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## Gold single atom-based aptananozyme as an ultrasensitive and selective colorimetric probe for detection of thrombin and C-reactive protein

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#### Abstract

An ultra-efficient biocatalytic peroxidase-like Au-based single-atom nanozyme (Au-SAzymes) has been synthesized from isolated Au atoms on black nitrogen doped carbon (Au–N-C) using a simple complexation-adsorption-pyrolysis method. The atomic structure of AuN<sub>4</sub> centers in black carbon was revealed by combined high-resolution transmission electron microscopy/high-angle annular dark-field scanning transmission electron microscopy. The Au-SAzymes showed a remarkable peroxidase activity with 1.7 nM as Michaelis–Menten constant, higher than most previously reported SAzyme activity. Density functional theory and Monte Carlo calculations revealed the adsorption of  $H_2O_2$  on AuN<sub>4</sub> with formation of OH\* and O\*. Molecular recognition was greatly enhanced via label-free integration of thiol-terminal aptamers on the surface of single Au atoms (Aptamer/Au-SAzyme) to design off–on ultrasensitive aptananozyme-based sensor for detecting thrombin and CRP with 550 pM and 500 pg mL<sup>-1</sup> limits of detection, respectively. The Aptamer/Au-SAzyme showed satisfactory accuracy and precision when applied to the serum and plasma of COVID-19 patients. Due to the maximum Au atom utilization, approximately 3636 samples can be run per 1 mg of gold, highlighting the commercialization potential of the developed Aptamer/Au-SAzyme approach.

Keywords Single-atom nanozyme · Aptamer · Aptananozyme · Single gold atom · Thrombin · CRP

## Introduction

Nanozymes, nanomaterials possessing inherent enzyme-like characteristics, have garnered significant interest in recent years due to their potential to overcome the limitations of natural enzymes, such as low stability, high cost, and challenging storage [1, 2]. This is due to their growing applications in biosensing [3, 4], imaging [5], therapy [6], and environmental remediations [7, 8]. Compared with natural enzymes, enzyme mimics stand out due to their distinct advantages such as great stability, flexibility, easy storage, and easy production [9]. In this area, the primary obstacle is

the limited catalytic efficiency of enzyme mimics, especially peroxidase mimics [10, 11].

Nanozymes are nano-sized materials exhibiting catalytic activity mimicking the functions of natural enzymes across various catalytic reactions, such as peroxidase, oxidase, catalase, and superoxide dismutase [2, 12–14]. Nanozymes showed higher advantageous over natural enzymes due to shelf-life stability, low-cost production, flexibility in tailoring, and easy functionalization [15]. Metal nanostructures have been used effectively as nanozyme starting from the first observation in Fe<sub>3</sub>O<sub>4</sub> [2, 14–16]. Nevertheless, the catalytic efficacy of nanozymes relies on multiple factors, one of them is the size of the particle [13, 16–19].

Recently, single-atom catalysts (SACs) have come to the forefront as a new field of heterogeneous catalysis in which different catalytic reactions can be carried out via a single atom on a surface [20–22]. SACs represent the smallest possible size for metal particles, featuring isolated metal atoms dispersed individually on supports. SACs optimize the utilization efficiency of metal atoms. Additionally, it is essential to have an appropriate substrate to prevent the aggregation

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of these isolated single atoms. Single atoms possess high surface energy, and maintaining their uniform dispersion is crucial for the catalytic stability of SACs [23]. Due to their distinct geometric structures, uniform active sites, and remarkable activity and selectivity, SACs can amalgamate the benefits inherent in both homogeneous and heterogeneous catalysts. Additionally, they exhibit high efficiency in atom utilization, excellent activity, and remarkable recyclability [24–26].

Following the ground-breaking finding of SACs by Zhang et al., [27] numerous reports were published using SACs for various applications, such as hydrogen evolution [28], oxygen reduction [28], and carbon dioxide reduction [29]. SACs with structure M-N-C (refers to metals, Pt, Co, Fe, Mn, etc.) have similar M-Nx sites like natural metalloenzymes. Therefore, it is proposed that M–N-C SACs can behave as bioinspired single-atom enzymes (SAzymes) to mimic the composition and catalytic activity of natural enzymes [30]. However, comparing to the use of SAC in other applications, yet, not much SACs are reported as biocatalyst (or SAzyme). Zhu et al. [31] prepared Fe–N-C SAzymes that showed peroxidase-mimic catalytic activity using FeCl<sub>3</sub>, glucose, and dicyandiamide as precursors under high temperature calcination. The Fe-N-C SAzymes were used for colorimetric detection of H<sub>2</sub>O<sub>2</sub> in via using chromogenic 3,3',5,5'-tetramethylbenzidine (TMB) substrate, demonstrating the acceptable specificity and sensitivity. This study provides evidence that the active sites of Fe-N-C SAzymes, which include atomically dispersed FeN<sub>4</sub>, play a critical role for peroxides-like action. SAzyme have shown high sensitivity due to extraordinary catalytic activity of single atoms, however, for biochemical sensing in real matrices, one should think about improving the selectivity. For this reason, combining SAzyme with aptamer would be an excellent choice to synergize the merits.

The global COVID-19 pandemic, triggered by the highly transmissible SARS-CoV-2 virus, manifests in respiratory symptoms of diverse severity. Initially identified in Wuhan, China, in December 2019, the virus rapidly spread worldwide, leading to a declared global pandemic in 2020 [32–35]. Numerous biomarkers have been associated with severe cases of COVID-19, suggesting an immunochemical profile consistent, elevation of proinflammatory cytokines, particularly interleukin (IL)-6 and tumor necrosis factor- $\alpha$  (TNF- $\alpha$ ) [36–38], ferritin [39], and C-reactive protein (CRP) [40, 41]. Those have all been linked to severe COVID-19 and can be used to measure disease severity [42]. Moreover, the (in)-direct effects of the viral illness have been linked to a number of cardiovascular [43, 44], hematologic [45, 46], and thrombotic problems [47].

Thrombin is an important and unique biomolecule with both procoagulant and anticoagulant properties [41, 48–50].

It has a multifunctional protein that serves as an inhibitor, activator, and regulator of the cellular coagulation process [51, 52] while CRP is a protein produced by the liver in a response to inflammation [53, 54], and it has a pentameric structure made up of five identical 23-kDa globular subunits [55]. During the acute phase of an inflammatory or viral condition, interleukin-6 (IL-6) primarily induces the production of CRP by influencing the gene responsible for CRP synthesis [56–58]. Detection of thrombin and CRP is very important among biomarkers of diagnosis of Covid-19.

There are numerous techniques and methods to determine thrombin and CRP, including electrochemical [59, 60], colorimetric [61, 62], and fluorescence [63-67]. Common and benchmark procedures like radio-assay- and immunoassay-based techniques are also in use [68, 69]. In clinical diagnostics, having a dependable, sensitive, selective, stable, and cost-effective method for detecting thrombin and CRP is crucial, particularly for on-site detection and diagnosis. Enzyme-linked immunosorbent assay (ELISA) has always been the "the best standard way" for quantification of CRP [70–73]. ELISA can produce false positive results due to cross-reactivity, may not be sensitive or specific enough, is time-consuming and expensive, and has a limited dynamic range. It is important to be aware of these limitations when using ELISA for diagnostic or research purposes. Thus, one should think of simple, low-cost, highly selective, sensitive, robust, and fast-response protocol for detection of CRP and thrombin.

Aptamers are single-stranded oligonucleotides (DNA or RNA), with a short length that attach to certain molecules. Due to the shape-forming ability of single-stranded oligonucleotides, they can bind to their targets with tremendous affinity and selectivity [74–78] the attractions of aptamer have included a broad range of target molecules, robust and strong binding capacity, high stability, long shelf-life, simplicity in synthesis, and easy modification [79–81].

In the present work, we use a complexation-adsorption-pyrolysis strategy to synthesize isolated Au-SAzyme on N-doped carbon. The resulting Au-SAzyme exhibited remarkably efficient peroxidase activity. To gain deeper insights, we investigated the peroxidase mechanism through theoretical calculations. Subsequently, thiol-terminal aptamers (ssDNA or RNA) were immobilized on the dispersed Au-SAzymes to inhibit the peroxidase enzymatic activity, resulting in the creation of Aptamer/Au-SAzyme hybrids. A biochemical sensor based on Aptamer/Au-SAzyme was utilized for detection of thrombo-inflammatory COVID-19 biomarkers, thrombin, and CRP. To the best of our knowledge, this is the first report on mimetic enzyme activity of isolated Au atoms combined with ssDNA/RNA. Scheme 1 shows the preparation roadmap of Au-SAzyme and Aptamer/ Au-SAzyme.



Scheme 1 A The schematic preparation steps of Au-SAzyme. B The mechanism of biocatalytic activity response of engineered aptamer on Au-SAzyme surfaces and its recovery

## **Experimental section**

## Chemicals

Chloroauric acid trihydrate (HAuCl<sub>4</sub>·3H<sub>2</sub>O) was purchased from Sigma Aldrich Co. (Hong Kong, China). Thiolmodified aptamer for thrombin was synthesized and purified by Generay biotech Co. Ltd. (Shanghai, China) with sequences for ssDNA (5-(SH)-(CH2)6-CCA TCT CCA CTT GGT TGG TGT GGT TGG-3) specific for thrombin. The sequence RNA strand is (5'-SH-(CH2)6-GCC UGU AAG GUG GUC GGU GUG GCG AGU GUG UUA GGA GAG AUU GC-3) specific for CRP. Thrombin protein extracted from human plasma, C-reactive protein (CRP), Dimethyl sulfoxide (DMSO), Dimethylformamide (DMF), 3, 3' 5, 5'-Tetramethylbenzidine (TMB) were all purchased from Merck-Sigma-Aldrich (Baden-Württemberg, Germany) and used as received. 1, 10 phenanthroline monohydrate and hydrogen peroxide ( $H_2O_2$ , 30%) were bought from Biochem chemopharma Co.Ltd. (Biochem, ZA Cosne sur Loire, France). TE buffer (Tris-EDTA, pH 8.0) was bought from EMB Corporation (EMB Co., USA). PBS (phosphate buffer saline, pH 7) was ordered from (CDH Co. Ltd. Delhi, India).

### Instruments

All UV–Vis absorption spectra and enzyme kinetics data were performed by a PG spectrophotometer (PG, UK). X-ray photoelectron spectroscopy (XPS) spectra were recorded using Thermo ESCALAB 250XI (Thermo Fisher, USA). X-ray diffraction (XRD) spectra were acquired by a D8 ADVANCE (Bruker, Germany). Raman spectra were recorded using a Renishaw 1000 micro-Raman system (Renishaw, UK). The aberration-corrected high-angle annular dark-field scanning transmission electron microscopy (HAADF-STEM) was performed by a Titan G2-600 (FEI, USA).

## **Preparation of Au-SAzyme**

The Au-SAzyme biocatalyst was prepared according to literature with some modifications [82]. In brief, a mixture of 0.076 g of HAuCl<sub>4</sub>·3H<sub>2</sub>O and 0.15 g of 1,10-phenanthroline monohydrate was combined and dissolved in 2.0 mL of dimethyl sulfoxide (DMSO), with stirring for about 20 min at room temperature. Following this, carbon black (69.6 mg) was introduced into the solution, and the resultant mixture underwent heating in water bath at 60 °C for

5 h with continuous stirring. The resulting dispersion was later subjected to heating at 80 °C in air for 12 h to remove the DMSO, resulting in the formation of a black solid product. The produced black solid was gently grounded using a mortar and pestle, then moved into a ceramic crucible and positioned inside a tube furnace for pyrolysis. The black product underwent pyrolysis under nitrogen gas at 600 °C for a duration of 5 h.

#### Integration of aptamer on Au-SAzyme surface

Au-SAzyme- thrombin binding ssDNA: To prepare a 100  $\mu$ g/mL solution of dispersed Au-SAzyme, 0.01 g of Au-SAzyme powder was dispersed in 100 mL of deionized water. Subsequently, 1.0 mL of the 100  $\mu$ g/mL Au-SAzyme solution was combined with 0.5 mL of 1.0  $\mu$ M thiolate terminal single-stranded DNA, stirred for 15 min, and then incubated for 1 h at room temperature.

Au-SAzyme- CRP binding RNA: 1.0 mL of  $100 \ \mu g.mL^{-1}$  of Au-SAzyme was mixed with 1.0 mL of 0.46  $\mu M$  thiolate terminal RNA (same as ssDNA) and stirred for 15 min, followed by incubation for 1 h at room temperature.

#### **Colorimetric analysis**

For the thrombin colorimetric analysis: 100  $\mu$ L of the mixture of Au-SAzyme/ssDNA probe was taken and mixed with certain amount of the thrombin and 100  $\mu$ L of 5 mM of TMB, 100  $\mu$ L of 30 mM H<sub>2</sub>O<sub>2</sub> added the volume was completed to 1 mL by buffer pH 4.

For the CRP colorimetric analysis: same as for thrombin analysis, 100  $\mu$ L of the mixture of Au-SAzyme/RNA probe was taken and mixed with certain amount of CRP and 100  $\mu$ L of 5 mM of TMB, 100  $\mu$ L of 30 mM H<sub>2</sub>O<sub>2</sub> added the volume was completed to 1 mL by buffer pH 4.

#### **DFT and Monte Carlo simulations**

Ab-initio calculations were conducted utilizing plane-wave basis and pseudopotential methods within the framework of Density Functional Theory (DFT). All computations were performed using the Dmol3 code and adsorption locator in material studio provided by BIOVIA. Nonlocal-functional generalized gradient approximations (GGA) have been applied. A  $4 \times 4 \times 2$  k-point Monkhorst–Pack grid was used to optimize the structure, as well as the SCF tolerance was  $1*10^{-6}$  eV/atom. For thermodynamic properties calculation  $3 \times 3x1$  k-point Monkhorst–Pack grid was applied, double numerical plus polarization and double numerical plus d-functions basis sets were used respectively. Monte Carlo (MC) simulations were performed to acquire more information about the interaction between the AuN4@C surface with hydrogen peroxide and reactive oxygen molecules. The absorption sites and energy (Eads) were calculated using adsorption locator code and universal force field integrated which carried out with Monte Carlo method. We have established that the entire system was in equilibrium until temperature and energy were equal. For this reason, the full simulation was carried out with AuN<sub>4</sub>@C (001) slab model with a lattice parameter of  $a = b = 29\text{\AA}$  and  $c = 20\text{\AA}$  in order to avoid the effect of interaction between layers.

## **Results and discussion**

#### Characterizations

Figure 1A shows Raman spectrum of the Au-SAzyme biocatalyst. Two well-resolved peaks are shown in the spectrum locating at 1350 and 1500 cm<sup>-1</sup>, which are normally assigned to D and G bands. The G-band (1597 cm<sup>-1</sup>) is related to stretching Eigen mode of the C–C bond in graphitic structure, and indicative of a typical sp<sup>2</sup>-hybridized carbon network. The D band at 1350 cm<sup>-1</sup> arises from a hybridized vibrational mode linked to graphene edges, signifying the existence of structural disorder in the graphene [83, 84].

The X-ray powder diffractometer (XRD) analysis of the as-synthesized materials confirmed that the sample synthesized with Au-SAzyme exhibits a graphitic packing structure. (Fig. 1B) [82, 85] The determination of the composition and electronic state of surface elements in the Au-SAzyme catalyst was initially conducted through X-ray photoelectron spectroscopy (XPS). The survey spectrum (depicted in Fig. 1C) exhibits distinct peaks corresponding to C, N, and Au elements. In the Au-SAzyme, carbon atoms are predominantly found in sp2-hybridized structure. (Fig. 1F). The Au XPS spectra were recorded under these optimized conditions (Fig. 1D). The high-resolution Au-4f XPS. A thorough analysis of the XPS spectra unveiled a notable presence at a binding energy (BE) of 83.38 eV, a value typically associated with Au<sup>0</sup>.On the contrary, also the Au<sup>3+</sup> species, located at 86.90, eV. The other two peaks in 83.13 and 86.76 are related to Au\* [86]. These peaks confirm presence of are a mixture of Au<sup>3+</sup>, Au<sup>0\*</sup> and Au<sup>0</sup>.

The high-resolution N 1 s spectrum (Fig. 1E) exhibits three peaks at 399.23, and 397.86, 397.02 eV, related to C = N, C-N–C, and C-NH<sub>2</sub>, respectively [87, 88]. Carbon atoms in Au-SAzyme are mainly present in sp2 -hybridized structure (Fig. 1F). According to the convoluted spectrum of C, two small peaks at 285.67 and 283.93 eV can be assigned to the C 1 s orbital of C=C and C-N respectively [89, 90].

Figure 2 illustrates the verification of the unique presence of individual gold atoms on the nitrogen-doped carbon platform using HAADF-STEM and the corresponding EDX mapping.



Fig. 1 A In situ Raman spectra of Au-SAzyme measured, B X-ray diffraction spectra, the X-ray diffraction spectra pattern, XPS spectra of Au-SAzyme, C XPS Survey spectrum, D Au-4f spectra, E N1s spectra, and F C 1 s spectra

To determine the elemental composition, EDX analysis was carried out on the Au-SAzyme, as shown in Fig. 2A-E, as well as area was 10 nm of mapping on Au- SAzyme. In the EDX analysis, it is shown that the co-existence of Au, C, and N corroborates the composite of gold single atoms hosted on N-doped carbon, derived by the controlled pyrolysis of Au-complex with 1,10-phenanthroline adsorbed on charcoal, under N<sub>2</sub> gas at 600 °C formation. HAADF-STEM was performed to observe dispersed isolated atoms (Fig. 2F and 2G), and is obvious that Au atoms are well-dispersed and isolated as indicated by are indicated by red circles, and they size are around 0.2 nm which is exactly close to atomic size of gold [91, 92]. Thus, HAADF-STEM images combined with EDX mapping confirmed the formation of isolated Au atoms.

#### **Enzymatic activity of Au-SAzyme**

Following formation of isolated Au atoms on the N-doped carbon platform, enzymatic activity was evaluated. Figure 3A (blue line) shows the enzymatic activity of Au-SAzyme (10  $\mu$ g/mL) using TMB as a chromogenic color in the presence of H<sub>2</sub>O<sub>2</sub>. Au-SAzyme showed remarkable excellent peroxidase-like activities. The best concentrations of TMB (substrate) were studied and it was between 0.16 and 1.6 mM for 10  $\mu$ g mL<sup>-1</sup> of Au-SAzyme as shown in Figure S1 (supporting information). Furthermore, kinetic parameters of peroxidase property of Au-SAzyme were investigated. For an enzyme activity, the Michaelis–Menten saturation curve illustrates the link between substrate

concentration and reaction rate. The kinetic parameters explain this relationship [93, 94]. The SAzyme's activities with different substrate concentrations were plotted, and the Michaelis–Menten's constants (Km) and the maximal reaction velocity (Vmax) were determined using the equation  $1/\nu = \text{Km/Vmax} \cdot [S] + 1/\text{Vmax}$ , where  $\nu$  denotes the initial velocity, [S] represents the substrate concentration, and Vmax is the maximum reaction velocity. As it shows in Fig. 3B, the value of Vmax was 5.5 nM min<sup>-1</sup>, and Km was 1.7 nM. The Km value obtained is exceptionally minute, and to the best of our knowledge, it represents the lowest reported in the literature (Table 1).

#### Proposed mechanism of the efficient peroxidase

Highly catalytic peroxidase was investigated based on the DFT and Monte' Carlo calculation as performed in the literature [101–104]. As shown in Fig. 3, the first step is adsorption of  $H_2O_2$  on the coordinately unsaturated  $AuN_4$  site. At first, as shown in Fig. 3A to C to F, H2O2 molecule was absorbed on Au–N4 active sites (i). The energy of adsorption was 2.266 eV based on DFT (Fig. 3E). The activated  $H_2O_2$  molecules easily dissociated by homolytic path, resulting in the generation of 2OH\* (ii). Step ii with energy of 2.024 eV for cleavage of  $H_2O_2$  to 2OH\*. Here the OH\* can desorb and oxidize TMB to oxTMB. Another probable step (step iii) is formation of H<sub>2</sub>O and singlet oxygen (O\*) with desorption of O\* to oxidize TMN to oxTMB (iv) as shown in Fig. 3D-E.



**Fig. 2 A–E** Energy dispersive X-ray spectroscopy elemental mapping results of Au-SAzyme, indicating the Au (yellow), C (red), N (blue) elements homogeneously distributed in pyrolyzed sample.

 $\mathbf{F}$ ,  $\mathbf{G}$  HAADF-STEM image of Au-SAzyme, suggesting Au single atoms was atomically dispersed on the substrate at 5 nm and 2 nm respectively

## Applications

The catalytic activity of peroxidase mimic was found by oxidizing colorless chromogenic TMB to blue ox-TMB in presence of  $H_2O_2$ . When the S- terminal ssDNA was added to the Au-SAzyme, due to the strong gold thiol bonds (Au–S), the gold single atoms were covered and lost their peroxidase catalytic activity. Later, when thrombin was added, and due to the high affinity of the ssDNA to the thrombin, thrombinssDNA was formed and the Au–S bond was weakened, then the blue color was restored within 15–20 min the (Fig. 4A). For the quantitative determination of the thrombin, different concentrations of standard thrombin were added (0.7, 1.5, 2, 5, 10, 20, 50, 100 nM). The increase of blue color intensity was noted with an increase in thrombin concentration (see Fig. 4B). A linear correlation was established between absorbance and thrombin concentration, revealing a remarkably low limit of detection (LOD) at 550 pM and a limit of quantification (LOQ) at 1.8 nM, as illustrated in Fig. 4C.

For the CRP analysis, like thrombin analysis mentioned above, the blue colored ox-TMB via Au-SAzyme was converted to colorless with adding thiol terminal RNA to the solution but a rapid color recovery has been detected with in adding CRP to the Au-SAzyme/RNA probe with 6–10 min (Fig. 4D). Quantification analysis was conducted with various standard CRP concentrations of CRP from 0.1 **Fig. 3 A** The enzymatic activity of the Au-SAzyme and the carbon@nitrogen as a blank in presence of TMB and  $H_2O_2$ . **B** The kinetic study of Au-SAzyme which include the Michaelis–Menten and Lineweaver–Burk plot for Au-SAzyme. **C**, **D** Schematic presentation of peroxidase on AuN<sub>4</sub>. **E** Energy distribution diagram from DFT calculation. **F** Energy level representation from Monte Carlo simulation



 Table 1
 Comparison of the Km for different SAzymes, nanoparticles, and Au-SAzyme, where Km is the Michaelis–Menten constant

SAzyme	Km (mM)	References
FeN <sub>4</sub> -SAzyme	$1.07 \times 10^{-3}$	[77]
Fe <sub>3</sub> O <sub>4</sub> nanozyme	$3.49 \times 10^{-2}$	[77]
Cu–N–C SAzymes	3.76	[78]
Zn-N-C SAzymes	$1.07 \times 10^{-3}$	[78]
FeN <sub>3</sub> P-SAzyme	$2.06 \times 10^{-3}$	[95]
HRP enzyme	0.434	[96]
Co-N-C SAzymes	5.06	[97]
MNx-C (M=Mn, Zn) SAzyme	5.550	[98]
CuN <sub>4</sub> SAzyme	0.101	[99]
PtN <sub>3</sub> PS-SAzyme	0.310	[100]
Au-SAzyme	$1.7 \times 10^{-6}$	This work

to 50.0 ng mL<sup>-1</sup>. The blue color was directly increases with increase of CRP concentrations. The recovery will be close to the Au-SAzyme signal (Fig. 4E). A linear relationship was obtained between the absorbance and CRP concentrations with very small LOD as 500 pg mL<sup>-1</sup>, and LOQ was

165,000 pg mL<sup>-1</sup> with S/N ratio = 3. The LOD and LOQ were established using the criteria of Signal-to-Noise (S/N) ratios of 3 and 10, respectively. The signal (S) was determined relative to the standard deviation (SD) from ten measurements, taken from the current intensity of the lowest concentrations in the calibration, which were 0.7 nM for thrombin and 0.1 ng mL<sup>-1</sup> for CRP, respectively (Fig. 4F).

#### Selectivity study

As the human sera from COVID-19 patients serve as the matrix for the proposed probe, the typical ions, biomolecules, and proteins present in the serum matrix are considered. The possible interferents' concentration were 100 times more than that of the target species, thrombin, and CRP should be examined (Fig. 5). In this study, the thrombin concentration was 10 nM, other interferences were 1  $\mu$ M. CRP was 10 ng mL<sup>-1</sup>, other interferents were 1.0  $\mu$ g mL<sup>-1</sup>. All selectivity experiments were performed in the presence of 0.5 mM TMB, 0.3 mM H<sub>2</sub>O<sub>2</sub>. The absorbance intensity of thrombin and CRP was notably high compared to that of other proteins and metals.



**Fig. 4** Au-SAzyme responses: **A** and **D** blue lines: Au-SAzyme signal 0.5 mM TMB, 0.3 mM  $H_2O_2$  and 10 µg/mL of Au-SAzyme; black line: integration with 0.5 mL of 1.0 µM ssDNA and 1.0 mL of 0.46 µM RNA; red line: signal recovery by thrombin and CRP respectively. **B**, **E** UV–Vis absorption spectra of oxidation of TMB in the presence of Au-SAzyme/ssDNA verse thrombin at various con-

centrations, Au-SAzyme /RNA verse different concentrations of CRP respectively. **C**, **F** A graph plotting the absorbance at 652 nm against the concentrations of thrombin and CRP, respectively, with error bars depicting the standard deviation from three independent measurements



The selectivity tests have shown that the aptanoanzymebased ssDNA is a highly selective probe to thrombin (Fig. 5A). Similarly, the aptanoanzyme-based RNA probe is highly selective for the CRP (Fig. 5B). These findings show that the Au-Sazyme/ssDNA and Au-SAzyme/RNA sensor has a high thrombin and CRP selectivity respectively, making it a suitable probe for a human serum fluid sample.

#### **Thrombin and CRP detection**

The developed Aptamer/Au-SAzyme biosensor was utilized for individuals diagnosed with COVID-19. Serum and plasma samples from COVID-19 patients were gathered from the emergency and intensive care units at Hospital of Shahid-Hemn in the Slemani City, Kurdistan Region,



Scheme 2 The quantity of gold need for individual-based SARS-CoV-2 biomarkers (thrombin and CRP) run test based on Au-SAzyme/aptamer probe

Iraq. The collection of serum samples adhered to the ethical standards and regulations set by the hospital. The hospital administration and the patients themselves were duly informed. Standard concentrations of thrombin or CRP were introduced into the plasma and serum samples from the patients. The plasma samples underwent a 1000-fold dilution for thrombin determination, while the serum samples were diluted by a factor of 10, all serum samples were diluted using PBS reagent and underwent measurements in triplicate (n = 3).

Table 2 shows spiking recoveries for both COVID-19 biomarkers (thrombin and CRP) ranged for colorimetric assay from 90.5 to 98.9% for thrombin, and for CRP the spiking ranges was 89.6–103.2%, with RSD 0.45–0.81%, and 0.70–2.38% for thrombin and CRP respectively. The intra-assay precision achieves an RSD of 2% (n=3) within the same day, and the inter-assay precision (reproducibility) is 2% (n=3) across different days.

# Commercialization potential of the Au-SAzyme-based aptasensor

Due to maximum atom utilization, SAzymes are expected to have a great potential for commercialization. At first, 33 mg of Au was taken (from HAuCl<sub>4</sub>. 4H<sub>2</sub>O), and 0.6 g of Au-SAzyme was obtained after pyrolysis. Then, 0.01 g of the Au-SAzyme was then dispersed in 100 mL of water to prepared stock Au-SAzyme solution. 1.0 mL of Au-SAzyme stock solution was mixed with 1 mL of aptamer to make the probe. Form the probe, 100  $\mu$ L was used for running the tests analysis. Scheme 2 shows the consumption rout of gold from starting precursor to the running test.

Based on this calculation, using 33 mg of gold, one can test 120,000 sample of a biomarker. This means for 1 g of gold approximately 3,636,000 samples can be analyzed. This assigns for highly valuable and economic of our SAzymebased aptasensor for real applications. This ascribed to the maximum atom utilization of the catalyst and high peroxidase activity. Thus, we can confidently state that our probe is suitable commercialization purpose.

## Conclusions

Highly efficient peroxidase mimic from Au dispersed single atoms on black carbon was synthesized using very costeffective strategy. Highly selective and sensitive biocatalyst of well-dispersed Au single atoms were prepared and used as mimicking enzymes (SAzymes). Simple and high-yield synthetic method based on complexation-adsorption-pyrolysis was used successfully. HAAD-STEM showed the Au atoms are well-dispersed and isolated on the carbon platform. The Au-SAzyme showed remarkable low Michaelis-Menten constant (KM) of 1.7 nM. Thiolate aptamers for thrombin and CRP combined with SAzyme enhanced the molecular recognition selectivity and showed remarkable results thrombin and CRP detection in human serum with COVID-19 patients. LOD as small as 550 pM and as 500 pg.mL<sup>-1</sup> were calculated for thrombin and CRP, respectively. Interestingly, our Aptamer/Au-SAzyme probe is very economic and one can commercialize it as approximately 360,000 tests can be done with 1.0 g of gold.

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#### Declarations

Conflict of interest The authors declare no competing interests.

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