

## Erratum to: Electrochemical and nanogravimetric studies of poly(copper phthalocyanine) microparticles immobilized on gold in aqueous solutions

Katalin Borsos<sup>1</sup> · György Inzelt<sup>1</sup>

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The original version of the article, unfortunately, contained a mistake in the caption of Fig. 2. The correct numbering of the curves on Fig. 2, i.e., the correct figure caption is as follows.

**Fig. 2** Cyclic voltammetric (a) and the simultaneously obtained EQCN frequency (b) responses for an Au | poly(CuPc) electrode in contact with electrolyte solutions of different pH values at constant 1 mol dm<sup>-3</sup> ionic strength: 0.001 mol dm<sup>-3</sup> perchloric acid+0.999 mol dm<sup>-3</sup> NaClO<sub>4</sub> (1), 0.01 mol dm<sup>-3</sup> perchloric acid+0.99 mol dm<sup>-3</sup> NaClO<sub>4</sub> (2), 0.1 mol dm<sup>-3</sup> perchloric acid+0.9 mol dm<sup>-3</sup> NaClO<sub>4</sub> (3), and 1 mol dm<sup>-3</sup> perchloric acid (4). Scan rate: 20 mV s<sup>-1</sup>

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The online version of the original article can be found at <http://dx.doi.org/10.1007/s10008-015-2770-6>.

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✉ György Inzelt  
inzeltgy@chem.elte.hu

<sup>1</sup> Department of Physical Chemistry, Institute of Chemistry, Eötvös Loránd University, Pázmány Péter sétány 1/A, 1117 Budapest, Hungary