

# D-Xylose Isomerase from a Marine Bacterium, *Vibrio* sp. Strain XY-214, and D-Xylulose Production from $\beta$ -1,3-Xylan

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**Abstract** The *xylA* gene from a marine bacterium, *Vibrio* sp. strain XY-214, encoding D-xylose isomerase (XylA) was cloned and expressed in *Escherichia coli*. The *xylA* gene consisted of 1,320-bp nucleotides encoding a protein of 439 amino acids with a predicted molecular weight of 49,264. XylA was classified into group II xylose isomerases. The native XylA was estimated to be a homotetramer with a molecular mass of 190 kDa. The purified recombinant XylA exhibited maximal activity at 60°C and pH 7.5. Its apparent  $K_m$  values for D-xylose and D-glucose were 7.93 and 187 mM, respectively. Furthermore, we carried out D-xylulose production from  $\beta$ -1,3-xylan, a major cell wall polysaccharide component of the killer alga *Caulerpa taxifolia*. The synergistic action of  $\beta$ -1,3-xylanase (TxyA) and  $\beta$ -1,3-xylosidase (XloA) from *Vibrio* sp. strain XY-214 enabled efficient saccharification of  $\beta$ -1,3-xylan to D-xylose. D-Xylose was then converted to D-xylulose by using XylA from the strain XY-214. The conversion rate of D-xylose to D-xylulose by XylA was found to be approximately 40% in the presence of 4 mM sodium tetraborate after 2 h of incubation. These results demonstrated that TxyA, XloA, and XylA from *Vibrio* sp. strain XY-214 are useful tools for D-xylulose production from  $\beta$ -1,3-xylan. Because D-xylulose can be used as a source for ethanol fermentation by yeast *Saccharomyces cerevisiae*, the present study will provide a basis for ethanol production from  $\beta$ -1,3-xylan.

**Keywords** *Caulerpa taxifolia* · D-Xylose isomerase ·  $\beta$ -1,3-Xylanase ·  $\beta$ -1,3-Xylosidase · *Vibrio* sp. strain XY-214 ·  $\beta$ -1,3-Xylan

## Introduction

In recent years, growing attention has been devoted to the conversion of biomass into the fuel ethanol, and significant advances have been made toward the technology of saccharification of carbohydrate polymers and ethanol fermentation (Lin and Tanaka 2006). In most cases, terrestrial plants, such as agricultural and forestry residues, are used for ethanol production because these contain large amounts of potentially fermentable sugars in the form of starch, cellulose, and  $\beta$ -1,4-xylan. In contrast, studies involved in ethanol production from algal biomass are very few.

The mutant *Caulerpa taxifolia* called as killer green alga was the first macrophyte invasion that drew widespread public attention. Its first breakout is attributed to an aquarium in Monaco in 1984 (Jousson et al. 1998; Olsen et al. 1998; Meusnier et al. 2001), and the invasive alga has now spread to the coasts of six countries bordering the Mediterranean Sea, i.e., Croatia, France, Italy, Monaco, Spain, and Tunisia (Meinesz and Hesse 1991; Meinesz et al. 2001). Furthermore, *C. taxifolia* was discovered for the first time in Californian coast in 2000 (Dalton 2000; Kaiser 2000). Additionally, new introduced populations of *C. taxifolia* were recognized in Australia (New South Wales; Schaffelke et al. 2002). Although *C. taxifolia* destroys the ecosystems of the sea environment through competition with native species, attempts at management of the invasive algae in the Mediterranean Sea remain rare.

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$\beta$ -1,3-Xylan is a major cell wall polysaccharide component of *Caulerpa* spp. (Iriki et al. 1960). It is also found in the cell walls of other green algae, such as *Bryopsis* and *Udotea* spp., and red algae (*Porphyra* and *Bangia* spp.; Iriki et al. 1960; McDowell 1967). Because  $\beta$ -1,3-xylan is a homopolysaccharide composed of  $\beta$ -1,3-linked D-xylose residues, it can be used as a source for ethanol fermentation. Similar to other biomass materials such as starch, cellulose, and  $\beta$ -1,4-xylan, the saccharification process using glycoside hydrolases is required for ethanol production from  $\beta$ -1,3-xylan. However, there are a few studies about  $\beta$ -1,3-xylan-degrading enzymes (Chen et al. 1986; Aoki et al. 1988; Yamaura et al. 1990; Araki et al. 1998, 1999).

*Vibrio* sp. strain XY-214 is a marine bacterium that can grow on  $\beta$ -1,3-xylan as a sole carbon source (Araki et al. 1999). To date, we have cloned and characterized the genes encoding a  $\beta$ -1,3-xylanase (TxyA; 1,3- $\beta$ -D-xylan xylanohydrolase; EC 3.2.1.32; Araki et al. 2000) and a  $\beta$ -1,3-xylosidase (XloA; Umemoto et al. 2008) from the bacterium. The synergistic action of these two types of enzymes enables the complete degradation of  $\beta$ -1,3-xylan to D-xylose. Although yeast *Saccharomyces cerevisiae* is the most effective ethanol-producing organism, it cannot ferment D-xylose as a carbon source. In contrast, the yeast can ferment D-xylulose, a keto isomer of D-xylose (Gong et al. 1981). Therefore, D-xylose released from  $\beta$ -1,3-xylan must be converted to D-xylulose before ethanol fermentation.

D-Xylose isomerase (XIs; EC 5.3.1.5) is an intracellular enzyme which catalyzes the reversible isomerization of D-xylose to D-xylulose in the first step of the D-xylose metabolism following the pentose phosphate pathway. The enzyme can also isomerize D-glucose to D-fructose. However,  $K_m$  is the lowest and  $V_{max}$  is the greatest when D-xylose is the substrate (Meng et al. 1991; Rangarajan and Hartley 1992). The XIs can be classified into two groups based on their amino acid sequences (group I enzymes are shorter by approximately 50 residues at the N terminus; Park and Batt 2004).

Fortunately, our previous study revealed that a truncated gene (*xylA*), which showed high sequence similarities to several bacterial XIs, is located in the upstream region of the  $\beta$ -1,3-xylosidase gene (*xloA*) from *Vibrio* sp. strain XY-214 (Umemoto et al. 2008). In the present study, therefore, we first cloned the complete *xylA* gene encoding a D-xylose isomerase from the genomic DNA of the strain XY-214 and characterized a recombinant D-xylose isomerase (rXylA) expressed in a transformed *Escherichia coli*. Second, we tried to produce D-xylulose from  $\beta$ -1,3-xylan by using three kinds of recombinant enzymes (rTxyA, rXloA, and rXylA). Our purpose is to provide information regarding the utilization of a killer alga, *C. taxifolia*, by the establishment of a basis for ethanol production from  $\beta$ -1,3-xylan.

## Materials and Methods

### Materials

$\beta$ -1,3-Xylan was prepared from a green alga, *Caulerpa racemosa* var. *laetevirens*, by the method of Iriki et al. (1960). D-Xylulose was purchased from Sigma Chemical (St. Louis, MO, USA). D-Xylose, D-glucose, and the other chemicals were purchased from Wako Pure Chemical Industries (Osaka, Japan).

### Bacterial Strains, Plasmids, and Culture Conditions

*Vibrio* sp. strain XY-214, isolated from sea mud at Ise bay in 1990, was grown in peptone medium (0.5% peptone, 0.1% yeast extract, 3% NaCl, 0.05% MgSO<sub>4</sub>, 0.2% K<sub>2</sub>HPO<sub>4</sub>, 0.04% KH<sub>2</sub>PO<sub>4</sub>; pH 7.6) containing 0.3%  $\beta$ -1,3-xylan and used as the source of chromosomal DNA. *E. coli* XL1-Blue and pBluescript II KS(-) plasmids served as cloning hosts and vectors (Stratagene, La Jolla, CA, USA), respectively. *E. coli* BL21 (DE3; Novagen, Madison, WI, USA) was used as the host for a derivative of pET22b(+) (Novagen) in the production of a recombinant protein. All *E. coli* strains were grown in Luria–Bertani (LB) medium supplemented with ampicillin (50  $\mu$ g ml<sup>-1</sup>). For agar medium, LB medium was solidified with 1.5% agar.

### Cloning of the *xylA* Gene

Previously, we have reported the cloning of a novel gene encoding  $\beta$ -1,3-xylosidase from *Vibrio* sp. strain XY-214. It has been revealed that a 4,206-bp DNA fragment, which was obtained by Southern hybridization analysis with a partial *xloA* fragment as the probe, contains the complete *xloA* gene and three possible open reading frames (ORFs; Umemoto et al. 2008). Sequence analysis of the DNA fragment revealed that an incomplete ORF encoding a putative XylA was located in the upstream region of the *xloA* gene. On the other hand, two complete ORFs named *gatA* and *alrA* are located in the downstream of the *xloA* gene. The *gatA* and *alrA* gene products showed similarity to a galactoside acetyltransferase (GAT) and an aldehyde reductase (AlrA), respectively. On the basis of the sequence of a partial *xylA* gene, one specific primer set (pxylA-F, 5'-CTAGAGAACGAGATCAAAGT-3', and pxylA-R, 5'-CTCTAGCGATAGTGCCATAG-3') was designed. The PCR product (330 bp) amplified from the plasmid *xloA*/pBluescript (*Xba*I-*Spe*I; Umemoto et al. 2008) with the primer set was used as the probe for cloning of the *xylA* gene after it was labeled with alkaline phosphatase (GE Healthcare, Buckinghamshire, England) according to the manufacturer's instructions.

To obtain the complete *xylA* gene, Southern hybridization with the probe was carried out. Digestion of *Vibrio* sp.

strain XY-214 genomic DNA with both *SacI* and *EcoRI* gave a 4.2-kbp fragment that was hybridized with the AlkPhos-labeled probe. The DNA fragments corresponding to 4.2-kbp were excised from the gel and purified with a Wizard gel and PCR clean-up system (Promega Co., Madison, WI, USA). These were ligated into the *SacI* and *EcoRI* sites of pBluescript II KS(-) (Stratagene), and the recombinant plasmids were introduced into *E. coli* XL1-Blue (Stratagene). One of the 123 colonies was selected as positive by colony hybridization with the AlkPhos-labeled probe. The cloned plasmid was named *xylA/pBluescript (SacI-EcoRI)*.

#### DNA Sequencing and Bioinformatics Approaches

Nucleotide sequence analysis of the DNA fragment inserted into the vector was carried out on a Beckman CEQ2000XL sequencer (Beckman Coulter, Fullerton, CA, USA) using a GenomeLab™ DTCS-Quick Start Kit (Beckman Coulter). Oligonucleotide primers designed on the basis of known sequences were also used in DNA sequencing. The nucleotide sequence data were analyzed using GENETYX-WIN computer software (Genetyx Corporation, Tokyo, Japan). Similarity searches were performed using the basic local alignment search tool algorithm at the National Center for Biotechnology Information server (Altschul et al. 1990). The molecular mass of a gene product was estimated by using the peptide mass tool at the ExPASy server of the Swiss Institute of Bioinformatics (Gasteiger et al. 2003).

#### Expression and Purification of rXylA

For the production of the rXylA in *E. coli*, the *xylA* gene was subcloned into a pET22b(+) vector (Novagen) as follows: The full-length *xylA* gene was amplified from *xylA/pBluescript (SacI-EcoRI)* as the template by PCR using the primers 5'-AGGGCCCATATGACTGAATTTTCAAAA-3' and 5'-AGCTCGAGCTTGTAGATGAAGCCATTTA-3', which contained artificial *NdeI* and *XhoI* sites (underlined), respectively. The PCR product was digested with *NdeI* and *XhoI* and ligated into a pET22b(+) vector linearized with the same enzymes to construct the *xylA/pET22b*. This plasmid provides the recombinant XylA with a six-His tag fused at its C terminus. The *xylA/pET22b* was transformed into *E. coli* BL21 (DE3)-competent cells (Novagen) and used for the production of rXylA. The absence of undesired mutations in the amplified DNA fragment was verified by DNA sequencing.

*E. coli* BL21 (DE3) transformants carrying *xylA/pET22b* were cultivated at 37°C in 800 ml of LB medium in the presence of ampicillin (50 µg ml<sup>-1</sup>). When the absorbance of the culture reached around 0.6 at 600 nm, the

temperature was changed to 25°C and isopropyl-β-D-galactopyranoside was added to the culture to give a final concentration of 1 mM for the induction of gene expression. After an additional incubation of 20 h at 25°C, the cells harvested by centrifugation were suspended in binding buffer (20 mM sodium phosphate, 0.5 M NaCl, and 10 mM imidazole; pH 7.4) and disrupted on ice by sonication. The supernatant of the cell lysate was collected by centrifugation and fractionated by a HiTrap chelating HP column (GE Healthcare) precharged with Ni<sup>2+</sup>. The recombinant protein was eluted with a linear gradient of imidazole (10–500 mM). The purified rXylA was dialyzed against 50 mM sodium phosphate buffer (pH 7.5) and stored at 4°C until use. Subunit molecular masses and enzyme purity were determined by sodium dodecyl sulfate–polyacrylamide gel electrophoresis (SDS-PAGE) using 12% polyacrylamide gels (Laemmli 1970). The protein concentration was measured by the method of Lowry et al. (1951) with bovine serum albumin (Sigma Chemicals Co.) as a standard.

#### Enzyme Assays

D-Xylose isomerase activity was determined by measuring the accumulation of D-xylulose. Unless otherwise indicated, the standard reaction mixture contained 5 mM D-xylose, 5 mM MgSO<sub>4</sub>, the enzyme solution at a suitable dilution, and 50 mM sodium phosphate buffer (pH 7.5) to bring the final volume to 400 µl. The reaction mixture was incubated at 40°C for 10 min followed by cooling the tubes on ice to stop the reaction. The D-xylulose formed was quantified by the cysteine–carbazole–sulfuric acid method (Dische and Borenfreund 1951), and the absorbance was measured at 540 nm. One unit of isomerase activity was defined as the amount of enzyme that produced 1 µmol of keto-sugar per minute under the assay conditions.

#### Characterization of rXylA

The effects of temperature on the activity of rXylA were determined for a range between 0°C to 80°C. With the exception of temperature, the assay conditions were the same as those described for the standard method. For the temperature stability measurements, the enzymes were preincubated at various temperatures at pH 7.5 for 20 min, and the residual activity was measured using the standard method. The effects of pH on the activity and stability of rXylA were examined at various pHs (3.0–12.0) in Britton and Robinson's universal buffer (40 mM phosphoric acid, 40 mM boric acid, and 40 mM acetic acid). The pH activities were assayed under the conditions described for the standard method. For the pH stability measurements, the enzymes were preincubated at various

pHs at 4°C for 12 h, and the residual activity was then measured using the standard method.

To determine the kinetic parameters, enzyme assays were performed in 50 mM sodium phosphate buffer (pH 7.5) containing 5 mM MgSO<sub>4</sub> and 1–64 mM D-xylose (25–300 mM for D-glucose). The reaction mixtures were incubated for 10 min at 40°C, and then reactions were stopped by cooling the tubes in ice. The D-xylulose and D-fructose formed were quantified by cysteine–carbazole–sulfuric acid method, and the absorbance at 540 and 560 nm were measured, respectively. The apparent  $K_m$  (millimolar per liter) and  $V_{max}$  (units per milligram protein) were calculated from Lineweaver–Burk plots.

For the determination of native molecular mass, samples were applied to a Superdex 200 HR 10/30 column (GE Healthcare) equilibrated with 50 mM sodium phosphate buffer (pH 7.5) containing 0.15 M NaCl. Molecular mass standards (GE Healthcare) included ferritin (440 kDa), aldolase (158 kDa), ovalbumin (43 kDa), and chymotrypsinogen A (25 kDa).

#### Effect of Metal Ions on rXylA Activity

Metal ions were removed from the purified rXylA by treatment with 10 mM EDTA at 4°C for 1 h, followed by overnight dialysis against 50 mM sodium phosphate buffer (pH 7.5) at 4°C with several changes of buffer solution. The effects of various metal ions were determined by adding 5 mM MgSO<sub>4</sub>, CoCl<sub>2</sub>, MnCl<sub>2</sub>, CaCl<sub>2</sub>, ZnSO<sub>4</sub>, FeSO<sub>4</sub>, or CuSO<sub>4</sub> to the dialyzed enzyme and assaying D-xylose isomerase activity under standard conditions without 5 mM MgSO<sub>4</sub>.

#### Saccharification of $\beta$ -1,3-Xylan to D-Xylose

$\beta$ -1,3-Xylanase (rTxyA) and  $\beta$ -1,3-xylosidase (rXloA) were used for the saccharification of  $\beta$ -1,3-xylan. Both enzymes were expressed in *E. coli* transformants and purified as described previously (Araki et al. 2000; Umemoto et al. 2008). Then, 0.5%  $\beta$ -1,3-xylan in 50 mM sodium phosphate buffer (pH 7.5) were placed in Eppendorf tubes in the presence of 2.5 U ml<sup>-1</sup> rTxyA and 0.15 U ml<sup>-1</sup> rXloA either alone or in combination (total volume 0.3 ml). The reaction was carried out at 37°C for 12 h. After the hydrolysis reactions were stopped by boiling the tubes for 5 min, the supernatants separated by centrifugation were used for analysis of hydrolysis products. The amount of reducing sugar released from  $\beta$ -1,3-xylan was measured by the Somogyi–Nelson method (Somogyi 1952). The hydrolysis products were detected by thin-layer chromatography (TLC) on a silica gel 60 plastic sheet (Merck, Darmstadt, Germany) with a solvent of *n*-butanol–acetic acid–water (10:5:1, v/v). Oligosaccharides were visualized by spraying

the plate with diphenylamine–aniline–phosphate reagent (Bailey and Bourne 1960).

#### D-Xylulose Production from D-Xylose

The reaction mixture containing 50 mM sodium phosphate buffer (pH 7.5), 1.5% D-xylose, 10 mM MgSO<sub>4</sub>, 4 mM sodium tetraborate, and 0.25 U rXylA in Eppendorf tubes (total volume 1 ml) were incubated at 40°C for 2 h. Sodium tetraborate has been reported to shift the equilibrium constant between D-xylose and D-xylulose to favor the formation of D-xylulose (Hsiao et al. 1982). The borate ion binds more tightly to D-xylulose than D-xylose, effectively reducing the free D-xylulose concentration, and thus leads to increased D-xylose isomerization. The binding ability of borate ion to D-xylulose is pH dependent, with higher pH (6.0–7.5) favoring binding (Hsiao et al. 1982). The borate ion dissociates from D-xylulose in lower pH. To analyze the conversion rates of D-xylose to D-xylulose, the reaction mixtures were boiled for 2 min to inactivate any residual D-xylose isomerase activity. The samples were then centrifuged and the pH of the supernatant was adjusted to 5.0 by 1 M acetic acid. These preparations were used for analyses of the reaction products. Sugar composition was analyzed by high-performance liquid chromatography (HPLC) using a refractive index (RI) detector (Shodex RI-72; Showa Denko, Tokyo, Japan) together with a Shodex Asahipak NH2P-50 4E column (4.6 mm i.d. × 250 mm; Showa Denko). Elution was performed at a flow rate of 1.0 ml min<sup>-1</sup> using acetonitrile–water (75:25, v/v) as the mobile phase. The column temperature was maintained at 40°C.

#### Nucleotide Sequence Accession Number

The nucleotide sequence of the *xylA* gene described here has been submitted to the DDBJ, EMBL, and GenBank databases and assigned the accession no. AB519447.

## Results

#### Cloning and Nucleotide Sequence Analysis of the *xylA* Gene

Nucleotide sequence analysis revealed that the plasmid *xylA*/pBluescript (*SacI*–*EcoRI*) contained 4,242-bp DNA fragment, including the complete *xylA* gene and the other two genes named *xylR* and *xylT*. The *xylA* gene was located in the upstream region of the *xloA* gene in the same direction (Fig. 1). There are only 80 nucleotides between the TAA termination codon of the *xylA* gene and the ATG initiation codon of the *xloA* gene. The *xylA* gene consisted of 1,320-bp nucleotides encoding a protein comprising 439



amino acids with a calculated molecular mass of 49,264 Da. The sequences TTGATG and TAAAAT with 15-bp spacing, having a certain homology to the  $-35$  and  $-10$  promoter consensus sequences of *E. coli*, were identified upstream of the coding region (Rosenberg and Court 1979). A putative ribosome-binding site sequence (GGAA) was located within a few bases upstream of the ATG start codon. A possible transcription terminator that consisted of a 36-bp palindrome sequence corresponding to an mRNA hairpin loop, followed by a T-rich portion, was found downstream of the TAA termination codon (Sugimoto et al. 1995). The *xylT* and *xylR* genes were located in the upstream region of the *xylA* gene in the opposite transcriptional directions, while the C-terminal region of the *xylR* gene was truncated. The N-terminal region (residues 4 to 137) of the *xylR* gene product showed similarities to xylose operon transcriptional regulators from *Klebsiella pneumoniae* 342 (31% identity, GenBank accession no. CP000964; residues 18 to 143) and *Yersinia pestis* KIM (30% identity, AALD0200013; residues 18 to 149). The *xylT* gene consisted of a 1,389-bp nucleotide sequence encoding a protein comprising 462 amino acids with a predicted molecular mass of 50,483 Da. The deduced amino acid sequence of the *xylT* gene product exhibits similarities to several sugar transporters, such as the  $\text{Na}^+$ /melibiose symporter from *Saccharophagus degradans* 2–40 (38% identity, CP000282; residues 1 to 538) and  $\text{Na}^+$ /xyloside symporter from *Lactobacillus brevis* ATCC 367 (31% identity, CP000416; residues 1 to 466).

#### Identification of the $\beta$ -1,3-Xylan Utilization Gene Cluster

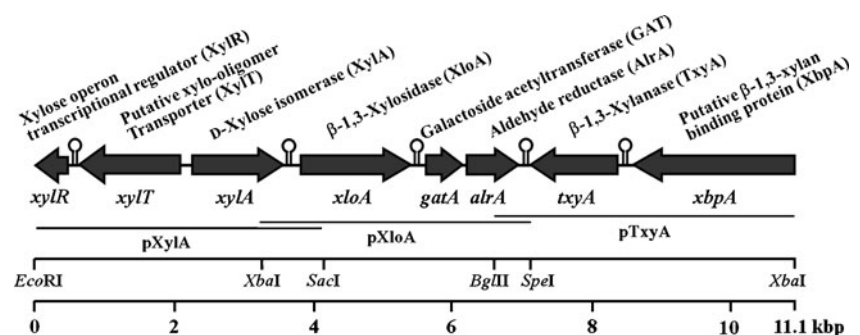
As shown in Fig. 1, the 7,267-bp DNA fragment, which consists of six different genes organized in the order *xylR*, *xylT*, *xylA*, *xloA*, *gatA*, and *alrA*, was generated from the 4,206-bp and 4,242-bp DNA fragments obtained for the cloning of the *xloA* and *xylA* genes, respectively. In addition, we found that the 3' region of *alrA* gene, which encodes an aldehyde reductase (Umemoto and Araki 2008), was overlapped with the 5' region of the 4,409-bp DNA fragment obtained for the cloning of a  $\beta$ -1,3-xylanase gene

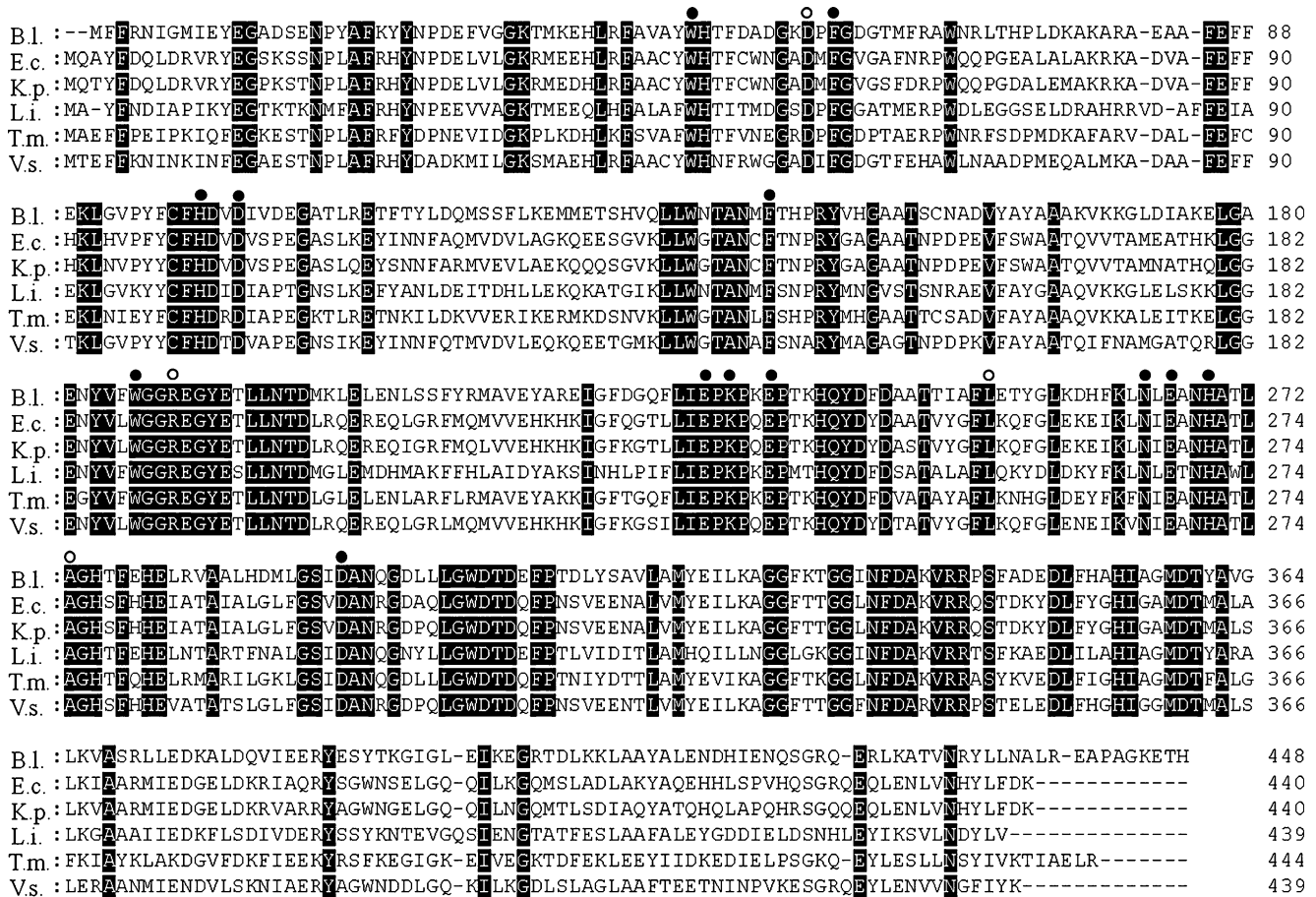
(*txyA*) from *Vibrio* sp. strain XY-214 (Araki et al. 2000). These results indicate that *Vibrio* sp. strain XY-214 has a gene cluster involved in the  $\beta$ -1,3-xylan degradation and D-xylose metabolism. The gene cluster obtained is 11,101-bp long and contains eight genes. There is an incomplete ORF that encodes a protein (XbpA) of unknown function in the upstream region of the *txyA* gene. However, the N-terminal sequence of the XbpA showed high sequence similarity with a family 31 carbohydrate-binding module (CBM31) of TxyA (Okazaki et al. 2002; Kiyohara et al. 2005). Carbohydrate-binding modules (CBMs) are non-catalytic protein modules found in many carbohydrate-degrading enzymes, and they function as recognition modules that convey the catalytic modules of these enzymes to the target substrate (Boraston et al. 2004). The CBM31 has a specific binding ability to  $\beta$ -1,3-xylan (Okazaki et al. 2002). Therefore, it is possible that XbpA is also involved in the  $\beta$ -1,3-xylan metabolism.

#### Similarity of the Deduced Amino Acid Sequence Encoded by the *xylA* Gene

The deduced amino acid sequence of the XylA from *Vibrio* sp. strain XY-214 was scanned with BLAST, and the protein showed similarities to other bacterial XIs. XylA exhibited the high sequence identities to the group II XIs (identity percentages are in parentheses) from *E. coli* (72%, GenBank accession no. AE014075), *K. pneumoniae* 342 (70%, CP000964), *Thermotoga maritima* MSB8 (52%, AE000512), *Bacillus licheniformis* DSM 13 (51%, AE017333), and *Lactococcus lactis* (46%, AF092042). In contrast, XylA showed low sequence identities to the group I XIs from *Streptomyces rubiginosus* (28%, M73789), *Actinoplanes missouriensis* (25%, X16042), and *Arthrobacter* sp. B-5-MG-1 (25%, AY937238). This suggested that XylA of *Vibrio* sp. strain XY-214 should be classified into the group II enzymes. Sequence alignment of XylA with other XIs revealed that all of the active-site residues (W49, F60, H101, D104, F145, W188, E232, K234, E237, N266, E268, H271, and D296) and residues mediating subunit interactions (D58, R191, L254, and A275; Park and Batt 2004) were conserved in XylA (Fig. 2).

**Fig. 1** Genomic organization of  $\beta$ -1,3-xylan utilization gene cluster in *Vibrio* sp. strain XY214. pTxyA, pXloA, and pXylA are the DNA fragments cloned for the *txyA*, *xloA*, and *xylA* genes, respectively. The *xylR* and *xbpA* genes are truncated. Hairpin-like marks indicate putative transcription terminators





**Fig. 2** Sequence alignment of the amino acid sequences of several D-xylose isomerases. From top to bottom: *B. licheniformis* DSM 13 (B.l.), *E. coli* (E.c.), *K. pneumoniae* 342 (K.p.), *L. lactis* (L.i.), *T. maritima* MSB8 (T.m.), and *Vibrio* sp. strain XY-214 (V.s.). Conserved amino

acid residues are *highlighted*. The conserved active-site residues and residues mediating subunit interactions are indicated by *closed and open circles*, respectively

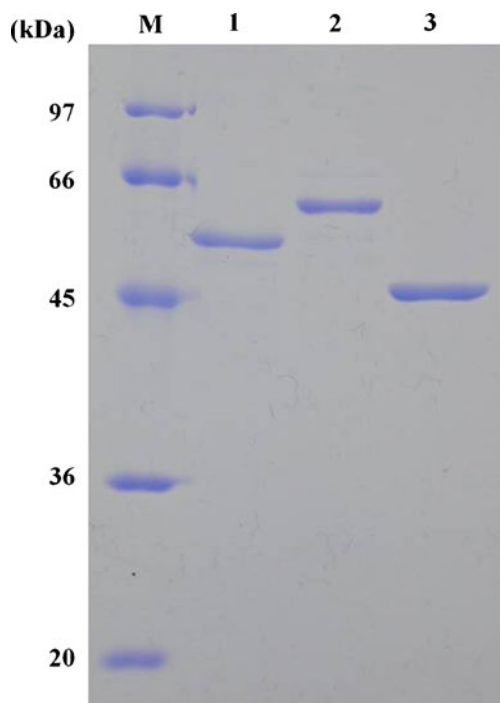
**Biochemical Properties of rXylA**

The purified rXylA was shown to be homogeneous by the detection of a single protein band on SDS-PAGE (Fig. 3, lane 3). Its relative molecular mass was estimated to be approximately 48 kDa. However, the relative molecular mass of the native enzyme was estimated to be 190 kDa by gel filtration chromatography on a Superdex 200 HR 10/30 column (data not shown). According to the nucleotide sequence, the relative molecular mass of the XylA monomer should be 49,264 Da. These results suggest that XylA is a homotetramer, which is the most common oligomer form of D-xylose isomerases (Hess et al. 1998).

The optimum pH of the rXylA was observed to be around 7.5 when the enzyme activity was assayed with D-xylose at 40°C for 20 min at various pHs (3.0–12.0) in Britton and Robinson’s universal buffer (Fig. 4a). The enzyme was stable at pH 6.0–11.0, retaining more than 80% of the maximal activity after preincubation in the buffer at various pHs at 4°C for 12 h (data not shown). The

enzyme was optimally active at around 60°C (Fig. 4b) and stable up to 50°C for 20 min. However, only 34% of the maximal activity was observed after preincubation at 60°C for 20 min, and no activity was observed after preincubation at 70°C for 20 min. The kinetic parameters were determined for D-xylose and D-glucose as substrates from Lineweaver–Burk plots. The  $K_m$ ,  $V_{max}$ ,  $K_{cat}$ , and  $K_{cat}/K_m$  for D-xylose were 7.93 mM, 14.3  $\text{Umg}^{-1}$ , 47.0  $\text{s}^{-1}$ , and  $5.93 \times 10^3 \text{ M}^{-1} \text{ s}^{-1}$ , respectively. On the other hand, the  $K_m$  and  $V_{max}$ ,  $K_{cat}$ , and  $K_{cat}/K_m$  for D-glucose were 187 mM, 1.38  $\text{Umg}^{-1}$ , 4.54  $\text{s}^{-1}$  and  $2.40 \times 10 \text{ M}^{-1} \text{ s}^{-1}$ , respectively.

Typically, two divalent cations ( $\text{Mg}^{2+}$ ,  $\text{Co}^{2+}$ , or  $\text{Mn}^{2+}$ ) per monomer are required for catalytic activity and stability of D-xylose isomerases (Carrell et al. 1989). However, the enzymes from different organisms require different metals for optimum activity. As shown in Table 1, rXylA requires the addition of  $\text{Mg}^{2+}$  for optimum activity, while the addition of  $\text{Co}^{2+}$  or  $\text{Mn}^{2+}$  gave lower activities. Addition of  $\text{Ca}^{2+}$ ,  $\text{Zn}^{2+}$ ,  $\text{Fe}^{2+}$ , or  $\text{Cu}^{2+}$  was not effective. No activity was observed in the absence of metal ions.

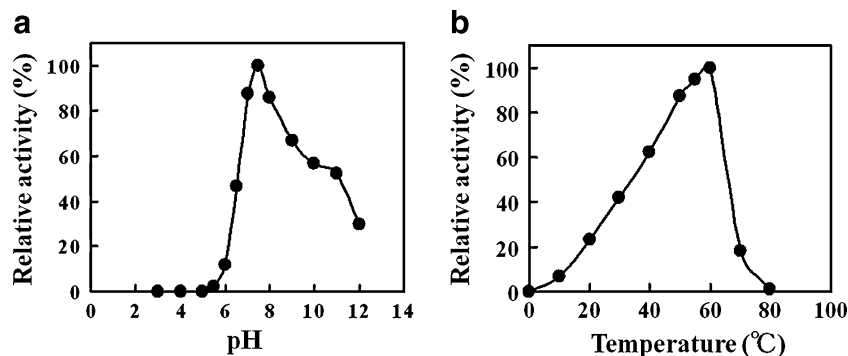


**Fig. 3** SDS-PAGE of purified rTxyA, rXloA, and rXylA. Lane M molecular weight markers, lane 1 rTxyA, lane 2 rXloA, lane 3 rXylA

#### Saccharification of $\beta$ -1,3-Xylan

Saccharification of  $\beta$ -1,3-xylan was carried out by using the purified rTxyA (Fig. 3, lane 1) and rXloA (Fig. 3, lane 2) at 37°C and pH 7.5. rTxyA hydrolyzes  $\beta$ -1,3-xylan to  $\beta$ -1,3-xylooligosaccharides, and then rXloA hydrolyzes  $\beta$ -1,3-xylooligosaccharides to D-xylose. Figure 5a shows the time course of the hydrolysis of  $\beta$ -1,3-xylan. A large amount of reducing sugar was released ( $2.62 \text{ g l}^{-1}$ ) by the co-action of rTxyA and rXloA against  $\beta$ -1,3-xylan, whereas the action of rTxyA alone gave poor hydrolysis ( $1.39 \text{ g l}^{-1}$ ) in a 12 h of hydrolysis. To examine the hydrolysis products released from  $\beta$ -1,3-xylan, TLC analysis was carried out. Samples subjected to a 12 h of hydrolysis were used for TLC analysis. As shown in Fig. 5b, D-xylose was released from  $\beta$ -1,3-xylan as the major hydrolysis product by the co-action of rTxyA and

**Fig. 4** Effects of pH (a) and temperature (b) on D-xylose isomerase activity of the purified rXylA. Each value represents the averages of at least duplicate measurements



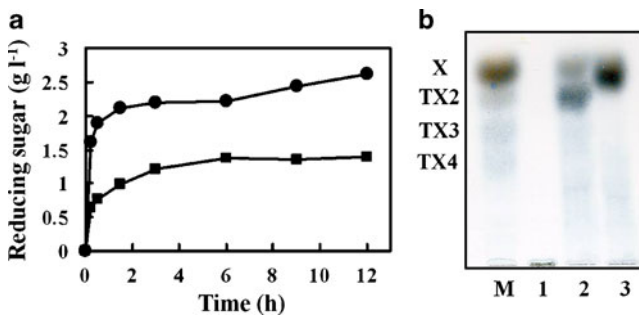
**Table 1** Metal ion requirement for XylA activity

Metal ion	Relative activity (%)
None	0
MgSO <sub>4</sub>	100
CoCl <sub>2</sub>	59
MnCl <sub>2</sub>	41
FeSO <sub>4</sub>	15
CuSO <sub>4</sub>	3
ZnSO <sub>4</sub>	1
CaCl <sub>2</sub>	0

rXloA, whereas the action of rTxyA alone released predominantly  $\beta$ -1,3-xylobiose with D-xylose and  $\beta$ -1,3-xylotriose as minor products. No hydrolysis products were observed by the action of rXloA alone.

#### Isomerization of D-Xylose to D-Xylulose

D-Xylose produced from  $\beta$ -1,3-xylan was converted to D-xylulose by using rXylA because *S. cerevisiae* is unable to ferment D-xylose but can ferment D-xylulose. Figure 6 shows the time course of D-xylulose production from D-xylose by the purified rXylA at 40°C and pH 7.5. After the enzyme reaction was stopped by boiling the tubes, sugar composition was analyzed by HPLC. Although XIs are the enzymes that can isomerize D-xylose to D-xylulose, the equilibrium ratio of D-xylose/D-xylulose is typically 80:20 (Chandrakant and Bisaria 2000). In concordance with this, the conversion rate of D-xylose to D-xylulose by rXylA was found to be approximately 15% after 2 h of incubation. On the other hand, the conversion rate was increased up to around 40% when 4 mM sodium tetraborate was added to the reaction mixture. We investigated the effect of borate ion concentration against the conversion rate of xylose to xylulose. When the xylose isomerase activity of the XylA was measured on the reaction mixtures containing several concentration of sodium tetraborate (0, 1, 2, 4, 6, and 8 mM), the activity increased linearly to 4 mM concen-

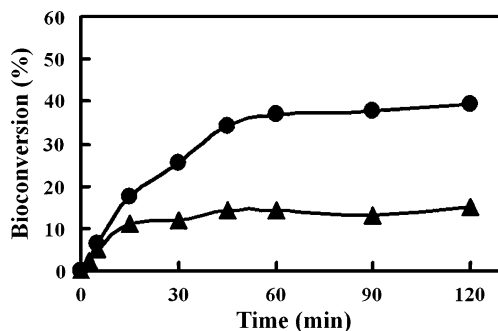


**Fig. 5** The time course of hydrolysis of  $\beta$ -1,3-xylan. **a** The amount of reducing sugar released was measured by the Somogyi–Nelson method; 0.5%  $\beta$ -1,3-xylan in 50 mM sodium phosphate buffer (pH 7.5) were placed in Eppendorf tubes in the presence of 2.5 U ml<sup>-1</sup> rTxyA and 0.15 U ml<sup>-1</sup> rXloA either alone or in combination (total volume 0.3 ml). The reaction was assayed at 37°C. Each value represents the averages of at least duplicate measurements. *Symbols: circles* co-action of rTxyA and rXloA, *squares* action of rTxyA alone, *triangles* action of rXloA alone. **b** TLC analysis of hydrolysis products released from  $\beta$ -1,3-xylan. The reaction was carried out at 37°C for 12 h. *Lane M* standard oligosaccharides, *lane 1* hydrolysis products by the action of rXloA, *lane 2* rTxyA, *lane 3* rTxyA and rXloA. TX2  $\beta$ -1,3-xylobiose, TX3  $\beta$ -1,3-xylotriose, TX4  $\beta$ -1,3-xyloetraose, respectively

tration and arrived at plateau more than 4 mM. Therefore, 4 mM sodium tetraborate was added to the reaction mixture.

## Discussion

The killer alga, mutant *C. taxifolia*, has spread uncontrollably in the Mediterranean Sea since 1984 and has destroyed local ecosystems in these waters. In this study, we tried to develop a method to produce ethanol from  $\beta$ -1,3-xylan, which is a major structural polysaccharide component of the *C. taxifolia* cell wall. To date, we have



**Fig. 6** The time course of D-xylulose production from D-xylose by rXylA. The reaction mixture containing 50 mM sodium phosphate buffer (pH 7.5), 1.5% D-xylose, 10 mM MgSO<sub>4</sub>, 4 mM sodium tetraborate, and 0.25 U rXylA in Eppendorf tubes (total volume 1 ml) were incubated at 40°C. *Symbols: triangles* isomerization in the absence of sodium tetraborate, *circles* isomerization in the presence of 4 mM sodium tetraborate. Each value represents the mean of duplicate measurements

cloned and characterized the genes encoding TxyA (Araki et al. 2000) and XloA (Umamoto et al. 2008) from the marine bacterium *Vibrio* sp. strain XY-214. These two enzymes are required for the saccharification of  $\beta$ -1,3-xylan to D-xylose. Because yeast *S. cerevisiae* cannot ferment D-xylose but can utilize D-xylulose, conversion of D-xylose to D-xylulose by using a D-xylose isomerase is required after the enzymatic saccharification of  $\beta$ -1,3-xylan. In our previous study of the cloning of the *xloA* gene from the strain XY-214, we have obtained a truncated gene (*xylA*) encoding a XylA in the upstream region of the *xloA* gene. In the present study, therefore, we first cloned the complete *xylA* gene from the strain XY-214 and characterized the gene product. Second, we produced D-xylulose from  $\beta$ -1,3-xylan by using three kinds of enzymes (TxyA, XloA, and XylA).

Cloning of the *xylA* gene and the sequence analysis revealed a gene cluster involved in the  $\beta$ -1,3-xylan degradation and D-xylose metabolism in the genomic DNA of *Vibrio* sp. strain XY-214 (Fig. 1). The gene cluster is 11,101-bp long and contains eight genes organized in the order *xylR*, *xylT*, *xylA*, *xloA*, *gatA*, *alrA*, *txyA*, and *xbpA*. To our knowledge, this is the first identification of a  $\beta$ -1,3-xylan utilization gene cluster, although there have been many reports regarding the gene cluster involved in the utilization of  $\beta$ -1,4-xylan (Shulami et al. 1999; Erlandson et al. 2001; Chow et al. 2007). In most cases, the D-xylose isomerase gene forms an operon with a gene encoding D-xylulose kinase (*xylB*), which converts D-xylulose to D-xylulose-5-phosphate (Lawlis et al. 1984; Lokman et al. 1991; Sizemore et al. 1991; Wong et al. 1991; Feldmann et al. 1992). However, we could not detect the *xylB* gene in the upstream and downstream of the *xylA* gene. This indicates that the *xylA* and *xylB* genes are transcribed separately in case of the strain XY-214. On the other hand, the *gatA* and *alrA* genes probably form an operon because they were separated by only 32-bp nucleotides and no transcription terminator was detected between these two genes. The deduced amino acid sequence of the *gatA* gene showed similarities to a GAT from *E. coli* (accession no. CP000243; 43% identity). GAT is encoded by the *lacA* gene of the *lac* operon of *E. coli*. The *lacA* gene is cotranscribed with the *lacZ* and *lacY* genes which encode  $\beta$ -galactosidase and lactose permease, respectively. Although the roles of the latter two enzymes in lactose metabolism have been clearly characterized, the cellular function of GAT remains unclear (Lewendon et al. 1995). The properties of the *alrA* gene product have been investigated previously (Umamoto and Araki 2008). It has been revealed that an AlrA encoded by the *alrA* gene belongs to the aldo-keto reductase superfamily although the cellular functions of AlrA also remain unclear. Therefore, further biochemical studies are required to understand the



roles of GAT and AlrA in the D-xylose metabolism. In order to understand the metabolic mechanism of  $\beta$ -1,3-xylan in *Vibrio* sp. strain XY-214, functional analysis of transcriptional regulation of the gene cluster will be required.

The *xylA* gene of *Vibrio* sp. strain XY-214 has been cloned and expressed in *E. coli*. Sequence analysis revealed that XylA is classified into the group II XIs, which differ from the group I XIs by the presence of approximately 50 additional residues at the N terminus. Although group II XIs share only 20–30% amino acid sequence identity with group I XIs, the active-site residues in group I and group II enzymes are highly conserved (Park and Batt 2004). Sequence alignment with other XIs revealed that all of the active-site residues and the residues having a role in the subunit interaction were conserved in XylA (Fig. 2). Similar to most of XIs reported so far, XylA of the strain XY-214 has been found to be a homotetramer with a molecular mass of 49,264 Da per subunit, whereas some XIs have been found to be dimeric (Batt et al. 1990; Meng et al. 1993). To date, the crystal structures of several group I XIs have been extensively characterized (Farber et al. 1987; Rey et al. 1988; Carrell et al. 1989; Henrik et al. 1989; Dauter et al. 1990; Whitlow et al. 1991; Rasmussen et al. 1994), and it has been revealed that each monomer of XIs consists of an  $(\alpha/\beta)_8$  barrel having an active site and a C-terminal loop lacking  $\beta$ -strands (Park and Batt 2004). The XylA was activated strongly by magnesium ion. Allen et al. describe that magnesium ion is essential for isomerization but not essential for ring opening of D-xylose isomerase structure (Allen et al. 1994). As do other XIs, XylA has a lower  $K_m$  for D-xylose (7.93 mM) than for D-glucose (187 mM). Although XylA shares common properties to most XIs from mesophilic bacteria, this is the first report on the characterization of XIs from *Vibrio* spp.

For the production of D-xylulose from  $\beta$ -1,3-xylan, the saccharification of  $\beta$ -1,3-xylan and isomerization of D-xylose to D-xylulose are required. In this study, each process was performed separately. Saccharification of  $\beta$ -1,3-xylan to D-xylose was carried out by using rTxyA and rXloA at 37°C and pH 7.5. As shown in Fig. 5b, D-xylose was produced as the major hydrolysis product by the co-action of rTxyA and rXloA, whereas the action of rTxyA alone predominantly produced  $\beta$ -1,3-xylobiose, with D-xylose and  $\beta$ -1,3-xylotriose as minor products. This indicated that rTxyA and rXloA are useful tools for the saccharification of  $\beta$ -1,3-xylan. D-Xylose prepared from  $\beta$ -1,3-xylan was then converted to D-xylulose by using rXylA at 40°C and pH 7.5. The conversion rate of D-xylose to D-xylulose reached approximately 15% after 2 h of incubation in the absence of sodium tetraborate. On the other hand, the conversion rate increased up to approximately 40% in the presence of 4 mM sodium tetraborate (Fig. 6). On the other hand, the conversion rate increased up to approximately 40% in the presence of 4 mM sodium tetraborate because borate

ion is combined with xylulose and forms a complex. As a result, free xylulose decreases in the reaction mixture and the conversion efficiency of xylose to xylulose rises (Fig. 6).

In summary, we successfully produced D-xylulose from  $\beta$ -1,3-xylan using three kinds of bacterial enzymes by two steps processing. We are going to develop technology to produce xylulose from  $\beta$ -1,3-xylan directly by one-step method with these enzymes. We have also confirmed that D-xylulose produced from  $\beta$ -1,3-xylan can be fermented to ethanol by yeast *S. cerevisiae* (data not shown). While further studies are required to improve the conversion rate of  $\beta$ -1,3-xylan to D-xylulose, the results of the present study will serve as the basis for the ethanol production from the killer alga *C. taxifolia*.

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