The origin of rutile-ilmenite aggregates ("nigrine") in alluvial-fluvial placers of the Hagendorf pegmatite province, NE Bavaria, Germany

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Summary

Titanium placer deposits occur in alluvial-fluvial drainage systems which dissect Moldanubian gneisses intruded by Late Variscan pegmatites (Hagendorf province) in southern Germany. Based upon their texture (zonation, exsolution lamellae, intergrowth), microchemical data (Nb, Cr, Ta, V, Fe, W, Sn) and mineral inclusions, two major grain types of intergrown rutile and ilmenite have been established. Grains of type A are always zoned and consist of rutile cores enveloped by ilmenite containing small inclusions of wolframite. A core-rim transition zone is characterized by complex relations of rutile and ilmenite, with rutile lamellae being rich in Nb, V and Fe. Types B1 and B2 aggregates consist of ilmenite with lamellae of niobian rutile and/or ilmenorutile, and additionally have inclusions of ferrocolumbite, pyrochlore, betafite, sphalerite, pyrrhotite and Fe oxides. Such grain types featuring an intimate intergrowth of rutile and ilmenite were called nigrine. Type-C grains are quite similar in their morphological appearance but consist of W-enriched rutile devoid of mineral inclusions and reaction products. Pseudorutile and leucoxene replacing minerals of the nigrine aggregates are presumably caused by supergene alteration under fluctuating redox conditions. Phosphate and aluminum remobilized by supergene processes led to the formation of hydrous Ti-rich phases containing Al, P and Fe. High Nb and W concentrations in nigrine aggregates and in rutile type C may be taken as a marker for highly differentiated granites or pegmatites. This has implications for both, heavy-mineral-based provenance analysis and stream sediment exploration.

Introduction

According to *Clark* (1993), "nigrine" (also: Eisenrutil, iron-rutile) is a variety of rutile. Nigrine is described either as black ferruginous rutile that is homogeneous

on a microscopic scale (*Ramdohr*, 1975), or as ilmenite pseudomorphosed by rutile (*Strübel* and *Zimmer*, 1982). The Latin word "nigrum" connotes a black mineral color and immediately directs thoughts to the most striking feature common to all these Ti heavy mineral aggregates, the dull black luster, by which they may easily be recognized in stream sediments. To avoid confusion, in the present study the term nigrine is used, without placing it in inverted commas, to describe aggregates of various Ti oxide minerals, including rutile and ilmenite, intimately intergrown with each other. The term is used irrespective of their composition and the quantity.

From the Pleystein area in NE Bavaria, Germany (Fig. 1), nigrine forming grains up to 2.5 cm in diameter was reported from alluvial-fluvial placer deposits by *Strunz* in 1961 (Fig. 2a–d). As he could not avail of any detailed analyses then, the author suspected cassiterite and rutile to be the main constituents of the placer deposits. Complex Ti oxide aggregates like those from SE Germany were also recorded from placer deposits elsewhere in the world, e.g., from Sierra Leone by *Raufuss* (1973) and from Malawi by *Bloomfield* (1958) and *Coakley* and *Mobbs* (2001). None of the papers quoted above reports on the complex nature of the placer mineral aggregates. Textural relationships, such as exsolution lamellae, grain size, intergrowth textures, mineral inclusions, the concentrations of certain trace elements, e.g., Cr, V and Nb, may fingerprint the formation of nigrine and whether it might be feasible for an industrial use (*Chernet*, 1999a, b). Information



Fig. 1. Sketch map showing (**a**) the position of the study area in NE Bavaria, Germany and (**b**) the regional geology (modified after *Forster*, 1965). Sampling sites: *1* Steininger Loh, *2* Zollgraben, *3* Pingermühle, *4* Pflaumbach, *5* Weißenstein

Sample Location	7134/1 Pingermühle (Zotthach)	7193/4 Zollgraben	7134/2 Pingermühle (Zotthach)	7194/4 Iglasreuth Tirschenreuth	7135 Pleystein (hedrock)	7193/3 Zollgraben	7134/2 Pingermühle (Zotthach)	7194/1 Iglasreuth Tirschenreuth	7194/4 Iglasreuth Tirschenreuth	7167 Pleystein	7134/1 Pingermühle (Zotthach)	7193/4 Zollgraben	7194/3 Iglasreuth Tirschenreuth
Analysis# Grain type	42 A	53 A	6 B1 B1	70 B2	A A A	12 A metro (Mb)	5 B1	B2 MHA	65 B2	6 C	47 A	34 1 A	36 B2
Mineral	umenue	umente	umenue	Ilmenite	runie (IND)	ruule (IND)	rume (ND)	rune (ND)	rune (ND)	rutile (w)	ш	ц	ш
TiO_2	52.38	52.85	49.30	51.29	89.39	87.78	85.53	66.87	95.70	93.22	98.17	98.19	98.80
Fe_2O_3	0.67	0.40	6.27	1.33									
FeO	45.33	44.95	41.71	43.48	2.79	2.94	4.20	6.85	0.65	2.07	0.30	0.24	0.38
MnO	1.17	2.21	2.34	2.16	bdl	0.07	0.03	0.02	bdl	lbdl	bdl	lbdl	0.04
Nb_2O_5	0.19	0.09	0.28	0.20	4.55	6.49	7.54	20.76	1.87	0.30	0.79	0.21	0.66
Ta_2O_5	lbdl	0.04	lbdl	0.04	0.39	0.38	0.56	1.18	0.03	lbdl	bdl	lbdl	0.04
Sc_2O_3	lbdl	bdl	0.02	bdl	0.03	lbd	lbdl	lbdl	0.02	0.02	bdl	lbdl	lbdl
V_2O_3	0.18	0.27	0.29	0.34	1.38	1.38	0.68	2.62	0.95	1.06	0.91	0.69	0.45
Cr_2O_3	0.03	0.02	lbdl	0.03	0.25	0.44	bdl	0.51	0.17	0.04	0.08	0.07	bdl
Al_2O_3	lbd	lbdl	lbd	lbdl	0.11	0.13	0.05	0.20	0.12	lbd	0.04	0.03	lbd
MgO	0.46	0.24	0.08	0.37	lbdl	0.02	bdl	lbdl	lbdl	lbdl	bdl	0.04	bdl
SiO_2	0.03	0.05	0.05	lbdl	bdl	lbdl	0.02	0.03	0.02	0.03	0.02	lþd	bdl
NiO	lþd	lbdl	0.02	lbdl	lbdl	lbdl	bdl	lbdl	lþd	0.02	bdl	lbd	lbd
SnO	lbdl	lbdl	lbd	lbdl	bdl	lbdl	0.11	bdl	bdl	lbdl	0.03	lþd	bdl
WO ₃	lbdl	lbd	lbd	bdl	1.37	0.50	0.31	1.33	0.14	4.87	0.04	lbdl	lbd
PbO	lbdl	lbdl	lbdl	bdl	lbdl	lbdl	0.19	lbd	bdl	0.11	bdl	lbdl	lbdl
Total	100.45	101.12	100.35	99.24	100.26	100.13	99.22	100.38	99.66	101.71	100.37	99.46	100.37
Ξ	1.974	1.983	1.875	1.961	1.848	1.821	1.806	1.480	1.941	1.900	1.965	1.980	1.977
Fe^{3+}	0.026	0.014	0.239	0.050									
Fe^{2+}	1.900	1.874	1.758	1.848	0.064	0.068	0.099	0.169	0.015	0.047	0.007	0.005	0.009
Mn	0.050	0.094	0.100	0.092		0.002	0.001	0.000					0.001
ЧN	0.004	0.002	0.007	0.005	0.057	0.081	0.096	0.276	0.023	0.004	0.009	0.003	0.008
Ta		0.001		0.001	0.003	0.003	0.004	0.009	0.000	0000			0.000
Sc	0000	0.00	0.001		0.001		1.000		0.000	0.000	0100	1000	0100
> č	0.000	010.0	0.012	0.014	0.030	0.031	c10.0	200.0	0.004	0.001	0.002	CIU.U	010.0
AI AI	100.0	1000		100.0	0.000	0.010	0.000	0.007	0.004	100.0	0.001	0.001	
Mg	0.034	0.018	0.006	0.028		0.001			-			0.001	
Si	0.001	0.003	0.002				0.001	0.001	0.000	0.001	0.001		
Ņ			0.001							0.000			
Sn							0.001				0.000		
₩ A					0.010	0.004	0.002	0.010	0.001	0.034 0.001	0.000		
1													
Total Cat	4.000	4.000	4.000	4.000	2.021	2.023	2.028	2.027	2.008	2.011	2.005	2.007	2.004

ucts	th reuth		utile																												(pənı
on prod	7194 Iglasreut Tirschen	B2 9	pseudori	lbdl	0.15	0.03	lbd	lbdl	Ipq			70.10	0.29 1541	ош 3.64	33.29	pql	0.23	lbdl	lbdl	bdl	lþd	lbd	95.37		0.015	0.003					(contin
of alterati	7193 Zollgraben	A 26	pseudorutile	lþd	0.22	0.09	0.03	bdl	Ipq		bdl 50.01	10.90	0.52 164	001 2,44	34.70	pqI	0.23	bdl	bdl	bdl	lþd	lbd	97.26		0.020	0.007	0.002				
analyses	7134/1 Pingermühle (Zottbach)	A 14	pseudorutile	lþd	0.52	0.07	0.03	lþd	lbdi			0C./C	0.02	cu.u 1 14	37,00	bdl	0.25	bdl	bdl	bdl	lþd	lbd	97.05		0.051	0.005	0.002				
nd average	7194/3 Iglasreuth Tirschenreuth 43-45	B2 3	leucoxene?	lþd	0.12	6.09	0.71	4.15	bdl î î î î	c0.0	0.32	/0./0	0.02 1-41	0.06	2.00	bdl	0.16	bdl	bdl	bdl	lbdl	lbdl	91.06		0.005	0.206	0.020	0.101	0.002	0.008	
lmenite, a	7193/1 Zollgraben 312–316	S A	leucoxene?	lþd	0.06	8.45	0.49	3.46	0.09	0.28	c0.0	/0.38	01.0	61.0	2.54	bdl	0.15	bdl	bdl	bdl	lþd	lþd	92.88		0.002	0.280	0.014	0.082 0.002	0.008	0.001	
utile and i	7134/2 Pingermühle (Zottbach) 13	B1	magnetite		0.03	0.02						5.22	1.0	0.00 0.00	90.12								93.67		0.013	0.007					
ncluded in 1	7134/2 Pingermühle (Zottbach) 35	Bl	altered betafite	0.08	bdl	2.23	4.68	3.02	0.07	0.30	0.04	20.87	0.18	0.00	4.07	bdl	42.72	0.27	13.61	bdl	0.99	3.48	96.85	0.010		0.160	0.285	0.156 0.004	0.024	0.002	
al phases i	7134/2 Pingermühle (Zottbach) 34	B1 2	betafite	1.29	bdl	0.10	0.09	lbdl	bdl	86.CI		0.20	0.00	0 09	0.68	bdl	32.14	0.29	4.28	0.60	1.00	29.11	102.42	0.178		0.008	0.007		1.178		
es of miner	7134/2 Pingermühle (Zottbach) 9	B1	pyrochlore	bdl	lþdl	0.03	0.29	lbdl	bdl	16./1	0.03	21.0	0.04	0.05	1.37	0.03	74.13	0.05	2.27	0.40	0.30	0.18	102.39			0.002	0.015		1.002	0.001	
obe analys	7194/1 Iglasreuth Tirschenreuth 9	B2	wolframite	lbdl	0.58	0.05	lbd	lbdl	lbd 		lbdl		0.00	0.09 1 99	21.20	bdl	6.51	lbdl	lbdl	62.89	lbd	lbd	94.37		0.044	0.003					
on micropi	7134/1 Pingermühle (Zottbach) 246	A	wolframite	lþd	0.98	lbdl	lbdl	lbdl	lbdi		0.15 11-1		D01	1 72	24.28	pdl	0.66	lbdl	lþdl	70.43	lþd	lbd	98.73		0.073					0.007	
ative electr	7194/1 Iglasreuth Tirschenreuth 10	B2	columbite	lbdl	0.50	bdl	0.06	bdl	bdl 2020	0.83	6 10	8.19	0.05	2.43	17.93	lpd	64.55	bdl	2.22	4.71	lþd	lbd	101.75		0.163		0.013		0.194		
Represent	7134/2 Pingermühle (Zottbach) 17	B1	columbite	lþd	0.26	0.03	0.10	lbdl	bdl 0.20	0.00	1.07	06.0	0.04 1541	041 3 54	16.89	bdl	69.76	0.14	2.58	0.69	lbd	lbd	101.72		0.084	0.00	0.021		0.141	0.204	
Table 2.	Sample Location Analysis#	Grain type N	Mineral	Na_2O	MgO	Al_2O_3	SiO_2	P_2O_5	$\frac{SO_2}{2}$	cau	Sc ₂ U ₃	11O ₂	V203 C= 03 C= 03 C C= 03 C C C= 03 C C C C C C C C C C C C C C C C C C C	CI2O3 MnO	FeO (tot)	NiO	Nb_2O_5	SnO	Ta_2O_5	WO_3	PbO	UO_2	Sum	Na	Mg	Al	Si	ч х 2	Ca	Sc	l

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Sample Location Analvsis#	7134/2 Pingermühle (Zottbach) 17	7194/1 Iglasreuth Tirschenreuth 10	7134/1 Pingermühle (Zottbach) 246	7194/1 Iglasreuth Tirschenreuth 9	7134/2 Pingermühle (Zottbach) 9	7134/2 Pingermühle (Zottbach) 34	7134/2 Pingermühle (Zottbach) 35	7134/2 Pingermühle (Zottbach) 13	7193/1 Zollgraben 312–316	7194/3 Iglasreuth Tirschenreuth 43-45	7134/1 Pingermühle (Zottbach)	7193 Zollgraben	7194 Iglasreuth Tirschenreuth
Grain type N	B1	B2	A	B2	B1	B1 2	Bl	Bl	5 A	B2 3	A 14	A 26	B2 9
Mineral	columbite	columbite	wolframite	wolframite	pyrochlore	betafite	altered betafite	magnetite	leucoxene?	leucoxene?	pseudorutile	pseudorutile	pseudorutile
Ti	0.976	1.348			0.201	0.928	0.956	0.745	1.618	1.651	2.832	2.907	2.895
>	0.007	0.039		0.012	0.002	0.005	0.00	0.041	0.012	0.014	0.015	0.017	0.015
Cr		0.009		0.003	0.001		0.003	0.006	0.004		0.001		
Mn	0.653	0.451	0.073	0.086	0.007	0.005	0.005	0.024	0.001	0.002	0.063	0.138	0.206
Fe^{3+}								14.439			2.024	1.875	1.859
Fe ²⁺ Ni	3.078	3.282	1.019	0.909	0.060	0.040	0.207	8.725	0.060	0.048	0.000	0.027	0.000
Nb	6.873	6.387	0.015	0.151	1.750	1.039	1.176		0.002	0.002	0.007	0.006	0.007
Sn	0.014				0.001	0.00	0.007						
Ta	0.153	0.132			0.032	0.083	0.225						
M	0.039	0.267	0.916	0.836	0.005	0.011							
Pb					0.004	0.019	0.016						
U					0.002	0.463	0.047						
Total Cat	12.265	12.303	2.124	2.087	3.084	3.510	3.245	24.000	2.090	2.060	5.000	5.000	5.000
Oxygen	24	24	4	4	9	7	7	32	4	4	6	6	6
<i>bdl</i> Belc High an	ow detectic alytical tot	on limits; N tals in small	number of grains are	analyses u due to exc	sed to calc itation fro	culate aver m ilmenite	age e						

Table 2 (continued)

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contained in nigrine from alluvial-fluvial placers is also of importance for sedimentologists for tracking back these heavy minerals to their provenance (*Morton*, 1984, 1985; *Dill*, 1998; *Zack* et al., 2002, 2004). The scope of the present study is to give an answer to these questions by providing mineralogical and chemical information on a detrital Ti mineralization genetically related to the emplacement of pegmatites.

Methodology

The coarse-grained "black sands" may be recovered by simple grab sampling and panning (Fig. 2c, d). The heavy minerals (density $>2.9 \text{ g/cm}^3$) were extracted from the 63-200 µm fractions by Na-polywolframate, a non-toxic heavy liquid. Grains mounted on a glass disk were identified and quantified under the petrographic microscope counting between 200 and 300 grains per sample. Examination of thin sections and X-ray diffraction analysis of particles larger than 200 µm supplemented the petrographic studies. Emphasis was placed on the study of polished sections under reflected light, in combination with electron microprobe analysis to identify the textural features and chemical composition, respectively. Electron microprobe analyses were carried out using a CAMECA SX100 equipped with five wavelength-dispersive spectrometers and a Princeton Gamma Tech energydispersive system. Oxide and silicate phases were analyzed at 20 kV acceleration voltage and 20 nA sample current (on brass). Natural minerals (albite, chromite, kaersutite, almandine, apatite, magnetite, pentlandite, biotite, rutile, rhodonite, galena) and pure metals were used as standards. Representative microprobe analyses are presented in Tables 1 and 2. Calculation of Fe³⁺/Fe²⁺ in ilmenite was done using the method of Droop (1987).

Geological setting

Forster (1965) gave a first comprehensive picture of the geological setting in this part of the NE Bavarian Basement (Fig. 1). The major part of the study area is underlain by Moldanubian paragneisses composed of variable amounts of biotite, sillimanite, cordierite, quartz, garnet, and feldspar. Second in abundance are felsic igneous rocks of Late Carboniferous age, including aplites and pegmatites of different grain size and mineral composition. These igneous rocks are mainly located in the center and at the eastern edge of the study area. Along the western, southern and northern margins of this cluster of igneous rocks, swarms of quartz veins intersect the basement gneisses in NW-SE direction. Their morphological expression in the field, texture and mineralogy are similar to those of the "Great Bavarian

Fig. 2. Titanium placer minerals in continental environments – past and present **a** Subangular nigrine intergrown with quartz (nigrine type A). Location: Pflaumbach near Pleystein. **b** Angular complex ilmenite-rutile intergrowth of nigrine type A. Location: Pflaumbach near Pleystein. **c** Bed load deposit of subrounded grains of nigrine (black) disseminated between gravel of vein quartz (bright) and gneiss (dark gray) in a creek near Pleystein. **d** Subrounded to rounded grains of nigrine concentrated manually by panning in the field. Location near Pleystein





Fig. 2 (continued)

Quartz Reef" (*Siebel* et al., 2005). Quartz veins are absent from the area intruded by the Late Variscan granites and shaped by the Cenozoic drainage system that gave host to the Ti placers under consideration.

Results

Mineralogy and textures of nigrine

In the following, three types of nigrine (A, B1, B2) found in placer deposits of NE Bavaria are presented (Fig. 2a–d). Type A comprises zoned grains consisting of a rutile core (host) and an ilmenite rim. Inclusions of wolframite are common. Rutile and ilmenite are intergrown crystallographically as flame-like ilmenite exsolutions in rutile and vice versa in a core-rim transition zone. Type B comprises grains of ilmenite hosting lamellae of niobian rutile. In subtype B1 rutile has exsolved Nb-bearing rutile lamellae, and the ilmenite has abundant hematite exsolutions and inclusions of ferrocolumbite and pyrochlore. In subtype B2, Nb-, Fe- and V-rich rutile exsolves directly from ilmenite. Another grain type, rutile C, is monomineralic but W-rich, lacking the complex intergrowth with ilmenite and any mineral inclusions observed in types A and B. Their characteristic textural and chemical peculiarities are summarized in Table 3.

Nigrine type A: Nigrine type A occurs in the placers as well rounded to rounded grains exceeding locally 10 mm in size. Optical examination under reflected light and the use of backscattered-electron images (BSE) reveals a complex intergrowth of twinned rutile which forms the core of the aggregates, with ilmenite forming a rim, up to 1.5 mm thick (Figs. 3a, b, 4a). Oriented lamellae of ilmenite are particularly abundant near the edge of the rutile core, and lamellae of rutile occur in the ilmenite rim. Wolframite and zircon are abundant inclusions and are restricted to the ilmenite rim. Very similar zoned grains of nigrine type A were found in the host rock outcrops around Pleystein (samples 7135, 7136; Table 3). These grains, 4–5 mm in size, are hosted by quartz, muscovite, orthoclase (with quartz-orthoclase symplectites), biotite and albite of much smaller grain size (<1 mm), with accessory Nbbearing rutile, Mn-bearing ilmenite, Fe oxides, apatite, zircon and pyrrhotite.

Chemical profiles across the rutile-ilmenite aggregates show a complex behavior of major and minor elements (Fig. 5), which is most likely controlled by intragrain diffusion processes. In most grains, three different zones are distinguished: a rutile core zone, an inclusion-rich rutile-ilmenite transition zone (several 100 µm thick), and an ilmenite rim. The rutile core is characterized by constantly high TiO_2 (>96 wt.%), up to 1 wt.% V₂O₃ and Nb₂O₅, some FeO (up to 0.7 wt.%), and lower Mn, Mg, Cr, Ta and W. Concentrations of Nb are fairly constant within one grain, e.g., 0.8-1.0 wt.% Nb₂O₅ in the example depicted in Fig. 5 (Table 1, columns 11–12; Fig. 6), but may also display weak continuous zoning with concentrations increasing in the transition zone. Similar behavior is observed for Cr (reaching 0.3 wt.% Cr₂O₃ in the transition zone), V, Mg, W, Sc and Mg, whereas Fe and Nb are lower than in the core (Fig. 7). The ilmenite rim has quite constant Ti/Fe ratios, and carries Mn and Mg as significant impurities. Minor elements present in the rutile core, e.g., Cr, V, and Ta, are low or close to the detection limits in the ilmenite. Scandium (up to $0.04 \text{ wt.}\% \text{ Sc}_2\text{O}_3$) is concentrated in ilmenite compared to rutile. Nb_2O_5 decreases to almost zero in the transition zone and increases gradually to values as high as 0.4 wt.% in the outer ilmenite rim (Table 1, columns 1-2). Iron, V and Mg follow similar enrichment trends. Tiny rutile grains and lamellae in the rutile-ilmenite transition zone may contain up to $7 \text{ wt.}\% \text{ Nb}_2\text{O}_5$ as well as

Alteration	S .'əlinrobuəzt'' ''ənəxcoue.''	x	х	ı, po	x			x	x				
	Ругосћіоге Zircon			spt	х		×		×			x mt	×
Inclusions	hm-lam Wolframite Columbite	x	×	x				x				x	x x
	Nb ₂ O ₅	<0.3	0.1	0.1	0.14	0.1	0.1-0.2	0.05-0.15	0.1 - 0.3	<0.1	0.09	0.22	0.3 - 0.5
	MgO	0.5	0.2	0.3	0.05	0.17	0.15	0.2	0.2-0.3	0.15	0.03	<0.1	0.25
Ilmenite	MnO	1.1	1.7	1.8	4	6	2.5-2.8	2.0-2.4	2.0-3.7	2.3–2.6	ю	2.4	1.5
ins	FeO	0.5-2.9	5	1:1	0.5-2		2.9	0.4	to 4.8	to 1.2		1.0-4.2	6-7
nellae/gra.	V ₂ O ₃	1.0-2.0	1.2	0.8	0.8-1.6		1.4	0.7	to 2.1	to 1.2		0.5–0.7	2.6
Rutile lar	Nb ₂ O ₅	1.1-7.4	3.3	1.1-3.5	1.4–5.8	n.a.	to 6.5	0.8	to 10	to 4		1.5-8.6	18 - 21
	FeO	0.2-6	0.4	0.5	0.3-0.7	0.3 - 0.4	0.4-0.7	0.2-0.3	0.3-0.4	0.2-0.3	0.6	0.4-0.8	
re	V ₂ O ₃	0.7-1.0	0.7	0.8	0.4-0.8	0.7	0.7-1	0.6–0.8	0.6–0.8	0.8-1.0	0.9	0.5	
Rutile co	Nb ₂ O ₅	0.8-1	0.5	0.3	0.1 - 0.8	0.4-0.8	1.1–1.4	0.2 - 0.3	0.5-0.7	0.4 - 0.6	0.96	0.3-0.4	
Size mm		11	S	4	8.4	6	6.4	10	7.2	9	0.4	9	3
Texture		zoned rutile grain with ilmenite rim	zoned rutile grain with ilmenite rim	zoned rutile grain with ilmenite lamellae and rim	zoned rutile grain with ilmenite rim,	strong atteration zoned rutile grain with thin ilmenite rim, flame-like ilm lamellae	zoned rutile grain with thin ilmenite rim, flame-like ilm lamellae	zoned rutile grain with thick ilmenite rim, flame-like ilm lamellae	zoned rutile grain with thick ilmenite rim, flame-like ilm lamellae	zoned rutile grain with thin ilmenite rim, flame-like ilm lamellae	rutile with ilmenite rim	ilmenite-rutile intergrowth with complex exsolution features	ilmenite grain with thin
Type		Υ	V	V	V	V	A	V	V	×	A	B1	B2
Sample		7134/1	7135	7136	7193/1	7193/2	7193/3	7193/4	7193/5	7193/6	7167/4	7134/2	7194/1

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Table 3 (continued)

Sample	Type	Texture	Size mm	Rutile co	ore		Rutile lame	llae/grains		Ilmenite			Inclusi	ons			Alt	eration
				Nb ₂ O ₅	V ₂ O ₃	FeO	Nb ₂ O ₅	V ₂ O ₃	FeO	MnO	MgO	Nb ₂ O ₅	msl-md ətimeriloW	Columbite	Pyrochlore	Othe	≈ Seudorutile"	.,əuəxoənəŢ,,
7194/2	B2	ilmenite grain with thin lamellae of Nb-rutile	2.5				13-15	>3	4-5	2.5	0.1	0.3-0.4	x			×	×	
7194/3	B2	altered rutile lamellae in ilmenite matrix	e				0.03 - 2.4	0.4 - 1.0	0.2-1.7	7–9	<0.05	0.12-0.15					х	x
7194/4	B2	ilmenite grain with thin lamellae of Nb-rutile, altered	б				1.9–10.2	0.8-1.6	0.6–3.5	2.3–2.5	0.1–0.4	0.1-0.25					×	
7167/5	B 2	ilmenite grain with thin lamellae of rutile	0.5				3.2	1.1	1.5	3.5	0.05	0.02				x		
7167/7	B 2	ilmenite grain with thin lamellae of rutile	0.4				4.1	1.4	1.5	3.5	0.05	0.06						
7167/2	U	rutile with exsolution lamellae of W-enriched rutile	0.5	0.15	0.6	0.5	0.3	1	7									
hm-la Sampl	n Her e loca	matite lamellae; <i>sph</i> llities for placers: Pi	sphalerit ngermüh	te; <i>po</i> p le (Zoti	yrrho tbach)	tite; <i>n</i> (713.	<i>nt</i> magnet 4); Pflaum	ite; x pr bach ne	esent; <i>n</i> ear Pleys	. <i>a</i> . not <i>a</i> stein (71	malyzec 67); Zo	l Ilgraben (7193)	i; Igla	Isreut	th, Tirs	chenr	euther

Waldnaab (7194) Sample localities for bedrock: Weißenstein (7135); Steiniger Loh, NW Fahrenberg (7136)

The origin of rutile-ilmenite aggregates



Fig. 3. Micrographs of nigrine types (Back-scattered electron images/BSE). **a** Rutile core with rim of ilmenite (ilm). Nigrine type A; sample 7135. **b** Oriented exsolution lamellae of ilmenite (ilm) in and near the edge of a rutile core. Nigrine type A; sample 7193/2. **c** Intergrowth of rutile (dark) and ilmenite (bright) in nigrine type B1. Subparallel wavy to discontinuous exsolution lamellae of Nb-enriched rutile are observed in matrix rutile; sample 7134/2. **d** Ilmenite with lamellae of niobian rutile in nigrine type B2. Surfaces of rutile are speckled with tiny crystals of columbite (white). Ilmenite is altered to pseudorutile (psrt), recognizable by its darker color in the BSE image; sample 7194/1. **e** Dendritic to graphic intergrowth of lath-shaped rutile and ilmenite in nigrine type B2. Ilmenite is altered to pseudorutile and to a Ti-rich phase (leucoxene) along the rutile surfaces; sample 7194/3. **f** Trellis-like patterns of W-enriched rutile exsolved from W-poor rutile; sample 7167/2



Fig. 4. Photomicrographs of nigrine grains taken in reflected light; all photos with one polar. **a** Nigrine type A: zoned rutile-ilmenite aggregate with attached bedrock (quartz-orthoclase-biotite); sample 7136. **b** Two individual ilmenite (ilm) crystals (pleochroic) with oriented lamellae of niobian rutile (Nb rt), and replaced by pseudorutile (grey phase along cracks). Nigrine type B2; sample 7194/2; oil immersion objective. **c** Ilmenite (ilm) with abundant hematite exsolution blebs (white) intergrown with rutile (rt); a few individual inclusions of magnetite (mt) and pyrochlore (dark). Nigrine type B1; sample 7134/2; oil immersion objective. **d** Large columbite (grey) included in ilmenite (ilm) with abundant hematite exsolution blebs (white), intergrown with rutile (rt). Nigrine type B1; sample 7134/2; oil immersion objective.

elevated V_2O_3 (>2 wt.%), Cr_2O_3 (>0.6 wt.%), Ta_2O_5 (up to 0.4 wt.%), WO_3 (up to 2 wt.%) and FeO (up to 2.7 wt.%) (Table 1, columns 5–6).

Nigrine type B1: One well-rounded grain consists of a complex intergrowth of ilmenite and rutile (Fig. 3c), and inclusions of ferrocolumbite and pyrochlore (Fig. 4c, d). The core-rim zonation of rutile and ilmenite, as it was recognized in type A nigrine, is not as strongly developed in this type; for this reason, it is termed B1 nigrine. BSE images and microprobe analyses revealed needle-shaped discontinuous exsolution lamellae of Nb-enriched rutile $(1.5-8.6 \text{ wt.\% Nb}_2O_5)$ within Nb-poor rutile $(0.3-0.4 \text{ wt.\% Nb}_2O_5)$ (Figs. 3c, 6, Table 1, column 7); such Nb-rich areas cannot be distinguished from normal rutile in reflected light. The rutile areas also have crystallographically oriented, very thin lamellae of ilmenite, whereas the grainy ilmenite carries abundant hematite exsolution lamellae (Fig. 4c, d). Compared to other nigrine types, the ilmenite in type B1 has lower Ti contents, and elevated Nb and V concentrations (Table 1, column 3; Fig. 8).



Fig. 5. Chemical transect across zoned nigrine grain (type A, sample 7134/1) consisting of a rutile core, ilmenite rims (ilm), and transition zones (T). Note the logarithmic scale for Nb and Mn. Spikes in the transition zone and the ilmenite rims are inclusions and lamellae of Nb-bearing rutile, and of ilmenite in the rutile phase, respectively

Nigrine type B2: Grains of nigrine B2 are smaller than those of B1 falling in the range 0.4-3 mm. They consist of an ilmenite host with oriented lamellae of rutile (Figs. 3d, e, 4b). The inclusion assemblage is dominated by wolframite, ferroco-lumbite and zircon. Rutile occurs as ovoids randomly distributed in the ilmenite (Fig. 3e) or as lath-shaped inclusions in an oriented intergrowth with ilmenite (Figs. 3d, 4b). In some B2 nigrine grains, rutile and ilmenite are intergrown in a patchwork fashion resembling trellis-type lamellae. Ilmenite is replaced by a more Ti-rich alteration product (pseudorutile, Figs. 3d, 4b). Rutile lamellae in nigrine type B2 may be exceptionally rich in Nb, Fe, and V with up to 21 wt.% Nb₂O₅, 2.6 wt.% V₂O₃ and as much as 7 wt.% FeO (Figs. 6, 7; Table 1, columns 8–9, 13). The ilmenite carries MnO (1.5–9 wt.%) as a major impurity, and Nb₂O₅ may reach 0.4 wt.% (Table 1, column 4).

Rutile C: Aggregates of rutile C are similar in size to aggregates of nigrine A and B but different in composition (Fig. 3f). In BSE images two different phases may be distinguished in the rutile aggregate. Needles of W-rich rutile ($5 \text{ wt.}\% \text{ WO}_3$) are encountered in a W-poor rutile host (Table 1, column 10).

Chemical composition of nigrine and rutile C aggregates

Rutile core: Rutile cores of nigrine aggregates are characterized by low, but consistent concentrations of Nb (up to $1.4 \text{ wt.\% Nb}_2O_5$), V (up to 1 wt.\% V_2O_3), Fe (up to 0.8 wt.% FeO), and Cr (up to $0.2 \text{ wt.\% Cr}_2O_3$). The concentrations of Ta, Sc, W, Sn, Mn, Mg are mostly below the detection limit of the electron microprobe (Table 1, columns 11–13). Some grains show continuous zoning of V, Cr, Fe, V, W, and Nb with increasing values towards the rutile-ilmenite transition zone (Fig. 5).

Rutile and ilmenorutile lamellae: Lamellae and grains in the rutile-ilmenite transition zone of type A nigrine grains have considerably higher concentrations of impurities than the core rutile, reaching 10 wt.% Nb₂O₅, 2 wt.% V₂O₃, 4.8 wt.% FeO, 0.6 wt.% Cr₂O₃, and 2.0 wt.% WO₃ (Table 1, columns 5–6). Lamellae in type B nigrine grains may contain exceptional contents of Nb₂O₅ (21 wt.%) and FeO (7 wt.%), as well as V₂O₃ (2.6 wt.%), Cr₂O₃ (0.7 wt.%) and 1.8 wt.% WO₃ (Table 1, columns 7–9). On the basis of four oxygen atoms, the trace element-rich niobian rutile has a formula of $(Ti_{1.46}Nb_{0.29}Fe_{0.18}V_{0.06})O_4$; thus, 25% of the metals are metals other than Ti. This "niobian rutile" corresponds to ilmenorutile [(Ti,Nb, Fe)O₂], which forms a solid-solution series with strüverite $[(Ti,Ta,Fe)O_2]$ (*Cerny* et al., 1981; Černy and Ercit, 1985; Černy et al., 1999). The negative correlation of Ti with other metals is illustrated for the lamellar inclusions in different nigrine types in Fig. 6, and compared to the more restricted composition of their respective rutile cores. The impurity elements Nb, Ta, Fe, V, Cr and W show reasonable to good positive correlations (Fig. 7). Cation totals normalized to 4 oxygens exceeding the ideal value of 2.0 (Table 1, columns 5-10) have been observed before, and are attributed either to oxygen deficiency $(Ti^{4+} \leftrightarrow 2Fe^{3+})$ or to $3\text{Ti}^{4+} \leftrightarrow \text{Fe}^{3+}$ substitution, or both (*Černy* et al., 1981).

Ilmenite: The composition of ilmenite in the nigrine aggregates varies in terms of Ti-Fe ratio and the concentrations of additional elements, especially of Mn and Mg. In most ilmenites analyzed, Ti/(Ti + Fe) ranges from 0.5 to 0.53, extending to 0.57 in places (Fig. 8); only type B1 ilmenite is significantly





Fig. 7. Variation of Nb versus Fe in rutile lamellae and cores of nigrine grains in atoms per formula unit based on 4 oxygens

enriched in Fe, probably due to submicroscopic inclusions of hematite (Fig. 4c, d, Table 1, column 3). The intra-grain variation of the Ti-Fe ratio is high, probably due to alteration processes removing and oxidizing Fe (see later). The Mn concentrations in ilmenite are homogeneous within grains. Most grains have between 1 and 4 wt.% MnO, with only one B2 type grain having 8–9 wt.% MnO (Fig. 8). MgO concentrations are usually between 0.1 and 0.2 wt.%, and reach 0.4–0.6 wt.% in few types A and B2 grains only. The concentrations of those elements that are considerably enriched in rutile (Nb, V, Cr, W) are lower, but often detectable. Nb₂O₅, for example, reaches 0.5 wt.% in some type B2 grains. V₂O₃ is fairly constant at 0.2–0.4 wt.% in all grains (Table 1, columns 1–4).

Mineral inclusions in nigrine

Wolframite: Wolframite is a mineral common to types A and B nigrine, occurring as small ($<10 \,\mu$ m) single-phase inclusions in ilmenite only. Based on the Fe/ (Fe + Mn) ratios between 0.91 and 0.93, the mineral may be described as almost pure ferberite (Table 2, columns 3–4).

Zircon and sulfides: Zircon inclusions are ubiquitous in nigrine and were not analyzed in detail. Sphalerite and pyrrhotite are rare constituents, only identified under the microscope in nigrine type A2.

Fig. 6. Composition of rutile in nigrine grains, in atoms per formula unit (apfu) calculated to 4 oxygen atoms. Lamellae are distinguished from homogeneous cores in types A, B1, B2 and C



Fig. 8. Compositional variation of ilmenite (ilm) and pseudorutile (psrt) in four groups of nigrine grains. Pseudorutile as defined here is characterized by lower totals than ilmenite; the transition from ilmenite to pseudorutile is set at Ti/(Ti + Fe) values of 0.55

Ferrocolumbite: Single inclusions and aggregates of ferrocolumbite of the general formula (Fe,Mn)(Nb,Ta) $_2O_6$, up to $40\,\mu\text{m}$ across (Fig. 4d), are restricted to types B1 and B2 nigrine. Within a given grain of nigrine, compositions are rather homogeneous, but differ slightly between grains. In type B1, ferrocolumbite is associated with pyrochlore-group minerals, and with wolframite in type B2. Weak zoning in some back-scatter electron images is due to slightly higher concentrations of Ta in cores compared to rims (e.g., 3.4 vs 3.0 wt.% Ta₂O₅). Ferrocolumbite in inclusions in nigrine aggregates are Nb- and Fe-rich, with 0.80 < Fe/(Fe + Mn) < 0.90 and 0.97 < Nb/(Nb + Ta) < 0.99 (Table 2, columns 1–2). Additional elements in significant concentrations are Sc (up to 2.5 wt.% Sc₂O₃), W (up to 3.2 wt.% WO₃), Sn (up to 0.5 wt.% SnO₂), Mg (up to 0.5 wt.% MgO) and Ca (up to 3.4 wt.% CaO). Vanadium, Cr, Al and Pb may occur in traces. The concentrations of TiO₂ range from 3 to 8 wt.%, but the effect of secondary fluorescence from surrounding ilmenite is not clear. Compared to ferrocolumbite included in type B1 nigrine, the ferrocolumbite from B2 nigrine is depleted in Nb and Mn but enriched in Ca, Ti and W. Iron-rich columbite is typically found in pegmatitic rocks poor in F and carrying, e.g., beryl (Cerny, 1992).

Pyrochlore: Inclusions of the pyrochlore-group $[A_{2-m}B_2X_6[(O,OH,F) = Y]_{1-n}^* pH_2O$ (*Hogarth*, 1977)] are confined to ilmenite in type B1 nigrine where they form grains up to 40 µm in size that may be associated with columbite (Fig. 4c). They are chemically dominated by Nb and Ca, with minor Ta (up to 3.1 wt.% Ta₂O₅), Ti, W, Fe, and traces of V, Mn, Sn, Sc, Al, Na, U and Pb (Table 2,

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column 5). The chemical requirements to qualify for a member of the pyrochlore group, i.e. Nb + Ta > 2Ti and Nb > Ta, are met. The calculated average formula of Ca(Nb_{1.78}Ta_{0.04}Ti_{0.16}Fe³⁺_{0.06})₂O₆ is close to the aeschynite-type [(Ce,Ca,Fe,Th) (Ti,Nb)₂(O,OH)₆]. In the same host grain of ilmenite, betafite [(Ca,U,Na)₂(Ti,Nb, Ta)₂(O,OH)₇, *Nickel* and *Nichols*, 2004] was found as zoned grains up to 30 µm in size, having 29 wt.% UO₂ and an average formula of $(Ca_{1.1}Na_{0.2}U_{0.4})_{\Sigma 1.7}(Nb_{1.0}Ta_{0.1}Ti_{0.9})_{\Sigma 2.0}(O,OH)_7$ (Table 2, column 6). Betafite is replaced at the margin by a hitherto unknown phase. The results obtained from microprobe analysis point to a Nb–Ti–Ta-rich phase with minor concentrations of Fe, U, P, Al, Si, Ca, Pb, Sn, V and Mn (Table 2, column 7); taking into account the low totals, H₂O or (OH)⁻ may be expected. Relative to betafite, Nb and Ta are enriched at the expense of Ca and U in the altered product. It might represent hydrated betafite.

Iron oxides: Magnetite inclusions are sporadically observed in the ilmenite phase of type B1 nigrine (Fig. 4c) only, occurring as subhedral to euhedral crystals up to 10 μ m in size. According to microprobe analyses they contain up to 3.2 wt.% TiO₂ (Table 2, column 8), corresponding to a low-Ti titanomagnetite. Ilmenite displays frequent, almost submicroscopic hematite exsolution lamellae only in nigrine type B1 (Fig. 4c, d).

Alteration products of nigrine

"Hydroilmenite", "leached ilmenite", pseudorutile: Ilmenite is commonly replaced by a more Ti-rich phase along the edges and fissures, commonly in a dendritic pattern, in nigrine types A and B (Figs. 3d, e, 4b, d). By definition, Ti/(Ti + Fe) ratios between 0.6 and 0.7 qualify Ti-Fe compounds as pseudorutile (Grey et al., 1994), while minerals with lower ratios are rather called "hydroilmenite" (Frost et al., 1983), a name not validated by the IMA. Mücke and Bhadra Chaudhuri (1991) introduced the term "leached ilmenite", and Chernet (1999a) used "pseudoilmenite" to describe a leaching product intermediate between ilmenite and pseudorutile showing similar color and anisotropy than ilmenite. In reflected light and using oil immersion, the alteration products are characterized by grayish color and a reflectance lower than ilmenite (Fig. 4b-d); compared to strongly bireflecting and anisotropic ilmenite, pleochroism and anisotropic effects are inconspicuous or absent. Microprobe analyses give low totals in the range of 94 to 99 wt.% and Ti/(Ti + Fe) ratios between 0.55 and 0.63 (Fig. 8). The analyses shown in Table 2 (columns 11-13) represent average values with most individual analyses ranging from 56 to 63 wt.% TiO₂ and 32 to 45 wt.% Fe₂O₃; some analyses having up to 76 wt.% TiO₂ at 23 wt.% Fe₂O₃ are transitional to leucoxene. The MnO and MgO contents are similar to their ilmenite precursor phases (1-9 wt.% MnO, 0.1–0.6 wt.% MgO, <0.5 wt.% Nb₂O₅, 0.2–0.4 wt.% V₂O₃), although Mn appears to decrease slightly with increasing Ti, i.e. with advancing alteration (Fig. 8). On the basis of formula calculations using 9 oxygen and 5 cations, the iron in altered ilmenite grains is almost completely oxidized to Fe³⁺, whereas the host ilmenites contain very little Fe³⁺. The low totals can be attributed to introduction of some H_2O or $(OH)^-$ (*Mücke* and *Bhadra Chaudhuri*, 1991). The mechanism of ilmenite hydration is not documented in the samples, but it may firmly be assumed that water was introduced along fractures.

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Leucoxene (?): An alteration phase similar in texture but different in composition to pseudorutile was exclusively found associated with rutile of some types A and B nigrines (Fig. 3e). The chemical composition points to a Ti-rich phase (76 wt.% TiO₂) with appreciable concentrations of impurities, e.g. 2.0–2.5 wt.% FeO, 6.1-8.5 wt.% Al₂O₃, 3.5-4.1 wt.% P₂O₅, and subordinate amounts of V, Si and Ca. Totals in the range 90 to 95 wt.% suggest considerable incorporation of H₂O or the (OH)⁻ complex (Table 2, columns 9–10). The phase may be considered a special type of leucoxene, i.e. submicroscopic intergrowths of TiO₂, Al-rich phosphates and silicates.

Discussion

The source of nigrine and W-enriched rutile

Accessory ilmenite and rutile in gneisses of the NE Bavarian basement are of much smaller grain size than the Ti aggregates in the stream sediments around Pleystein (*Propach*, 1969). Moreover, single grains of ilmenite or rutile are more widespread in basement rocks than aggregates of both minerals as they were found in grains of nigrine A and B from the alluvial-fluvial sediments. Rutile grains from metapelitic and mafic rocks usually carry 0.1-0.3 wt.% and 0-0.18 wt.% Nb, respectively (*Zack* et al., 2004). This level is well below the Nb contents analyzed from rutile of the nigrine placer deposits. Therefore metamorphic rocks may definitely be ruled out as a source for nigrine and highly differentiated igneous rocks are most likely the source for the nigrine and rutile.

During the collisional stage of the Variscan orogeny several pegmatites were emplaced near Hagendorf and Pleystein (Fig. 1), containing Li phosphates, Nb-, Ta-, U-oxides and numerous other rare element minerals (*Teuscher* and *Weinelt*, 1972; *Forster* and *Kummer*, 1974; *Strunz* et al., 1975; *Tennyson*, 1981; *Mücke*, 1987; *Mücke* et al., 1990).

The Late Variscan pegmatites between Pleystein and Hagendorf are Nb-selective with Nb contents exceeding the Ta contents by some orders of magnitude (Teuscher and Weinelt, 1972). Such high Nb/Ta ratios were observed in the rutile component of nigrine grains, and in columbite from type-B nigrine. Niobium becomes a marker element, connecting the pegmatite-borne mineralizations with the detrital nigrine in the placer deposits. Titanium minerals, common to the placer deposits, are rare constituents of the mineral assemblage of the Hagendorf pegmatite province. In the upper portions at Hagendorf, above the 90 m level, an older pegmatite carrying abundant potassic feldspar is intruded by a younger quartzenriched pegmatitic root zone below (Uebel, 1975). Mücke (2000) subdivided the mineralization at the lower level in two paragenetic associations: (1) pyrrhotite stage, and (2) iron-oxide stage. Ilmenite and rutile are rare inclusions in magnetite. Minerals encountered in the older alkali feldspar-quartz pegmatites at an intermediate mining level or near surface have no equivalent among the secondary mineral assemblage in the placer deposits, with the exception of ferrocolumbite. Feldspar and the phosphate minerals bearing divalent Fe and Mn from the older pegmatite are very susceptible to chemical weathering and, hence, decomposed during post-Variscan times when a subtropical climate provoked a pervasive chemical weathering and erosion capped the NE Bavarian basement to great depth (*Dill*, 1995). Rutile and nigrine are more resistant to these conditions of chemical weathering than feldspar and phosphate minerals and, locally, acted as a shelter for less resistant minerals such as ferberite and pyrochlore. The placer deposits around Pleystein mirror the evolution of the Late Variscan pegmatites at the western edge of the Bohemian Massif. Different stages of the pegmatite evolution may be deduced from the composition and texture of the different types of nigrine A and B as well as from the W-enriched rutile C: (1) low-Nb quartz vein stage (A), (2) high-Nb pegmatite stage (B), (3) high-W stage (C). Furthermore, there is a wide range of compositional changes on transport or storage that left their imprints on grains of nigrine and rutile and allow for another subdivision of alteration processes, namely a (4) ferric-ferrous titanium alteration stage, and a (5) phosphate-titanium alteration stage.

Pegmatite evolution reflected by detrital nigrine and rutile

Low-niobium quartz vein stage: The nigrine grains are characterized by a specific element assemblage (in decreasing order of abundance: Ti, Fe, V, Nb, Mn, W, Mg, Cr, Ta, Sc, Sn, U) that reflects both their source as well as the process of crystallization. If we assume that the zoned and exsolved nigrine grains once formed from homogeneous grains, chemical traverses (e.g., Fig. 5) may be used to estimate their original composition. For zoned nigrine grains of type A this simplified approach yields the following ranges: 75-85 wt.% TiO2, 13-17 wt.% FeO, 0.5-0.7 wt.% V₂O₃, 0.4–0.9 wt.% MnO, 0.2–0.6 wt.% Nb₂O₅, 0.03–0.11 wt.% WO₃, 0.07– 0.18 wt.% MgO, 0.02-0.03 wt.% Ta₂O₅. The atomic ratios are 4-6 for Ti/Fe, 13–38 for Fe/Mn and 17–38 for Nb/Ta. Nigrine grains of type B1 are too heterogeneous for estimation of the original composition. Ilmenite grains with exsolution lamellae of niobian rutile or ilmenorutile (nigrine B2) will have lower Ti/Fe ratios than type A grains, and higher Nb concentrations. For comparison, recalculation of the original composition of niobian rutile from the McGuire pegmatite, Colorado, gave (simplified) $(Ti_{2,3}Nb_{0,8}Ta_{0,1}Fe^{3+}_{0,4}Fe^{2+}_{0,4})_4O_8$ or 20 mol% of Nb component (*Černy* et al., 1999). There, an original rutile has completely exsolved into Nbbearing pseudorutile I, that broke down to Nb-rich pseudorutile II embedded in Nb-Fe bearing "ferropseudobrookite", and later exsolved into niobian rutile, titanian ferrocolumbite and ilmenite.

The primary source of type-A nigrine lies in the NW-oriented quartz veins (Figs. 1, 2a, b). Mineral inclusions, textural features and the chemical signature between placer- and quartz-hosted nigrine are almost identical. *Brenan* et al. (1994) measured rutile from high-pressure quartz veins that contained only 50 ppm Nb and *Zack* et al. (2002) investigated rutile from Disentis, Switzerland, with ca. 150 ppm Nb, but with substantial amounts of Cr (1200 ppm Cr). Low Nb concentrations are explained by a low capacity of fluids for transporting high field strength elements (*Xiong* et al., 2005).

The ilmenite exsolution in rutile and the rim around rutile are controlled by the ferric/ferrous iron ratio and the f_{O2} . The buffering reaction is as follows: 2 Fe₂O₃+4 TiO₂ \leftrightarrow 4 FeTiO₃+O₂ (*Zhao* et al., 1998). At present we cannot provide fully quantitative data on the Fe³⁺/Fe²⁺ ratio due to the limitations placed by the EMP analyses. However, formula calculations on the basis of 4 cations and

6 oxygens indicate very low Fe^{3+}/Fe^{2+} ratios in all ilmenites except those of type B1, where the contribution of Fe^{3+} to total Fe averages to 9%. In types A and B2 nigrines, Fe^{3+}/Fe^{tot} averages to only 0.2 to 0.6. An indirect way to get an idea on the f_{O2} may be provided by minerals included in A-type nigrine. Apart from highfield strength elements Zr, Nb and Ta which are compatible in rutile, there are polyvalent elements like W which are very sensitive to the oxygen fugacity (Klemme et al., 2005). Although the W content of A-type nigrine is rather low, wolframite was identified, attesting to a rather moderate oxygen fugacity during this stage of quartz-rutile-ilmenite mineralization. Oscillation of elements during core-margin transition in A-type nigrine, shown by the chemical transect in Fig. 5 is interpreted in terms of a strong diffusion process, which is a common process in the pegmatite stage. Despite some uncertainties about the precise composition of the original phase, the entire process of exsolution or differentiation was less prominent during the low-niobium quartz vein stage than during the stage discussed in the following chapter. The low-niobium quartz vein stage is representative of the younger pegmatite transitional into the barren vein quartz.

High-niobium pegmatite stage: For the high-niobium nigrine of type B, no equivalent source rock was identified close-by. The "rune-like" or "graphic" intergrowth in nigrine is fostered by the epitactic relations between the original rutile and ilmenite and similar in appearance to the graphic feldspar-quartz intergrowth known from the core-margin zone of the host K-feldspar-quartz pegmatites of the Hagendorf province. Some of the B-type nigrine inclusions resemble Nb-Ti-U mineralization in the core zones of the late Variscan pegmatites of the Hagendorf pegmatite province. Therefore the term pegmatite stage is justified. Type-B placer mineralization is similar to the mineralization recorded from the 115 m-mining level of the feldspar pegmatite upward. Exsolution and separation of Ti phases prograded and diffusion processes at the rutile-ilmenite boundaries are more pronounced than during the previous quartz stage, generating true niobian rutile intergrown with ilmenite enriched in Mn and containing inclusions of columbite and, in places, pyrochlore-group minerals.

The tungsten contents in columbite inclusions, although extremely variable, are very high with as much as 8.89 wt.% WO₃. They resemble W contents reported from the Vezná pegmatite, Czech Republic and the McGuire pegmatite, USA (*Černy* et al., 1999, 2000). Wolframite included by rutile of type B nigrine corroborates correlation of type-B nigrine from the placer with the sulfide stage in the pegmatite which developed at approximately 450 °C (*Mücke* et al., 1990).

Based upon the Fe/(Fe + Mn) ratios ranging from 0.82 to 0.89 and the Nb/(Nb + Ta) ratios of 0.98 this titanian ferrocolumbite is typical of columbites from beryl-bearing F-impoverished pegmatites (*Černy*, 1992). In general, these Nb-Ta oxides exhibit decreasing Nb/Ta ratios with progressive fractionation (*Aleksandrov* 1963; *Linnen* and *Keppler*, 1997). Columbite included in type-B nigrine is still more selective for Nb than equivalent columbites reported by *Forster* (1965) from the nearby pegmatite (56.5 wt.% Nb and 22.8 wt.% Ta). The rather high Nb/Ta ratio determined from columbite inclusions in type-B nigrine suggests that the host environment for these nigrine-columbite mineral aggregates was less strongly fractionated than the adjacent Hagendorf pegmatite that was taken for reference.

Columbite is replaced by pyrochlore in type-B nigrine. A mineral assemblage, composed of Nb-rutile, columbite, aeschynite and pyrochlore, similar to that from B-type nigrine occurs in some of the veins associated with the Bayan Obo ironrare earth-niobium deposits, Inner Mongolia, China (*Drew* et al., 1990). In the Bayan Obo deposit, this mineral association represents the waning stages of the main mineralizing system, and is related to a Hercynian heating event. The preponderance of W of type-B placer nigrine which has no match in the pegmatites is considered as a mineralization transitional from the pegmatitic into the hydrothermal stages. It is considered as a stage distal to the older Hagendorf pegmatite, *sensu Uebel* (1975). *Linnen* and *Keppler* (1997) have shown that Ta/Nb solubility in the melt is an important factor for separation of Nb and Ta. During type-B nigrine mineralization in the source area, the differentiation of a granitic melt has no longer played a crucial role.

High-tungsten stage: Detrital type-C rutile stands out among the Ti compounds in the study area by its high WO₃ contents (up to 5 wt.%). Rutile grains in alluvial tin concentrates, mostly from the Western Tin Belt from Peninsular Malaysia contain tungsten in the range 0.5-1.0 wt.% WO₃ (*Hassan*, 1994). Rutile associated with biotite in the Potrerillos granite, Argentine, formed during post-magmatic alteration, ranges up to 7.14% WO₃ (*Sardi*, 2003). High-W rutile C is held to be of (pneumatolytic)-hydrothermal origin. Its comparatively low Nb contents resemble what has been observed in type-A nigrine samples from the NW-SE striking quartz veins in the western part of the study area. Conclusively, the high-W rutile represents a transitional stage between the barren quartz veins proximal to the Late Variscan pegmatite province and the wolframite-(gold) quartz veins at Neualbenreuth far off that pegmatitic zone near the Czech-German border (*Dill*, 1985, 1989).

Supergene alteration of nigrine

Ferric-ferrous titanium alteration stage: Pseudorutile and leucoxene are the products of ilmenite alteration under fluctuating redox conditions. Dendritic pseudorutile alters the host ilmenite under oxidizing conditions (Figs. 3d, e, 4c, d). The question, if this conversion of bivalent into trivalent iron went on under hypogene or supergene conditions is difficult to answer (*Anand* and *Gilkes*, 1985). Pseudorutile and leucoxene replacing ilmenite along fissures and cracks took place simultaneously in a slightly acidic environment near the water table of an old peneplain during the Cenozoic or even Late Mesozoic (*Dill*, 1995). Judging by the preponderance of pseudorutile over leucoxene oxidizing conditions prevailed most of the time.

Phosphate-titanium alteration stage: Primary phosphates such as apatite, triphylite or triplite are common in the pegmatites of the Hagendorf pegmatite province. They were attributed to the group of primary phosphates (*Forster*, 1965). The Al-Fe-P-bearing "leucoxene" compound may be ascribed to a supergene process during which Fe and Al were remobilized under slightly acidic and permanently oxidizing conditions. Saprolites which may contain such an aluminum-phosphate-sulfate mineralization formed throughout the Cenozoic and Late Mesozoic on various types of bedrock of the NE Bavarian Basement, examples of which have also been found among the secondary phosphates in the Hagendorf province (*Dill*, 2001).

Conclusions

Detrital grains of nigrine and associated W-enriched rutile in placer deposits in the Hagendorf pegmatite province provide clear evidence of a pegmatitic to hydrothermal derivation. Nigrine A is the detrital mirror image of the quartzose root zone of the younger pegmatite. Nigrine B is representative of the marginal facies of the older pegmatite stage found at an intermediate level in the in Late Variscan pegmatites.

Rutile C, albeit its paragenetic position could not precisely be fixed in the mineral succession, is the missing link between the pegmatitic nigrine mineralization mentioned above and the skarn-type scheelite mineralization nearby (*Jakob*, 1979) and hydrothermal wolframite-quartz veins (*Dill*, 1985, 1989). Nigrine grains underwent supergene alteration near and above the water table of an ancient palaeoaquifer.

Nigrine and rutile being abnormally rich in W may lead exploration geologists to rare-element pegmatites and may also be useful for sedimentologists who want to learn more about the source areas of heavy minerals in clastic sediments.

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