



Promoted performance of microbial fuel cells using *Escherichia coli* cells with multiple-knockout of central metabolism genes

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Abstract

The effect of central metabolic activity of *Escherichia coli* cells acting as biocatalysts on the performance of microbial fuel cells (MFCs) was studied with glucose used as the energy source. Milliliter-scale two-chambered MFCs were used with 2-hydroxy-1,4-naphthoquinone (HNQ) as an electron mediator. Among the single-gene deletions examined, *frdA*, *pdhR*, *ldhA*, and *adhE* increased the average power output of the constructed MFC. Next, multiple-gene knockout mutants were constructed using P1 transduction. The $\Delta 5$ ($\Delta frdA \Delta pdhR \Delta ldhA \Delta adhE \Delta pta$) strain showed the highest ave. power output (1.82 mW) and coulombic efficiency (21.3%). Our results show that the combination of multiple-gene knockout in *E. coli* cells leads to the development of an excellent catalyst for MFCs. Finally, preventing a decrease in the pH of the anodic solution was a key factor for improving the power output of the $\Delta 5$ strain, and a maximum ave. power output of 2.21 mW was achieved with 5% NaHCO₃ in the buffer. The ave. power density of the constructed MFC was 0.27 mW/cm³, which is comparable to an enzymatic fuel cell of a Milliliter-scale using glucose dehydrogenase.

Keywords Microbial fuel cell · *Escherichia coli* · Central metabolism · Knockout mutant · Glucose

Introduction

Clean energy production from renewable substrates is essential for achieving sustainable energy supplies. Microbial fuel cells (MFCs) are devices that convert the chemical energy of biological fuels into electrical energy using microorganisms as biocatalysts [1–3]. MFCs have several key advantages over conventional fuel cells. For example, MFCs can be

operated under mild conditions and do not require expensive inorganic catalysts or purification of the biocatalysts [4], which allows for large-scale, long-lifetime operation.

Looking back at the long history of MFC research, the idea of using microorganisms to produce electricity was, remarkably, first conceived and reported in 1911, using *E. coli* cells as the biocatalyst [5]. *E. coli* shows promise as a biocatalyst for MFCs because this bacterium can utilize a broad spectrum of organic compounds. However, electron mediators are necessary for *E. coli* cells to be used in MFC systems [6–8], because *E. coli* cells do not possess the extracellular electron transfer (EET) pathways found in electricigenic bacteria such as *Shewanella oneidensis* or *Geobacter sulfurreducens* [9, 10]. Due to the lack of EET pathways in *E. coli* cells, many researchers have attempted to promote the performance of *E. coli*-based MFCs by endowing *E. coli* cells with an EET pathway either via gene modification [11–15] or by modifying the electrode or the MFC system [16–18] for more efficient electron transfer. However, modifying the central metabolic activity of *E. coli* on the performance of MFCs have not been well studied.

In the present study, the performance of *E. coli*-based MFCs using *E. coli* cells that had multiple central metabolism genes

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knocked out was examined. In the first step, single-gene knock-out mutants were compared in terms of their effect on the output of a constructed MFC system. Next, multiple-gene knock-out mutants were constructed by sequential knockout of genes that had increased the MFC power output, and the combination of multiple genes knocked out with highest power output was determined. Furthermore, the effect of 2-hydroxy-1,4-naphthoquinone (HNQ) and sodium hydrogen carbonate concentrations on the output of the $\Delta 5$ strain was examined to consider the electron transfer efficiency and pH value at the anode.

Materials and methods

Bacterial strains

The *E. coli* strains used in this study are listed in Table 1. *E. coli* K-12 strain BW25113 and its derivatives [19] were obtained from the National BioResource Project (NIG, Mishima, Japan). The BW25113(DE3) strain was constructed in previous work [20]. In our study, multiple-deficiency mutant strains were constructed from the BW25113(DE3) strain by P1 transduction using P1kc phage [20]. Briefly, P1 transduction was performed on a kanamycin (Km)-sensitive recipient strain, with the donor strain harboring a Km cassette

between flippase recognition target sites (FRT-Km-FRT cassette), which was inserted at the desired gene on the chromosome. Km resistance was used to select the recipient strain. The Km cassette was eliminated from the recipient strain by FLP/FRT (Flippase/Flippase Recognition Target) recombination. For multiple-deficiency mutant strains, the P1 transduction and elimination of the Km cassette steps were repeated. The deletion was confirmed by PCR using loci-specific primers (Supplementary Table S1).

Construction of MFCs

Milliliter-scale two-chambered MFCs were fabricated according to a method described previously with minor modifications [21]. Briefly, anode and cathode tanks of 8.3 mL capacity were constructed from an acrylic rod of diameter 3.4 cm. Commercial carbon rods with diameters of 0.5 cm were used as cathodes. Carbon fiber bundles (Mitsubishi Rayon Co. Ltd, Japan) were used as high-performance anodes. The diameter of one filament was 7 μm and one bundle consisted of approximately 12,000 filaments. For one anode, 10 bundles approximately 8 cm long were used to maintain contact with *E. coli* cells in the anode solution. A cation-specific membrane, GORE-SELECT (Japan Gore-Tex Inc. Japan), of thickness 30 μm , was used to separate the two electrode chambers.

Table 1 *E. coli* strains used in this study

| Strain | Note | Reference |
|-------------------------------|---|------------------|
| <i>E. coli</i> | | |
| BW25113 | <i>rrnBT14</i> Δ <i>lacZ</i> WJ16 <i>hsdR514</i> Δ <i>araBADAH33</i> | Baba et al. [19] |
| Δ <i>frdA</i> (JW4115) | BW25113, Δ <i>frdA</i> ::FRT-Km-FRT | Baba et al. [19] |
| Δ <i>pdhR</i> (JW0109) | BW25113, Δ <i>pdhR</i> ::FRT-Km-FRT | Baba et al. [19] |
| Δ <i>ldhA</i> (JW1375) | BW25113, Δ <i>ldhA</i> ::FRT-Km-FRT | Baba et al. [19] |
| Δ <i>adhE</i> (JW1228) | BW25113, Δ <i>adhE</i> ::FRT-Km-FRT | Baba et al. [19] |
| Δ <i>pta</i> (JW2294) | BW25113, Δ <i>pta</i> ::FRT-Km-FRT | Baba et al. [19] |
| Δ <i>ackA</i> (JW2293) | BW25113, Δ <i>ackA</i> ::FRT-Km-FRT | Baba et al. [19] |
| Δ <i>poxB</i> (JW0855) | BW25113, Δ <i>poxB</i> ::FRT-Km-FRT | Baba et al. [19] |
| Δ <i>acs</i> (JW4030) | BW25113, Δ <i>acs</i> ::FRT-Km-FRT | Baba et al. [19] |
| Δ <i>pflB</i> (JW0886) | BW25113, Δ <i>pflB</i> ::FRT-Km-FRT | Baba et al. [19] |
| Δ <i>adhP</i> (JW1474) | BW25113, Δ <i>adhP</i> ::FRT-Km-FRT | Baba et al. [19] |
| Δ <i>arcA</i> (JW4364) | BW25113, Δ <i>arcA</i> ::FRT-Km-FRT | Baba et al. [19] |
| Δ <i>arcB</i> (JW5536) | BW25113, Δ <i>arcB</i> ::FRT-Km-FRT | Baba et al. [19] |
| Δ <i>rpoS</i> (JW5437) | BW25113, Δ <i>rpoS</i> ::FRT-Km-FRT | Baba et al. [19] |
| BW25113(DE3) | Derived from BW25113, λ DE3 phage lysogen | Koma et al. [20] |
| Δ <i>frdA</i> (DE3) | BW25113(DE3), Δ <i>frdA</i> ::FRT-Km-FRT | This study |
| $\Delta 2$ | Derived from Δ <i>frdA</i> (DE3), Δ <i>frdA</i> Δ <i>pdhR</i> :: FRT-Km-FRT | This study |
| $\Delta 3$ | Δ <i>frdA</i> Δ <i>pdhR</i> Δ <i>ldhA</i> ::FRT-Km-FRT | This study |
| $\Delta 4$ | Δ <i>frdA</i> Δ <i>pdhR</i> Δ <i>ldhA</i> Δ <i>adhE</i> ::FRT-Km-FRT | This study |
| $\Delta 5$ | Δ <i>frdA</i> Δ <i>pdhR</i> Δ <i>ldhA</i> Δ <i>adhE</i> Δ <i>pta</i> ::FRT-Km-FRT | This study |
| $\Delta 6-1$ | Δ <i>frdA</i> Δ <i>pdhR</i> Δ <i>ldhA</i> Δ <i>adhE</i> Δ <i>pta</i> Δ <i>poxB</i> ::FRT-Km-FRT | This study |
| $\Delta 6-2$ | Δ <i>frdA</i> Δ <i>pdhR</i> Δ <i>ldhA</i> Δ <i>adhE</i> Δ <i>pta</i> Δ <i>poxB</i> ::FRT-Km-FRT | This study |
| $\Delta 6-3$ | Δ <i>frdA</i> Δ <i>pdhR</i> Δ <i>ldhA</i> Δ <i>adhE</i> Δ <i>pta</i> Δ <i>arcA</i> ::FRT-Km-FRT | This study |

Operation of MFCs

The fundamental anode solution (7.5 mL) was composed of 50 g/L glucose, 10 g/L sodium hydrogen carbonate (NaHCO₃), 0.2 g/L ammonium sulfate, and 0.5 M phosphate buffer (pH 8.0) in reverse osmosis water. This solution also contained 1 g/L 2-hydroxyl-1,4-naphthoquinone (HNQ) as an electron transfer mediator [21]. The cathodic chamber was filled with 7.5 mL of deionized water containing 158 g/L potassium ferri-cyanide, K₃[Fe(CN)₆]. Each *E. coli* strain was pre-cultured in lysogeny broth (LB) medium at 30 °C. After 24 h, glucose was added to the culture broth at a concentration of 50 g/L and cultured for 1 h as a warm-up culture for the MFC operation.

E. coli cells were harvested by centrifugation (10,000×g, 5 min). The harvested *E. coli* cells were resuspended using the anode solution to an OD₆₀₀=64 and added to the anodic chamber. The MFC was connected to an ammeter and a voltmeter (Sanwa Electric Instrument, Tokyo, Japan) and operated at 37 °C. Current and voltage were recorded via a computer using PC Kink 7 software (Sanwa Electric Instrument) through a connection with a 100 Ω resistor.

Analyses

Samples of 0.2 mL were withdrawn from the anode solution of a fuel cell for measuring the glucose concentration. After centrifugation (10,000×g, 5 min), the glucose concentration was determined by HPLC (GL Sciences Inc. Tokyo, Japan) using an ULTRON PS-80 N column and ultrapure water as the solvent with a flow rate of 0.6 mL/min at 60 °C. Glucose was detected using a refractive index detector.

The coulombic efficiency was calculated as $100 \times C_p / C_T$ [22], where C_p is the total coulombs calculated by integrating the current over time, and C_T is the theoretical amount of coulombs available from glucose, which is given by the following equation:

$$C_T = \frac{FN\Delta S\nu}{M},$$

where F is Faraday's constant (98,485 C per mol of electrons), N is the number of mol of electrons generated per mol of glucose (24 electrons per mol), ΔS is the concentration of glucose consumed, ν is the volume of liquid, and M is the molecular mass of glucose.

Results and discussion

Performance of the MFC with single-gene knockout mutants

Figure 1 shows an overview of the modification of *E. coli* central metabolism as a biocatalyst for improving the

power output in a constructed MFC system. In total, 13 genes associated with the central metabolism of *E. coli* cells were selected for single-gene deletion. Eight genes, *ldhA*, *adhE*, *adhP*, *pta*, *poxB*, *ackA*, *acs* and *pflB*, are involved in the synthesis or degradation of organic acids generated by branching from glycolysis [23]. Four genes, *pdhR*, *arcA*, *arcB* and *rpoS*, are regulators to repress the activity of the TCA cycle under microaerobic or anaerobic conditions [24]. The *frdA* gene encodes the enzyme which catalyzes the reduction of fumarate to succinate in the TCA cycle [25]. We expected that knockouts of these genes would lead to more NADH (i.e., electrons) generation from the TCA cycle.

Figure 2 shows the time course of power outputs from whole fuel cells composed of various single-gene knockout mutants of *E. coli* strains. For the wild-type (WT) BW25113 strain, the power started at 1.5 mW and reached a peak value of 2.3 mW after 60 min. This value is comparable to that of a similar MFC system that used yeast cells [21], indicating that *E. coli* can be used as a biocatalyst in the examined MFC system. After 60 min, the power output decreased in a linear manner to 0.5 mW at the end of the reaction (1080 min). Furthermore, for the single-gene knockout mutant strains, while the peak values and the slopes of decreasing power differed, the time courses showed a similar tendency. Among the mutant strains, the $\Delta frdA$, $\Delta pdhR$, $\Delta ldhA$ and $\Delta adhE$ strains had higher peak values of power (approximately 2.5 mW) when compared with that of the WT strain, and these values were maintained at a higher level throughout the MFC operation period. As for the genes involved in acetate metabolism (Δpta , $\Delta ackA$, $\Delta poxB$ and Δacs), the $\Delta poxB$ strain showed a relatively higher peak value; however, the decrease in power output was much steeper than for other strains, especially during the latter stages of the operating period. In contrast, the Δpta strain showed the smallest peak value of 2.0 mW at 60 min. The deletion of genes related to formate and ethanol production ($\Delta adhP$ and $\Delta pflB$) and regulators of the TCA cycle did not improve the power output drastically.

The ave. power outputs of whole fuel cells for 18 h are summarized in Fig. 3. The ave. power of the WT strain was 1.25 mW, whereas four strains ($\Delta frdA$, $\Delta pdhR$, $\Delta ldhA$, and $\Delta adhE$) showed significantly higher ave. power values. The maximum ave. value was 1.49 mW, which was achieved by the $\Delta adhE$ strain. The power outputs of the other nine strains (Δpta , $\Delta ackA$, $\Delta poxB$, Δacs , $\Delta adhP$, $\Delta pflB$, $\Delta arcA$, $\Delta arcB$ and $\Delta rpoS$) were the same or less than the WT strain. The ave. power output of the $\Delta rpoS$ strain was significantly lower than the other strains, at 1.09 mW.

The anodic chamber of the MFC constructed was considered to be under oxygen-limited conditions because there

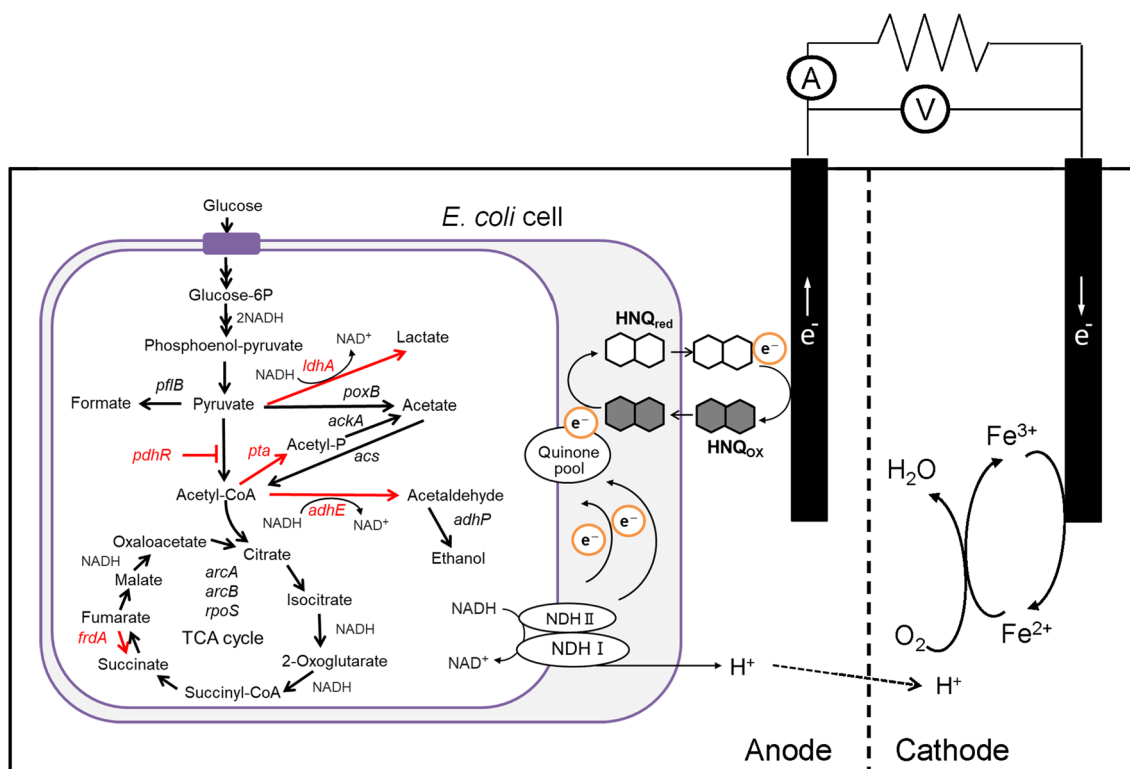


Fig. 1 Overview of the modification of *E. coli* central metabolism as a biocatalyst for improving power output in a constructed MFC system. This overview includes the hypothetical mechanism of electron

transfer, in which HNQ obtains electrons from a quinone pool such as ubiquinone in the cell membrane of *E. coli* cells

was minimal oxygen supply and the cell concentration was extremely high ($OD_{600} = 64$). In *E. coli*, under oxygen-limited conditions, the *frdA* gene encodes fumarate reductase, which catalyzes the reduction of fumarate accompanied by the oxidation of reduced menaquinone [25]. Deletion of the *frdA* gene has been reported to significantly suppress succinate production and slightly increases lactate production [26, 27]. These results suggest that the accumulation of reduced compounds arising from *frdA* deletion promoted the performance of MFC. The pyruvate dehydrogenase complex regulator (PdhR) is a transcriptional regulator that negatively controls the formation of the pyruvate dehydrogenase complex (PDHc), NADH dehydrogenase (NDH)-2 and cytochrome bo3 oxidase in *E. coli*. Under oxygen-limited conditions, it has been reported that *pdhR*-deficient mutants grow and consume glucose more efficiently than the WT strain, with enhanced respiration because of the increased activities of PDHc and NDH-2 [28]. These reports support our result of improved MFC output following *pdhR* deletion. The *ldhA* and *adhE* genes encode lactate dehydrogenase and alcohol dehydrogenase, respectively [23]. Given that these enzymes consume NADH to produce anaerobic byproducts, it is not unexpected that the deletion of these genes increased the power output of *E. coli* MFCs (Fig. 1).

Improvement of MFC performance with multiple-knockout mutants

Next, the cumulative effects of multiple-gene deletions in *E. coli* cells on MFC performance were examined. The BW25113(DE3) strain containing the λ DE3 lysogen was used as the parent strain in a series of experiments. First, four genes (*frdA*, *pdhR*, *ldhA* and *adhE*) were selected for multiple-gene deletions, because deletion of each of these genes increased MFC the power. Although single-gene deletions of *pta*, *poxB* and *pflB* had no great effect on MFC power output, these enzymes are involved in the acetate and formate production pathways and play an important role in central, anaerobic carbon metabolism reactions. These genes were also selected for deletion because the consequences may differ when these genes are cumulatively deleted. The *arcA* gene was also chosen for deletion because a previous report described improved MFC performance using the Δ *arcA* strain [6]. In that report, the Δ *arcA* strain increased the power output because of the activation of key enzymes in the TCA cycle.

Figure 4 shows the time course of MFC power outputs using multiple-knockout mutants. Compared with the WT (BW25113(DE3)) strain, all multiple-knockout mutants showed much higher peak values of power output. While

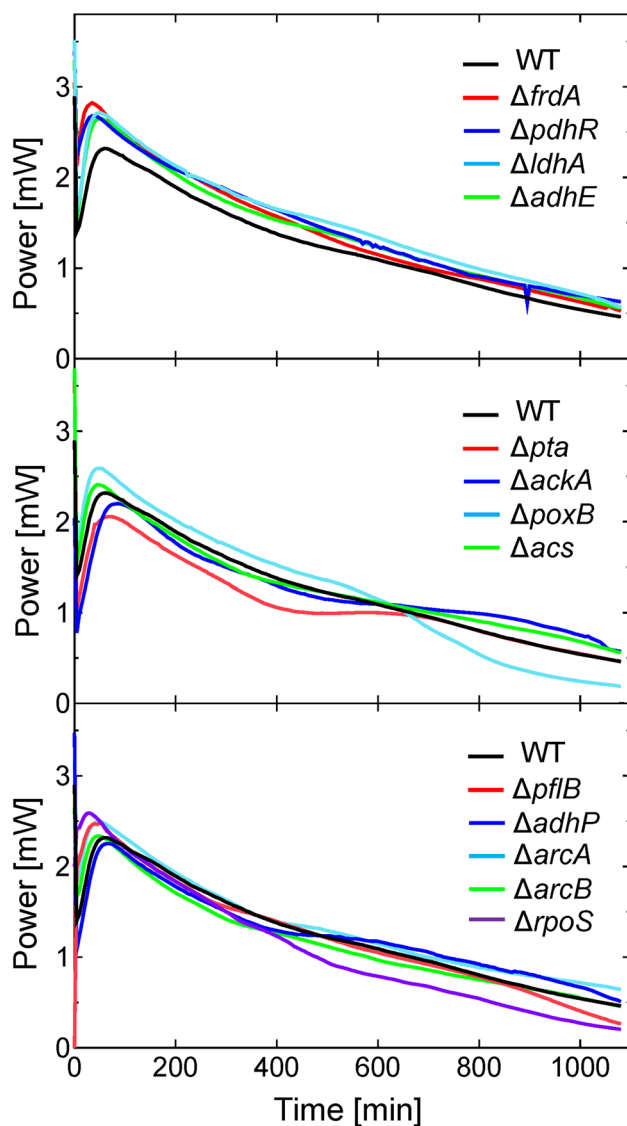


Fig. 2 Time course of power output of whole fuel cells composed of various single-gene knockout mutant *E. coli* strains. Power was calculated based on the measurement of voltage and current. The lines show the average value of power obtained from more than three independent experiments

the double-knockout mutant $\Delta 2$ ($\Delta frdA \Delta pdhR$) showed a rapid decrease in power output after 200 min, the triple-knockout mutant $\Delta 3$ ($\Delta frdA \Delta pdhR \Delta ldhA$) slightly suppressed the power drop during the latter stages of the operation period. The $\Delta 4$ ($\Delta frdA \Delta pdhR \Delta ldhA \Delta adhE$) and $\Delta 5$ ($\Delta frdA \Delta pdhR \Delta ldhA \Delta adhE \Delta pta$) strains maintained relatively high power outputs and their time profiles overlapped. The sixth mutation of each of *pflB*, *poxB*, and *arcA* in the $\Delta 5$ strain did not improve the power profile.

The ave. power outputs are summarized in Fig. 5. The ave. power of BW25113(DE3) was 1.23 mW, which was almost equal to that of BW25113 (1.25 mW). This

result suggests that the insertion of λ DE3 lysogen into the genomic DNA of BW25113 did not affect its performance in the MFC. The ave. power of $\Delta 2$ was 1.46 mW, which was slightly higher than that of the single-knockout mutants, $\Delta frdA$ (1.41 mW) and $\Delta pdhR$ (1.40 mW). The additional deletion of *ldhA* ($\Delta 3$) increased the ave. power to 1.57 mW. A cumulative effect of multiple-gene deletions on power output was observed. Further deletion of the *adhE* gene ($\Delta 4$) significantly promoted the performance of the MFC (1.81 mW) due to the slower decrease in power during the latter stages of operation period. Thus, the deletion of four genes (*frdA*, *pdhR*, *ldhA*, and *adhE*) resulted in a cumulative effect that increased the MFC power output. In addition, the ave. power of the $\Delta 5$ strain was 1.82 mW, and the additional deletion of the *pta* gene did not increase the MFC performance. Furthermore, a sixth mutation of each of *pflB*, *poxB*, and *arcA* in the $\Delta 5$ strain negatively affected the power output. These results suggest that the cumulative effects of multiple-gene deletions on MFC performance became saturated. The coulombic efficiency of these multiple-gene mutants was calculated based on the amount of glucose consumed (Fig. 5). The coulombic efficiencies of the WT, $\Delta 2$, and $\Delta 3$ strains were 8.1%, 9.5% and 9.1%, respectively, which were not significant. In contrast, the coulombic efficiency of $\Delta 4$ was 14%, which was significantly higher than that of $\Delta 3$. In addition to the increased ave. power, the repression of glucose consumption in the $\Delta 4$ strain is also a key reason for the higher coulombic efficiency (Fig. 5). Furthermore, the coulombic efficiency of the $\Delta 5$ strain reached 21.3%, even though the ave. power of this strain was almost equal to that of $\Delta 4$. This result is also because of the further decrease in glucose consumption by the $\Delta 5$ strain. In terms of fuel economy, $\Delta 5$ is a superior strain when compared with that of $\Delta 4$, although their power outputs are at the same level. Further multiple-gene deletion did not drastically increase coulombic efficiency. Considering both ave. power and the coulombic efficiency, it was concluded that the $\Delta 5$ strain with multiple-gene knockout is the best biocatalyst among the examined strains.

Effect of HNQ concentration on the performance of the $\Delta 5$ strain

Although the $\Delta 5$ strain was selected as the best multiple-gene knockout *E. coli* for MFC, repression of glucose consumption still remained as a serious problem. We hypothesized that the reason for the decrease in glucose consumption was that the electron transfer efficiency was not sufficient for *E. coli* to regenerate the intracellular oxidation state at the anode. To confirm the effect of the mediator concentration on the power and glucose consumption for the $\Delta 5$ strain, lower (0.05%) and higher (0.20%) concentrations of HNQ (basal condition

Fig. 3 Ave. power output of whole fuel cells for 18 h operation with each knockout mutant strain of *E. coli*. Data were obtained from more than three independent experiments. Vertical bars indicate standard deviations. Statistically significant differences from the WT strain ($p < 0.05$) are marked with asterisks

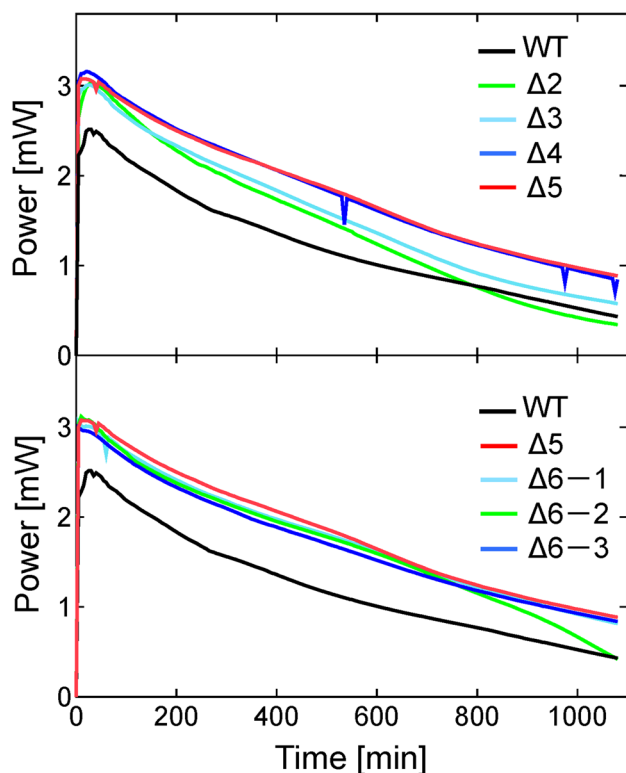
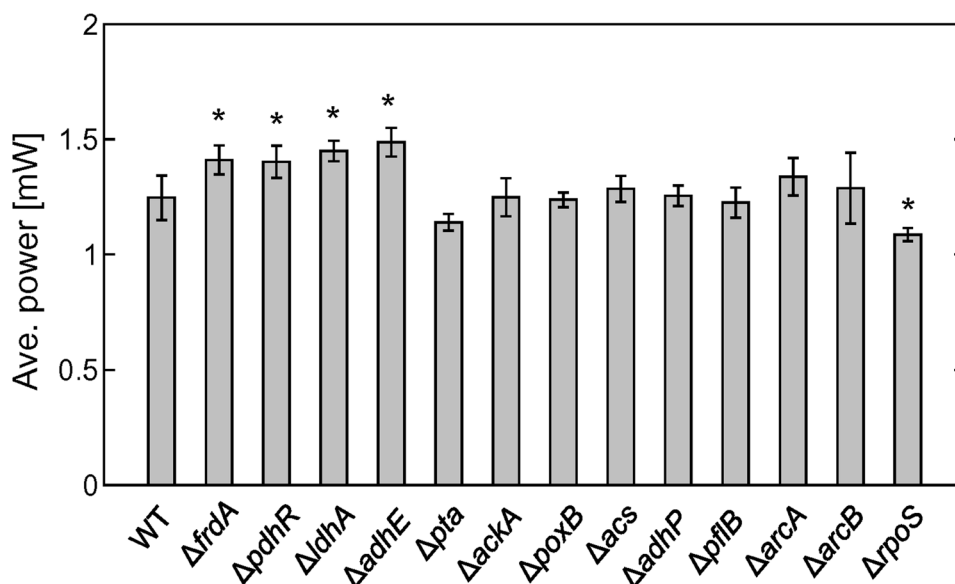


Fig. 4 Time courses of power outputs of whole fuel cells composed of various multiple-gene knockout mutant strains of *E. coli*. The definitions of the strains are given in Table 1. The lines show the average value of the power obtained from more than three independent experiments

0.10%) were added to the anodic chamber. Figure 6a shows the time course of power outputs using different mediator concentrations. In the presence of 0.05% HNQ, the power was clearly lower than that obtained with 0.10% HNQ. The ave. power with 0.05% HNQ was 1.56 mW (Fig. 6a), which was significantly lower when compared with that of 0.10% HNQ (1.82 mW). These results indicate that 0.05% HNQ is not sufficient to transfer all electrons generated from the $\Delta 5$ cells. In contrast, the time courses of power output using 0.10% and 0.20% HNQ almost overlapped and the ave. power did not increase with 0.20% HNQ. Based on these results, we concluded that the basal condition (0.10% HNQ) provided sufficient electron transfer efficiency and further increases in the HNQ concentration did not improve the power output and glucose consumption of $\Delta 5$ cells.

Effect of the NaHCO_3 concentration on the pH value of the anodic solution and performance of the $\Delta 5$ strain

The pH of the anodic solution with the $\Delta 5$ strain was found to decrease considerably during the operating period. Although the pH of the anodic solution was initially adjusted with phosphate buffer (pH 8.0) and 1% NaHCO_3 , the pH value was observed to decrease to 5.1 over the 18 h operation period (Fig. 7a). This result suggests that the considerable pH drop decreased the activity of the $\Delta 5$ strain, which caused a decrease in power and

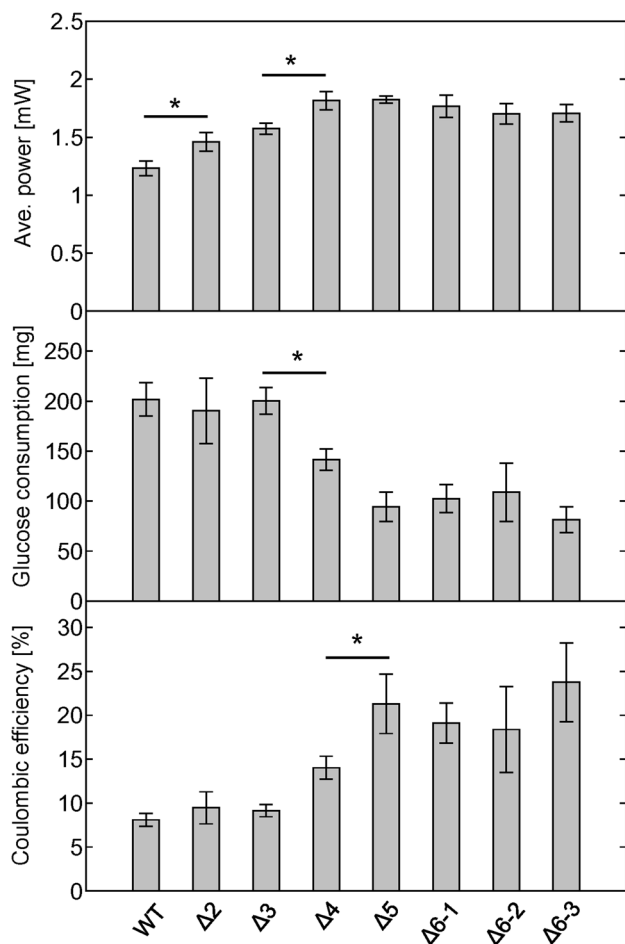


Fig. 5 Ave. power output, glucose consumption, and coulombic efficiency of whole fuel cells for 18 h operation with various multiple-gene knockout mutant strains of *E. coli*. The definitions of the strains are given in Table 1. Data were obtained from more than three independent experiments. Vertical bars indicate standard deviations. Asterisks show statistically significant differences ($p < 0.05$)

glucose consumption. Higher concentrations of NaHCO_3 were added to the anode (3–7%) to prevent the pH drop in the anodic solution of $\Delta 5$ strain. As shown in Fig. 7a, the pH of the anodic solution over 18 h gradually increased with increasing NaHCO_3 concentrations used. The pH with 3% and 5% NaHCO_3 was 6.1 and 6.8, respectively, and reached 7.4 with 7% NaHCO_3 added, indicating that a drop in pH was prevented with higher NaHCO_3 concentrations. Figure 7b shows the time course of MFC power outputs with different NaHCO_3 concentrations. Compared with the basal condition (1% NaHCO_3), the power with 3% NaHCO_3 showed the same level of peak value and the power drop was strongly prevented during the latter stages of the operation period. With 5% NaHCO_3 , the value was maintained at a higher level throughout the MFC operation period. These results show that preventing the decrease in pH in the anodic solution effectively improved the

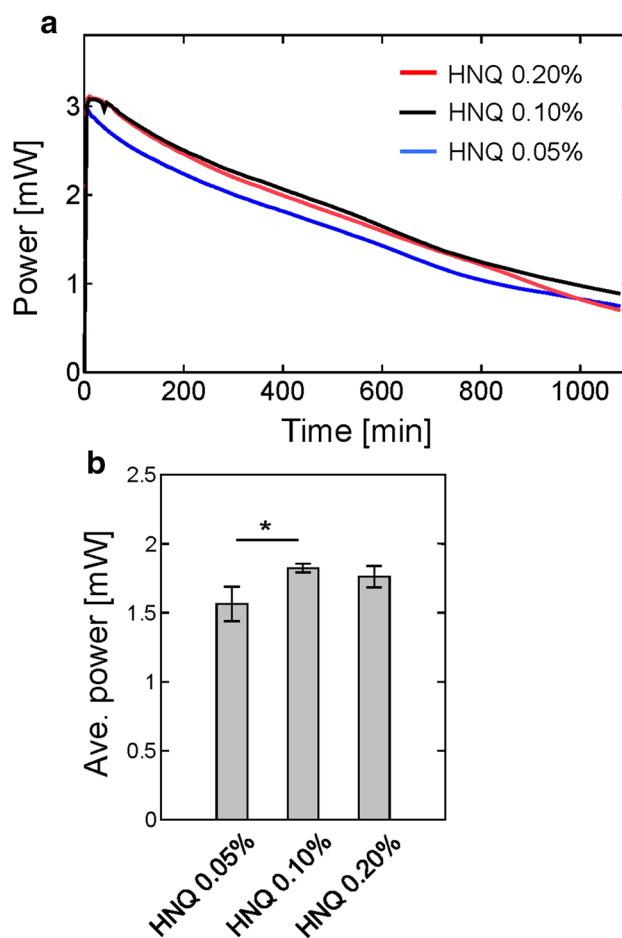


Fig. 6 Effect of the mediator (HNQ) concentration on the performance of MFCs with the *E. coli* $\Delta 5$ strain. **a** Power output of MFCs with 0.05%, 0.10% and 0.20% HNQ. **b** Ave. power outputs of MFCs for 18 h operation with different HNQ concentrations. Data were obtained from more than three independent experiments and vertical bars indicate standard deviations. Asterisks show statistically significant differences ($p < 0.05$)

power output of the $\Delta 5$ strain. A further increase of the NaHCO_3 concentration to 7% showed a smaller peak value of 2.5 mW at 60 min and the power output was maintained at ~2.0 mW throughout the operation period. Therefore, the power output was more constant as the concentration of NaHCO_3 increased. The values of the ave. power output, glucose consumption and coulombic efficiency are summarized in Fig. 7c. The ave. power with 3% NaHCO_3 was 2.08 mW, which was significantly higher than that with 1% NaHCO_3 (1.82 mW). With 5% NaHCO_3 , the ave. power increased slightly to 2.21 mW and was saturated with 7% NaHCO_3 (2.20 mW). Glucose consumption with 3% NaHCO_3 was 180 mg, which was two times higher than that with 1% NaHCO_3 . These results suggested that glucose consumption was strongly promoted by preventing a decrease in pH and this improved the power output

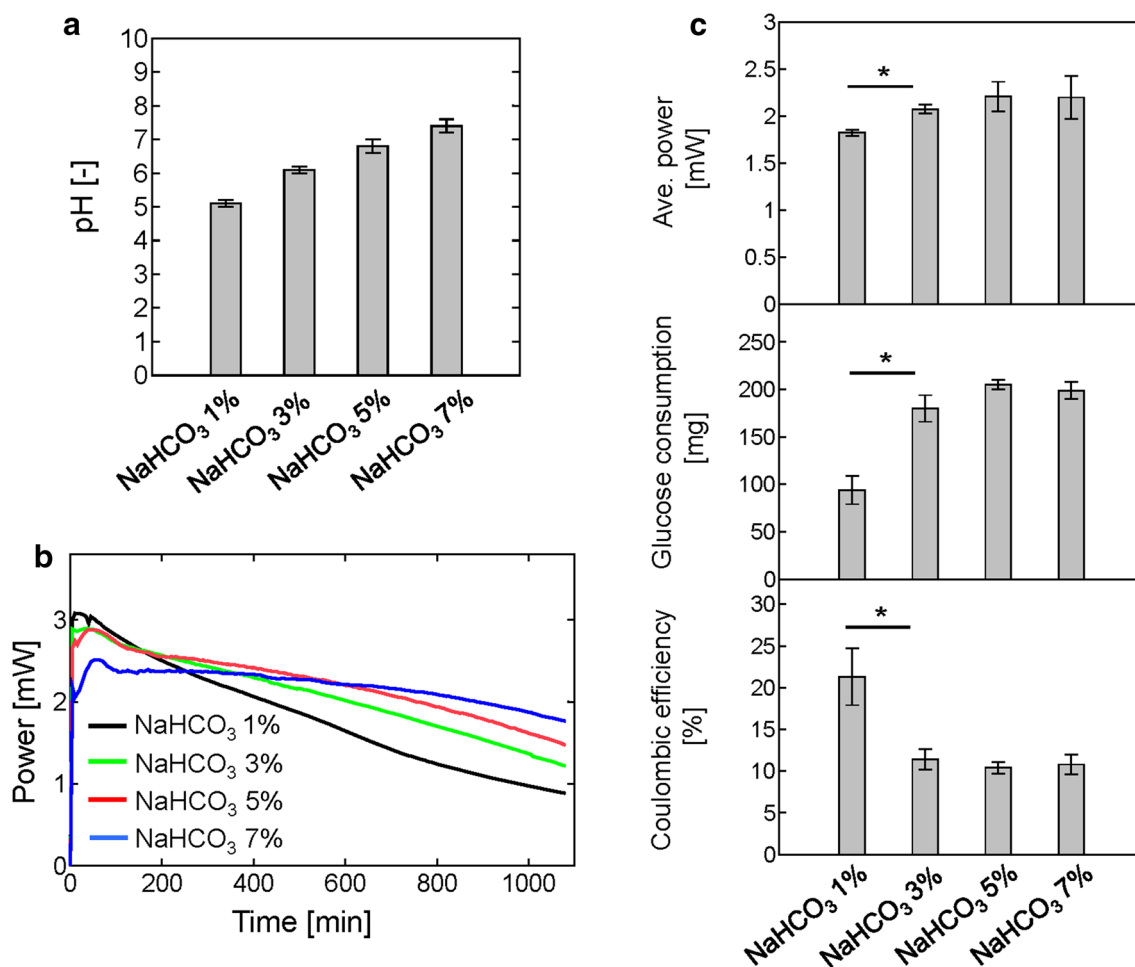


Fig. 7 Effect of NaHCO₃ concentration on the pH value of the anodic solution and performance of the *E. coli* $\Delta 5$ strain. **a** pH value of the anodic solution over 18 h. **b** Time courses of power outputs of the *E. coli* $\Delta 5$ strain. **c** Ave. power output, glucose consumption, and coulombic efficiency of whole fuel cells for 18 h operation. Data were obtained from more than three independent experiments and vertical bars indicate standard deviations. Asterisks show statistically significant differences ($p < 0.05$)

of the $\Delta 5$ strain. Coulombic efficiency decreased with an increase in NaHCO₃ concentration, which was inversely proportional to glucose consumption (Fig. 7c). Thus, controlling the pH was a key factor to improve the power output of the $\Delta 5$ strain and a maximum output of 2.21 mW was achieved with 5% NaHCO₃.

Comparison of the performance of an enzymatic biofuel cell and the MFC prepared in this study

Biofuel cells are categorized based on the type of catalyst used; enzymes or microorganisms. In general, an enzymatic fuel cell (EFC) using glucose oxidase (GOx) or glucose dehydrogenase (GDH) shows a higher power density than a MFC because of the efficient reaction provided by the condensed pure enzymes. In contrast, a MFC can obtain a higher coulombic efficiency because microbial glucose degradation can theoretically obtain 24 electrons, whereas

enzymatic glucose degradation is a 2-electron reaction. The MFC system in the present study has a simple anode component using glucose as a single substrate and that is very similar to EFCs. The results of the MFC used in this study were compared with an enzymatic fuel cell with comparable cell volume (6.5 cm³), operating time (16.6 h) and glucose concentration (1.2 M) [29]. Table 2 shows the comparison of the performance of the EFC and MFC used in this study. The calculated ave. power density of the $\Delta 5$ strain with 5% NaHCO₃ was 0.27 mW/cm³, which reached approximately 35% that obtained by the in EFC (0.77 mW/cm³) of a previous report [29]. The highest coulombic efficiency was 21.3% obtained from the $\Delta 5$ strain with 1% NaHCO₃, which is five times or more when compared with that of the EFC (3.7%). Thus, although the power output is lower when compared with EFC, the constructed MFC provides a considerable improvement on previous MFCs and has reached a comparable level to EFCs.

Table 2 Comparison of the performance of the enzymatic biofuel cell and the microbial fuel cell in this study

| Biocatalyst | Cell volume (cm ³) | Operating time (h) | Ave. power (mW) | Ave. power density (mW/cm ³) | Coulombic efficiency ^a (%) | Reference |
|------------------------------------|--------------------------------|--------------------|-----------------|--|---------------------------------------|------------|
| Glucose dehydrogenase | 6.5 | 16.6 | 4.98 | 0.77 | 3.7 | [29] |
| BW25113DE3 (1%NaHCO ₃) | 8.3 | 18.0 | 1.23 | 0.15 | 8.2 | This study |
| Δ5 (1%NaHCO ₃) | 8.3 | 18.0 | 1.82 | 0.22 | 21.3 | This study |
| Δ5 (5%NaHCO ₃) | 8.3 | 18.0 | 2.20 | 0.27 | 10.4 | This study |

^aCalculation of the theoretical value in terms of a 24-electron reaction by glucose degradation

Conclusions

In this study, the effect of central metabolic activity of *E. coli* cells, using glucose as an energy source on the performance of a MFC was studied. Single-gene deletions of *frdA*, *pdhR*, *ldhA*, and *adhE* increased the power output of the constructed MFC. Multiple-gene knockout mutants created using P1 transduction showed that the Δ5 (*ΔfrdAΔpdhRΔldhAΔadhEΔpta*) strain resulted in the highest ave. power (1.82 mW) and coulombic efficiency (21.3%). Finally, controlling the pH value was found to be a key factor in improving the power output of the Δ5 strain and a maximum ave. power of 2.21 mW was achieved with 5% NaHCO₃. The ave. power density of the constructed MFC was 0.27 mW/cm³, which is comparable in performance to an enzymatic fuel cell (0.77 mW/cm³).

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Compliance with ethical standards

Conflict of interest The authors declare that they have no conflict of interest.

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