



Correction to: Protein-bound calcium phosphate in uremic rat serum: a quantitative study

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Correction to:

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In the original article, few equations and units were published incorrectly. The correct equations and units are given below:

Page 5: In the first line of the note under Table 2, “Mean ± standard deviation.” should be “mean ± standard deviation (unit: mmol/L).”

Page 7: In Eq. (4), “ $\frac{1}{k_j \cdot t \cdot [\text{pr}]}$ ” should be “ $\frac{1}{K_j \cdot t \cdot [\text{pr}]}$ ”. The whole Eq. (4) should be present as:

$$\frac{1}{[\text{prCa}]} = \frac{1}{K_j \cdot t \cdot [\text{pr}]} \cdot \frac{1}{f\text{Ca}} + \frac{1}{t \cdot [\text{pr}]} \quad (4)$$

The original article can be found online at <https://doi.org/10.1007/s00775-020-01807-x>.

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Page 7: In Eq. (5), “0.63610–30.83” should be “ $0.636 \times 10^{-3} \times 0.83$ ”. The whole Eq. (5) should be present as:

$$t = (1/\text{intercept})/[\text{pr}] \\ = (1/103)/(0.636 \times 10^{-3} \times 0.83) = 18.4 \approx 18 \quad (5)$$

Page 8: In line 13 from bottom, “1.76” should be “1.76 mmol/L”.

Page 8: In line 12 from bottom, “ 0.636×10^{-3} ” should be “ 0.636×10^{-3} mol/L”.

Page 8: In Eq. (9b), “ $\frac{K_{a2}(\text{uPi}-\text{cPi})}{[\text{H}^+] + K_{a2}}$ ” should be “ $\frac{K_{a2}(\text{uPi}-\text{cPi})}{[\text{H}^+] + K_{a2}}$ ”. The whole Eq. (9b) should be present as:

$$[\text{HPO}_4^{2-}] = \frac{K_{a2}(\text{uPi} - \text{cPi})}{[\text{H}^+] + K_{a2}} \\ = \frac{(6.17 \times 10^{-8})(\text{uPi} - \text{cPi})}{(3.98 + 6.17) \cdot 10^{-8}} = 0.608(\text{uPi} - \text{cPi}) \quad (9b)$$

Page 9: In the end of the paragraph above Eq. (12), “0.15 mmol/L.” should be “0.15 mmol/L.” A full stop.

Page 9: In Eq. (14), “ 1.7310^{-3} ” should be “ 1.73×10^{-3} ”. The whole Eq. (14) should be present as:

$$[\text{HPO}_4^{2-}]_0 \equiv -\text{Intercept}/K_m = 0.361/209 = 1.73 \times 10^{-3} \text{ mol/L} \quad (14)$$

Page 9: In Eq. (15), the two semicolons (;) should be deleted. The whole Eq. (15) should be present as:

$$[\text{HPO}_4^{2-}] - [\text{HPO}_4^{2-}]_0 > 0 \quad ([\text{prPi}] > 0) \quad (15)$$

Page 10: In Eq. (17a), “ $\frac{1}{[\text{Ca}^{2+}([\text{HPO}_4^{2-}] - [\text{HPO}_4^{2-}]_0)]}$ ” should be “ $\frac{1}{[\text{Ca}^{2+}([\text{HPO}_4^{2-}] - [\text{HPO}_4^{2-}]_0)]}$ ”. There is a subscript “0”. The whole Eq. (17a) should be present as:

$$K = K_j K_m = \frac{[\text{prCa}]}{([\text{t} \cdot \text{pr}] - [\text{prCa}])} \cdot \frac{[\text{prPi}]}{([\text{prCa}] - [\text{prPi}])} \cdot \frac{1}{[\text{Ca}] \cdot ([\text{HPO}_4^{2-}] - [\text{HPO}_4^{2-}]_0)} \quad (17a)$$

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