



# Enhanced production of poly(3-hydroxybutyrate-co-3-hydroxyvalerate) biopolymer by recombinant *Bacillus megaterium* in fed-batch bioreactors

Murat Akdoğan<sup>1</sup> · Eda Çelik<sup>1,2</sup>

Received: 16 July 2020 / Accepted: 17 September 2020 / Published online: 29 September 2020  
© Springer-Verlag GmbH Germany, part of Springer Nature 2020

## Abstract

Polyhydroxyalkanoates (PHAs) are biodegradable polyesters accumulated in a wide variety of microorganisms as intracellular carbon and energy storage compounds. Poly(3-hydroxybutyrate-co-3-hydroxyvalerate) (PHBV) is one of the most valuable biopolymers because of its superior mechanical properties. Here, we developed a bioprocess utilizing recombinant *Bacillus megaterium* strain for PHBV over-production from glucose, without any precursor addition. PHA production was performed in a controlled bioreactor by batch and fed-batch modes using wild-type *B. megaterium* and rec-*B. megaterium* cells overexpressing the native *phaC* gene. The effect of oxygen transfer rate on biomass formation and PHA accumulation was also investigated, under different dissolved oxygen levels. Structural and thermal properties of PHA were characterized by GC-FID, <sup>1</sup>H-NMR, TGA and DSC analyses. Significantly, the copolymer produced from glucose as the carbon source in rec-*B. megaterium* was composed of 58 mol% of 3-hydroxyvalerate monomers. After 66 h, rec-*B. megaterium* cells in fed-batch fermentation with a pre-determined growth rate  $\mu_0 = 0.1 \text{ h}^{-1}$  produced the highest CDW ( $7.7 \text{ g L}^{-1}$ ) and PHA concentration ( $6.1 \text{ g L}^{-1}$ ). Moreover, an exponential glucose feeding profile resulted in 2.2-fold increase in PHA yield compared to batch cultivation. Overall, this study paves the way to an enhanced biopolymer production process in *B. megaterium* cells, where the highest product yield on cell was obtained as  $Y_{p/X} = 0.8 \text{ g g}^{-1}$ .

**Keywords** Poly(3-hydroxybutyrate-co-3-hydroxyvalerate) · Recombinant *B. megaterium* · Fed-batch fermentation · Oxygen transfer · Copolymer

## Introduction

Polyhydroxyalkanoates (PHAs) have been considered as an eco-friendly alternative to petroleum-based plastics owing to their biorenewability, biocompatibility and biodegradability. PHAs are biodegradable polymers, accumulated in wide variety of microorganisms, including Gram-negative bacteria, Gram-positive bacteria, cyanobacteria and archaea

as intracellular cytoplasmic inclusions for carbon and energy storage material, under environmental and nutritional stress conditions (limited nitrogen, phosphorus or oxygen) and excess carbon source [1, 2]. PHAs can be classified as short-chain length (SCL) (C3–C5), medium-chain length (MCL) (C6–C14) and SCL–MCL groups (C3 to C6–C14) depending on the monomer structure. Poly (3-hydroxybutyrate) (PHB) and poly (3-hydroxybutyrate-co-3-hydroxyvalerate) (PHBV) are the most well-known polyesters of the SCL-PHA family. PHB is a highly crystalline and brittle biopolymer with limited processability [3]. Hence, there have been attempts to improve the processability of PHB by incorporation of 3-hydroxyvalerate (3HV) through fermentation using expensive precursors, such as sodium valerate [4, 5] or propionic acid [6, 7], unrelated carbon sources [8–11] or via limited number of attempts in metabolic engineering of *Escherichia coli* [12–14], *Haloflex mediterranei* [15], *Rhodospirillum rubrum* [16] or *Ralstonia eutropha* [17]. On the other hand, *B. megaterium* is a natural producer of P(3HB-co-3HV)

**Electronic supplementary material** The online version of this article (<https://doi.org/10.1007/s00449-020-02452-z>) contains supplementary material, which is available to authorized users.

✉ Eda Çelik  
edacelik@hacettepe.edu.tr

<sup>1</sup> Department of Chemical Engineering, Hacettepe University, Beytepe, 06800 Ankara, Turkey

<sup>2</sup> Institute of Science, Bioengineering Division, Hacettepe University, Beytepe, 06800 Ankara, Turkey

from unrelated carbon sources [8, 10, 18–20] with a yield of up to 16.6 mol% of 3HV fraction, and this fraction goes up to 33 mol% for *B. thuringiensis* [21].

The incorporation of 3HV unit into the PHB crystal structure leads to dramatic changes in material properties of synthesized PHBV copolymers, such as low melting temperature, low crystallinity and toughness [22]. PHBV synthesis occurs via three-step enzymatic process catalyzed by  $\beta$ -ketothiolase (PhaA),  $\beta$ -ketoacyl-CoA reductase (PhaB) and PHA synthase (PhaC) sequentially [23]. PhaC enzyme is responsible for polymerization of PHA polyesters [24].

Currently, the major drawback for PHA production at large scale is the high production cost and low product yields compared with petroleum-based plastics [25]. Therefore, efforts have also been carried out to reduce the production cost using inexpensive and renewable carbon sources, the development of genetically-engineered better PHA-synthesizing strains and more efficient and sustainable bioprocess strategies for higher productivities [26–29].

There are various factors, such as microorganism selection, growth medium, nitrogen sources, carbon-to-nitrogen ratio, temperature, pH and dissolved oxygen demand, to obtain a higher polymer yield and volumetric productivity [25]. Various fed-batch fermentation strategies have been successfully used for PHA production by *B. megaterium* strains to achieve high cell densities and high PHA productivity [30, 31]. On the other hand, to maximize volumetric PHA productivity, oxygen transfer rate (OTR) is an important factor in fed-batch cultivation due to high oxygen demand [32]. Various studies have reported the increase in SCL-PHA synthesis in several microorganisms from sugars at low DO conditions [33, 34].

This study aimed to develop an engineered *Bacillus megaterium* NRRL B-14308 strain by overexpressing the native *phaC* gene for PHA production from glucose. Furthermore, the engineered *B. megaterium* strain was evaluated for PHBV production in laboratory-scale bioreactors by batch and fed-batch modes and compared with the wild-type *B.*

*megaterium* NRRL B-14308 strain. Generally, PHBV copolymers were biologically produced by feeding its petroleum-based precursors like propionate or valeric acid, resulting in limited actual production due to their high cost and toxicity. This study is significant for production of PHBV by *B. megaterium* strain with the highest 3HV content from structurally unrelated carbon sources, such as glucose, without a 3HV precursor addition. The influence of oxygen transfer rate on the biomass production and PHA synthesis in laboratory-scale bioreactors under different DO concentrations was also investigated.

## Materials and methods

### Bacterial strains and plasmids

All bacterial strains and constructed plasmids used in this study are listed in Table 1. For construction of the recombinant plasmid of pC-HIS1623hp-*phaC*, the PHA synthase gene (*phaC*) (2590 bp) of *B. megaterium* NRRL B-14308 strain (GenBank Accession No. KGJ86215.1) was amplified from genomic DNA by colony PCR method, using the specific primers *phaC\_F* (ATGACTACTAGTAAGGAGGTGAATATACAATGGCAATTCCTTACG-TGCAAG) and *phaC\_R* (ATGATCGCATGCTTAGTGATGGTGATGTGA-TGAGAACCGCCTTTAGAGCGTTTTTCTAGC) (underlined sequences indicate *SpeI* for *phaC\_F* and *SphI* for *phaC\_R*, respectively).

The amplified *phaC* fragment was digested with *SpeI* and *SphI* followed by purification, ligation and transformation of *Escherichia coli* DH5 $\alpha$  by CaCl<sub>2</sub> method, with 100  $\mu\text{g mL}^{-1}$  ampicillin in selective solid medium. Single colonies were picked and the purified plasmids were confirmed via restriction digestion and DNA sequence analysis. Molecular biology protocols used in this study were as outlined by Sambrook and Russell [35]. *B. megaterium* protoplasts were transformed by a modified minimal media protoplast

**Table 1** Bacterial strains and plasmids used in this study

Name	Description	References
<i>Strains</i>		
<i>E. coli</i> DH5 $\alpha$	F- (f80dlacZ $\Delta$ M15) $\Delta$ (lacIZYA -argF) U169 deoR recA1 endA1 hsdR17 (rk-, mk+) supE44, thi-1, gyrA96, relA1	Invitrogen, USA
<i>B. megaterium</i> NRRL B-14308	Wild type	ARS, USA
<i>Plasmids</i>		
pC-HIS1623hp	Expression vector, Amp <sup>R</sup> ( <i>E. coli</i> ) and Tet <sup>R</sup> ( <i>Bacillus</i> )	MoBiTec GmbH, Germany
pC-HIS1623hp- <i>phaC</i>	Derivative of pC-HIS1623hp with <i>phaC</i> downstream of a xylose inducible promoter with Gly-Gly-Ser linker and 6xHis-tag	This study

transformation protocol [36], and the selected colony was stored in glycerol stocks at  $-86\text{ }^{\circ}\text{C}$ .

### Culture medium and growth conditions

The recombinant *B. megaterium* were grown in lysogeny broth (LB) medium containing  $10\text{ g L}^{-1}$  peptone,  $5\text{ g L}^{-1}$  yeast extract and  $10\text{ g L}^{-1}$  NaCl and selective antibiotics to maintain the plasmid stability. The bacterial culture was inoculated using a 1:100 inoculum ratio into 20 mL LB medium in 250 mL Erlenmeyer flasks. The cells were centrifuged and re-suspended in 50 mL Minimal Medium (MM) [20] and the flasks were incubated at  $37\text{ }^{\circ}\text{C}$  and 200 rpm for 66 h. After fermentation, the cells were harvested by centrifugation at  $10,000\times g$  for 10 min.

### Recombinant protein analysis

Whole-cell lysates were used for recombinant protein analysis, where total protein concentration was determined in duplicates using Bradford assay with bovine serum albumin (BSA, Sigma-Aldrich) as standard. The samples were subjected to SDS-PAGE (12%) and Western Blotting analysis. Proteins were visualized with Coomassie Blue (Bio-Rad) staining by GelDoc EZ imaging system (Bio-Rad). Then, proteins were transblotted onto PVDF membranes, blocked with blocking reagent (Bio-Rad) at room temperature (RT), and probed with anti-His (1:1000, Abcam) followed by anti-rabbit-HRP (1:2500, Promega) antibodies in 1% BSA-PBST. The immunoreactive bands were visualized using Opti-4CN colorimetric kit (Bio-Rad) according to manufacturer's instructions.

### Bioreactor operation in batch and fed-batch modes

All bioreactor experiments were carried out in a 500 mL fully controlled bioreactor system (My-Control miniBio, Applikon Biotechnology, Delft, Netherlands) with a working volume of 300 mL. The Minimal Medium (MM) was used as initial fermentation medium. A glucose solution of  $100\text{ g L}^{-1}$  was fed separately for fed-batch cultivations. The 500 mL bioreactor was assembled with a pH sensor, optical DO sensor, L-type gas sparger and 2 Rushton impellers. The temperature and pH were maintained at  $37\text{ }^{\circ}\text{C}$  and  $7.0\pm 0.1$ , respectively. The pH was controlled by 1 M HCl and 1 M NaOH solutions. The initial agitation rate was set at 300 rpm. Three different dissolved oxygen values (10%, 20% and 30% DO) were evaluated in this study and adjusted via agitation and/or aeration rates.

Fed-batch cultivations were performed using wild-type and recombinant *Bacillus megaterium* strains. The culture was initially carried out in a batch mode for 42 h and then fed-batch mode was initiated with exponential glucose

feeding at a pre-determined specific growth rate ( $\mu_0$ ) to achieve a higher PHA yields. During the exponential glucose feeding stage, feeding profile,  $F(t)$ , was controlled based on the following equation [37]:

$$F(t) = \frac{\mu_0 V_0 C_{X0}}{C_{S0} Y_{X/S}} \exp(\mu_0 t),$$

where  $F(t)$  ( $\text{L h}^{-1}$ ) is the total substrate feeding rate,  $\mu_0$  ( $\text{h}^{-1}$ ) is the desired specific growth rate,  $V_0$  (L) is the initial volume,  $C_{X0}$  ( $\text{g L}^{-1}$ ) is the initial biomass concentration,  $C_{S0}$  ( $\text{g L}^{-1}$ ) is the feed substrate concentration and  $Y_{X/S}$  ( $\text{g g}^{-1}$ ) is the biomass yield on substrate. In the fed-batch phase,  $Y_{X/S}$  for the feeding profile equation was set as  $0.36\text{ g g}^{-1}$ , pre-determined from batch experiments. All runs were performed in duplicates.

### Determination of oxygen transfer parameters

To determine the oxygen transfer parameters, such as volumetric mass transfer coefficient ( $k_L a$ ), biological enhancement factor ( $E$ ), oxygen uptake rate (OUR), and oxygen transfer rate (OTR) for PHA production, the Dynamic method [38] was used.

### Analytical procedures

Cell growth was analyzed by measuring the optical density at 600 nm ( $\text{OD}_{600}$ ) using a UV–visible spectrophotometer (Thermo Fisher Scientific Genesys 10S UV–VIS, USA). The cell dry weight was determined gravimetrically. The residual glucose concentration was determined spectrophotometrically using a modified microplate 3,5-dinitrosalicylic acid (DNS) assay [39]. PHA was measured spectrofluorometrically with fluorescence spectrophotometer (Varian Cary Eclipse, Agilent, Santa Clara, CA, USA) as given in our previous study [20].

### PHA content and monomer determination

The PHA content quantification and monomer composition were determined by gas chromatography (GC) of methyl esters and verified by  $^1\text{H-NMR}$ . The lyophilized biomass was subjected to acidic methanolysis reaction [40]. The 3-hydroxyalkanoic acid methyl esters were analyzed by GC using an Agilent Technologies 6890 N chromatograph equipped with an HP-5 capillary column ( $25\text{ m}\times 0.32\text{ mm}\times 0.25\text{ }\mu\text{m}$ ) and a flame ionization detector (FID). The temperatures of injection and detector were  $300\text{ }^{\circ}\text{C}$ . The oven temperature profile was: from 50 up to  $100\text{ }^{\circ}\text{C}$  at a rate of  $5\text{ }^{\circ}\text{C min}^{-1}$ , then from 100 up to  $300\text{ }^{\circ}\text{C}$  at a rate of  $20\text{ }^{\circ}\text{C min}^{-1}$ , and finally  $300\text{ }^{\circ}\text{C}$  for 5 min [41]. Helium was used as a carrier gas at a flow rate

of  $1 \text{ mL min}^{-1}$ . Methyl benzoate was used as an internal standard.

### Polymer characterization

$^1\text{H-NMR}$  spectroscopy was used for determination of the chemical structure of PHA polymers. Briefly, about 20 mg of PHA sample was dissolved in deuterated chloroform and  $^1\text{H-NMR}$  spectra were recorded at  $25 \text{ }^\circ\text{C}$  on Bruker (Billerica, MA, USA) 300 MHz spectrometer.

The thermal properties of PHA samples were analyzed by TGA (SII Exstar 6000 TG–DTA 6300, Perkin Elmer Inc, Waltham, MA, USA) and DSC (Diamond DSC, PerkinElmer Inc., Waltham, MA, USA). To determine the thermal degradation profile of the polymer, about 10 mg of PHA sample was heated from room temperature to  $500 \text{ }^\circ\text{C}$  at a heating rate of  $10 \text{ }^\circ\text{C min}^{-1}$  under nitrogen gas.

The melting temperatures ( $T_m$ ) of the PHA polymers were determined by DSC analysis. Briefly, about 10 mg of PHA sample was encapsulated in an aluminum DSC pan and then heated from  $-30 \text{ }^\circ\text{C}$  to  $200 \text{ }^\circ\text{C}$  at a heating rate of  $10 \text{ }^\circ\text{C min}^{-1}$  under nitrogen gas.

### Statistical data analysis

Average and standard deviation values were calculated according to standard procedures and the results were analyzed by ANOVA test. Tukey's test was used to compare mean values and to evaluate the significance of the differences between mean values to assess the PHA production. All statistical analyses were carried out using GraphPad Prism 8 software.

## Results and discussion

The Gram-positive bacteria display the advantage of producing endotoxin-free PHAs, in contrast to Gram-negative organisms like *E. coli*. *B. megaterium* can grow on a wide variety of carbon sources and has no obvious alkaline proteases degrading recombinant gene products [42]. In the present study, the native PHA synthase genes of *B. megaterium* NRRL B-14308 were amplified by PCR and cloned into pC-HIS1623hp expression vector to construct the recombinant plasmid overexpressing its native PhaC enzyme, with the aim of enhancing PHA production. Successful construction of the recombinant plasmid pC-HIS1623hp-*phaC* was confirmed via DNA sequencing. The PhaC synthase was then overexpressed in the *B. megaterium*/pC-HIS1623hp-*phaC* system. SDS-PAGE analysis revealed the presence of the recombinant PhaC synthase gene with the expected molecular weight of 42 kDa (Fig. S1). The results were further confirmed by Western blot analysis using anti-His and

anti-rabbit HRP antibodies and wild-type *B. megaterium* strain as the negative control.

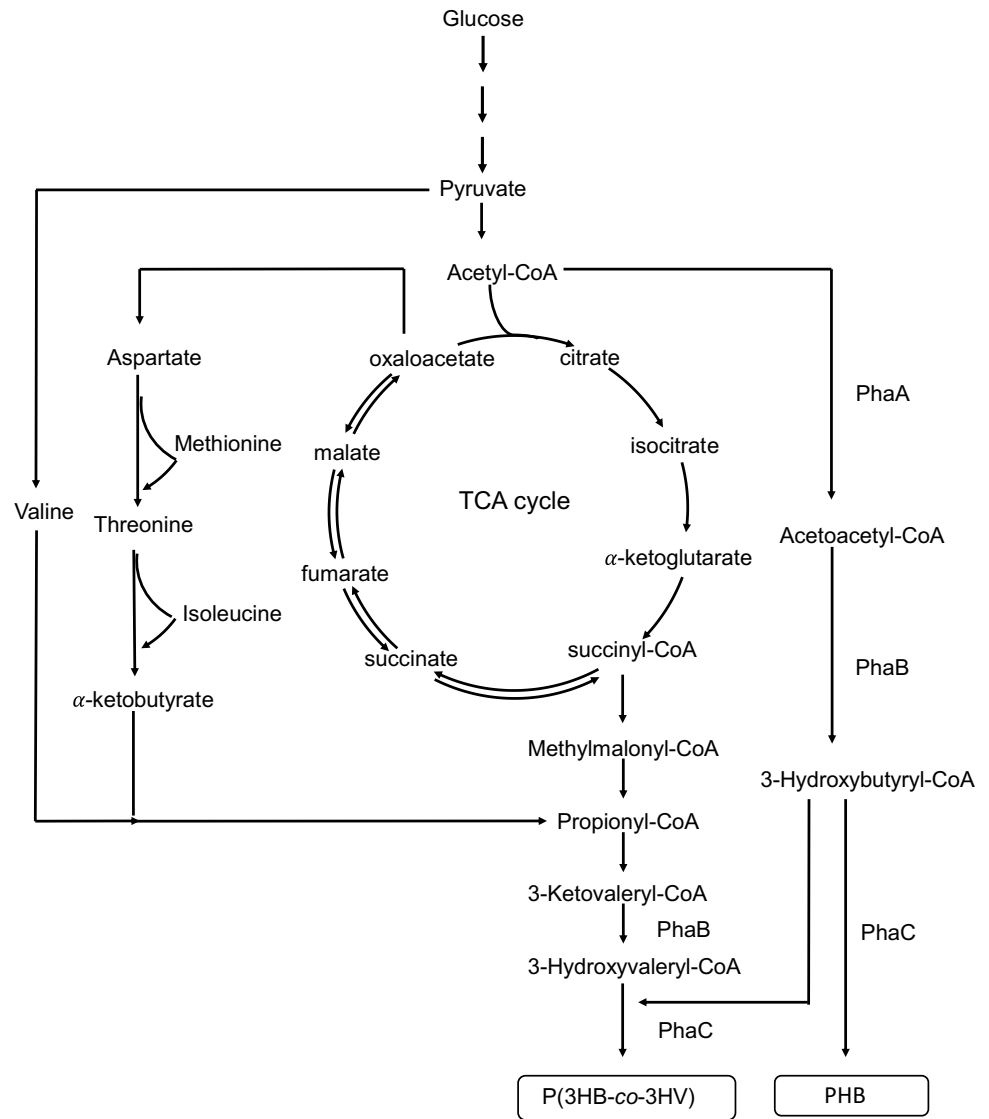
The PHBV formation in the *B. megaterium* involves two parallel pathways, leading to  $\text{C}_4$  monomer (3-hydroxybutyrate) and  $\text{C}_5$  monomer (3-hydroxyvalerate) in the copolymer (Fig. 1). In the PHBV biosynthesis, first, two acetyl-CoA moieties or acetyl-CoA and propionyl-CoA are condensed to form thioester intermediates acetoacetyl-CoA and 3-ketovaleryl-CoA, respectively, by expressing *phaA*. Then, the thioester intermediates are reduced to two PHA monomers 3HB-CoA and 3HV-CoA by expressing *phaB* gene. Finally, by expressing *phaC* gene, the two PHA monomers are randomly polymerized to form PHBV copolymer [13, 43]. *B. megaterium* polyhydroxyalkanoate synthase gene complex is a heterodimer protein composed of PhaR and PhaC (catalytically active unit) subunits [44]. In this study, we overexpressed only the catalytically active unit, PhaC, where PhaR is not required for the expression of *phaRBC* operon. Similarly, there are several studies on overexpression of a single subunit from the heterodimeric forms [45–47]. Provided that the monomers are present, overexpressing this final enzyme in the pathway, PhaC, could enhance the PHA yields and change the copolymer composition. The former would especially hold true, if the polymerization step is the rate limiting step, and the latter if the PhaC enzyme has different affinities toward different monomers.

### PHA production in batch mode

It has been clearly known that dissolved oxygen (DO) is one of the most important key parameters for cell growth and PHA accumulation. PHA synthesis can be triggered under oxygen-limited conditions by *Bacillus megaterium* strains [48]. The limitation of DO concentration results in the deviation of carbon flux from biomass production towards PHA accumulation [49].

*B. megaterium* NRRL B-14308 wild-type strain was cultivated in 500 mL scale, fully controlled bioreactor for 66 h using glucose as the carbon source to observe cell growth and PHA accumulation under different DO levels (10%, 20% and 30% DO) (Fig. S2) in batch cultivations. The fermentation kinetics and yields related to PHA production in the batch mode are shown in Table 2. The results showed that the highest PHA concentration and intracellular PHA content were reached at 20% DO set point compared to other DO levels. The 20% DO level showed higher cell dry weight than 10% DO level, whereas the lowest cell dry weight and PHA accumulation were obtained at 30% DO set point. On the other hand, the highest PHA yield,  $Y_{P/S}$ , was obtained for 20% DO set point. Based on these results, 20% DO was selected for further studies in bioreactor cultivations at batch

**Fig. 1** Schematic representation of the metabolic pathways involved in the biosynthesis of the P(3HB-co-3HV) copolymers in *Bacillus megaterium*



**Table 2** Kinetic and stoichiometric parameters of PHA produced from *B. megaterium* NRRL B-14308 wild-type strain and recombinant *B. megaterium* strain by batch cultivation mode at various dissolved oxygen (% DO) levels

Fermentation	CDW (g L <sup>-1</sup> )	PHAm <sub>ax</sub> (g L <sup>-1</sup> )	PHA (%)	Y <sub>X/S</sub> (g g <sup>-1</sup> )	Y <sub>P/S</sub> (g g <sup>-1</sup> )	Y <sub>P/X</sub> (g g <sup>-1</sup> )	qPHA (g L <sup>-1</sup> h <sup>-1</sup> )
<i>B. megaterium</i> wild-type strain							
10% DO	2.66 ± 0.03	2.08 ± 0.03	78 ± 1	0.32 ± 0.03	0.26 ± 0.03	0.78 ± 0.02	0.11 ± 0.02
20% DO	3.06 ± 0.04	2.43 ± 0.03	79 ± 2	0.38 ± 0.02	0.31 ± 0.02	0.76 ± 0.03	0.12 ± 0.03
30% DO	2.49 ± 0.03	1.96 ± 0.04	78 ± 2	0.31 ± 0.02	0.26 ± 0.03	0.77 ± 0.02	0.09 ± 0.02
<i>Recombinant B. megaterium</i>							
20% DO	3.52 ± 0.04	2.76 ± 0.05	78 ± 2	0.44 ± 0.03	0.35 ± 0.02	0.78 ± 0.03	0.15 ± 0.02

and fed-batch modes with wild-type and rec-*B. megaterium* strains, to maximize the efficiency of the PHA accumulation.

The total cell dry weight (CDW), maximum PHA accumulation, PHA content, biomass yield ( $Y_{X/S}$ ), product yields ( $Y_{P/S}$  and  $Y_{P/X}$ ) and PHA volumetric productivity (qPHA) obtained from batch cultivations of wild-type *B.*

*megaterium* and rec-*B. megaterium* cells are also summarized in Table 2. With a defined fermentation medium, cell dry weight reached 3.06 g L<sup>-1</sup> with 2.43 g L<sup>-1</sup> PHA accumulation for the wild-type *B. megaterium* strain. However, the highest CDW (3.52 g L<sup>-1</sup>) and maximum PHA accumulation (2.76 g L<sup>-1</sup>) were achieved when fermentation



was carried out using rec-*B. megaterium* cells after 66 h of cultivation. The highest final PHA content and the highest PHA volumetric productivity were also attained using rec-*B. megaterium* cells as 78 wt% and  $0.15 \text{ g L}^{-1} \text{ h}^{-1}$ , respectively. Consequently, there is 1.2-fold increase in the PHA final concentration for rec-*B. megaterium* cells compared to wild-type *B. megaterium* cells in the batch mode.

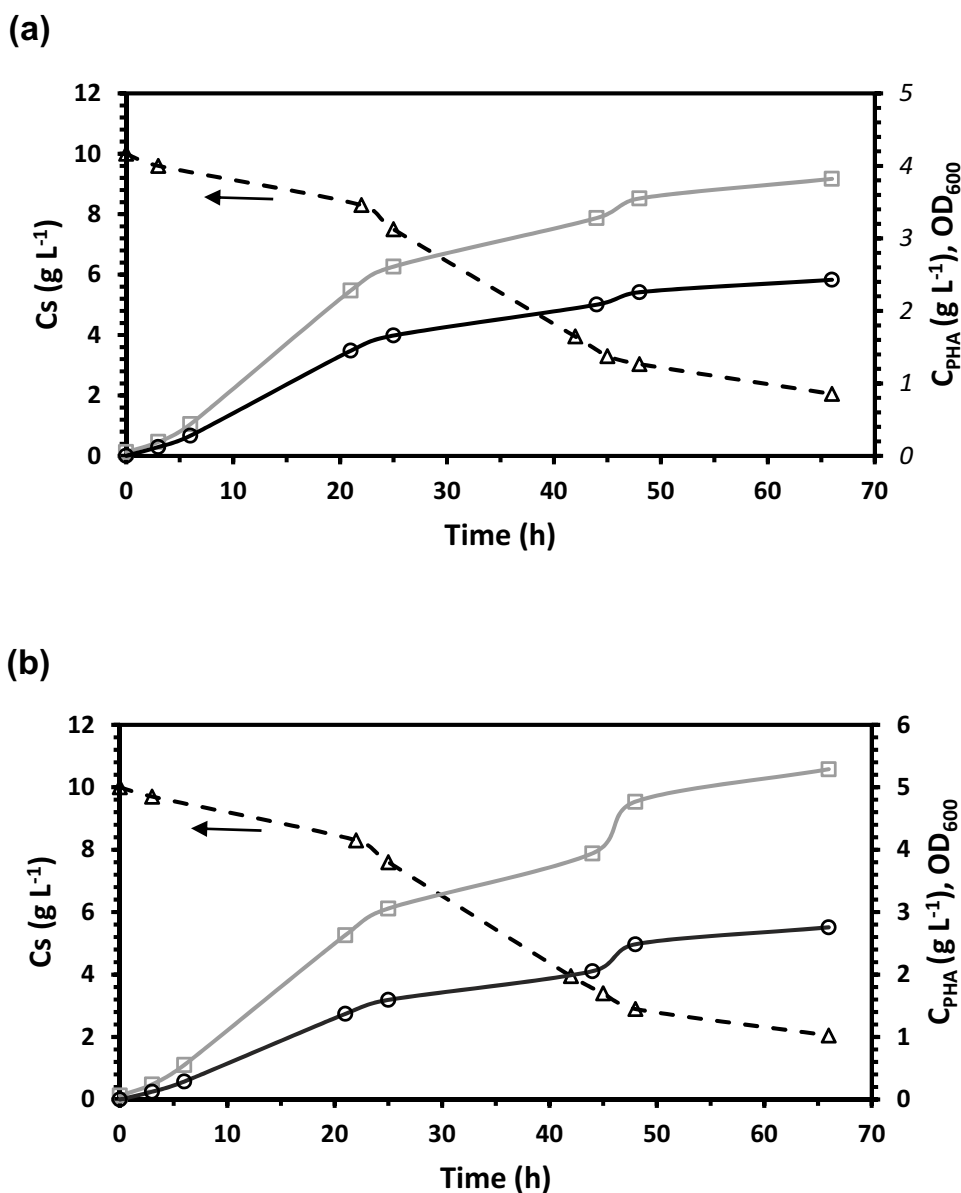
PHA accumulation reached a steady state after approximately 66 h of batch cultivation, even when glucose was not consumed totally by the cells, confirming that glucose was not the limiting nutrient in the minimal medium (Fig. 2). PHA synthesis continued in the stationary phase ( $t=42\text{--}66 \text{ h}$ ), confirming that *B. megaterium* accumulated PHAs by adopting growth-associated and non-growth-associated mechanisms [50, 51].

## Fed-batch cultivations for PHA production

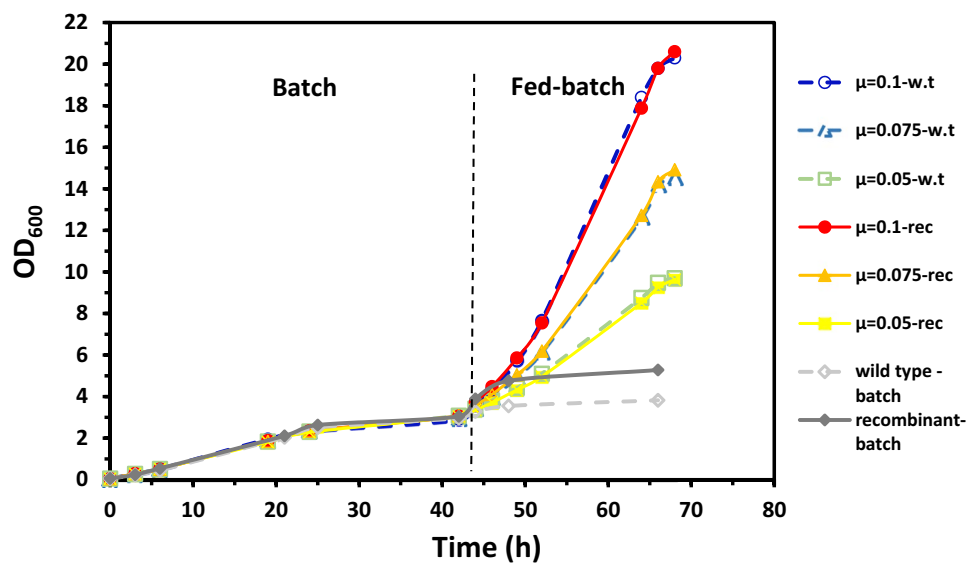
Fed-batch cultivations have been extensively employed to reach high cell density cultures and improve polymer productivity by monitoring DO, pH and carbon levels as feedback control parameters [52]. In this study, exponential glucose feeding profiles at several pre-determined specific growth rates were implemented to achieve higher PHA production in fed-batch fermentation.

The cultivation was initiated with a glucose concentration of  $10 \text{ g L}^{-1}$ , and the reactor was operated in a batch mode for 42 h, followed by a fed-batch phase (42–66 h) (Fig. 3). Ammonium sulfate ( $1 \text{ g L}^{-1}$ ) as the nitrogen source was only supplemented at the beginning of the cultivation, to promote cell growth.

**Fig. 2** Glucose consumption ( $C_s$ , dashed line), cell ( $\text{OD}_{600}$ , gray line) and PHA ( $C_{\text{PHA}}$ , solid line) accumulations during batch production by **a** *B. megaterium* NRRL B-14308 wild-type strain; **b** recombinant *B. megaterium* strain. Data expressed are the mean values and all experiments were performed in duplicate ( $n=2$ , maximum SD =  $\pm 21.2\%$ , minimum  $\pm 1.9\%$ )



**Fig. 3** Cell growth ( $OD_{600}$ ) during fed-batch production by wild-type and recombinant *B. megaterium* strains, using different exponential glucose feeding profiles starting at  $t = 42$  h. The experimental data are the mean data from two data points ( $n = 2$ , maximum SD =  $\pm 28.3\%$ , minimum  $\pm 4.2\%$ )

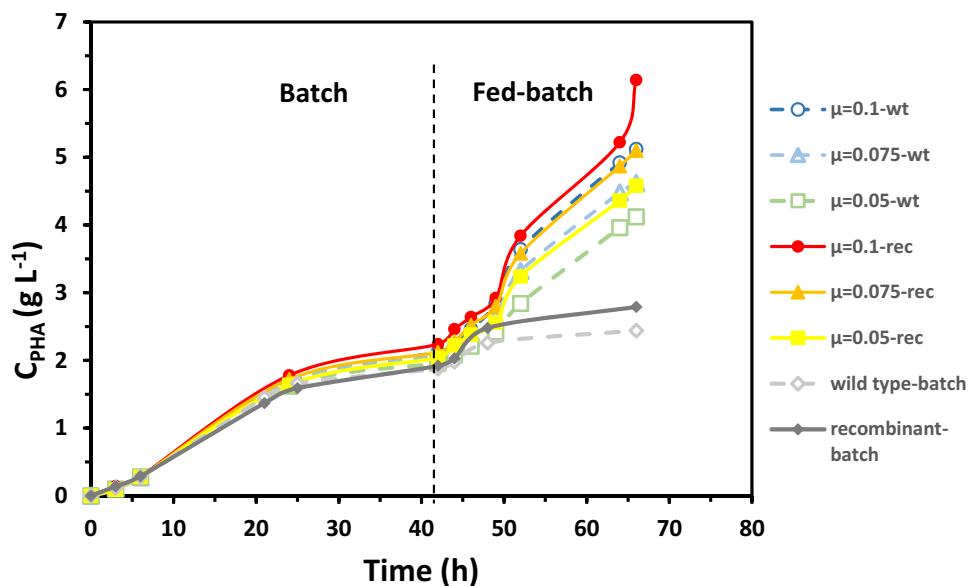


The fed-batch phase was initiated after the batch phase (at  $t = 42$  h) by automatically feeding the fermentation medium with glucose solution under exponential feeding profiles, controlled 20% DO and pH 7.0 for wild-type and rec-*B. megaterium* cells.

The three different exponential glucose feeding profiles were based on the pre-determined specific growth rates:  $\mu_0 = 0.05 \text{ h}^{-1}$ ,  $\mu_0 = 0.075 \text{ h}^{-1}$ ,  $\mu_0 = 0.1 \text{ h}^{-1}$ . Therefore, a series of batch cultivations in the bioreactor were performed to determine the initial biomass yield on substrate ( $Y_{X/S}$ ), as  $Y_{X/S} = 0.36 \text{ g biomass g}^{-1} \text{ glucose}$ . This value was applied in fed-batch experiments to evaluate the different exponential glucose feeding strategies and was consistent with previous studies [53, 54] for PHA production.

Comparing the fermentation profiles, the trends of cell growth (Fig. 3) and PHA concentration (Fig. 4) were similar in the batch phase, whereas the accumulation curves diverged for the fed-batch phase, i.e. after 42 h, because of the increasing PHA accumulation and a reduced biomass formation as a consequence of the exponential glucose feeding and nitrogen limitation. During the fed-batch stage, the PHA concentration increased exponentially to  $6.15 \text{ g L}^{-1}$  as the highest level, by the end of the fermentation (Fig. 4). The high rate of PHA accumulation observed in rec-*B. megaterium* strain at the higher pre-determined specific growth rate ( $\mu_0 = 0.1 \text{ h}^{-1}$ ) towards the end of the process was potentially due to the high exponential glucose overfeeding and nitrogen depletion in the growth medium.

**Fig. 4** PHA ( $C_{\text{PHA}}$ ) accumulation during fed-batch production by wild-type and recombinant *B. megaterium* strains, using different exponential glucose feeding profiles starting at  $t = 42$  h. The experimental data are the mean data from two data points ( $n = 2$ , maximum SD =  $\pm 8.5\%$ , minimum  $\pm 2.1\%$ )

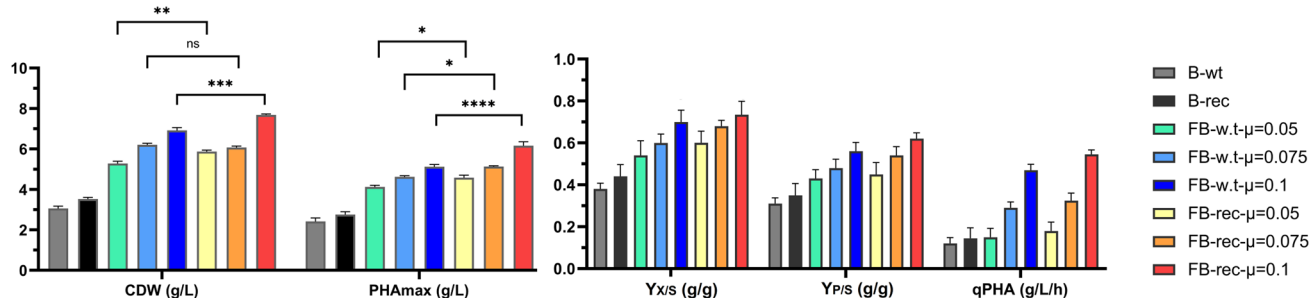


The kinetic and stoichiometric parameters for the fed-batch cultivation of wild-type and *rec-B. megaterium* cells at different exponential glucose feeding rates are demonstrated in Fig. 5. The exponential fed-batch strategy resulted in a 2.2-fold increase in PHA production compared to batch cultivations. The highest final cell concentration ( $7.68 \text{ g L}^{-1}$ ), PHA content of the cells (80%), volumetric PHA productivity ( $0.54 \text{ g L}^{-1} \text{ h}^{-1}$ ), as well as cell and product yields on substrate ( $0.74 \text{ g g}^{-1}$  and  $0.62 \text{ g g}^{-1}$ , respectively) were obtained with *rec-B. megaterium* cells at the higher pre-determined specific growth rate ( $\mu_0 = 0.1 \text{ h}^{-1}$ ), confirming the positive effect of genetic modification for PHA biosynthesis. The results also showed that *rec-B. megaterium* NRRL B-14308 strain could accumulate higher amount of PHA content compared to other *Bacillus* strains (Table 3). Moreover, if a high cell density fed-batch strategy is applied, the PHA yields could be even further improved. Nevertheless, a relatively high  $Y_{P/X}$  value,  $0.8 \text{ g g}^{-1}$ , was obtained via strain and bioprocess improvements.

## Oxygen transfer parameters of fed-batch fermentations

During PHA production by wild-type and *rec-B. megaterium* cells in fed-batch cultivation using different glucose feeding rates, the dynamic method was applied to determine the oxygen transfer parameters, such as volumetric mass transfer coefficient ( $k_L a$ ), enhancement factor (E), oxygen uptake rate (OUR) and oxygen transfer rate (OTR). The oxygen transfer parameters were determined for all fed-batch fermentations and are summarized in Table 4.

The  $k_L a$  values first increased and then decreased with the cultivation time. The  $k_L a$  values are considerably affected by a lot of factors, including geometrical characteristics of bioreactors, viscosity and surface tension of the broth, temperature, aeration rate, agitation speed, foam formation, and microorganisms' morphology [55]. The bioreactor operational parameters, such as temperature, agitation speed, aeration rate, were kept constant



**Fig. 5** The kinetic and stoichiometric parameters for the fed-batch cultivation of wild-type and *rec-B. megaterium* cells at different exponential glucose feeding rates. Asterisks indicate the significance lev-

els of two-way ANOVA test comparisons with Tukey test, \* $p < 0.1$ , \*\* $p < 0.01$ , \*\*\* $p < 0.001$ , \*\*\*\* $p < 0.0001$

**Table 3** A comparison of PHA production by *Bacillus* strains using various fermentation techniques

Microorganism	Culture conditions	Carbon substrate	CDW ( $\text{g L}^{-1}$ )	PHA (% w/w)	PHA ( $\text{g L}^{-1}$ )	3HV (mol %)	References
<i>B. megaterium</i>	Batch	Red-algae	4.1	30	1.24	n.d	[69]
<i>B. megaterium</i> BBST4	Shake-flask batch	Starch	1.72	24	0.42	9.8	[10]
<i>B. megaterium</i> BBST4	Shake-flask batch	Glycerol	2.60	20	0.53	7.8	[19]
<i>B. megaterium</i> OU303A	Shake-flask batch	Glucose	n.d	58	n.d	2.5	[18]
<i>B. megaterium</i>	Fed-batch	Acid treated red algae	8.2	53	4.37	n.d	[70]
<i>B. megaterium</i> uyuni S29	Fed-batch	Glucose	28.6	30	8.58	n.d	[71]
<i>B. megaterium</i> DSM 32	Batch	Sucrose	5.32	62	3.30	n.d	[48]
<i>B. megaterium</i> NCIM 5472	Batch	Cheese Whey Permeate	4.2	87	3.64	16.6	[7]
<i>B. megaterium</i> strain A1	Shake- flask batch	Molasses	n.a	78	n.a	11	[8]
<i>B. cereus</i> SPV	Shake-flask batch	Glucose	1.35	38	0.52	10	[11]
<i>B. thuringiensis</i>	Batch	Glucose, glycerol and propionic acid	4.5	46	2.07	58	[21]
Recombinant <i>B. megaterium</i> NRRL B-14308	Fed-batch	Glucose	7.68	80	6.15	58	This study



**Table 4** The variations in oxygen transfer parameters with different glucose feeding profiles for fed-batch PHA production

Fed-batch strategy	$t$ (h)	$k_L a$ ( $s^{-1}$ )	$E$ ( $k_L a/k_L a_0$ )	$OTR \times 10^3$ ( $mol\ m^{-3}\ s^{-1}$ )	$OTR_{max} \times 10^3$ ( $mol\ m^{-3}\ s^{-1}$ )	$OUR \times 10^3$ ( $mol\ m^{-3}\ s^{-1}$ )	
<i>B. megaterium</i> wt							
$\mu_0=0.05\ h^{-1}$	0	0.009	1.00	–	–	–	
	2	0.027	3.00	3.40	5.40	3.40	
	4	0.026	2.89	4.20	5.20	4.20	
	18	0.024	2.67	4.80	4.80	4.80	
	24	0.024	2.67	4.60	4.70	4.60	
	42	0.021	2.33	4.20	4.20	4.20	
	49	0.020	2.22	4.00	4.00	4.00	
	$\mu_0=0.075\ h^{-1}$	0	0.010	1.11	–	–	–
		2	0.031	3.44	4.30	6.30	4.30
		4	0.029	3.22	5.20	6.00	5.20
18		0.026	2.88	5.80	5.80	5.80	
24		0.025	2.77	5.40	5.40	5.40	
42		0.023	2.56	4.80	4.90	4.80	
49		0.021	2.33	4.50	4.50	4.50	
$\mu_0=0.1\ h^{-1}$	0	0.010	1	–	–	–	
	2	0.036	3.60	5.40	8.80	5.40	
	4	0.032	3.20	6.70	8.10	6.70	
	18	0.027	2.70	7.80	7.80	7.80	
	24	0.027	2.70	7.20	7.20	7.20	
	42	0.024	2.40	6.50	6.60	6.50	
	49	0.021	2.10	5.90	5.90	5.90	
<i>Recombinant B. megaterium</i>							
$\mu_0=0.05\ h^{-1}$	0	0.010	1.00	–	–	–	
	2	0.030	3.00	3.80	6.20	3.80	
	4	0.027	2.70	4.40	5.60	4.40	
	18	0.026	2.60	4.90	4.90	4.90	
	24	0.026	2.60	4.60	4.60	4.60	
	42	0.023	2.30	4.20	4.20	4.20	
	49	0.021	2.10	3.90	3.90	3.90	
$\mu_0=0.075\ h^{-1}$	0	0.011	1.10	–	–	–	
	2	0.036	3.60	4.56	6.70	4.56	
	4	0.030	3.00	5.40	6.20	5.40	
	18	0.027	2.70	6.08	6.0	6.10	
	24	0.025	2.50	5.75	5.80	5.75	
	42	0.022	2.20	5.20	5.20	5.20	
	49	0.020	2.00	4.80	4.80	4.80	
$\mu_0=0.1\ h^{-1}$	0	0.011	1.00	–	–	–	
	2	0.039	3.54	5.90	9.54	5.90	
	4	0.034	3.09	7.10	8.70	7.10	
	18	0.029	2.64	8.30	8.30	8.30	
	24	0.027	2.45	7.70	7.72	7.70	
	42	0.024	2.18	6.54	6.54	6.54	
	49	0.020	1.82	5.90	5.90	5.90	

throughout the bioprocesses and an antifoam agent was not used. Therefore, this decrease can be related to the viscosity of the fermentation broth. The viscosity of fermentation broth increased as the *B. megaterium* cells were in

exponential growth phase and during PHA accumulation. The increase in broth viscosity reduced oxygen transfer to the cells, resulting in a resistance zone for mass transfer [37, 56].

The oxygen uptake rate (OUR) is one of the most fundamental parameters in fermentation processes and depends on metabolic activity of cells. OUR increases in the exponential growth phase due to the high substrate consumption at the start of the cultivation process. The OUR was at its maximum for 18 h of the fermentation processes because of the high-specific oxygen demand for culture viability. After exponential growth phase, OUR decreased because of decreasing metabolic activity of cells [57]. The oxygen transfer rate (OTR) in fermentation processes has been related to volumetric mass transfer coefficient,  $k_L a$  and oxygen consumption by the microorganism. The OTR increased because of high oxygen demand for 18 h. The OTR is a critical factor for PHA biosynthesis, because OTR is equal to OUR under oxygen limitation. Thus, variation in cellular respiration could increase the reducing power (NADPH/NADP<sup>+</sup>), which is an important cofactor involved in PHA biosynthesis and regulation [5].

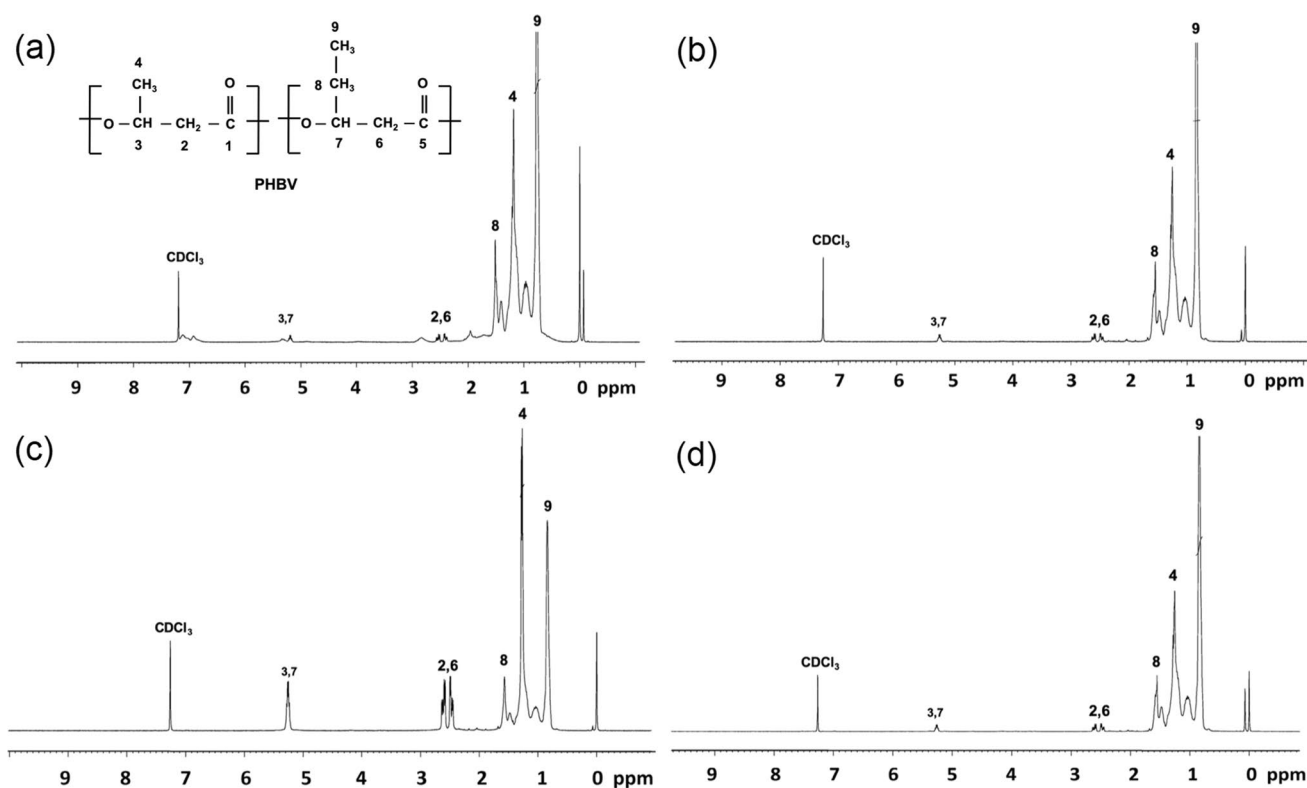
Besides,  $k_L a$ , OTR and OUR values in *rec-B. megaterium* cells were generally higher than wild-type cells. This can be explained by *rec-B. megaterium* cells having a higher metabolic activity, especially in biopolymer synthesis. Moreover, with the increasing glucose feeding rates, higher oxygen transfer parameters were attained, due to the increased oxygen demand of the cells, as expected.

## Polymer characterization

<sup>1</sup>H-NMR spectroscopy was performed to identify the chemical structure of the produced PHA polymers, as shown in Fig. 6. The peak at 0.90 ppm corresponds to the protons of the terminal methyl group of 3HV monomer unit. The peaks at 1.25 ppm, 1.55 ppm, 2.58 ppm and 5.27 ppm correspond to the methyl group of 3HB unit, internal –CH<sub>2</sub> group of 3HV unit, the –CH<sub>2</sub> groups and the –CH groups of 3HB and 3HV monomer units, respectively [58]. The molar fraction of 3HV monomer unit was determined from the relative intensities of methyl groups of 3HV (0.90 ppm) and 3HB (1.25 ppm) monomer units in the <sup>1</sup>H-NMR spectra according to the following equation [53]:

$$\% \text{ 3HV} = \frac{\text{area CH}_3(3\text{HV})}{\text{area CH}_3(3\text{HV}) + \text{area CH}_3(3\text{HB})} \times 100\% .$$

The <sup>1</sup>H-NMR spectrum confirms that the chemical structure of the biopolymers produced in this study corresponds to PHBV copolymer. According to the above equation, integrating the area under the peaks at 0.90 ppm and 1.25 ppm, the PHBV copolymer was found to be composed of 58 mol% of 3HV monomer units. The content of 3HV



**Fig. 6** <sup>1</sup>H-NMR spectrum of PHA synthesized in wild-type and recombinant *B. megaterium* NRRL B-14308 cells under batch and fed-batch cultivations, using: **a** PHA-B-wt, **b** PHA-B-rec, **c** PHA-FB-wt, **d** PHA-FB-rec

monomer units in PHBV copolymer is important for industrial applications because P(3HB-co-3HV) copolymers containing more than 20 mol% 3HV monomers exhibit superior material properties, such as impact strength and polymer flexibility, suitable for manufacturing of films and fibers [59, 60]. This study is significant for production of PHBV by *B. megaterium* strain with the highest 3HV content from unrelated carbon sources, without a precursor addition (Table 3).

Although there is no clear evidence from metabolic pathway analysis on the increase of 3HV content in recombinant *B. megaterium* cells only overexpressing *phaC* gene, it could be explained by the increment in microbial growth, which may digest organic acids and release odd-fatty acid compounds as metabolites, precursors for 3HV production. On the other hand, the observed increase in the 3HV fraction could also be due to the higher PHA synthase activity from the additional copy of *phaC*. In recombinant strains, the higher PHA synthase activity would promote the polymerization of 3HV-CoA in the medium. Briefly, the higher PhaC activity in recombinant *B. megaterium* cells pulls more C5 intermediates into P(3HB-co-3HV) synthesis prior to conversion into shorter C4 intermediates, resulting in a higher 3HV fraction than wild-type strain [61, 62]. Moreover, a further study on enzyme  $k_{cat}$  values for each enzyme in the PHA synthesis pathway, especially for PhaA, PhaB and PhaC, the phenomenon could be better explained, and accordingly plan a better strategy for metabolic engineering. This could also be performed via an initial metabolic flux analysis.

To determine the PHA content and monomer composition within bacterial cells, GC analysis was performed after methanolysis. GC analysis also confirmed the produced polymer to be PHBV copolymer, as two different peaks were observed at retention times of ~0.9 min and ~1.7 min, corresponding to 3HB and 3HV methyl esters in the PHA sample, respectively (Fig. S3). From the GC peak areas, the PHA produced from recombinant *B. megaterium* cells on fed-batch fermentation mode was found to contain 54 mol% HV and 46 mol% HB, where the monomer ratios were consistent with the results obtained from  $^1\text{H-NMR}$  analysis.

The thermophysical properties of the produced PHA polymers were analyzed by TGA and DSC (Table 5). PHA polymers showed two main degradation temperature ranges in agreement with previous studies [20, 63, 64]. The TGA curves for produced PHA showed a gradual weight loss with the increasing temperature, which started at around 260 °C and entirely degrading at around 455 °C. Two-step degradation temperatures are likely due to the incorporation of different monomers, such as 3HV, in the PHA samples. The first-step degradation temperature can be ascribed to decomposition of the crosslinked polymer chain. The second-step degradation is due to the decomposition of the main block of PHBV polymer chain [65].

**Table 5** Thermophysical properties of produced PHA polymers from wild-type (-wt) and recombinant (-rec) *B. megaterium* strains by batch (-B) and fed-batch (-FB) mode of operation and commercial PHB (PHB-C)

Polymer	$T_{d1}$ (°C)	$T_{d2}$ (°C)	$T_{m1}$ (°C)	$T_{m2}$ (°C)	%3HV
PHA-FB-wt	274	452	124	156	52
PHA-B-wt	266	449	129	163	42
PHA-FB-rec	279	455	122	152	58
PHA-B-rec	261	441	126	159	49
PHB-C	287	–	–	172	–

The melting temperatures of PHBV polymers were examined using DSC analysis. The DSC thermogram showed two melting points for the produced PHBV polymers (Table 5). It can be seen that the two melting temperatures were both lower than the melting temperatures of PHB homopolymer, ca. 170 °C [66]. As the proportion of 3HV monomer units in the PHBV polymer chain increases, the melting temperature of the polymer decreases, resulting in an improvement in impact strength and polymer flexibility and broader applications [67]. The occurrence of two melting temperatures could be observed due to melting–re-crystallization–re-melting process of PHBV polymers [68].

## Conclusion

PHAs are currently being produced at about 270,000 tons year<sup>-1</sup> with an increasing demand [29]. The present study aimed to produce PHA biopolymer from *Bacillus megaterium* NRRL B-14308 strain with a higher production efficiency, by investigating the bioprocess design parameters, including dissolved oxygen level and fermentation mode. Also, aiming to increase the production of PHA synthase enzyme (PhaC), recombinant strains were constructed via *phaC* overexpression in *B. megaterium*.

The batch experiments showed that the highest PHA concentration and intracellular PHA content were reached at 20% DO set point. Furthermore, in the batch fermentation mode, there was 1.2-fold increase in the PHA final concentration for rec-*B. megaterium* cells against wild-type *B. megaterium* cells. To maximize PHA productivity, fed-batch fermentations with different exponential glucose feeding rates based on desired specific growth rate were performed. The results demonstrated that fed-batch cultivation increased the volumetric PHA productivity 2.2 times compared to batch cultivation. By controlling the DO level, the highest cell biomass (7.7 g L<sup>-1</sup>), final PHA concentration (6.1 g L<sup>-1</sup>), volumetric PHA productivity (0.54 g L<sup>-1</sup> h<sup>-1</sup>) as well as cell and product yields on substrate (0.74 g g<sup>-1</sup> and 0.62 g g<sup>-1</sup>, respectively) were achieved at the end of the fed-batch cultivation by rec-*B. megaterium* strain at  $\mu_0 = 0.1 \text{ h}^{-1}$ .

Also, the experimental data indicated that well-tuned DO level and availability of excess carbon source and nitrogen limitation enhance the synthesis of PHAs in *B. megaterium*. The characterization of synthesized PHAs was performed by <sup>1</sup>H-NMR, GC-FID, TGA and DSC analyses. The results revealed that the synthesized PHAs composed of 42 mol% of 3HB and 58 mol% of 3HV monomers. Thus, this study is significant for the production of PHBV copolymer with a high 3HV content by *B. megaterium* from an unrelated, simple carbon source, glucose, with no need of precursor addition.

**Acknowledgements** This work was supported by Hacettepe University Scientific Research Projects Coordination Unit (Grant number FBB-2016-11851). We would like to thank Belgin Aslan for her help in TG and DSC analyses and Beray Temelli for her help in <sup>1</sup>H-NMR analysis.

### Compliance with ethical standards

**Conflict of interest** The authors declare that they have no conflict of interest.

### References

- Krishnan S, Chinnadurai GS, Perumal P (2017) Polyhydroxybutyrate by *Streptomyces* sp.: production and characterization. *Int J Biol Macromol* 104:1165–1171
- Kumar M, Singhal A, Verma PK, Thakur IS (2017) Production and characterization of polyhydroxyalkanoate from lignin derivatives by *Pandora* sp. *ISTKB ACS Omega* 2:9156–9163
- Ferre-Guell A, Winterburn J (2018) Biosynthesis and characterization of polyhydroxyalkanoates with controlled composition and microstructure. *Biomacromol* 19:996–1005
- Koller M, Hesse P, Bona R, Kutschera C, Atlci A, BrauneGG G (2007) Potential of various archae- and eubacterial strains as industrial polyhydroxyalkanoate producers from whey. *Macromol Biosci* 7:218–226
- Urtuvia V, Maturana N, Peña C, Díaz-Barrera A (2020) Accumulation of poly(3-hydroxybutyrate-co-3-hydroxyvalerate) by *Azotobacter vinelandii* with different 3HV fraction in shake flasks and bioreactor. *Bioprocess Biosyst Eng* 43:1469
- Marangoni C, Furigo A, de Aragão GMF (2002) Production of poly(3-hydroxybutyrate-co-3-hydroxyvalerate) by *Ralstonia eutropha* in whey and inverted sugar with propionic acid feeding. *Process Biochem* 38:137–141
- Suhazsini P, Keshav R, Narayanan S, Chaudhuri A, Radha P (2020) A study on the synthesis of poly (3-hydroxybutyrate-co-3-hydroxyvalerate) by *Bacillus megaterium* utilizing cheese whey permeate. *J Polym Environ* 28:1390–1405
- Güngörmedi G, Demirbilek M, Mutlu MB, Denkbaş EB, Çabuk A (2014) Polyhydroxybutyrate and hydroxyvalerate production by *Bacillus megaterium* strain A1 isolated from hydrocarbon-contaminated soil. *J Appl Polym Sci* 131:40530
- Haywood GW, Anderson AJ, Roger Williams D, Dawes EA, Ewing DF (1991) Accumulation of a poly(hydroxyalkanoate) copolymer containing primarily 3-hydroxyvalerate from simple carbohydrate substrates by *Rhodococcus* sp. NCIMB 40126. *Int J Biol Macromol* 13:83–88
- Porrás MA, Ramos FD, Diaz MS, Cubitto MA, Villar MA (2019) Modeling the bioconversion of starch to P(HB-co-HV) optimized by experimental design using *Bacillus megaterium* BBST4 strain. *Environ Technol* 40:1185–1202
- Valappil SP, Rai R, Bucke C, Roy I (2008) Polyhydroxyalkanoate biosynthesis in *Bacillus cereus* SPV under varied limiting conditions and an insight into the biosynthetic genes involved. *J Appl Microbiol* 104:1624–1635
- Aldor IS, Kim S-W, Prather KLJ, Keasling JD (2002) Metabolic engineering of a novel propionate-independent pathway for the production of poly(3-hydroxybutyrate-co-3-hydroxyvalerate) in recombinant *Salmonella enterica* serovar typhimurium. *Appl Environ Microbiol* 68:3848–3854
- Chen Q, Wang Q, Wei G, Liang Q, Qi Q (2011) Production in *Escherichia coli* of poly(3-hydroxybutyrate-co-3-hydroxyvalerate) with differing monomer compositions from unrelated carbon sources. *Appl Environ Microbiol* 77:4886–4893
- Yang JE, Choi YJ, Lee SJ, Kang K-H, Lee H, Oh YH, Lee SH, Park SJ, Lee SY (2014) Metabolic engineering of *Escherichia coli* for biosynthesis of poly(3-hydroxybutyrate-co-3-hydroxyvalerate) from glucose. *Appl Microbiol Biotechnol* 98:95–104
- Han J, Hou J, Zhang F, Ai GM, Li M, Cai SF, Liu HL, Wang L, Wang ZJ, Zhang SL, Cai L, Zhao DH, Zhou J, Xiang H (2013) Multiple propionyl coenzyme a-supplying pathways for production of the bioplastic poly(3-hydroxybutyrate-co-3-hydroxyvalerate) in *Haloferax mediterranei*. *Appl Environ Microbiol* 79:2922–2931
- Heinrich D, Raberg M, Steinbuechel A (2015) Synthesis of poly(3-hydroxybutyrate-co-3-hydroxyvalerate) from unrelated carbon sources in engineered *Rhodospirillum rubrum*. *FEMS Microbiol Lett* 362:fnv038
- Zhang YZ, Liu GM, Weng WQ, Ding JY, Liu SJ (2015) Engineering of *Ralstonia eutropha* for the production of poly(3-hydroxybutyrate-co-3-hydroxyvalerate) from glucose. *J Biotechnol* 195:82–88
- Reddy SV, Thirumala M, Mahmood S (2009) Production of PHB and P (3HB-co-3HV) biopolymers by *Bacillus megaterium* strain OU303A isolated from municipal sewage sludge. *World J Microbiol Biotechnol* 25:391–397
- Porrás MA, Vitale C, Villar MA, Cubitto MA (2017) Bioconversion of glycerol to poly(HB-co-HV) copolymer in an inexpensive medium by a *Bacillus megaterium* strain isolated from marine sediments. *J Environ Chem Eng* 5:1–9
- Akdoğan M, Çelik E (2018) Purification and characterization of polyhydroxyalkanoate (PHA) from a *Bacillus megaterium* strain using various dehydration techniques. *J Chem Technol Biotechnol* 93:2292–2298
- Ray S, Kalia VC (2017) Co-metabolism of substrates by *Bacillus thuringiensis* regulates polyhydroxyalkanoate co-polymer composition. *Bioresour Technol* 224:743–747
- Luzier WD (1992) Materials derived from biomass/biodegradable materials. *Proc Natl Acad Sci USA* 89:839–842
- Sim SJ, Snell KD, Hogan SA, Stubbe J, Rha C, Sinskey AJ (1997) PHA synthase activity controls the molecular weight and polydispersity of polyhydroxybutyrate in vivo. *Nat Biotechnol* 15:63–67
- Chek MF, Hiroe A, Hakoshima T, Sudesh K, Taguchi S (2019) PHA synthase (PhaC): interpreting the functions of bioplastic-producing enzyme from a structural perspective. *Appl Microbiol Biotechnol* 103:1131–1141
- Raza ZA, Tariq MR, Majeed MI, Banat IM (2019) Recent developments in bioreactor scale production of bacterial polyhydroxyalkanoates. *Bioprocess Biosyst Eng* 42:901–919
- Huschner F, Grousseau E, Brigham CJ, Plassmeier J, Popovic M, Rha C, Sinskey AJ (2015) Development of a feeding strategy for high cell and PHA density fed-batch fermentation of *Ralstonia eutropha* H16 from organic acids and their salts. *Process Biochem* 50:165–172

27. Suwannasing W, Imai T, Kaewkannetra P (2015) Cost-effective defined medium for the production of polyhydroxyalkanoates using agricultural raw materials. *Bioresour Technol* 194:67–74
28. Choi SY, Rhie MN, Kim HT, Joo JC, Cho IJ, Son J, Jo SY, Sohn YJ, Baritugo K-A, Pyo J, Lee Y, Lee SY, Park SJ (2020) Metabolic engineering for the synthesis of polyesters: a 100-year journey from polyhydroxyalkanoates to non-natural microbial polyesters. *Metab Eng* 58:47–81
29. Choi SY, Cho IJ, Lee Y, Kim Y-J, Kim K-J, Lee SY (2020) Microbial polyhydroxyalkanoates and nonnatural polyesters. *Adv Mater* 32:1907138
30. Kanjanachumpol P, Kulpreecha S, Tolieng V, Thongchul N (2013) Enhancing polyhydroxybutyrate production from high cell density fed-batch fermentation of *Bacillus megaterium* BA-019. *Bioprocess Biosyst Eng* 36:1463–1474
31. RamKumar Pandian S, Deepak V, Kalishwaralal K, Rameshkumar N, Jeyaraj M, Gurunathan S (2010) Optimization and fed-batch production of PHB utilizing dairy waste and sea water as nutrient sources by *Bacillus megaterium* SRKP-3. *Bioresour Technol* 101:705–711
32. Blunt W, Sparling R, Gapes DJ, Levin DB, Cicek N (2018) The role of dissolved oxygen content as a modulator of microbial polyhydroxyalkanoate synthesis. *World J Microbiol Biotechnol* 34:106
33. de Almeida A, Giordano AM, Nikel PI, Pettinari MJ (2010) Effects of aeration on the synthesis of poly(3-hydroxybutyrate) from glycerol and glucose in recombinant *Escherichia coli*. *Appl Environ Microbiol* 76:2036
34. Senior PJ, Beech GA, Ritchie GA, Dawes EA (1972) The role of oxygen limitation in the formation of poly- $\beta$ -hydroxybutyrate during batch and continuous culture of *Azotobacter beijerinckii*. *Biochem J* 128:1193–1201
35. Sambrook JF, Russell DW (2001) *Molecular cloning: a laboratory manual*, 3rd edn. Cold Spring Harbor Laboratory Press, Woodbury
36. Moore SJ, Lawrence AD, Biedendieck R, Deery E, Frank S, Howard MJ, Rigby SEJ, Warren MJ (2013) Elucidation of the anaerobic pathway for the corrin component of cobalamin (vitamin B12). *Proc Natl Acad Sci* 110:14906
37. Çelik E, Çalık P, Oliver SG (2009) Fed-batch methanol feeding strategy for recombinant protein production by *Pichiapastoris* in the presence of co-substrate sorbitol. *Yeast* 26:473–484
38. Taguchi H (1966) Dynamic measurement of the volumetric oxygen transfer coefficient in a fermentation system. *J Ferment Technol* 44:881–889
39. Yıldırım Z, Çelik E (2017) Periplasmic and extracellular production of cellulase from recombinant *Escherichia coli* cells. *J Chem Technol Biotechnol* 92:319–324
40. Braunegg G, Sonnleitner B, Lafferty RM (1978) A rapid gas chromatographic method for the determination of poly- $\beta$ -hydroxybutyric acid in microbial biomass. *Eur J Appl Microbiol Biotechnol* 6:29–37
41. Fei T, Cazeneuve S, Wen Z, Wu L, Wang T (2016) Effective recovery of poly- $\beta$ -hydroxybutyrate (PHB) biopolymer from *Cupriavidus necator* using a novel and environmentally friendly solvent system. *Biotechnol Progress* 32:678–685
42. Bunk B, Schulz A, Stammen S, Münch R, Warren MJ, Rohde M, Jahn D, Biedendieck R (2010) A short story about a big magic bug. *Bioeng Bugs* 1:85–91
43. Srirangan K, Liu X, Tran TT, Charles TC, Moo-Young M, Chou CP (2016) Engineering of *Escherichia coli* for direct and modulated biosynthesis of poly(3-hydroxybutyrate-co-3-hydroxyvalerate) copolymer using unrelated carbon sources. *Sci Rep* 6:36470
44. McCool GJ, Cannon MC (2001) PhaC and PhaR are required for polyhydroxyalkanoic acid synthase activity in *Bacillus megaterium*. *J Bacteriol* 183:4235–4243
45. Chabes A, Domkin V, Larsson G, Liu A, Gräslund A, Wijmenga S, Thelander L (2000) Yeast ribonucleotide reductase has a heterodimeric iron-radical-containing subunit. *Proc Natl Acad Sci* 97:2474
46. Laszlo V, Hoda MA, Garay T, Pirker C, Ghanim B, Klikovits T, Dong YW, Rozsas A, Kenessey I, Szirtes I, Grusch M, Jakopovic M, Samarzija M, Brcic L, Kern I, Rozman A, Popper H, Zöchbauer-Müller S, Heller G, Altenberger C, Ziegler B, Klepetko W, Berger W, Dome B, Hegedus B (2015) Epigenetic down-regulation of integrin  $\alpha 7$  increases migratory potential and confers poor prognosis in malignant pleural mesothelioma. *J Pathol* 237:203–214
47. Liberal V, Martinsson-Ahlzén H-S, Liberal J, Spruck CH, Wid-schwendter M, McGowan CH, Reed SI (2012) Cyclin-dependent kinase subunit (Cks) 1 or Cks2 overexpression overrides the DNA damage response barrier triggered by activated oncoproteins. *Proc Natl Acad Sci* 109:2754
48. Faccin DJL, Rech R, Secchi AR, Cardozo NSM, Ayub MAZ (2013) Influence of oxygen transfer rate on the accumulation of poly(3-hydroxybutyrate) by *Bacillus megaterium*. *Process Biochem* 48:420–425
49. Koller M, Braunegg G (2015) Potential and prospects of continuous polyhydroxyalkanoate (PHA) production. *Bioengineering* 2:94–121
50. Mohapatra S, Maity S, Dash HR, Das S, Pattnaik S, Rath CC, Samantaray D (2017) *Bacillus* and biopolymer: prospects and challenges. *Biochem Biophys Rep* 12:206–213
51. Lee SY (1996) Plastic bacteria? Progress and prospects for polyhydroxyalkanoate production in bacteria. *Trends Biotechnol* 14:431–438
52. Ryu HW, Hahn SK, Chang YK, Chang HN (1997) Production of poly(3-hydroxybutyrate) by high cell density fed-batch culture of *Alcaligenes eutrophus* with phosphate limitation. *Biotechnol Bioeng* 55:28–32
53. García C, Alcaraz W, Acosta-Cárdenas A, Ochoa S (2019) Application of process system engineering tools to the fed-batch production of poly(3-hydroxybutyrate-co-3-hydroxyvalerate) from a vinasses–molasses mixture. *Bioprocess Biosyst Eng* 42:1023–1037
54. Sun ZY, Ramsay JA, Guay M, Ramsay BA (2006) Automated feeding strategies for high-cell-density fed-batch cultivation of *Pseudomonas putida* KT2440. *Appl Microbiol Biotechnol* 71:423–431
55. Zhou Y, Han L-R, He H-W, Sang B, Yu D-L, Feng J, Zhang X (2018) Effects of agitation, aeration and temperature on production of a novel glycoprotein GP-1 by *Streptomyces kanasensis* ZX01 and scale-up based on volumetric oxygen transfer coefficient. *Molecules* 23:125
56. Sinha J, Tae Bae J, Pil Park J, Hyun Song C, Won Yun J (2001) Effect of substrate concentration on broth rheology and fungal morphology during exo-biopolymer production by *Paecilomyces japonica* in a batch bioreactor. *Enzyme Microb Technol* 29:392–399
57. Garcia-Ochoa F, Gomez E, Santos VE, Merchuk JC (2010) Oxygen uptake rate in microbial processes: an overview. *Biochem Eng J* 49:289–307
58. Wang C, Zheng Y, Sun Y, Fan J, Qin Q, Zhao Z (2016) A novel biodegradable polyurethane based on poly(3-hydroxybutyrate-co-3-hydroxyvalerate) and poly(ethylene glycol) as promising biomaterials with the improvement of mechanical properties and hemocompatibility. *Polym Chem* 7:6120–6132
59. Alsafadi D, Al-Mashaqbeh O (2017) A one-stage cultivation process for the production of poly-3-(hydroxybutyrate-co-hydroxyvalerate) from olive mill wastewater by *Haloferax mediterranei*. *N Biotechnol* 34:47–53

60. de Paula FC, de Paula CBC, Gomez JGC, Steinbüchel A, Contiero J (2017) Poly(3-hydroxybutyrate-co-3-hydroxyvalerate) production from biodiesel by-product and propionic acid by mutant strains of *Pandoraea* sp. *Biotechnol Progress* 33:1077–1084
61. Han J, Qiu Y-Z, Liu D-C, Chen G-Q (2004) Engineered *Aeromonas hydrophila* for enhanced production of poly(3-hydroxybutyrate-co-3-hydroxyhexanoate) with alterable monomers composition. *FEMS Microbiol Lett* 239:195–201
62. Fukui T, Kichise T, Iwata T, Doi Y (2001) Characterization of 13 kDa granule-associated protein in *Aeromonascaviae* and biosynthesis of polyhydroxyalkanoates with altered molar composition by recombinant bacteria. *Biomacromol* 2:148–153
63. Kuciel S, Mazur K, Jakubowska P (2019) Novel biorenewable composites based on poly (3-hydroxybutyrate-co-3-hydroxyvalerate) with natural fillers. *J Polym Environ* 27:803–815
64. Singh S, Mohanty AK, Sugie T, Takai Y, Hamada H (2008) Renewable resource based biocomposites from natural fiber and polyhydroxybutyrate-co-valerate (PHBV) bioplastic. *Compos A* 39:875–886
65. Hasan SK, Zainuddin S, Tanthongsack J, Hosur M, Allen L (2018) A study of poly (3-hydroxybutyrate-co-3-hydroxyvalerate) biofilms' thermal and biodegradable properties reinforced with halloysite nanotubes. *J Compos Mater* 52:3199–3207
66. Salgaonkar BB, Bragança JM (2015) Biosynthesis of poly(3-hydroxybutyrate-co-3-hydroxyvalerate) by *Halogeometricum borinquense* strain E3. *Int J Biol Macromol* 78:339–346
67. Luangthongkam P, Laycock B, Evans P, Lant P, Pratt S (2019) Thermophilic production of poly(3-hydroxybutyrate-co-3-hydroxyvalerate) by a mixed methane-utilizing culture. *N Biotechnol* 53:49–56
68. Don T-M, Chen CW, Chan T-H (2006) Preparation and characterization of poly(hydroxyalkanoate) from the fermentation of *Haloferaxmediterranei*. *J Biomater Sci Polym Ed* 17:1425–1438
69. Alkotaini B, Sathiyamoorthi E, Kim BS (2015) Potential of *Bacillusmegaterium* for production of polyhydroxyalkanoates using the red algae *Gelidiummamsii*. *Biotechnol Bioprocess Eng* 20:856–860
70. Alkotaini B, Koo H, Kim BS (2016) Production of polyhydroxyalkanoates by batch and fed-batch cultivations of *Bacillusmegaterium* from acid-treated red algae. *Korean J Chem Eng* 33:1669–1673
71. Rodriguez-Contreras A, Koller M, Dias MMD, Calafell-Monfort M, Braunegg G, Marques-Calvo MS (2013) High production of poly(3-hydroxybutyrate) from a wild *Bacillusmegaterium* Bolivian strain. *J Appl Microbiol* 114:1378–1387

**Publisher's Note** Springer Nature remains neutral with regard to jurisdictional claims in published maps and institutional affiliations.