

Morphology, pore size distribution, and nutrient characteristics in biochars under different pyrolysis temperatures and atmospheres

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Abstract To evaluate the agronomic potential of biochar, we prepared a series of biochars using rice straw waste under the limited oxygen cracking condition (CO₂ or N₂) and the different pyrolysis temperatures including 300, 400, 600, and 800 °C. The results showed that morphology structure, specific surface area, pore size distribution, and element contents of the biochars were superior to the biochars prepared under traditional inert atmosphere (N₂) and the same four pyrolysis temperatures. In comparison with the rice straw, pore structure of biochars was mainly mesoporous and more developed, average pore size decreased, and BET-specific surface area increased with the increase of temperature from 300 to 800 °C. Biochars distributed abundant mesopores and macropores under 400 and 600 °C; the maximum macropores of the biochar were shaped under 600 °C. Concentration of phosphorus (P) and potassium (K) increased significantly with increasing temperature, while that of nitrogen (N) first increased and then decreased and reached a maximum at 400 °C. In addition, taking these physiochemical properties into consideration, we drew a conclusion that the optimum quality biochar could acquire under the work conditions of 400 °C and CO₂ atmosphere, which was supposed to provide theoretical guidance for biochar returning to soil.

Keywords CO₂ atmosphere · Biochars · Nutrient content · Pore structure

Introduction

Biochar is the solid product of biomass pyrolysis when the feedstocks are heated under complete or partial exclusion of oxygen [1–3]. Various common residual forms of the biomass, including wood, rice husk, straw, bamboo, and animal wastes, can be utilized as a feedstock of biochar for the pyrolysis process [4]. In recent years, biochar has received considerable interest in recent years due to its potential roles in carbon sequestration and soil amendment [5–7]. Biochar characteristics change widely according to preparation conditions (e.g., temperature, atmosphere, raw material, pyrolysis rate, residence time, etc.) [8, 9]; therefore, a wide diversity of its composition and property is expected [10, 11]. Biochar derived from wood has higher carbon (C) amount than those obtained from herbaceous feedstocks [12, 13], but they are depleted in nitrogen (N) [14]. Conversely, livestock manures are more enriched in nitrogen, phosphorus, and micronutrients, and as a consequence, their pyrolysis produces biochars with higher nutrient content [15]. Extensive analyses contribute to a better understanding of thermochemical transformations that take place during pyrolysis, providing support to the consumers on the use of a specific biochar for each particular agro-ecosystem. Table 1 summarizes the studies about the effects of preparation conditions on the physicochemical properties of biochar in recent years.

Seen from Table 1, scholars have studied mostly about the effects of feedstocks and pyrolysis temperatures on the physicochemical properties of biochars, and the preparation atmosphere they choose mostly are N₂ atmosphere or anoxybiotic conditions. CO₂, one of the weak oxidation gases, has been widely used as the gasification agent for the synthesis gas and the activator for the preparation of activating CO₂, while it is not widely used in the biochar preparation process as the pyrolysis atmosphere (especially

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Table 1 Evolution of biochar characteristics/functions along different preparation conditions (temperatures, feedstocks, and atmospheres)

Characteristics	Temperatures: low → high	References
Carbon categories	Transition char → amorphous char → composite char → turbostratic char (pine wood shavings and tall fescue grass, 100–700 °C, oxygen-limited conditions)	[8]
Electrical conductivity	Increased first and then decreased and reached its peak at 400 °C (sewage sludge, 300–700 °C, N ₂ atmosphere)	[25]
Formation mechanism	Dehydration → condensation/aromatization (honey mesquite, 200–650 °C, oxygen-limited conditions)	[40]
Field capacity	Increased first and then decreased (corn stover, magnolia leaves, and apple wood, 300–700 °C, oxygen-limited conditions, high field capacity, and low hydrophobicity at pyrolysis temperatures between 400 and 600 °C)	[41]
Specific surface area	Increased constantly (Douglas fir wood, Douglas fir bark, and poplar wood, 350–600 °C, N ₂ atmosphere)	[42]
	Increased first and then decreased and reached its peak at 600 °C (pine wood, green waste, wheat straw, and spray-dried algae, 300, 450, 600, and 750 °C, N ₂ atmosphere)	[43]
Organic carbon content	Decreased constantly (pine wood, 300–700 °C, oxygen-limited conditions)	[44]
Transport	Decreased constantly (wheat straw and pine needle, 350 and 550 °C, anaerobic conditions)	[45]
Water-soluble intermediates	Increased to a maximum and then decreased (amorphous and crystalline cellulose, 100–350 °C, Ar atmosphere, get a maximum at 250 and 270 °C, respectively)	[46]
Nitrogen content	Increased to a maximum and then decreased (pine wood, 100–700 °C, oxygen-limited conditions, achieved a maximum at 400 °C)	[8]
Inorganic composition	Silicon: amorphous to crystalline (rice straw, 150–700 °C, oxygen-limited conditions)	[47]
	Phosphorus: phytate → orthophosphate/pyrophosphate (plant and manure, 350, 500, 650 and 800 °C, N ₂ atmosphere)	[48]
	Calcium: whewellite → calcite (<i>Eucalyptus saligna</i> wood, 400 °C and 550 °C)	[49]
	Potassium leaching rate increased (wheat straw, poplar wood, and spruce wood, 400, 460 and 525 °C, oxygen-limited conditions)	[50]
	Sulfur: inorganic sulfur decreased, organosulfur increased (oak and corn stover, 500, 600, and 850 °C, N ₂ atmosphere)	[51]

using straw as the feedstocks), considered as a good carrier gas in biochar preparation for its simple storage, safety, nontoxicity, and extensive sources. Rice straw is selected as a feedstock for preparing biochar due to its abundance in China, it is worth noting that the preparation atmospheres are N₂ or oxygen-limited conditions, and there have been scarce studies reporting about the characteristics of rice straw biochar prepared under CO₂ atmosphere. At present, many uncertainties exist in defining “biochar quality.” For example, biochar used for carbon sequestration purposes would be evaluated based on its decomposition rates and its effect on the native soil organic matter pool, whereas biochar used as an adsorbent would be evaluated on its pore structure and high specific surface area [16–20]. However, biochar value for the purpose of returning has always been the focus in scenic areas. The highly aromatic structure of biochar and its specific surface characteristics along with its high porosity have been considered as the main reason for altering the soil structure (density, pore size distribution, and aggregate stability), influencing the water dynamics, affecting soil pore size distribution, and water holding capacity [21]. Pore characteristics are a major contributor to the properties of biochar; it can be considered as the

high-quality biochar through soil improvement (promotion of soil porosity and fertility) after applying biochar to soil. Given this, this article chooses rice straw as the feedstock to prepare a series of biochars at a range of 300–800 °C under CO₂ atmosphere, at the same time, the other four rice straw biochars were prepared under the inert atmosphere (N₂) and the same four pyrolysis temperatures, separately, and focuses on the characteristics of the products through biochar yield calculation, FESEM–EDS, BET-specific surface, and other analysis methods to figure out the preparation condition of high-quality biochar, which is expected to provide a theoretical reference for the preparation condition of biochar and its reasonable use in agriculture.

Materials and methods

Materials

Rice straw is known as the agricultural biomass waste, obtained from the experimental field of Huazhong Agricultural University, Wuhan, China. At first, the rice straw containing more than 15 wt% of moisture was dried naturally

Table 2 Yields of biochars prepared under different conditions

Pyrolysis temperature (°C)	Yields in CO ₂ atmosphere (%)	Yields in N ₂ atmosphere (%)
300	54.67	56.26
400	40.35	40.42
600	33.17	33.49
800	28.38	29.13

and cut into rice straw chips (10–20 mm long); the sample size of 20 g per batch experiment was used for the biochar preparation, marked B0, and set aside. Then, a part of the rice straws was crushed and sieved to obtain a sample of particle size less than 1 mm for subsequent analysis.

Biochar preparation

The pyrolysis of rice straw chips (10–20 mm long) was carried out in a fluidized bed pyrolysis furnace. For each trial, about 20 g of sample was loaded into the sample carrier. The biochars were produced using a stepwise procedure, which was heated up gradually to the target

temperature of 300, 400, 600, and 800 °C, and the heating rate was 20 °C/min) with CO₂ (99.99%, 40 L/h) purging to provide a reductive atmosphere in the pyrolysis process, and remained at the target temperatures for about 20 min. The resultant biochar sample was allowed to cool to ambient environment, and then was weighed and recorded as the solid charcoal yield, as shown in Table 2. The biochars were labeled as BC300, BC400, BC600, and BC800 accordingly. BC represented biochars prepared under CO₂ atmosphere, while the suffix number represents the carbonization temperature. The preparation steps of biochars prepared under N₂ atmosphere mostly were the same as the biochars preparation; CO₂ would be replaced by N₂ (99.999%) at a flow of 40 L/h to provide a reductive atmosphere in the furnace, and then, the biochar products were marked BN300, BN400, BN600, and BN800, separately. BN represented biochars prepared under N₂ atmosphere, and 300, 400, 600, and 800 represented the pyrolysis temperature, as well. All the biochars were stored in a desiccator. They were then homogeneously subsampled for analyses. Straw and the biochars prepared under different conditions can be shown in Fig. 1. Element analysis of rice straw and biochar was done as shown in Table 3.

Fig. 1 Straw and the biochars prepared under different conditions**Table 3** Elemental analysis of rice straw and biochars

Samples	C (wt%)	H (wt%)	O (wt%)	N (wt%)	P (wt%)	K (wt%)
B0	36.8	5.51	53.65	1.3	0.27	2.47
BC300	44.35	3.99	45.56	1.65	0.41	4.04
BC400	46.64	2.49	43.03	1.48	0.55	5.81
BC600	47.03	1.05	43.67	1.13	0.62	6.50
BC800	44.82	0.52	46.15	0.94	0.66	6.91
BN300	45.73	4.08	44.64	1.68	0.36	3.50
BN400	48.05	2.71	41.92	1.42	0.52	5.38
BN600	47.80	1.24	42.83	1.04	0.62	6.47
BN800	45.19	0.57	46.24	0.85	0.62	6.53

Samples (rice straw, biochars) characterization

A number of methods were used to investigate the physicochemical characteristics (e.g., morphology, nutrient contents, specific surface area, and pore size distribution) of the biochars. All analyses were conducted in duplicate. Morphology and nutrient contents: Field emission scanning electron microscopy (FESEM) with energy dispersive spectrometer (EDS) was used to determine the morphology and nutrient contents of virgin biomass and char particles. The samples for cross-sectional images were prepared by placing them into epoxy resins and were polished with platinum to ensure electrical conductivity of the surface and were then observed under the same magnifications by FESEM. The EDS line spectra were acquired by drawing arrows at 1000× magnification for 300 s to achieve a sufficient signal-to-noise ratio. The results were autocorrected and optimized by Inca software.

BET-specific surface area and pore size distribution: specific surface area and pore size distribution of the biochar were determined using a BET- N_2 ; according to BET multiplayer adsorption theory, the isothermal adsorption of N_2 at 77 K was studied using an automatic adsorption equipment (Micromeritics, ASAP 2020, USA) to analyze the pore structure characteristics of the samples. Before the adsorption measurements, the sample was degassed at 150 °C under a vacuum for 6 h. The total pore volume was calculated at a relative pressure of 0.99. The BET surface area (S_{BET}), average pore size, and total volume (V) of the samples were determined by application of the Brunauer–Emmett–Teller (BET) equation, t plot method, and single point adsorption total pore volume analysis, respectively. The pore size

distribution was analyzed by the density-function theory (DFT) method, which can be used to calculate the distribution of micro-, meso-, and macropores, simultaneously.

Results and discussion

Effect of pyrolysis temperature and atmosphere on the physical characteristics of biochar

Effect of pyrolysis temperature and atmosphere on biochar morphology characteristics

Temperature FESEM was employed to observe the surface physical morphology of the samples. Surface morphology evaluation of the biochars produced at different pyrolysis temperatures under N_2 atmosphere can be compared from the FESEM micrographs (Fig. 2). It can be seen that the structure of the parent material (rice straw) was disordered consisted of solid cells strongly bonded, and slits derived from the interior of the straw and no micropores were observed, and just several mesopores and macropores existed, which generated for the mass transmission during the growth process as a plant, giving an irregular and rough appearance instead of the pore structure. Morphology of the biochars varied with the different temperatures; with increase in pyrolysis temperature, the fissures and slits in biochar surface were found to be more prominent. The external surface of BN300 displayed several irregular pore structures of different sizes, and the pore shape was not complete; parts of the straw structures still maintained in BN300, indicating incomplete pyrolysis at 300 °C. When the pyrolysis

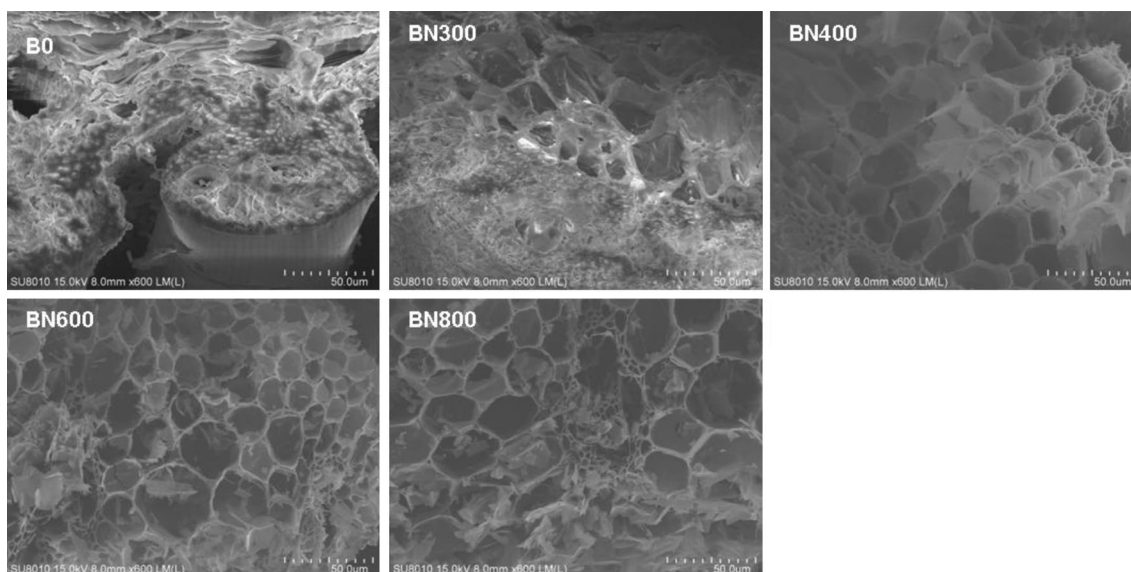


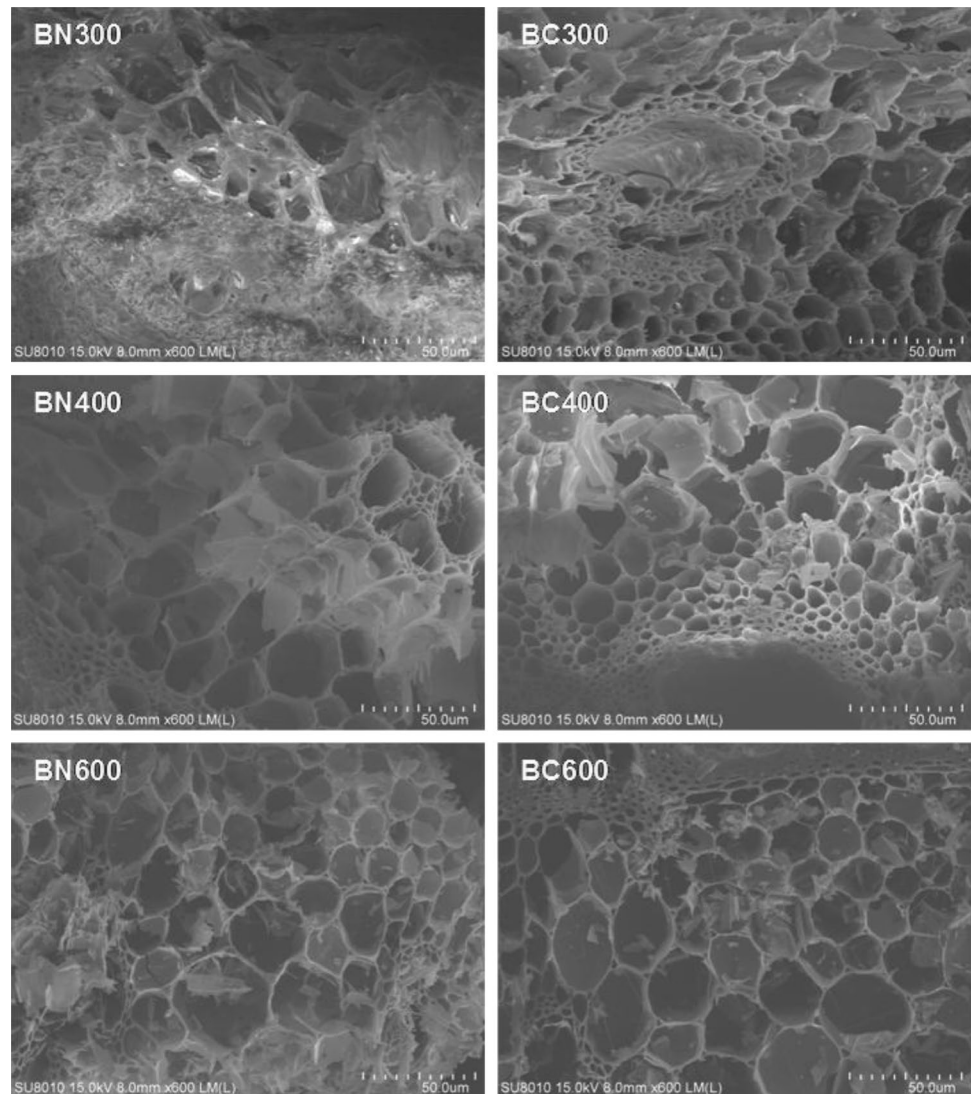
Fig. 2 Cross-sectional FESEM images of the rice straw and biochars prepared under different pyrolysis temperatures (N_2)

temperature rose to 400 °C, the pore structure exhibited in order, increasing pores, compact pore structure, developed porosity, and shaped integrity in BN400. With the temperature rising from 400 to 600 °C, the pores in the marrow of biochar enlarged and started to crack outward, leading the appearance of larger pores, a certain amount of micropores generated and developed close to the macropores. At temperatures higher than 600 °C, micropores developed further, the collapse of pore structure was more obvious, marrows fell in the way of clusters, but most of the structure was still reserved. The results showed the pyrolysis process facilitated the pore structure; however, some macropores with large pore size would collapse at high temperatures in accompany with biochar micropore development.

Atmosphere Figure 3 shows the effects of different pyrolysis atmospheres on the morphology of the rice straw biochars. The main char structure of the biochar was more and

more obvious and rich with increasing temperature, similar to the morphology change of municipal sewage sludge biochar produced at 400–600 °C [22]. Pituello et al. [23] observed the similar changes and thought that the interactions among components of the biomass might affect pyrolysis behavior of the various constituents and increase the heterogeneity of biochars. However, biochars prepared under the two atmospheres differed in the morphology characteristics. At temperatures higher than 300 °C, micropores of the biochars (N₂) formed mostly in the adjacent marrow of macropores, while the micropores would generated densely in somewhere of the biochars (CO₂) other than appeared in the adjacent marrow of macropores. At the temperature of 300 °C, it can be seen parts of primary rice straw tissues in BN300 instead of BC300, indicating that CO₂ improved the pyrolysis process and presented more pores in BC300, that was to say, pore structure in BC300 was more developed than BN300. In both atmospheres at 400 °C, the process

Fig. 3 Cross-sectional FESEM images of the biochars prepared under different pyrolysis atmospheres



of volatile release produced a highly porous char structure in BC400 and BN400, but it was much thicker of the pore walls in BC400 compared with BN400, which deserves higher mechanical strength relative to BN400, and the char structure of BC400 would not be destroyed instantly when incorporated into soil, pore sizes varied from micropores to macropores in BC400, while seen simple pore sizes distribution in BN400 (macropores and micropores close to the macropores), showing that CO₂ atmosphere affected the formation of pores. At temperatures higher than 400 °C, char structure in BC600 started to collapse; however, BN600 depicted slight collapse, leading a poorer mechanical strength as well. All the morphology structures presented a certain difference in the biochars under the two atmospheres. Morphology of BC showed more developed and firmer char structures, higher mechanical strength, and more extensive forms of the pore sizes. Lee et al. [24] reported similar results that micro- and mesopores were highly developed in biochar prepared under CO₂ instead of N₂ at the temperature of 650 °C. We concluded that the quality of biochars obtained under CO₂ atmosphere was superior to that of N₂ atmosphere.

Effects of pyrolysis temperature and atmosphere on the specific surface area of samples (rice straw, biochars)

Temperature Biochar pores were divided into micropores (<2 nm), mesopores (2–50 nm), and macropores (>50 nm), according to pore size [25]. As shown in Table 4, pore structure of rice straw and biochar products was mainly mesoporous; for the biochars, the average pore size decreased, but the BET-specific surface area increased with an increase of pyrolysis temperature. Due to the devolatilization characteristics of rice straw in heating process, specific surface area and pore volume of the rice straw sample were as small as 1.167 m²/g and 0.00479 cm³/g, and they increased gradually from 1.809 m²/g and 0.00670 cm³/g to 24.435 m²/g and 0.0211 cm³/g, respectively, with the rise of the pyrolysis temperature from 300 to 800 °C, which was similar to the previous observations for the biochars pro-

Table 4 Comparisons of the surface structure features between rice straw and biochars at different temperatures

Samples	S _{BET} (m ² /g)	Total pore volume (cm ³ /g)	Average pore size (nm)	Pore volume of micropore (cm ³ /g)
B0	1.167	0.00479	16.407	0.000432
BN300	1.809	0.00670	14.807	0.000711
BN400	4.649	0.0142	12.242	0.00184
BN600	10.273	0.0148	5.613	0.00426
BN800	24.435	0.0211	3.455	0.00959

duced from cotton stalk [26], reed [27], and rice straw [28–30]. Compared with raw rice straw, the release of the mass as volatile matter only enlarged the surface area of chars at 300 °C, to a small extent, and the surface area only increased to 1.809 m²/g, which may be due to the degradation of the hemicellulose, and thus caused the formation of a few pores for the release of volatile matter and discharge of gaseous product, leading the slight increase of the surface area. Continuous increase in the temperature up to 400 °C and the surface area of BN400 increased to 4.649 m²/g, showing an improvement by a large margin for some micropores appearance. Several researchers suggested that the degradation of cellulose was occurring during 250–450 °C [31–33], and the appearance of micropores showed that the cellulose may be decomposed. However, with the temperature increasing further to 800 °C, a continuous increase of the surface area, causing the total surface area to increase from 4.649 to 24.435 m²/g. This increase may be attributed to the increasing micropores for the decomposition of hemicellulose, cellulose, and lignin at high temperatures. From the comparison of these dates, it can be concluded that higher temperatures were favourable to the preparation of biochars with high specific surface area. And FESEM image presented evidence that the surface area of biochar increased with pyrolysis temperature, as well; however, similar observations by Li et al. [34] reported that pyrolysis at the higher temperature causes a larger specific surface area in biochars.

Atmosphere Table 5 gave the comparisons of the surface structure features of the biochars produced from different pyrolysis atmospheres. It was observed that the biochars prepared under two different atmospheres had different surface structure features, though they had similar variation trend with increasing temperature. The specific surface areas of BC were higher than BN at the same temperature; at temperatures lower than 600 °C, the effects of the two atmospheres on biochar-specific surface area were not as obvious as that at temperatures higher than 600 °C. Under

Table 5 Comparisons of the surface structure features between biochars at different atmospheres

Samples	S _{BET} (m ² /g)	Total pore volume (cm ³ /g)	Average pore size (nm)	Pore volume of micropore (cm ³ /g)
BC300	2.105	0.00957	18.178	0.000832
BC400	4.985	0.0146	11.684	0.00196
BC600	13.511	0.0158	4.673	0.00535
BC800	303.731	0.161	2.114	0.122
BN300	1.809	0.00670	14.807	0.000711
BN400	4.649	0.0142	12.242	0.00184
BN600	10.273	0.0148	5.613	0.00426
BN800	24.435	0.0211	3.4551	0.00959

CO₂ atmosphere, when the pyrolysis temperature rose from 600 to 800 °C, the specific surface area had a significant increase from 13.511 to 303.731 m²/g, while under N₂ atmosphere, the specific surface area just increased from 10.273 to 24.435 m²/g, showing an increase in small scale compared with the former, which may be due to the different properties of the two atmospheres. N₂, an inert gas, had limited effects on the char structure. Unlike N₂, CO₂ can serve as an activator, CO₂ activation will become more vigorous once at a higher temperature, the presence of CO₂ was benefit for improvement of the micropores, and specific surface area of BC800 surpassed BN800 markedly.

Effects of pyrolysis time, temperature, and atmosphere on pore size distribution of biochars

Time Residence time affects biochar properties. For example, the average pore size of biochar is shown in Fig. 4. The average pore size of biochar increases with increasing residence time from 10 to 30 min at 600 °C and in the N₂ atmosphere. When the residence time is 20 min, the average pore size of biochar reaches 0.1232 μm, corresponding to an 8.45% higher pore size compared to 10 min. Increasing the residence time from 20 to 30 min is associated with a decrease in the average pore size. The pore size decreases 3.57% between 20 and 30 min. The pore structure of biochar is closely connected to its average pore size, and the pore structure is a primary factor affecting its adsorption ability. Therefore, the pore structure of biochar can be adjusted by controlling the residence time, because there is a critical time for biomass material carbonization.

Temperature Pore size distributions of the rice straw sample and biochars represented by BN were showed in Fig. 5. There were almost no micropores in the rice straw, showing rare mesopores and macropores, indicating that the structure of rice straw was extremely underdeveloped. In all the biochars, the number of micropores, mesopores, and macropores was enhanced after heating the rice straw

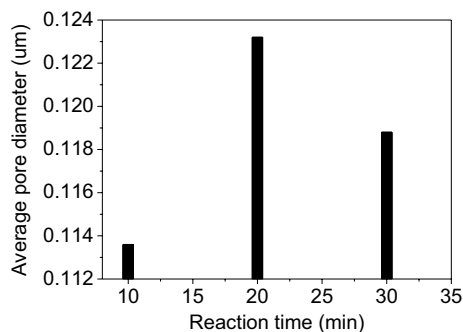


Fig. 4 Effect of residence time on the average pore size of biochar

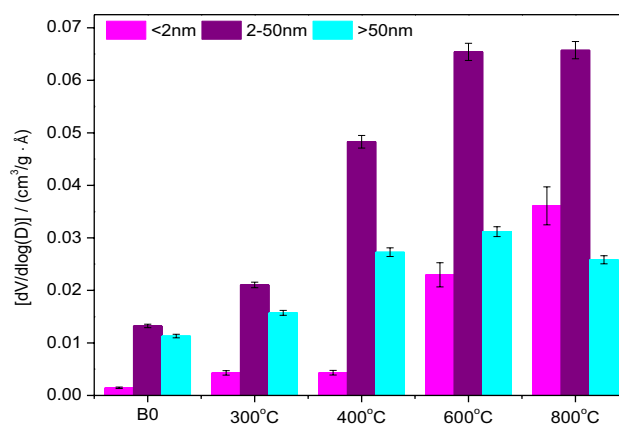


Fig. 5 Changes of the pore size distribution in rice straw sample and biochars prepared at different temperatures (N₂)

at different temperatures. Contrast to the rice straw, the pore structures of all the biochars were relative developed with different degree. Micropores was generally low for biochars produced at all temperatures, especially in low temperatures (300, 400 °C), which were slightly higher than that of rice straw, number of micropores only changed a little, when the pyrolysis temperature rose from 300 to 400 °C, micropores in BN300 were nearly equal to that in BN400. It showed a continuous increase of micropores with the pyrolysis temperature increasing from 400 to 800 °C, and a significant improvement of micropores occurred between 300 and 400 °C, which were consistent with FESEM- and BET-specific surface analysis. Mesopores were the dominant pores in all the biochars, which increased greatly with an increase in temperature from 300 to 600 °C; then, with further increase in temperature, the increasing extent of mesopores was lowered, reaching its maximum value at 800 °C. In comparison with micropores and mesopores, macropores has the variation that increased first and then decreased with increasing temperature and reached its peak at 600 °C. The number of macropores had the order of BN600 > BN400 > BN800 > BN300 > rice straw, which achieved the largest growth at temperature ranged from 300 to 400 °C, indicating that it was not conducive to the formation of macropores at the temperature which was too high or too low. Therefore, biochar produced at high temperature was beneficial to the generation of micropores, while mesopores and macropores had extensive pore size distribution at moderate temperature. Hence, the pore structure of the biochar was relatively developed at 400 °C.

Atmosphere To describe the effects of atmospheres on the pore characteristics during pyrolysis, a pore size distribution profile was obtained based on the DFT, as shown in Fig. 6. All the biochars had the pore size distributions which varied from 1.7 to 300 nm, CO₂ pyrolysis generated biochars

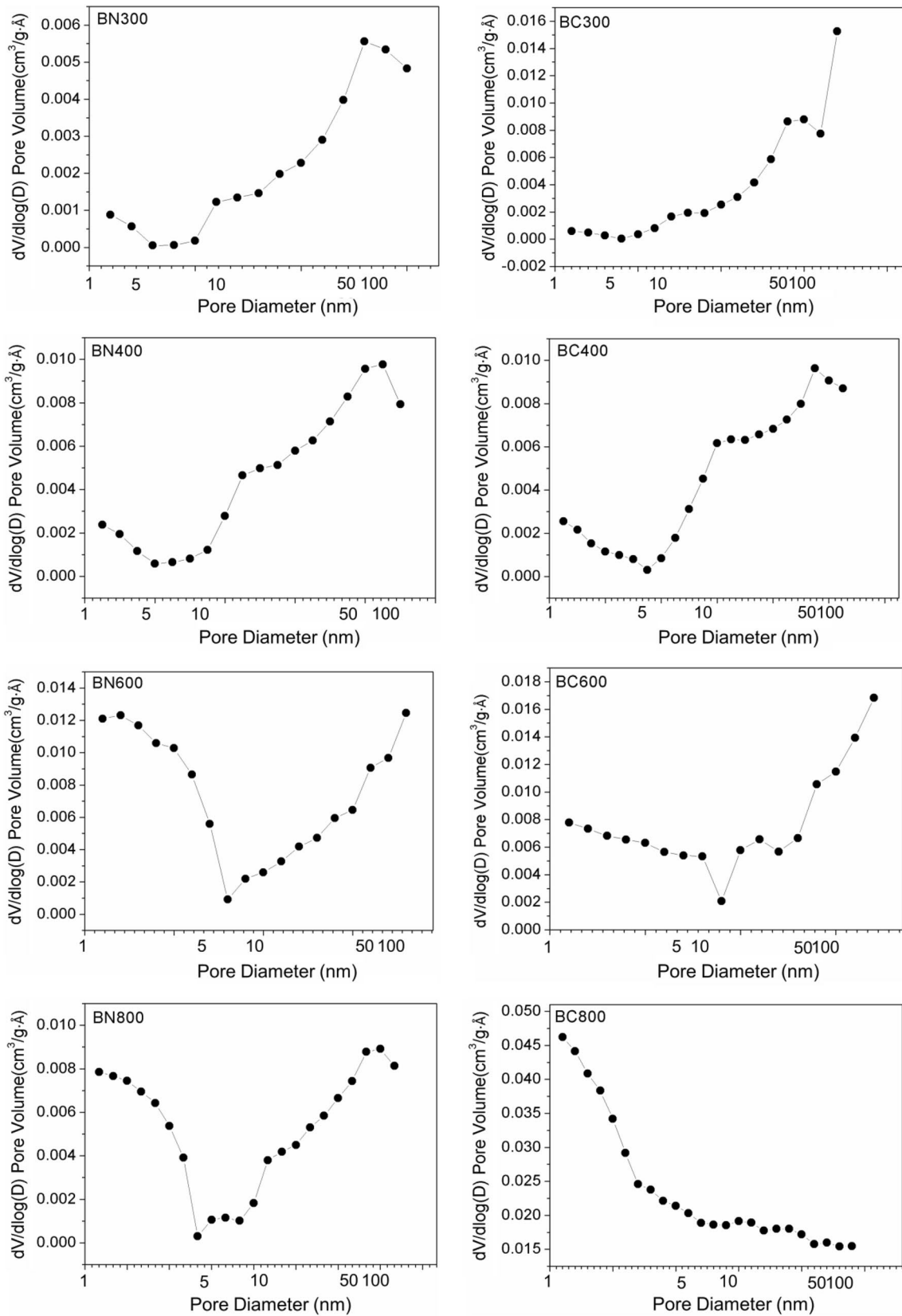


Fig. 6 Effects of the pyrolysis atmosphere on the pore size distribution of the biochars

with higher micropores, mesopores, and macropores as observed from Fig. 6. It can be concluded that the porosity in the range of micropores, mesopores, and macropores was better developed under CO₂ atmosphere. Pore size distribution began to present a certain differences even at low temperatures in BC and BN, especially for the mesopores. Pore size distribution of BC was relatively sparse at pore diameters above 15 nm and the majority of macropores found between 75 and 300 nm; among the four biochars (BC300, BC400, BC600, and BC800), BC600 had the largest member of macropores. While pore size distribution of BN was relatively sparse at pore diameters 5–15 nm, macropores focused mostly between the pores of sizes 75 and 275 nm, which showed more extensive mesopores and macropores distribution in CO₂ pyrolysis generated biochars. On increasing the temperature, there was an increasing disparity (in the quantities) between the macropores in BC and BN, which was the same as mesopores–mesopores and micropores–micropores. It can be seen that the numbers of pores (micropores, mesopores, and macropores) in BC800 were significantly higher than the numbers of corresponding pores in BN800. The pore size distribution had a significant increase of micropores in BC800, and in the range of mesopores (2–15 nm), more intensive pores size distribution of BC800 contrast with BC800, which was attributed to the fact that CO₂ could serve as an important activator and react with the char structure of biochars at high temperatures, showing a promotion of the pore structure, especially the

micropores. BC800 has the largest numbers of mesopores and macropores, while the numbers of micropores were up to the maximum in BC600.

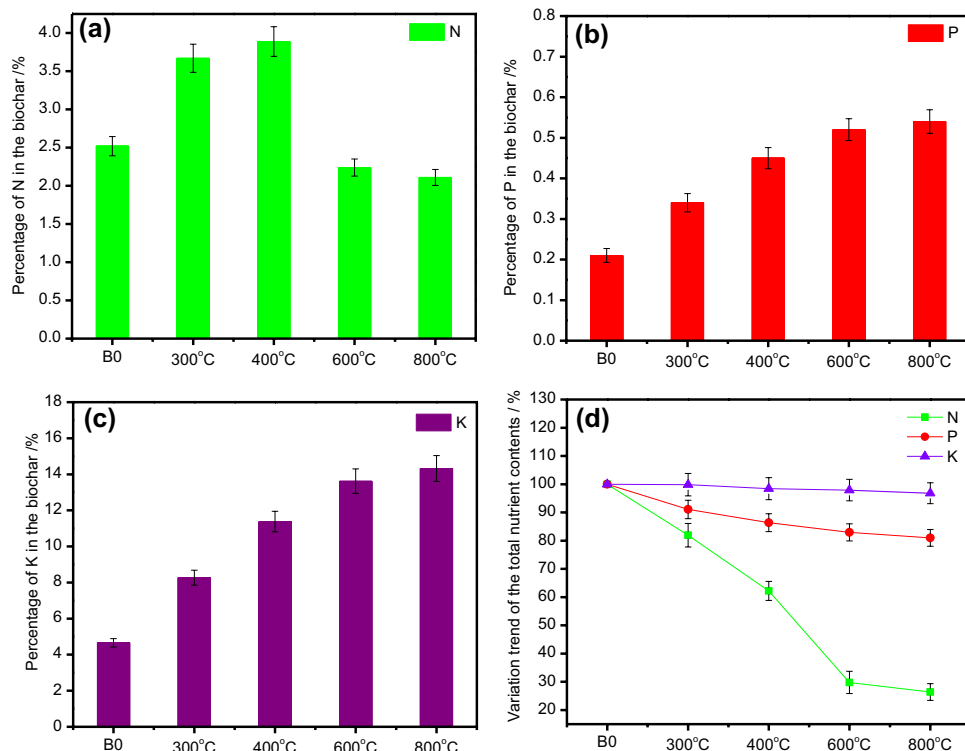
Effects of pyrolysis temperature and atmosphere on the nutrients in biochars

Pyrolysis temperature effects on variation of nutrients (N, P, and K) in biochars (N₂)

EDS technique was applied to detect the variation of nutrients (N, P, and K) in rice straw and biochars produced at various temperatures and in N₂ atmosphere (shown in Fig. 7a–c) and the RR (retention rate shown in Fig. 7d), which was between 0 and 100%. The lower RR indicated fewer elements remained in biochars. It was known that the rice straw and biochars were rich in K, and followed by N nutrient, while P content was relatively low. All the three kinds of nutrient elements showed regular changes with the rise of pyrolysis temperature.

Total N content in biochar significantly increased initially in low-temperature biochars (300 and 400 °C), and then significantly decreased for high-temperature biochars (600 and 800 °C), getting a maximum and minimum at 400 and 800 °C, respectively. This may be due to the loss rate of N which was less than the mass loss rate of rice straw at the temperature below 400 °C, leading to the enrichment of nitrogen in low-temperature biochars (300 and 400 °C).

Fig. 7 a–c Changes in total N, P, and K concentration in rice straw sample and biochars at various temperatures, respectively, **d** RR for the nutrient contents of rice straw sample and biochars at various temperatures



However, with the temperature increasing further, the volatilisation of N became more vigorous; a decrease of total N in the biochar was observed in high-temperature biochars (600 and 800 °C). Total N content of 400 °C biochar increased from the feedstock concentrations by 54.37%; for the 800 °C biochar compared to feedstock, total N content decreases by 16.27%. As for the retention rate of N (NRR), with the pyrolysis temperature increasing from 300 to 800 °C, NRR decreased constantly from 81.92 to 26.38%; a sharp decrease was observed during 400–600 °C. The previous research has shown that N was removed through loss of the $\text{NH}_4\text{-N}$ and $\text{NO}_3\text{-N}$ fraction as well as the loss of volatile matter containing N groups [35]. Our study indicated that relatively high proportion of nitrogen would be conserved at low pyrolysis temperatures, and high-temperature pyrolysis would present a massive loss of N.

Total content of P in the rice straw sample was 0.21%. Total content of P in biochars increased from 0.34 to 0.54% with the pyrolysis temperature increasing from 300 to 800 °C, indicating that P was associated with the inorganic fraction of the rice straw, showing a similar trend in sludge pyrolysis biochar reported by Mustafa [36]. The maximum P was 0.54% in BN800 at 800 °C, which increased 157% compared to the rice straw. The RR of P indicated 81.00–91.07% of P was retained in biochars, which might be attributed to the fact that P contained in rice straw was associated with thermo-stable phosphate minerals that was difficult to decompose and became more crystallized with increasing temperature [37].

Total content of K in the rice straw sample was 4.66%. With the pyrolysis temperature increasing from 300 to 700 °C, the content of K in the biochars increased from 8.27 to 14.32%. This result indicated that the enrichment of K in biochars was intensified remarkably with the rise of pyrolysis temperature, which has the similar trend as compared to P. And the RR of K indicated that 96.8–99.83% of K was retained in biochars. The concentration of K in biochars increased with increasing pyrolysis temperatures and this increment may be due to its high vaporization temperature.

Pyrolysis atmosphere (CO_2 and N_2) effects on variation of nutrients (N, P, and K) in biochars

The EDS spectra of the biochars prepared under CO_2 and N_2 atmospheres revealed their elements variation (Fig. 8). It showed that a relatively C content in biochars prepared under CO_2 atmosphere decreased by 6.58–17.15%; C content in biochars prepared under the two atmospheres had a same variation trend with increasing temperature (increased first and then decreased and reached its peak at 400 °C), while O content of BC surpassed BN, which was attributed to the pyrolysis and gasification reaction of biochars prepared at CO_2 atmosphere, the element of O can

be introduced into the biochar through a series of oxidation reactions during the gasification process. The higher nutrient contents (N, P, and K) and according RRs were observed in BC suggested better nutrient characteristics. N content in BC increased by 3.57–8.17% compared with N_2 atmosphere. Ren indicated that the formation of H free radicals of biochars can make CO_2 inhibit the production of gases (HCN, H₂CO, etc.) through the pyrolysis process [38]. There were almost no volatilization of P and K in the rice straw pyrolysis process, which differed from N. P and K would be intensified with the rise of pyrolysis temperature; however, biochar yields (Table 2) were lower BC than BN, hence causing the high content of P and K in BC.

Optimum preparation approach of the biochar for its returning use

Nutrients were easy to lose during the pyrolysis of rice straw; biochars also varied in morphology features and pore structure with the preparation conditions. Through the analysis we have characterized above, the theoretical mechanism of the high-quality biochar for the purpose of returning was obtained, as shown in Fig. 9. Rice straw heated in a relatively low temperature (300 and 400 °C) would get larger retention rate of the nutrients (N, P, and K) in biochars. It was porous structure of the biochar morphology that pyrolyzing at temperatures above 300 °C (400, 600, and 800 °C). Extensive pore size distributions were observed at temperatures higher than 300 °C (400, 600, and 800 °C), all the pores of the rice straw biochars were mesopores, macropores were only second to only mesopores other than the biochar prepared at 800 °C, and with the temperature increasing further to 800 °C, micropores would increase significantly. The effects of these pores on soil quality are different. Specifically, macropores might improve soil aeration and water infiltration; mesopores could adsorb macromolecules which cannot enter the microporous structure, and provided the carrier for the preservation of soil fertility, micropores might adsorb and retain nutrients to improve soil fertility. However, studies have shown that the larger macropores and mesopores can play a relatively obvious role in soil amendment compared to the micropores [25, 39]. Then, combining with economic energy consumption, it was suggested that 400 °C was the optimum temperature for the preparation of biochars. In comparison with the two atmospheres (N_2 , CO_2), we found that characteristics (morphology, pore size distribution, and nutrients content) of BC were superior to BN at the same pyrolysis temperature, and hence, CO_2 atmosphere was more appropriate than N_2 atmosphere to produce biochars. Therefore, the optimum preparation approach of the biochar for its returning use should be in 400 °C and CO_2 atmosphere.

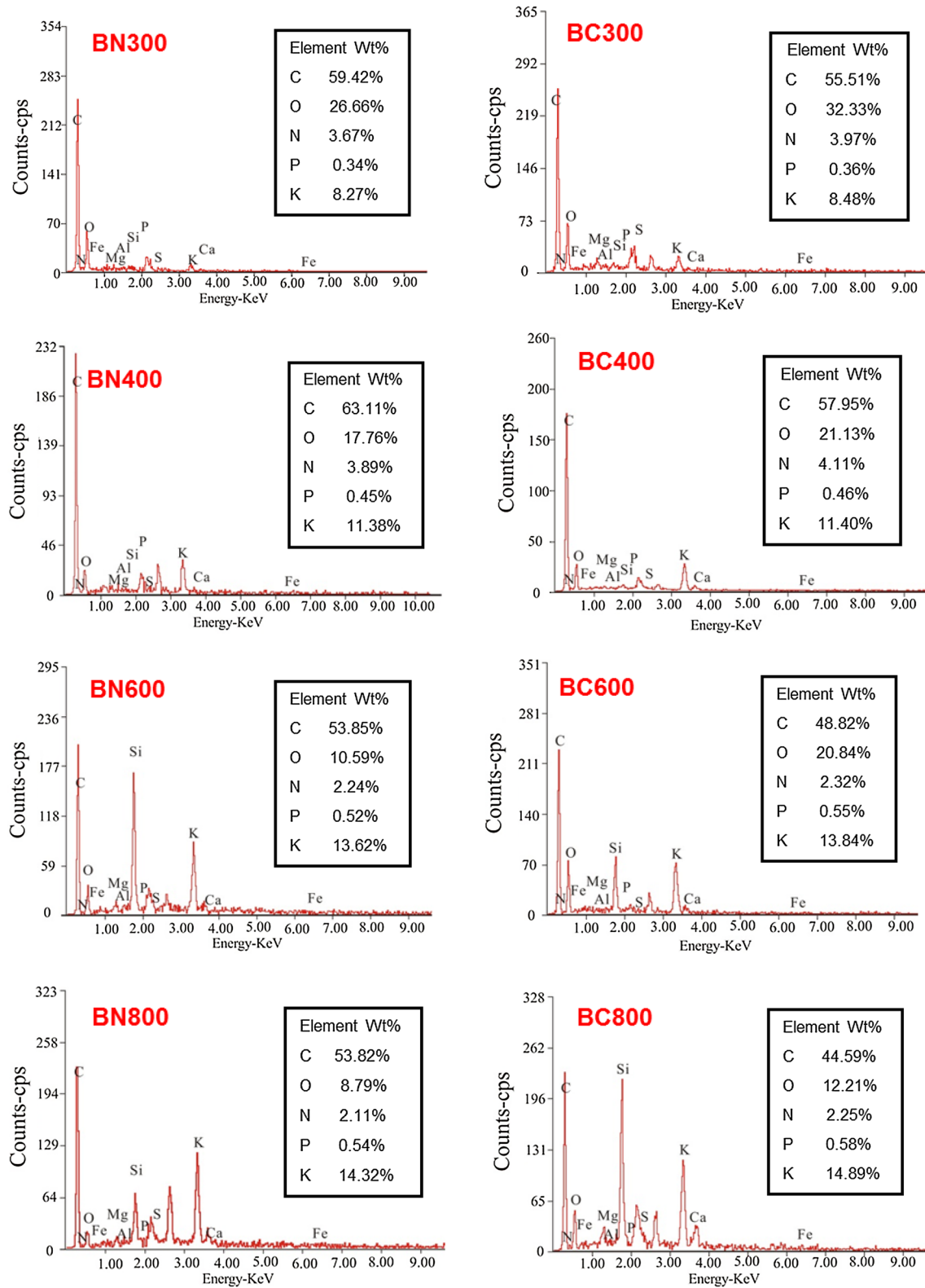


Fig. 8 EDS spectra quantification of the biochars (BN300, BN400, BN600, BN800, BC300, BC400, BC600, and BC800), respectively

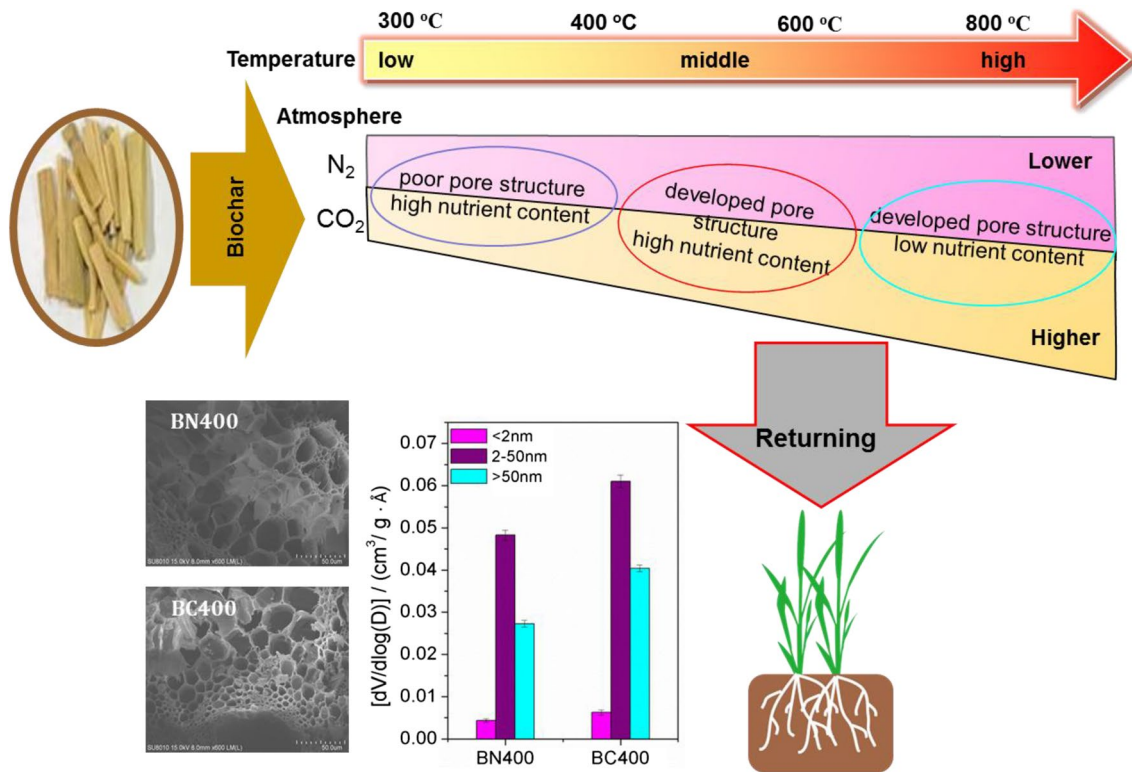


Fig. 9 Effect of residence time on the average pore size of biochar at 600 °C and in the N₂ atmosphere [52]

Conclusions

The results of FESEM showed that both the pyrolysis temperature and atmosphere have effects on biochar morphology, the raw rice straw could not be pyrolyzed completely at 300 °C, char structure gradually developed, more developed pore structure in biochars prepared at intermediate temperature (400 °C), and a collapse was seen in high temperatures (600 and 800 °C). CO₂ prepared biochars had more developed pore structure, thicker wall, and higher mechanical strength which was not easy to destroy after their incorporation into soil. Total pore volume and specific surface area increased with increasing temperature, CO₂ prepared biochars exhibited a high total pore volume and specific surface area than N₂ prepared biochars. The rice straw-derived biochars were rich in K, and followed by N nutrient, while P content was relatively low. With the rise of pyrolysis temperature, the content of N in the biochars increased first and then decreased and reached its peak at 400 °C, while the content of P and K always increased; all the three nutrients’ content showed relatively lower content in N₂ prepared biochars than that of CO₂. The result from BET analysis indicated that extensive pore size distribution in biochars pyrolyzes at temperatures above 300 °C. There were significant increases of macropores and mesopores at intermediate temperatures (400 and 600 °C), and porosity in the range of micropores,

mesopores, and macropores was better developed under CO₂ atmosphere. 400 °C and CO₂ atmosphere were considered as the optimum operating conditions of high-quality rice straw-derived biochars for returning.

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