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# **Branched‑chain alcohol formation by thermophilic bacteria within the genera of** *Thermoanaerobacter* **and** *Caldanaerobacter*

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**Abstract** Fifty-six thermophilic strains including members of *Caldanaerobacter*, *Caldicellulosiruptor*, *Caloramator*, *Clostridium*, *Thermoanaerobacter*, and *Thermoanaerobacterium*, were investigated for branched-chain amino acid degradation in the presence of thiosulfate in batch culture. All of the *Thermoanaerobacter* and *Caldanaerobacter* strains (24) degraded the branched-chain amino acids (leucine, isoleucine, and valine) to a mixture of their corresponding branched-chain fatty acids and branchedchain alcohols. Only one *Caloramator* strain degraded the branched-chain amino acids to the corresponding branched-chain fatty acids. The ratio of branched-chain fatty acid production over branched-chain alcohol production for *Thermoanaerobacter* was 7.15, 6.61, and 11.53 for leucine, isoleucine, and valine, respectively. These values for *Caldanaerobacter* were 3.49, 4.13, and 7.31, respectively. This indicates that members within *Caldanaerobacter* produce proportionally more of the alcohols as compared with *Thermoanaerobacter*. No species within other genera investigated produced branched-chain alcohols from branched-chain amino acids in the presence of thiosulfate.

**Keywords** Branched-chain amino acids · Thermophiles · Thermoanaerobes · Amino acid catabolism · Biofuel

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### **Introduction**

Thermophilic species within Class *Clostridia* from the genera of *Caldicellulosiruptor*, *Caldanaerobacter*, *Clostridium*, *Thermoanaerobacter*, and *Thermoanaerobacterium* have been intensively investigated in the context of biofuel production due to their broad substrate spectrum, especially among the sugars present in lignocellulosic biomass, and because of their high ethanol and hydrogen yields (Taylor et al. [2009](#page-9-0); Ren et al. [2009](#page-9-1); Chang and Yao [2011](#page-8-0); Scully and Orlygsson [2015\)](#page-9-2). The protein and amino acid catabolism of thermophilic *Clostridia* has received considerably less attention likely due to the abundance of lignocellulosic biomass. However, several investigations have been performed mainly focusing upon the thermodynamics of the catabolism and end-product formation from specific amino acids (Elsden and Hilton [1978;](#page-8-1) Barker [1981;](#page-8-2) McInerney [1988](#page-9-3); Orlygsson et al. [1994](#page-9-4); [1995](#page-9-5); Faudon et al. [1995](#page-9-6); Fardeau et al. [1997\)](#page-8-3). Thermoanaerobes can catabolize amino acids by oxidative and/or reductive deamination and decarboxylation (Elsden and Hilton [1978](#page-8-1); Andreesen et al. [1989;](#page-8-4) Orlygsson et al. [1995](#page-9-5)). Thus, the oxidative fermentation of amino acids yields the corresponding  $\alpha$ -keto acid in the first step which is thereafter decarboxylated to one carbon shorter volatile fatty acid. For instance, alanine is deaminated to pyruvate, which is then decarboxylated to acetate.

The amino acids that undergo oxidative deamination usually have reduced oxidation states, including the branched-chain amino acids (BCAAs), alanine, and glutamate (Andreesen et al. [1989;](#page-8-4) Fardeau et al. [1997\)](#page-8-3). These amino acids can only be fermented if the electrons produced from the initial deamination step are scavenged due to the unfavorable thermodynamics involved (Orlygsson

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et al. [1995](#page-9-5); Fardeau et al. [1997](#page-8-3)). For instance, the ∆*G*°′ for the degradation of the BCAAs is between  $+4.2$  and +9.7 kJ/mol (Fardeau et al. [1997\)](#page-8-3). The addition of thiosulfate (which can be reduced to  $H_2S$  or  $S^0$ ) to the medium or co-cultivating the amino acid degrader with hydrogen scavenging bacteria, such as a methanogen or sulfate reducer, drives the reaction forward resulting in the amino acids being degraded to their corresponding branched-chain fatty acids (BCFA) in addition to  $CO<sub>2</sub>$  and ammonium (Fardeau et al. [1997\)](#page-8-3).

The degradation of BCAAs to their corresponding BCFAs by *Thermoanaerobacter brockii* has been investigated in some detail (Fardeau et al. [1997\)](#page-8-3). Under hydrogen scavenging conditions, leucine, isoleucine, and valine are degraded to 3-methylbutyrate, 2-methylbutyrate, and 2-methylpropionate, respectively. A recent investigation in our laboratory demonstrated that *T. brockii* (DSM 1457) and *Caldanaerobacter subterraneus* subsp*. yonseiensis* (DSM 13777) degrade BCAAs to a mixture of the corresponding BCFAs and branched chain alcohols (BCOHs) when cultivated in the presence of thiosulfate but only to the corresponding BCFAs when co-cultured with a hydrogenotrophic methanogen (Scully and Orlygsson [2014\)](#page-9-7).

The majority of studies on the catabolism of the BCAAs have focused on aerobic bacteria such as species of *Staphylococcus* and *Enterococcus* (Beck et al. [2004](#page-8-5); Ward et al. [1999](#page-9-8)), aerotolerant anaerobes including *Lactobacillus sakei* (Gutsche et al. [2012\)](#page-9-9), or yeasts that use the so-called Ehrlich pathway (Ehrlich [1907;](#page-8-6) Hazelwood et al. [2008\)](#page-9-10). These studies have often focused on the formation of compounds that contribute to the flavor profile of foods and beverages (branched-chain and aromatic aldehydes, alcohols, and acids) (Smit et al. [2005,](#page-9-11) [2009\)](#page-9-12). Recently, some studies have focused on the production of branched-chain alcohols from protein-rich waste using genetically engineered *Escherichia coli* and *Bacillus subtilis* with the main focus being that BCOHs are promising biofuel candidates (Huo et al. [2011](#page-9-13); Choi et al. [2014\)](#page-8-7). Additionally, BCOHs can be used as building blocks for the production of higher chemicals (Sakuragi et al. [2011](#page-9-14)).

In this study, the production of BCOHs from the BCAAs was investigated within various genera of thermophilic bacteria to gain a broader knowledge whether this phenomenon is genus specific within the genera of *Thermoanaerobacter* and *Caldanaerobacter*. Thus, the present investigation examines the ability of thermophilic anaerobes from several genera to produce BCOHs from BCAAs. Fortyeight (48) thermophilic anaerobic bacteria within Class *Clostridia* isolated from various hot springs in Iceland were investigated. Additionally, eight type strains from other thermophilic genera within Class *Clostridia* were used for comparison.

### **Materials and methods**

### **Medium and cultivation**

The BM medium consisted of (per liter):  $NaH<sub>2</sub>PO<sub>4</sub> 2.34 g$ ,  $Na<sub>2</sub>HPO<sub>4</sub>$  3.33 g, NH<sub>4</sub>Cl 2.2 g, NaCl 3.0 g, CaCl<sub>2</sub> 8.8 g, MgCl<sub>2</sub>·6H<sub>2</sub>O 0.8 g, yeast extract 2.0 g, resazurin 1 mg, trace element solution 1 ml, vitamin solution (DSM141) 1 ml, and NaHCO<sub>3</sub> 0.8 g. The trace element solution consisted of the following on a per liter basis:  $FeCl<sub>2</sub>·4H<sub>2</sub>O$ 2.0 g, EDTA 0.5 g, CuCl<sub>2</sub> 0.03 g, H<sub>3</sub>BO<sub>3</sub>, ZnCl<sub>2</sub>,  $MnCl_2·4H_2O$ ,  $(NH_4)Mo_7O_{24}$ ,  $AlCl_3$ ,  $CoCl_2·6H_2O$ ,  $NiCl_2$ , and  $0.05$  g, Na<sub>2</sub>S $\cdot$ 9H<sub>2</sub>O 0.3 g, and 1 mL of concentrated HCl. Carbon and energy sources were 20 mM. The medium was prepared by adding the buffer to distilled water containing resazurin and then boiled for 10 min and cooled under nitrogen flushing. The mixture was then transferred to serum bottles using the Hungate technique (Hungate [1969](#page-9-15)) and autoclaved for 60 min. All other components of the medium were added separately through filter  $(0.45 \mu m)$ sterilized solutions. All experiments were conducted at 65 °C and at pH 7.0. In all cases, experiments were performed in duplicate.

#### **Bacterial strains**

Forty-eight (48) strains from our culture collection (hereafter called AK strains) used in this investigation were isolated as previously described on various carbon sources (Orlygsson and Baldursson [2007;](#page-9-16) Orlygsson et al. [2010](#page-9-17)). Eleven strains belong to the genus *Thermoanaerobacter*, 9 to *Caldanaerobacter*, 14 to *Thermoanaerobacterium*, 7 to *Clostridium*, 4 to *Caloramator* and 3 to *Caldicellulosiruptor*. Additionally, the following strains were obtained from Deutsche Sammlung von Mikroorganismen und Zellkulturen (DSMZ): *Thermoanaerobacter wiegelii* (DSM 10319), *Thermoanaerobacter thermohydrosulfuricus* (DSM 567), *Thermoanaerobacter ethanolicus* (DSM 2246), *Caldanaerobacter subterraneus* subspecies *subterraneus* (DSM 13054), *Caldicellulosiruptor saccharolyticum* (DSM 8903), *Caldicellulosiruptor owensis* (DSM 13100), *Thermoanaerobacterium saccharolyticum* (DSM 571), and *Caloramator viterbiensis* (DSM 13723).

# **Screening for branched‑chain fatty acid and branched‑chain alcohol formation**

All strains (AK and DSM) were screened for branchedchain amino acid degradation. The strains were inoculated from frozen (−20 °C) cultures stored in 30 % (v/v) glycerol and reactivated on BM medium containing glucose (20 mM). Reactivated cultures were inoculated (2  $\%$  v/v)

from exponential growth phase to 25 mL serum bottles (liquid–gas ratio was 1:1) containing a mixture of leucine, isoleucine, and valine (9 mM each) in the presence of thiosulfate (20 mM). Cultures were grown for 5 days and thereafter 1 mL of liquid and 0.2 mL of gases were removed for analysis of end-products.

# **Production of branched‑chain fatty acids and branched‑chain alcohols from individual branched‑chain amino acids in the presence of thiosulfate**

All strains showing positive growth on BCAAs in the presence of thiosulfate from the initial screening experiment (mixture of BCAAs) were cultivated in 25 mL serum bottles (L–G ratio 1:1) on 20 mM of individual BCAAs (leucine, isoleucine, and valine) containing 20 mM of thiosulfate.

#### **Analytical methods**

Hydrogen was analyzed using a Perkin Elmer gas chromatograph equipped with a thermo conductivity detector. Nitrogen was used as carrier gas at a rate of 3 ml/min, with another 10 ml/min as make-up gas in the detectors. The separation was performed on a Supelco 1010 Carboxen CC Plot Capillary Column. The oven temperature was 80 °C and the injector and detector temperatures were 200 °C. Volatile fatty acids and alcohols were analyzed by gas chromatograph (Perkin Elmer Clarus 580) using a FID detector with 30 m DB-FFAP capillary column (Agilent Industries Inc, Palo Alto, CA, US). Samples (1 mL) were centrifuged for 20 min at 6000*g*. The supernatants were acidified with 25 % formic acid, and crotonic acid was used as the internal standard. Thiosulfate was analyzed by the method of Westley ([1987\)](#page-9-18) with modifications as described by Scully and Orlygsson [\(2014](#page-9-7)); the background interference was measured by adding 10 µL of 0.25 M KCN and 10 µL of  $dH<sub>2</sub>O$  to 180 µL of sample; after mixing at 1000 rpm for 20 s, 100 µL of Sorbo's ferric nitrate reagent (10 % w/v Fe(NO<sub>3</sub>)<sub>3</sub>.9H<sub>2</sub>O and 20 % v/v HNO<sub>3</sub>) was added. Samples were allowed to clarify and the absorbance read at 460 nm against a water blank. The analyte was then determined by repeating the procedure and substituting 10 µL of  $0.20$  M CuSO<sub>4</sub> for dH<sub>2</sub>O. Thiosulfate concentrations were calculated as the background absorbance–analyte against a standard curve generated using thiosulfate concentrations between 0.1 and 10 mM. Amino acids were analyzed using the ninhydrin method by adding 100 µL of a sample and 100 µL of 1 % (w/v) ninhydrin reagent (60 % v/v 2-propanol and 40 mM acetate buffer, pH 5.5) in a microtiter plate and incubated at 100 °C for 20 min; relevant amino acids (leucine, isoleucine, and valine) were used as

standards. After cooling to ambient temperature, 200 µL of 50 % (v/v) 2-propanol was added and the absorbance read at 580 nm on a Bioscreen C (Oy Growth Curves Ab, Finland). Hydrogen sulfide was quantified as described by Cord-Ruwisch [\(1985](#page-8-8)). Briefly, the pH of the medium was adjusted to 10 with the addition of 6 M NaOH; samples were then allowed to stand for 60 min with periodic mixing. 50 µL of sample was then transferred into a cuvette containing 1.95 mL of 5 mM  $CuSO<sub>4</sub>$  in 50 mM HCl under rapid stirring. After 5 s of stirring, the cuvette was transferred to a Perkin Elmer Lambda 25 UV–Vis spectrophotometer and the absorbance read at a wavelength of 480 nm against a dH<sub>2</sub>O blank. Other sulfur species were not analyzed in the present investigation.

### **Molecular methods**

DNA was extracted from the strains and used as templates in 16S rRNA PCR reactions. 16S rRNA genes were amplified with primers F9 and R1544, which are specific for bacterial genes (Skirnisdottir et al. [2000](#page-9-19)). The PCR products were sequenced with the universal 16S rRNA primers F9, F515, F1392, R357, R805, R1195, and R1544 (Skirnisdottir et al. [2000](#page-9-19)) using a Big Dye Terminator Cycle Sequencing Ready Reaction Kit (Applied Biosystems). Subsequently, the DNA was analyzed with a 3730 DNA analyzer from Applied Biosystems. The nucleotide sequence was displayed and analyzed with Sequencer (Gene Code Corporation). Sequences from 16S rRNA analysis were submitted to the NCBI database (National Center for Biotechnology Information) using the nucleotide–nucleotide BLAST tool (Altschul et al. [1997](#page-8-9)). The most similar sequences obtained were aligned with sequencing results in MEGA6 (Tamura et al. [2013](#page-9-20)) and the maximum likelihood method based on the Tamura–Nei model was used to generate a phylogenetic tree (Tamura and Nei [1993](#page-9-21)). Evolutionary analyses were conducted in MEGA6 (Tamura et al. [2013](#page-9-20)).

# **Results**

# **Screening of the culture collection for branched‑chain amino acid degradation**

Fifty-six strains of thermophilic bacteria, either isolated from various hot springs in Iceland (48) or obtained from culture collections (8) were investigated in present study. The origin of AK strains showing temperature and pH at site of isolation, together with end product formation from glucose is showed in Supplementary Table 1. Additionally, substrate spectra of strains are presented. All of the *Thermoanaerobacter* (14) and *Caldanaerobacter* (10) strains were positive for BCFA and BCOH formation (Table [1](#page-3-0)). Only

<span id="page-3-0"></span>**Table 1** Relative production of BCOH and BCFA from BCAA by positive strains. All strains were cultivated in the presence of a mixture of leucine (9 mM), isoleucine (9 mM), and valine (9 mM) in the presence of thiosulfate (20 mM). Closest similarity of strains from

partial 16S rRNA is shown within *brackets* as well as the length of the 16S rRNA used for phylogenetic analysis. NCBI accession numbers for all AK strains are also shown



For the BCOH formation:  $- = <0.1$  mM,  $+ = 0.1$ –1.0 mM,  $++ = 1.0$ –3.0 mM,  $++ + = 3.0$ –5.0 mM;  $++ + + = >5.0$  mM For BCFA formation:  $+ = 0.1 - 5.0$  mM;  $++ = 5.1 - 13.0$  mM,  $++ + = 13.1 - 20.0$  mM,  $++ + + = 220.0$  mM

one other strain was positive for BCFA formation, *Caloramator* strain AK49, but it did not produce BCOHs. Other strains from the genera *Clostridium*, *Caldicellulosiruptor* and *Thermoanaerobacterium* did not degrade the BCAAs and thus did not produce BCFAs or BCOHs (are thus not included in Table [1](#page-3-0)). The relative amount of BCFAs and BCOHs produced by the 25 strains that were positive on BCAA degradation is shown in Table [1](#page-3-0) together with phylogenetic data showing the closest relative of each strain. In general, the highest amounts of BCFAs (>20 mM of all three fatty acids) and BCOHs (>3 mM of all three alcohols) were produced by the genera of *Caldanaerobacter*. Most of the *Thermoanaerobacter* species produced between 5 and 13 mM of the BCFA and between 1 and 3 mM of the BCOH. The best alcohol producer, *Caldanaerobacter* strain AK72, most closely related to *C. subterraneus* subsp*. yonseiensis*, produced  $24.5 \pm 1.3$  mM of the BCFAs (sum of 3-methylbutyrate, 2-methylbutyrate, and 2-methylpropionate) and  $5.7 \pm 0.3$  mM of BCOH (summary of 3-methylbutanol, 2-methylbutanol, and 2-methylpropanol).

# **Phylogenetic relationship of selected** *Caldanaerobacter* **and** *Thermoanaerobacter* **species**

Figure [1](#page-4-0) shows the phylogenetic relationship between all *Thermoanaerobacter* and *Caldanaerobacter* species with standing in nomenclature as well as the twenty AK strains belonging to these two genera. Eleven (11) of the AK strains belong to the genus *Thermoanaerobacter* and nine (9) belong to *Caldanaerobacter*. Most of the *Thermoanaerobacter* AK strains were most closely related to *T. thermohydrosulfuricus* and *T. ethanolicus* (8 strains) and *T. uzonensis* (3 strains). Six of the *Caldanaerobacter* strains were most closely related to *Caldanaerobacter*



<span id="page-4-0"></span>**Fig. 1** Maximum likelihood phylogenetic dendrogram based on 16S rRNA gene sequences showing the position of BCOH-forming thermophilic strains within the class *Clostridia*. Bootstrap values (>50 %) based on 1000 replicates are shown at branch nodes. *Bar* represents 1 substitution per 100 nucleotides

*subterraneus* subsp. *subterraneus* and three to *Caldanaerobacter subterraneus* subsp. *yonseiensis*. It is noteworthy that *Caldanaerobacter* AK113 is phylogenetically distant from its nearest neighbor *Caldanaerobacter subterraneus* subsp. *yonseiensis*. The reason could be that this strain has a relatively short 16S sequenced (406 nt). Table [1](#page-3-0) shows the most closely related species assigned for each AK strain together with accession numbers obtained from NCBI.

### **End product formation from individual branched‑chain amino acids**

All strains positive for growth on a mixture of BCAAs were tested on for growth and end product formation in the presence of thiosulfate on individual BCAAs. Data showing end product formation from these experiments are shown for *Thermoanaerobacter* and *Caldanaerobacter* species in Supplementary Tables 2 and 3, respectively. In all cases, thiosulfate was nearly completely utilized (data not shown). All strains were positive for BCAA degradation although some variations in amounts of end products were observed. For simplicity, data is presented for three representative *Thermoanaerobacter* strains (*T. wiegelii*, AK15, and AK68; Fig. [2](#page-5-0)a–c) and three *Caldanaerobacter* strains (*C. subterraneus* subsp. *subterraneus*, AK102, and AK112; Fig. [3](#page-6-0)a–c).

*Thermoanaerobacter wiegelii* (DSM 10319) produced between 12.0 and 14.0 mM of the BCFAs and 1.8 to 3.0 mM of the BCOHs (Fig. [2a](#page-5-0)). Hydrogen concentrations were from 5.3 to 7.8 mmol/L for the three amino acids and hydrogen sulfide from 7.8 to 12.6 mmol/L. In all cases, more than 90 % of the BCAAs were degraded and carbon balances for leucine, isoleucine, and valine were 80.5, 88.5, and 84.0 %, respectively (Fig. [2](#page-5-0)a; Supplementary Table 2). *Thermoanaerobacter* strain AK 68 (most closely related to *T. uzonensis*) almost completely degraded all three amino acids and produced similar amounts of BCFA (12.2 to 13.0 mM) and BCOH (2.3 to 4.0 mM) (Fig. [2](#page-5-0)b). Carbon balances ranged from 83.3 to 93.8 % (Supplementary Table 2). Hydrogen concentrations were very low or between 0.2 and 0.6 mmol/L and more than 11 mmol/L of H2S were produced in all three cases. *Thermoanaerobacter* strain AK15 (most closely related to *T. uzonensis*) also almost completely degraded all three amino acids with carbon recoveries ranging from 82.0 % (isoleucine) to 88.4 % (leucine) as summarized in Supplementary Table 2. Similar amounts of BCFAs and BCOHs were observed as for *T. wiegelii*, except for BCOH formation from isoleucine which was considerable lower (Fig. [2](#page-5-0)c). Hydrogen was between  $0.8$  and  $1.8$  mmol/L and  $H<sub>2</sub>S$  produced between 11.2 and 12.1 mmol/L.

*Caldanaerobacter subterraneus* subsp. *subterraneus* (DSM 13054) showed similar fermentation spectrum on all three amino acids (Fig. [3a](#page-6-0)); with the production of BCFAs



<span id="page-5-0"></span>**Fig. 2** Production profile from BCAA by **a** *Thermoanaerobacter wiegelii* (DSM 10319) of 20 mM thiosulfate. **b** *Thermoanaerobacter* strain AK68, and **c** *Thermoanaerobacter* strain AK15

between 14.0 and 15.6mM and the BCOHs formation between 3.4 and 5.7 mM with carbon recoveries between 91.1 to 99.0 % (Supplementary Table 3). Very low concentrations of both H<sub>2</sub> (<1 mmol/L) and H<sub>2</sub>S (<1.3 mmol/L) were observed. Figure [3b](#page-6-0) shows end product formation for *Caldanaerobacter* strain AK102 (most closely related to *Caldanaerobacter subterraneus* subsp. *subterraneus*). This strain produces lower amounts of both BCFAs and BCOHs as compared to *Caldanaerobacter subterraneus* subsp. *subterraneus* (DSM 13054) but between 5.7 and 11.0 mM of the amino acids was not degraded by this strain but carbon balances were between 85.5 and 88.5 % (Supplementary Table 3). Hydrogen concentrations were between 4.1 and 8.4 mmol/L and  $H_2S$  concentrations were between 5.8 and 8.0 mmol/L. *Caldanaerobacter* strain AK112 is most



<span id="page-6-0"></span>**Fig. 3** Production profile from BCAA by **a** *Caldanaerobacter subterraneus* subsp. *subterraneus* in presence the of 20 mM thiosulfate. **b** strain *Caldanaerobacter* strain AK102, and **c** *Caldanaerobacter* strain AK112

closely related to *C. subterraneus* subsp. *yonseiensis* produced between 11.7 and 12.9 mM and between 1.4 and 4.0 of the BCFA and BCOHs, respectively (Fig. [3](#page-6-0)C) and carbon balances were between 82.8 and 89.3 (Supplementary Table 3). Hydrogen and  $H<sub>2</sub>S$  concentrations ranged from 9.6 to 15.4 mmol/L and 7.5 to 10.2 mmol/L, respectively.

Statistical tests were done to reveal the difference between the ratio of BCFA production versus the BCOH production for all alcohol forming species of *Thermoanaerobacter* and *Caldanaerobacter.* The difference proved statistically significant. The data points used for the test are presented in Supplementary Tables 2 and 3. A preliminary test for equality of variances for the ratio of 3-methylbutanol and 3-methylbutanol (products from leucine degradation) indicates that there is a clear difference between the strains within the two genera  $(F = 4.11, p = 0.015)$ . Therefore, a two-sample *t* test (95 % confidence interval) was performed that does not assume equal variances. The mean score for *Thermoanaerobacter* ( $M = 7.15$ ,  $SD = 2.31$ ,  $N = 14$ ) was significantly higher than the scores for *Caldanaerobacter*  $(M = 3.28, SD = 1.14, N = 11)$ . Similarly, *t* test showed a clear difference between the ratio of 2-methylbutyrate and 2-methylbutanol (from isoleucine) and the ratio of 2-methylpropionate and 2-methylpropanol (from valine).

### **Discussion**

The present investigation examines the distribution of BCAA metabolism among selected thermophilic *Clostridia* with the main emphasis of the genera *Thermoanaerobacter* and *Caldanaerobacter*. The degradation of the BCAAs to their corresponding BCFAs been shown to be possible by *Thermoanaerobacter* species when grown under electron scavenging systems (Fardeau et al. [1997;](#page-8-3) Scully and Orlygsson [2014](#page-9-7)). Although bacteria have the enzymes responsible for the degradation of these amino acids, hydrogen accumulation from initial deamination, and decarboxylation steps inhibits further degradation. By adding thiosulfate to the cultures, *Thermoanaerobacter* and *Caldanaerobacter* species can use it as an electron acceptor producing elemental S,  $H_2S$ , or possibly other sulfur species (Lee et al. [1993](#page-9-22); Cann et al. [2001](#page-8-10); Kozianowski et al. [1997\)](#page-9-23). Other means of scavenging electrons is by cocultivating the amino acid degraders either with hydrogenotrophic methanogens (Fardeau et al. [1997;](#page-8-3) Scully and Orlygsson [2014](#page-9-7)) and sulfate reducers (Stams and Hansen [1984](#page-9-24)).

A recent investigation in our laboratory demonstrated that *Thermoanarobacter brockii and Caldanaerobacter subterraneus* subsp. *yonseiensis* produce a mixture of BCFAs and BCOHs from BCAAs in the presence of thiosulfate (Scully and Orlygsson [2014\)](#page-9-7). Thus, these strains produced a mixture of 3-methylbutyrate and 3-methylbutanol from leucine, 2-methylbutyrate and 2-methylbutanol from isoleucine and 2-methyl-propionate and 2-methylpropanol from valine. No BCOH formation occurred when the strains were co-cultivated with the hydrogenotrophic methanogen *Methanothermobacter* strain M39. The present investigation focused on other well-known species within these two genera as well as with other species within other genera to see if this ability of BCOH was genus specific or not.

### **Phylogenetic considerations**

Recently, two species within *Thermoanaerobacter* were moved to a new genus, *Caldanaerobacter*. Thus, currently, the *Thermoanaerobacter* genus has 15 species and

five sub-species (Euzéby [1997](#page-8-11); Parte [2014\)](#page-9-25) and *Caldanaerobacter* (Fardeau et al. [2004\)](#page-8-12) has two species and 4 sub-species. The phylogenetic tree of most of the type species within *Thermoanaerobacter* and *Caldanaerobacter* together with all AK strains positive for BCAA degradation is presented in Fig. [1](#page-4-0). Many of the *Thermoanaerobacter* AK strains were closely related with *T. thermohydrosulfuricus*, *T. ethanolicus*, and *T. pseudoethanolicus*. Comparing data from Table [1](#page-3-0) with phylogenetic relationship of good BCOH producers the only four strains producing more than 20 mM of BCFA and more than 5 mM of BCOH belong to the genus *Caldanaerobacter*. Additionally, only three strains that produced less than 1 mM of BCOH all belong to the genus *Thermoanaerobacter*. Thus, in general species within *Caldanaerobacter* produce higher concentrations of the alcohol as compared to the *Thermoanaerobacter* species investigated. This was further analyzed statistically with a *t* test in present study. Interestingly, there was also a clear difference between the ratio of BCFAs over BCOHs from different amino acids degraded. Thus, from leucine and isoleucine, this ratio was between 7.15  $\pm$  2.31 and 6.61  $\pm$  2.77 for *Thermoanaerobacter*, but 3.49  $\pm$  1.18 to 4.13 ± 1.30 for *Caldanaerobacter*. This value for valine was, however, much higher (11.53  $\pm$  4.65 and 7.31  $\pm$  1.74 for *Thermoanaerobacter* and *Caldanaerobacter*, respectively) indicating proportionally more alcohol formation from the 6 carbon amino acids (leucine and isoleucine) as compared to valine (5 carbons). This could be due to the enzyme specificity of the different enzymes used for the degradation of the BCAAs and needs further investigation.

#### **Degradation on single branched‑chain amino acids**

A great variation was observed on end product formation between strains using single BCAAs as substrates with 20 mM of thiosulfate (Supplementary Tables 2 and 3). Selected data are presented for only six strains to emphasize this variation (Figs. [2](#page-5-0), [3\)](#page-6-0). For instance, *Caldanaerobacter subterraneus* subsp. *subterraneus* showed good yields in BCFAs and BCOHs (between 91.1 and 99.0 in carbon balances) and most of the hydrogen was scavenged by thiosulfate reduction. However, the balances for sulfur species are incomplete since very little  $H_2S$  was detected at the end of cultivation. Lee and co-workers stated that *Thermoanaerobacter* species reduce thiosulfate to hydrogen sulfide, whereas *Thermoanaerobacterium* reduce it to elementary sulfur (Lee et al. [1993\)](#page-9-22). Later, it became clear that this is not a unified characteristic of these genera when two species within *Thermoanaerobacterium* were shown to be unable to reduce thiosulfate to sulfur (Cann et al. [2001\)](#page-8-10). Additionally, it is known that *Thermoanaerobacter italicus* produces a mixture of sulfur and hydrogen

sulfide from thiosulfate (Kozianowski et al. [1997](#page-9-23)) while *Thermoanaerobacter sulfurigignens* produces only sulfur (Lee et al. [2007](#page-9-26)). In all cases, only trace amounts of thiosulfate were detected after end of cultivation in present study suggesting that all thiosulfate was reduced to other sulfur species. Sulfur was not analyzed but a strong yellow color and microscopic observation showed the presence of sulfur in some of the culture bottles, especially when  $H_2S$ was found in low concentrations. Theoretically, 20 mM of a BCAA should yield 40 mmol/L of hydrogen and if thiosulfate (20 mM) is the only electron acceptor it should produce 20 mmol/L of hydrogen sulfide. These yields are in good correlation for most of the strains tested; usually the  $H<sub>2</sub>S$  was little lower as compared with the amount of BCFA produced (Figs. [2](#page-5-0)a–c, [3](#page-6-0)b, c; Supplementary Tables 2 and 3). However, in some cases this was not the case and very low H<sub>2</sub>S concentrations were indeed observed. Clearly, the different species use different sulfur metabolism for scavenging the electrons and this can also been seen by different hydrogen concentrations observed. Earlier data from *Caldanaerobacter subterraneus* subsp. *yonseiensis* and *Thermoanarobacter brockii* showed H<sub>2</sub>S yields varying from 11.9 to 17.0 mmol/L from 20 mM of BCAA in the presence of 40 mM of thiosulfate and from 7.5 to 9.3 mmol/L for *C. subterraneus* subsp. *yonseiensis* and *T. brockii*, respectively (Scully and Orlygsson [2014](#page-9-7)). Clearly, there is a need for further studies to better explore the differences in the sulfur metabolism of these two genera.

Most *Thermoanaerobacter* and *Caldanaerobacter* strains grow optimally at neutral pH and between 65 and 70 °C. Thus, for simplicity reasons, in present investigation, all strains belonging to these two genera were cultivated at the same pH and temperature (pH 7.0 and at 65  $^{\circ}$ C). This may, however, not be their optimum growth conditions and results should be taken with caution concerning yields end product formation. For the *Thermoanaerobacter* and *Caldanaerobacter* strains, a more detailed study using not only variation in the liquid–gas phase ratios and the concentration of thiosulfate, but also an investigation into the role of culture conditions such as temperature and pH is needed to better understand conditions leading to the production of BCOHs rather than the corresponding BCFAs. From the study of *Caldanaerobacter subterraneus* subsp. *yonseiensis*, it was clear that both the concentration of thiosulfate as well as the ratio of liquid gas phases was of great importance for BCOH formation (Scully and Orlygsson [2014](#page-9-7)). Thus, using a high liquid–gas ratios increases the relative amount of the BCOH over the BCFA for this strain.

The majority of studies on the catabolism of the BCAAs have focused on aerobic bacteria such as species of *Staphylococcus* and *Enterococcus* (Beck et al. [2004](#page-8-5); Ward et al. [1999](#page-9-8)), aerotolerant anaerobes including *Lactobacillus sakei*



<span id="page-8-13"></span>**Fig. 4** Proposed pathway of BCFA and BCOH formation for *Thermoanaerobacter* and *Caldanaerobacter* species

(Gutsche et al. [2012\)](#page-9-9), or yeasts that use the Ehrlich pathway as shown in Fig. [4](#page-8-13) (Ehrlich [1907;](#page-8-6) Hazelwood et al. [2008](#page-9-10)). Here, we report the production of these products using species of *Thermoanaerobacter*. The phenomenon may be common among other members of class *Clostridia*, although the production of BCOHs may have escaped detection, as culture conditions require electron scavenging systems and low  $pH<sub>2</sub>$ . The degradation of amino acids presents a renewable route to potentially important feedstock chemicals. The interest in the production of BCOHs has often been directed to the formation of flavor compounds (branched- and aromatic chain aldehydes, alcohols, and acids) in food and beverage products (Hazelwood et al. [2008](#page-9-10)). Additionally, (*S*)-2-methylbutanol is a potential biofuel (Peralta-Yahya and Keasling [2010\)](#page-9-27) and some of the BCOHs may serve as building blocks (Sakuragi et al. [2011](#page-9-14)). A crude protein extract from fish waste or whey protein might be used as a good source of BCAAs for the production of these compounds.

### **Conclusions**

The production of BCOHs during the fermentation of BCAAs by thermophilic *Clostridia* appears to be limited to the genera of *Thermoanaerobacter* and *Caldanaerobacter*. The amount of BCOH was found to be proportionally higher within the genus *Caldanaerobacter* compared with *Thermoanaerobacter* and more profound from leucine and isoleucine as compared with valine. Clearly, there is a difference in thiosulfate reduction between investigated species. The role of  $pH_2$  and the enzymes involved requires further investigations to elucidate the pathway these bacteria use for BCOH formation.

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**Conflict of interest** The authors declare that they have no competing interests.

### **References**

- <span id="page-8-9"></span>Altschul S, Madden T, Schaffer A, Zhang J, Zhang Z, Miller W, Lipman D (1997) Gapped BLAST and PSI-BLAST: a new generation of protein database search programs. Nucleic Acids Res 25:3389–3402
- <span id="page-8-4"></span>Andreesen JR, Bahl H, Gottschalk G (1989) Introduction to the physiology and biochemistry of the genus *Clostridium*. In: Minton NP, Clarke DJ (eds) Clostridia. Plenum Press, New York, pp 27–62
- <span id="page-8-2"></span>Barker HA (1981) Amino acid degradation by anaerobic bacteria. Annu Rev Biochem 50:23–40
- <span id="page-8-5"></span>Beck HC, Hansen AM, Lauritsen FR (2004) Catabolism of leucine to branched-chain fatty acids in *Staphylococcus xylosus*. J Appl Microbiol 96:1185–1193
- <span id="page-8-10"></span>Cann IKO, Stroot PG, Mackie KR, White BA, Mackie RI (2001) Characterization of two novel saccharolytic, anaerobic thermophiles, *Thermoanaerobacterium polysaccharolyticum* sp. nov. and *Thermoanaerobacterium zeae* sp. nov., and emendation of the genus *Thermoanaerobacterium*. Int J Sys Evol Microbiol 51:293–302
- <span id="page-8-0"></span>Chang T, Yao S (2011) Thermophilic, lingocellulolytic bacteria for ethanol production: current state and perspectives. Appl Micorbiol Biotechnol 92:13–27
- <span id="page-8-7"></span>Choi KY, Wernick DG, Tat CA, Liao JC (2014) Consolidated conversion of protein waste into biofuels and ammonia using Bacillus subtilis. Metab Eng 23:53–61
- <span id="page-8-8"></span>Cord-Ruwisch R (1985) A quick method for determination of dissolved and precipitated sulphides in cultures of sulphate-reducing bacteria. J Microbiol Methods 4:33–36
- <span id="page-8-6"></span>Ehrlich F (1907) Uber die Bedingungen der Fuselöbildung und uber ihren Zusammenhang mit dem Eiweissaufbau der Hefe. Ber Dtsch Chem Ges 40:1027–1047
- <span id="page-8-1"></span>Elsden SR, Hilton MG (1978) Volatile acid production from threonine, valine, leucine and isoleucine by clostridia. Arch Microbiol 117:165–172
- <span id="page-8-11"></span>Euzéby JP (1997) List of bacterial names with standing in nomenclature: a folder available on the internet. Int J Syst Evol Microbiol 47:590–592
- <span id="page-8-3"></span>Fardeau M-L, Patel BKC, Magot M, Ollivier B (1997) Utilization of serine, leucine, isoleucine, and valine by *Thermoanaerobacter brockii* in the presence of thiosulfate or *Methanobacterium* sp. as electron acceptors. Anaerobe 3:405–410
- <span id="page-8-12"></span>Fardeau M-L, Salinas MB, L'Haridon S, Jeanthon C, Verhé F, Cayol J-L, Patel BKC, Garcia J-L, Ollivier B (2004) Isolation from oil resevoirs of novel thermophilic anaerobes phylogenetically related to *Thermoanaerobacter subterraneus*: reassignment of *T. subterraneus*, *Thermoanaerobacter yonseiensis*, *Thermoanaerobacter tengcongensis* and *Carboxydibrachium pacificum* to *Caldanaerobacter subterraneous* gen. nov., sp. nov., com. nov. as four novel subspecies. Int J Syst Evol Microbiol 54:467–474
- <span id="page-9-6"></span>Faudon C, Fardeau ML, Heim J, Patel B, Magot M, Ollivier B (1995) Peptide and amino acid oxidation in the presence of thiosulfate by members of the genus *Thermoanaerobacter*. Curr Microbiol 31:152–157
- <span id="page-9-9"></span>Gutsche KA, Tran TBT, Vogel RF (2012) Production of volatile compounds by *Lactobacillus sakei* from branched chain alpha-keto acids. Food Microbiol 29:224–228
- <span id="page-9-10"></span>Hazelwood LA, Daran J-M, van Maris AJA, Pronk JT, Dickinson JR (2008) The Ehrlich pathway for fusel alcohol production: a century of research on *Saccharomyces cerevisae*. Appl Environ Microbiol 74:2259–2266
- <span id="page-9-15"></span>Hungate RE (1969) A roll tube method for cultivation of strict anaerobes. In: Norris JR, Ribbons DW (eds) Methods in microbiology, vol 3B. Academic Press, New York, pp 117–132
- <span id="page-9-13"></span>Huo YX, Cho KM, Rivera JGL, Monte E, Shen CR, Yan YJ, Liao JC (2011) Conversion of proteins into biofuels by engineering nitrogen flux. Nat Biotechnol 29:346–351
- <span id="page-9-23"></span>Kozianowski G, Canganella F, Rainey FA, Hippe H, Antranikian G (1997) Purification and characterization of thermostable pectatelyases from a newly isolated thermophilic bacterium, *Thermoanaerobacter italicus* sp. nov. Extremophiles 1:171–182
- <span id="page-9-22"></span>Lee YE, Jain MK, Lee CY, Lowe SE, Zeikus JG (1993) Taxonomic distinction of saccharolytic thermophilic anaerobes—description of *Thermoanaerobacterium xylanolyticum* gen-nov, sp-nov, and *Thermoanaerobacterium saccharolyticum* gen-nov, sp-nov - reclassification of *Thermoanaerobium brockii*, *Clostridium thermosulfurogenes*, and *Clostridium thermohydrosulfuricum* E100-69 as *Thermoanaerobacter brockii* comb-nov, *Thermoanaerobacterium thermosulfurigenes* comb-nov, and *Thermoanaerobacter thermohydrosulfuricus* comb-nov, respectively—and transfer of *Clostridium thermohydrosulfuricum* 39E to *Thermoanaerobacter ethanolicus*. Int J Syst Bacteriol 43:41–51
- <span id="page-9-26"></span>Lee Y-J, Dashti M, Prange A, Rainey FA, Rohde M, Whitman WB, Wiegel J (2007) *Thermoanaerobacter sulfurigignens* sp. nov., an anaerobic thermophilic bacterium that reduces 1 M thiosulfate to elemental sulfur and tolerates 90 mM sulfite. Int J Syst Evol Microbiol 57:1429–1434
- <span id="page-9-3"></span>McInerney MJ (1988) Anaerobic hydrolysis and fermentation of fats and proteins. In: Zehnder AJB (ed) Biology of anarobic microorganisms. Wiley, New York, pp 373–415
- <span id="page-9-16"></span>Orlygsson J, Baldursson SRB (2007) Phylogenetic and physiological studies of four hydrogen-producing thermoanaerobes from Icelandic geothermal areas. Icelandic Agric Sci 20:93–106
- <span id="page-9-4"></span>Orlygsson J, Houwen FP, Svensson BH (1994) Influence of hydrogenotrophic methane formation on the thermophilic anaerobic degradation of proteins and amino acids. FEMS Microbiol Ecol 13:327–334
- <span id="page-9-5"></span>Orlygsson J, Houwen FP, Svensson BH (1995) Thermophilic anaerobic amino acid degradation–deamination rates and end product formation. Appl Microbiol Biot 43:235–241
- <span id="page-9-17"></span>Orlygsson J, Sigurbjornsdottir MA, Bakken HE (2010) Bioprospecting thermophilic ethanol and hydrogen producing bacteria from hot springs in Iceland. Icelandic Agricult Sci 23:73–85
- <span id="page-9-25"></span>Parte AC (2014) LPSN—list of prokaryotic names with standing in nomenclature. Nucl Acid Res 42:D613–D616
- <span id="page-9-27"></span>Peralta-Yahya PP, Keasling JD (2010) Advanced biofuel production in microbes. Biotechnol J 5:147–162
- <span id="page-9-1"></span>Ren N, Wang A, Cao G, Xu J, Gao L (2009) Bioconversion of lignocellulosic biomass to hydrogen: potential and challenges. Biotechnol Adv 27:1051–1060
- <span id="page-9-14"></span>Sakuragi H, Kuroda K, Ueda M (2011) Molecular breeding of advanced microorganism for biofuel production. J Biomed Biotechnol. doi:[10.1155/2011/416931](http://dx.doi.org/10.1155/2011/416931)
- <span id="page-9-7"></span>Scully SM, Orlygsson J (2014) Branched-chain alcohol formation from branched-chain amino acids by *Thermoanaerobacter brockii* and *Thermoanaerobacter yonseiensis*. Anaerobe 30:82–84
- <span id="page-9-2"></span>Scully SM, Orlygsson J (2015) Recent advances in second generation ethanol production by thermophilic bacteria. Energies 8:1–30
- <span id="page-9-19"></span>Skirnisdottir S, Hreggvidsson GO, Hjorleifsdottir S, Marteinsson VT, Petursdottir SK, Holst O, Kristjansson JK (2000) Influence of sulfide and temperature on species composition and community structure of hot spring microbial mats. Appl Environ Microbiol 66:2835–2841
- <span id="page-9-11"></span>Smit G, Smit A, Engels WJM (2005) Flavour formation by lactic acid bacteria and biochemical flavor profiling of cheese products. FEMS Microbiol Rev 29:591–610
- <span id="page-9-12"></span>Smit BA, Engels WJM, Smit G (2009) Branched chain aldehydes: production and breakdown pathways and relevance for flavor in foods. Appl Microbil Biot 81:987–999
- <span id="page-9-24"></span>Stams AJM, Hansen JA (1984) Fermentation of glutamate and other compounds by *Acidaminobacter hydrogenoformans* gen. nov. sp. nov., an obligate anaerobe isolated from black mud. Studies with pure cultures and mixed cultures with sulfate-reducing and methanogenic bacteria. Arch Microbiol 137:329–337
- <span id="page-9-21"></span>Tamura K, Nei M (1993) Estimation of the number of nucleotide substitutions in the control region of mitochondrial DNA in humans and chimpanzees. Mol Bio Evol 10:512–526
- <span id="page-9-20"></span>Tamura K, Stecher G, Peterson A, Kumar S (2013) MEGA6: molecular evolutionary genetic analysis. Version 6.0. Mol Biol Evol 30:2725–2729
- <span id="page-9-0"></span>Taylor MP, Eley KL, Martin S, Tuffin M, Burton SG, Cowan DA (2009) Thermophilic ethanologenesis: future prospects for second-generation bioethanol production. Trends Biotechnol 27:398–405
- <span id="page-9-8"></span>Ward DE, Ross P, van der Weijden CC, Snoep JL, Claiborne A (1999) Catabolism of branched-chain alpha-keto acids in *Entercoccus faecalis*: the bkd gene cluster, enzymes, and metabolic route. J Bacteriol 181:5433–5442
- <span id="page-9-18"></span>Westley J (1987) Thiocyanate and thiosulfate. Methods in Enzym 143:22–25