



# A nanohybrid composed of Prussian Blue and graphitic C<sub>3</sub>N<sub>4</sub> nanosheets as the signal-generating tag in an enzyme-free electrochemical immunoassay for the neutrophil gelatinase-associated lipocalin

Fengling Zhang<sup>1</sup> · Hongbin Zhong<sup>1</sup> · Ying Lin<sup>1</sup> · Miaoxuan Chen<sup>1</sup> · Qingshui Wang<sup>2</sup> · Yao Lin<sup>2</sup> · Jiyi Huang<sup>3</sup>

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## Abstract

An enzyme-free electrochemical immunoassay is described for the neutrophil gelatinase-associated lipocalin (NGAL; a biomarker of kidney disease). Prussian Blue (PB) nanoparticles with redox activity were deposited on graphitic C<sub>3</sub>N<sub>4</sub> nanosheets (g-C<sub>3</sub>N<sub>4</sub>) by in-situ reduction. A screen printed electrode (SPCE) was modified with antibody against NGAL, and the PB-g-C<sub>3</sub>N<sub>4</sub> nanohybrid was used as the signal-generating tag for the secondary antibody against NGAL. Upon addition of target NGAL and of secondary antibody, a sandwich is formed on the SPCE. At an applied potential of typically 0.13 V (vs. Ag/AgCl), a well-defined voltammetric peak is observed that results from the presence of PB on the secondary antibody. Under optimal conditions, the peak current increases linearly in the 0.01 to 10 ng·mL<sup>-1</sup> NGAL concentration range, and the detection limit is 2.8 pg·mL<sup>-1</sup>. An average precision of <12% was accomplished in the batch-to-batch mode. Other disease-related biomarkers do not interfere. The accuracy and inter-laboratory validation of this method were evaluated for target NGAL detection in spiked human serum by using a commercial ELISA. The results obtained by the two methods are in good accordance.

**Keywords** Kidney biomarker · Transmission electron microscopy · Scanning electron microscopy · Method accuracy · Human serum specimens

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Fengling Zhang and Hongbin Zhong contributed equally to this work.

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✉ Yao Lin  
yaolin@fjnu.edu.cn

✉ Jiyi Huang  
hji0602@163.com

<sup>1</sup> Xiang'an Branch, The First Affiliated Hospital of Xiamen University, Xiamen University, Xiamen 361003, China

<sup>2</sup> Provincial University Key Laboratory of Cellular Stress Response and Metabolic Regulation, College of Life Sciences, Fujian Normal University, Qishan Campus, Fuzhou 350117, China

<sup>3</sup> Medical College of Xiamen University, Xiang'an Branch, The First Affiliated Hospital of Xiamen University, Xiamen University, Xiamen 361003, China

## Introduction

Neutrophil gelatinase-associated lipocalin (NGAL, a biomarker of kidney health, is a 24 kDa glycoprotein originally isolated from mouse kidney cells infected with a simian virus, SV-40. It shows promise in allowing a timely diagnosis of AKI since it is believed to be involved in renal development [1–3]. Although first discovered as a component of the late granules of neutrophils, NGAL is an indicator of kidney injury and released only when the kidney has come under duress from inflammation [4]. Urinary NGAL is produced in the renal tubule of the thick ascending limb of Henle and the collecting ducts [5]. Numerous studies have shown that the NGAL level in serum and urine is relative with the AKI severity. So, NGAL appears to fulfill many characteristics of an appropriate 'real-time' biomarker for AKI detection [6]. Early intervention significantly improves the prognosis.

Different schemes and strategies have been developed for the quantitative or qualitative determination of NGAL on the basis of various signal-generation principles, e.g., by up-

converting phosphor technology-based lateral flow assay [7], tangential flow filtration and ion-exchange chromatography [8] and solid-phase proximity ligation assay [9]. Li et al. presented a label-free photoelectrochemical immunosensor for the NGAL detection by utilizing a biotinylated anti-NGAL nanobody-oriented TiO<sub>2</sub> electrode [10]. Kannan et al. developed a label-free electrochemical immunosensor for the NGAL detection by immobilizing rabbit polyclonal lipocalin-2 antibody on gold nanoparticles attached on PAMAM dendrimer-modified electrode [11]. Among these methods, electrochemical immunosensor have attracted great attention because of specific antigen-antibody reaction [12, 13]. For successful development of high-efficient electrochemical immunoassays, one of the most concerns for obtaining low limits of detection and quantification is to exploit advanced signal-generation tags [14]. Routine protocols usually involve in ligand-conjugated enzyme labels or nano labels. Undoubtedly, natural enzymes (e.g., horseradish peroxidase and alkaline phosphatase) are often employed as the signal-generation tags [15–17]. Unfavorably, the catalytic efficiency can be adequately implemented in the presence of electron mediators (e.g., ferricyanide, thionine or ferrocene derivatives) during electrochemical measurement [18, 19]. In this regard, the detection system may be contaminated, and the signal generation is unavoidable of interferences. Enzyme-substrate reaction is susceptible to incubation temperature and inhibitory substances [20]. Hence, an alternative immunosensing strategy that is based on an electrochemical principle and does not require an enzyme label would be advantageous. Prussian blue (PB; a typical hexacyanoferrate coordination compound with a low energy gap) exhibits unique electrocatalytic function since each neighboring iron in the chemical structure has two various valances, Fe(II) and Fe(III), for building the skeleton with the CN group [21, 22]. Interestingly, every adjacent twelve Fe-CN-Fe edges of the unit cell can form a large cavity to allow the entrance of alkali metal ions (e.g., Na<sup>+</sup> and K<sup>+</sup>) for the increasing conductivity [23, 24]. To this end, our motivation of this study is to

synthesize nanoheterostructures with Prussian blue nanoparticles (PB-NPs) on two-dimensional g-C<sub>3</sub>N<sub>4</sub> nanosheets (i.e., a metal-free and nontoxic semiconductor with high thermal and chemical stability) thanks to atomic layer thickness, extraordinary physicochemical and biocompatible unique properties.

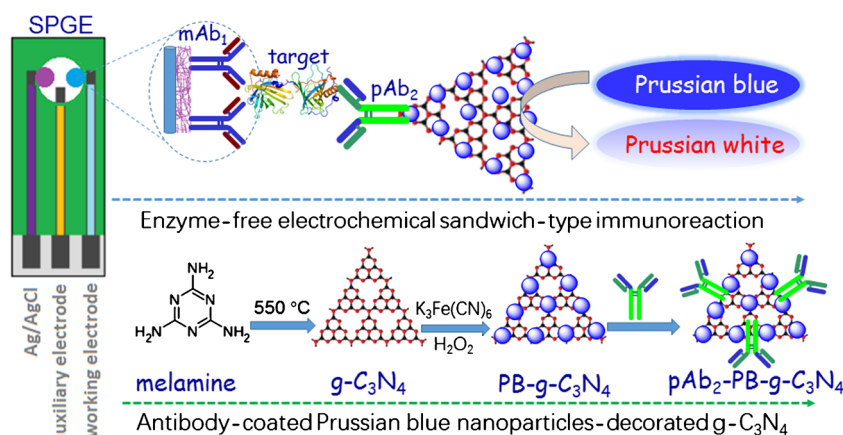
For facile ligand functionalization and biospecific interaction of electrochemical immunoassay, another important issue is to design a simple feasible sensing interface. Screen-printing technology seems to be one of the most suitable ways owing to its highly simplified reproducibility in serial production and inherent characteristics (e.g., miniaturization and inexpensive sensor production) [25]. The small-sized and integrated system with three electrodes in a single piece makes screen-printed electrodes ideal devices for fabrication of on-site electrochemical sensors [26]. Inspired by these advantages, we herein report on the proof-of-concept of simple and powerful immunosensing scheme for enzyme-free electrochemical detection of the NGAL (Scheme 1). PB-NP-decorated g-C<sub>3</sub>N<sub>4</sub> nanosheets are first synthesized and functionalized with the anti-NGAL secondary antibody, whilst the immunosensor is fabricated on screen-printed graphite electrode (SPGE) by conjugating capture antibody with BSA in a saturated glutaraldehyde vapor. Accompanying the formation of the sandwich-type immunocomplex, the voltammetric signal derives from the labeled PB-NPs with the redox activity within the applied potentials. This work aims to develop an enzyme-free and on-site amplified electrochemical immunosensing protocol for the sensitive detection of low-abundance disease-related biomarker without the requirement of natural enzymes.

## Materials and methods

### Chemicals and reagents

Monoclonal mouse anti-lipocalin-2/NGAL antibody (mAb<sub>1</sub>; Unconjugated; Reactivity: human; Clone: 5G5; Cat#:

**Scheme 1** Schematic illustration of the electrochemical immunoassay toward neutrophil gelatinase-associated lipocalin (NGAL) on monoclonal anti-NGAL antibody (mAb<sub>1</sub>)-coated screen-printed graphite electrode (SPGE) by using polyclonal anti-NGAL antibody (pAb<sub>2</sub>)-coated PB-NP-decorated g-C<sub>3</sub>N<sub>4</sub> nanosheets (pAb<sub>2</sub>/PB-g-C<sub>3</sub>N<sub>4</sub>)



ab23477) (note: Lipocalin-2 is also known as neutrophil gelatinase-associated lipocalin), polyclonal rabbit anti-lipocalin-2/NGAL antibody (pAb<sub>2</sub>; Unconjugated; Reactivity: human; Cat#: ab41105) and human lipocalin-2 ELISA kit (Reactivity: human; Cat#: ab21554; Sensitivity: 14.8 pg mL<sup>-1</sup>; Linear range: 46.9–3000 pg mL<sup>-1</sup>) including the NGAL standards were purchased from Abcam (Cambridge, MA, USA [www.abcam.cn](http://www.abcam.cn)). Glutaraldehyde solution (Grade I, 25 wt% in H<sub>2</sub>O), bovine serum albumin (BSA; Vetec™, reagent grade, ≥98%) and melamine were obtained from Alfa (Shanghai, China [www.alfa.com](http://www.alfa.com)). These materials were used for preparation of screen-printing graphite electrode as follows: cyclohexanone (CA, No. C10, 218–0; Sigma-Aldrich), graphite power (No. 50870; Fluka), silver-based ink (GEM-Gwent, Pontypool, UK) and solid paraffin substrate (Alfa, Shanghai, China [www.alfa.com](http://www.alfa.com)). All other chemicals were of analytical grade and used as received. Ultrapure water obtained from a Millipore water purification system at 18.2 MΩ·cm (Milli-Q, Millipore [www.merckmillipore.com](http://www.merckmillipore.com)) was used throughout this work. Phosphate buffered saline (PBS, 10 mM) with different pH values and containing 0.9% NaCl was used as the supporting electrolyte.

### Synthesis of Prussian Blue nanoparticle-decorated g-C<sub>3</sub>N<sub>4</sub> nanohybrids (PB-g-C<sub>3</sub>N<sub>4</sub>)

Prior to decoration with Prussian blue, water-dispersible g-C<sub>3</sub>N<sub>4</sub> nanosheets were first prepared on the basis of typical wet-chemistry method (Please see the process in the Electronic Supporting Material) [27]. Then, PB-NP-decorated g-C<sub>3</sub>N<sub>4</sub> nanohybrids (PB-g-C<sub>3</sub>N<sub>4</sub>) were synthesized via a hydrothermal method. Generally, 5.0 mL of aqueous mixture containing 2.5-mg g-C<sub>3</sub>N<sub>4</sub> and 1.5-g K<sub>3</sub>Fe(CN)<sub>6</sub> was kept into a vial under vigorous stirring for 30 min in an ice bath. Thereafter, 1.0 mL of 0.3 wt% H<sub>2</sub>O<sub>2</sub> (pH 2.0) was added to the mixture, and continuously stirred for another 3 h. Following that, the reaction mixture was washed with pure water, centrifuged to remove the remaining reagents and dried at 60 °C. The prepared PB-g-C<sub>3</sub>N<sub>4</sub> nanohybrids were stored in drying oven when not in use.

### Preparation of mAb<sub>1</sub> antibody-coated screen-printed graphite electrode (mAb<sub>1</sub>-SPGE)

First, screen-printed graphite electrode (SPGE) was prepared on a semi-automatic screen-printing machine (Please see the process in the Electronic Supporting Material). Then, the SPGE was modified with mAb<sub>1</sub> antibody by cross-linkage of the antibody with BSA in a saturated glutaraldehyde vapor. A homogenous mixture containing pAb<sub>2</sub> antibody (10 μL, 1.0 mg mL<sup>-1</sup>) and BSA (10 μL, 10 wt%) was firstly deposited on the active area of the working electrodes via a drop method,

then the resulting SPGE was placed in saturated glutaraldehyde vapor for 20 min, and then the resultant SPGE was dried at 4 °C for 12 h. Finally, the mAb<sub>1</sub>-SPGE immunosensor was stored at 4 °C for use.

### Immunoreaction protocol and voltammetric measurement

Scheme 1 represents schematic illustration of PB-g-C<sub>3</sub>N<sub>4</sub>-based electrochemical immunoassay for target NGAL on mAb<sub>1</sub>-SPGE, using pAb<sub>2</sub>/PB-g-C<sub>3</sub>N<sub>4</sub> as the signal tag with a sandwiched reaction format. Prior to measurement, conjugation of PB-g-C<sub>3</sub>N<sub>4</sub> nanohybrids with pAb<sub>2</sub> secondary antibody were prepared referring to the literature (Please see the process in the Electronic Supporting Material) [28]. The detection process toward NGAL was depicted as follows: (i) the mAb<sub>1</sub>-SPGE was incubated with 30-μL PSA standard/sample for 35 min at room temperature; (ii) the resulting SPGE was submerged into the suspension containing pAb<sub>2</sub>/PB-g-C<sub>3</sub>N<sub>4</sub> (200 μL) and incubated for another 35 min under the same conditions; and (iii) square wave voltammetric (SWV) measurement from -50 mV to 350 mV (vs. Ag/AgCl) (Amplitude: 25 mV; Frequency: 15 Hz; Increase *E*: 4 mV) was collected on an AutoLab workstation (μAutIII.Fra2.v, Eco Chemie, Netherlands [www.ecochemie.nl](http://www.ecochemie.nl)) and registered in a PBS (pH 6.8) as the signal of the immunosensor relative to NGAL level. After each step, the immunosensor was washed with pH 7.4 PBS. Impedimetric measurement was monitored at frequency 10<sup>-2</sup> to 10<sup>5</sup> Hz at the formal potential of 220 mV, using alternating voltage of 10 mV. A Nyquist plot (*Z*<sub>re</sub> vs. *Z*<sub>im</sub>) was drawn to analyze impedance results. All measurements were carried out at room temperature (25 ± 1.0 °C). The signals in all experiments referred to average responses of immunoreaction with corresponding standard deviations in triplicate, unless otherwise indicated.

## Results and discussion

### Choice of materials

To construct an electrochemical immunosensing platform for sensitive detection of low-abundance proteins, great attention has been focused on the signal amplification without the use of natural enzymes. Recent research has looked to develop innovative and powerful nanomaterials-based enzymatic mimics. In this case, the successful preparation of nanomaterials is very crucial in terms of reproducibility. Herein, mAb capture antibody is immobilized on the SPGE through a simple cross-linkage method with BSA in a saturated glutaraldehyde vapor, whilst pAb<sub>2</sub> secondary antibody is labeled to PB-NP-decorated g-C<sub>3</sub>N<sub>4</sub> nanosheets by the

physical adsorption and entrapped technique. PB-NPs are utilized as the indicators with redox activity during the electrochemical measurement without the requirement of natural enzymes. Relative to the redox-active organic polymer nanoparticles alone, e.g., poly(Methylene Blue) or simple Prussian blue nanoparticles, introduction of  $g\text{-C}_3\text{N}_4$  nanosheets is expected to enhance the amount of Prussian blue nanoparticles thanks to large surface coverage, thereby resulting in the amplification of the electrochemical signal. With a sandwich-type immunoassay format, the carried PB- $g\text{-C}_3\text{N}_4$  accompanying the secondary antibody increases with the increment of target NGAL. In this case, each of nanoparticle on the  $g\text{-C}_3\text{N}_4$  nanosheets produces an electronic signal within the applied potentials. The change in the signal relies on the concentration of target NGAL in the sample with high sensitivity.

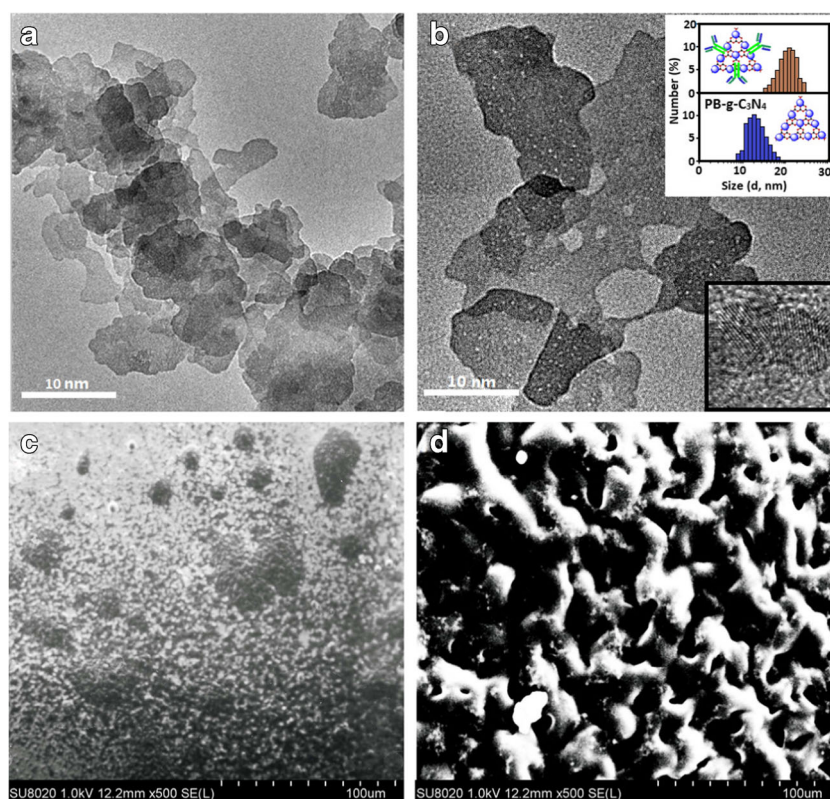
### Characterization of PB- $g\text{-C}_3\text{N}_4$ -based electrochemical immunoassay

To realize our design, preparation of  $m\text{Ab}_1$ -SPGE and  $p\text{Ab}_2$ /PB- $g\text{-C}_3\text{N}_4$  were very crucial for the successful development of the sandwich-type immunosensing system. Firstly, we used transmission electron microscopy (TEM, H-7605, Hitachi Instrumentation, Japan [www.hitachi.co.jp](http://www.hitachi.co.jp)) to characterize the synthesized  $g\text{-C}_3\text{N}_4$  nanosheets (Fig. 1a) before and (Fig. 1b) after modification with Prussian blue. Figure 1a shows typical TEM image of  $g\text{-C}_3\text{N}_4$  nanostructures, which exhibited a planar sheet-like morphology, suggested that the

nanoscales were neither particles nor nanotubes. Such a sheet-like structure provides a large surface coverage for the immobilization of the PB-NPs. In contrast, the morphology of  $g\text{-C}_3\text{N}_4$  nanosheets after reaction with  $\text{K}_3\text{Fe}(\text{CN})_6$  and  $\text{H}_2\text{O}_2$  was obviously different from  $g\text{-C}_3\text{N}_4$  alone (Fig. 1b vs. Fig. 1a). Numerous nanoparticles were coated on the surface of  $g\text{-C}_3\text{N}_4$  nanosheets. Vaguely we seemed to observe that the lattice and shape of PB-NPs, as seen from high-resolution transmission electron microscopy (HRTEM) (Fig. 1b, left inset). Since BSA protein was used for the synthesis of  $p\text{Ab}_2$ /PB- $g\text{-C}_3\text{N}_4$ , bioconjugation of  $p\text{Ab}_2$  antibody with PB- $g\text{-C}_3\text{N}_4$  nanosheets was very difficultly verified by using UV-vis adsorption spectroscopy and Fourier transform infrared spectroscopy. In this regard, we also used dynamic light scattering (DLS, Zetasizer Nano S90, Malvern, UK [www.malvern.com](http://www.malvern.com)) to characterize PB- $g\text{-C}_3\text{N}_4$  before and after modification with  $p\text{Ab}_2$  secondary antibody (Fig. 1b, top inset). Obviously, the average size of PB- $g\text{-C}_3\text{N}_4$  nanohybrids increased from  $11.3 \pm 1.2$  nm to  $20.6 \pm 1.5$  nm after coating with the antibody. These results revealed the successful preparation of PB- $g\text{-C}_3\text{N}_4$  for the labeling of  $p\text{Ab}_2$  secondary antibody.

The fabrication process of the  $m\text{Ab}_1$ -SPGE immunosensor was characterized by using scanning electron microscopy (SEM; Hitachi SU8020, Japan [www.hitachi.co.jp](http://www.hitachi.co.jp)). Figure 1c gives typical TEM image of the newly screen-printed graphite electrode by using the cutting technique with the coated working electrode. The surface was very rough, which was not comparative with commercial glassy carbon electrode.

**Fig. 1** Typical TEM images of **a**  $g\text{-C}_3\text{N}_4$  nanosheets and **b** PB- $g\text{-C}_3\text{N}_4$  nanohybrids [insets: (top) DLS data of PB- $g\text{-C}_3\text{N}_4$  and  $p\text{Ab}_2$ /PB- $g\text{-C}_3\text{N}_4$ ; (bottom) HRTEM image of PB- $g\text{-C}_3\text{N}_4$  nanohybrids]; TEM images of **c** the newly SPGE and **d**  $p\text{Ab}_2$ -SPGE



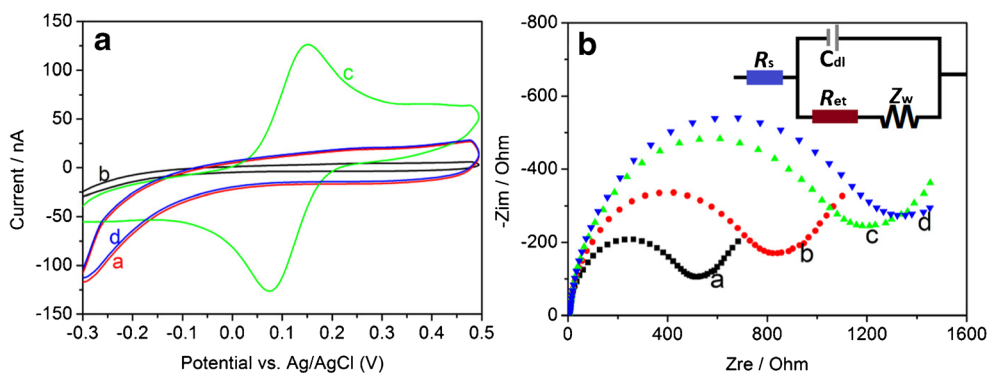
Actually, such a rough surface was favorable for the immobilization of biomolecules in this work by the cross-linkage of the antibody with BSA in a saturated glutaraldehyde vapor. As shown in Fig. 1d, some additional materials were attached to the screen-printed graphite electrode after using the vapor-deposition technique, which derived from the cross-linked pAb<sub>2</sub> antibody with BSA through the glutaraldehyde. Thus, the successful preparation of mAb<sub>1</sub>-SPGE and pAb<sub>2</sub>/PB-g-C<sub>3</sub>N<sub>4</sub> provided a necessary prerequisite for the development of electrochemical immunoassay.

### Evaluation of feasibility and control experiments

Logically, one question arises as to whether the mAb<sub>1</sub>-SPGE and pAb<sub>2</sub>/PB-g-C<sub>3</sub>N<sub>4</sub> can be used for the detection of target NGAL by using electrochemical detection technique. To verify this issue, cyclic voltammetric characteristics of the mAb<sub>1</sub>-SPGE were monitored in PBS (pH 6.8) at 50 mV s<sup>-1</sup> within the applied potentials from -300 mV to 500 mV after reaction with target NGAL and pAb<sub>2</sub>/PB-g-C<sub>3</sub>N<sub>4</sub> in sequence (note: 0.1 ng·mL<sup>-1</sup> NGAL used as an example in this case) (Fig. 2a). No cyclic voltammetric peaks of interest were obtained at both mAb<sub>2</sub>-SPGE (curve 'a') and mAb<sub>1</sub>-SPGE/NGAL (curve 'b'). Introduction of target NGAL on mAb<sub>1</sub>-SPGE caused the decreasing background currents, indicating that NGAL as a kind of biomacromolecules inhibited the electron communication between the solution and the electrode. When the resulting SPGE reacted again with pAb<sub>2</sub>/PB-g-C<sub>3</sub>N<sub>4</sub>, significantly, a pair of well-defined redox peaks were appeared at 82 mV and 143 mV (vs. Ag/AgCl), respectively (curve 'c'), which originated from the coated nanoparticles on g-C<sub>3</sub>N<sub>4</sub> nano-sheets. To further investigate whether the redox peaks derived from the nonspecific adsorption of the mAb<sub>1</sub>-SPGE toward pAb<sub>2</sub>/PB-g-C<sub>3</sub>N<sub>4</sub>, the immunosensor was used for direct incubation with pAb<sub>2</sub>/PB-g-C<sub>3</sub>N<sub>4</sub> in the absence of target NGAL. As seen from curve 'd', the cyclic voltammogram was almost

the same as that of mAb<sub>1</sub>-SPGE alone (curve 'a'). The result indicated that pAb<sub>2</sub>/PB-g-C<sub>3</sub>N<sub>4</sub> is not non-specifically adsorbed to the mAb<sub>1</sub>-SPGE, further suggested that introduction of pAb<sub>2</sub>/PB-g-C<sub>3</sub>N<sub>4</sub> on the mAb<sub>1</sub>-SPGE stemmed from the specific antigen-antibody reaction.

In addition to voltammetric characterization, the immunoreaction at each step were monitored by using electrochemical impedance spectroscopy (EIS) in 10 mM Fe(CN)<sub>6</sub><sup>4-/3-</sup> + 0.1 M KCl with the range from 10<sup>-2</sup> Hz to 10<sup>5</sup> Hz at an alternate voltage of 5 mV (Fig. 2b). A Randles equivalent circuit is fitted on the basis of the EIS data (Fig. 2b, inset). Typically, the equivalent circuit involves in four components including electrolyte resistance (*R<sub>s</sub>*), the lipid bilayer capacitance (*C<sub>dl</sub>*), charge transfer resistance (*R<sub>et</sub>*) and Warburg element (*Z<sub>w</sub>*). Among these components, the lipid bilayer capacitance and charge transfer resistance are dependent on the dielectric and insulating features at the electrode/electrolyte interface. Hence, the diameter of semicircular Nyquist diagram stands for the electron transfer resistance (*R<sub>et</sub>*). Plots 'a' gives the Nyquist diagram of the newly screen-printed graphite electrode, and the resistance was ~490 Ohm. After formation of mAb<sub>1</sub>-SPGE, the resistance increased to ~812 Ohm (diagram 'b'), implied that the coated mAb<sub>1</sub> capture antibody inhibited the electron transfer of Fe(CN)<sub>6</sub><sup>4-/3-</sup> from the solution to the SPGE surface. Further, the combination of target NGAL with capture antibody increased the resistance (diagram 'c'), which was in accordance with that of cyclic voltammetric result (Fig. 2a, curve 'b'). Although PB-NPs display certain electrochemical activity, reaction of pAb<sub>2</sub>/PB-g-C<sub>3</sub>N<sub>4</sub> with the analyte also caused the increase in the resistance (diagram 'd' vs. diagram 'c'). The reason might be attributed to the fact that the nanohybrids and the labeled pAb<sub>2</sub> antibody acted as the inert layer to inhibit the electron transfer. Based on the results of cyclic voltammetric and EIS measurement, we might make a conclusion that the mAb<sub>1</sub>-SPGE and pAb<sub>2</sub>/PB-g-C<sub>3</sub>N<sub>4</sub> can be employed for the detection of target NGAL.



**Fig. 2** **a** Cyclic voltammograms of (a) mAb<sub>1</sub>-SPGE, (b) mAb<sub>1</sub>-SPGE + 0.1 ng·mL<sup>-1</sup> NGAL, (c) sensor 'b' + pAb<sub>2</sub>/PB-g-C<sub>3</sub>N<sub>4</sub> and (d) mAb<sub>1</sub>-SPGE + pAb<sub>2</sub>/PB-g-C<sub>3</sub>N<sub>4</sub> at 50 mV·s<sup>-1</sup> in PBS, pH 6.8; and **b** Nyquist diagrams for (a) SPGE, (b) mAb<sub>1</sub>-SPGE, (c) mAb<sub>1</sub>-SPGE + 0.1 ng·mL<sup>-1</sup>

NGAL, (d) sensor 'c' + pAb<sub>2</sub>/PB-g-C<sub>3</sub>N<sub>4</sub> in 10 mM Fe(CN)<sub>6</sub><sup>4-/3-</sup> + 0.1 M KCl with the range from 10<sup>-2</sup> Hz to 10<sup>5</sup> Hz at an alternate voltage of 5 mV (inset: equivalent circuit)

## Optimization of method

The following parameters were optimized: pH value, concentration of PB-NPs and incubation times. Respective data and Figures are given in the Electronic Supporting Material. The following experimental conditions were found to give best results: pH 6.8 PBS; 5.0 mL of an aqueous mixture containing 2.5-mg g-C<sub>3</sub>N<sub>4</sub> and 1.5-g K<sub>3</sub>Fe(CN)<sub>6</sub> for preparation of PB-g-C<sub>3</sub>N<sub>4</sub>, and 35 min for the antigen-antibody reaction.

## Calibration plot

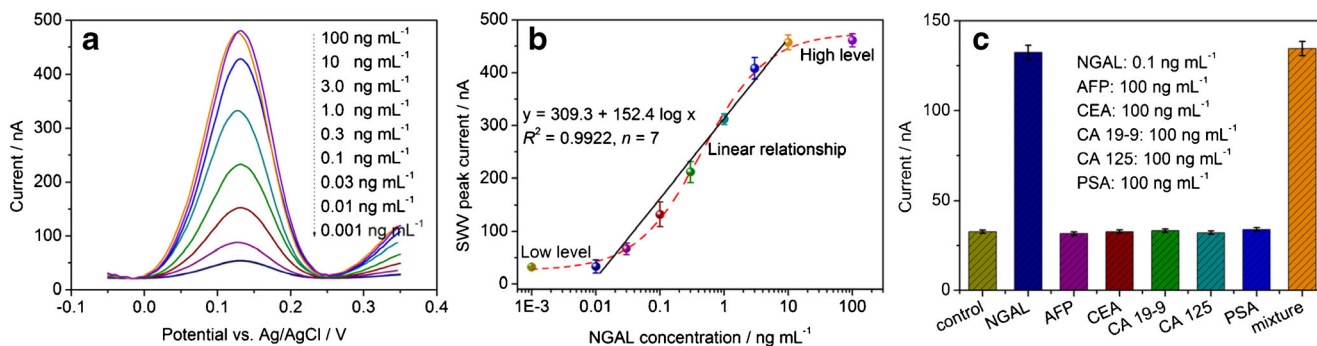
Under above-optimized conditions, the analytical properties of PB-g-C<sub>3</sub>N<sub>4</sub>-based electrochemical immunoassay were investigated toward target NGAL standards with different concentrations on the mAb<sub>1</sub>-SPGE by using pAb<sub>2</sub>/PB-g-C<sub>3</sub>N<sub>4</sub> as the signal-generation tag with a competitive-type assay mode. SWV measurement was carried out in pH 6.8 PBS after the sandwiched immunocomplex. As shown in Fig. 3a, the SWV peak currents increased with the increasing NGAL concentration in the detection solution. Within the range of low-concentration or high-concentration NGAL samples, the voltammetric responses were not obvious, and exhibited a sigmoidal “S” relationship between the peak current and NGAL level. A good linear relationship between SWV peak current (nA) and the logarithm of NGAL concentration (ng mL<sup>-1</sup>) was acquired within the dynamic range from 0.01 to 10 ng·mL<sup>-1</sup> (Fig. 3b). The regression equation can be fit as  $y$  (nA) = 152.4 × log C<sub>[NGAL]</sub> + 309.3 (ng·mL<sup>-1</sup>, R<sup>2</sup> = 0.9922, n = 7). The limit of detection (LOD) was estimated to 2.8 pg·mL<sup>-1</sup>, which was based on the 3σ/m criterion (where σ is the standard deviation of the blank and m is the slope of calibration plot). The sensitivity was ~544.2 μA·μM<sup>-1</sup>·cm<sup>-2</sup>. Moreover, the LOD of our strategy was comparable with commercial human lipocalin-2/NGAL ELISA kits from various companies [e.g., 5.0 pg·mL<sup>-1</sup> for Sigma (Cat#: RAB0332);

4.0 pg·mL<sup>-1</sup> for Abcam (4.1–1000 pg·mL<sup>-1</sup>; Cat#: ab113326); 14.8 pg·mL<sup>-1</sup> for Abcam (46.9–3000 pg·mL<sup>-1</sup>; Cat#: ab215541); 10 pg·mL<sup>-1</sup> for Abcam (156–10,000 pg·mL<sup>-1</sup>; Cat#: ab119600); 0.25 pg·mL<sup>-1</sup> for Abcam (41.1–90,000 pg·mL<sup>-1</sup>; Cat#: ab216816); 0.4 pg·mL<sup>-1</sup> for Abcam (1.52–3333.33 pg mL<sup>-1</sup>; Cat#: AB219024)], and other detection schemes (Table 1).

## Precision, reproducibility, specificity and stability

The precision and reproducibility of PB-g-C<sub>3</sub>N<sub>4</sub>-based immunoassay were studied toward NGAL standards within six repeatedly assays by using the same-batch or different-batch mAb<sub>1</sub>-SPGE and pAb<sub>2</sub>/PB-g-C<sub>3</sub>N<sub>4</sub>. Three NGAL standards including 0.01, 0.1 and 10 ng·mL<sup>-1</sup> were used in this case. The judgment was confirmed on the basis of relative standard deviation (RSD) during six analyses per sample. As indicated from experimental results, the RSDs was as follows: 9.2%, 4.3 and 5.6% (n = 6) for the intra-assays, and 10.7%, 9.8 and 11.2% (n = 6) for the inter-assays toward the aforementioned standards, respectively, thus suggested good reproducibility.

Next, the specificity of PB-g-C<sub>3</sub>N<sub>4</sub>-based electrochemical immunoassay was investigated toward target NGAL against other possible disease-related biomarkers in human serum, e.g., cancer antigen 15–3 (CA 15–3), alpha-fetoprotein (AFP), carcinoembryonic antigen (CEA), cancer antigen 19–9 (CA 19–9), cancer antigen 125 (CA 125) and prostate-specific antigen (PSA). These non-target analytes alone or co-existed with NGAL were determined by the electrochemical immunoassay. In this case, the ratio in the concentration between target NGAL and non-target analyte was 1: 100. As shown from Fig. 3c, the SWV peak currents toward non-target analytes were almost the same as that of background signal, and the strong current was obtained for target NGAL. Moreover, the coexistence of non-target analytes with NGAL did not significantly interfere the electrochemical



**Fig. 3** **a** SWV response curves of mAb<sub>1</sub>-SPGE-based immunosensor toward target NGAL with different levels from 0.001 to 100 ng·mL<sup>-1</sup> by using pAb<sub>2</sub>/PB-g-C<sub>3</sub>N<sub>4</sub> as the signal-generation tag in PBS (pH 6.8); **b** the corresponding calibration plots; and **c** the specificity of PB-g-C<sub>3</sub>N<sub>4</sub>-based electrochemical immunoassay was investigated toward

target NGAL against other possible disease-related biomarkers in human serum, e.g., cancer antigen 15–3 (CA 15–3), alpha-fetoprotein (AFP), carcinoembryonic antigen (CEA), cancer antigen 19–9 (CA 19–9), cancer antigen 125 (CA 125) and prostate-specific antigen (PSA) (note: The mixture contained the above-mentioned analytes) (n = 3)

**Table 1** An overview on recently reported nanomaterial-based electrochemical methods for determination of lipocalin-2/NGAL

Materials used <sup>a</sup>	Method applied	Linear range	LOD	Ref.
Upconverting nanoparticles	Lateral flow assay	7.68–1000 ng·mL <sup>-1</sup>	7.68 ng·mL <sup>-1</sup>	[7]
Enzymatic DNA ligation	Solid-phase proximity ligation assay	–	6.4 pg·mL <sup>-1</sup>	[9]
Nanobody	Photoelectrochemical immunosensor	0.001–500 ng·mL <sup>-1</sup>	0.6 pg·mL <sup>-1</sup>	[10]
AuNPs/PAMAM	Electrochemical immunosensor	50–250 ng·mL <sup>-1</sup>	1.0 ng·mL <sup>-1</sup>	[11]
PB-g-C <sub>3</sub> N <sub>4</sub>	Electrochemical immunosensor	0.01–10 ng·mL <sup>-1</sup>	2.8 pg·mL <sup>-1</sup>	This work

<sup>a</sup> AuNPs/PAMAM: gold nanoparticles attached on generation-1 poly(amidoamine) dendrimer

signal of the immunoassay. Hence, PB-g-C<sub>3</sub>N<sub>4</sub>-based immunoassay had good specificity.

Additionally, the lifetime of the mAb<sub>1</sub>-SPGE by using pAb<sub>2</sub>/PB-g-C<sub>3</sub>N<sub>4</sub> was also monitored by keeping them in refrigerator at 4 °C when not in use. They were used for the detection of 10 ng·mL<sup>-1</sup> NGAL (used as an example) every five days. Results indicated that the currents was maintained 98.1, 97.8, 95.3, 93.2 and 90.8% of the initial signal after storage of 20, 40, 60, 80 and 100 days with satisfactory stability.

### Monitoring of target NGAL in human serum samples and urine

To elucidate the method accuracy relative to commercial human lipocalin-2/NGAL ELISA kit, PB-g-C<sub>3</sub>N<sub>4</sub>-based electrochemical immunoassay was used for analysis of target NGAL in human serum samples and urine (note: Human serum specimens and urine were obtained from our hospital according to the rules of the local ethical committee; Informed consent was obtained from all individual participants including in this study). Typically, NGAL is secreted into the urine by the thick ascending limb of loop of Henle and collecting ducts of kidney, with synthesis occurring in the distal nephron [29].

Owing its small molecular size, NGAL is freely filtered and can be easily detected in urine. Thereafter, the results from our strategy were compared with those from commercial human lipocalin-2/NGAL ELISA kit by the application of an *F*-test and the linear regression equation between the average values per sample. As shown from Table 2, the *t*<sub>exp</sub> values in all cases were below *t*<sub>crit</sub> (*t*<sub>crit</sub>[4, 0.05] = 2.77), and the regression equation was fitted to  $y = 0.8982x + 0.1512$  ( $R^2 = 0.9977$ ,  $n = 8$ ) (note: *x*-axis for the electrochemical immunoassay and *y*-axis for human lipocalin-2/NGAL ELISA kit). The slope and intercept of the regression equation were close to the ideal '1' and '0', respectively. Thus, the method accuracy of our system was acceptable for the detection of complex biological fluids.

### Conclusions

In summary, this work demonstrated the development of an electrochemical immunoassay by using Prussian blue-decorated g-C<sub>3</sub>N<sub>4</sub> nanohybrids for signal amplification. Introduction of Prussian blue nanoparticles as the signal-generation tags efficiently avoided the participation of natural natures. The g-C<sub>3</sub>N<sub>4</sub> with high surface coverage provides a large room for conjugation of the indicator and biomolecule.

**Table 2** Evaluation of method accuracy with commercial human lipocalin-2/NGAL ELISA kit for human serum specimens and urine

Type	No.	Method; Conc. [mean ± SD (RSD), ng·mL <sup>-1</sup> , <i>n</i> = 3] <sup>a</sup>		<i>t</i> <sub>exp</sub>
		Immunosensor	ELISA kit	
Human serum	1	3.2 ± 0.11 (3.44%)	3.3 ± 0.09 (2.73%)	1.22
	2	0.12 ± 0.02 (16.67%)	0.16 ± 0.02 (12.5%)	2.45
	3	4.5 ± 0.3 (6.67%)	4.2 ± 0.2 (4.76%)	1.44
	4	12.3 ± 1.1 (8.94%)	11.1 ± 1.1 (9.91%)	1.34
Human urine	1	0.34 ± 0.02 (5.88%)	0.29 ± 0.03 (10.34%)	2.41
	2	1.2 ± 0.2 (16.67%)	1.4 ± 0.1 (7.14%)	1.55
	3	5.6 ± 0.6 (10.71%)	5.3 ± 0.5 (9.43%)	0.67
	4	2.4 ± 0.3 (12.5%)	2.1 ± 0.2 (9.23%)	1.44

<sup>a</sup> Analyses were made in triplicate according to the diluted ratio, and the data were obtained as mean values of three assays (*n* = 3)

Relative to graphene nanosheets, the existed N element in g-C<sub>3</sub>N<sub>4</sub> nanosheets enhanced electrocatalytic efficiency of the nanohybrids thanks to the lone-pair electrons [30, 31]. Although the present work focused on detection of target NGAL, it easily extends to determine other cancer biomarkers by controlling the corresponding antibodies. Nevertheless, one limitation of this work is the relatively long assay time per sample. Therefore, future work should focus on the improvement of the detection steps.

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