites can be discussed in terms of three factors. The main reason for the observed results is that the "small skeleton" formed by the Si-MWCNTs and char layer after PR thermal degradation and the "large skeleton" formed by PI-PMOEs fbers jointly establish a continuous and dense network protective layer, which reduces the transportation efficiency of combustible gases such as oxygen. Second, the PI-PMOEs fbers have poor thermal conductivity and conjugated structures of aromatic heterocycles in the macromolecular chains. This provides high thermal oxygen stability for the fbers and inhibits the mass transfer from the matrix to the heat source. Finally, the noncombustible nitrogencontaining inert gas will be released during the thermal degradation of PI-PMOEs fbers, which dilutes the concentration of the combustible gas. The conditions that support combustion are destroyed, thereby preventing further thermal degradation inside the composite. The above results demonstrate that PI-PMOEs fbers have great potential as efective fre-retardant fllers for composites and can be used as a continuous layered structure to hinder the spread of fames.

#### **3.4 Electrical Properties of PI‑PMOEs Fiber‑Reinforced Si‑MWCNTs/PR Composites**

We also measured the electrical conductivities of Si-MWCNTs/PR and PI-PMOEs/ Si-MWCNTs/PR composites, as listed in Table [5.](#page-19-0) Since Si-MWCNTs/PR composite is prepared in the mould, it has almost the same electrical conductivity in the plane and the thickness directions. It also shows good electrical conductivity. When PI-PMOES fiber is used as reinforcement filler, the electrical conductivity of the composite only decreases slightly in the plane direction. However, the electrical conductivity decreased a lot in the thickness direction. Even when the fiber paper reached 40 wt.% (the base weight of the paper was  $80g/m^2$ ), the electrical conductivity decreased to  $5.4 \times 10^{-12}$  S/cm. The fibers in this article are composite with matrix in the form of paper. According to the seepage theory, when the fiber fillers are less, Si-MWCNTs nanoparticles can form a bridge and network along the thickness direction, and a complete conductive pathway can be formed. According to the percolation theory, we infer that Si-MWCNTs nanoparticles bridges and networks are formed along the thickness direction when the fiber filler is less [[57–](#page-23-16)[62\]](#page-24-0). Finally, a complete conductive pathway is formed. However, the filler reaches the percolation threshold with the increase of the proportion of fiber paper and the composite changes from the electrical conductor to insulator. In addition, PI-PMOES fiber and aramid fiber are the fillers with excellent insulation properties, which further reduce the conductivity of the composite in the thickness direction.

<span id="page-19-0"></span>

In conclusion, we have demonstrated a simple and efective strategy for preparing a novel PI-PMOEs fber-reinforced Si-MWCNTs/PR composite with excellent mechanical and fame-retardant properties by using PI-PMOEs fbers catalysed by HRP as fllers. A new reinforcement method of polymer composite is introduced, in which the fber is introduced into the composite material in the form of fber paper. It was observed that PMOEs were closely connected to the surface of PI fber. The PMOEs network on the fber surface could be controlled by the concentration of the monomer. Furthermore, the surface roughness and active sites of the PI-PMOEs fber increased. This improved the interfacial adhesion between the fber and polymer matrix. Moreover, the pore size distribution of the fber paper prepared by vacuum-assisted wet papermaking technology was uniform, which was attributed to the enhancement of the hydrophilicity of the fber and was conducive to improving the stress transfer efficiency. Compared with that of the Si-MWCNTs/PR composite, the tensile strength of the 80PI-PMOEs/Si-MWCNTs/ PR composite increased signifcantly by 267.9% when the fber paper content was 40 wt.%. The composite also exhibited good fame retardancy, with a char residue rate of 48.8% and an LOI value of 35.0% at 800℃. It maintained the original physical shape after high-temperature burning of the ethanol fame. However, the electrical property of the composite became poor due to the percolation threshold and the insulating property of PI fiber. This work provides an eco-friendly and efficient method to improve the surface activity of PI fbers while maintaining their excellent original properties. It is expected that this work will provide a fundamental understanding of the remarkable action of the high-performance chemical synthetic fber incorporated with carbon nanotube polymer composites on the substantial improvement of mechanical properties and fame retardancy.

## **5 Supporting Information**

Electronic Supplementary Information (ESI) available: synthesis of phosphate monoester (PMOE) and silane functionalization of multiwalled carbon nanotubes (MWCNTs) in Section I and Section II, respectively. Additionally, the burning process of Si-MWCNTs/PR composite and 80PI-PMOEs/Si-MWCNTs/PR composite are shown in Movie S1.

**Supplementary Information** The online version contains supplementary material available at [https://doi.](https://doi.org/10.1007/s10443-021-09878-y) [org/10.1007/s10443-021-09878-y](https://doi.org/10.1007/s10443-021-09878-y).

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## **Declarations**

**Confict of Interest** The authors declare no competing fnancial interest.

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## **Authors and Affiliations**

## Hongyang Dang<sup>1</sup> · Yunfeng Cao<sup>2</sup> · Ruqiang Zhang<sup>1</sup> · Zhu Long<sup>1,3</sup> · Shuai Guo<sup>3</sup> · **Shihua Wang3,4 · Zhiqiang Li3,4**

- <sup>1</sup> Key Laboratory of Eco-Textiles, Ministry of Education, College of Textile Science and Engineering, Jiangnan University, Jiangsu Wuxi 214122, China
- <sup>2</sup> College of Light Industry and Food Science, Nanjing Forestry University, Nanjing 210037, China
- <sup>3</sup> Lianyungang Industrial Investment Group Co., Ltd, Jiangsu Lianyungang 222002, China
- <sup>4</sup> Lianyungang Fiber New Materials Research Institute Co., Ltd, Jiangsu Lianyungang 222002, China