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Titanium(IV)-functionalized zirconium-organic frameworks as dual-metal affinity probe for recognition of endogenous phosphopeptides prior to mass spectrometric quantification

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Abstract

A zirconium-organic framework was modified with titanium(IV) ions to obtain a modified framework that is shown to be a viable sorbent for selective capture of phosphopeptides. This dual-metal affinity probe exhibits 0.1 fM limits of detection and excellent size-exclusion effect (the mass ratio of β-casein digests/BSA/intact β-casein is 1:1000:1000). This is attributed to abundant Ti(IV) and Zr(IV) coordination sites and high porosity. The performance of the sorbent for extracting endogenous phosphopeptides from human serum and saliva was investigated. Especially, 105 endogenous phosphopeptides from saliva were captured specifically. In addition, the amino acid frequency of the enriched phosphopeptides was analyzed. Conservation of sequence around the identified phosphorylated sites from saliva confirmed that phosphorylation took place in the proline-directed motifs.

Keywords Metal-organic frameworks . Nanomaterials . Post-functionalization . Dual-metal ions . Size-exclusion . Human saliva . Phosphoproteome . MALDI-TOF MS . Immobilized metal ion affinity chromatography . Post-translational modification

Introduction

Reversible phosphorylation is a key mechanism regulating signal transduction in biological processes. Many diseases are associated with abnormal phosphorylation of protein at specific sites [[1](#page-9-0), [2](#page-9-0)]. Global and exhaustive analysis of protein phosphorylation sites is crucial that can assist us in looking for potential biomarkers related to cancers. Mass spectrometry (MS)-based phosphoproteomics analysis becomes the most attractive strategy due to its high throughput [[3\]](#page-9-0). Substoichiometric abundance and low ionization efficiency of phosphopeptides, however, impedes direct detection through MS [[4](#page-9-0)–[8](#page-9-0)]. As a result, enrich-

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ment and separation of phosphopeptides prior to MS analysis is prerequisite.

Various strategies have been developed to specifically capture phosphopeptides from complex biosamples, including solid-phase extraction (SPE) [\[9](#page-9-0)], strong cation exchange chromatography (SCX) [\[10](#page-9-0)], metal oxide affinity chromatography (MOAC) [[11](#page-9-0)] and immobilized metal affinity chromatography (IMAC) [[12,](#page-9-0) [13\]](#page-9-0). Due to the high affinity between metal and phosphoate groups, MOAC and IMAC methods have been utilized for enriching phosphopeptides extensively. Each metal extracts a unique set of phosphorylated peptides, causing distinct phosphorylation profile for same organism [\[14\]](#page-9-0). To increase coverage of phosphorylated peptides, materials functionalized with various metal ions that exhibit complementary capture ability, have been exploited as phosphopeptides affinity probe (PPAP) [\[15\]](#page-9-0). In addition, high porosity that endows PPAPs with the function of size-exclusion is also vital $[16–18]$ $[16–18]$ $[16–18]$ $[16–18]$, especially for enriching endogenous phosphopeptides from complex biosamples. High abundance and large molecular weight proteins can always be found in them. It is necessary to fabricate novel porous materials with multi-metal ions

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for extracting endogenous phosphopeptides from organism.

Metal-organic frameworks (MOFs) have been developed rapidly in many fields, including gas separation [[19](#page-10-0)], catalytic degradation [[20\]](#page-10-0), luminescence sensing [\[21,](#page-10-0) [22\]](#page-10-0) and drug delivery [\[23](#page-10-0)]. Remarkably broad applications should be attributed to their high surface areas, tunable properties and further modification. MOFs have also been applied in phosphopeptides enrichment because of the existence of metal centers and high porosity [\[24](#page-10-0)]. However, single metal ion and limited coordination sites cause that extraction of phosphorylated peptides, through MOFs without further modification, has low selectivity [[17](#page-10-0)]. Immobilizing other metal ions on parent MOFs, which creates more abundant metal sites, provides a solution for such a predicament. For example, Peng et al. fabricated a novel dual-metal centered MOF for specifically enriching phosphopeptides via covalent-coordination co-operative modification of Uio-66-NH₂ [\[18\]](#page-10-0). Liu et al. synthesized core-shell multi-sites MOF as phosphopeptides affinity probe by titanium atom exchange with zirconium atom inside Uio-66 [[16\]](#page-9-0). Despite of the effort to prepare multi metal-based MOFs for PPAP, the above mentioned synthesized approaches need multi-steps or harsh reaction condition, which can lead to the collapse of frameworks. In a consequence, developing a novel modification strategy is extremely urgent for fabricating multi-metal MOF as PPAP.

Titanium(IV) is one of the most promising candidates for extracting phosphopeptides from intricate biosamples, because of its robust affinity with phosphoate. Consequently, titanium ions functionalized zirconiumorganic framework $(Ti^{4+}@Zr-MOF)$ was prepared at mild reaction condition through Ti^{4+} chelation to bipyridine chelation sites within Zr-MOF. This dual-metal PPAP that possessed more abundant coordination sites showed better phosphopeptides extraction efficiency compared with parent Zr-MOF. Due to the vast interaction with phosphopeptides and open 3D frameworks of Ti^{4+} @Zr-MOF, it exhibited ultra-low limit of detection and outstanding size-exclusion effect towards capture of phosphopeptides. As expected, for practical application, when human serum and saliva were incubated with $Ti⁴⁺@Zr-MOF, 4 endogenous phosphopeptides and 105$ endogenous phosphopeptides were identified respectively. Amino acid frequency of the captured phosphopeptides by $Ti^{4+}@Zr$ -MOF from saliva was analyzed. The results uncovered that phosphorylation occurred at proline-directed motifs, which provided the awareness of kinase recognition in salivary proteins. Ti^{4+} @Zr-MOF with robust affinity to phosphopeptides has great potential for large-scale phosphoproteomics research.

Experimental section

Materials and reagents

Zirconium tetrachloride (ZrCl4), terephthalic acid (TPA), tetraisopropoxytitanium (TIPT) and 2,2[']-bipyridine-5,5'-dicarboxylic acid (H₂bpydc) were purchased from Adamas-beta [\(http://www.adamas-beta.com](http://www.adamas-beta.com)). Trifluoroacetic acid (TFA), 2,5-dihydroxybenzoic acid (DHB), α-Casein, β-Casein, bovine serum albumin (BSA) and trypsin from bovine pancreas were purchased from Sigma-Aldrich ([https://www.](https://www.sigmaaldrich.com) [sigmaaldrich.com](https://www.sigmaaldrich.com), USA). Acetonitrile (ACN) was purchased from Merck [\(https://www.merckgroup.com](https://www.merckgroup.com), Darmstadt, Germany). Milli-Q water ([http://www.merckmillipore.com,](http://www.merckmillipore.com) Millipore, Bedford, MA) was used in all process. All of the other starting materials and reagents were also purchased from commercial sources and used without further purification. Human serum and saliva were obtained from Shanghai Zhongshan Hospital.

Instrumentation

The power X-ray diffraction (PXRD) were recorded with a Bruker D8 Advance diffractometer using CuK α radiation with 40 mA and 40 kV, with a scan range of 2 theta from 5° to 50° . Fourier transform infrared spectra (FTIR) were collected on a Nicolet IS10 infrared spectrum radiometer in the range of 4000–400 cm^{-1} using the KBr pellets. X-ray photoelectron spectra (XPS) were obtained on a Perkin Elmer PHI 5000C & PHI 5300 system by using the Mg $K\alpha$ anode. All binding energies were calibrated by using contaminant carbon(C 1 s = 284.6 eV). Nitrogen adsorption/desorption isotherms were measured at liquid nitrogen temperature using a Tristar 3020 analyser. Thermogravimetric analysis (TGA) was measured using a TGA 8000 system at a heating rate of 5 K min−¹ under nitrogen protection. Energy dispersive X-ray (EDX) was performed on Philips XL30. Transmission electron microscopy (TEM) images were taken on JEOL 2011 microscope.

Preparation of Zr-MOF and Ti^{4+} @Zr-MOF

Zr-MOF was synthesized according to previous reported method [\[25](#page-10-0)] with some modifications. Detailed procedures to synthesize Zr-MOF are displayed in Electronic Supporting Material. Ti^{4+} @Zr-MOF composite was prepared by soaking Zr-MOF (100 mg) into 20 mL methanol solution of $Ti(SO₄)₂$ (0.1 mM) for 6 h at room temperature under stirring. The resultant Ti^{4+} @ Zr-MOF was washed with methanol fully. The product was then dried at 50 \degree C for 12 h under vacuum.

Preparation of $Al^{3+}@Zr$ -MOF and Fe $3+@Zr$ -MOF

To compare the enrichment efficiency of Ti^{4+} with other metal ions, Al^{3+} and Fe^{3+} were also introduced into the parent MOFs (denoted as $Al^{3+}@Zr$ -MOF and $Fe^{3+}@Zr$ -MOF, respectively). $Al^{3+}@Zr-MOF$ and $Fe^{3+}@Zr-MOF$ was fabricated similarly to Ti^{4+} @ Zr-MOF besides that $Ti(SO_4)$ ₂ was displaced by the same mole of $AICl_3 \cdot 6H_2O$ and $FeCl_3 \cdot 6H_2O$ correspondingly.

Preparation of MIL-53(Al), MIL-53(Fe) and MIL-125(Ti)

To compare the enrichment efficiency towards phosphopeptides with single metal-based MOF, typical series MOF of Materials Institut Lavoisiers (MIL-53(Al), MIL-53(Fe) and MIL-125(Ti)) that have uniform organic linker terephthalic acid (TPA), were also prepared. MIL-53(Al), MIL-53(Fe) and MIL-125(Ti) were fabricated via solvothermal approach by mixing TPA and corresponding metal salts. For MIL-53(Al), 0.151 g AlCl₃ \cdot 6H₂O and 0.104 g TPA were mixed in 7.5 mL DMF. After being stirred vigorously, the mixture was transferred into 20 mL Teflonlined stainless steel container and heated at 180 °C for 12 h. The resulting powder was washed with DMF for three times and dried in vacuum. For MIL-53(Fe), the synthesized process was same as MIL-53(Al) except that the metal ions were replaced by Fe^{3+} . MIL-125(Ti) was prepared via reported protocol [[26](#page-10-0)] with some modifications. The concrete process to fabricate MIL-125(Ti) was also depicted in Electronic Supporting Material.

Digestion of standard proteins and preparation of human serum and saliva

The standard β-casein, α-casein phosphoprotein and BSA were dispersed into 25 mM $NH₄CO₃$ buffer and heated at 100 °C for 10 min. After cooling down to 25 °C, trypsin was added into the solution at an enzyme/substrate ratio of 1:40 (w/w). The mixture was then incubated at 37 \degree C for 16 h under shaking. The digests were diluted to certain concentration for further enrichment process.

Human serum and saliva sample were prepared in similar methods. 1 mL serum or saliva were diluted with 1 mL 0.2% TFA and centrifuged for 10 min at the speed of 5000 rpm. The supernatant were collected for further utilization.

Selective enrichment of phosphopeptides from standard peptides with Ti^{4+} @Zr-MOF

200 μg Ti⁴⁺ @Zr-MOF were dispersed into 100 μL loading buffer $(ACN/H₂O/TFA = 50/48/2)$ containing standard phosphoprotein digests. The suspension was incubated at 37 °C for 30 min. After being separated by centrifugation, the materials

were washed with 100 μL loading buffer for three times. Subsequently, the captured phosphopeptides were eluted by $10\% \text{ NH}_3 \cdot H_2$ O for 20 min. Finally, the eluent was mixed with matrix solution for MALDI-TOF analysis.

For complex biosamples, 200 μg Ti^{4+} @Zr-MOF was suspended in loading buffer which contains 10 μ L human serum or saliva. Then, the enrichment, washing and eluting process was the same as standard digests. The final eluent was analyzed by MALDI-TOF directly or lyophilized for next nano-LC-MS/MS analysis.

MS analysis and data search

Detailed procedures of MALDI-TOF and nano-LC-MS/MS analysis are shown in support information. Data search are also listed in support information.

Results and discussion

Choice of materials

Among the thousands of reported MOFs, Zr-based MOFs with excellent thermal and chemical stability are suitable for postfunctionalization. More important, they can be explored as sorbents for extracting phosphopeptides. On the other hand, H2bpydc as organic linkers can anchor other metal ions that is attributed to its 2,2′-bipyridyl moiety [\[27](#page-10-0)]. Therefore, the reaction of ZrCl4 with H2bpydc was carried out via solvothermal method to prepare parent Zr-MOF. To take advantages of free Lewis basic sites, titanium (IV) ions, possessing robust affinity with phosphoate and complementary extraction ability with Zr^{4+} [\[14](#page-9-0)], were encapsulated into Zr-MOF $(Ti^{4+}@Zr\text{-MOF})$ through chelation interaction. To confirm better capture efficiency of titanium (IV) ions towards phosphopeptides compared with other IMAC ions, $Fe^{3+}@Zr$ -MOF and $Al^{3+}@Zr$ -MOF were also synthesized and utilized for enriching phosphopeptides. In addition, three analogous single-metal MOFs (MIL-53(Al), MIL-53(Fe) and MIL-125(Ti)) were prepared as PPAPs too, which have uniform organic linker terephthalic acid (TPA) [\[26](#page-10-0)]. Their enrichment efficiency were compared with the corresponding dualmetal MOF in order to pinpoint the dual-metal synergistic effects.

Synthesis and characterization Ti^{4+} @Zr-MOF

The procedure for synthesis of post-functionalized $Ti^{4+}@Zr$ -MOF is presented in Scheme [1a](#page-3-0). In brief, zirconium-based MOF (Zr-MOF) possessing abundant chelation sites and robust rigidity was selected as the parent MOF. Subsequently, Ti^{4+} was introduced into Zr-MOF through coordination post-synthesized modification. Zr-MOF was fabricated via solvothermal strategy by mixing H₂bpydc and ZrCl₄ at 120 $\rm{^{\circ}C}$ according to previous

Non-Phosphopeptides help Phosphopeptides and Proteins

reports [\[25\]](#page-10-0). Power X-ray diffraction (PXRD) pattern (Fig. [1a\)](#page-4-0) agrees well with the simulated one, which confirms the successful synthesis of Zr-MOF. Ti⁴⁺@Zr-MOF was prepared via incorporating Ti^{4+} into bipyridine sites of Zr-MOF through postsynthesized modification (PSM). The PXRD pattern (Fig. [1a\)](#page-4-0) of Ti^{4+} @Zr-MOF is similar to the parent Zr-MOF, which suggests that the encapsulation of Ti^{4+} has little influence to the 3D frameworks. EDX (Fig. S1) was performed to certificate the successful introduction of Ti^{4+} . The atomic percentage and weight percentage of Ti^{4+} are 1.60 and 4.32 (Table S1), respectively. N₂ adsorption-desorption isotherms of Ti^{4+} @Zr-MOF are shown in Fig. $S2a$. It shows porosity towards N_2 and the Brunauer-Emmett-Teller (BET) surface area derived from isotherms is calculated as 158 m² g^{-1} . The pore size of Ti^{4+} @ Zr-MOF is 1.3 nm (Fig. S2b) that is consistent with the parent Zr-MOF which holds the tetrahedral cages and the octahedral cages with the diameter of 1.2 nm and 1.6 nm [\[25](#page-10-0)], respectively. The analogous pore size also indicates intact rigidity after Ti^{4+} was encapsulated into Zr-MOF.

In order to confirm Ti^{4+} is coordinated with bipyridine sites rather than physical absorption within the pore,

FTIR and XPS were performed. As shown in Fig. [1b,](#page-4-0) comparing with the IR spectra of Zr -MOF and Ti^{4+} @Zr-MOF, the stretching vibration of C=N is shifted from 1658 cm−¹ to 1689 cm−¹ . This should be attributed to the coordination interaction between Ti^{4+} and bipyridine sites, which hinders the vibration of C=N. Then, XPS analysis was also carried out to prove the interactions between loaded Ti^{4+} and bipyridyl moieties. The spectra of Ti^{4+} @Zr-MOF (Fig. [1c\)](#page-4-0) appear the peak of Ti $2p^3$ that can't be detected in the Zr-MOF, suggesting the successful encapsulation of Ti^{4+} . The binding energy of Ti $2p^3$ of Ti^{4+} @Zr-MOF is lower than the Ti(SO₄)₂ (Fig. [1d](#page-4-0)), implying the different coordination environment of Ti^{4+} . On the other hand, the binding energy of $N₁ s$ in the spectra (Fig. S3) of Ti^{4+} @Zr-MOF also has a slight shift compared to primitive Zr-MOF. The above results reflect that Ti^{4+} is supposed to coordinate with bipyridine sites within Zr-MOF instead of adsorption. Transmission electron microscopy demonstrates the octahedral-shaped crystals of Zr-MOF and it has nanoscale dimensions (Fig. [1e\)](#page-4-0). No obvious change of morphology was observed when Ti^{4+} is coordinated with bipyridine sites Fig. 1 a XRD patterns of Zr-MOF and Ti^{4+} @ Zr-MOF; b FTIR spectra of Zr-MOF and $Ti^{4+}@Zr$ -MOF; c XPS spectra of Ti $(SO_4)_{2}$, Ti⁴⁺ @ Zr-MOF and Zr-MOF; **d** Ti $2p$ XPS spectra of Ti $(SO₄)₂$ and Ti⁴⁺@Zr-MOF (e) TEM images of Zr-MOF; f TEM images of Ti^{4+} @Zr-MOF

within Zr-MOF (Fig. 1f). Zeta-potential analysis were also been operated to confirm the introduction of $Ti⁴⁺$. Zeta-potential of Ti^{4+} @Zr-MOF increased from 10.1 mV to 14.0 mV compared to parent Zr-MOF (Fig. S4). Thermogravimetric analysis (TGA) suggests that $Ti⁴⁺@Zr-MOF$ still possesses good stability even though $Ti⁴⁺$ is introduced into parent Zr-MOF (Fig. S5). For the sake of comparing the capture efficiency of single-metal PPAP with dual-metal PPAP, MIL-125(Ti), MIL-53(Al) and MIL-53(Fe) were also fabricated successfully (Fig. S6 and Fig. S7).

Enrichment of phosphopeptides from standard protein digests

In order to obtain best efficiency, the percentage of TFA (0.5%, 1%, 2%, 4%) within loading buffer and incubation time (10 min, 20 min, 30 min, 40 min) were optimized. As shown in Fig. S8, different percentage of TFA showed different results towards phosphopeptides enrichment. According to the peaks intensities, 2% TFA within loading buffer was selected. With the increment of incubation time before 30 min, the signal intensities are raised significantly (Fig.

S9). Increasing the incubation time from 30 min to 40 min, the peak intensities are increased limitedly. Therefore, 2% TFA within loading buffer and 30 min incubation time were used in subsequent experiment.

To investigate the capture efficiency of Ti^{4+} @Zr-MOF towards phosphorylated peptides through MS analysis, β-Casein digests and α -Casein digests were used as model phosphopeptides. At a concentration of 200 fmol· μL^{-1} β-Casein digests, without enrichment, the signal peaks of low abundance phosphopeptides were severely suppressed (Fig. 2a), and almost no phosphopeptides were detected. After treatment with Ti^{4+} @Zr-MOF, the mass spectra (Fig. 2b) were dominated by phosphorylated fragments with high S/N of 7385 and 12 phosphopeptides were counted. Then, α -Casein digests were also incubated with Ti^{4+} @Zr-MOF. As shown in Fig. S10, a total of 24 phosphopeptides were identified and almost all of the non-phosphopeptides were excluded, which were significantly enhanced compared with the α -Casein digests without enrichment. The above results suggested Ti^{4+} @Zr-MOF possesses high affinity with phosphortlated peptides that is attributed to abundant Ti^{4+} and Zr^{4+} coordination sites. Detailed sequence information of captured phosphopeptides from β-Casein digests and α-Casein digests are shown in Table S2 and Table S3, respectively.

To validate dual-metal centers Ti^{4+} @Zr-MOF has better enrichment performance compared with single-metal MOF, parent Zr-MOF and Ti-based MOF (MIL-125(Ti)), which only contain single metal ions, were also utilized to extract phosphopeptides from β-Casein digests. When treated with

Fig. 2 MALDI-TOF-MS for the phosphopeptides from 200 fmol· μL^{-1} β-Casein digests: a before enrichment; **b** after enrichment by T_1^{4+} @ Zr-MOF; Phosphopeptides were marked with \blacktriangledown and dephosphopeptides were marked with \triangle

Zr-MOF, although five phosphopeptides were detected (Fig. S11), the peak intensities were low and the spectra had interference of non-phosphopeptides. On the other hand, after treatment with MIL-125(Ti), non-phosphopeptides were observed obviously in the mass spectra (Fig. S12), revealing it had low specificity towards phosphopeptides. The results showed that $Ti^{4+}@Zr-MOF$ was superior to Zr-MOF and MIL-125(Ti) towards phosphopeptides extraction, which is due to the dual-metal synergistic effects within $Ti^{4+}@Zr$ -MOF. Complementary enrichment capability of dual-metal makes Ti^{4+} @ Zr-MOF captures more numbers of phosphopeptides.

To further confirm dual-metal synergistic effects play a crucial role in increasing enrichment efficiency of phosphopeptides, $Al^{3+}@Zr-MOF$ and $Fe^{3+}@Zr-MOF$ were fabricated via chelation interaction between bipyridine sites and metal ions. Then they were utilized for assessing the capture efficiency of phosphopeptides from β-Casein digests. Simultaneously, MIL-53(Al) and MIL-53(Fe) that belong to single-metal MOF were also synthesized successfully and used for comparison of enrichment performance with Al^{3+} @ Zr-MOF and Fe³⁺ @ Zr-MOF correspondingly. Al^{3+} @Zr-MOF and Fe³⁺ @Zr-MOF captured 8 and 9 phosphopeptides with high S/N exceeding 4000 from β-Casein digests, respectively. However, MIL-53(Al) and MIL-53(Fe) only captures 4 and 6 phosphopeptides, respectively (Fig. S13). In comparison with MIL-53(Al) and MIL-53(Fe), more numbers of phosphopeptides captured by $Al^{3+}@Zr-MOF$ and Fe³⁺@Zr-MOF should also be ascribed to the synergistic effects of the dual metals. Numbers of enriched phosphorylated peptides from standard β-Casein digests by all of the dual-metal MOFs and single-metal MOFs are listed in Table S4. The results clearly present enhanced enrichment efficiency of dual-metal integrated MOF compared with single-metal MOF.

Investigating limitation of detection of phosphopeptides (LOD) is also significant, because phosphorylated peptides are always at low abundance in biosamples. Ti^{4+} @Zr-MOF was incubated with different concentrations of β-Casein digests to test LOD. As depicted in Fig. [3,](#page-6-0) when the concentration of digests was reduced to 5 fmol· μL^{-1} , 6 phosphopeptide fragments with high signal intensity were determined. Three characteristic peaks of phosphopeptides (m/z = 2061, 2556, 3122) derived from β-Casein can still be observed with an S/N ratio of 25, while the concentration was 1 fmol· μL^{-1} . Even when the concentration was as low as 0.1 fmol· μL^{-1} , phosphorylated fragments (m/z = 2556) were also detected. The ultra-low detection towards phosphopeptides of Ti^{4+} @Zr-MOF is attributed to the abundant metal sites that provide robust interaction with phosphate.

In addition, the interference of other components in biosamples is also one of the main problems for enriching

Fig. 3 MALDI-TOF-MS for phosphopeptides enriched by $Ti^{4+}@Zr$ -MOF from β-Casein digests with different concentration: a 5 fmol· μL^{-1} ; **b** 1 fmol· μL^{-1} ; **c** 0.1 fmol· μL^{-1} . Phosphopeptides were marked with \blacktriangledown and dephosphopeptides were marked with \blacktriangledown

endogenous phosphopeptides. Although ions, biogenic amines and drugs can be founded in biological organism, they have lower molecular weight compared with peptides. In mass spectra, their signal suppresses phosphopeptide signal insignificantly. In a consequence, non-phosphopeptides having adjacent molecular weight are considered merely. BSA, typical non-phosphorylated protein, was digested and used as the interfering reagents. Phosphopeptides were inhibited completely by the peaks of BSA digests without enrichment (Fig. 4a), when the mass ratio was 40:1 (BSA: β-Casein). After being treated with Ti^{4+} @ Zr-MOF, however, the mass spectrum (Fig. 4b) was occupied by phosphopeptides and the peaks of BSA digests were almost vanished. Even when the ratio was increased to 80:1, two phosphorylated fragments were also detected with high intensity (Fig. S14), indicating the high selectivity of Ti^{4+} @Zr-MOF towards phosphopeptides. This is ascribed to strong interaction between metal sites and phosphopeptides and weak affinity with nonphosphopeptides under the optimal buffer condition.

Excluding large molecular proteins and enriching phosphopeptides simultaneously are also crucial, especially

Fig. 4 MALDI-TOF-MS for phosphopeptides from mixture of β-Casein digests and BSA digests at a mass ratio of 1:40: a before enrichment; b after enrichment by Ti^{4+} @Zr-MOF. Phosphopeptides were marked with \blacksquare and dephosphopeptides were marked with \blacksquare

for endogenous peptides. Ti^{4+} @Zr-MOF that remains the open 3D frameworks of parent Zr-MOF provides the ability for size-exclusion. Phosphorylated protein (β-Casein, MW = $24,000$) and non-phosphorylated protein (BSA, MW = 66,000) were employed as interference proteins to study the size-exclusion effect. As shown in Fig. [5](#page-7-0), no phosphopeptides were detected by MALDI-TOF directly, when the mass ratio of β-Casein, β-Casein digests and BSA is 1:800:800. After enrichment, the phosphorylated peptides in the eluent with high intensity appeared in the mass spectra. Notably, the signal of protein was not observed in the eluent (Fig. [5c\)](#page-7-0). In contrast, when the supernatant was analyzed using mass spectrum, the signal of protein was detected successfully. Increasing the ratio of mixture to 1:1000:1000, the mass spectra (Fig. S15) were also occupied by phosphopeptides when the washing buffer was analyzed. These results demonstrate that Ti^{4+} @Zr-MOF with high porosity has the ability of excluding large molecular protein off the composites and enabling the target phosphopeptides to interact with metal sites within the pore.

To evaluate the binding efficiency of the materials, $Ti⁴⁺@Zr-MOF$ was utilized for three cycles. As displayed in Fig. S16, the mass spectra were almost identical and have minor loss of phosphopeptides, indicating $Ti^{4+}@Zr-MOF$ has outstanding binding efficiency. To estimate the stability of Ti^{4+} @ Zr-MOF, it was utilized for enriching phosphopeptides again after being washed with methanol for several times and exposed to air for one month. As shown in Fig. S17, the mass spectra from β-Casein digests was analogous to that treated with freshly prepared material, confirming

Fig. 5 MALDI-TOF-MS from mixture of intact BSA, intact β-Casein and β-Casein digests at a mass ratio of 1:800:800: a without enrichment; **b** supernatant after enrichment by T_1^{4+} @Zr-MOF; c and (d) elution after

the good stability of the material due to robust chelation interaction between bipyridine sites and $Ti⁴⁺$.

Enrichment of phosphopeptides from complex biosamples

Excellent performance of Ti^{4+} @Zr-MOF towards capturing standard phosphopeptides motivates us to utilize the material for enriching endogenous phosphorylated peptides from complex biosamples. Human serum and saliva, both of which belong to clinical specimens and are easy to obtain, were utilized for assessing the enrichment efficiency of Ti^{4+} @Zr-MOF. The eluent were analyzed by MALDI-TOF MS and nano-LC-MS/MS simultaneously. As shown in Fig. S18, for human serum, after treatment with Ti^{4+} @Zr-MOF, four characteristic endogenous phosphopeptides peaks were observed with high intensity when the eluent were analyzed by MALDI-TOF MS. However, the mass spectrum was dominated by non-phosphopeptides before enrichment. The reports of nano-LC-MS/MS also provided the support. Detailed information of captured phosphopeptides from serum is shown in Table S5. The results were identical with the previous work [\[28\]](#page-10-0).

For human saliva, no phosphorylated peptides were observed without enrichment in the mass spectrum (Fig. 6a). After incubation with Ti^{4+} @Zr-MOF, however, 25 endogenous phosphopeptides were detected by MALDI-TOF MS. MS/MS analysis (Fig. S19) was performed to identify these peaks belongs to phosphopeptides, because mass loss of 98 of the dephosphorylated fragments were observed in the MS/MS spectra in comparison with the corresponding peaks in MS

enrichment by Ti^{4+} @Zr-MOF. Phosphopeptides were marked with and dephosphopeptides were marked with \bigstar

spectra. The reports (Table S6) of nano-LC-MS showed that a total of 105 endogenous phosphopeptides with 55 unique phosphorylated sites were identified directly from human saliva without enzymolysis. The detected phosphopeptides are consisted of 84 mono-phosphopeptides and 21 multiphosphopeptides.

In order to have an insight into the motif composition of the identified phosphorylated sites from human saliva, we aligned

Fig. 6 MALDI-TOF-MS for endogenous phosphopeptides from human saliva without enrichment (a) and after enrichment by Ti^{4+} @Zr-MOF (b). Phosphopeptides were marked with \blacktriangledown and dephosphopeptides were marked with \blacktriangleright

Fig.7 a Sequence logo of the motif of the identified phosphorylated sites from human saliva. b Distribution of phosphorylated sites in serine, threonine and tyrosine

the amino acid sequence from N-terminal to C-terminal. The conserved sequence patterns were obtained via Weblogo [[29\]](#page-10-0). As shown in Fig. 7, the distributions of identified phosphorylated sites (serine, threonine and tyrosine) were 70.9%, 21.8% and 7.3%, respectively. It was notable that proline appeared frequently on upstream or downstream of phosphorylated sites, which indicates proline-directed motif composition. There also was a high level of glutamic acid that belongs to acidic residue. The above amino acid frequency analysis near the identified phosphorylated sites suggests proline-directed and acidic-directed motif within salivary phosphoproteins that was also reported in previous research [[30\]](#page-10-0).

In comparison with other reported materials for phosphopeptides enrichment, results were summarized in

Materials	Method	LOD (β -casein)	Specificity (molar ratio of β-casein to BSA)	Size exclusion effect	Practical sample	Numbers of peptides	Ref.
Ti^{4+} @Zr-MOF	IMAC	0.1 fmol	1:80		Human saliva	105	This work
T ₂ M	MOAC&IMAC	10 fmol	1:800	X	Human saliva	30	$\lceil 31 \rceil$
Ti-PA-MNPs	IMAC	0.8 fmol	1:2000	X	Rat liver lysate	1568	$\lceil 32 \rceil$
SiO ₂ @PDA@Zr-MOF	MOAC	4 fmol	1:1000	V	Human saliva	240	$\lceil 33 \rceil$
$Fe3O4$ @mTiO ₂ -MSA	MOAC	0.05 fmol	1:800	V	Human saliva	307	$\left[34\right]$
Ti^{4+} @PDA@GA	IMAC		1:200	\times			$[7]$
Fe_3O_4 @PDA-Ti/Nb	IMAC	2 fmol	1:1000	\times	Non-fat milk	19	$[15]$

Table 1 The comparison of Ti^{4+} @Zr-MOF with other materials reported previously for extraction of phosphopeptides

Table [1.](#page-8-0) Ti^{4+} @Zr-MOF has advantages in LOD and sizeexclusion effect. Due to the complementary capture capability of Ti^{4+} and Zr^{4+} for phosphopeptides, Ti^{4+} @Zr-MOF performed quite well as sorbents for extracting endogenous phosphopeptides from human saliva. Of course, some challenges remain to be resolved. To detect the lower abundance of phosphorylated peptides, selectivity towards phosphopeptides of the material would be further advanced. In the future work, we will promote the specificity of the method in more complex biosamples.

Conclusion

A dual-metal centered Ti^{4+} @ Zr-MOF was fabricated successfully with a novel post-functionalization strategy. The sorbent was applied to capture model phosphopeptides and endogenous phosphopeptides from biological fluids. Ti^{4+} @Zr-MOF exhibits high sensitivity and high selectivity, which is due to the abundant $Ti⁴⁺$ and $Zr⁴⁺$ coordination sites endowing the sorbent robust interaction with phosphate group of phosphopeptides. The high porosity entrust Ti^{4+} @Zr-MOF with good size-exclusion effect. Remarkably, 4 endogenous phosphorylated peptides and 105 phosphopeptides were extracted from human serum and saliva, respectively. Amino acid frequency analysis near the identified phosphorylated sites from saliva uncovered phosphorylation occurred at proline-directed motifs obviously in salivary proteins. We believe Ti^{4+} @ Zr-MOF is promising for large-scale phosphoproteomics and further discovery of disease biomarkers via bioinformatics process.

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Compliance with ethical standards

Conflict of interest All the experiments in this work were carried out in compliance with the ethical standards, and conducted according to the Declaration of Helsinki and approved by the Ethics Committee of Fudan University.

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