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DNAzyme-Au nanoprobe coupled with graphene-oxide–loaded hybridization chain reaction signal amplification for fluorometric determination of alkaline phosphatase

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Abstract

A sensing platform is presented for the determination of alkaline phosphatase (ALP) activity based on the cooperation of DNAzyme-Au spherical nucleic acid nanoprobe with the graphene-oxide–loaded hybridization chain reaction (HCR/GO) system to achieve good detection sensitivity and specificity. This assay takes advantage of the strong affinity of pyrophosphate (PPi) to Cu^{2+} ions and the fact that ALP can hydrolyze pyrophosphate (PPi) to release free Cu^{2+} ions. In the presence of ALP, the released Cu^{2+} can promote the Cu^{2+} -dependent DNAzyme to cleave the substrate that generates a shorter DNA fragment, which is responsible for further triggering the HCR/GO system to form a long fluorescence dsDNA and thereby giving an amplified fluorescence signal. Linear calibration range was obtained from 0.2 to 10 U L⁻¹, and the limit of detection (LOD) is about 0.14 U L⁻¹. The feasibility of the proposed method was validated by spiking ALP standards in bovine serum. The recovery ranged from 97.2 to 104.6%, and a coefficient of variation (CV) of less than 8% (n = 3) was obtained. This assay strategy was also applied to evaluate the ALP inhibitor efficiency, which indicates that the assay has potential for drug screening.

Keywords Alkaline phosphatase · DNAzyme · HCR/GO · Pyrophosphate · ALP inhibitor

Introduction

As a critical mediator of biological systems, alkaline phosphatase (ALP) actively participates in the physiological processes of the body, which plays an integral role in metabolism, signal transduction, molecule transportation, and genetic expression information [1–4]. The normal level of ALP activity in human serum is in a range of 40 to 150 U L⁻¹ [5]. Abnormal ALP activity has been universally recognized as a marker for various

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² Hunan Provincial Key Laboratory of Micro & Nano Materials Interface Science, College of Chemistry and Chemical Engineering, Central South University, Changsha 410083, China diseases such as bone disease, liver dysfunction, breast, and prostate cancer [4–7]. Of note, disease spectrums associated with abnormal ALP activity are still developing, which causes a higher demand for ALP activity detection to further understand ALP's role in physiological processes [5, 8, 9]. Therefore, developing analytical methods for measuring ALP activity with high sensitivity and precision is of considerable significance in clinical diagnoses.

To date, many strategies have been developed for ALP activity assay based on the dephosphorylation by ALP. In general, specially designed substrates such as *p*-nitrophenyl phosphate, 4methylumbellyferyl phosphate, and *p*-aminophenyl phosphate are dephosphorylated by ALP to produce colored (*p*-nitrophenol), highly fluorescent (4-methylumbellyferyl), or electroactive (*p*-aminophenyl) products that are straightly detected by spectroscopic or electrochemical methods [6, 10, 11]. Despite being characterized by straight-forward and fast response, these methods are limited to clinical tests due to low sensitivity. Apart from the synthetic small molecules that contain phosphate groups, pyrophosphate (PPi), a natural substrate of ALP is often utilized for ALP activity assay based on the strong binding of PPi to specific metal ions (including Al³⁺, Zn²⁺, Cu²⁺, Fe³⁺, and Ce³⁺) and the fact that ALP can efficiently catalyze the hydrolysis of PPi to phosphate (Pi), a process that generates uncomplexed metal ions [6, 12–14]. According to the design, the amount of released metal ions reflects the activity of ALP and is consistent with the concentration of ALP in the samples, which is determined by the quenching or stimulating influences on various foreign fluorescent labels, such as quantum dots (QDs) [15], metal-based nanomaterials [12], and upconversion materials [16]. However, most of these methods suffer from false-positive signals as a consequence of its sensing mechanism toward metal ions [6], although they have made many achievements.

DNAzymes, which was obtained via in vitro selection, have received great significance due to their unique recognition ability toward specific metal ions and the catalytic properties [17]. Because of such high metal ion selectivity, in combination with the excellent chemical stability, synthetic accessibility, as well as ease of modification with various functional molecules, these DNAzymes have been a powerful tool in the design of sensors for many metal ions, such as Zn^{2+} , Cu^{2+} , UO_2^{2+} , and Pb²⁺ [18–21]. More importantly, as ssDNA strands, DNAzymes can be further incorporated into other nucleic acid amplification techniques (such as RCA, SDA, and hybridization chain reaction (HCR)) for enhancing the detection sensitivity [21–24].

Herein, taking advantage of these excellent properties of DNAzyme, we report a sensing platform for ALP activity assaying. As shown in Scheme 1a, a Cu^{2+} -dependent DNAzyme (Cu-Enz)-functionalized gold nanoparticle (DNAzyme-Au) nanoprobe was used as the recognition component for monitoring the free Cu^{2+} ions generated via ALP-catalyzed hydrolysis of PPi. To construct this nanoprobe, 20 nm AuNPs was chosen as a scaffold for assembling and orienting the thiol-modified substrate stands into a dense arrangement, followed by hybridizing with the Cu-Enz. Upon hydrolyzation by ALP, free Cu^{2+} ions are present that activate the DNAzymes to cleave the intact substrate DNA strands (Cu-Sub), which results in the disassociation of a shorter DNA fragment (named as trigger-DNA) from the AuNPs due to lack of thermal stability.

However, regarding the DNAzyme-Au nanoprobe, the fluorescence signal was switched on in a one-to-one form with the ALP catalytic event, which would limit the detection sensitivity. Therefore, a graphene-oxide–loaded hybridization chain reaction (HCR/GO) system that was initiated by the trigger DNA, was incorporated with nanoprobe for tandem signal amplification. The kinetics-controlled HCR reaction is an ideal signal amplification tool for detecting specific nucleic acids because it does not require any enzyme reactions [25]. At the same time, the graphene oxide (GO) can be used as the carrier to accumulate the HCR products that enrich the fluorescence signal [26, 27]. As a result of the

amplification, the signal-to-background value (S/N) was improved more than 12-fold compared to using the single DNAzyme-Au nanoprobe. Incorporation of the HCR/CO system with the nanoprobe leads to a novel ALP activity assay platform that possesses excellent specificity and sensitivity.

Experiment section

Chemicals and instruments

Alkaline phosphatase from bovine intestinal mucosa was bought from Sigma-Aldrich. Chloroauric acid (HAuCl₄· 4H₂O, 99.99%) was obtained from Shanghai Chemical Reagent Company (Shanghai, China). The GO was bought from XFNANO Materials Tech Co., Ltd. (Nanjing, China), which was then ultrasound according to the instruction from the manufacturer to obtain the GO nanosheets (0.5 mg mL^{-1}) . Oligonucleotides used in this work were custom synthesized by Shanghai Sangon Biological Engineering Technology & Services Co., Ltd. (Shanghai, China), and the sequences and modifications are listed in Table S1. All the other chemicals were of analytical grade and obtained from commercial sources without further purification. The aqueous solutions of all the reagents were prepared using deionized water (\geq 18 M Ω , Watsons). Transmission electron microscopy (TEM) images of AuNPs was carried out on JEM-2100 TEM (JEOL Ltd., Japan). A Hitachi U-3010 UV-vis spectrometer (Japan) was used to measure the absorption spectra. Fluorescence signal was obtained on a Hitachi F-7000 spectrometer (Hitachi, Japan).

Preparation of DNAzyme functionalized AuNPs

Twenty-nanometer AuNPs was used to construct the DNAzyme-Au nanoprobe, and the synthesis process is shown in the "Supplementary information." AuNPs were first functionalized with thiolated Cu-Sub, followed by hybridizing with the Cu-Enz. Specifically, 10 µL of Cu-Sub (100 µM), 10 µL of TCEP (10 mM), and 1 µL of NaAc-HAc buffer (500 mM, pH 5.2) were mixed and kept for 1 h to reduce the disulfide bond. The reduced Cu-Sub was mixed with 750 µL of AuNP solution, adding 75 µL 20% Tween20, and incubated for 5 min; then adding 562 µL 5 M NaCl (2 M final concentration) to the mixture to shield the charge repulsion among the neighboring DNAs, thereby enhancing the DNA loading amounts. After the addition of NaCl, the following steps were repeated three times consisting of sonication for 2 min, followed by incubation for 30 min. Unconjugated DNA was separated from the AuNPs by centrifugation (13,000 \times g for 30 min), and the red precipitate was washed three times with 10 mM PBS buffer (pH 7.4)

Scheme 1 (a) Schematic illustration of ALP activity detection principle. (b) The structure of the DNAzvme



containing 0.3 M NaCl, which was then resuspended in 200 μ L of the same buffer. In the subsequent hybridization process, 10 μ L 100 μ M Cu-Enz was added to the Cu-Sub-AuNP solution. The solution was maintained at 55 °C for 5 min and allowed to cool to room temperature for 1 h to achieve full hybridization. The product DNAzyme-Au nanoprobe was isolated by centrifugation (13,000×g, 20 min), then washed twice with 10 mM PBS buffer (200 mM NaCl, pH 7.4) and finally resuspended in this PBS for further use.

Optimize the working concentration of Cu²⁺ and PPi

To achieve the optimized response to ALP, the optimal working concentration of Cu^{2+} ions was first explored by examining the intensities of the fluorescence signals arising from the DNAzyme-Au nanoprobe. A series of concentrations of Cu^{2+} stock solution was added into the nanoprobe solution, which was incubated for 20 min at room temperature, and the intensities of their fluorescence signals were recorded. The optimal concentration of PPi required to block Cu^{2+} was determined next; the solution of PPi with different concentrations was mixed with 10 μ M of Cu^{2+} , followed by incubation for 30 min at room temperature to enable a sufficient combination of PPi and Cu^{2+} . Then, FAM-modified nanoprobe solution (0.5 nM) was added to this mixture. Finally, the fluorescence changes were recorded after incubation for 20 min at 37 $^{\circ}$ C.

Preparation of HCR/GO system

The HCR/GO system was synthesized as previously described [26]. Five microliters of each FAM-labeled hairpin DNA (10 μ M) was first annealed to form the stable hairpin structure, which was mixed with a GO solution (500 μ g mL⁻¹ of 60 μ L) and incubated for 1 h at room temperature. The mixture was centrifuged for 15 min at 15,000 rpm to remove excess hairpin DNA strands. The sediments were resuspended in 100 μ L of PBS buffer and stored at 4 °C before used.

Detection procedure of ALP activity

In the typical experiment for ALP activity assay, 1 μ L of Cu²⁺ (1 mM), 6 μ L of PPi (1 mM), and 10 μ L of a sample containing a certain concentration of ALP were mixed in a reaction buffer (10 mM, Tris-HCl, 200 mM NaCl), making a total volume of 80 μ L. This mixture was incubated for 30 min at 37 °C for the ALP catalytic reaction. Then, 20 μ L DNAzyme-AuNP probe (final concentration was 0.5 nM) was added, followed by incubation for 20 min at room temperature. A solution (30 μ L) of HCR/GO (300 μ g mL⁻¹) and 3 μ L of

EDTA (10 mM) was finally added to the above solutions, and the resulting solution was held at 37 $^{\circ}$ C for 2 h before the fluorescence spectra were recorded. In the whole analysis process, the fluorescence spectrums were collected between 500 and 600 nm using the maximal excitation wavelength at 492 nm.

ALP detection procedure in serum samples

For ALP detection in bovine serum, 10 μ L bovine serum was spiked with 10 μ L of solutions containing varying concentrations of ALP and then diluted with 980 μ L of reaction buffer (10 mM, Tris-HCl, 200 mM NaCl) to give a total volume of 1 mL. One microliter of Cu²⁺ (1 mM) and 6 μ L of PPi (1 mM) were mixed with aliquots (73 μ L) of the diluted serum solutions (1%) containing ALP, which is then incubated at 37 °C for 30 min for the ALP catalytic reaction. The detection procedure was the same as described in the "Detection procedure of ALP activity" section.

Results and discussion

Characterization of the DNAzyme-Au Nanoprobe

Since the pioneering work on DNA-AuNP nanoconjugates by Mirkin [28], DNA-AuNP has been one of the most versatile hybrid nanomaterials for bioanalysis. Compared with those linear oligonucleotide-based sensors, the spherical nucleic acid nanoprobe that is heavily loaded with oligonucleotide has significant advantages in detecting sensitivity and specificity [29]. Thus, we utilize 20 nm AuNP as the scaffold, followed by functionalizing the AuNPs with DNAzyme and its substrate to construct the nanoprobe. The synthesis of AuNPs is shown in the "Supplementary information." Bare AuNPs were synthesized and characterized by TEM. The TEM image (Fig. S1a) indicates that the spherical AuNPs were well prepared with an average size of 20 nm. Subsequently, the Cu-Sub strands modified with a thiol group in 3' side were immobilized onto the AuNPs' surface via Au-S bond and then hybridized with Cu-Enz strands. To enhance the accessibility of the Cu-Sub strand to the Cu-Enz, a 20-thymine (T) spacer was added between the thiol moiety and the enzyme strands. The UV absorption spectra of the DNA-modified AuNPs (Fig. S1b) show a slight shift compared to the typical AuNP peak at 520 nm. But the DLS-determined hydrodynamic size of AuNPs changed obviously (Fig. S1c). For fluorophore-labeled DNAzyme-AuNP nanoprobe, the 5' end of the substrate is labeled with a FAM molecule whose fluorescence is quenched by the AuNP. The amount of hybridized DNAzyme covering the per AuNP surface was determined by a fluorescence-based method that employs a mercaptoethanol competing experiment. Fluorescence quantification assay based on the standard linear calibration curves of FAMlabeled substrate stands (Fig. S2) shows that there are approximately 140 hybridized DNAzyme strands on each AuNP. Thus, the parameters for the DNAzyme-Au nanoprobe is set at a stoichiometry of approximately 140 DNA strands per particle. Such a dense loading of DNAzymes and efficient hybridization between the DNAzyme and their substrate strand allows for maximum dynamic range.

Optimization of the concentration of Cu²⁺ ions

The free Cu^{2+} that is generated via ALP hydrolyzing of the PPi-Cu²⁺ complex can be regarded as an indicator to reflect the activity of ALP in samples. Therefore, the nanoprobes that can reliably monitor the concentration of Cu^{2+} is crucial for ALP activity assay. We thus investigated the fluorescence increment of the FAM-modified DNAzyme-AuNPs (0.5 nM particle concentration, corresponding to 70 nM substrate) in the presence of different concentrations of Cu^{2+} (Fig. 1a) to test the response of nanoprobe to Cu²⁺. As shown in Fig. 1b, the fluorescence intensity increased gradually with increasing concentration of Cu²⁺ and exhibited good linear correlation $(R^2 = 0.9923)$ when the Cu²⁺ concentration lies between 0.01 and 10 µM, suggesting that the DNAzyme-Au nanoprobe is reliable for monitoring the Cu²⁺ ions. While it should be noted that the fluorescence intensities decreased if the Cu²⁺ concentration was more than 30 µM, which is reasonable to believe that the high Cu²⁺ concentration would reduce the system stability. Therefore, the best performance of the DNAzyme-Au nanoprobe can be achieved at 10 μ M Cu²⁺. We also tested the nanoprobe in response to various biologically relevant metal ions (including Mg²⁺, Fe³⁺, Ca²⁺, K⁺, Zn²⁺, and Mn^{2+}) to evaluate its selectivity to Cu^{2+} . As Fig. 1c shows, 10 μ M of Cu²⁺ can induce about 60% cleavage efficiency of DNAzyme that is much higher than other competing metal ions, indicating that the nanoprobe has a good selectivity to Cu^{2+} .

Optimization of the concentration of PPi

We then investigated the effect of PPi concentration to obtain the optimum working concentration of PPi for ALP assay. The amount of PPi is a crucial factor that influences the background signal and detection sensitivity. In principle, an appropriate level of PPi should be required to completely block Cu^{2+} that prevents the DNAzyme-Au nanoprobe from being active in the absence of ALP. Hence, the working concentration of PPi was determined by using the optimized Cu^{2+} level. With the increase of PPi from 0 to 60 μ M, as shown in Fig. 1d, the fluorescence intensity of nanoprobe gradually decreases because PPi binds to Cu^{2+} . The fluorescence intensity reaches the lowest value at PPi concentrations above 60 μ M, implying that the Cu^{2+} can be completely bound when the concentration



Fig. 1 a Fluorescence response of DNAzyme-AuNPs at different concentrations of Cu²⁺ (λ_{ex} = 492 nm). **b** The relationship of the fluorescence enhancement with the Cu²⁺ concentration. Inset: the linear responses of DNAzyme-Au nanoprobe to Cu2+. Error bars show the standard deviations of three experiments. c Specificity of the nanoprobe over Cu²⁺ and

of PPi is above 60 µM. Then, 60 µM of PPi was thus used for the following experiment.

Studies of catalytic kinetic of ALP with DNAzyme-Au nanoprobes

The catalytic kinetic of ALP was studied by investigating the time-dependent fluorescence enhancements of the FAMmodified nanoprobe in the presence of different concentrations of ALP (Fig. 2). Within the reaction time of 30 min, the rate of fluorescence enhancement was increased with the increase of ALP concentration. The fluorescence spectrum of the DNAzyme-Au nanoprobe with different concentrations of ALP was recorded at 30 min (Fig. 3a), and a curve between fluorescence change ratio (F/F_0-1) and ALP concentration ranging from 0.2 to 10 UL^{-1} is plotted in Fig. 3b. An excellent linear relationship $F/F_0-1 = 0.293$ [ALP] + 0.580 ($R^2 =$ 0.9913) exists in this plot in the range of 0.2 to 10 U L^{-1} . However, within the nanoprobe detection system, each DNAzyme strand is only responsible for cutting one



600

580

Wavelength (nm)

b

500

n

800

600

400

200

0

520

d

1200

F-F_n (a.u.)

40

(a.u.

Intensity

⁻luorescence

0 µM

90 µM

560

20

540

R²=0.9923

60

Concentration of Cu2+ (µM)

Concentration of Cu2+ (µM)

80

10 20 30 40 50 60 70 80

Concentration of PPi (µM)

620

100

660

640



Fig. 2 Time-dependent fluorescence responses of DNAzyme-Au nanoprobe within 30 min in the presence of different concentrations of ALP



Fig. 3 a Fluorescence response of DNAzyme-Au nanoprobe after reaction of 30 min in the presence of different concentrations of ALP ($\lambda_{ex} = 492$ nm). b Plot of fluorescence change ratio (*F*/*F*₀–1) value vs. the

concentration of ALP that ranges from 0 to 40 U L⁻¹. Inset: plot of linear region from 0.2 to 10 U L⁻¹. Error bars show the standard deviations of three experiments

fluorescent substrate strand and releases one fluorophore from the AuNP surface, which would offer limited sensitivity.

The design of amplified sensing system for ALP

Therefore, a GO-load HCR system was further proposed as a signal amplification tool to improve detection sensitivity, in which a single trigger DNA that arises from the DNAzyme cleaves its substrate strand, could trigger the self-assembly of HCR. As strong as noncovalent binding of GO with nucleobases and aromatic compounds was reported [25-27], the GO was used to bind FAM-labeled hairpin DNA and quenched the fluorescence of FAM. The proposed HCR/GO system was constructed by absorbing two FAM-modified hairpin sequences (H1 and H2, respectively) onto GO surface via π stacking. Once the hybridization of the trigger DNA to hairpin DNA initiates the HCR, a long dsDNA product with accumulated fluorescence signals is formed followed by a desorption from GO nanosheet due to the weaker affinity between GO nanosheet and dsDNA. Based on this design principle, the DNA structures used for constructing the DNAzyme-Au nanoprobe was designed and schematically shown in Scheme 1b. The hybridization of Cu-Enz strand to Cu-Sub strand forms a complex asymmetric manner, in which the 3' portion of the substrate is hybridized with the DNAzyme through Watson-Crick base pairs, and the 5' part contains 8 bases that bind to DNAzyme via the formation of a triple-helix conformation. When treating with Cu^{2+} , the DNAzyme is activated, cleaving the substrate strand at the guanine site, generating two DNA fragments F1 and F2. The F1 is what we call the trigger DNA responsible for trigger HCR/GO, which was dissociated with arm 2 and released from the AuNP surface.

We thus verified if the protocol of the sensing platform with HCR/GO is competent for ALP assay, especially on

whether the HCR/GO could keep the nanoprobe silent in the absence of ALP. A fluorescence test was conducted to evaluate the leakage phenomenon of the sensing platform by coincubating the nanoprobe with HCR/GO. The result in Fig. 4a shows that there is no significant change in the fluorescence intensity of the system after the HCR/GO co-incubation with nanoprobe for 1 h, suggesting that the HCR/GO is inert to unactivated nanoprobe. We also performed the same experiments on the incubation of HCR/GO with PPi and Cu²⁺ separately. Similar results show that these individual components cannot cause interference for ALP assay. When ALP is introduced, the Cu²⁺ cofactor of DNAzyme was released by ALP hydrolyzing the PPi; the nanoprobe was activated to cleave the substrate strand and releases the trigger DNA, thereby initiating the HCR to recover the fluorophore that is quenched by GO. This process is demonstrated by the significantly increased fluorescence of the sensing system, which further confirmed that the sensing system is ALP activity dependent. More importantly, the enhanced fluorescent intensity accompanied by the DNAzyme-Au nanoprobe and the HCR/GO system illustrated higher sensitivity that could be obtained by the combination sensing platform.

Analytical performance of the sensing system for ALP activity assay

The sensitivity and selectivity of the entire assay were systematically investigated. The activity of ALP was plotted against the fluorescence change ratio (F/F_0-1), and they were fitted to a linear curve $F/F_0-1 = 4.230$ [ALP] + 1.534 ($R^2 = 0.9907$) with a dynamic range from 0.2 to 10 U L⁻¹ (Fig. 4b). The detection limit is estimated to be 0.144 U L⁻¹ based on $3\sigma/$ slope. Additionally, there is a nearly 12-fold enhancement of S/N ratio for the AuNP-HCR/GO sensing system compared with that of the DNAzyme-Au nanoprobe alone. We prepared





Fig. 4 a Fluorescence responses of control experiments under different conditions: (a) DNAzyme-Au nanoprobe+HCR/GO system (no ALP); (b) DNAzyme-Au nanoprobe+HCR/GO system (10 U L^{-1} ALP); (c) DNAzyme-Au nanoprobe (no ALP); (d) DNAzyme-Au nanoprobe (10 U L^{-1} ALP); (e) PPi+HCR/GO system; and (f) Cu^{2+} +HCR/GO system. For all experiments, the concentration of PPi and Cu^{2+} was 10 μ M

and 60 μ M, respectively. **b** Fluorescence spectra of the AuNP-HCR/GO system with addition of different concentrations of ALP. Inset: the linear curve between the fluorescence change ratio (F/F_0 -1) value and ALP concentration ranging from 0.2 to 10 U L⁻¹. Error bars show the standard deviations of three experiments

several interference proteins, including bovine serum albumin (BSA), horseradish peroxidase (HPR), glucose oxidase (GOx), human serum albumin (HSA), and protein kinase (PKA), to evaluate the selectivity of the sensing platform. As shown in Fig. S6, these five proteins at 10 μ g mL⁻¹ concentrations did not yield obvious fluorescence signals, whereas the ALP at a concentration 200 times lower (≈ 5 ng mL⁻¹) resulted in a large fluorescence increase, which demonstrates that the sensing platform is selective for ALP.

ALP activity assay in biological samples

We thus examined the response of the sensing system to ALP in diluted human serum (1%). However, the activity of ALP in unspiked bovine serum is almost undetectable though ALP is a component in the serum, which perhaps allows ALP to lose its activity during bovine serum storage. We further studied the recovery of this method by adding a known amount of ALP into the serum samples (Table 1). Varying concentrations of ALP was spiked into the bovine serum follow by diluting the sample to 1%, and an acceptable recovery ratio that ranges from 97.2 to 104.6% was achieved, with a coefficient of variation (CV) that is less than 8% (n = 3). These results indicate that our sensing platform could be potentially used to evaluate the ALP activity in practical biological samples.

Evaluate the ALP inhibitor efficiency

Moreover, the practicability of the sensing platform was demonstrated by its utility in screening the ALP inhibitors. Here, NaF was chosen as a model inhibitor, and its inhibition effect on the ALP was tested by using the developed method. ALP (10 U L⁻¹) was first treated with different concentrations of NaF for 30 min, and then the activity of NaF-treated ALP was measured. In Fig. 5, a profile was obtained by plotting the system fluorescence intensity against the NaF concentration. The measured IC₅₀ value of NaF was calculated to be about 1.1 ± 0.2 mM, which is consistent with previous reports [10, 30]. This result demonstrated that the sensing platform could be used to evaluate the inhibitor efficiency and may further be applied to screen other ALP inhibitors.

Table 1	Results of the recovery
experim	ent for ALP in spiked
bovine s	erum

Measured ALP ^a (U L^{-1})	SD^b	CV ^c (%)	Recovery (%)
_	_	_	_
0.52	0.04	7.59	104.6
1.94	0.10	5.26	97.2
8.14	0.28	3.44	101.8
	Measured ALP ^a (U L ⁻¹) - 0.52 1.94 8.14	Measured ALP ^a (U L ⁻¹) SD ^b - - 0.52 0.04 1.94 0.10 8.14 0.28	Measured ALP^a (U L ⁻¹)SD ^b CV^c (%)0.520.047.591.940.105.268.140.283.44

^a Mean of three measurements

^b Standard deviation of three measurements

^c Coefficient of variation = SD / mean \times 100



Fig. 5 a Fluorescence spectra of the sensing system with addition of different concentrations of NaF in the presence of 10 U L⁻¹ ALP ($\lambda_{ex} = 492$ nm). b Effect of NaF concentration on ALP inhibition efficiency. Error bars show the standard deviations of three experiments

Comparison of the current method to other techniques

Also, various sensing platforms with different mechanisms have been explored to determine ALP activity. To assess the performance of ALP detection, we compared this assay with other methods. As shown in Table S1, the LOD and linear range of ALP based on colorimetric, electrochemical, and fluorescence are summarized. Most of these strategies and our method acquired a satisfying analytical performance for determining ALP levels in biological samples. Compared with other methods, our sensing platform is a DNA-based fluorescence strategy, which shows several obvious advantages, such as simplicity for design and synthesis, low cost, and excellent biocompatibility. Meanwhile, given the recognition ability of DNAzyme-Au nanoprobe and HCR/GO to act as a signal amplifier, our sensing platform may be easily adapted to be extended to detect other targets using DNA hybridization technology. Nevertheless, the operation of this sensing platform in practical applications is slightly cumbersome since the analysis process requires the participation of two components.

Conclusion

In this work, we established a novel signal amplification sensing platform for highly sensitive ALP activity detection. The sensing platform was built on the mechanism that ALP can regulate the concentration of free Cu^{2+} released from the PPi- Cu^{2+} complex. The platform contains a DNAzyme-based nanoprobe as the Cu^{2+} ion recognition component and an HCR/GO system as the signal amplifier. These designs together allowed a high sensitivity for ALP assay and resulted in an approximately 12-fold increase in the S/N ratio compared with the DNAzyme-Au nanoprobe alone. The practical application in biological samples of this strategy was

confirmed by using a spiked bovine serum as the ALPpositive sample. This sensing platform contains the first successful integration of DNAzyme-Au nanoprobe with the HCR/GO system for the determination of ALP activity. Given the performance of high sensitivity and reliability, this sensing platform has been successfully used in evaluating the ALP inhibitor efficiency. The assay has the potential as a platform for clinical ALP assay or for screening new ALP inhibitors.

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Compliance with ethical standards

Conflict of interest The authors declare that they have no competing interests.

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