



# Rebalancing metal dyshomeostasis for Alzheimer's disease therapy

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## Abstract

Alzheimer's disease (AD) is a type of neurodegenerative malady that is associated with the accumulation of amyloid plaques. Metal ions are critical for the development and upkeep of brain activity, but metal dyshomeostasis can contribute to the development of neurodegenerative diseases, including AD. This review highlights the association between metal dyshomeostasis and AD pathology, the feasibility of rebalancing metal homeostasis as a therapeutic strategy for AD, and a survey of current drugs that action via rebalancing metal homeostasis. Finally, we discuss the challenges that should be overcome by researchers in the future to enable the practical use of metal homeostasis rebalancing agents for clinical application.

**Keywords** Metal dyshomeostasis · Metal ions · Alzheimer's disease · Metal chelators · A $\beta$  aggregation inhibitors · Multifunctional agents

## Abbreviations

AD	Alzheimer's disease
A $\beta$	Amyloid-beta
BBB	Blood–brain barrier
CNS	Central nervous system
APP	Amyloid precursor protein
ROS	Reactive oxygen species
8HQ	8-Hydroxyquinoline
CQ	Clioquinol
DFP	Deferiprone
TETA	Trientine
DFO	Desferrioxamine
EGCG	(–)-Epigallocatechin-3-gallate
BACE1	$\beta$ -Secretase
AChE	Acetylcholinesterase
NMDA	<i>N</i> -Methyl-D-aspartic acid

## Introduction

Metals can be broadly classified into toxicological metals and biometals based on their physiological functions, or lack thereof [1]. Toxicological metals including lead, mercury, aluminum, and cadmium have no beneficial biological function and are toxic to the human body [1]. Interestingly, these metals are concentrated in the brain after absorption into the bloodstream [1]. Meanwhile, biometals are metal ions that are essential for normal body function, including brain function. Critically, metal dyshomeostasis has been linked to the etiology of a range of neurodegenerative diseases, including Alzheimer's disease (AD) in particular [2, 3].

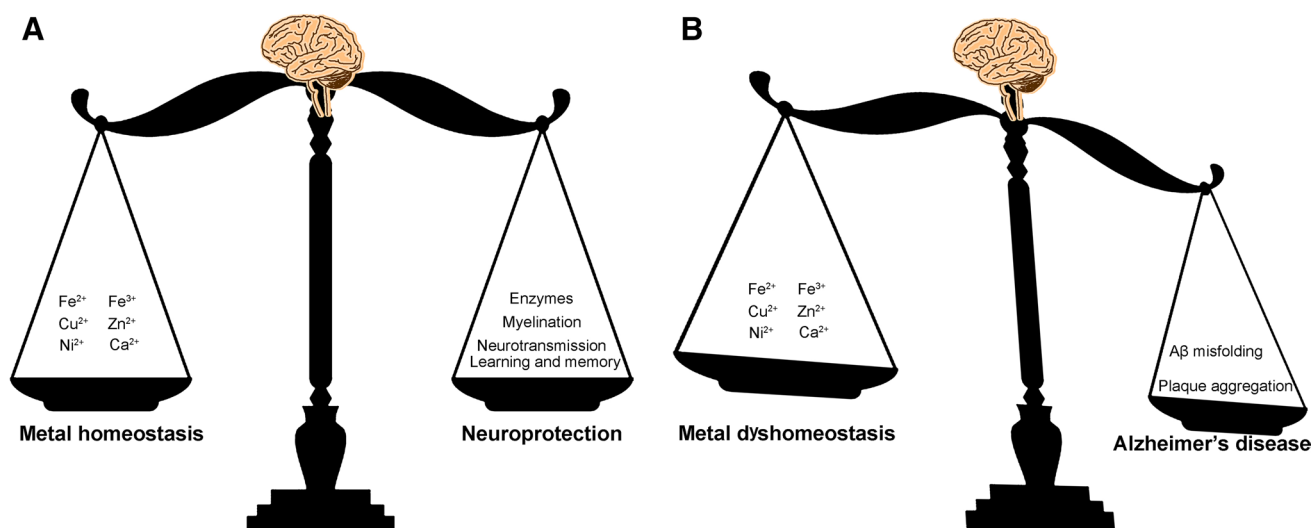
AD is a progressive neurodegenerative disease that afflicts almost 36 million individuals worldwide [4–6]. AD is recognized by the presence of amyloid-beta (A $\beta$ ) plaque deposits in the brain. Critically, the A $\beta$  misfolding has been found to be significantly affected by the presence of metals, which are found in both the interior and periphery of established AD plaques [7–10]. Recently, there has been renewed attention given to the function of metal dyshomeostasis as a causative agent for AD. A number of key studies have shown that re-establishing proper metal ion balance in the brain can halt A $\beta$  aggregation, disassemble amyloid plaques, and slow down AD-related cognitive decline in both AD patients and AD transgenic mice (Fig. 1) [11–15]. Here, we review the recent literature on the role of metal ion imbalance on the pathogenesis of AD. We also discuss new possible strategies

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**Fig. 1** **a** Normal metal homeostasis is essential for healthy brain function. **b** Metal dyshomeostasis can lead to AD

for treating AD that rely on restoring metal balance in the brain. Finally, we provide our critical view and outlook of the challenges that must be overcome by researchers seeking to progress metal-targeting drugs into the clinic.

### Metal homeostasis in brain regulation

Metals such as zinc, iron and copper are essential for the biologically vital processes of the human body, including catalysis, stabilization of protein structure, signal transmission, and metabolism [16, 17]. Hence, living systems have developed intricate strategies for regulating the levels of ions and tightly control the entry, exit and storage of metal ions both within the cell and within sub-cellular organelles [18]. Metals can be imported into the brain through two pathways, either via the blood–brain barrier (BBB), or through the brain–cerebrospinal fluid barrier. Metals pass through the BBB in the form of free ions, or they can be carried by ion-specific transporters, or alternatively, they can traverse through the BBB by complexing with cysteine or other amino acids [19, 20]. Within the brain, metal ions are mostly concentrated within membrane metalloproteins or inside synaptic vesicles [21]. The concentration of metal ions can fluctuate in different areas of the brain [22] or subcellular compartments. Astrocytes in the brain help to maintain metal homeostasis [23–25], as they are the first kind of cell to interact with metals when they pass through the BBB. Also, astrocytes possess large amounts of metal-binding biomolecules, including metallothioneins and glutathione, which enable them to sequester metal ions and limit their neurotoxicity [26].

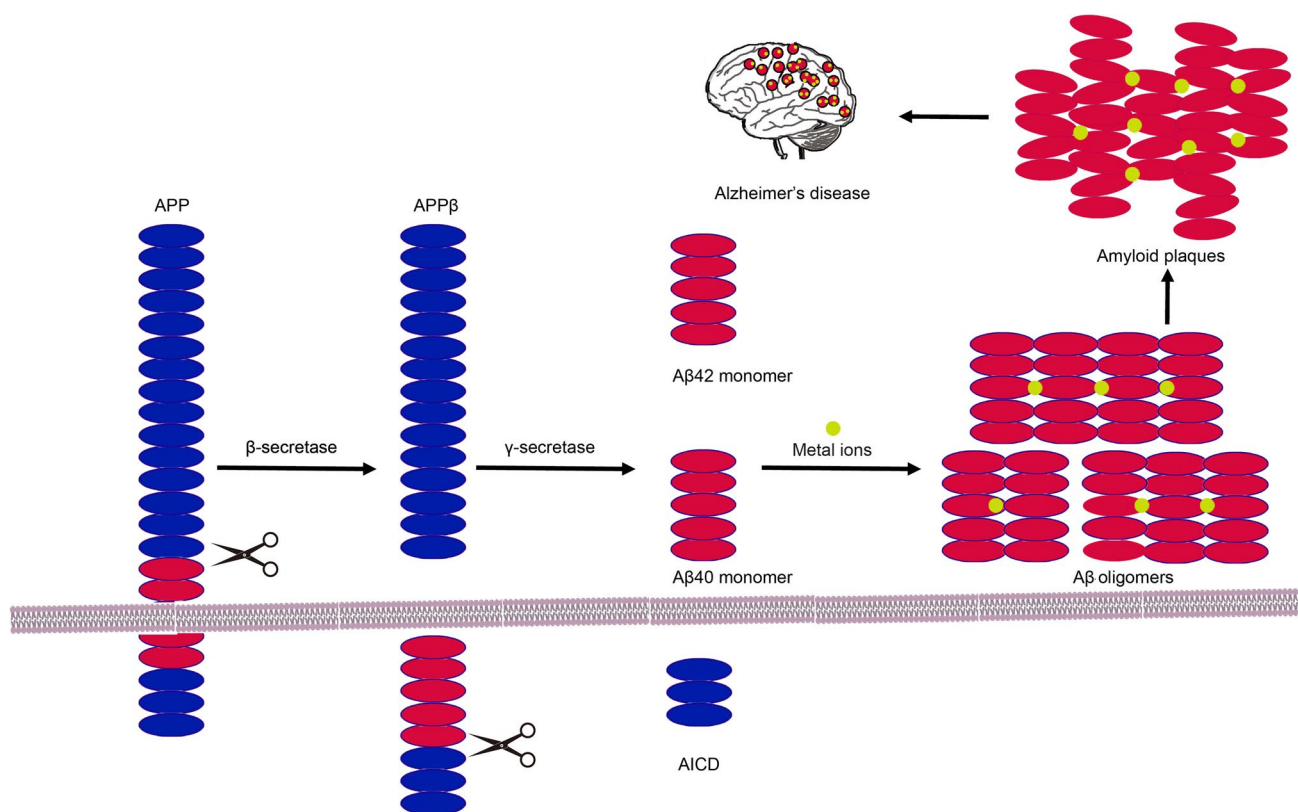
Due to the critical role of metals in the brain, the tight regulation of metal homeostasis in the brain is essential. For

instance, the correct development of the brain is dependent on appropriate copper regulation [27], as dysfunction copper levels can trigger development defects and abnormalities of the brain. Copper and zinc also play important functions in neurotransmission in the central nervous system (CNS) [28]. Lastly, biometals regulate the potentiation of synapses in the long term, as well as the pharmacology of synaptic receptors [29].

### The pathogenic hypothesis of AD

The “metal hypothesis” describes the theory that an imbalance in the levels and localization of metal ions contributes to AD pathology [30]. This theory posits that the binding of metals to A $\beta$  can affect peptide conversion and promote its aggregation into fibrils, eventually forming plaques [30] (Fig. 2). In the amyloid precursor protein (APP) sequence, there are 2 or 4 metal-binding sites, depending on the metal ion recognized [31, 32]. APP has sites that are selective for binding to zinc and copper, and these sites also regulate redox activity to cause aggregation of A $\beta$  even at low levels [33, 34]. Meanwhile, A $\beta$  has both strong and weak-affinity selective sites for binding metals. They can bind equal ratios of zinc and copper, but copper can completely replace zinc in the binding site under acidosis [35, 36]. A $\beta$  reduces metal ions by delivering electrons to O $_2$ , with the concomitant production of hydrogen peroxide.

Interestingly, while the degree of deposits of A $\beta$  is age related, there does not appear to be a relationship between A $\beta$  production and age. This may suggest that other changes that are dependent on age, such as aberration of metal homeostasis, may instead be responsible for the pathogenesis of A $\beta$ . As both zinc and copper and zinc modulate



**Fig. 2** The pathophysiology of AD, showing the production of Aβ monomers from APP, which aggregates into oligomers and eventually plaques

neurotransmission [30], deficiencies in metal balance can adversely affect transmission across synapses, via altering the reuptake or storages of metals in the synaptic cleft. Taken together, these data support the metal hypothesis of AD [30], which is based on the idea that the pathogenesis of AD is driven by the interaction of Aβ with particular metal ions. Additionally, a consensus has emerged on the view that transition metals can trigger oxidation stress, leading to another pathological factor driving AD. Accumulating evidence in the literature has ensured that the metal hypothesis of AD is now well established in the scientific community [37–43].

### Metal dyshomeostasis and pathophysiology of AD

The deposition in the brain of Aβ plaques is a key hallmark of AD. The plaques consist of aggregated Aβ peptides ranging between 39 and 43 amino acids in length. These peptides are generated from the transmembrane precursor, APP [44]. Unbalanced metal ion concentrations can disrupt the normal activity of enzymes that regulate the cleavage of APP, leading to aberrant Aβ formation [45]. The 1–40 and 1–42 isoforms are the most abundant isoforms of Aβ. The latter isoform has a greater propensity for fibrillation and, thus, plaque formation [46, 47]. Amyloid

fibrillation is reversible in the early stages (Fig. 2). However, whether neurotoxicity is primarily dependent on plaque formation or alternatively to monomeric Aβ toxicity, is still not fully known [48]. In contrast, some studies have shown the neuroprotective effect of monomeric Aβ [49]. Experiments have implicated oxidative stress in AD pathology [48], which could promote the apoptotic death of neurons either via caspase-3-dependent or independent pathways [50]. The presence of copper, iron, and zinc in Aβ aggregates has also been recently linked to neurotoxicity [51, 52]. Studies have shown that these metal ions can interact with and induce the aggregation of Aβ. Moreover, the redox activity of metal ions can trigger a cascade that leads to the production of reactive oxygen species (ROS) [53, 54]. Another mechanism that has been implicated in the pathogenesis of AD is the failure of the ubiquitin–proteasome system (UPS) [55, 56]. Metal dyshomeostasis can lead to the abnormal functioning of the UPS and trigger neurodegeneration via inhibiting self-polyubiquitination reactions [57]. Metal dyshomeostasis has also been suggested to induce CNS damage via modulating autophagy [55, 58–62]. In cellulo experiments also showed that bio-metal dyshomeostasis is connected to mitochondrial dysfunction [63, 64] and lysosomal storage disorders [65–67] in neurons. This suggests the possibility of targeting metal

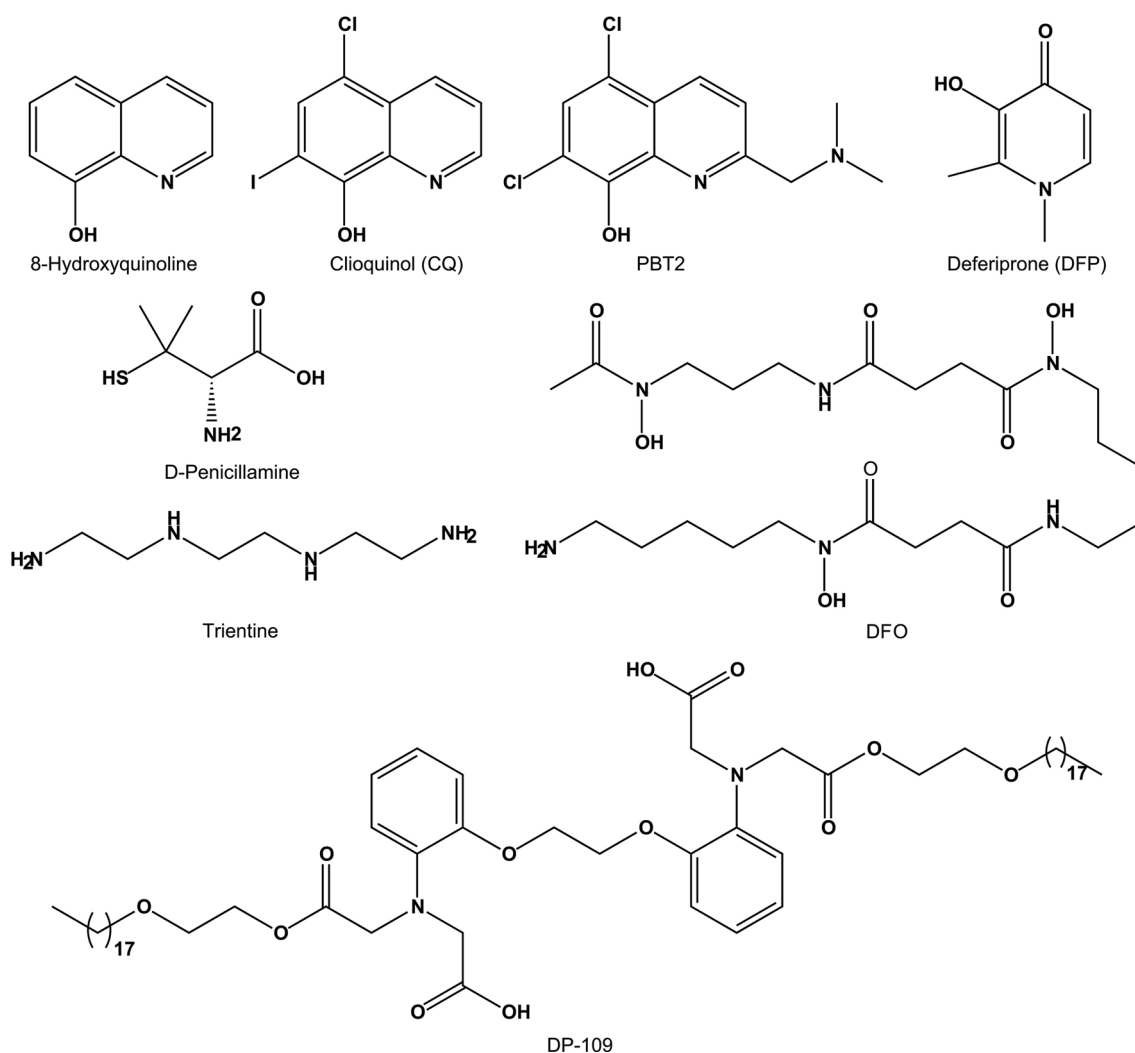
dyshomeostasis in the brain as a potential therapeutic approach.

## Targeting metal dyshomeostasis

The putative role of metal dyshomeostasis in promoting the pathology of AD suggests that rebalancing metal homeostasis may be a potential strategy for AD therapy [68]. Metal chelators can sequester metal ions and remove them from the site of the lesion, thus preventing them from participating in redox chemistry or promoting A $\beta$  aggregation. Meanwhile, other compounds can target specific compartments or organelles where metal ions are either relatively abundant or deficient [1]. The following sections discuss recent examples of compounds that target metal dyshomeostasis as potential therapeutic agents for AD.

## Metal chelators

Metal chelators are molecules that can capture and bind metal ions via forming two or more coordinate bonds to a single metal ion. They can deplete the total pool of bioavailable metals extracellularly or compete with endogenous ligand as for metal ions as ionophores [1]. 8-Hydroxyquinolines (8HQ), a lipophilic molecule from plants, has been reported to exhibit potent anti-AD effects via chelating metal ions (Fig. 3) [25]. Clioquinol (CQ), a derivative of (8HQ), has entered into the phase II clinical trials for AD therapy [1]. PBT2, another drug for AD treatment in phase II clinical trials, was documented to promote A $\beta$  degradation and phosphorylation of GSK3 through its metal chelator activity [69]. Deferiprone (DFP), an oral iron chelator that has been previously used for the treatment of thalassemia syndrome, exhibits neuroprotective action via alleviating the phosphorylation levels of A $\beta$  and tau without affect the generation of



**Fig. 3** Metal chelating agents for AD

ROS [70]. Trientine (TETA), a tetradentate chelator, inhibits  $\beta$ -secretase (BACE1) and slows amyloidosis via targeting the AGE/RAGE/NF- $\kappa$ B signaling cascade in a murine model of AD [71]. D-penicillamine, a Cu/Zn chelator used as a clinical drug for Wilson's disease and rheumatoid arthritis [72–74], can delay AD via reducing serum oxidative stress [75]. Desferrioxamine (DFO), a preferential iron chelator, was shown to impede APP holoprotein translation via IRE in the 5' UTR of the *APP* transcript and significantly improve some cognitive functions in AD patients injected intramuscularly [76]. The preclinical chelator DP-109 has been successfully tested on AD models. In 3-month mouse experiments with Tg2576 mice, DP-109 appeared to induce the increase the solubility of A $\beta$  in the cerebrum, thereby decreasing the amount of A $\beta$  aggregates [77]. Moreover, several metal chelating drugs (e.g., metformin and cyclodipeptides), already approved and used for other purposes, have been reported to be effectively repurposed for the treatment of neurodegenerative diseases, including AD [78].

Although several metal ion chelators have been approved for clinical use for metal overexposure diseases such as for Wilson's disease [79], lead toxicity [80], and rheumatoid arthritis, to date no chelators have been approved for use in treating AD. Obstacles to the use of metal chelators are the potential adverse effects resulting from the removal of essential metal ions from the brain, as well as their poor

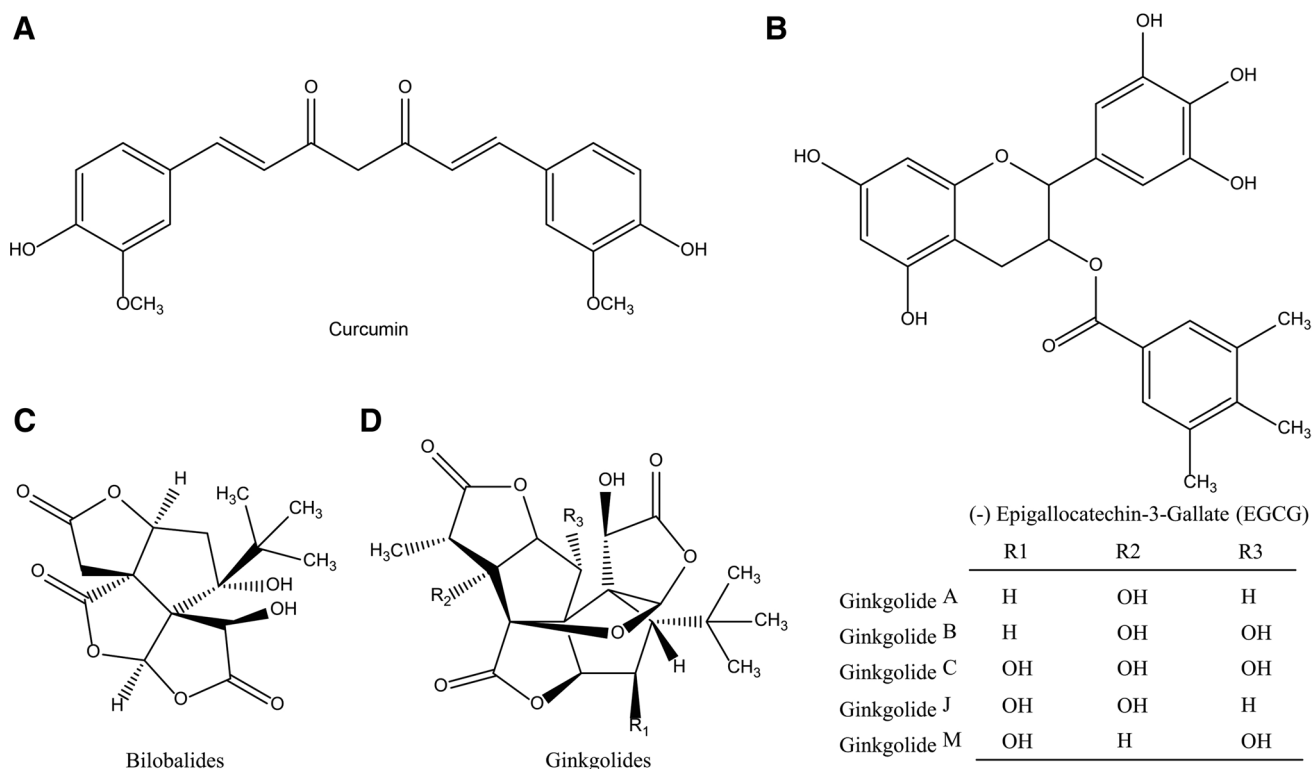
BBB permeability due to their hydrophilic nature [61, 62]. Hence, further research is needed into the development of more selective metal ion chelators that can effectively enter the brain to target metal dyshomeostasis without causing systemic effects.

## A $\beta$ aggregation inhibitors

An increasing amount of evidence supports the hypothesis that metal ions alter the kinetic pathway of A $\beta$ , directing its aggregation away from a more stable fibrillar structure and toward more neurotoxic structures [81]. It is also believed that fibrils and plaques during AD development result from metal binding to A $\beta$ , causing its aggregation [30]. Thus, inhibition of A $\beta$  aggregation is also a potential strategy to combat metal dyshomeostasis-induced AD.

## Natural product-based A $\beta$ aggregation inhibitors

Natural products are a source of chemical scaffolds with abundant activity profiles and moderate toxicity [82, 83]. Several natural products have been explored for their anti-AD activity (Fig. 4). Curcumin and its analogs such as calebin-A and dimethoxycurcumin have shown activity against fibril formation and extension, and promoted the destabilization of pre-aggregated A $\beta$  peptides in SH-SY5Y cells and



**Fig. 4** Natural product-based A $\beta$  aggregation inhibitors

in vivo [84–86]. (–)-Epigallocatechin-3-gallate (EGCG), a polyphenolic constituent of green tea, has been studied for the prevention of age-related AD. A number of cell and animal experiments have shown that EGCG impeded A $\beta$ -induced cell death at micromolar concentrations via decreasing soluble and insoluble levels of A $\beta_{40}$  and A $\beta_{42}$  and alleviating plaque load [87–89]. Bilobalide, a compound extracted from the Chinese medicinal plant *Ginkgo biloba*, reduced A $\beta$ -stimulated apoptosis via downregulating ROS and blocking NF- $\kappa$ B activation [90, 91]. Ginkgolides A, B, C, J and M were also active against AD in part via anti-inflammatory and antioxidative properties [92, 93]. Ginkgolide A prevented A $\beta$ -induced depolarization of cortical neurons by targeting the NMDA receptor [92]. Ginkgolide B reduced A $\beta_{42}$ -induced oxidative damage and restored the long-term antioxidant activities of enzymes in SH-SY5Y cells [94]. Meanwhile, ginkgolide J can enhance memory via inhibiting A $\beta_{42}$ -induced cell death in rodent hippocampal neurons [95].

### Peptide-based A $\beta$ aggregation inhibitors

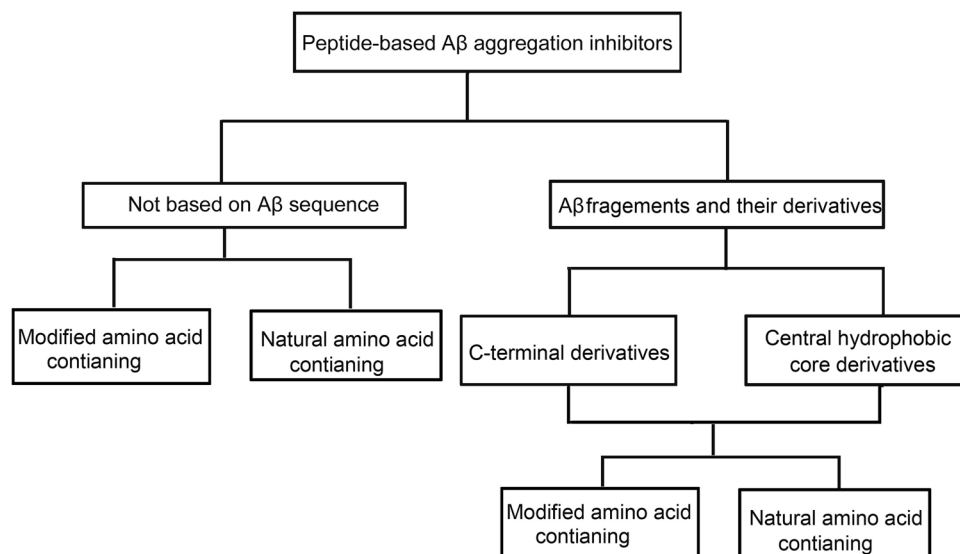
Peptide inhibitors have also been used to develop potent pharmacological agents for AD treatment due to their high specificity, low toxicity, BBB permeability, and high chemical and biological diversity [96]. To date, many peptide-based aggregation inhibitors have been developed, which can be classified several subgroups based on their design principle (Fig. 5). Peptide fragments that consist of the hydrophobic core of A $\beta$  [such as residues (15–22), (16–23), (17–24), (25–35)] or the C-terminal of A $\beta$  [such as residues (25–35), (28–38), (39–42)] have been reported to block A $\beta$  aggregation via binding to full-length A $\beta$  [96–98]. Interestingly, the ability of the A $\beta$ -derived peptide inhibitors to bind

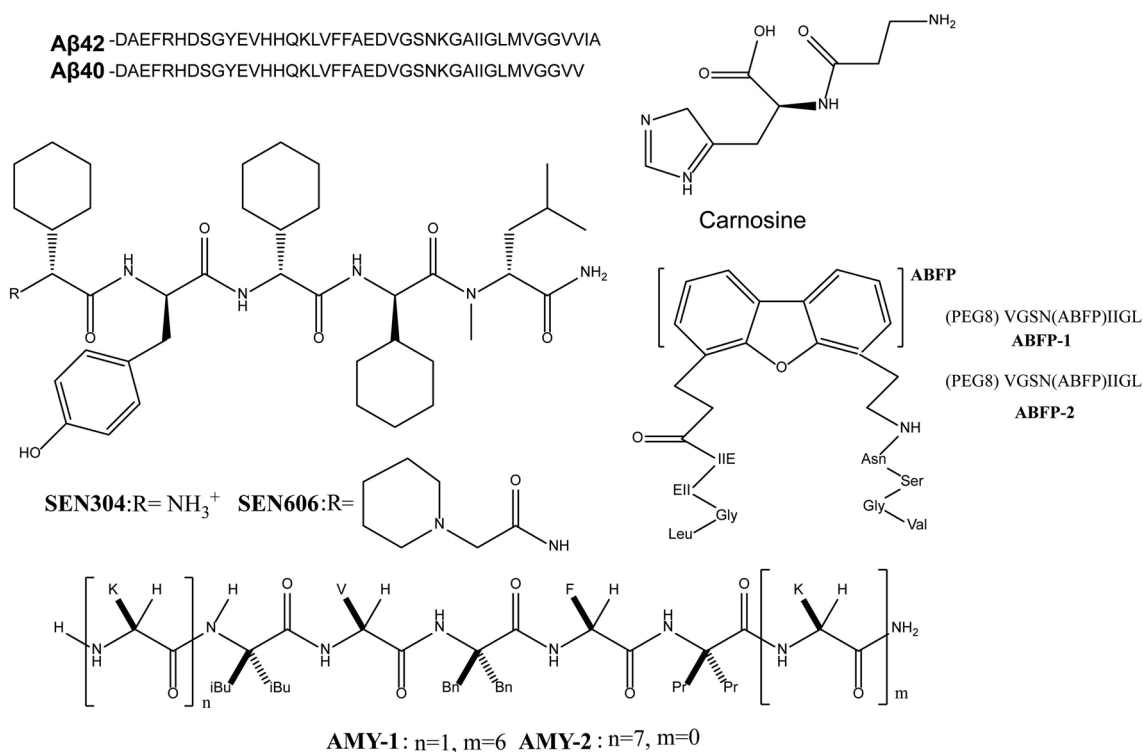
to A $\beta$  and block its aggregation depends on their hydrophobicity, which helps them to incorporate into the  $\beta$ -sheet structure of A $\beta$ . A metalloporphyrin–peptide conjugate is an effective inhibitor of amyloid- $\beta$  peptide fibrillation and cytotoxicity [99].

Alternatively, neuroprotective peptides that are not based on the A $\beta$  sequence have also been reported. NAP (amino acid sequence: NAPVSIPQ), a peptide drug in phase II clinical trials, was documented to prevent formation of fibrils via abrogating the assembly and inducing disaggregation of A $\beta$  [100]. A 12-mer peptide (PWRWQLWWHNWS), which was identified using the phage display technique, selectively bound to A $\beta$  (1–10) and thus interdicted A $\beta$  fibrillation by maintaining a steady-state equilibrium between monomeric A $\beta$  monomer and soluble plaques, leading to an increase in the proportion of soluble A $\beta$  [101]. An endogenous dipeptide, carnosine (Fig. 6), interacts with monomeric A $\beta$  via impeding intermolecular salt bridge formation, thus blocking the aggregation event. This dipeptide significantly reduced A $\beta$  accumulation and greatly relieved AD- and age-related mitochondrial dysfunction in a transgenic mice model of AD [102, 103].

Natural amino acid-containing A $\beta$  fragments, while effectively inhibiting A $\beta$  aggregation, themselves have high risk of self-associating into fibrils and also exhibit low resistance to cellular proteolytic enzymes [96]. To overcome these problems, many modified peptides have been developed. SEN304 (Fig. 6), a modified derivative of the KLVFF sequence, inhibited A $\beta$  aggregation via directly binding to A $\beta_{40}$  and A $\beta_{42}$ , retarded  $\beta$ -sheet formation, and induced formation of oligomers in a nontoxic form conformation [104]. SEN606, a derivative of SEN304, was reported to exhibit similar nanomolar activity in preclinical trials. **AMY-1** and **AMY-2**, two peptide analogs of the hydrophobic interior of

**Fig. 5** Classifications of peptide aggregation inhibitors of A $\beta$  aggregation inhibitors





**Fig. 6** Aβ sequences and their peptide inhibitors

Aβ, bound avidly to fibrils and abrogated its further assembly via forming a blocking surface [105]. Meanwhile, tongue-type mimic peptides (**AFBP**, **AFBP-1**, and **AFBP-2**) mimic the turn motif of Aβ oligomers and could form β-sheets with oligomerized Aβ and to inhibit Aβ aggregation [106].

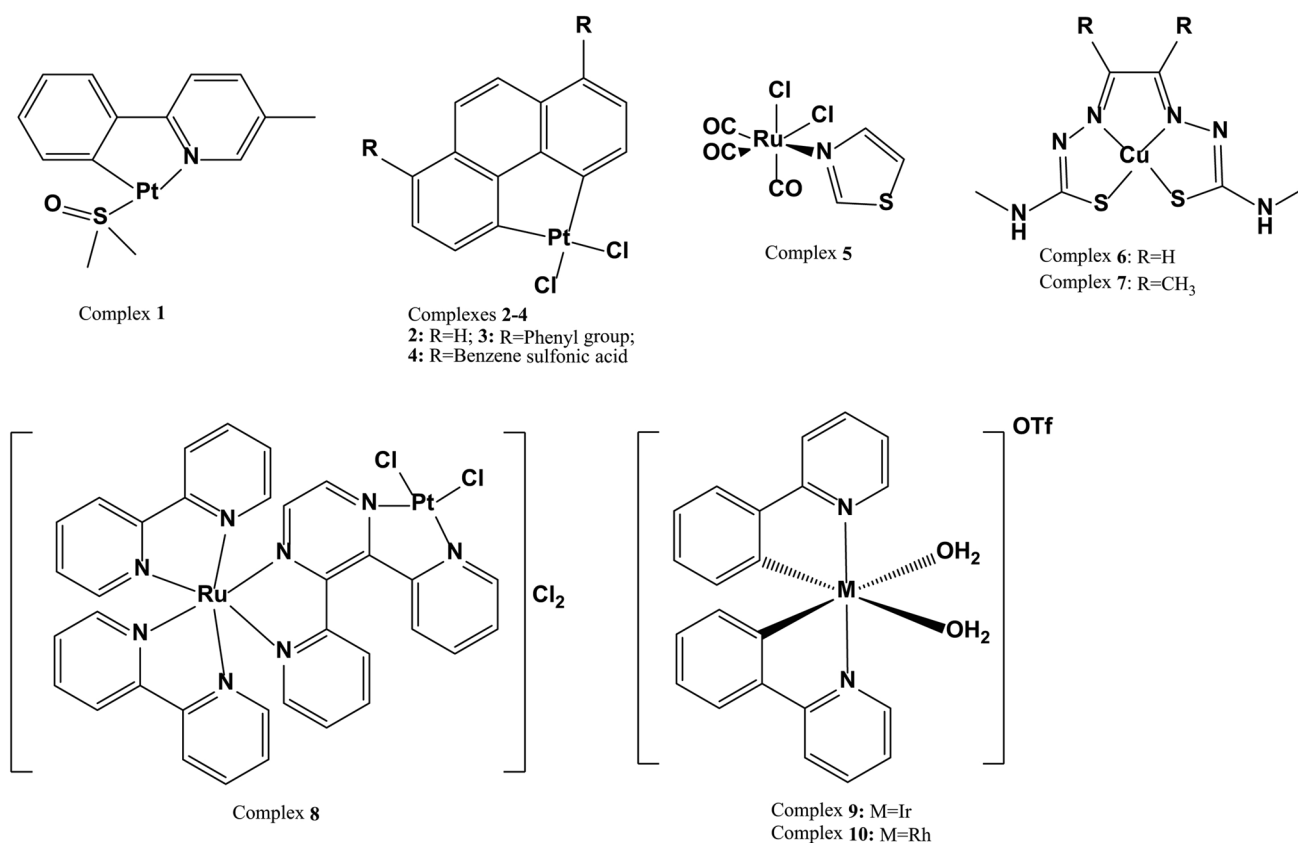
### Metal-based Aβ aggregation inhibitors

Transition metal complexes have also emerged as a viable alternative for the treatment of AD [107, 108]. Organometallic complexes can display great structural diversity of geometrical shapes via variations in assembly of the metal center and its co-ligands [83], allowing the effective targeting of the active sites of proteins or enzymes through shape-specific interactions [107]. The high-affinity metal binding site in Aβ peptides binds with several metal ions such as zinc, copper, and iron to mediate peptide aggregation and toxicity [108, 109]. Therefore, occupying this targeting site on Aβ peptides may be a possible therapeutic strategy for AD treatment. Based on this idea, three platinum (Pt) complexes (complexes **1–3**) containing phenanthroline ligands were developed as Aβ inhibitors [110] (Fig. 7). Another cisplatin-based complex **4** was found to interrupt the binding between Aβ<sub>16</sub> peptides and Cu(II) via binding to the histidine imidazole moiety of Aβ<sub>16</sub> [111]. Similarly, an Ru-based complex **5** was developed that could bind to both Aβ<sub>28</sub> and Aβ<sub>42</sub> via interacting with His-13 and His-14 [112]. Interestingly, a

Pt/Ru dual metal core complex **6** also exhibited selective binding to Aβ and impeded amyloid fibril formation via binding to Aβ<sub>42</sub> in a ratio of 2:1 [113]. Two Cu complexes containing bis(thiosemicarbazone) (**7** and **8**) have also been shown to reduce the progression of AD. Unlike the above metal complexes, these two metal complexes exhibited their bioactivity by reducing the oxidation state of the intracellular copper from Cu<sup>2+</sup> to Cu<sup>+</sup>, thus abrogating their ability to bind to Aβ peptides [114]. Two group 9 metal-based complexes (**9** and **10**) were identified by our group which slowed the aggregation of Aβ<sub>40</sub> via binding to His imidazole [107]. Interestingly, complex **10** (rhodium core) was more active than complex **9** (iridium core), suggesting that the interaction with the metal complex with Aβ is strongly metal dependent [107]. Another report described the ability of metal complexes with cyclam glycoconjugates to protect against metal dyshomeostasis-induced amyloid aggregation [115].

### Multifunctional agents against AD

It is believed that many factors contribute to the pathogenesis of AD. Pharmacological agents that target only one factor are often incapable of exerting sufficient therapeutic effect to reverse the progression of AD [116]. Thus, many multifunctional agents have been studied for anti-AD activity in cellulo or even in vivo (Fig. 8). Myricetin, a



**Fig. 7** Metal-based A $\beta$  aggregation inhibitors

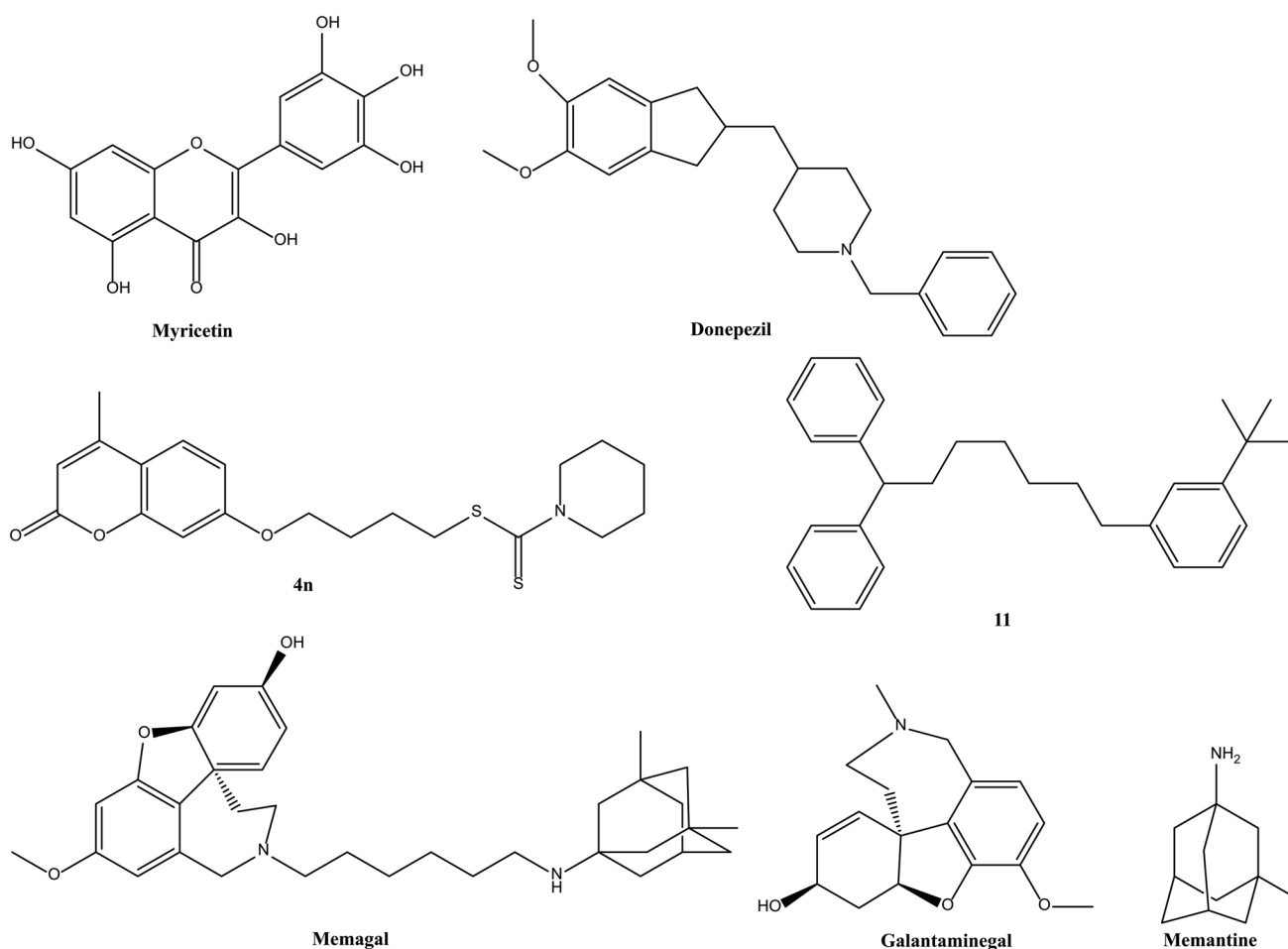
natural polyphenol, exhibited neuroprotection via regulating A $\beta$  conformation and reducing the enzyme activity of secretases [117]. Donepezil is an inhibitor of A $\beta$  aggregation and BACE1 enzyme, which activates the production of A $\beta$ . Donepezil also binds to sigma-1 receptors, which have anti-amnesic activity [118–120]. Apart from natural products, many synthetic compounds have also been developed as multifunctional agents against AD. A multifunctional compound (**4n**) was produced by hybridizing coumarin, an acetylcholinesterase (AChE) and A $\beta$  aggregation inhibitor, and dithiocarbamate, an AChE inhibitor. The compound exhibited much higher activity against AChE inhibition than either individual molecule and greatly reduced A $\beta$  aggregation [121]. The combination of AChE inhibitors with an *N*-methyl-D-aspartic acid (NMDA) receptor antagonist is the current recommended standard for AD treatment [122]. A hybrid compound, memagal, formed from the hybridization of galantamine, AChE inhibitor, and memantine, an NMDA receptor antagonist, showed multifunctional activity in cellulo [122]. A 1-benzylamino-2-hydroxyalkyl derivative (**11**) showed anti-AD activity via combining inhibition against butyrylcholinesterase, BACE1,  $\beta$ -amyloid aggregation, and tau aggregation in a mouse model [123]. In addition, based on drug repurposing, several antipsychotic drugs such as

pimozide, benperidol, and anisopirrol were also found to inhibit multiple targets involved in AD [124].

### Challenges for targeting metal dyshomeostasis against AD

With dramatic rise in the number of AD patients, developing the drugs which can prevent, or cure AD is urgently needed. However, even though many compounds have shown efficacy in treating AD in animal models or clinical trials, few have succeeded in the clinic in terms of showing sustained therapeutic effect. There are many challenges for identifying novel agents against AD. Firstly, some of compounds may have side effects for patients. For example, metal chelators without selectivity might lead to brain impairment via removing essential metal ions [61]. Poor BBB permeability is another obstacle for targeting metal dyshomeostasis against AD. Some compounds are hydrophilic and might exhibit good therapeutic effects in vitro or in cellulo, but they have almost no activity in vivo due to their inability to cross the BBB [62]. Therefore, developing efficient technologies to deliver drugs across the BBB are also urgently needed [125]. Recently, several compounds in clinical trials





**Fig. 8** Multifunctional inhibitors against AD

against AD have been declared failures, which suggest that once AD has entered certain stage, it may be irreversible and cannot be repaired by targeting metal dyshomeostasis [126, 127]. These failures have prompted some researchers turn to AD prevention in the early stages of disease [126, 127]. Last but not least, AD is a multifactor disease, and targeting metal dyshomeostasis via a single pathway alone may not be a guarantee of efficacy in the clinic [128]. Currently, the first diagnosis and therapy for AD patients used separate therapeutic and diagnostic agents, which may make patients miss the optimal therapeutic time window and greatly reduced the efficacy.

### Concluding remarks

Although more than a century has elapsed since the first diagnosis of AD, the development of practical treatments for AD is still difficult [129]. Metal ions play critical roles in various critical neurological processes and, therefore, an aberration in their homeostasis can have catastrophic

consequences [130]. However, according to the current research, metal ion dyshomeostasis not only leads to neurotoxicity and A $\beta$  aggregation but can also lead to changes in apoptosis [49] and autophagy [54–56], mitochondrial dysfunction [57, 58], and lysosomal storage disorders [59–61] in neurons. Elucidating the precise mechanisms in which metal homeostasis is affected in each disease of interest is central to the development of new pharmacological agents. It is also extremely critical to appreciate the different functions of various cell types in regulating metal homeostasis, in order to be able to direct the therapeutic modalities to the appropriate region. [131]. Moreover, more effort should be devoted to reversing the onset of AD at an early stage, since the neurological damage caused by AD is irreversible once the disease enters into late stage [126, 127]. Another avenue of investigation could be the combination of diagnostic and therapeutic functions into a single modality, known as a “theranostic”, in order to reduce side effects and potential drug–drug interactions [132]. Finally, although extensive research has implicated the involvement of metal ions in AD, their precise mechanism in the neuropathogenesis of

the disease is still unclear [133]. In this context, rebalancing metal homeostasis via developing multifunctional agents has shown great potential in both animal models of AD and in early-stage clinical trials.

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## Compliance with ethical standards

**Conflict of interest** Authors declare that there are no conflicts of interest.

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