

# An efficient synthesis of fluorescent spiro[benzopyrazoloquinoline-indoline]triones and spiro[acenaphthylenebenzopyrazoloquinoline]triones

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**Abstract** An efficient and simple synthesis of spiro[11*H*-benzo[*g*]pyrazolo[4,3-*b*]quinoline-11,3'-[3*H*]indole]-2',5,10(1'*H*)-triones and spiro[acenaphthylene-1(2*H*),11'-[11*H*]-benzo[*g*]pyrazolo[4,3-*b*]quinoline]-2,5',10'-triones by a three-component condensation reaction of 2-hydroxy-1,4-naphthoquinone, pyrazol-5-amines, and isatins or acenaphthylene-1,2-dione in the presence of *p*-TSA as an inexpensive and available catalyst in refluxing ethanol is reported.

**Keywords** Isatin · Spirooxindole · Aminopyrazole · Spiro[benzopyrazoloquinoline-indoline]

## Introduction

Fluorescent heterocyclic compounds are of interest in many disciplines; for example, they are used as emitters for electroluminescence devices [1], molecular probes for biochemical research [2], in traditional textile and polymer fields [3], in fluorescent whitening agents [4], and in photoconducting materials [5].

Molecules with a naphthoquinone structure, due to their biological properties, constitute one of the most interesting classes of compounds in organic chemistry. They have also industrial applications and can potentially be used as intermediates in the synthesis of heterocycles [6]. A series of related naphthoquinone pigments (streptocarpone,  $\alpha$ -

dunnione, dunninol, and dunnione) from *Streptocarpus dunnii* have been isolated and characterized [7, 8]. On the other hand, pyrazolo[3,4-*b*]quinolines belong to a class of highly fluorescent compounds which emit mostly in the blue spectral range and have been classified as promising materials for optoelectronics [9]. Attempts have been made to produce redshifts in their fluorescence spectra by changing substituents [10].

Indole and indoline fragments are important moieties in a large number of natural biologically active compounds [11], and some indolines that are spiro-annulated with heterocycles at the 3 position have shown high biological activity [12–15]. The spirooxindole system is the core structure of many pharmacological agents and natural alkaloids [16–19]. Therefore, a number of methods have been reported for the preparation of spirooxindoles [20–23].

Considering the above reports, and as part of our program aiming at the synthesis of heterocycles [24–33], we are currently investigating the synthesis of new spirooxindoles containing pyrazolo[3,4-*b*]quinoline via a facile, atom-economical, and one-pot condensation reaction.

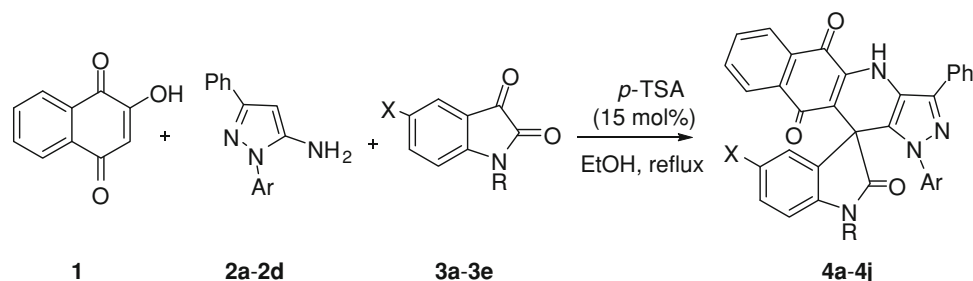
## Results and discussion

We found that a mixture of 2-hydroxy-1,4-naphthoquinone (1), pyrazol-5-amines 2a–2d, and isatins 3a–3e in the presence of a catalytic amount of *p*-TSA (15 mol%) in refluxing ethanol for 4–5.5 h afforded spiro[11*H*-benzo[*g*]pyrazolo[4,3-*b*]quinoline-11,3'-[3*H*]indole]-2',5,10(1'*H*)-triones 4 in good yields (Scheme 1). Using this method, some new spirooxindoles 4a–4j were selectively synthesized by a three-component condensation reaction. The results are summarized in Table 1.

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Scheme 1

**Table 1** Synthesis of spiro[indoline-benzopyrazoloquinoline] derivatives **4a–j**

Derivative	Ar	X	R	Yield/% <sup>a</sup>
<b>4a</b>	C <sub>6</sub> H <sub>5</sub>	H	H	94
<b>4b</b>	4-MeO-C <sub>6</sub> H <sub>4</sub>	H	H	87
<b>4c</b>	4-Br-C <sub>6</sub> H <sub>4</sub>	H	H	83
<b>4d</b>	4-NO <sub>2</sub> -C <sub>6</sub> H <sub>4</sub>	H	H	90
<b>4e</b>	C <sub>6</sub> H <sub>5</sub>	Br	Et	80
<b>4f</b>	4-MeO-C <sub>6</sub> H <sub>4</sub>	Br	Et	81
<b>4g</b>	C <sub>6</sub> H <sub>5</sub>	NO <sub>2</sub>	H	80
<b>4h</b>	4-NO <sub>2</sub> -C <sub>6</sub> H <sub>4</sub>	NO <sub>2</sub>	H	73
<b>4i</b>	C <sub>6</sub> H <sub>5</sub>	Me	H	81
<b>4j</b>	4-Br-C <sub>6</sub> H <sub>4</sub>	H	PhCH <sub>2</sub>	85

<sup>a</sup> Isolated yields

To the best of our knowledge, this new procedure provides the first example of a synthesis of spirooxindole-annulated benzopyrazoloquinolines. The reactions in ethanol are relatively safe, nontoxic, environmentally friendly, and inexpensive. This method is most simple and convenient, and would be applicable for the synthesis of different types of spiro[benzopyrazoloquinoline-indoline] derivatives. In addition, the workup for these very clean reactions involves only a filtration and a simple washing step with EtOH.

We have not established an exact mechanism for the formation of product **4**, but a reasonable possibility is shown in Scheme 2.

As expected, when the isatin **3** was replaced by acenaphthylene-1,2-dione (**5**), the spiro[acenaphthylene-1(2*H*),11'-[11*H*]benzo[*g*]pyrazolo[4,3-*b*]quinoline]-2,5',10'-triones **6** were obtained in good yields under the same reaction conditions (Scheme 3).

Compounds **4** and **6** are stable solids whose structures were established by IR and <sup>1</sup>H NMR spectroscopy and

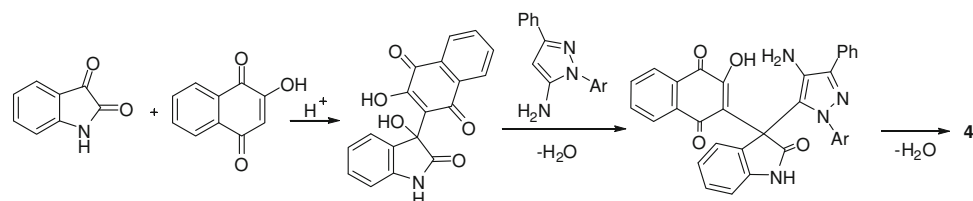
elemental analysis. Electronic absorption and fluorescence spectra of  $5 \times 10^{-5}$  M solutions of the selected products **4** and **6** in methanol were measured, and the results are shown in Table 2. A fluorescence excitation ( $\lambda_{\text{ex}}$ ) wavelength of 465 nm was used for compounds **4** and **6**. Figures 1, and 2 show the visible and emission spectra of the selected products. From Fig. 2, it can be observed that the products are fluorescent in solution. The  $\lambda_{\text{em}}$  values are presented in Table 2.

In conclusion, we have described a facile three-component method for the synthesis of spiro[11*H*-benzo[*g*]pyrazolo[4,3-*b*]quinoline-11,3'-[3*H*]indole]-2',5,10(1'*H*)-triones and spiro[acenaphthylene-1(2*H*),11'-[11*H*]benzo[*g*]pyrazolo[4,3-*b*]quinoline]-2,5',10'-triones in ethanol using readily available starting materials. Prominent among the advantages of this new method are operational simplicity, good yields in short reaction times, and the easy work-up procedures employed.

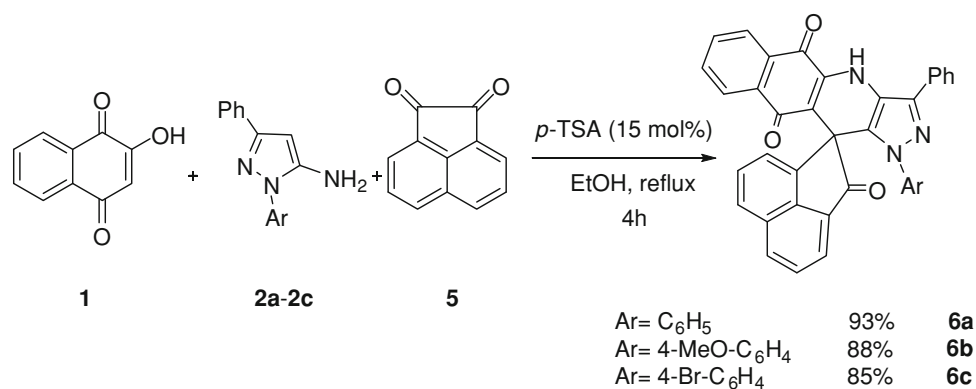
## Experimental

Melting points were measured on an Electrothermal 9100 apparatus. IR spectra were recorded on a Shimadzu IR-470 spectrometer. <sup>1</sup>H NMR spectra were recorded on a Bruker DRX-300 Avance spectrometer at 300.13 MHz. MS spectra were recorded on a Shimadzu QP 1100EX mass spectrometer operating at an ionization potential of 70 eV. Elemental analyses were performed using a Heraeus CHN-O-Rapid analyzer, and results agreed favorably with calculated values. Absorption spectra were recorded on a Shimadzu UV-2100 spectrophotometer. Fluorescence spectra were recorded using a PerkinElmer LS45 spectrofluorophotometer. Chemicals were purchased from Fluka or Merck and used as received.

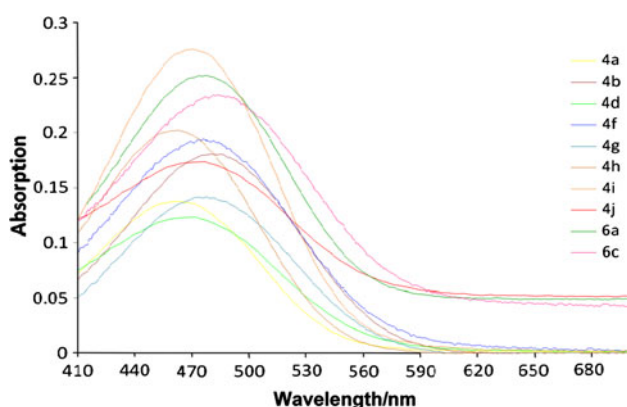
Scheme 2



Scheme 3

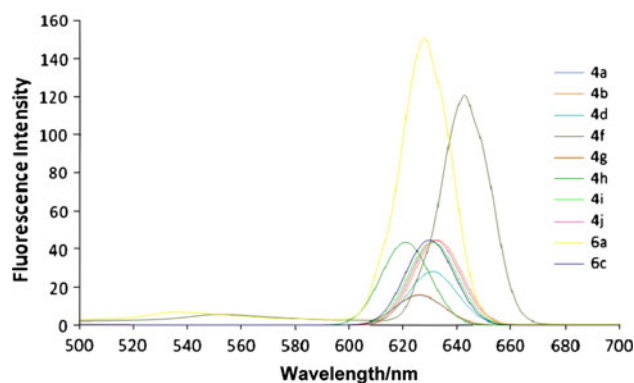
**Table 2** UV-vis and fluorescence emission data for compounds **4** and **6** in methanol

Entry	Product	$\lambda_{\text{abs}}/\text{nm}$	$\log \varepsilon$	$\lambda_{\text{em}}/\text{nm}$
1	<b>4a</b>	462	3.44	628
2	<b>4b</b>	475	3.55	626
3	<b>4c</b>	471	3.41	629
4	<b>4d</b>	471	3.39	630
5	<b>4e</b>	473	3.60	631
6	<b>4f</b>	476	3.58	626
7	<b>4g</b>	477	3.45	635
8	<b>4h</b>	463	3.60	619
9	<b>4i</b>	470	3.74	628
10	<b>4j</b>	477	3.54	633
11	<b>6a</b>	478	3.70	626
12	<b>6b</b>	477	3.49	633
13	<b>6c</b>	484	3.67	648

**Fig. 1** The visible spectra of the selected compounds **4** and **6** ( $5 \times 10^{-5}$  M in methanol)

#### Typical procedure for the preparation of compounds **4** and **6**

A mixture of isatin or acenaphthylene-1,2-dione (1 mmol), 1*H*-pyrazol-5-amine (1 mmol), 2-hydroxy-1,4-naphthoquinone (1 mmol), and *p*-TSA (15 mol%) in 5 cm<sup>3</sup>

**Fig. 2** The emission spectra of the selected compounds **4** and **6** ( $5 \times 10^{-5}$  M in methanol)

refluxing EtOH was stirred for 4 h. After the reaction had completed, as confirmed by TLC (eluent: EtOAc/*n*-hexane 1:3), the reaction mixture was filtered and the precipitate washed with water and then 5 cm<sup>3</sup> EtOH to afford the pure products. Due to the very low solubility of the products, we are unable to report the <sup>13</sup>C NMR data for these products.

#### 1,4-Dihydro-1,3-diphenylspiro[11*H*-benzo[*g*]pyrazolo[4,3-*b*]quinoline-11,3'-[3*H*]indole]-2',5,10(1'*H*)-trione (**4a**, C<sub>33</sub>H<sub>20</sub>N<sub>4</sub>O<sub>3</sub>)

Red powder; m.p.: 267–270 °C; IR (KBr):  $\bar{\nu}$  = 3,426, 3,343, 1,755, 1,727, 1,678 cm<sup>-1</sup>; <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>):  $\delta$  = 6.61–8.57 (18H, m, arom), 10.46 (1H, s, NH), 11.47 (1H, s, NH) ppm; MS: *m/z* = 520 (M<sup>+</sup>).

#### 1,4-Dihydro-1-(4-methoxyphenyl)-3-phenylspiro[11*H*-benzo[*g*]pyrazolo[4,3-*b*]quinoline-11,3'-[3*H*]indole]-2',5,10(1'*H*)-trione (**4b**, C<sub>34</sub>H<sub>22</sub>N<sub>4</sub>O<sub>4</sub>)

Dark red powder; m.p.: 260–262 °C; IR (KBr):  $\bar{\nu}$  = 3,343, 3,200, 1,745, 1,650, 1,615 cm<sup>-1</sup>; <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>):  $\delta$  = 3.86 (1H, s, OCH<sub>3</sub>), 6.67 (3H, d, *J* = 4.6 Hz, arom), 7.15–7.39 (8H, m, arom), 7.60 (2H, d, *J* = 8.1 Hz, arom), 7.79 (3H, bs, arom), 8.07 (1H, d, *J* = 6.5 Hz, arom), 9.90 (1H, s, NH), 10.09 (1H, s, NH) ppm; MS: *m/z* = 550 (M<sup>+</sup>).

*1-(4-Bromophenyl)-1,4-dihydro-3-phenylspiro[11H-benzo[g]pyrazolo[4,3-b]quinoline-11,3'-[3H]indole]-2',5,10(1'H)-trione (4c, C<sub>34</sub>H<sub>21</sub>BrN<sub>4</sub>O<sub>3</sub>)*

Orange powder; m.p.: 254–257 °C; IR (KBr):  $\bar{\nu}$  = 3,150, 3,056, 1,711, 1,679, 1,644 cm<sup>-1</sup>; <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>):  $\delta$  = 6.44 (1H, d, *J* = 8.6 Hz, arom), 6.63 (2H, d, *J* = 7.3 Hz, arom), 6.90–7.82 (13H, m, arom), 8.07 (1H, d, *J* = 8.7 Hz, arom), 12.10 (1H, s, NH), 12.49 (1H, s, NH) ppm; MS: *m/z* = 600 ([M + 2]<sup>+</sup>), 598 (M<sup>+</sup>).

*1,4-Dihydro-1-(4-nitrophenyl)-3-phenylspiro[11H-benzo[g]pyrazolo[4,3-b]quinoline-11,3'-[3H]indole]-2',5,10(1'H)-trione (4d, C<sub>33</sub>H<sub>19</sub>N<sub>5</sub>O<sub>5</sub>)*

Red powder; m.p.: >300 °C; IR (KBr):  $\bar{\nu}$  = 3,210, 3,061, 1,715, 1,676, 1,644 cm<sup>-1</sup>; <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>):  $\delta$  = 8.72 (1H, d, *J* = 9.0 Hz, arom), 8.94 (2H, d, *J* = 8.1 Hz, arom), 9.36–10.03 (13H, m, arom), 10.25 (1H, d, *J* = 8.9 Hz, arom), 10.05 (1H, s, NH), 10.18 (1H, s, NH) ppm; MS: *m/z* = 565 (M<sup>+</sup>).

*5'-Bromo-1'-ethyl-1,4-dihydro-1,3-diphenylspiro[11H-benzo[g]pyrazolo[4,3-b]quinoline-11,3'-[3H]indole]-2',5,10(1'H)-trione (4e, C<sub>35</sub>H<sub>23</sub>BrN<sub>4</sub>O<sub>3</sub>)*

Red powder; m.p.: 268–271 °C; IR (KBr):  $\bar{\nu}$  = 3,255, 1,722, 1,705, 1,677 cm<sup>-1</sup>; <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>):  $\delta$  = 0.83 (3H, bs, CH<sub>3</sub>), 3.15 (2H, bs, CH<sub>2</sub>), 6.63–8.05 (17H, m, arom), 9.98 (1H, s, NH) ppm; MS: *m/z* = 628 ([M + 2]<sup>+</sup>), 626 (M<sup>+</sup>).

*5'-Bromo-1'-ethyl-1,4-dihydro-1-(4-methoxyphenyl)-3-phenylspiro[11H-benzo[g]pyrazolo[4,3-b]quinoline-11,3'-[3H]indole]-2',5,10(1'H)-trione (4f, C<sub>36</sub>H<sub>25</sub>BrN<sub>4</sub>O<sub>4</sub>)*

Dark red powder; m.p.: 253–256 °C; IR (KBr):  $\bar{\nu}$  = 3,360, 1,749, 1,705, 1,639 cm<sup>-1</sup>; <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>):  $\delta$  = 1.05 (3H, bs, CH<sub>3</sub>), 3.42 (2H, bs, CH<sub>2</sub>), 3.85 (3H, s, OCH<sub>3</sub>), 6.68–8.06 (17H, m, arom), 10.01 (1H, s, NH) ppm; MS: *m/z* = 658 ([M + 2]<sup>+</sup>), 656 (M<sup>+</sup>).

*1,4-Dihydro-5'-nitro-1,3-diphenylspiro[11H-benzo[g]pyrazolo[4,3-b]quinoline-11,3'-[3H]indole]-2',5,10(1'H)-trione (4g, C<sub>33</sub>H<sub>19</sub>N<sub>5</sub>O<sub>5</sub>)*

Dark red powder; m.p.: 270–273 °C; IR (KBr):  $\bar{\nu}$  = 3,426, 3,327, 1,705, 1,672, 1,650 cm<sup>-1</sup>; <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>):  $\delta$  = 6.70 (3H, d, *J* = 7.4 Hz, arom), 7.15–8.06 (13H, m, arom), 8.51 (1H, d, *J* = 6.4 Hz, arom), 10.84 (1H, s, NH), 11.16 (1H, s, NH) ppm; MS: *m/z* = 565 (M<sup>+</sup>).

*1,4-Dihydro-5'-nitro-1-(4-nitrophenyl)-3-phenylspiro[11H-benzo[g]pyrazolo[4,3-b]quinoline-11,3'-[3H]indole]-2',5,10(1'H)-trione (4h, C<sub>33</sub>H<sub>18</sub>N<sub>6</sub>O<sub>7</sub>)*

Dark red powder; m.p.: >300 °C; IR (KBr):  $\bar{\nu}$  = 3,418, 3,329, 1,711, 1,670, 1,653 cm<sup>-1</sup>; <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>):  $\delta$  = 6.51 (1H, d, *J* = 8.3 Hz, arom), 6.75 (2H, d, *J* = 7.9 Hz, arom), 7.12–7.82 (12H, m, arom), 8.06 (1H, d, *J* = 6.4 Hz, arom), 10.28 (1H, s, NH), 10.59 (1H, s, NH) ppm; MS: *m/z* = 610 (M<sup>+</sup>).

*1,4-Dihydro-5'-methyl-1,3-diphenylspiro[11H-benzo[g]pyrazolo[4,3-b]quinoline-11,3'-[3H]indole]-2',5,10(1'H)-trione (4i, C<sub>34</sub>H<sub>22</sub>N<sub>4</sub>O<sub>3</sub>)*

Red powder; m.p.: >300 °C; IR (KBr):  $\bar{\nu}$  = 3,194, 3,056, 1,716, 1,678 cm<sup>-1</sup>; <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>):  $\delta$  = 2.17 (3H, s, CH<sub>3</sub>), 6.49–8.05 (17H, m, arom), 10.02 (1H, s, NH), 10.07 (1H, s, NH) ppm; MS: *m/z* = 534 (M<sup>+</sup>).

*1-(4-Bromophenyl)-1,4-dihydro-3-phenyl-1'-(phenylmethyl)spiro[11H-benzo[g]pyrazolo[4,3-b]quinoline-11,3'-[3H]indole]-2',5,10(1'H)-trione (4j, C<sub>40</sub>H<sub>25</sub>BrN<sub>4</sub>O<sub>3</sub>)*

Red powder; m.p.: >300 °C; IR (KBr):  $\bar{\nu}$  = 3,211, 1,710, 1,681, 1,623 cm<sup>-1</sup>; <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>):  $\delta$  = 3.91 (1H, d, *J* = 16 Hz, CH<sub>2</sub>), 4.79 (1H, d, *J* = 16 Hz, CH<sub>2</sub>), 6.42 (1H, d, *J* = 8.9 Hz, arom), 6.62 (2H, d, *J* = 6.5 Hz, arom), 6.91–7.83 (18H, m, arom), 8.09 (1H, d, *J* = 8.9 Hz, arom), 10.11 (1H, s, NH) ppm; MS: *m/z* = 690 ([M + 2]<sup>+</sup>), 688 (M<sup>+</sup>).

*1',4'-Dihydro-1',3'-diphenylspiro[acenaphthylene-1(2H),11'-[11H]benzo[g]pyrazolo[4,3-b]quinoline]-2,5',10'-trione (6a, C<sub>37</sub>H<sub>21</sub>N<sub>3</sub>O<sub>3</sub>)*

Red powder; m.p.: 252–255 °C; IR (KBr):  $\bar{\nu}$  = 3,205, 1,718, 1,670, 1,607 cm<sup>-1</sup>; <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>):  $\delta$  = 6.17 (2H, d, *J* = 6.6 Hz, arom), 6.60 (2H, t, *J* = 6.3 Hz, arom), 6.85 (1H, t, *J* = 6.1 Hz, arom), 7.49–8.12 (15H, m, arom), 10.05 (1H, s, NH) ppm; MS: *m/z* = 555 (M<sup>+</sup>).

*1',4'-Dihydro-1'-(4-methoxyphenyl)-3'-phenylspiro[acenaphthylene-1(2H),11'-[11H]benzo[g]pyrazolo[4,3-b]quinoline]-2,5',10'-trione (6b, C<sub>38</sub>H<sub>23</sub>N<sub>3</sub>O<sub>4</sub>)*

Red powder; m.p.: 263–266 °C; IR (KBr):  $\bar{\nu}$  = 3,183, 1,705, 1,678, 1,611 cm<sup>-1</sup>; <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>):  $\delta$  = 3.86 (3H, s, CH<sub>3</sub>), 6.15 (2H, d, *J* = 8.7 Hz, arom), 6.59 (2H, t, *J* = 8.9 Hz, arom), 6.85 (1H, t, *J* = 8.4 Hz, arom), 7.15 (2H, d, *J* = 8.9 Hz, arom), 7.48–8.13 (12H, m, arom), 9.92 (1H, s, NH) ppm; MS: *m/z* = 585 (M<sup>+</sup>).

*1'-(4-Bromophenyl)-1',4'-dihydro-3'-phenylspiro[acenaphthylene-1(2H),11'-[11H]benzo[g]pyrazolo[4,3-b]quinoline]-2,5',10'-trione (6c, C<sub>37</sub>H<sub>20</sub>BrN<sub>3</sub>O<sub>3</sub>)*

Red powder; m.p.: 266–269 °C; IR (KBr):  $\bar{\nu}$  = 3,200, 1,711, 1,678, 1,617 cm<sup>-1</sup>; <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>):  $\delta$  = 6.14–8.08 (19H, m, arom), 10.26 (1H, s, NH) ppm; MS: *m/z* = 635 ([M + 2]<sup>+</sup>), 633 (M<sup>+</sup>).

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## References

- Hunger K (2003) Industrial dyes. Wiley-VCH, Weinheim, p 569
- Dmitry A, Pavel A (2003) Chem Commun 12:1394

3. Gold H (1971) In: Venkataraman H (ed) The chemistry of synthetic dyes. Pergamon, New York, p 535
4. Belgodere E, Bossio R, Chimichi S, Passini V, Pepino R (1985) *Dyes Pigment* 4:59
5. Kalle AG (1962) British Patent 1962;895(001)
6. Thomson RH (1997) Naturally occurring quinines, 4th edn. Chapman & Hall, London
7. Perez AL, Lamoureux G, Sanchez-Kopper A (2007) *Tetrahedron Lett* 48:3735
8. Inoue K, Ueda S, Nayeshiro H, Inouye H (1982) *Chem Pharm Bull* 30:2265
9. Tao YT, Balasubramaniam E, Danel A, Wisla A, Tomasik P (2001) *J Mater Chem* 11:768
10. Gondek E, Kityk IV, Danel A, Wisla A, Pokladko M, Sanetra J, Sahraoui B (2006) *Mater Lett* 60:3301
11. Sundberg RJ (1996) The chemistry of indoles. Academic, New York
12. Joshi KC, Chand P (1982) *Pharmazie* 37:1
13. Da Silva JFM, Garden SJ, Pinto AC (2001) *J Braz Chem Soc* 12:273
14. Abdel-Rahman AH, Keshk EM, Hanna MA, El-Bady SM (2004) *Bioorg Med Chem* 12:2483
15. Zhu SL, Ji SJ, Yong Z (2007) *Tetrahedron* 63:9365
16. Kang TH, Matsumoto K, Tohda M, Murakami Y, Takayama H, Kitajima M, Aimi N, Watanabe H (2002) *Eur J Pharmacol* 444:39
17. Ma J, Hecht SM (2004) *Chem Commun* 1190
18. Usui T, Kondoh M, Cui CB, Mayumi T, Osada H (1998) *Biochem J* 333:543
19. Khafagy MM, Abd El-Wahab AHF, Eid FA, El-Agrody AM (2002) *Farmaco* 57:715
20. Nandakumar A, Thirumurugan P, Perumal PT, Vembu P, Ponuswamy MN, Ramesh P (2010) *Bioorg Med Chem Lett* 20:4252
21. Chen H, Shi D (2010) *J Comb Chem* 12:571
22. Chen T, Xu XP, Ji SJ (2010) *J Comb Chem* 12:659
23. Li Y, Chen H, Shi C, Shi D, Ji S (2010) *J Comb Chem* 12:231
24. Bazgir A, Seyyedhamzeh M, Yasaei Z, Mirzaei P (2007) *Tetrahedron Lett* 48:8790
25. Sayyafi M, Seyyedhamzeh M, Khavasi HR, Bazgir A (2008) *Tetrahedron* 64:2375
26. Dabiri M, Delbari AS, Bazgir A (2007) *Synlett* 821
27. Dabiri M, Arvin-Nezhad H, Khavasi HR, Bazgir A (2007) *Tetrahedron* 63:1770
28. Ghahremanzadeh R, Imani Shakibaei G, Bazgir A (2008) *Synlett* 1129
29. Bazgir A, Noroozi Tisseh Z, Mirzaei P (2008) *Tetrahedron Lett* 49:5165
30. Ghahremanzadeh R, Sayyafi M, Ahadi S, Bazgir A (2009) *J Comb Chem* 11:393
31. Ghahremanzadeh R, Ahadi S, Imani Shakibaei G, Bazgir A (2010) *Tetrahedron Lett* 51:499
32. Ghahremanzadeh R, Imani Shakibaei GH, Ahadi S, Bazgir A (2010) *J Comb Chem* 12:191
33. Jadidi K, Ghahremanzadeh R, Bazgir A (2009) *J Comb Chem* 11:341