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# Electrochemical sensor based on a carbon nanotube-modified imprinted sol–gel for selective and sensitive determination of ß2-agonists

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Abstract We describe a molecularly imprinted electrochemical sensor for selective and sensitive determination of  $\beta_2$ -agonists. It is making use of a combination of single-wall carbon nanotubes (SWNTs) with a molecularly imprinted sol– gel. The SWNTs were introduced in order to enhance electron transport and sensitivity. The imprinted sol–gel film with its specific binding sites acts as a selective recognition element and as a preconcentrator for  $\beta_2$ -agonists. The morphology of the imprinted film was characterized by scanning electron microscopy. The optimized sensor displays high sensitivity and excellent selectivity for the  $\beta_2$ -agonists as shown for their determination in human serum samples.

Keywords Molecularly imprinted polymer . Single-wall carbon nanotubes  $\cdot$  Sol–gel  $\cdot$   $\beta$ <sub>2</sub>-agonists

## Introduction

 $\beta_2$ -agonists are phenylethanolamines with different substituent groups on the aromatic ring and the terminal amino group (Table [1\)](#page-1-0). They are used in the symptomatic treatment of asthma and chronic bronchitis. They are also used in the prevention of exercise-induced asthma [\[1](#page-6-0)]. However, the drug residues accumulated in animal tissues may pose acute

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poisoning when consumed by humans, with symptoms of muscular tremor, cardiac palpitation, nervousness, headache, muscular pain, dizziness, nausea, vomiting, fever, and chills [\[2](#page-6-0), [3](#page-6-0)]. For this reason, the use of  $\beta_2$ -agonists in animal feeds has been banned in most countries. So it is necessary to set up accurate and quick methods to detect  $\beta_2$ -agonists.

Several technologies including capillary electrophoresis [\[4](#page-6-0), [5\]](#page-6-0), chromatography [[6](#page-6-0)–[10](#page-6-0)], spectrophotometry [\[11](#page-6-0), [12\]](#page-6-0), fluorescence and chemiluminescence [[13](#page-6-0)], immunoassays [\[14](#page-6-0), [15](#page-6-0)] have been adopted for the determination of  $\beta_2$ agonists in the biological sample. The general drawbacks of capillary electrophoresis and chromatography include expensive instrumentation and time-consuming. Spectrophotometry tends to lack suitable antibodies, stability and repeatability for the determination of  $\beta_2$ -agonists in biological samples. All of these limitations have stimulated the research and development of new devices for quick and accurate detection of drug residues. A new trend in analytical chemistry is to develop novel sensors based on molecularly imprinted polymers (MIPs) or molecularly imprinted membranes [[16,](#page-6-0) [17](#page-6-0)]. Molecular imprinting is an excellent technique for the preparation of MIP with a tailor-made affinity and specificity for the template molecule [\[18](#page-6-0)–[20](#page-6-0)], and it has been widely used in several applications involving chromatography [\[21](#page-6-0)], solid phase extraction [\[22](#page-6-0)] and sensor [[23\]](#page-6-0). Specifically, MIP-based sensors attract increasing attention on account of the advantages including low cost, high specificity, stability and robustness [[24](#page-6-0)–[27\]](#page-6-0). Zhou et al. reported a chemiluminescence sensor using molecularly imprinted polymer as recognition elements for detection of salbutamol [\[28](#page-6-0)]. Zhao et al. prepared a molecular imprinted film based on chitosan/nafion/nano-silver/polyquercetin for clenbuterol sensing [\[29](#page-6-0)]. Andrea et al. developed a solid binding matrix/molecularly imprinted polymer-based sensor system

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## <span id="page-1-0"></span>Table 1 Linear relationships and sensitivity of  $\beta_2$ -agonists



for the determination of clenbuterol in bovine liver using differential-pulse voltammetry [\[30](#page-6-0)]. Liang et al. reported a potentiometric sensor based on molecularly imprinted polymers for rapid determination of clenbuterol in pig urine [\[31](#page-6-0)]. Among all of the developed imprinted materials, imprinted sol–gel, which is a polymeric material formed by the acid-or base-catalyzed hydrolysis and condensation of metal alkoxides, has been widely used in the development of imprinted sol–gel sensors.

The aim of this work is to construct electrochemical sensors for the detection of  $\beta_2$ -agonists by using molecularly imprinted technique and sol–gel rigidity network. SWNTs were introduced during the preparation of imprinted sensor owing to the unique properties involving huge surface area, subtle electronic properties and catalytic ability for enhancing the sensitivity of the sensor. The MIP electrochemical sensors have the advantage of convenience, high sensitivity and low detection limit. In addition, the sensors were successfully applied to the determination of  $β_2$ -agonists in human serum samples.

## Experimental

#### Reagents and materials

SWNTs were obtained from Chengdu Organic Chemicals Co. (Sichuan, China. [www.cioc.ac.cn](http://www.cioc.ac.cn/)). All  $\beta_2$ -agonists (Salbutamol, clenbuterol, terbutaline, Metaproterenol, Fenoterol, Cimbuterol, Mabuterol, Isoxsuprine, Brombuterol, Ractopamine and Ritodrine hydrochloride) were purchased from the National Institute for the Control of Pharmaceutical and Biological Products (China. [www.nicpbp.org.cn](http://www.nicpbp.org.cn/)). Tetraethoxysilane (TEOS) was purchased from Sigma (USA. [www.sigma-aldrich.com\)](http://www.sigma-aldrich.com/). Bistriethoxysilane endcapped polyurethane/urea (SPU) was obtained from College of Material Science and Engineering, Hubei University. All chemicals were of reagent grade and used as received. All solution was prepared with ultra-pure water (18.25 M $\Omega$  cm) by Aquapro Ultra-pure water system (Chongqi, China. [www.aquapro-china.com](http://www.aquapro-china.com/)).

## Apparatus and procedures

A computer-controlled electrochemical workstation (CHI660C, CH Instruments, Chenhua Co., Shanghai, China. [www.chinstr.com](http://www.chinstr.com/)) was carried out for voltammetry measurement. A conventional three-electrode system was used in the measurements, with a bare GCE (3 mm diameter) or a modified GCE as the working electrode, a saturated calomel electrode as the reference electrode and a platinum electrode as the auxiliary electrode. The pH value of electrolyte was determined by using a 320-S acidity meter (Mettler-Toledo, Switzerland. [www.mt.com](http://www.mt.com/)). A JEOL JSM-5510LV scanning electron microscopy (SEM, Japan. [www.jeol.com\)](http://www.jeol.com/) was applied for characterizing the sensors. All measurements were carried out at a room temperature.

#### Preparation of imprinted sensor

1.5 mg purified SWNTs were dispersed into 3.0 mL 5.0 mM sodium dodecyl sulfate by ultrasonic agitation for 40 min to give a homogeneous SWNTs suspension. Then, 5.0 μL of SWNTs suspension was coated on the surface of a cleaned GCE and allowed to evaporate solvent at room temperature.

The molecularly imprinted sol–gel was prepared according to the following produce. Firstly, 1.0 mL of SPU was mixed with 10.0 mL HCl (0.01 M). The mixed solution was stirred 5 h at 30 °C to obtain a homogeneous sol. Then, a total of 1.0 mL of this sol mixed with 100.0 μL of TEOS, 300.0 μL of  $\beta_2$ -agonists (0.01 M) was stirred for 2 h at 25  $\rm{^{\circ}C}$  to obtain an imprinted sol– gel solution, while non-imprinted sol–gel film was prepared under the same experimental conditions without the addition of  $\beta_2$ -agonists. Finally, the MIP/sol–gel/SWNTs sensor was fabricated by placing 5.0 μL of sol–gel imprinted polymer on the surface of SWNTs/GCE (Scheme. 1). The decorated sensor was left to deposit overnight at room temperature. Removal of the template molecular was achieved by immersing the dried MIP/sol–gel/SWNTs/GCE into the mixture of ethanol-pure water (1:1, v/v), and the completely removal of  $\beta_2$ -agonists was verified by electrochemical measurement.

#### Results and discussion

Characterization of molecularly imprinting film

In order to characterize the morphology of SWNTs/GCE and MIP/sol–gel/SWNTs/GCE, SEM was performed. As shown in Fig. [1a,](#page-3-0) SWNTs were formed as boundless tubes because of



Scheme 1 The preparation of the MIP/sol-gel/SWNTs/GCE

<span id="page-3-0"></span>Van der Waals forces, and distributed uniformly in sodium dodecyl sulfate. From the image of MIP/sol–gel/SWNTs film, it was clear to see that the film was uniformly cast onto the electrode surface (Fig. 1b). And the surface of the film was more complanate and compact, revealing that the fabrication sensor was achieved perfectly.

To further confirm the successful synthesis of MIP/sol–gel, FT-IR spectra measurements were conducted. Figure 2 shows the spectra of pure SAL (a), imprinting film without SAL (b) and imprinting film (c), respectively. It can be seen pure SAL shows the characteristic bands of benzene at 1,500 cm−<sup>1</sup> and the absorption band of its –NH at  $3,300 \text{ cm}^{-1}$ . Besides, the absorption appeared at 3,000 cm<sup>-1</sup> is ascribed to the -OH group. For the imprinting film without SAL and imprinting films, the common absorption peak of amide (−NH) and urethane group can be observed at  $3,300 \text{ cm}^{-1}$  and  $1,639 \text{ cm}^{-1}$ . The band around 1,069 cm−<sup>1</sup> indicates Si–O–Si stretching vibrations, conforming the existence of an extensive ethoxysilane groups in the film. The distinct difference between the spectra of imprinting film without SAL and imprinting films is the characteristic bands of SAL at 1,500 cm<sup>-1</sup>, which are discernable for the imprinting film, indicating that MIP/sol–gel was achieved perfectly.

#### Electrochemical behaviors of imprinted sensor

To characterize the properties of the imprinted sensor, cyclic voltammetry of  $β_2$ -agonists imprinted film before and after the



Fig. 1 SEM images of the SWNTs/GCE (a) and MIP/sol–gel/SWNTs/ GCE (b)



Fig. 2 FT-IR spectra of the pure SAL (a), imprinting film without SAL (b) and SAL imprinting film (c)

removal of β<sub>2</sub>-agonists were investigated in phosphate buffer solution. All of the 11  $\beta_2$ -agonists showed similar electrochemical response in the potential range of 0.6~0.8 V. SAL was taken as an example: an apparent response current peak at



Fig. 3 (A) Cyclic voltammograms of the MIP/sol–gel/SWNTs/GCE: before (a) and after (b) extraction of SAL. (B) Cyclic voltammograms of different electrodes incubated in 5.0×10−<sup>6</sup> M SAL solution for 10 min: MIP/sol–gel/SWNTs/GCE (a), MIP/sol–gel/GCE (b) and NIP/sol–gel/SWNTs/GCE (c). Scan rate: 0.1 V s−<sup>1</sup> . Buffer solution: 0.1 M PBS, pH 7.0

0.62 V was observed in Fig. [3A-a](#page-3-0). However, when MIP/sol– gel/SWNTs/GCE was eluted repeatedly with the mixture of ethanol-pure water, tiny current was observed (Fig. [3A-b\)](#page-3-0). This suggested that SAL was successfully removed.

The rebinding ability of MIP/sol–gel/SWNTs/GCE with SAL was investigated by cyclic voltammetry. After being incubated in 5.0 μM SAL solution for 10 min, the MIP/sol– gel/SWNTs/GCE (Fig. [3B-a\)](#page-3-0) exhibited a much higher current response than that of the GCE just modified with the imprinted sol–gel film (Fig. [3B-b](#page-3-0)) or the non-imprinted film (Fig. [3B-c\)](#page-3-0). The excellent sensitivity of the MIP/sol–gel/SWNTs/GCE towards SAL was ascribed to the presence of the SWNTs functional monolayer and the imprinted sol–gel film. When SWNTs were introduced into the MIP/sol–gel/SWNTs/GCE, the current response was amplified further than that of the MIP/sol–gel/GCE. It is inferred that SWNTs can expand the current response signal by enhancing electron transfer and increasing specific surface area on the electrode interface. Owing to the inherent specificity of the imprinted film, SAL could interact with it selectivity in the cavities with rebinding groups. Thus, the MIP/sol–gel/SWNTs/GCE showed much higher responses than the NIP/sol–gel/SWNTs/GCE.

Optimization of the molecularly imprinted sol–gel polymer

## Influence of the dosage of tetraethoxysilane

The role of functional monomers in sol–gel films is to assist the creation of the specific binding cavity by leaving interacting chemical functions after the polymerization, which is situated within the cavity in an optimal position for rebinding similar to active site in an enzyme. SPU was acted as functionalized monomer for molecularly imprinted, which can interact with the template  $\beta_2$ -agonists via hydrogen-bond. TEOS was allowed for acting as cross-linker to form a polymeric network around the template, so the ratio of SPU and TEOS is very important for the sol–gel films. SAL was taken as an example: Different ratios of SPU and TEOS in the imprinting polymerization, including 20:1, 10:1, 5:1, 2:1, 1:1, 1:5, 1:10 were studied by cyclic voltammetry. It was found that the current of the imprinted SAL increased with the increasing of the dosage of TEOS in the imprinting polymerization, indicating that more SAL had been imprinted. But the sol–gel films became unstable with the increasing of the dosage of TEOS in polymerization process. Considering the above two contradiction aspects comprehensively, sol–gel film formed by mixture of SPU and TEOS (10:1, v/v) was taken as a compromise for the polymerization.

#### Influence of incubation time

During electrochemical determination, accumulation is an effective and simple way to improve the sensitivity. In this study, MIP/sol–gel/SWNTs/GCE was firstly immersed in the standard

solution of  $\beta_2$ -agonists at  $5.0 \times 10^{-7}$  M for different time. Then the peak current was investigated quantitatively by cyclic voltammetry. SAL was taken as an example: as shown in Fig. 4, when the incubation time gradually increased from 1 min to 15 min, the response currents increased gradually. The results suggested that more and more SAL molecules were re-embedded on the MIP/sol–gel/SWNTs/GCE with increment of the incubation time. However, the peak current remained constant with the incubation time up to 15 min. As a result, a time of 15 min was chosen for the MIP/sol–gel/SWNTs/GCE. In addition, 15 min was also chosen for the NIP/sol– gel/SWNTs/GCE in order to make the comparison under the same condition.

## Calibration curve

The current response is directly correlated with the concentration of the  $\beta_2$ -agonists added. SAL was taken as an example: The linear sweep voltammetry responses of SAL with stepwise increases of SAL on the MIP/sol–gel/SWNTs/GCE are shown in Fig. [5](#page-5-0). The inset of Fig. [5](#page-5-0) shows the calibration curves of the MIP/sol–gel/SWNTs/GCE and the NIP/sol– gel/SWNTs/GCE between the corresponding current and SAL concentration. As it showed, for the imprinted sensor, the peak currents were proportional to the concentration of SAL in the range of  $9.9 \times 10^{-9} \sim 8.3 \times 10^{-7}$  M with the detection limit of  $3.0 \times 10^{-9}$  M. However, compared with the MIP/sol– gel/SWNTs/GCE, the NIP sensor gave no homologous response which can be explained by the lack of specific binding site and porous in the NIP film. The linear range and limit of quantitation of [1](#page-1-0)1  $\beta_2$ -agonists are listed in Table 1. Besides, the calibration curves of other 10  $\beta_2$ -agonists were shown in figures S1 to S10 (supplementary materials).

Additionally, the analytical performance of the molecularly imprinted sol–gel sensor has been compared with that of other developed  $\beta_2$ -agonists sensors reported in the literatures [[28,](#page-6-0) [29](#page-6-0), [31\]](#page-6-0). The performance features are



Fig. 4 Influence of incubation time on the response current

<span id="page-5-0"></span>

Fig. 5 Linear sweep voltammetry of MIP/sol–gel/SWNTs/GCE in pH 7.0 PBS after incubation in different concentrations of SAL solution for 15 min. Concentration of SAL (a to g):  $9.9 \times 10^{-9}$ ,  $4.9 \times 10^{-8}$ ,  $7.9 \times$  $10^{-8}$ ,  $1.9 \times 10^{-7}$ ,  $4.5 \times 10^{-7}$ ,  $6.9 \times 10^{-7}$ ,  $8.3 \times 10^{-7}$  M. The inset shows the calibration curves of SAL at imprinted sol–gel/SWNTs/GCE and nonimprinted sol–gel/SWNTs/GCE

summarized in Table 2. As can be seen, the proposed sensor exhibit a wide linear range and low detection limit for  $\beta_2$ agonists.

### Selectivity and stability of the imprinted sensor

Selective recognition of the template molecule is an important merit for a MIP sensor. In this work, SAL was taken as an example: the selectivity of the imprinted sensor to SAL was evaluated by testing its cyclic voltammetric response of SAL or SAL in the presence of some analogues and potential interfering substances, including clenbuterol (CLE), terbutaline (TER), dopamine (DA), glucose (Glu), ascorbic acid (AA), uric acid (UA) and acetic acid (AcA). As shown in Fig. 6, the excess of interfering substances mentioned above hardly caused any significant change of peak current of SAL. The tolerance limit was taken as the maximum concentration of the foreign substances which caused an approximately  $\pm$ 5 % relative error in the determination. The selectivity of the imprinted sensor to other  $\beta_2$ agonists was shown in Fig. S11 to Fig. S20 (supplementary materials). In view of the above facts, satisfactory selectivity to

**Table 2** Comparison of the analytical methods for the detection of  $\beta$ <sub>2</sub>agonists

Analytical method	Linear Range $(\mu M)$	Limit of Detection $(\mu M)$	Reference
Chemiluminescence sensor	$0.208 - 41.7$	0.067	[28]
Electrochemical sensor	$0.3 - 50$	0.01	[29]
Potentiometric sensor	$0.1 - 100$	0.07	$\lceil 31 \rceil$
Electrochemical sensor	$0.0099 - 0.83$	0.003	This work



Fig. 6 Selectivity of the imprinted sol–gel/SWNTs/GCE. Peak current response of imprinted sol–gel/SWNTs/GCE incubated in (a) 5.0×  $10^{-7}$  M SAL in 0.1 mol L<sup>-1</sup> PBS, (b) a+7.5×10<sup>-5</sup> M CLE, (c) a+  $7.5 \times 10^{-5}$  M TER, (d) a+2.5×10<sup>-4</sup> M DA, (e) a+5.0×10<sup>-4</sup> M Glu, (f)  $a+1.0\times10^{-4}$  M AA, (g)  $a+1.0\times10^{-4}$  M UA, (h)  $a+1.0\times10^{-4}$  M AcA. Incubation time: 15 min

 $\beta_2$ -agonists was obtained by such a kind of sensor. This can be explained by the suitable molecular contours cavity and unique binding, which resulted from hydrogen bonds between the recognition site and  $\beta_2$ -agonists.

The stability was evaluated for the imprinted electrode. Detailed experiments revealed that the response of  $\beta_2$ -agonists at MIP/sol–gel/SWNTs/GCE hardly changed after the imprinted electrode was used at least 50 times with subsequent washing and measuring operations.

#### Sample analysis

The practical analytical utility of MIP/sol–gel/SWNTs/GCE was assessed by measurement of  $\beta_2$ -agonists in human serum sample. SAL was taken as an example: The measurement was performed by investigating human serum samples with SAL at three different levels (20, 100 and 200 nM). For each concentration, three different samples were independently processed. On the basis of the calibration curves prepared in phosphate buffer solution, it was possible to calculate the recovery of the analyte, which ranged from 96.3 % to 100.3 % [[32\]](#page-6-0). The statistical results are listed in Table 3. The excellent recoveries indicate that this method has good accuracy and great potential in the practical sample analysis.





 $*_{n=3}$ 

#### <span id="page-6-0"></span>Conclusions

In this work, an electrochemical imprinted sensor for the determination of 11  $\beta_2$ -agonists was constructed via stepwise modification of SWNTs and a sol–gel imprinted film on GCE. The excellent performance of the imprinted sol– gel/SWNTs/GCE towards  $\beta_2$ -agonists can be ascribed to the SWNTs functional layer with electrochemical catalytic activities and the imprinted film with plentiful selective binding sites. It offers advantages by its easy preparation, high sensitivity and low detection limits. With high selectivity and sensitivity, the electrochemical sensor had great potential application in the real samples analysis.

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