



# Antioxidant and antibacterial activities of a starch film with bioextracts microencapsulated from cactus fruits (*Opuntia oligacantha*)

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**Abstract** The use of unconventional sources is very relevant in the food area. In the present study the development of active films with the addition of bioextract (BE) or microencapsulated bioextract (MBE) from xoconostle (*Opuntia oligacantha*) on chayotextle starch was investigated. The film formulations were: 4 g of chayotextle starch, 2 g of glycerol and 180 g of water, three films with BE added (0.4, 0.8 and 1.2 g) and three films with MBE added (0.4, 0.8 and 1.2 g). Total phenols, total flavonoids, antioxidant activity (ABTS and DPPH), *Salmonella typhimurium* inhibition, color and mechanical properties of the films were analyzed. The film with 1.2 g of MBE showed high concentration of total phenols ( $54.12 \pm 0.77$  mg

EAG/100 g), total flavonoids ( $16.65 \pm 0.10$  mg QE/100 g) and antioxidant activity ( $29.11 \pm 0.48$  and  $41.42 \pm 1.81$  mg EAA for ABTS and DPPH respectively). The addition of bioextract from xoconostle is an option for the development of active films with antioxidant properties.

**Keywords** Active film · Chayotextle · Phenols · Flavonoids · Xoconostle

## Introduction

The development and constant improvement of packaging materials are very important to maintain food quality (Souza et al., 2013). Increasingly, great efforts are being made to develop biodegradable and environmentally-friendly materials from low-cost natural resources, such as polysaccharides and proteins. The design to edible films should provide improvements in the properties of the matrix and should consider other synergistic effects between components (Gao et al., 2017). The application of natural antioxidants in starch-based films to maintain food quality through antioxidant action is an innovative concept in the food industry (Reis et al., 2015).

Film based carbohydrates are particularly attractive because they have good film forming ability due to their unique colloidal properties (Ghasemlou et al., 2013). Starch is completely biodegradable, without toxic components, low in cost and widely available and (Luchese et al., 2017). The use of new starch alternatives for the development of new materials is of great interest for the industry, such as chayotextle starch (Palma-Rodríguez et al., 2018). Chayotextle has been isolated in central Mexico. This tuber contains an adequate amount of starch

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(60% dry basis), but the use of this tuber is limited to local consumers (Aila-Suárez et al., 2013).

The cactus pear “xoconostle” is cultivated in countries such as Mexico and have high commercial value due to are used for the manufacture of food products such juices, alcoholic beverages, jams, and natural liquid sweeteners (Kıvrak et al., 2018). It is a source of functional ingredients including phenolic compounds, betacyanins and vulgaxanthins (Hernández-Fuentes et al., 2015). The consumption of xoconostle, due to its content of polyphenols, can help improve human health; it may contribute to the prevention of chronic diseases and other problems in humans such as diabetes, obesity and respiratory illnesses (Morales et al., 2012).

The objective of this study was to develop edible films from an unconventional source of starch (chayotextle) with the incorporation of bioextract (BE) or microencapsulated bioextract (MBE) from *O. oligacantha* and characterize the antioxidant, antimicrobial and mechanical properties of the resulting films.

## Materials and methods

The xoconostle variety *Opuntia oligacantha* was used with commercial maturity and was obtained from Tezontepec de Aldama, Hidalgo, Mexico, in March 2016, a batch of 100 fruits morphologically homogeneous were used. The chayotextle (*Sechium edule Sw.*) was obtained in the municipal market of Tulancingo, Hidalgo, Mexico, in March 2016, and homogeneous tubers were used in weight and shape (300 g approximately each). The reagents used were as follows: methanol (HPLC grade), 2, 2'-azino-bis (3-ethylbenzothiazoline-6-sulphonic acid) (ABTS) from Sigma-Aldrich (St. Louis, MO, USA), potassium persulphate (analytical-reagent grade), Folin-Ciocalteu reagent (analytical-reagent grade) (Sigma-Aldrich), sodium carbonate (analytical-reagent grade) from J.T. Baker (USA), gallic acid (analytical-reagent grade) (Sigma-Aldrich), ascorbic acid (analytical-reagent grade) (Sigma-Aldrich), 1,1-diphenyl-2-picrylhydrazyl (DPPH) (Sigma-Aldrich), ethanol (HPLC grade), glycerol (analytical-reagent grade), acetone (gradient grade). Rutin (95%), ferulic acid (98%), vanillic acid (98%), quercetin (95%), hydroxybenzoic acid (99%), apigenin (98%), caffeic acid (98%) and kaempferol (98%) from Sigma-Aldrich (St. Louis, MO, USA). Maltodextrin with 10% dextrose from Grain Processing Corporation (Muscatine, IA, USA) and gum arabic from Frutarom (Winter Haven, FL, USA) were also used.

## Microencapsulation

Xoconostle fruits morphologically homogeneous and commercial maturity were manually selected and the whole fruit was mixed with water to a proportion 1:2 and homogenized in an industrial blender HGBSS (Waring Commercial, Torrington, CT, USA) for approximately 3 min. The product was vacuum filtered using Whatman paper number 1 to obtain cactus pear bioextract (BE), with 4% of total solids. After, it was mixed with two polymers used as wall materials: maltodextrin and gum arabic (biopolymers mixed between in a proportion 50-50) to obtain 30% total solids. The mixture was placed in a container and stirred for 24 h at room temperature to dissolve the gum; the mixture was then homogenized and dried on a Mini Spray Dryer (Büchi B-290, Switzerland) following the method developed by Martínez et al. (2015) with some modifications. The drying conditions were as follows: inlet temperature of 160 °C, 4 bar pressure and 10 mL/min inflow. The microencapsulated bioextract (MBE) of cactus pear was stored in sealed amber bags and refrigerated until use.

## Isolation of chayotextle starch

The starch was isolated following the methodology of Aila-Suárez et al. (2013). The tubers were cut into 2 × 2 cm cubes and macerated at a low speed (500 g of tuber with 500 g of water) for 2 min. The homogenate was washed and sieved to #50, #100, #200, #270 and #325 mesh (300, 150, 75, 53 and 45 µm, respectively) until the wastewater was observed to be transparent; then the homogenate was pelleted overnight and decanted. This material was dried in a convection oven at 35 °C overnight. The dried starch was ground in a blender to a powder; then, it was passed through a #100 mesh standard sieve and stored in a sealed container until use.

## Starch film

The films were made following the methodology of Aila-Suárez et al. (2013) with some modifications. First, 4 g of starch was mixed with 180 mL of water and 2 g of glycerol; this mixture was stirred at 40 °C for 3 min to achieve homogenization. Then, we proceeded to gelatinize the starch, raising the temperature to 90 °C and stirring continuously for 10 min. The BE or MBE was added when the filmogenic solution reached 50 °C (Table 1). After, the mixture was transferred to a plate coated with Teflon and subsequently dried in an oven for 24 h at 35 °C.

**Table 1** Formulation of different films from chayotextle starch with added bioextracts or microencapsulated bioextracts from cactus fruits (*Opuntia oligacantha*)

	Starch of chayotextle (g)	Glycerol (g)	Water (g)	BE (g)	MBE (g)
Starch film	4	2	180	–	–
0.4 BE	4	2	180	0.4	–
0.8 BE	4	2	180	0.8	–
1.2 BE	4	2	180	1.2	–
0.4 MBE	4	2	180	–	0.4
0.8 MBE	4	2	180	–	0.8
1.2 MBE	4	2	180	–	1.2

BE, bioextracts from cactus fruits; MBE, microencapsulated from bioextracts of cactus fruits

### Extraction of phenolic compounds from the films

Extraction was realized according to reported by Pérez-Alonso et al. (2015). First, 2 g of film was placed in a 100 mL beaker, and 20 mL of ethanol solution was added (50/50). Then, the mixture was stirred for 10 min, and the supernatant was centrifuged at  $10,000 \text{ min}^{-1}$  for 10 min at  $4^\circ\text{C}$  in a centrifuge Z 36 HK (HERMLE Labor Technik GmbH, Wehingen, Germany). The pellet obtained in the previous step was mixed with 20 mL of acetone (70/30) for 10 min and centrifuged at  $10,000 \text{ min}^{-1}$  for 10 min at  $4^\circ\text{C}$ . The ethanol and acetone supernatants were combined, stirred for 5 min and centrifuged under the conditions described above. The samples were refrigerated in the dark until analysis.

### Profile of phenolic compounds

The phenolic compounds were analyzed by capillary electrophoresis using a P/ACE<sup>TM</sup> MDQ instrument (Beckman Coulter, Indianapolis, USA) by the method described by Cao et al. (2004) with some modifications. The extract obtained in the previous section was mixed with 15 mL of HPLC grade methanol for 10 min in an ultrasonic bath. Then, the sample was centrifuged and filtered, and the supernatant was refrigerated. The temperature was kept at  $25^\circ\text{C}$  and the wavelength of UV detector was set at 214 nm. Pressure injection (30 mbar) was for 12 s. A 60.0 cm (distance between injector and detector 50.0 cm)  $\times$  75  $\mu\text{m}$  I. D. fused-silica capillary was used. A 5 min wash cycle with 0.1 M NaOH followed by 2 min distilled water, and 5 min running borate buffer pH 8.7 was necessary to condition the capillary. The running time was standardized at 20 min. For the results a curve for rutin, ferulic acid, vanillic acid, quercetin, hydroxybenzoic acid, apigenin, caffeic acid and kaempferol was elaborated with its respective standard, according to reported by Osorio-Esquivel et al. (2011) for phenolic compounds in *Opuntia jocosostle* fruits. Method validation was performed according to guidelines set by International Union of Pure and Applied Chemistry (IUPAC, 2002). The precision was

estimated by making five replicate injections of a standard mixture solution. The relative standard derivations (RSDs) peak areas are 4.17%, 4.21%, 4.19%, 4.85%, 4.42%, 4.11% and 4.65% for rutin, ferulic acid, vanillic acid, quercetin, 4-hydroxybenzoic acid, apigenin, caffeic acid and kaempferol, respectively.

### Total phenolic content

The phenolic content was assessed using the Folin–Ciocalteu method (Singleton et al., 1999) with some modifications. First, 2 g of film was mixed with 0.5 mL of Folin Ciocalteu (Sigma-Aldrich, USA). and 4.5 mL of water; after 5 min, the mixture was homogenized with 4 mL of 20% sodium carbonate. Then, the resulting mixture was measured at 765 nm after 2 h using Jenway 6715 UV–Vis spectrophotometer (USA). The total phenols are expressed in mg gallic acid equivalents GAE/100 g of film.

### Estimation of total flavonoids

Total flavonoid content was determined using the method described by Arvouet-Grand et al. (1994). A solution of aluminum trichloride ( $\text{AlCl}_3$ ) in 2% methanol was used. A total of 1 mL of the supernatant obtained in the extraction of the film was added to a 10-mL volumetric flask with methanol and stirred. Then, 2 mL of the sample mixture was added to 2 mL of  $\text{AlCl}_3$ . The samples were read at 415 nm in a spectrophotometer Jenway 6715 UV–Vis (USA). Total flavonoid content was expressed as mg quercetin equivalents QE/100 g of film.

### ABTS free radical scavenging activity

The antioxidant activity was determined using the chromogenic compound ABTS according to the protocol of Re et al. (1999) with some minor modifications. This method is based on studying the discoloration of the free radical ABTS by anti-oxidants. The ABTS (Sigma-Aldrich, USA) 7 mM was combined with potassium persulphate to generate the ABTS<sup>+</sup> radical (2.45 mM). Then, 980  $\mu\text{L}$  of

ABTS was mixed with 20  $\mu\text{L}$  of the supernatant obtained in the extraction of the film. In total, 2 readings per sample were performed; the first measurement was performed at 1 min, and the second measurement was performed at 6 min, both at a wavelength of 740 nm in a spectrophotometer Jenway 6715 UV–Vis (USA). The results were calculated as mg ascorbic acid equivalents of AAE/100 g of film.

#### DPPH free radical scavenging activity

The antioxidant activity of DPPH was studied according to the protocol previously described by Kuskoski et al. (2005) with some modifications. This method is based on studying the discoloration of DPPH by anti-oxidants. In total, 2.7 mL of DPPH (Sigma-Aldrich, USA) prepared in 80% methanol was added to 0.3 mL of the supernatant obtained in the extraction of the film. Two readings per sample were performed (at 1 min and 1 h) using a spectrophotometer Jenway 6715 UV–Vis (USA) at 515 nm. The results were calculated as mg ascorbic acid equivalents AAE/100 g of film.

#### Antibacterial activity

Effect against *Salmonella typhimurium* was determined according to descript to Seydim and Sarikus (2006) with some modifications. The zone of inhibition assay on solid media was used to determine the antimicrobial effects of the films. *S. typhimurium* (ATCC 43971) was activated using 1 mL of bacteria in 99 mL of nutritious broth until a concentration of approximately  $10^7$  CFU/mL (24 h, 37 °C) was obtained. The concentration of the bacteria was determined using the plate count method. Edible film discs were cut to 1 cm diameters using a sterile cutter and then placed in petri dishes containing *Salmonella* and *Shigella* agar. After, the petri dishes were inoculated with 100  $\mu\text{L}$  of bacteria and incubated at 37 °C for 24 h. Then, the area of the whole zone was calculated and then subtracted from the film disc area; this difference in area was reported as the zone of inhibition. Inhibition was monitored until the growth of bacteria occurred in the initial halo. The days of protection were determined by observing when the halo of inhibition disappeared.

#### Film color

The color of the films was determined following the methodology of Saberi et al. (2017), using a Minolta CM-508d colorimeter (Japan). Calibration was performed prior to the sample analysis using the color white. The measurements were taken directly on the glass surface. Three shots were performed for each measurement, and the

obtained parameters were  $L^*$   $a^*$   $b^*$  and total color change ( $\Delta E$ ).

#### Mechanical properties

The thickness of the films was evaluated using a manual micrometer (Mitutoyo Co., Kawasaki, Kanagawa, JP) at 5 random positions of the film, according to reported by Fakhouri et al. (2015). The average value was used to calculate the area of the cross-sections of the films, where the area is the width per each film thickness. The percentage elongation, tensile strength and Young's modulus of each film was determined. The mechanical properties were measured according to a standard method, (ASTM, 1995), using CT3 Brookfield texture analyzer with a 50 kg load cell (AMETEC, Harlow, Essex, UK). For the stress tests, the films were cut into rectangles 10 cm long and 1 cm wide. The rectangles were kept in a desiccator for 14 days with saturated sodium bromide (57% RH) solution at 25 °C 5 cm apart to maintain the constant moisture of films until the mechanical tests were performed. The ends of the films were fixed to clamps. The strain rate was 0.13 mm/s.

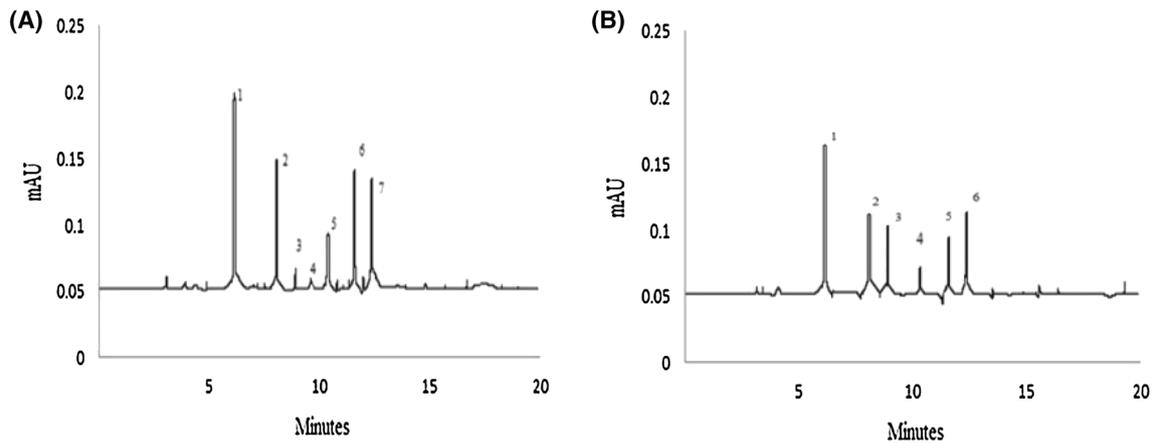
#### Statistical analysis

The experiments were arranged in a completely randomized design and were performed in triplicate. The results were processed by analysis of variance. Media comparison was conducted using Tukey's multiple test ( $\alpha = 0.05$ ) (Steel and Torrie, 1960).

## Results and discussion

#### Phenolic compounds

The yield obtained after spray drying was 30%, that is, for every 100 g of sample that was introduced to the dryer (bioextract and gums), 30 g of microencapsulation was obtained, which had a particle size of 14.47  $\mu\text{m}$ . The bioextract (BE) and microencapsulated bioextract (MBE) from *Opuntia oligacantha* showed rutin, quercetin, kaempferol, apigenin, caffeic acid and ferulic acid (Fig. 1), according to electropherograms, the compound that is in the highest concentration is rutin, while ferulic acid and kaempferol are well preserved once encapsulated. 4-hydroxybenzoic acid was not identified in BME (Fig. 1B), neither showed presence starch films. The films with MBE had the highest concentrations of phenolic compounds, this result suggest that the extract when encapsulated receives protection and is in an adequate form to act as soon as it is released. Rutin had the highest concentration



**Fig. 1** (A) Electropherogram of phenolic compounds of bioextract (BE) from *Opuntia oligacantha* at 214 nm. Identified peaks: (1) rutin, (2) ferulic acid, (3) quercetin, (4) 4-hydroxybenzoic acid, (5) apigenin, (6) caffeic acid, (7) kaempferol. (B) Electropherogram of

phenolic compounds of microencapsulated bioextract (MBE) from *Opuntia oligacantha* at 214 nm. Identified peaks: (1) rutin, (2) ferulic acid, (3) quercetin (4) apigenin, (5) caffeic acid, (6) kaempferol

( $0.97 \pm 0.12$  mg/g) in the film with 1.2 g of MBE (Table 2). Our results are similar to those of previous reports regarding 4-hydroxybenzoic acid, caffeic acid, rutin and quercetin (Cortez-García et al., 2015) in samples of *O. joconostle*, all of which are indicators of possible antioxidant or antimicrobial activity. Phenolic compounds identified in BE and MBE have been associated with some biological activity, antioxidant activity or antimicrobial activity among others, resulting successful this identification method. The results obtained in the present study reaffirm that xoconostle is an important source of phenolic compounds.

The total phenolic content showed significant differences between the films ( $p < 0.05$ ) (Table 3). The film with the addition of 1.2 g of BE had  $28.06 \pm 0.71$  mg GAE/100 g, while the film with the addition of 1.2 g of MBE had  $54.12 \pm 0.77$  mg GAE/100 g, a more than 1.5-fold increase. Cortez-García et al. (2015) found that the

total phenols decreased in *O. joconostle* when subjected to boiling. However, the microencapsulation preserved the phenolic compounds for application in food products, because the biopolymers of the wall of the microcapsules have a thermoprotector effect (Pérez-Alonso et al., 2015). The increase in total phenols in relation to the addition of BE and MBE are consistent with the results reported by Wang et al. (2013) in making chitosan films with different concentrations of polyphenols from tea. Phenolic compounds help protect plants against ultraviolet light, and act as defenses against pathogenic microorganisms (Osorio-Esquivel et al., 2011).

Significant differences between the films were found in terms of total flavonoids ( $p < 0.05$ ), showing the highest concentration in the film with 1.2 g of MBE ( $11.65 \pm 0.17$  mg QE/100 g) (Table 3). According to Reis et al. (2015) the antioxidant activity of flavonoids is completely related to their structure, acting as primary

**Table 2** Phenolic compounds identified in chayotextle starch films with added bioextracts or microencapsulated bioextracts from cactus fruits (*Opuntia oligacantha*)

Film	Rutine (mg/g)	Ferulic acid (mg/g)	Quercetin (mg/g)	Apigenin (mg/g)	Caffeic acid (mg/g)	Kaempferol (mg/g)	Total (mg/g)
Starch film	ND	ND	ND	ND	ND	ND	0.0
0.4 BE	$0.08 \pm 0.02^a$	$0.10 \pm 0.05^a$	ND	ND	ND	ND	0.18
0.8 BE	$0.31 \pm 0.09^c$	$0.18 \pm 0.08^b$	$0.07 \pm 0.04^{ab}$	$0.04 \pm 0.02^a$	$0.03 \pm 0.01^a$	$0.02 \pm 0.01^a$	0.65
1.2 BE	$0.46 \pm 0.08^c$	$0.24 \pm 0.01^c$	$0.10 \pm 0.02^b$	$0.06 \pm 0.03^a$	$0.06 \pm 0.02^{ab}$	$0.05 \pm 0.03^a$	0.97
0.4 MBE	$0.17 \pm 0.03^b$	$0.15 \pm 0.10^{ab}$	$0.03 \pm 0.02^a$	$0.05 \pm 0.03^a$	$0.05 \pm 0.02^{ab}$	$0.05 \pm 0.02^a$	0.5
0.8 MBE	$0.82 \pm 0.07^d$	$0.23 \pm 0.06^c$	$0.12 \pm 0.03^b$	$0.11 \pm 0.02^b$	$0.07 \pm 0.02^b$	$0.08 \pm 0.01^{ab}$	1.43
1.2 MBE	$0.97 \pm 0.12^d$	$0.36 \pm 0.09^d$	$0.22 \pm 0.10^c$	$0.15 \pm 0.04^b$	$0.16 \pm 0.10^c$	$0.10 \pm 0.01^b$	1.96

Results are expressed in mean  $\pm$  standard deviation

BE, bioextracts from cactus fruits; MBE, microencapsulated bioextracts from cactus fruits; ND, no detected

<sup>a,b,c,d</sup>Different letters in same column indicate differences between the films ( $p < 0.05$ ) using Tukey's test of comparison of means

**Table 3** Total phenols, total flavonoids, antioxidant activity (ABTS and DPPH) and inhibition of *Salmonella typhimurium* in chayotextle starch films with added of bioextracts or microencapsulated bioextracts from cactus fruits (*Opuntia oligacantha*)

Film	Total Phenols	Total flavonoids	ABTS	DPPH	mm of inhibition against <i>S. typhimurium</i>	Duration of halo of inhibition (days)
Starch film	1.80 ± 0.30 <sup>a</sup>	0.60 ± 0.10 <sup>a</sup>	2.10 ± 0.30 <sup>a</sup>	3.60 ± 0.20 <sup>a</sup>	–	–
0.4 BE	27.28 ± 0.95 <sup>b</sup>	10.72 ± 0.73 <sup>b</sup>	12.71 ± 0.44 <sup>c</sup>	14.20 ± 1.22 <sup>b</sup>	–	–
0.8 BE	28.52 ± 1.49 <sup>b</sup>	11.30 ± 0.20 <sup>b</sup>	15.30 ± 0.93 <sup>d</sup>	16.64 ± 1.42 <sup>bc</sup>	1	2
1.2 BE	28.06 ± 0.71 <sup>b</sup>	11.66 ± 0.22 <sup>b</sup>	15.71 ± 0.39 <sup>d</sup>	16.93 ± 3.39 <sup>bc</sup>	1	2
0.4 MBE	34.45 ± 0.36 <sup>c</sup>	10.42 ± 0.12 <sup>b</sup>	10.80 ± 1.17 <sup>b</sup>	20.99 ± 1.61 <sup>c</sup>	–	–
0.8 MBE	44.01 ± 0.62 <sup>d</sup>	13.76 ± 0.51 <sup>c</sup>	20.79 ± 0.39 <sup>e</sup>	32.49 ± 1.84 <sup>d</sup>	1	4.33
1.2 MBE	54.12 ± 0.77 <sup>e</sup>	16.65 ± 0.10 <sup>d</sup>	29.11 ± 0.48 <sup>f</sup>	41.42 ± 1.81 <sup>e</sup>	2	4.66

Results are expressed in mean ± standard deviation. Total phenols expressed (mg GAE/100 g of film), total flavonoids expressed (mg QE/100 g of film). ABTS and DPPH expressed (mg AAE/100 g film)

BE, bioextracts from cactus fruits; MBE, microencapsulated bioextracts from cactus fruits

<sup>a,b,c,d,e</sup>Different letters in same column indicate differences between the films \*( $p < 0.05$ ) using Tukey's test of comparison of means

antioxidants when donating a hydrogen atom and they can also act as chelating agents.

### Antioxidant activity

It is well known that the anti-oxidant activity of plant extracts is derived primarily from phenolic compounds, which are highly effective antioxidants and free radical scavengers (Wang et al., 2013). Osorio-Esquivel et al. (2011) mention that kaempferol and rutin are powerful antioxidants in the oxidation of LDL. The antioxidant activity measured for ABTS and DPPH were similar. The films showed significant differences ( $p < 0.05$ ) with the addition of BE or MBE (Table 3) against starch film. High antioxidant activity was observed for films treated with 1.2 g of MBE, 29.11 ± 0.48 mg EAA/100 g for ABTS, and 41.42 ± 1.81 mg EAA/100 g for DPPH. *O. oligacantha* has been demonstrated to have antioxidant activity due to the phenols and flavonoids it contains (Hernández-Fuentes et al., 2015). Moradi et al. (2012) added the phenolic compounds of grape seeds in a matrix based on chitosan and observed that the antioxidant properties of the matrix increased considerably.

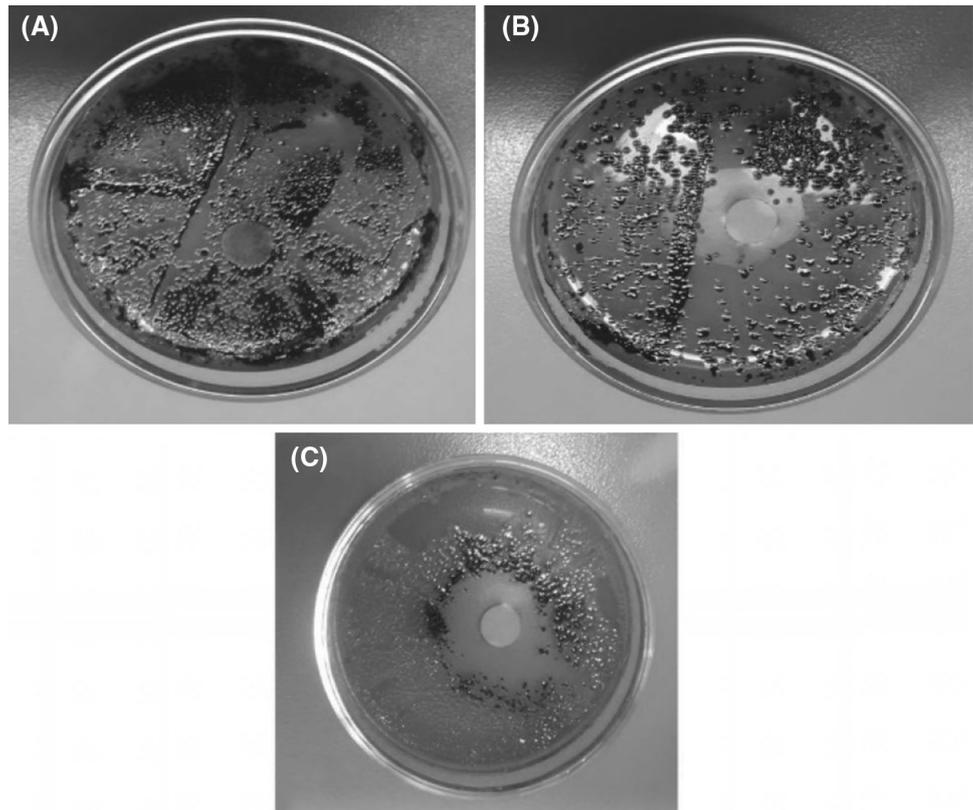
The films with added BE showed no significant differences ( $p > 0.05$ ) in DPPH (Table 3), perhaps due to the interaction between the phenols and polymers of the carbohydrates of the film. Saura-Calixto (2010) reported interactions between phenolic acids and polysaccharides. These interactions could be affecting the antioxidant activity of the phenolic acid so the increase of BE would not be reflected in increment of antioxidant activity, Corrales et al. (2009) mention that hydroxyl groups of phenolic compounds, mainly responsible for antioxidant properties, may interact with OH<sup>-</sup> groups from starch films resulting in the reduction of their properties. The films with 0.4, 0.8

and 1.2 g of MBE showed significant differences ( $p < 0.05$ ), which may be due to the interactions between amylose and maltodextrin/gum arabic allowing the phenolic compounds are free for provide a greater antioxidant capacity. The results of antioxidant activity in the added films indicate that the incorporation of phenolic compounds from natural sources are an effective alternative for their addition in complex matrices.

### Antibacterial activity

Figure 2 shows the zones of inhibition resulting in the films by the addition of bioextract, in the starch film no effect is observed, showing growth of the colonies even in the contact zone (a), in section b of the image can be observed that at the maximum concentration of BE the zone of inhibition is well defined, although few colonies managed to develop there, and section c shows a zone of inhibition that is greater and better defined with respect to the others. The films with MBE exhibited protective effects for a longer time than films with added BE. The films with 0.4 g of BE and 0.4 g of MBE showed no had inhibitory against *S. typhimurium* (Table 3). The maximum protection of a film against *S. typhimurium* was found for the film containing 1.2 g of MBE, which exhibited a halo of inhibition of 2 mm for 4.66 days, this due to the micro-encapsulates free the anti-bacterial compounds more slowly. Hayek and Ibrahim (2012) reported the anti-microbial effect of *O. matudae* against Escherichia coli O157: H7, showing halos of inhibition of up to 9.8 ± 1.01 mm. Further, Espinosa-Muñoz et al. (2017) reported that extracts from *O. oligacantha* contain phenols and flavonoids with inhibitory effects on *S. typhimurium* and *Staphylococcus aureus*. The inhibitory effect of BE and MBE is due of the phenolic compounds that contain, which attack cell walls and cell

**Fig. 2** Image of the inhibition halo of *Salmonella typhimurium*: (A) starch film; (B) film with addition of 1.2 g of bioextract; (C) film with addition of 1.2 g of microencapsulated bioextract



membranes, thereby affecting their permeability and the release of intracellular constituents causing the death of pathogenic bacteria (Bajpai et al., 2008). According to the results it found that the higher the antioxidant activity increased antimicrobial activity was observed, therefore, can be attributed to the antibacterial effect synergy of different phenolic compounds added to the films by BE and MBE, demonstrating that the bioactive compounds present in the xococonostle provide beneficial effects for use in the food industry.

### Film color

Table 4 shows the results obtained for the color properties, where  $L^*$ ,  $a^*$  and  $b^*$  presented significant differences ( $p < 0.05$ ) among starch films. Parameter  $L^*$ ,  $a^*$  and  $\Delta E$  increased with the addition of BE and MBE. The films with BE and MBE presented negative values for parameter  $b^*$ . The BE from *O. oligacantha* have phenols and flavonoids that confer color to the extract.

### Mechanical properties

Mechanical properties were also altered, depending on the levels of BE and MBE incorporated. Table 4 shows the significant ( $p < 0.05$ ) differences between the starch film, the film with 0.4 and 0.8 of BE added, the film with 1.2 of

BE added and the films with 0.4, 0.8 and 1.2 of MBE added in terms of thickness, which is related to breaking resistance. The films with 0.4 g and 0.8 g of BE were not significantly different ( $p < 0.05$ ) from the starch film (Table 4) in terms of tensile strength. The films with major phenol and flavonoid contents showed decreased tensile strength. Thus, the molecular properties of phenolic compounds affect the strength of the film matrix considerably, the interaction between the BE and MBE, and starch promotes the formation of discontinuities in the structure of the film. For example, the tensile strength of zein films reduces as the phenolic concentration in the films increases (Arcan and Yemenicioğlu, 2011). Similar results were observed by Moradi et al. (2012) with chitosan films incorporating grape seed extract.

The films with 1.2 g of BE and those with 0.4, 0.8 and 1.2 g of MBE showed significant differences ( $p < 0.05$ ) from the starch film in terms of the percentage elongation (Table 4). The lowest percentages of elongation may be attributed to the natural free sugars in *O. oligacantha*, which interact with the film network causing a percentage elongation less, as reported by Reis et al. (2015) in their study packaging films based on cassava starch with added yerba mate and mango pulp.

The films incorporating 1.2 g of BE and those with 0.8 and 1.2 g of MBE showed decreases ( $p < 0.05$ ) in the Young's modulus compared to the starch film. Arcan and

**Table 4** Thickness, mechanical properties (tensile strength, Young's modulus and elongation percentage) and color coordinates L\*, a\*, b\* and ΔE in chayotextile starch films with added bioextracts or microencapsulated bioextracts from cactus fruits (*Opuntia oligacantha*)

Film	Thickness (mm)	Tensile strength (MPa)	Percentage elongation (%)	Young's modulus (MPa)	Color		ΔE
					L*	a*	
Starch film	0.17 ± 0.03 <sup>ab</sup>	18.46 ± 2.22 <sup>d</sup>	0.89 ± 0.12 <sup>cd</sup>	0.92 ± 0.09 <sup>cd</sup>	83.89 ± 1.63 <sup>a</sup>	0.29 ± 0.09 <sup>a</sup>	0.29 ± 0.03 <sup>d</sup>
0.4 BE	0.17 ± 0.02 <sup>a</sup>	16.80 ± 2.80 <sup>cd</sup>	1.07 ± 0.20 <sup>d</sup>	1.01 ± 0.22 <sup>d</sup>	85.28 ± 1.67 <sup>b</sup>	0.17 ± 0.07 <sup>c</sup>	− 1.26 ± 0.17 <sup>b</sup>
0.8 BE	0.17 ± 0.01 <sup>a</sup>	15.95 ± 1.57 <sup>bcd</sup>	0.78 ± 0.17 <sup>bc</sup>	0.77 ± 0.16 <sup>bc</sup>	88.37 ± 1.60 <sup>c</sup>	0.67 ± 0.02 <sup>d</sup>	− 1.47 ± 0.15 <sup>b</sup>
1.2 BE	0.16 ± 0.01 <sup>b</sup>	14.91 ± 1.65 <sup>bc</sup>	0.67 ± 0.10 <sup>ab</sup>	0.66 ± 0.08 <sup>ab</sup>	91.95 ± 1.70 <sup>d</sup>	0.72 ± 0.17 <sup>d</sup>	− 2.52 ± 0.37 <sup>a</sup>
0.4 MBE	0.16 ± 0.02 <sup>b</sup>	13.6 ± 0.78 <sup>b</sup>	0.94 ± 0.13 <sup>cd</sup>	0.95 ± 0.14 <sup>cd</sup>	87.59 ± 3.83 <sup>c</sup>	0.07 ± 0.02 <sup>b</sup>	− 0.62 ± 0.05 <sup>c</sup>
0.8 MBE	0.16 ± 0.01 <sup>b</sup>	8.29 ± 1.17 <sup>a</sup>	0.47 ± 0.12 <sup>a</sup>	0.48 ± 0.13 <sup>a</sup>	93.43 ± 1.59 <sup>e</sup>	0.19 ± 0.06 <sup>c</sup>	− 0.80 ± 0.10 <sup>bc</sup>
1.2 MBE	0.16 ± 0.01 <sup>b</sup>	9.21 ± 1.65 <sup>a</sup>	0.51 ± 0.08 <sup>a</sup>	0.50 ± 0.10 <sup>a</sup>	94.34 ± 1.55 <sup>e</sup>	0.20 ± 0.06 <sup>c</sup>	− 0.82 ± 0.39 <sup>bc</sup>

Results are expressed in mean ± standard deviation

BE, bioextracts from cactus fruits; MBE, microencapsulated bioextracts from cactus fruits

<sup>a,b,c,d,e</sup>Different letters in same column indicate differences between the films \*( $p < 0.05$ ) using Tukey's test of comparison of means

Yemenicioğlu (2011) reported that the Young's modulus of zein films reduced as the phenolic concentration in the films increased, and the films in our study with high concentrations of phenols and flavonoids showed the same effect. The plant extracts rich in polyphenols affect the properties of starch in different ways. For example, they either increased or decreased setback and cool paste viscosity. The effect appears to be dependent on the type of starch, chemical composition of the extract or the structure of the specific phenolic compound (Zhu, 2015).

In Conclusion, it is possible to develop starch films added with bioextract of *O. oligacantha*. The addition of bioextract to films resulted in higher antioxidant activity, being higher which were added with BME. Only the films with 0.8 and 1.2 of BE and MBE had antibacterial properties against *S. typhimurium*. The films with MBE had higher functional properties than films with added BE. The films with 0.4 and 0.8 of BE retained all the mechanical properties of starch film. Thus, the addition of BE or MBE is an option for the development of active films with functional properties.

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**Compliance with ethical standards**

**Conflict of interest** The authors declare no conflict of interest.

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