# ORIGINAL ARTICLE

# Biocatalysts for multicomponent Biginelli reaction: bovine serum albumin triggered waste-free synthesis of 3,4-dihydropyrimidin-2-(1H)-ones

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Abstract Bovine serum albumin (BSA) promoted simple and efficient one-pot procedure was developed for the direct synthesis of  $3,4$ -dihydropyrimidin-2(1H)-ones including potent mitotic kinesin Eg5 inhibitor monastrol under mild reaction conditions. The catalyst recyclability and gram scale synthesis have also been demonstrated to enhance the practical utility of process.

Keywords Biginelli condensation - Bovine Serum Albumin - Biocatalysis - Sustainable Chemistry

#### Abbreviations



# Introduction

The considerable interest in employing environmentally acceptable processes has motivated the use of biocatalysts for the production of different chemical structures. Thus,

many researchers are utilizing these natural catalysts for the smart insertion of chirality and selectivity with regard to regio- and stereo-chemistry for producing optically active, higher value molecules under milder reaction conditions (Santacoloma et al. [2011;](#page-5-0) Hudlicky and Reed [2009](#page-5-0); Pollard and Woodley [2007](#page-5-0); Schmid et al. [2001](#page-5-0); Clouthier and Pelletier [2012](#page-4-0); Patel [2008;](#page-5-0) Van Rantwijk and Sheldon [2007](#page-6-0)).The Biginelli reaction (Biginelli [1893\)](#page-4-0) involving condensation of aldehyde,  $\beta$ -ketoester and urea ranks as one of the most recognized multicomponent reaction (Dömling  $2006$ ; Zhu and Bienaymé  $2005$ ; Wipf et al.  $2003$ ; Biggs-Houck et al. [2010\)](#page-4-0) and widely employed reaction for the preparation of  $3,4$ -dihydropyrimidin-2- $(1H)$ -ones (DHPMs). DHPM derivatives exhibit a wide range of biological and pharmacological properties such as antiviral, antimitotic, anticarcinogenic, antihypertensive and most importantly, as calcium channel modulators (Kappe [1998,](#page-5-0) [2000](#page-5-0); Yarim et al. [2003;](#page-6-0) Kidwai et al. [2005;](#page-5-0) Kumar et al. [2009](#page-5-0); Jain et al. [2008](#page-5-0); Mayer et al. [1999](#page-5-0)). Additionally, the biological activity of potent HIV gp-120-CD4 inhibitor Batzelladine A and B has also been attributed to DHPM moiety (Hojati et al. [2010\)](#page-5-0).

In the past decade, a series of procedures has been developed (Gore et al. [2011](#page-4-0); Ramalingan et al. [2010;](#page-5-0) Shen et al. [2010;](#page-5-0) Tamaddon et al. [2010](#page-5-0); Chitra and Pandiarajan [2009](#page-4-0); Lannou et al. [2008](#page-5-0); Chari et al. [2009](#page-4-0); Suzuki et al. [2008](#page-5-0); Polshettiwar and Varma [2007;](#page-5-0) Yu et al. [2007](#page-6-0); Debache et al. [2006,](#page-4-0) [2008](#page-4-0); Bhosale et al. [2004;](#page-4-0) Rafiee and Jafari [2006](#page-5-0); Banik et al. [2007](#page-4-0); Bose et al. [2004;](#page-4-0) Li et al. [2003](#page-5-0); Saha and Moorthy [2011\)](#page-5-0) to overcome the relatively harsh acidic conditions of the original Biginelli reaction. However, in spite of their potential utility, most of these methods involve expensive reagents, stoichiometric amount of catalyst, longer reaction times, unsatisfactory yields and generation of waste materials besides

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	CH <sub>O</sub> O Et <sub>O</sub>	CH <sub>3</sub> $H_2N$	EtO <sup>-</sup> NH biocatalyst (50 mg) NH <sub>2</sub> 6 days		
	1a	$\mathbf{3}$ $\boldsymbol{2}$		$\mathbf H$ 1 <sub>b</sub>	
S. no.	Enzyme	Solvent	Temperature (°C)	Yield $(\%)^a$	ee $(\%)$
1	$CAL-B$	EtOH	28	28	5 <sup>5</sup>
$\overline{c}$	$\ensuremath{\textup{CCL}}\xspace$	EtOH	28	24	$<$ 5
3	<b>PPL</b>	EtOH	28	28	$<$ 5
4	$\text{MJL}$	EtOH	28	12	6
5	<b>TLL</b>	EtOH	28	15	$<$ 5
6	<b>CRL</b>	EtOH	28	15	6
7	$CAL-B$	$CH_2Cl_2$	28	24	5
8	$CAL-B$	$C_6H_5CH_3$	28	19	6
9	$CAL-B$	$H_2O$	28	15	$<$ 5
10	$CAL-B$	<b>ACN</b>	28	18	5
11	$CAL-B$	<b>THF</b>	28	10	6
12	$CAL-B$	<b>DMSO</b>	28	23	$\leq$
13	$CAL-B$	MeOH	28	26	$\leq$
14	$CAL-B$	EtOH	40	44	5 <sup>5</sup>
15	$CAL-B$	EtOH	60	62	$\leq$
16	Denatured CAL-B <sup>b</sup>	EtOH	60	52	$<$ 5
17	<b>BSA</b>	EtOH	60	80	$<$ 5
18	Protein albumin	EtOH	60	77	
19	Egg albumin	EtOH	60	74	
20	Denatured BSA <sup>b</sup>	EtOH	60	28	
21	Acetylated BSA	EtOH	60	7	

<span id="page-1-0"></span>Table 1 Screening of different enzymes/proteins and solvents for asymmetric Biginelli reaction

Experimental conditions: 0.25 mmol 1a, 1 equiv. ethyl acetoacetate, 1 equiv. urea, 3 mL solvent

CAL-B Candida antarctica lipase B, CCL C. cylindracea lipase, PPL Porcine pancreas lipase, MJL Mucor javanicus lipase, TLL Thermomyces lanuginosus lipase, CRL C. rugosa lipase

<sup>a</sup> On the basis of HPLC

 $<sup>b</sup>$  Pre-treated with urea and thiourea (7:1) at 100oC for 2 h</sup>

cumbersome product isolation procedures and incompatibility with certain functional groups, thus are not closer to the principles of green chemistry. We envisioned that an improvement addressing these issues might be achieved by the use of a biocatalyst particularly for the asymmetric version.

Exploiting the promiscuous activity of a biocatalyst to achieve multicomponent reaction has always remained challenging (Znabet et al. [2010a](#page-6-0), [b;](#page-6-0) Wang et al. [2011](#page-6-0)) and in this case too, only few reports are there where enzymatic systems have been used for the resolution of DHPM esters (Prasad et al. [2009;](#page-5-0) Schnell et al. [2000\)](#page-5-0) or towards the synthesis of racemic DHPMs (Borse et al. [2012](#page-4-0); Kumar and Maurya [2007](#page-5-0)). Following our recent report on role of amino acid based ionic liquid (Sharma et al. [2012a](#page-5-0)) in DHPMs synthesis and exploitation of biocatalyst for various organic transformations (Sharma et al. [2009](#page-5-0), [2011a](#page-5-0), [b,](#page-5-0) [2012b](#page-5-0); Kasana et al. [2007](#page-5-0)), we targeted the waste-free synthesis of DHPMs involving a biocatalytic procedure where water ought to be the only by-product.

Initially, lipases (Table 1; entries 1–6) were screened for obtaining enantiomeric pure DHPM (1b) by cyclocondensation of 0.25 mmol of benzaldehyde (1a), ethyl acetoacetate (2) (1 equiv.) and urea (3) (1 equiv.) in EtOH at 28 °C for 6 days. Among all, Candida antarctica lipase-B (CAL-B) and porcine pancreas lipase (PPL) provided 1b in 28 % yields (entries 1 and 3) with inferior enantioselectivity (5 % ee) at room temperature. Even the use of <span id="page-2-0"></span>Table 2 Optimization of BSA catalyzed Biginelli reaction

$$
C_6H_5
$$
  $CHO$  + EtO  $CH_3 + H_2N$   $NH_2$   $BSA$  (25-100mg) EtO NHH  
solvent, 40-80°C



Experimental conditions: 0.25 mmol 1a, 1 equiv. ethyl acetoacetate, 1 equiv. urea, 3 mL solvent

<sup>a</sup> Isolated yield

Table 3 Substrate scope of BSA catalysed Biginelli reaction

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	$R$ <sup>-CHO</sup> $+$ $R'$ O a	X O $CH_3 +$ $H_2N$ NH <sub>2</sub> $X = O$ or S	BSA (50mg) EtOH, 60°C	R'O' <b>NH</b> Χ $\mathbf b$ H					
S. no.	$\mathbb{R}$	R	$\mathbf X$	Time (h)	Yield $(\%)^a$				
	$C_6H_5$	Et	$\mathbf{O}$	8	83				
2	$4$ -OMe $C_6H_4$	Et	$\Omega$	8	78				
3	$2,4,5-(OMe)3C6H2$	Et	$\mathbf{O}$	8	73				
4	$3,4-(-OCH2O-$ $C_6H_3$	Et	$\mathbf{O}$	8	76				
5	$3$ -OMe, $4$ -OHC $_6$ H <sub>3</sub>	Et	$\mathbf{O}$	8	$72\,$				
6	$4-OHC6H4$	Et	$\mathbf{O}$	8	70				
7	$3$ -OHC $_6$ H <sub>4</sub>	Et	$\mathbf{O}$	8	69				
8	$4$ -ClC $6H_4$	Et	$\Omega$	8	79				
9	$3-{\rm BrC}_6{\rm H}_4$	Et	$\mathbf{O}$	8	80				
10	$4-NO2C6H4$	Et	$\Omega$	8	78				
11	$\rm C_{10}H_7$	Et	$\mathbf{O}$	8	74				
12	$4-N, N(CH_3)_2C_6H_4$	Et	$\Omega$	8	76				
13	$4-MeC6H4$	Et	$\Omega$	8	78				
14	$C_6H_5$	CH <sub>3</sub>	$\Omega$	8	76				
15	$C_6H_5$	$t$ -Bu	$\Omega$	8	75				
16	$C_6H_5$	Et	S	12	$72\,$				
17	$3\mbox{-}\mathrm{OHC}_6\mathrm{H}_4$	Et	S	12	$70\,$				

Experimental conditions: 0.25 mmol of substituted benzaldehyde, 1 equiv. dicarbonyl compound, 1 equiv. urea or thiourea, in 3 mL of EtOH, BSA 50 mg <sup>a</sup> Isolated yield (after recrystallization). All compounds have been characterized by NMR and HRMS data (available in the supporting information)

 $C_6H_5$ 

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<span id="page-3-0"></span>Scheme 1 Plausible reaction mechanism for BSA catalysed Biginelli reaction





Scheme 2 BSA catalysed gram scale synthesis of bioactive monastrol (17b)

different solvents with CAL-B as catalyst, did not improve the yield and % ee of chiral **[1](#page-1-0)b** (Table 1; entries  $7-13$ ). Thereafter, the effect of temperature was probed to stir the reaction in desired direction. It was observed that though increase in temperature from 28  $\degree$ C to 60  $\degree$ C brought about a linear change on the yield (62 %), yet it could not exert any positive effect on % ee (Table [1](#page-1-0); entry 1 vs 14, 15). In order to exclude background activity, we decided to perform some control experiments where denatured CAL-B or bovine serum albumin (BSA) was used to mediate the desired transformation (Table [1;](#page-1-0) entry 16, 17).

Surprisingly, 52 % conversion was reached with denatured CAL-B (entry 16) and a very high conversion value (80 %) using BSA (entry 17). With these results in hand (lack of enantioselectivity together with considerable yield observed in case of both denatured lipase and BSA), we hypothesized the role of protein catalysis instead of specific promiscuous catalysis by enzymes which is in contrast to some recent reports (Borse et al. [2012;](#page-4-0) Kumar and Maurya [2007\)](#page-5-0) where the possibility of protein catalysis has not been considered. Moreover, in comparison to enzyme CAL-B (Walker [2005\)](#page-6-0), BSA have a rich diversity of surface amino acids which might account for the enhanced yield of 1b. Replacing the BSA with porcine albumin and egg albumin also provided the desired DHPM (1b) in 77 % and 74 % yield (entries 18, 19), respectively whereas denatured BSA (Norberto et al. [2012\)](#page-5-0) provided 1b in 28 %

yield (entry 20).To validate the above hypothesis, further optimization of reaction conditions using BSA as catalyst was carried out. The effect of solvent, temperature, reaction time and the amount of catalyst on the yield of 1b was examined (Table [2](#page-2-0)). Best results were obtained when the reaction of 1a was carried out with 50 mg BSA in EtOH at 60 °C with shorter reaction time of 8 h in 82  $%$  yield (Table [2,](#page-2-0) entry13).

Afterwards, to extend the substrate scope of BSA triggered Biginelli reaction, various benzaldehydes 1a–15a were treated with urea and 1,3 dicarbonyl compounds for the synthesis of dihydropyrimidones 1b–15b (Table [3,](#page-2-0) entries 1–15). All the obtained products were characterized by NMR and HRMS data (see supporting information). Thiourea was also used with similar success to provide the corresponding dihydropyrimidine- $(2H)$ -thiones (Table [3,](#page-2-0) entries 16, 17), which are also of much interest with regard to their biological activity (e.g. monastrol Table [3,](#page-2-0) entry 17). As shown in Table  $3$  (entries 1–17), yields of this onepot protocol following recrystallization from ethanol were of the order 69–83 %, which is quite favourable. In particular, the catalyst exhibited remarkable activity for Cl, Br, OH,  $NO<sub>2</sub>$  and OCH<sub>3</sub> functional group containing compounds. In order to check the standard deviation (SD), three independent experiments were conducted for the substrate with electron neutral, withdrawing and releasing groups (Table [3](#page-2-0), entries 1, 10 and 13) where deviation in the yield of the products were  $83 \pm 3$ ,  $78 \pm 5$  and  $78 \pm 2$ percent, respectively.

A plausible mechanism (Scheme 1) suggested the probable role of amino acid side chain in BSA as the catalytic base for the reaction where basic character of the amino group might be responsible for the catalysis as per prior reports (Taylor et al. [1975](#page-5-0); Riva et al. [1998](#page-5-0); Hollfelder et al. [1996;](#page-5-0) Boucher et al. [2005](#page-4-0); Klein and Reymond [1998](#page-5-0); Reetz et al. [2007;](#page-5-0) Strohmeier et al. [2009\)](#page-5-0). Firstly, the urea attacks on the aldehyde group of substrate leading to

<span id="page-4-0"></span>the formation of an iminium ion (Kappe [1997;](#page-5-0) De Souza et al. 2009) followed by the removal of proton from active methylene group of ethylacetoacetate (Sharma et al. [2011b\)](#page-5-0) by the free basic amino group of BSA which subsequently condensed with iminium ion to give desired product as shown in scheme [1](#page-3-0). As expected, use of acetylated BSA drastically reduced the formation of 1b in 7 % yield only (Table [1,](#page-1-0) entry 21) which clearly indicates the importance of free amino group present in BSA indispensable for synthesis of DHPM.

Subsequently, we studied the recyclability of the BSA for the above reaction. Complete conversion was observed up to third reaction cycle with benzaldehyde (1a), ethyl acetoacetate (2) and urea (3) as reaction substrates. However, slight loss of activity (74 %) was observed after fourth catalytic cycle onwards. Further, to demonstrate the practical applicability of the developed method, preparative scale reaction (1 g batch) of 3-hydroxybenzaldehyde (17a) with ethyl acetoacetate and thiourea was effectively accomplished using BSA as catalyst leading to the formation of a potent mitotic kinesin Eg5 inhibitor monastrol (Mayer et al. [1999\)](#page-5-0) (17b) in good yield (Scheme [2](#page-3-0)).

#### Conclusion

In conclusion, we have successfully developed a simple, inexpensive and waste-free methodology for the synthesis of bioactive 3,4-dihydropyrimidin-2(1H)-ones using "offthe shelf'' protein- bovine serum albumin (BSA) while exploring the promiscuous lipase catalysed system for the asymmetric version of Biginelli reaction. Moreover, catalyst recyclability and gram scale synthesis have also been demonstrated to enhance the practical utility of process.

#### Experimental Section

General Procedure for the synthesis of 3,4 dihydropyrimidin-2(1H)-ones  $(1b-17b)$ from substituted benzaldehydes (Table [3](#page-2-0), entries 1–17 and Scheme [2\)](#page-3-0)

Substituted benzaldehyde (1a–17a, 0.25 mmol), ethyl acetoacetate (1 equiv.), urea or thiourea (1 equiv.) and BSA (50 mg) was taken in 3 mL ethanol in a round bottom flask and the reaction mixture was incubated in a orbital shaker at 60  $\degree$ C for 8–12 h. Afterwards, the reaction mixture was worked up and analyzed with the HPLC. Further, the crude product was purified by recrystallization from water–ethanol mixture giving an isolated yield of 1b–17b in the range of 69–83 %. <sup>1</sup>H and <sup>13</sup>C NMR spectra were recorded

and matched with reported values and further confirmed by HRMS/MS.

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