REVIEW

Recent microencapsulation trends for enhancing the stability and functionality of anthocyanins: a review

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Abstract

Anthocyanins (ACNs) are water-soluble pigments in various fruits and vegetables known for their high antioxidant activity. They are used as natural food colorants and preservatives and have several medicinal benefts. However, their application in functional foods and nutraceuticals is often compromised by their low stability to heat, oxygen, enzymes, light, pH changes, and solubility issues. Spray drying has emerged as an efective microencapsulation technique to enhance the shelf life, quality, and stability of ACNs. This manuscript reviews the latest scientifc developments in spray drying microencapsulation of ACNs-rich fruit extracts. Process optimization and the stability and physicochemical properties of the spray-dried, microencapsulated ACNs-rich powders are discussed. This review also covers functional food and nutraceutical applications and introduces novel encapsulation methods, such as freeze-drying, supercritical carbon dioxide (SC-CO₂), coacervation, drum drying, and electrospraying, highlighting their potential in improving the utility of ACNs-rich fruit extracts.

Keywords Anthocyanins stability · Encapsulating agents · Functional food value · Optimized spray drying · Powder product

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Introduction

In the present era of optimal nutrition, the growing demand for healthier foods has led to the consumption of natural antioxidant-containing foods, such as polyphenol-rich fruits and vegetables. Anthocyanins (ACNs) are water-soluble phenolic compounds renowned for their potent natural antioxidant properties. Their unique structure, characterized by a favonoid backbone, multiple hydroxyl groups, conjugated double bonds, chelation sites, glycosylation, and varied substitution patterns, collectively contributes to their exceptional antioxidant properties. These structural attributes enable ACNs to efectively neutralize free radicals, inhibit metal ion-induced oxidative reactions, and protect cells and tissues from oxidative damage, making them valuable natural antioxidants (Enaru et al., [2021;](#page-22-0) Tarone et al., [2020](#page-24-0)).

ACNs are responsible for plants' orange, red, violet, and blue colors and thus hold great potential as natural colorants (Azman et al., [2022a](#page-21-0), [b\)](#page-21-1). Grapes, blackcurrants, blueberries,

and bilberries are among the most popular ACNs-rich fruits. In contrast, red cabbages (Machado et al., [2022](#page-23-0)), eggplants (Demiray et al., [2023](#page-22-1)), and red onions (Ali et al., [2016\)](#page-21-2) are typical examples of ACNs-rich vegetables. In addition to antioxidants, ACNs provide a broad spectrum of therapeutic effects and biological activities, including anticarcinogenic, infammatory, immune-stimulating, antibacterial, antiallergic, and antiviral properties (Nawawi et al., [2023a](#page-23-1); Riberio et al., [2019](#page-23-2); Salehi et al., [2020\)](#page-24-1). However, the biological properties and nutraceutical value of ACNs are often compromised by their instability. The stability of ACNs is afected by multiple factors such as light, temperature, pH, enzymatic reactions, oxidation, presence of metal ions, humidity, microorganisms, moisture, solvents, and the presence of co-pigments, causing degradation or partial loss of these compounds (Azman et al., [2022a,](#page-21-0) [b;](#page-21-1) Feitosa et al., [2023](#page-22-2); Nawawi et al., [2023a;](#page-23-1) Nawawi et al., [2023b](#page-23-3)). In this regard, diferent delivery processes have been developed to improve the stability of ACNs and coloring pigments, such as microencapsulation, nanoencapsulation, and protein complex formation (Feitosa et al., [2023\)](#page-22-2).

Microencapsulation is a viable alternative that efficiently preserves thermally sensitive compounds such as phenolic compounds (Abdel‐Aty et al., [2022;](#page-21-3) Gawalek, [2022\)](#page-22-3). This technique entraps bioactive compounds within the encapsulating agent and transforms liquids into powders for easier handling (Rocha et al., [2019;](#page-22-4) Singh et al., [2023a\)](#page-24-2). Encapsulating agents such as maltodextrin (MD), gum arabic (GA), carrageenan, alginate, waxes, and phospholipids are added to facilitate powder production, protect bioactive compounds from oxygen, light, or other unfavorable conditions, and increase stability (Li et al., [2017;](#page-23-4) Ligarda-Samanez et al., [2023](#page-23-5)).

Many investigations on the microencapsulation of ACNsrich fruit extracts, such as berries, pomegranates, jaboticaba, and cherries, have focused on storage, color stability, and the quality of the fnal fruit powder (Gawalek, [2022](#page-22-3); Halahlah et al., [2023;](#page-22-5) Mahdavi et al., [2016b](#page-23-6); Shwetha and Preetha, [2016](#page-24-3)). Spray drying, freeze-drying, and supercritical carbon dioxide $(SC$ – $CO₂$) technology are commonly used microencapsulation techniques, and spray drying is the most popular among food industries owing to its cost-efectiveness, efficiency, and high yield of good-quality powder products (Ravichandran et al., [2023;](#page-23-7) Da Rosa et al., [2019\)](#page-22-6). Over the years, various parameters of spray-drying microencapsulation techniques have been studied for ACNs from various fruit types. New updates are required to determine which parameters can be applied to best suit specifc fruits.

Therefore, the primary goal of this review was to assess the current trends and advances in spray drying microencapsulation of ACNs-rich fruit extracts. The effects of different parameters, such as types of encapsulating agents, the ratio of encapsulating agent to feed, drying temperature, and feed

fow rate, on the physiochemical and stability characteristics and functional food quality of the end-use ACNs powder product are discussed. Furthermore, an overview of innovative microencapsulation techniques, such as freeze-drying, $SCCO₂$, coacervation, drum drying, and electrospraying, is provided.

An overview of ACNs

The word ACNs has been coined from two Greek words, namely *Anthos* (flower) and *kyanos* (blue) (Koop et al., [2022\)](#page-23-8). As a class of polyphenols and sub-class of favonoids, ACNs are widely distributed in diferent parts, especially leaves, fruits, and fowers of various plants belonging to the *Rosaceae*, *Vitaceae*, *Saxifragaceae*, *Ericaceae*, *Cruciferae*, *Fabaceae* and *Caprifoliaceae* families, among others (Li et al., [2021](#page-23-9); Mazza and Miniati, [2018\)](#page-23-10). ACNs tend to be located in the cells of plant vacuoles and cell walls and can be acquired through extraction (Ijod et al., [2022](#page-22-7)) (Fig. [1](#page-2-0)). ACNs yield in fruits depends on several factors such as processing and extraction methods, growing location and weather, fruit maturity levels, and storage conditions before analysis (Mazza and Miniati, [2018](#page-23-10); Nawawi et al., [2023a\)](#page-23-1).

The concentration of ACNs varies from 0.1 to 1.0% across various plant materials (Ercoli et al., [2021\)](#page-22-8). Vegetables (e.g., red potato, purple sweet potato, eggplant, carrot, purple corn, red onion, and red cabbage) and fruits (e.g., berries,

cherries, blood oranges, blackcurrants, pomegranate, grapes, plums, and apples) are essential sources of ACNs for human nutrition (Karacabey et al., [2023;](#page-23-11) Mohammed and Khan, [2022](#page-23-12); Tan et al., [2023\)](#page-24-4). Berries, grapes, blackcurrants, and some tropical fruits have been identifed as abundant sources of ACNs (Khoo et al., [2017\)](#page-23-13). Due to their broad range of nutra-pharmaceutical attributes and chemo-preventive efects, as evidenced by clinical trials, ACNs have emerged as promising natural compounds with the potential to replace synthetic additives or colorants (Thakur et al., [2023](#page-24-5)).

ACNs consist of two aromatic rings attached to three carbons in an oxygenated heterocycle and a chromane ring with a second aromatic ring (Tarone et al., [2020\)](#page-24-0) **(**Fig. [2](#page-3-0)**)**. ACNs have sugar at R_3 positions called oligosaccharide side chains, normally in a single glucoside unit. The amount of hydroxyl and methoxy groups indicates the level of intensity and type of color of the ACNs. The higher amounts of hydroxyl and methoxy groups contribute to the extracts' intense bluish and redness colors, respectively (Enaru et al., [2021](#page-22-0)).

ACNs are also known as the glycosylated (aglycone) forms of anthocyanidins. Chemically, anthocyanidins are sugar-free counterparts of ACNs. As natural dyes, these compounds are responsible for the color of many fruits (Khoo et al., [2017\)](#page-23-13). There are six main types of glycosides derivative ACNs: pelargonidin, cyanidin, delphinidin, peonidin, petunidin, and malvidin. Interestingly, pelargonidin displays a red-colored pigment in fruits and berries but appears orange in fowers. Besides, cyanidin appears reddish-purple

Fig. 1 ACNs move from the synthesis site, the endoplasmic reticulum (ER), to the vacuole for storage. ER-derived vesicles facilitate anthocyanin transport to the vacuole, where they bind to the membrane via soluble N-ethylmaleimide-sensitive factor attachment protein receptors (SNARE) and release ACNs during the micro (1) and macro (2) autophagy processes. Several membrane proteins (multidrug and toxic compound extrusion (MATE), ATP-binding cassette

(ABC), and bilitranslocase (BTL-like transporters) aid in the transport of ACNs into vacuoles and their sequestration in vacuolar inclusions (AVIs) in the membrane transporter-mediated pathway (3), Glutathione S-transferases (GSTs) mediate the conjugation of ACNs to generate the glutathione-ACNs conjugate, which serves as an intact and efficient means of transport from the ER to the vacuole. Adapted from Nistor et al. [\(2022](#page-23-14)) with modifcation

Fig. 2 Microencapsulation of ACNs extracts from various food sources and general chemical structure of ACNs and their anthocyanidins. R3=sugar (glucose, arabinose, galactose, etc.). Adapted from Kozłowska and Dzierżanowski [\(2021](#page-23-15)) with modifcation

or magenta, while delphinidin causes a blue-reddish or purple pigment in plants. Examples of methylated anthocyanidins include peonidin, malvidin, and petunidin. Peonidin contributes to the magenta color in grapes, berries, and red wines. Also, malvidin displays as a darker rusty red pigment in red wine, while petunidin appears as a dark red or purple pigment in blackcurrants and purple petals of the fower (Khoo et al., [2017](#page-23-13)). These color variations can result from the complex interplay between pH, co-pigments, metal ions, genetics, environmental factors, chemical modifcations, and concentration levels (Enaru et al., [2021](#page-22-0)).

Stability of ACNs

ACNs, which are hydrophilic, are highly unstable compounds and are quickly degraded due to diferent factors. The stability of ACNs is infuenced by their concentration, pH, storage temperature, chemical structure, the presence of enzymes, proteins, favonoids, metal ions, oxygen, and light (Enaru et al., [2021;](#page-22-0) Jafari et al., [2016\)](#page-22-9). Their stability is also afected by the presence of hydroxyl or methoxy functional groups in the structure, as the presence of these clusters decreases the stability of the compound in an organic solvent or aqueous solution (Khoo et al., [2017](#page-23-13)).

The quality of ACNs degrades during processing and storage, thereby reducing the efectiveness of their potential role in the food and pharmaceutical industries. Therefore, preventive measures must be employed to maintain the quality of these compounds, particularly during thermal processing (Ijod et al., [2022](#page-22-7)). The degradation of ACNs occurs when they are exposed to temperatures exceeding 60 °C for prolonged periods, typically exceeding 60 min. Such conditions can break covalent bonds within ACN molecules and induce oxidation processes, resulting in a loss of color and potential health benefts (Ali et al., [2016](#page-21-2)). In blanched purple potatoes (95–97 °C/2 min), a 63% loss in total monomeric ACNs was recorded compared to that in fresh potatoes (Karacabey et al., [2023](#page-23-11)). Also, hot drying (60℃, 10 h), hot water blanching (5 min), and steaming (100℃, 5 min) of red cabbages resulted in ACNs losses of 60, 23 and 13%, respectively (Tan et al., [2023\)](#page-24-4).

Improving ACNs stability: chemical and biological approaches

The therapeutic applications of ACNs are often limited because of their reduced stability and low solubility in aqueous and organic media. Interestingly, ACNs can be transformed into acylated or glycosylated derivatives using enzymatic, chemical, or chemoenzymatic approaches. For example, converting ACNs into bioconjugates via fatty acid (FAs) acylation may offer the opportunity to positively modify these compounds' physicochemical properties and biological functionalities (Khoo et al., [2017](#page-23-13)). Bioconjugates, a novel class of hybrid materials consisting of a synthetic macromolecule linked to a biomolecule/ biological entity such as peptides or proteins, vitamins, and nucleic acids, are gaining increasing importance in the felds of medicine, biotechnology, and nanotechnology (Li and Mahato, [2017](#page-23-16)).

The stability of ACNs can also be improved through intramolecular copigmentation, wherein the color intensity of anthocyanidins/ACNs is increased or reinforced in the presence of other favonoids as cofactors or copigments (Azman et al., [2022a,](#page-21-0) [b](#page-21-1)). In this phenomenon, due to color intensifcation, an increase in color intensity with spectral shifts towards higher wavelengths can be observed with the addition of a copigment to acidic (preferably), neutral, and even slightly alkaline ACNs aqueous solutions.

Improving ACNs stability: physical approaches

In addition to bioconjugation and copigmentation, the stability and functional properties of ACNs can be improved using microencapsulation. Microencapsulation is a viable alternative for enhancing the stability and practical uses of ACNs on an industrial scale. In this process, ACNs, being the core material (active agent), are entrapped within another substance (wall material/coating material) or encapsulants such as starches, gelatin, GA, and MD (Estupiñan-Amaya et al., [2023;](#page-22-10) Mahdavi et al., [2016a](#page-23-17); Rocha et al., [2019\)](#page-22-4).

Optimization of various parameters (type and concentration of the encapsulating agent, concentration of the active agent, ratio of coating/active material, process time, temperature, etc.) is required to establish the best conditions for microencapsulation, offering better yield and good quality of the end-use powdered product, which has been extensively reviewed by Tarone et al. ([2020](#page-24-0)). In this context, conventional optimization methods are now being replaced by modern approaches, such as response surface methodology (RSM) and in silico (computer simulation) modeling studies, to devise an optimized microencapsulation process (Machado et al., [2022;](#page-23-0) Mahdavi et al., [2016a](#page-23-17); Tarone et al., [2020](#page-24-0)).

Microencapsulation provides many advantages to the fnal product by diminishing the surroundings and protecting bioactive compounds/ACNs from side effects caused by air, light, moisture, and heat. This process can deteriorate the vapor of the inside material to the outside environment and modify the physical characteristics to make ACNs or related bioactive materials more convenient to use. Also, microencapsulation is efficient in masking the inside material's flavor and forming two phases when mixed with liquid or semisolid products (Ray et al., [2016\)](#page-23-18). Powdered particle size can be categorized as macro ($>$ 5000 µm), micro (1–5000 µm), and nano $(<1 \mu m)$ (Jafari et al., [2008\)](#page-22-11).

Spray drying microencapsulation

Spray drying, due to its versatility, cost-efectiveness, and ease of operation, is the most common and widely applicable technique for the encapsulation of bioactive such as ACNs-rich extracts (Da Rosa et al., [2019;](#page-22-6) Ravichandran et al., [2023\)](#page-23-7) (Fig. [2\)](#page-3-0). Spray drying process is the traditional method for forming powders from semi-solid or liquid forms. Most of the 15,000 industries have used spray dryers to form products, such as luminescence materials, oxides, chemicals, fertilizers, and dried foods (Nandiyanto et al., [2019\)](#page-23-19).

Additionally, it is a process in which the industry can manage acceptable levels of deterioration and decomposition of volatile compounds such as fruit juice. Spray drying, which involves encapsulation by creating protective 'walls' around sensitive ingredients, converts liquids into solids and enhances their shelf-life and color stability while safeguarding them against oxidation (Vasile et al., [2023](#page-24-6)). The reduced volume resulting from spray drying also simplifes the handling and storage of the compounds.

The principles of spray drying include preparation, homogenization, atomization, dispersion, and dehydration of the liquid solution. A critical issue during this process is wall deposition, which may afect the quality and quantity of the product that needs to be achieved. The occurrence of wall deposition depends on the spray dryer's type, size, and operating parameters. Appropriate measures and controls are needed to avoid wall deposition, thus preventing the high maintenance cost and lowering the powder yield (Tarone et al., 2020). The efficiency of spray drying depends on the selection of parameters, such as the encapsulating agent and its concentration added to the feed (active agent), inlet and outlet temperatures, atomization speed or pressure, and feed fow rate (Gawalek, [2022;](#page-22-3) Pan et al., [2022;](#page-23-20) Vasile et al., [2023\)](#page-24-6). These factors must be controlled and manipulated for acceptable physical properties and a higher powder yield.

Encapsulating agents

The limitation of the spray drying technique is the use of high temperatures for drying and air access (Bednarska and Janiszewska‐Turak, [2019](#page-22-12)), which may decompose thermally sensitive ACNs (Gawalek, [2022\)](#page-22-3). Therefore, encapsulating agents are introduced in the spray drying to facilitate powder production, especially for ripe fruits with high °Bx value. This high °Bx value corresponded to a low glass transition temperature (Tg). When ripe fruits dehydrate above Tg, they may exhibit stickiness and adhere to dryer walls, resulting in a reduction in the fnal yield of the product (Zotarelli et al., [2017\)](#page-25-0). To prevent stickiness, one option is to control the glass transition temperature so that it remains below $Tg + 20$ °C. High molecular weight encapsulating agents can also improve the product's glass transition (Machado et al., [2022](#page-23-0)).

Encapsulating agents can be divided into groups of carbohydrates, proteins, or a combination of both. Carbohydrates act as a protective barrier from the external environment to the inside material, such as gums, starch, modifed starches, dextrins, and cellulose (Halahlah et al., [2023](#page-22-5); Lacerda et al., [2016](#page-23-21); Villacrez et al., [2013](#page-24-7)). The most essential characteristics of this group are their emulsifying activity and solubility properties. These agents should exhibit characteristics such as low viscosity, nonhygroscopic, bland favor/tasteless, non-reactive with core materials, soluble in aqueous solvents, inexpensive, foodgrade, fexible, rigid, thin, and pliable (Tan et al., [2015\)](#page-24-8).

MD has good water solubility and low viscosity and is produced by enzymatic or acid hydrolysis of starch (Lacerda et al., [2016](#page-23-21)). The low emulsifying properties of MD can be counteracted by substituting it with sodium octenyl succinate (OSA) starch. OSA starch introduces lipophilic elements, resulting in amphiphilic properties that enhance emulsifcation (Lacerda et al., [2016](#page-23-21); Sweedman et al., [2013\)](#page-24-9). Another encapsulating agent in carbohydrates is inulin, which is a polysaccharide. This polysaccharide consists of fructose units linked by β -(2,1) bonds with glucose in the chain. Inulin is derived from chicory and has dietary fber and prebiotic efects on consumers (El-Kholy et al., [2020](#page-22-13); Lacerda et al., [2016](#page-23-21)).

Examples of protein types include whey protein isolates (WPI) and soy protein isolates (SPI) (Robert and Fredes, [2015\)](#page-24-10). These proteins excel as encapsulating agents because of their distinct attributes and adaptability. Whey protein isolates sourced from whey, a byproduct of cheese production, are renowned for their high nutritional quality as complete proteins containing all essential amino acids. They also exhibit exceptional emulsifying properties, stabilize emulsions, and efficiently encapsulate lipophilic compounds. Moreover, WPI has a low allergenicity, making it suitable for a wide array of applications, and its neutral favor accommodates the encapsulation of diverse ingredients. On the other hand, SPI, derived from soybeans, offers versatility by encapsulating a broad spectrum of ingredients, including favors, vitamins, minerals, and lipids. With its high protein content, SPI enhances the nutritional value of encapsulated products. In contrast, its functional properties, such as emulsifcation and flm formation, further contribute to its efectiveness in encapsulation applications. Being gluten-free and sustainably sourced from soybeans, SPI aligns with dietary preferences and environmentally conscious practices (Bian et al., [2022\)](#page-22-14).

The selection of encapsulating agents depends on the solubility of the bioactive compound of interest, which is either hydro- or lipo-soluble. GA is the best encapsulating agent for liposoluble bioactive compounds, and hydro-soluble compounds such as ACNs and MD with diferent dextrose equivalent (DE) values, GA, or modifed starch are commonly used. Due to its low hygroscopicity, the prominent MD used in spray drying is MD 10-DE (De Souza et al., [2015](#page-22-15)).

Encapsulating agents from natural polymers can prevent the degradation of ACNs and aid in their delivery to the human body for nutraceutical applications (Vergara et al., [2020](#page-24-11)). Combining OSA starch, inulin, and MD as encapsulating agents with an encapsulating carbohydrates (EC) ratio of 2:1:1 (2/3:1/6:1/6) produced jussara pulp microparticles with favorable properties. Specifcally, these microparticles exhibited excellent color, antioxidant activity, and ACNs content, making them a promising encapsulation approach for preserving the quality and functionality of the jussara pulp (Lacerda et al., [2016\)](#page-23-21).

The production of ACNs-rich powders from various fruit sources using spray drying is summarized in Table [1](#page-6-0). Overall, the SPI encapsulation technique was highly efficient for polyphenols (catechins, ellagitannins, gallitannins, and quercetin glycosides). This is due to the nature of the charge of bioactive compounds, where polyphenols and ACNs have negative and positive charges, respectively. Also, the polyelectrolyte structure (density and charge type) contributes to the interaction of SPI with the bioactive polymer. Meanwhile, MD was more efective in entrapping ACNs owing to the production of larger and smoother particles of fruit powder than GA, which produced smaller particles and wrinkled surfaces, thus making it easy to be exposed to oxygen (Ferrari et al., [2013](#page-22-16)).

Specific inlet and outlet temperatures are needed to maximize encapsulation efficiency and prevent the acceleration of ACNs, polyphenols, and antioxidant degradation (Gawalek, [2022\)](#page-22-3). Higher inlet temperatures may produce lower-quality ACNs powder, such as a dense surface layer, thereby reducing the efficiency of powder reconstitution (Jafari et al., [2017\)](#page-23-22). In some instances, spray drying using a high inlet temperature and aspiration rate may contribute to the high yield of microparticles. Consequently, this can reduce the stickiness of fruit powder on the cyclone wall (Yingngam et al., [2018\)](#page-24-12). A more signifcant loss of ACNs was detected when the air inlet temperature was higher than 180℃ (Gawalek, [2022\)](#page-22-3). A higher inlet temperature with a lower concentration of encapsulating agents below 30% (w/w) leads to decreased ACNs stability due to non-enzymatic browning and pro-anthocyanidin deterioration (De Souza et al., [2015\)](#page-22-15).

A combination of encapsulating agents, such as protein-based products, protects ACNs better than a single

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encapsulating agent. Combining encapsulating agents is assumed to create a strong interaction, leading to a synergistic efect in improving stability. The temperature and presence of light also infuence the stability of ACNs during storage. Low temperatures and dark conditions during storage improve the stability of ACNs and prolong their shelf-life. Overall, optimizing spray drying conditions, particularly drying temperature, is crucial to ensure polyphenol encapsulation, as presented in Table [1](#page-6-0).

Physicochemical, functional food, and nutraceutical attributes of encapsulated ACNs–rich powder

Moisture content, water activity, and particle size are the most critical parameters for producing microencapsulated ACNs powders. The low moisture content of microencapsulated fruit powder is vital to achieve excellent stickiness, flowability, and storage stability and prevent microbial growth. This is due to the higher water activity, which provides more free water space for microbial growth. Notably, the water activity level of the microencapsulated powder produced by spray drying is below 0.3, efectively inhibiting microbial growth (Todorović et al., [2022](#page-24-22)).

Furthermore, the particle size of the powder affects the texture and nutritional properties of the food product. The optimal size range for microencapsulated powder produced by spray drying is 10−100 μm (Da Rosa et al., [2019](#page-22-6)). Solubility is also salient because it provides desirable properties such as dispersibility, solubility, and wettability. These properties contribute to the rehydration of food ingredients; thus, a low moisture content of the powder is desired (Mahdavi et al., [2016a](#page-23-17)). As shown in Fig. [3,](#page-11-0) good solubility of the microencapsulated pomegranate powder was achieved when GA was used as an encapsulating agent. However, its optical properties are poor compared to encapsulation with waxy starch and MD and without cellulose (Yousefi et al., [2010](#page-25-1)).

Santos et al. ([2019](#page-24-17)) found that using MD in blackberry spray drying efficiently preserved the physicochemical characteristics of the end-use powder product. This was due to the MD characteristics having a less hygroscopic nature, which resulted in improved ACNs retention, low moisture content, and excellent powder reconstitution. With these properties, this spray-dried blackberry powder is easily soluble in water and can be efectively applied to produce juice fruit powder.

The microencapsulation of ACNs provides multiple benefits to human health. For example, the addition of encapsulated ACNs powders into yogurt, ice cream, or other desserts can serve as a prebiotic, assisting in addressing digestive system issues. Using inulin to encapsulate ACNs in cornelian cherry fruit extract has an antidiabetic effect at 1 mg/mL. It is suitable as a food ingredient for diabetic consumers (Enache et al., [2020\)](#page-22-21). Besides, this product suits a vegan diet and consumers with lactose intolerance and diabetic problems (Dias et al., [2020](#page-22-22); Enache et al., [2020\)](#page-22-21). In the simulated gastrointestinal studies by da Rosa et al. ([2019](#page-22-6)), microencapsulated blueberry extracts successfully improved ACNs digestion than unencapsulated blueberry extracts. Oancea et al. ([2018\)](#page-23-26) also reported that using WPI as an encapsulating agent could facilitate the release of ACNs into the intestine. The presence of various enzymes and different pH levels

Fig. 3 Efect of encapsulating agent on the microstructure of microencapsulated ACNs powder with (**a**) GA (12%, w/v), (**b**) waxy starch $(12\%, w/v)$, and (**c**) MD (12%, w/v) without cel-lulose from Yousefi et al. [\(2010](#page-25-1))

in the human digestive system makes it a complex system. This affects the stability and bioavailability of the ACNs. Combining encapsulating agents, as opposed to single encapsulating agents, is a successful approach. The synergistic effects of these encapsulating agents improve the stability and bioavailability of the ACNs. This implementation enables a controlled-release mechanism that efficiently administrates ACNs to targeted organs or systems, thereby promoting the overall improvement of human health (Fig. [4\)](#page-12-0) (Da Rosa et al., [2019;](#page-22-6) Enache et al., [2020](#page-22-21); Mansour et al., [2020](#page-23-27); Oancea et al., [2018\)](#page-23-26). Given these potential health advantages, producing microencapsulated ACNs powder with high solubility is essential and possesses sustained-released properties for its effective use in dry mixes or instant health foods**.**

Other innovative microencapsulation techniques

In addition to spray drying, various microencapsulation approaches have been studied and verified for their efficiency in encapsulating ACNs. Promising techniques for ACNs include freeze-drying, SC – $CO₂$, coacervation, drum-drying, and electrospraying microencapsulation. Each approach has unique benefits regarding quality, compound preservation, and encapsulation effectiveness. The following section will discuss their principles, benefits, and prospective uses for protecting the durability and efficacy of ACNs in detail.

Freeze‑drying

The freeze-drying technique **(**Fig. [5](#page-13-0)a**)** relies on the sublimation of water from frozen material and has been explored as an alternative approach for encapsulating ACNs (Fredes et al., [2018\)](#page-22-19). This process involves freezing, sublimation, desorption, and storage (Bhatta et al., [2020\)](#page-22-23). The sublimation phase efficiently extends shelf life and preserves the quality of heat-sensitive food materials. As a simple technique for encapsulating water-soluble essences such as ACNs and other natural aromas or medications, freeze-drying is one of the most convenient methods for drying thermosensitive substances that are unstable in aqueous solution (Azarpazhooh et al., [2018](#page-21-5); Estupiñan-Amaya et al., [2020\)](#page-22-24) (Table [2](#page-14-0)**).**

As shown in Table [2](#page-14-0), MD frequently provided excellent ACNs retention and stability results, particularly at low DE values. A low DE generates low hygroscopicity in the ACNs-rich powder, which minimizes moisture absorption and accelerates powder deterioration. Moreover, MD is highly soluble, blending effortlessly with water and producing potent combinations when combined with ACNs.

The concentration of the encapsulating agent affects ACNs stability, as a high amount of the encapsulating agent serves as a solid wall to protect the core molecule. Regular integration of encapsulating agents improves ACNs retention during storage. This combination initiates the construction of a dual-property wall that regulates compound delivery while simultaneously increasing compound stability. This synergistic efect was observed when MD was combined with the diferent encapsulation agents.

Fig. 5 Schematic diagram of (**a**) a typical freeze-drying mechanism, (**b**) supercritical conditions, (**c**) ACNs coacervation, (**d**) double drum drying mechanism, (**e**) a typical electrospraying mechanism. Adapted from Bigazzi et al. ([2020\)](#page-22-26) with modifcations

Supercritical carbon dioxide (SC-CO₂)

 $SC\text{-}CO₂$ is an emerging method that is inert, non-toxic, non-fammable, low cost, environmentally friendly, versatile, and free from toxic excess in the yield of the product formed (Fig. [5b](#page-13-0)) (Da Fonseca Machado et al., [2018](#page-22-25)). $SCCO₂$ has excellent solvents such as carbon dioxide $(CO₂)$, ethane, water, propane, and dimethyl ether, which can be categorized as having gas-like low viscosity, intermediate difusivity, and liquid-like high density. This technique may overcome the disadvantages of the conventional spray drying technique by applying the processing medium with the conditions above its critical point, 31.1 °C and 7.4 mPa (Jang and Koh, [2023\)](#page-23-28), to precipitate and encapsulate the ACNs

(Da Fonseca Machado et al., [2018\)](#page-22-25). A supercritical fuid is above its critical point where gas and liquid exist in the equilibrium phase, and this fuid is known as a pure substance (Wang et al., 2020). This technique's efficiency depends on the active ingredient's thermodynamic properties, encapsulating agents, suitable co-solvents, and the materials used.

As shown in Table [2](#page-14-0), the ACNs exhibited more excellent stability when encapsulated in polyethylene glycol (PEG) and polyvinylpyrrolidone (PVP) using $CO₂$ as the solvent and ethanol as a co-solvent. It is worth noting that these two solvents are generally recognized as safe (GRAS). The improved ACNs retention during SC – $CO₂$ encapsulation can be attributed to the solubility of PEG and PVP in ethanol. PVP and PEG are both soluble in ethanol, which facilitates their combination with ACNs to create a stable solution before encapsulation.

Coacervation

Coacervation microencapsulation involves the separation of one or more hydrocolloids from the original solution. Following this separation, the newly created coacervate phase surrounds and encapsulates the active ingredient, which is either suspended or emulsifed within the same reaction medium (Fang and Bhandari, [2010\)](#page-22-29) (Fig. [5c](#page-13-0)). Coacervation encapsulation can be accomplished using a single colloidal solute such as gelatin, or through a more intricate method involving substances such as gelatin and gum acacia. Although complex coacervation often lacks specifc shapes and is deemed an expensive method for encapsulating food items, it is important to weigh its potential advantages. Specifcally, they can be valuable for encapsulating sensitive and high-value functional ingredients, including ACNs (Devi et al., [2023\)](#page-22-30).

Devi et al. [\(2023](#page-22-30)) also found that using the dual emulsion method followed by complex coacervation with gelatin and acacia gum enhanced the microencapsulation of ACNs from black rice bran. This approach improved the encapsulation efficiency and thermal stability and ensured better stability of ACNs during storage at both 7℃ and 37℃. The microcapsules exhibited decreased moisture content, hygroscopicity, and solubility. Additionally, their appearance was characterized by smooth, circular, or intact surfaces and frm and agglomerated structures. The fndings in Table [2](#page-14-0) indicate that the coacervation formulation efectively extended the shelf life of ACNs, especially under high-temperature conditions, compared to ACNs that were not encapsulated within a coacervation complex. This indicates that the coacervation formulation formed between the diferent encapsulating agents strengthens their heat stability, thereby improving the protection of the core materials. Also, the selection of gumbased materials signifcantly infuences the complexity and stability of coacervation formulations. This factor is crucial for dealing with sensitive compounds.

Drum drying

Drum drying is widely employed in food and chemical sectors to produce powdered or granular substances. This process involves spreading a liquid or slurry in a thin, even layer on the surface of a heated revolving drum to dry the material. The material underwent drying upon contact with the internally heated surface of the rotating drum (Fig. [5d](#page-13-0)) (Sakulnarmrat et al., [2021b;](#page-24-27) Sakulnarmrat and Konczak, [2022](#page-24-28)). Encapsulating agents are often used to protect ACNs during microencapsulation. Meanwhile, Senevirathna et al. ([2021\)](#page-24-30) produced purple sweet potato powder with the addition of citric acid rather than encapsulating it with encapsulating agents. In comparison to the control, the powder made with 0.6% citric acid had a higher concentration of ACNs, antioxidant activity, and an intense red color. As a result, it can be determined that factors such as steam pressure, drum rotation speed, and citric acid content infuence the powder quality, which can be optimized using RSM.

Based on the fndings in Table [2,](#page-14-0) it can be summarized that even though a similar combination of encapsulating agents was incorporated during drum drying, the encapsulation efficiency was different, probably because of the different sources of ACNs, which possess diferent properties and interactions with encapsulating agents. Therefore, optimization must be performed to select the best concentration of both encapsulating agents to address this issue.

Electrospraying

Electrospraying encapsulation involves the formation of nanodroplets by using a high-voltage electric feld. The voltage, solution feed rate, solution properties, humidity, temperature, and separation from the needle tip to the collector are only a few variables that might infuence the ultimate output (Atay et al., [2018](#page-21-6)). The electrospraying equipment included a syringe pump, voltage power supply, collector, and syringe (Fig. [5e](#page-13-0)). The electrospraying process requires injecting a syringe-fed mixture of ACNs and encapsulating agents into a liquid medium before an electric feld is applied at the nozzle. Therefore, it overcomes the surface tension and produces a cone-shaped droplet called a Taylor cone. Increasing the electric feld causes the Taylor cone to become fragile and expels tiny droplets of encapsulated ACNs. Therefore, the solvent evaporates in the air, causing the droplets to coagulate as small particles and accumulate in the collector (Atay et al., [2018;](#page-21-6) González-Cruz et al., [2020\)](#page-22-28).

The solution properties, including viscosity, surface tension, pH, and electrical conductivity, are crucial elements that must be considered to ensure that ACNs can undergo electrospraying. Improper solution properties cause undesirable particles to form, eventually reducing encapsulation efficiency (Atay et al., [2018\)](#page-21-6). As reported by González-Cruz et al. (2020) (2020) , the addition of $10 - 20\%$ zein successfully encapsulated ACNs from blueberries; further increases in concentration caused instability and clogging in the nozzle. Adding ACNs to $10 - 30\%$ agave fructans or WPI resulted in instability during the electrospraying process. Based on Table [2](#page-14-0), it can be inferred that various factors must be considered during the encapsulation of ACNs by electrospraying. Using a combination of encapsulating agents has improved protective properties and the controlled release of ACNs. Nonetheless, when employing a single encapsulating agent, the concentration and molecular weight of the agent play critical roles in achieving high-quality encapsulated ACNs.

Encapsulation efficiency across various microencapsulation techniques

The encapsulation efficiency (EE) is a crucial parameter that determines the ability of the microencapsulation process to protect targeted compounds. The data presented by various authors in Table [1](#page-6-0) and Table [2](#page-14-0) for various microencapsulation techniques employed for encapsulating ACNs were used to illustrate the potential of their encapsulation efficiency (Fig. [6](#page-18-0)). SD and drum drying consistently showed high EE

Fig. 6 Percentage encapsulation efficiency (EE) of various encapsulated ACN powders by referring to the optimal operation of diferent encapsulation techniques as reported by diferent authors. The straight line indicates 90% of EE. Diferent bar colors indicate the different techniques. Yellow, spray drying; blue, freeze-drying; purple, SC-CO₂; orange, coacervation; green, drum drying; dark red, electrospraying; numbers 1–17 referred to diferent authors. 1 – Fredes et al. [\(2018](#page-22-19)), 2 – Yingngam et al. ([2018\)](#page-24-12), 3 – Ribeiro et al. [\(2019](#page-23-2)), 4 – Xue et al. ([2019\)](#page-24-15), 5 – Pan et al. ([2022\)](#page-23-20), 6 – Laureanti et al. [\(2023](#page-23-25)), 7 – Zahed et al. ([2023\)](#page-25-2), 8 – Jang and Koh ([2023\)](#page-23-28), 9 – Nguyen et al. ([2022\)](#page-23-24), 10 – Xue et al. ([2019\)](#page-24-15), 11 – Santos et al. [\(2013](#page-24-23)), 12 – Gharanjig et al. [\(2020](#page-22-27)) 13 – Sarkar et al. ([2020\)](#page-24-26), 14 – Sakulnarmrat and Konczak [\(2022](#page-24-28)), 15 – Sakulnarmrat et al. ([2021b](#page-24-27)), 16 – Atay et al. ([2018\)](#page-21-6), 17 – González-Cruz et al. [\(2020](#page-22-28))

values ranging from 91.14%–99.80% and 98.85%–98.86%, respectively. In contrast, FD exhibited slightly more variability in EE, ranging from 85.00 to 98.33%. Although FD yielded a high EE in some studies (98.33%), some authors reported values lower than 90%, indicating an inconsistency in achieving at least 90% EE. Other techniques, such as $SC\text{-}CO₂$, showed an EE higher than 90%. Coacervation and electrospraying display varying degrees of efficiency, indicating the infuence of the process parameters and formulation characteristics. The aforementioned techniques had the lowest EE, ranging from 44.77%–86.00% and 52.65%–76.90%, respectively. Considering the EE data, SD is the best and most promising choice for ACNs encapsulation, owing to its consistently high efficiency, relatively straightforward process, and cost-efectiveness.

Current status of product application with microencapsulation techniques

The increasing awareness and demand for healthy products among consumers requires the food industry and researchers to determine a solution for incorporating and protecting bioactive compounds, such as ACNs, in products. Therefore, microencapsulation is a promising solution for protecting ACNs. Many researchers have compared and determined the efects of diferent parameters and microencapsulation techniques on the stability of ACNs after incorporation into the product (Mihalcea et al., [2020;](#page-23-31) Sakulnarmrat et al., [2021a](#page-24-19); Santos et al., [2022\)](#page-24-31). For instance, Sakulnarmrat and Konczak [\(2022\)](#page-24-28) incorporated ACNs from lamduan into gummy jellies after double-drum drying with diferent encapsulating agents. The combination of MD and GA (60:40) was selected as the best combination and applied to gummy jellies. The shelf-life stability of the gummy jellies was studied for eight weeks at diferent temperatures (25℃ and 35℃). The highest lamduan encapsulated powder (30 g/kg) added to gummy jellies showed the most extended shelf life and retention of ACNs at both temperatures after eight weeks of storage. These fndings corroborated the idea that encapsulating ACNs before application to food products retained and enhanced their stability and functionality. Other examples of ACNs sources, microencapsulation techniques, and their applications in various products are shown in Table [3.](#page-19-0)

Advantages and disadvantages of spray drying compared to other methods

Like any other method, spray drying has unique strengths and limitations compared to alternative techniques. It is quicker, more afordable, versatile, and appropriate for largescale production than freeze-drying, SC–CO2, coacervation,

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drum drying, and electrospraying microencapsulation. However, the potential deterioration of heat-sensitive substances and low encapsulation efficiency during the encapsulation process are noteworthy disadvantages of spray drying. However, the choice of method always depends on the specifc properties of the product being dried and the desired characteristics of the fnal dried product.

Conventional spray drying is the most popular and practically viable technique for microencapsulating ACNs and other bioactive components. High-quality powders with low moisture content, water activity, particle size, and morphology can be produced using spray drying. The selection of encapsulating agents may vary from case to case and mainly depends on the nature of the feed material/active agent to be microencapsulated. Various encapsulating agents are used on specifc fruits at specifc concentrations or ratios. It can be concluded that the best encapsulating agent is MD, with an inlet temperature of<180℃. This was due to MD characteristics with low viscosity, moisture, hygroscopicity, and good solubility. These can help reconstitute and entrap bioactive compounds from deterioration during processing. Thus, it can easily be used as a food ingredient. Although there are many emerging technologies for microencapsulation, spray drying is still widely practiced in most industries due to its low operation cost, high yield, speed, and efficiency.

Overall, spray drying is more efective than other techniques for microencapsulation due to its ability to encapsulate bioactive compounds rapidly and individually. However, ACNs extraction requires longer, and the solvents are expensive and risky. Moreover, it is recommended that appropriate encapsulating agents for specifc fruits be used to increase the stability of ACNs-rich powder, which is costly. Storage surroundings are also necessary for stabilizing the ACN compound; thus, suitable storage, such as vacuum-pack packaging, can be used to prevent oxidation. Therefore, it is recommended to add an antioxidant agent, such as tocopherol, into the fruit extract and encapsulating agents before the spray drying. Studies have focused on powdered products' physicochemical and nutritional characteristics derived from the spray-drying microencapsulation of ACNs-rich fruits. However, functional foods and nutraceutical qualities are frequently overlooked. Therefore, there is a need to analyze and explore the functional food and nutra-pharmaceutical prospects of ACNs-rich microencapsulated powdered products for specifc food applications. Moreover, it can be noted that diferent process variables for spray drying microencapsulation of ACNs-rich fruits have been optimized by the researchers using conventional methods. However, with new software development, modern tools such as RSM and computer simulation techniques are needed to allow rapid evaluation and optimization of the plans and design of the spray drying-based microencapsulation process. Using artifcial neural networking (ANN) to validate such designs can provide more value for optimizing such processes.

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Declarations

Conflict of interest The authors declare that they have no competing interests.

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