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Relative oxidation states of magmas inferred from Ce(IV)/Ce(III) in zircon: application to porphyry copper deposits of northern Chile

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Abstract Major- and trace-element compositions of zircons and whole rocks from 14 barren and seven ore-bearing calc-alkaline intrusions from the Chuquicamata-El Abra porphyry copper belt of northern Chile have been measured in situ by excimer laser ablation (ELA) ICP-MS. These data permit the Ce(IV)/Ce(III) ratio within zircon to be calculated using a lattice-strain model for mineral-melt partitioning of Ce(IV) and Ce(III). Zircon Ce(IV)/Ce(III) and Eu_N/Eu_N^* ratios, and by inference magmatic oxidation states, generally increase from older, mafic to younger, felsic units. Within this sequence, porphyry copper mineralization is directly associated only with intrusions with zircon Ce(IV)/Ce(III) > 300 and $Eu_N/Eu_N^* > 0.4$. Such trends can be understood in terms of interdependent relations between oxygen fugacity, sulfur speciation and solubility, and chalcophile element partitioning in silicate magmas. Because zircon occurs in most calc-alkaline intrusions and is resistant to subsolidus alteration, zircon Ce(IV)/Ce(III) ratios provide a useful tool for evaluating the economic potential of such rocks for magmatic-hydrothermal Cu ± Au mineralization. The approach is general and may provide a means to infer relative oxidation state in a wide range of intermediate to felsic igneous rocks.

Introduction

Studies of magmatic-hydrothermal Cu ± Au deposits reveal a persistent spatial association with oxidized calc-alkaline intrusions (e.g., Candela 1992; Blevin and Chappell 1992; Hedenquist and Lowenstern 1994). This association suggests that a genetic link exists between oxidized magmas and processes leading to Cu ± Au mineralization, most likely involving redox control of the speciation and solubility of magmatic sulfur and its influence on the fractionation of chalcophile elements. From the perspective of mineral exploration, this association means that the oxidation state of intrusions could be used as an empirical criterion to distinguish between ore-bearing and barren magmatic-hydrothermal systems. It is, therefore, of considerable economic as well as scientific interest to determine the magmatic oxidation state of such intrusions. Unfortunately, quantitative oxygen barometers based on Fe–Ti oxides are prone to resetting during slow cooling of intrusive rocks, and empirical redox indicators such as the whole-rock Fe(III)/Fe(II) ratio or the presence of anhydrite rarely survive processes of hydrothermal alteration and surficial weathering. Here, we present a method of determining the Ce(IV)/Ce(III) ratio in zircon and use it to examine the temporal evolution of oxidation state in a suite of barren and ore-bearing intrusions. Zircon is well suited to this task, as it is a widespread accessory mineral in intermediate to felsic igneous rocks and is resistant to hydrothermal alteration and physical and chemical weathering. Zircon partitions Ce(IV) in strong preference to Ce(III), which means the Ce(IV)/Ce(III) ratio should be a sensitive and robust measure of magmatic oxidation state.

This study is based on the Los Picos-Fortuna/Pajonal-El Abra igneous complex, a suite of spatially and temporally associated calc-alkaline intrusions within the Chuquicamata-El Abra porphyry copper belt of northern Chile. The complex is cut by a major strike-slip fault, the West Fault, with the Pajonal-El Abra intrusive suite

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displaced ca. 35 km to the north of the correlative Los Picos-Fortuna intrusive suite. The Los Picos-Fortuna/Pajonal-El Abra complex was emplaced over a period of about 5.5×10^6 years (Ballard et al. 2003) and shows a systematic compositional evolution with time through a sequence of barren intrusions, culminating in the ore-bearing El Abra porphyry. This igneous complex provides an ideal opportunity to examine the role of oxidation state in a magmatic system which eventually produced a porphyry copper deposit. The study is supplemented by analyses of zircons from other ore-bearing porphyries in the belt, including the super-giant Chuquicamata copper deposit, as well as the smaller Radomiro Tomic and Opache deposits, to establish whether there is a systematic difference between ore-bearing and barren intrusions.

Geological setting

Geology of the Chuquicamata-El Abra belt

The Chuquicamata-El Abra belt in northern Chile (Fig. 1) hosts several important porphyry copper

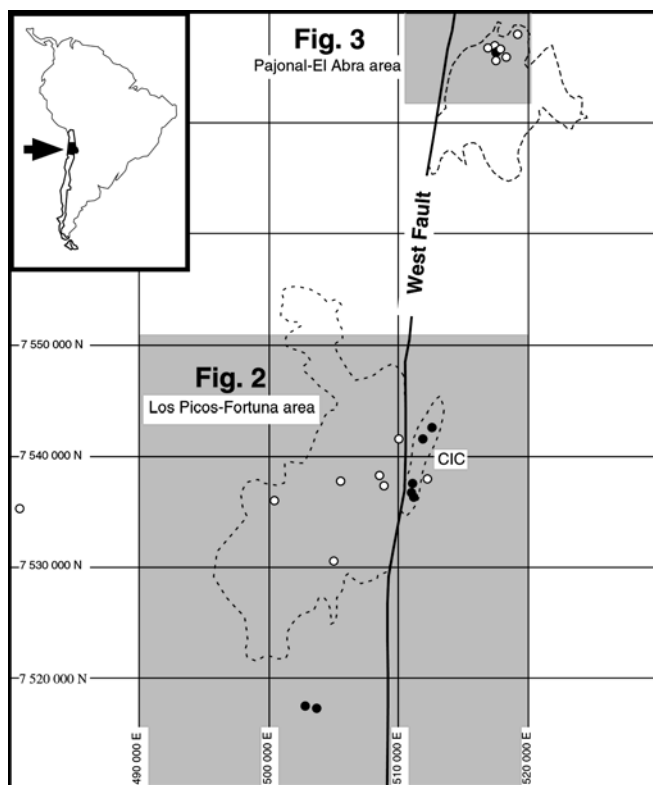


Fig. 1 Map of the Chuquicamata-El Abra porphyry copper belt in northern Chile (modified from Maksaev and Tomlinson 1995; Tomlinson et al. 1995; Reynolds et al. 1998; Pardo and Rivera 1999) showing the approximate outline of the Los Picos-Fortuna and Pajonal-El Abra igneous suites, the Chuquicamata Intrusive Complex, and the locations of igneous rock samples used for zircon trace-element analysis. *Open symbols* indicate barren units, and *closed symbols* are ore-bearing intrusions

deposits, including one of the world's largest, Chuquicamata. It forms part of a more extensive north-south-trending corridor of porphyry copper deposits of Eocene-Oligocene age which includes Collahuasi, Escondida and Potrerillos (Sillitoe 1992). The basement of the belt is composed of a variety of pre-Cenozoic igneous, sedimentary and metamorphic rocks. Calc-alkaline igneous activity dominated the late Cretaceous to Oligocene history of the Chuquicamata-El Abra belt, but only Eocene to Oligocene intrusive rocks host porphyry copper mineralization. The major mines and prospects include Chuquicamata, El Abra, Radomiro Tomic, Opache, and Mansa Mina (Sillitoe et al. 1996; Cuadra et al. 1997; Ossandón and Zentilli 1997; Pardo and Rivera 1999). Miocene to Holocene alluvial, colluvial and eolian sediments as well as some volcanic flows were deposited after emplacement of the porphyries and now mantle the greater part of the region (Boric et al. 1990; Maksaev et al. 1994; Maksaev and Tomlinson 1995; Tomlinson et al. 1995).

Since the Jurassic, regional compression associated with subduction of the Phoenix, Farallon and then Nazca plates (Duncan and Hargraves 1984) has been responsible for deformation along the western margin of South America. Changing patterns of stress have led to extensional, thrust and strike-slip faulting, as well as widespread folding. The West Fault zone is the major structure in the Chuquicamata-El Abra belt, forming part of the north-south-orientated Domeyko Fault System which extends for over 2,000 km in northern Chile (Lindsay et al. 1995; Reutter et al. 1996; Dilles et al. 1997; McInnes et al. 1999).

Los Picos-Fortuna and Pajonal-El Abra igneous suites

The Los Picos-Fortuna igneous suite outcrops west of the Chuquicamata mine across the West Fault and consists of Los Picos quartz monzodiorite (Dilles et al. 1997) and the Fortuna intrusive complex (Fig. 2). Los Picos is thought to consist of multiple plutons, evidenced by monzonitic bodies intruding diorites (Rosas 1999). It outcrops as a north-south-trending belt along the western extremity of the Fortuna complex, and as isolated roof pendants within these intrusions. The Fortuna complex intrudes Los Picos along its eastern and northern margins, forming a continuous linear outcrop approximately 20 km (north-south) by 5 km (east-west). It is composed of Gris granodiorite and monzonite, Clara hornblende granodiorite, small leucocratic intrusions, and the San Lorenzo rhyodacite porphyry (Rosas 1999). Broadly, the Fortuna complex becomes younger from west (Gris) to east (Clara; Rosas 1999).

Forty-five kilometers north of the Chuquicamata mine and 2 km east of the West Fault, the El Abra complex intrudes upper Carboniferous to middle Triassic plutonic and volcanic rocks and Cretaceous continental sediments (Boric et al. 1990; Tomlinson et al. 1995; Maksaev and Tomlinson 1995; Fig. 3).

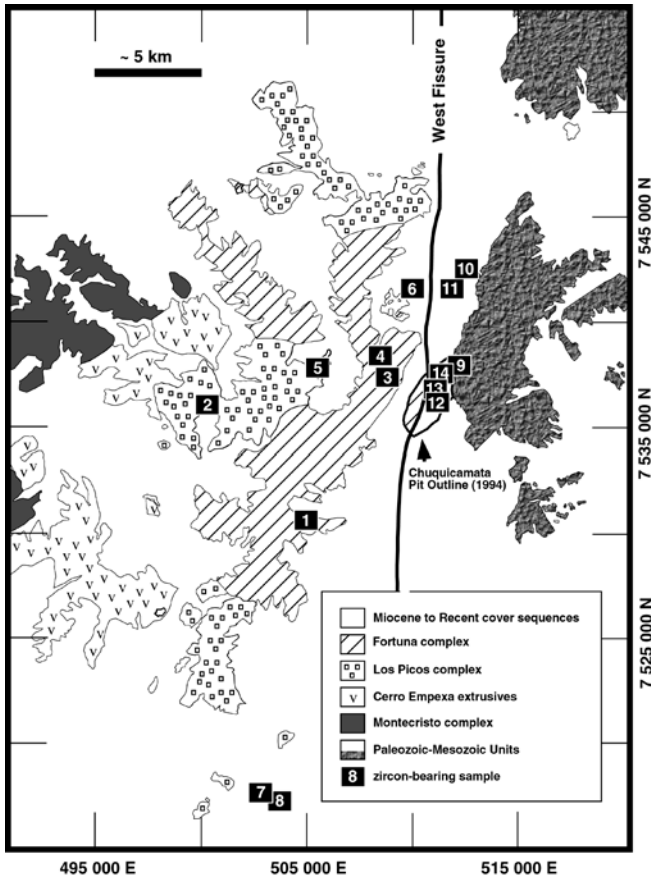


Fig. 2 Geologic map of the Chuquicamata mine and surrounding area (modified from Chong and Pardo 1994). Locations and identities of samples are as follows: 1 and 2 Los Picos monzodiorite, 3 Gris granodiorite, 4 Clara granodiorite, 5 San Lorenzo rhyodacite porphyry, 6 leucocratic intrusive, 7 and 8 Opache mine porphyry, 9 Elena granodiorite, 10 Radomiro Tomic major facies porphyry and 11 minor facies porphyry. Samples of East (12), West (13) and Bench (14) porphyries come from within the Chuquicamata mine

The northern extent of the El Abra complex is defined by a sharp contact made with older Pajonal diorite (Dilles et al. 1997). Together, these comprise the Pajonal-El Abra igneous suite. The El Abra complex consists of quartz and biotite diorites (Dark and Central diorite), Equis monzodiorite, South granodiorite, El Abra quartz monzonite porphyry, Apolo alkali biotite leucogranite, aplite, and undifferentiated intrusive breccia (Graichen et al. 1995). Diorite and porphyry host the bulk of the ore within the El Abra mine (Ambrus 1977; Maksaev et al. 1994), although the source of the copper has not been proven.

Based on previous geological mapping, structural reconstructions, and geochemical data (Graichen et al. 1995; Dilles et al. 1997; Rosas 1999), the Los Picos-Fortuna and Pajonal-El Abra igneous suites are interpreted to have formed part of a contiguous batholith prior to net sinistral displacement along the West Fault. Our recent zircon geochronology and geochemical analysis confirm this interpretation and indicate that this

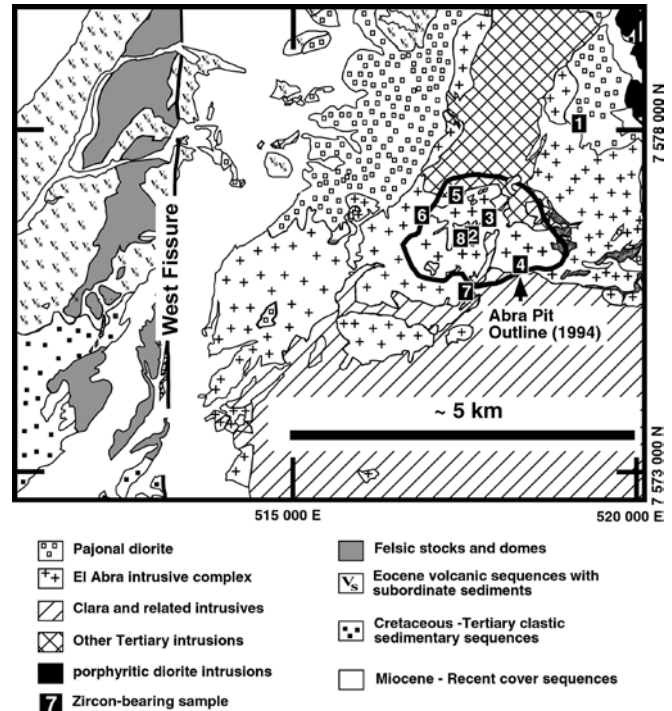


Fig. 3 Geologic map of the El Abra area (modified from Tomlinson et al. 1995, and Graichen et al. 1995). Locations of samples are as follows: 1 Pajonal diorite, 2 Apolo leucogranite, 3 Dark diorite, 4 Central diorite, 5 El Abra aplite, 6 Equis monzodiorite, 7 South granodiorite, and 8 El Abra mine porphyry

composite body was emplaced over 5.5×10^6 years between 42.9 and 37.4 Ma (Ballard et al. 2003). In addition, the zircon geochronology allows the emplacement sequence of the individual plutons of the Los Picos-Fortuna/Pajonal-El Abra complex to be established, and shows that the compositionally evolved and ore-bearing El Abra porphyry is the youngest unit. The spatial association of a significant copper deposit with the youngest porphyry suggests a genetic link between magmatic-hydrothermal mineralization and the final stages of the magmatic evolution of this large igneous complex.

Other Eocene porphyry copper deposits

Immediately east of the Fortuna intrusive complex, the super-giant Chuquicamata copper deposit (Fig. 2) lies at the southern end of the 14 by 2 km, north-south-trending Chuquicamata Igneous Complex (CIC; Zentilli et al. 1994; Tomlinson et al. 1995; Maksaev and Tomlinson 1995; Pardo and Rivera 1999). In the Chuquicamata mine, the CIC consists of three intrusions, the East, West and Bench porphyries, which can be distinguished on the basis of texture and varying proportions of plagioclase, quartz, orthoclase and biotite phenocrysts (Aracena 1981). The Radomiro Tomic copper deposit is located 7 km north of the Chuquicamata mine, also within the CIC (Cuadra et al. 1997). At Radomiro

Tomic, copper mineralization is hosted within two distinct porphyry intrusions which are texturally very similar to the Chuquicamata porphyries.

The Opache prospect is located 6 km west of the West Fissure, about 20 km south-southwest of Chuquicamata (Fig. 2). Preliminary work suggests that Opache mineralization consists of hydrothermal breccias which cross cut a series of hypabyssal dacite and tonalite porphyries which intrude andesite volcanics (Williams 1997). Basement in this part of the district is thought to consist of Paleozoic to Cretaceous metamorphic rocks, which is supported by their inclusion as clasts in some hydrothermal breccia samples (Williams 1997). A limestone sequence covers the Opache prospect, locally reaching a thickness of over 200 m (Pardo and Rivera 1999).

Indicators of magmatic oxidation state

Aerially extensive hydrothermal alteration has destroyed many of the primary mineralogical and chemical indicators of oxidation state in igneous rocks of the Chuquicamata-El Abra porphyry copper belt. Iron-titanium oxides are rarely preserved, primary mafic phases are widely replaced by chlorite, and most biotite is Mg-rich and probably of hydrothermal origin. An exception is Los Picos diorite in which exsolved hematite-ilmenite coexists with Ti-poor magnetite and Mg-rich biotite. Based on reconstructed hemo-ilmenite compositions and assuming a crystallization temperature of ca. 900 °C, the estimated oxygen fugacities for this unit lie about one log unit above FMQ (fayalite-magnetite-quartz), using the calibration of Anderson et al. (1993).

Methods

Zircon trace-element analysis

Zircons were separated by standard magnetic and density methods (Allen et al. 1998) from 2-kg samples collected from drill core and outcrop (Figs. 1, 2, 3). The separated grains were mounted in epoxy and polished to expose their centers. Using a combination of cathodoluminescence and optical microscopy, the clearest, least fractured rims of the zircon crystals were selected as suitable targets for laser ablation.

Sample mounts were placed in a custom-designed sample cell flushed with Ar and He (Loucks et al. 1995; Eggins et al. 1998). Laser ablation was accomplished using a pulsed Lambda Physik LPX 120I UV ArF excimer laser operated at a constant voltage (between 21 and 23 kV) at 5 Hz, with a spot diameter of 29 μm . The ablated material was carried by the He-Ar gas via a custom-made flow homogenizer to the ICP-MS. Raw count rates for the indicated isotopes were recorded in time-resolved mode, peak hopping in pulse counting mode. Data were acquired for 20 s with the laser off, and 40 s with the laser on, giving approximately 120 mass scans for a penetration depth of ca. 20 μm . Each block of eight unknowns was bracketed by analysis of a standard and took approximately 20 min. Data were acquired during three separate analytical sessions. The first employed a Perkin-Elmer ELAN 6000 ICP-MS, whereas the final two sessions used an Agilent 7500 ICP-MS. Four

samples were repeated to ensure that both instruments produced consistent analytical results. Fused whole-rock glasses were analyzed in an identical way to the zircons.

Raw elemental count rates were converted to concentrations (in parts per million) offline using a spreadsheet on a PC. Corrections were made for mass bias drift which was evaluated by reference to standard glass NIST 612 (Pearce et al. 1997). After triggering, it took three to four mass scans for the count rates to reach a steady state, so these initial data were excluded from the calculations. Trace-element concentrations were obtained by normalizing count rates for each analyzed element to those for Si, and assuming SiO_2 to be stoichiometric in zircon with a concentration of ca. 32.8 wt%.

The laser ablation spots in zircon are carefully selected to avoid mineral and glass inclusions, inherited cores, and cracks, and most grains displayed smooth elemental patterns. However, it is difficult to avoid all subsurface heterogeneities. Apatite is a common inclusion in zircon which can strongly perturb LREE concentrations in zircon. To monitor for apatite inclusions, phosphorus was included in the list of elements measured. Zircons with high P count rates almost always displayed correspondingly high levels of LREEs, usually within discrete depth intervals (Fig. 4), consistent with the apatite inclusion hypothesis. These data intervals were excluded from the calculations.

Whole-rock trace element analysis

Whole-rock powders were mixed with a thulium-spiked lithium metaborate flux, fused at 1,200 °C, and quenched to produce a homogeneous glass beads. Glass chips were set in an epoxy mount and the ELA-ICP-MS analyses followed a similar approach to zircon analysis. Whole-rock glasses were ablated using a spot diameter of 103 μm , background data were acquired for 30 s with the laser off, and trace-element data were acquired for 60 s with the laser on. This gave a total of approximately 180 mass scans for a penetration depth of ca. 30 μm . Whole-rock trace-element concentrations were obtained by normalizing counts for each analyzed element to the doped Tm concentration of ca. 27 ppm. These were checked against XRF-based CaO compositions in sample replicates and well-documented standards.

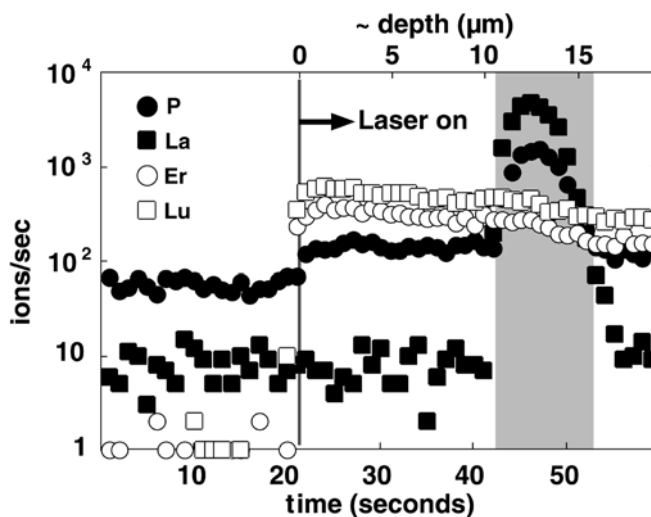


Fig. 4 Raw signal intensities (ions/s) for ^{31}P , ^{139}La , ^{166}Er and ^{175}Lu obtained from ELA-ICP-MS analysis of a zircon from the El Abra complex. The rapid rise and fall of the ^{31}P and ^{139}La count rates in the shaded interval (42–52 s, 11–16 μm depth), with no effect on the HREE counts, is interpreted to be due to an inclusion of LREE-enriched apatite. Such data intervals are excluded from the Ce(IV)/Ce(III) calculations described in the text

Theory

Zircon (ZrSiO_4) is a common accessory mineral in igneous rocks of intermediate to felsic composition. Tetravalent cations of appropriate size such as Hf, Th, and U readily substitute for Zr, as do the trivalent heavy rare-earth elements (HREEs), although the exact mechanism of charge compensation is uncertain (Hanchar et al. 2001). By contrast, the light rare-earth elements (LREEs) are strongly excluded by zircon, except for Ce which usually exhibits a chondrite-normalized concentration higher than those of the adjacent LREEs, La and Pr (Fig. 5). Such a positive Ce anomaly is unique among igneous minerals and indicates that Ce(IV) is partitioned into zircon in strong preference to Ce(III) because of its identical charge and similar size in eight-fold coordination (0.97 Å) to Zr ($r = 0.84$ Å). By similar reasoning, Eu(III) should be incorporated preferentially into zircon relative to Eu(II), although in this case the redox signal may be complicated by the effects of plagioclase crystallization.

In order to use the Ce anomaly in zircon as a measure of oxidation state, the Ce(IV)/Ce(III) ratio must be determined. Spectroscopic methods are inappropriate because Ce is present at trace concentration levels in zircon (e.g., ppm levels) and valence states are unlikely to be retained subsequent to crystallization. Instead, we employ an indirect approach described by Philpotts (1970) for determining the Eu(III)/Eu(II) ratio in plagioclase based on chemical analysis of two equilibrated phases modified for the case of Ce(IV) and Ce(III) in zircon. The total concentrations of Ce in zircon and coexisting melt can be considered as sums of Ce(IV) and Ce(III) in each phase:

$$\text{Ce}_{\text{zircon}} = \text{Ce(III)}_{\text{zircon}} + \text{Ce(IV)}_{\text{zircon}} \quad (1)$$

$$\text{Ce}_{\text{melt}} = \text{Ce(III)}_{\text{melt}} + \text{Ce(IV)}_{\text{melt}} \quad (2)$$

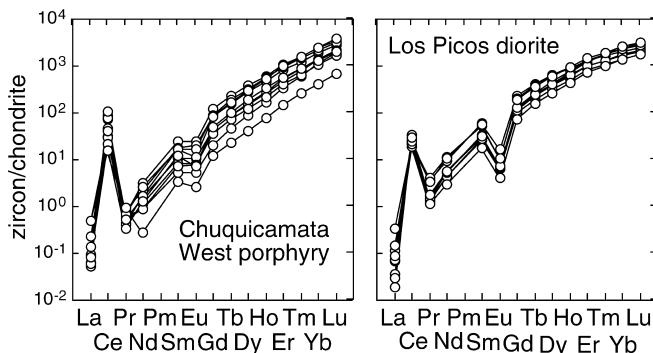


Fig. 5 Chondrite-normalized REE patterns for zircon from the Chuquicamata West porphyry (sample 98-609) and the Los Picos diorite (sample 98-521). Note the larger (more positive) Ce anomaly and smaller (less negative) Eu anomaly displayed by zircons of the West porphyry compared to those of the Los Picos diorite

By introducing separate distribution coefficients for Ce(IV) and Ce(III), these two equations can be combined to give:

$$\text{Ce}_{\text{melt}} = \frac{(\text{Ce}_{\text{zircon}} - \text{Ce(IV)}_{\text{zircon}})}{D_{\text{Ce(III)}}^{\text{zircon/melt}}} + \frac{\text{Ce(IV)}_{\text{zircon}}}{D_{\text{Ce(IV)}}^{\text{zircon/melt}}} \quad (3)$$

which upon rearrangement yields:

$$\text{Ce(IV)}_{\text{zircon}} = \frac{\text{Ce}_{\text{melt}} - \frac{1}{D_{\text{Ce(III)}}^{\text{zircon/rock}}} \text{Ce}_{\text{zircon}}}{\frac{1}{D_{\text{Ce(IV)}}^{\text{zircon/rock}}} - \frac{1}{D_{\text{Ce(III)}}^{\text{zircon/rock}}}} \quad (4)$$

Substituting Eq. (4) into Eq. (1) and solving for the concentration of Ce(III) in zircon gives:

$$\text{Ce(III)}_{\text{zircon}} = \text{Ce}_{\text{zircon}} - \frac{\text{Ce}_{\text{melt}} - \frac{1}{D_{\text{Ce(III)}}^{\text{zircon/rock}}} \text{Ce}_{\text{zircon}}}{\frac{1}{D_{\text{Ce(IV)}}^{\text{zircon/rock}}} - \frac{1}{D_{\text{Ce(III)}}^{\text{zircon/rock}}}} \quad (5)$$

Equations (4) and (5) can then be combined, rearranged and simplified to produce the following expression for the Ce(IV)/Ce(III) ratio in zircon:

$$(\text{Ce(IV)/Ce(III)})_{\text{zircon}} = \frac{\text{Ce}_{\text{melt}} - \frac{\text{Ce}_{\text{zircon}}}{D_{\text{Ce(III)}}^{\text{zircon/melt}}}}{\frac{\text{Ce}_{\text{zircon}}}{D_{\text{Ce(IV)}}^{\text{zircon/melt}}} - \text{Ce}_{\text{melt}}} \quad (6)$$

The analogous expression for the Eu(III)/Eu(II) ratio in zircon is:

$$(\text{Eu(III)/Eu(II)})_{\text{zircon}} = \frac{\text{Eu}_{\text{melt}} - \frac{\text{Eu}_{\text{zircon}}}{D_{\text{Eu(II)}}^{\text{zircon/melt}}}}{\frac{\text{Eu}_{\text{zircon}}}{D_{\text{Eu(III)}}^{\text{zircon/melt}}} - \text{Eu}_{\text{melt}}} \quad (7)$$

To solve Eq. (6) for the Ce(IV)/Ce(III) ratio in zircon, four values are needed: concentrations of Ce in zircon and melt, and zircon-melt distribution coefficients for Ce(IV) and Ce(III). The first value is measured directly by in-situ chemical analysis of zircon using ELA-ICP-MS, the second value is assumed to be equivalent to the whole-rock (as a proxy for melt) concentration of Ce also measured by ELA-ICP-MS, and the partition coefficients for Ce(IV) and Ce(III) can be estimated on the basis of crystal chemical constraints on trace-element partitioning.

It has been known for sometime that when mineral-melt distribution coefficients for isovalent cations are plotted as a function of ionic radius, they form smooth, concave-down parabolic curves (Onuma et al. 1968; Fig. 6). The apex of each parabola in these so-called Onuma diagrams corresponds to the optimal size of the crystal site into which the cations substitute, and its curvature reflects the elasticity of the surrounding lattice. Blundy and Wood (1994) quantified this relation using a thermodynamically based model relating the mineral-melt distribution coefficient (D_i) for a cation to the lattice-strain energy (ΔG) induced by its misfit into a given crystal site:

$$D_i = D_0 e^{-\Delta G/RT} \quad (8)$$

where D_0 is the hypothetical strain-free distribution coefficient for a cation of optimal radius, R is the gas

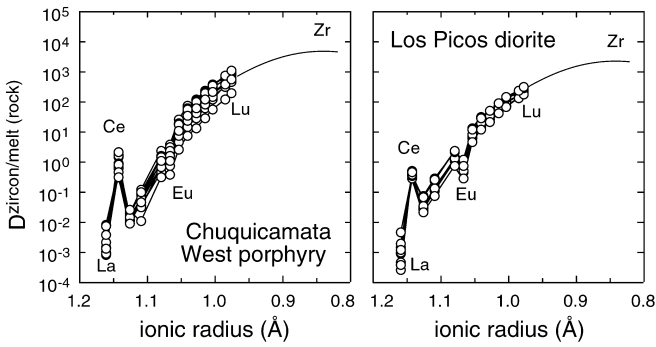


Fig. 6 Calculated distribution coefficients for REEs between individual zircons and average rock (as a melt proxy) for the Chuquicamata West porphyry and Los Picos diorite plotted versus ionic radii of the REEs as trivalent cations in eight-fold coordination. The curved trends form steep limbs of parabolic curves centered on the ionic radius of Zr (0.84 Å) which are characteristic of crystal chemically controlled trace-element partitioning (Onuma et al. 1968)

constant, and T is absolute temperature. The lattice-strain energy depends on the difference between the ionic radius of the cation (r_i) and the optimal site radius (r_0) through the Young's modulus (E) of the host crystal:

$$\Delta G = 4\pi EN_A \left[\frac{r_0}{2} (r_i - r_0)^2 + \frac{1}{3} (r_i - r_0)^3 \right] \quad (9)$$

where N_A = Avogadro's number. Although the lattice-strain model may not explain all partitioning behavior (e.g., Ghaderi et al. 1999; Klein et al. 2000), it captures

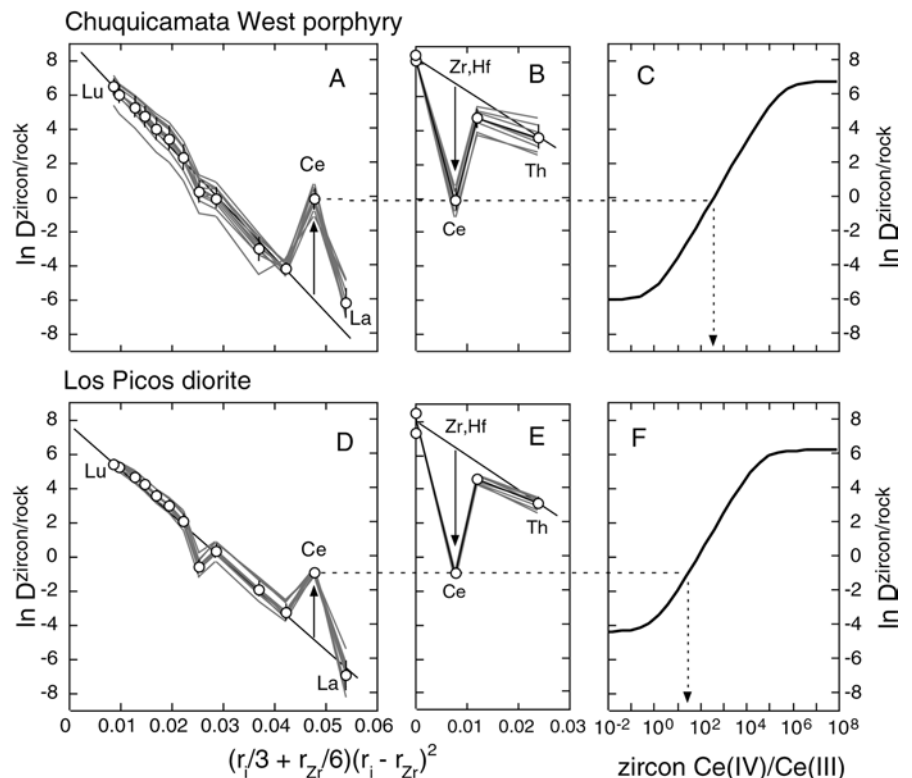
the first-order features, especially for structurally and compositionally simple minerals such as zircon.

Combining Eqs. (8) and (9), rearranging and taking logarithms yields a linear relation between mineral-melt distribution coefficients for a series of isoivalent elements and a lattice-strain term involving their ionic radii:

$$\ln D_i = \ln D_0 - \frac{4\pi EN_A}{RT} \left(\frac{r_i}{3} + \frac{r_0}{6} \right) (r_i - r_0)^2 \quad (10)$$

When plotted according to Eq. (10) and assuming $r_0 = 0.84$ Å for Zr in eight-fold coordination, zircon-melt (or rock as a proxy for melt) distribution coefficients for tetravalent and trivalent cations form linear arrays (Fig. 7). These straight lines correspond to parabolas in an Onuma diagram. Because Ce and Eu are present as mixtures of Ce(IV) and Ce(III) and Eu(III) and Eu(II) in terrestrial magmas, the distribution coefficients for bulk Ce and Eu will fall somewhere between the tetravalent and trivalent and trivalent and divalent trends, respectively (Fig. 7). If all Ce were present in zircon as Ce(III), it would plot on the line defined by the trivalent REEs. Conversely, if all the Ce was Ce(IV), it would fall on the tetravalent line along with Zr, Hf, U(IV) and Th. By fitting a line to the trivalent REE array, a value for $D_{\text{Ce(III)}}^{\text{zircon/melt}}$ can be estimated. A similar procedure can be applied to the tetravalent (Zr–Hf–U–Th) trend to estimate $D_{\text{Ce(IV)}}^{\text{zircon/melt}}$. The resulting distribution coefficients for Ce(IV) and Ce(III) can be substituted into Eq. (6) and used in the calculation of the Ce(IV)/Ce(III) ratio in zircon. Because of the extremely slow diffusivities of trivalent and

Fig. 7A–F Natural logarithm of zircon/rock distribution coefficients for the Chuquicamata West porphyry and Los Picos diorite plotted versus a lattice-strain parameter (modified from Blundy and Wood 1994) for trivalent (A, D) and tetravalent cations (B, E). As discussed in the text, linear arrays are predicted using these coordinates. Results from individual zircon grains are shown in *shading* and sample averages are shown in *bold* (with 1 standard deviation errors). Measured zircon/rock distribution coefficients for total Ce lie between those expected for Ce(III) and Ce(IV) based on the observed trends for trivalent and tetravalent cations, and can be used to calculate the Ce(IV)/Ce(III) ratio in zircon as illustrated (C, F). Note the high La value in A which is indicative of contamination by an LREE-enriched inclusion



tetravalent cations in zircon (Cherniak et al. 1997), such elemental characteristics should survive as long as the zircon crystal remains intact and thereby record the Ce(IV)/Ce(III) at the time of crystallization.

It is important to point out that zircon Ce(IV)/Ce(III) ratios greater than one do not imply the same for the parental melt. Zircon is the only common igneous mineral to exhibit a Ce anomaly and it does so as much because of an extreme discrimination against Ce(III) as a preference for Ce(IV). Hinton and Upton (1991) note that a positive Ce anomaly in zircon is consistent with crystallization from a melt with a Ce(IV)/Ce(III) ratio on the order of 10^{-3} . Melt Ce(IV)/Ce(III) ratios calculated using the zircon Ce(IV)/Ce(III) ratio and $D_{\text{Ce(III)}}^{\text{zircon/melt}}$ and $D_{\text{Ce(IV)}}^{\text{zircon/melt}}$ values determined for each sample of this study range from 0.5×10^{-3} to 1.2×10^{-3} . Mass-balance considerations indicate that fractional crystallization or accumulation of zircon are incapable of generating measurable Ce anomalies in magmas. Conversely, fractionation of LREE-enriched phases such as xenotime and monazite, although depleting residual melt in the LREEs, will not affect the calculation of Ce(IV)/Ce(III) ratios as described above because it is based on zircon-melt partitioning of REEs.

Results

Zircon trace-element data are presented in Tables 1 and 2 together with average whole-rock MgO, SiO₂, and trace-element data from Ballard et al. (unpublished data) and zircon U–Pb ages (Ballard et al. 2003). Calculated zircon Ce(IV)/Ce(III) and Eu_N/Eu_N* ratios are also listed in Table 2 (where the subscript indicates chondrite normalization, and Eu_N* = (Sm_NGd_N)^{1/2}). In cases where zircon REE concentrations were close to the detection limit of the ICP-MS, no Ce(IV)/Ce(III) ratio has been calculated because of the large uncertainties involved. Individual grain and sample average zircon Ce(IV)/Ce(III) and Eu_N/Eu_N* ratios are plotted versus host intrusion zircon U–Pb age in Fig. 8. We do not have an adequate explanation for the large range of ratios observed in zircons from individual samples. In several cases, outliers may represent inherited grains or portions of grains. It was not possible to simultaneously determine U–Pb ages and measure a full suite of trace elements in the same spot, so we cannot prove this was always the case. Nonetheless, older units have zircons with lower Ce(IV)/Ce(III) and Eu_N/Eu_N* values than their younger counterparts, and these redox-sensitive ratios increase with decreasing emplacement age, with the notable exception of the leucocratic unit of the Fortuna complex. These data also show that the intrusions directly associated with porphyry copper deposits are characterized by having the highest zircon Ce(IV)/Ce(III) (>300) and Eu_N/Eu_N* ratios (>0.4).

Discussion

Oxidation during magmatic differentiation

The major- and trace-element compositions of the Los Picos-Fortuna/Pajonal-El Abra igneous complex vary systematically with time (Ballard et al., unpublished data). This point is illustrated in Fig. 9, which shows that average MgO values of individual units within the complex decrease with age, although there is considerable range within units and overlap between units. Our preferred explanation for these data is that all intrusions were sourced from a single, large magma reservoir located deeper in the crust which was episodically replenished by mafic magma while undergoing progressive fractional crystallization and assimilation. We suggest that this deep, parental magma reservoir remained at least partially molten over the 5.5×10^6 year history of the complex. In this way, resident evolved melt could mix with injections of new mafic magma to form hybrid products which retained a chemical imprint of prior differentiation (Ballard et al., unpublished data). Importantly, the increase in the average Ce(IV)/Ce(III) ratios of zircon with time (Fig. 10) suggests that oxidation state also evolved progressively in the magma system. The Pajonal diorite represents the oldest, least oxidized stage of this evolution, and the ore-bearing El Abra mine porphyry is the youngest, most oxidized stage.

The other ore-bearing porphyries analyzed in this study, from Chuquicamata, Radomiro Tomic and Opache, do not have obvious precursor-barren intrusive sequences as does the Los Picos-Fortuna/Pajonal-El Abra complex. Nevertheless, the high Ce(IV)/Ce(III) ratios of the zircons from the ore-bearing units and their geochemical resemblance to the El Abra mine porphyry suggest that these intrusions may have experienced a similar magmatic evolution. It is therefore likely that the Chuquicamata, Radomiro Tomic and Opache porphyries are underlain by long-lived magma systems similar to that hypothesized for the Los Picos-Fortuna/Pajonal-El Abra complex.

The progressive increase in average zircon Ce(IV)/Ce(III) ratios in the Los Picos-Fortuna/Pajonal-El Abra complex is accompanied by a similar trend in average zircon Eu_N/Eu_N* toward smaller (less “negative”) Eu anomalies (Fig. 8D). These data are consistent with an increase in magmatic oxidation state and the Eu(III)/Eu(II) ratio over time if, as is expected on the basis of crystal chemical arguments given above, Eu(III) is partitioned preferentially into zircon relative to Eu(II). Such a hypothesis is supported by trends in whole-rock Eu_N/Eu_N* from the complex. Because Eu(II) is compatible in plagioclase, with $D_{\text{Eu}}^{\text{albite/melt}} > D_{\text{Eu}}^{\text{anorthite/melt}} > 1$, fractional crystallization usually produces melts with pronounced, negative Eu anomalies, that is, Eu_N/Eu_N* < 1 (Nash and Crecraft 1985; Bacon and Druitt 1988; Dunn and Sen 1994). However, Eu becomes

Table 1 (Contd.)

Unit	Analysis ^a	Age (Ma)	±SE	SiO ₂	±SD	MgO	±SD	P	Sr	Y	La	Ce	Pr	Nd	Sm	Eu	Gd
	165-5							89	0.14	310	0	9	0.022	0.43	1.21	0.23	5.8
	165-6							129	0.25	630	0.002	20	0.059	1.07	2.1	0.48	12.5
	165-7							102	0.13	296	0.007	11	0.013	0.42	1.51	0.41	5.1
	165-8							64	0.21	267	b.d.l.	11	0.03	0.51	0.76	0.41	4.3
	165-9							113	0.22	507	0.004	16	0.037	1.04	2.49	0.71	11.8
	Av. rock	38	0.3	64.8	3.8	2.16	1.39	811	505	8.87	20.8	40.3	4.33	16.1	2.71	0.768	2.04
El Abra	176-10							60	0.13	186	b.d.l.	10	-0.008	0.16	0.52	0.17	2.7
mine	176-2							61	0.14	287	0.008	10	0.021	0.52	0.89	0.29	4.5
por-	176-3							56	0.19	230	0.001	11	0.023	0.19	0.78	0.24	3.9
phyry	176-4							124	1.88	749	0.021	16	0.039	0.74	1.67	0.49	10.4
	176-5							65	0.19	319	0.004	13	0.003	0.17	0.96	0.25	4.9
	176-6							124	0.34	791	0.018	28	0.068	1.18	1.68	0.66	11.5
	176-7							149	0.17	290	0.142	12	0.031	0.62	0.67	0.19	5.4
	176-8							247	1.11	1,956	1.746	54	0.336	1.74	3.21	0.69	26.1
	176-9							51	0.3	365	0.056	11	0.081	0.63	1.65	0.54	7.1
	Av. rock	37.4	0.3	66.64	3.14	1.41	0.79	637	545	9	19.8	39.6	4.35	16.3	2.7	0.761	2.1
Los Picos-Fortuna suite																	
Los	521-02							338	0.32	1,515	0.006	28	0.204	3.54	6.11	0.48	32.6
Picos	521-03							290	0.23	1,226	0.011	21	0.172	2.68	5.34	0.48	27.1
diorite	521-04							318	0.31	2,033	0.037	19	0.425	5.95	11.9	1.3	60.2
	521-05							351	0.37	1,794	0.009	25	0.156	2.76	5.44	0.51	30.6
	521-06							388	0.39	1,849	0.021	25	0.183	3.31	6.58	0.57	33.9
	521-07							259	0.33	1,498	0.024	15	0.211	3.49	6.43	0.54	35.7
	521-08							230	0.58	1,068	0.106	18	0.136	1.8	3.57	0.31	19.6
	521-09							237	0.42	2,249	0.047	20	0.485	6.88	10.76	0.83	54
	521-10							226	0.46	2,275	0.029	25	0.413	6.68	11.47	0.79	50.3
	Av. rock	42.3	0.2	62.11	3.43	2.5	0.73	784	527	20.34	26.4	56.2	6.56	25.6	4.98	1.014	4.22
Fortuna	98-576.1								0.54		0.014	16	0.069	3.47	3.24	0.41	13.3
Gris	98-576.2								0.49		0.333	9	0.296	2.52	1.02	0.35	7.3
	98-576.3								0.93		0.044	20	0.51	6.45	8.75	1.69	38.8
	98-576.4								0.22		0.019	14	0.37	4.43	5.6	0.77	15.8
	98-576.5								0.76		0.182	13	0.307	2.74	3.14	0.34	10.2
	98-576.6								0.88		0.009	13	0.053	2.63	1.81	0.33	10.4
	98-576.7								0.45		0.703	11	0.221	3.52	2.69	0.66	14.1
	98-576.8								0.17		0.018	9	0.428	4.21	2.35	0.39	12.7
	98-576.9								0.34		0.327	20	0.431	8.06	9.66	1.98	44.2
	98-576.10								0.56		0.004	12	0.219	2.74	2	0.43	6.2
	Av. rock	39.1	0.4	65.28	3.12	1.77	0.62	765	488	16.78	25.9	52.8	6.03	23.1	4.26	0.878	3.41
Fortuna	98-570.1								0.45		0.139	11	0.141	1.7	1.01	0.5	4.5
Clara	98-570.2								0.43		0.152	11	0.46	3.24	3.52	1.29	16.1
	98-570.3								0.51		0.106	15	0.534	2.67	2.92	0.44	9.1
	98-570.4								1.3		0.98	16	0.496	3.32	1.86	0.27	8
	98-570.5								1.55		0.139	11	0.042	2.46	2.33	0.5	8.1
	98-570.6								0.2		0.025	13	0.619	5.64	6.59	1.8	23.6
	98-570.7								0.72		0.011	14	0.142	2.58	1.9	0.61	7.2
	98-570.8								0.31		0.336	20	0.242	3.59	2.32	0.36	8.7
	98-570.9								0.87		0.01	14	0.301	2.94	3.61	0.83	12.2
	98-570.10								0.72		0.156	13	0.501	2.74	1.78	0.24	6
	Av. rock	38.5	0.4	66.49	1.13	1.38	0.2	730	672	9.72	20.6	41.2	4.75	18	3.22	0.811	2.36
Fortuna	98-565.1								0.63		0.033	14	0.517	5.42	6.96	1.38	27.9
leuco-	98-565.2								0.4		0.349	10	0.291	2.07	1.54	0.48	8.1
ganite	98-565.3								1.1		0.365	15	0.301	3.35	3.51	0.61	8.9
	98-565.4								1.83		0.029	12	0.455	2.53	1.35	0.59	8
	98-565.5								1.14		0.004	12	0.174	1.9	2.15	0.77	11.8
	98-565.6								0.99		0.332	15	0.283	2.83	1.62	0.32	9.7
	98-565.7								1.16		0.005	11	0.195	2.38	2.89	0.59	9
	98-565.8								0.51		0.362	12	0.405	4.37	6.72	1.06	26.1
	98-565.9								1.22		0.172	11	0.307	4.18	3.37	0.82	18
	98-565.10								0.49		0.177	13	0.323	5.16	6.22	1.33	26
	Av. rock	38.1	0.4	75.83	3.09	0.17	0.15	95	239	15.27	31.1	66.2	7.51	26.8	4.58	0.572	3.22
San	98-530.1								6.04		8.87	37	2.9	14.17	5.54	0.72	13.9
Lore-	98-530.2								0.9		0.291	16	0.481	3.42	1.34	0.32	7.3
nzo	98-530.3								0.2		0.022	17	0.072	3.47	2.62	0.81	17.2
por-	98-530.4								0.61		0.005	13	0.275	2.61	1.49	0.27	7.4
phyry	98-530.5								0.6		0.004	13	0.273	2.76	1.42	0.28	7.6
	98-530.6								0.59		0.004	26	0.315	3.85	2.56	0.36	13.6
	98-530.7								0.87		0.077	9	0.022	2.32	1.37	0.26	4.9
	98-530.8								1.21		0.143	8	0.153	1.45	0.89	0.54	5.9

Table 1 (Contd.)

Unit	Analysis ^a	Age (Ma)	±SE	SiO ₂	±SD	MgO	±SD	P	Sr	Y	La	Ce	Pr	Nd	Sm	Eu	Gd	
	626-9								177	0.22	564	0.056	8	0.075	0.99	2.04	0.34	12.4
	Av. rock	63.2	0.5	63.52	3.49	2.23	0.62	1,039	362	27.8	31.4	67.3	7.8	30.2	5.97	1.02	5.33	
Opache mine por- phyry	067-1								58	0.28	483	0.001	18	0.015	0.38	0.96	0.31	5.8
	067-10								116	0.26	995	0.001	15	0.093	2.27	5.17	1.15	26
	067-2								61	0.13	275	0.004	15	b.d.l.	0.23	0.85	0.19	4.3
	067-3								99	0.32	757	0.014	29	0.054	1.33	2.29	0.97	13.3
	067-4								91	0.17	296	0.024	7	0.053	0.63	1.16	0.14	5.6
	067-5								84	0.17	312	0.009	14	0.008	0.47	1	0.32	5.3
	067-6								100	0.19	333	0.005	13	0.025	0.67	1.45	0.39	7.2
	067-7								67	0.17	290	b.d.l.	12	0.017	0.2	0.75	0.32	4.5
	067-8								123	0.27	745	0.005	13	0.114	1.87	3.66	0.93	16.8
	067-9								70	0.22	326	0.006	13	0.028	0.48	0.99	0.39	5.7
698-1								84	0.15	252	0.005	11	0.014	0.43	0.97	0.3	4.6	
698-10								114	0.22	609	0.006	22	0.073	1.33	2.58	0.66	12.4	
698-2								93	0.18	446	0.007	12	0.014	0.6	1.35	0.46	8.4	
698-3								99	0.19	407	0.009	12	0.039	0.5	1.49	0.33	7.8	
698-4								79	0.17	274	b.d.l.	11	0.027	0.38	0.87	0.32	5	
698-5								81	0.23	562	0.009	28	0.05	0.95	1.73	0.49	8.9	
698-6								82	0.18	374	0.002	10	0.068	1.33	2.06	0.68	9	
698-7								54	0.16	259	0.005	9	0.041	0.51	0.66	0.3	4.6	
698-8								64	0.25	558	0.004	12	0.104	1.92	3.5	1.16	14.1	
698-9								75	0.16	252	0.004	12	0.023	0.34	0.96	0.37	4.6	
	Av. rock	37.9	0.2						543	7.33	21	41.6	4.51	17.1	2.81	0.809	1.99	
Elena grano- diorite	679-1								77	0.27	487	0.029	18	0.037	0.64	1.21	0.66	8.5
	679-4								49	0.41	233	0.137	11	0.054	0.43	0.39	0.21	2.9
	679-5								81	0.33	746	0.006	33	0.041	1.05	2.33	0.81	13.3
	679-6								52	0.14	172	0.007	9	0.013	0.12	0.5	0.14	2.9
	679-7								81	0.22	461	0.098	12	0.097	1.15	2.23	0.64	9.9
	679-8								58	0.28	513	0	25	0.027	0.42	1.23	0.46	7.1
	679-9								80	0.21	328	0.029	13	0.022	0.44	0.88	0.34	5.3
		Av. rock	37.7	1						532	6.37	19.9	36.1	3.7	13.3	2.15	0.561	1.5

^aSiO₂ and MgO are given in wt%, all other concentrations are reported in ppm. SE, Standard error; SD, standard deviation; b.d.l., below detection limit

increasingly incompatible in plagioclase at higher oxidation states, reflecting a larger proportion of Eu(III) in the magma (Drake 1975; Wilke and Behrens 1999). Thus, if plagioclase fractionation is accompanied by progressive oxidation, the expected trend of decreasing Eu_N/Eu_N^* during differentiation can be retarded, which is what is observed in the the Los Picos-Fortuna/Pajonal-El Abra complex (Ballard et al., unpublished data).

Although both the Ce(IV)/Ce(III) and Eu_N/Eu_N^* ratios in zircon can be used as a measure of magmatic oxidation state, we prefer Ce(IV)/Ce(III) over Eu_N/Eu_N^* for two reasons. First, the Ce(IV)/Ce(III) ratio is more sensitive, varying by almost two orders of magnitude in the zircons of this study compared to a factor of five range in Eu_N/Eu_N^* (Fig. 8). Second, the interpretation of Eu_N/Eu_N^* in zircon is complicated by the competing influences of plagioclase crystallization and assimilation of plagioclase-bearing rocks, which can change the Eu_N/Eu_N^* of the melt.

Mechanism of magma oxidation

The zircon REE data suggest that the Los Picos-Fortuna/Pajonal-El Abra igneous complex became progressively more oxidized over time. Previous workers have attributed positive shifts in magmatic oxidation state to

(1) metasomatism of a mantle source region prior to or during partial melting, induced by hydrous fluids or silicate melts derived from subducted oceanic crust and enriched in oxidized components such as Fe(III), CO₂ or sulfate (McInnes and Cameron 1994); (2) fractional crystallization of Fe(II)-bearing silicate, oxide or sulfide phases in a closed-system magma chamber (Carmichael and Ghiorso 1986); (3) influx of SO₂ into an evolved felsic magma chamber from degassing mafic magma ponded below it (Matthews et al. 1992; Pallister et al. 1992); and (4) degassing of hydrous magma at low pressures with preferential removal of reduced volatile species such as H₂ or H₂S (Czamanski and Wones 1973; Candela 1986; Ague and Brimhall 1988). Assimilation of oxidized country rock could be added to this list. As noted above, we interpret the Los Picos-Fortuna/Pajonal-El Abra igneous complex to have been sourced from an episodically replenished, long-lived magma reservoir at depth. If this interpretation is correct, then the systematic increase in zircon Ce(IV)/Ce(III) ratios with time (Fig. 10) means that progressive oxidation within the parental magma reservoir was not interrupted or reversed by injection of new magma or extraction of resident magma. This argues that oxidation did not occur in association with high-level emplacement of individual intrusions, but by processes within or below the hypothesized deep magma reservoir.

Table 2 (Contd.)

Unit	Analysis ^a	Tb	Dy	Ho	Er	Tm	Yb	Lu	Th	U	Zr	Hf	Eu/ Eu*	±SD	Ce(IV)/ Ce(III)	±SD	
Rado- miro	695-1	4.11	48.4	18.7	95.8	25.3	285	59.2	130	268		12,587	0.47		313		
	695-10	2.69	34.7	15.3	83.6	23.4	282	65.4	119	268		11,693	0.49		1,571		
	Tomic major facies	695-2	4.54	55.5	24.1	131.5	36.3	445	103.8	276	595		12,857	0.63		1,684	
		695-3	4	47	18.2	90	22.7	248	53	438	487		10,365	0.48		485	
	por- phyry	695-4	5.33	64.1	23.8	119.1	29.4	325	66.5	342	392		11,790	0.53		262	
		695-5	2.05	27.3	12.3	65.2	17.9	216	49	90	228		9,998	0.46		1,249	
		695-6	3.44	43.7	19.7	110.6	30.9	387	85.7	213	452		12,702	0.57		1,820	
		695-7	2.38	28.9	12.1	64.5	17.2	210	47.5	171	338		12,525	0.57		1,457	
		695-8	4.01	50.8	21.1	107.9	28.8	334	70.6	158	298		11,680	0.42		795	
		695-9	10.52	126.2	51	259.8	67.8	772	162.7	2,628	2,194		12,072	0.42		560	
Av. zircon													0.5	0.07	1,020	601	
Av. rock		0.767	0.137		0.401		0.484	0.076	5.23	2.86	89.1	2.49					
Others																	
Monte- cristo	626-1	11.33	114	37.5	154.2	32.1	277	47.1	147	137		8,613	0.21		6		
	626-10	6.47	71.3	24.4	108.1	22.8	210	35.5	92	103		9,348	0.2		31		
	Intru- sive	626-2	6.04	67.1	22.6	100.5	21.6	196	34	87	100		8,562	0.21		25	
		626-3	11.66	117.9	38.3	157.3	32.4	280	47.2	150	134		8,701	0.17		6	
	626-4	7.21	76.2	25.6	108.6	23.4	207	35.7	93	102		8,992	0.21		12		
	626-5	3.54	40.9	14.6	65.3	14.9	137	24.8	66	92		8,803	0.2		64		
	626-6	6.29	69.5	23.4	102.9	21.9	199	34.7	90	100		8,854	0.19		23		
	626-7	6.2	69.2	22.9	99.7	21.2	191	33	86	97		8,890	0.18		15		
	626-8	2.96	34.2	12.5	58.2	12.9	123	21.9	55	82		9,032	0.19		71		
	626-9	4.5	51.7	17.9	81.5	17.9	166	28.8	73	90		8,773	0.21		56		
Av. zircon												0.2	0.01	31	24		
Av. rock	0.8	5.05	1.01		2.86		3.01	0.44	24	6.98	284	8.49					
Opache mine	067-1	2.41	31.2	13	76.2	22.1	266	60.9	315	677		11,562	0.4		391		
	067-10	9.11	95.6	33.2	143.2	31.6	292	51.2	108	119		9,687	0.3		201		
	por- phyry	067-2	1.83	20.4	7.9	42	11.6	126	25.1	67	134		12,763	0.3		342	
		067-3	5.1	64.2	25.4	133.7	33.2	386	75.4	222	321		12,647	0.54		222	
	067-4	2.06	24.2	9.1	43.5	10.4	107	21.9	84	436		11,317	0.17		551		
	067-5	1.99	23.9	9.4	48.8	12.5	140	29.7	128	207		10,673	0.42		728		
	067-6	2.73	30	10.7	51.3	12.2	120	23.6	44	61		8,236	0.36		109		
	067-7	1.85	21.6	8.5	45.1	11.8	132	27.5	72	131		10,251	0.53		752		
	067-8	6.21	67.5	24.5	113.9	26.5	264	48.6	93	112		9,966	0.36		57		
	067-9	2.07	23.8	9.7	50.9	13.8	162	35.3	169	406		10,454	0.49		411		
698-1	1.69	19.5	7.6	38.1	10	114	24	73	112		9,591	0.43		1,698			
698-10	4.37	50.1	18.9	90	21.9	219	42	159	184		9,943	0.36		40			
698-2	3.1	36.9	13.9	67.7	17	179	35.9	105	150		10,206	0.42		713			
698-3	2.91	33.3	12.5	59.9	14.8	151	29.8	69	111		9,544	0.29		591			
698-4	1.82	21.2	8.4	42	11.2	125	26.2	68	116		10,077	0.46		155			
698-5	3.47	42.2	16.9	85.2	23.7	261	55	469	501		10,480	0.38		593			
698-6	2.81	32	11.7	56.9	14.2	149	31.5	75	104		9,115	0.48		203			
698-7	1.61	20.7	8	40.5	11.2	121	24.9	50	95		11,295	0.52		858			
698-8	4.24	48.2	16.5	78	19.5	200	39.7	106	135		10,626	0.5		68			
698-9	1.69	20.7	7.9	38.8	9.9	109	22.8	106	132		9,539	0.54		695			
Av. zircon												0.41	0.1	469	393		
Av. rock		1.416	0.257		0.715		0.751	0.112	6	2.09	98.6	2.74					
Elena grano- diorite	679-1	3.07	37.2	15	79.4	20.1	224	48.4	101	195		1,1835	0.63		548		
	679-4	1.12	12.9	5.6	31.6	9.2	118	31.3	72	183		9,777	0.61		2,205		
	679-5	5.09	61.1	23.4	118.4	29.5	310	61.7	165	241		11,433	0.44		349		
	679-6	1.07	13.4	5.2	28.7	7.4	87	18.2	51	111		11,791	0.35		642		
	679-7	3.37	38.8	14.1	70.8	17.7	194	38.6	96	129		10,116	0.42		91		
	679-8	2.88	36.9	15.1	81.5	21.1	247	56.4	122	229		11,663	0.47		867		
	679-9	2.13	25.1	10	51.8	13.1	147	29.7	73	131		10,713	0.49		488		
Av. zircon												0.49	0.1	741	689		
Av. rock		1.171	0.22		0.638		0.724	0.112	5.1	1.49	107.6	3.03					

^aAll concentrations are reported in ppm. SD, Standard deviation

Sulfur speciation and its control on chalcophile element behavior in magmas

Although there is a well-known empirical association between Cu±Au mineralization and oxidized calc-alkaline intrusions, the cause of this relationship is

uncertain. Oxidation state controls sulfur speciation and hence strongly influences sulfur solubility in magmas. At low oxygen fugacities, magmatic sulfur exists principally as sulfide (S²⁻) which has a relatively low solubility in silicate melts (Carroll and Rutherford 1985, 1988). Crystallization under such conditions can lead to early

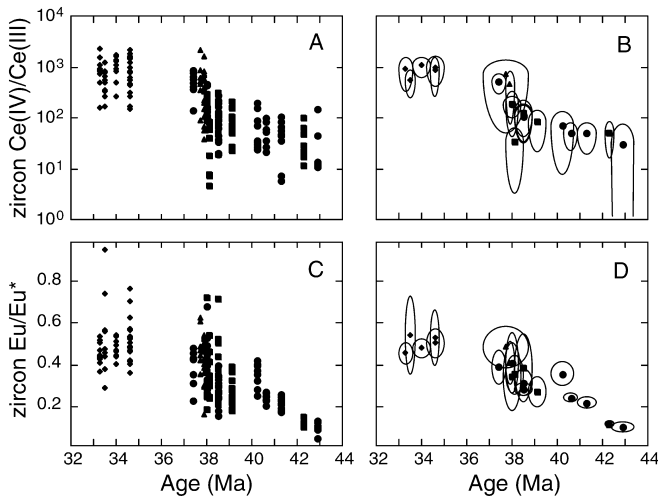


Fig. 8A–D Individual grain (A, C) and average (B, D) zircon Ce(IV)/Ce(III) and $\text{Eu}_N/\text{Eu}_N^*$ ratios plotted versus zircon U–Pb age for intrusions of the Chuquicamata–El Abra porphyry copper belt (Ballard et al. 2001, 2003). The Pajonal–El Abra igneous suite is indicated by *circles*, the Los Picos–Fortuna suite by *squares*, the Chuquicamata Intrusive Complex (including the East, West, and Bench porphyries in the Chuquicamata mine and the major and minor facies porphyries of the Radomiro Tomic mine) by *diamonds*, and other units by *triangles*. Uncertainty envelopes are 2 standard errors for ages, and 1 standard deviation for Ce(IV)/Ce(III) and $\text{Eu}_N/\text{Eu}_N^*$ ratios. Values of zircon Ce(IV)/Ce(III) = 300 and $\text{Eu}_N/\text{Eu}_N^* = 0.4$ effectively distinguish the ore-bearing and barren units examined in this study

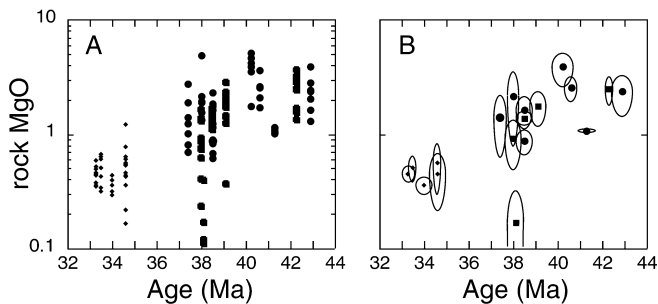


Fig. 9 Individual sample (A) and unit average (B) whole-rock MgO values (wt%) plotted versus zircon U–Pb age for intrusions of the Chuquicamata–El Abra porphyry copper belt (Ballard et al. 2001, 2003). The Pajonal–El Abra igneous suite is indicated by *circles*, the Los Picos–Fortuna suite by *squares*, the Chuquicamata Intrusive Complex (including the East, West, and Bench porphyries in the Chuquicamata mine and the major and minor facies porphyries of the Radomiro Tomic mine) by *diamonds*. Uncertainty envelopes are 2 standard errors for ages, and 1 standard deviation for MgO values. Successive intrusions of the Los Picos–Fortuna/Pajonal–El Abra complex display significant compositional overlap but unit average MgO values decrease with age

saturation of a separate sulfide phase. Chalcophile elements such as Cu and Au partition strongly into magmatic sulfide phases, in contrast to their general incompatible behavior in silicate and oxide minerals (Lynton et al. 1993; Fleet et al. 1996; Jugo et al., 1999). If a magma experiences early sulfide saturation, most of the Cu and Au will be sequestered in sulfides trapped in

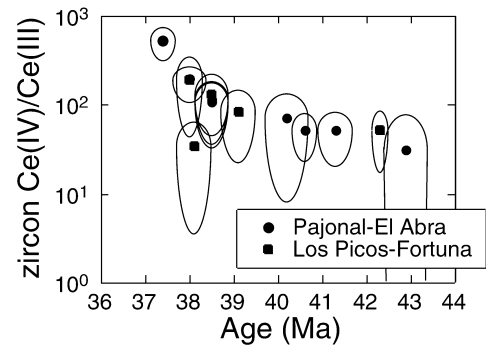


Fig. 10 Average zircon Ce(IV)/Ce(III) ratios increase systematically from the oldest, most mafic, barren intrusions of the Los Picos–Fortuna/Pajonal–El Abra igneous complex to the youngest, most felsic, and ore-bearing unit, the El Abra mine porphyry. The one apparent outlier to this trend is the Fortuna leucocratic intrusion which also displays trace-element evidence which distinguishes it from the main sequence. Zircon U–Pb ages are taken from Ballard et al. (2003)

cumulates where they will be unavailable to a late-stage magmatic-hydrothermal fluid.

At higher oxygen fugacities, magmatic sulfur exists mainly as sulfate (SO_4^{2-}) which has a much higher solubility in silicate melts than sulfide (Carroll and Rutherford 1985) and will tend to delay or even prevent saturation of a magmatic sulfide phase. The transition between sulfide-dominant and sulfate-dominant melt compositions is quite abrupt, occurring over about one log unit of oxygen fugacity (Matthews et al. 1999). Copper and Au will accumulate in the melt during differentiation under oxidized conditions, and partition into a magmatic-hydrothermal fluid should the magma reach fluid saturation (Ulrich et al. 1999). If S saturation is reached, it will involve anhydrite (CaSO_4) into which the chalcophile elements do not partition. Thus, all other factors being equal, oxidized sulfate-bearing hydrous magmas will contain higher concentrations of Cu, Au, and S when they exsolve a late-stage fluid phase compared to their more reduced, sulfide-saturated counterparts. As a consequence, these magmas will release magmatic-hydrothermal fluids with higher concentrations of Cu, Au, and S which, given appropriate depositional conditions, may be better able to generate Cu ± Au mineralization (Cline and Bodnar 1991; Pasteris 1996).

The seven intrusions directly associated with porphyry copper deposits are the most oxidized units of those examined here on the basis of both their individual grain and average zircon Ce(IV)/Ce(III) and $\text{Eu}_N/\text{Eu}_N^*$ ratios (Fig. 11). This correlation provides strong support for the hypothesis that oxidation state is a key factor in determining the potential of calc-alkaline magmas for forming Cu mineralization. If the explanation for the correlation involves redox-related speciation of sulfur as described above, it implies that these ore-bearing intrusions approached or passed through the sulfate–sulfide transition at least in the latter stages of their differentiation history. Conversely, the barren intrusions,

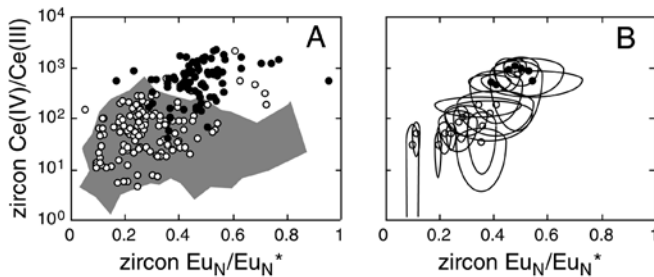


Fig. 11 Individual grain (A) and average (B) Ce(IV)/Ce(III) and $\text{Eu}_N/\text{Eu}_N^*$ ratios in zircons from intrusions of the Chuquicamata-El Abra porphyry copper belt, illustrating the positive correlation between these two indicators of oxidation state. Data for ore-bearing intrusions are shown as closed circles, and for barren units as open circles. The patterned field in A is occupied by unpublished data for 436 detrital zircons from Paleozoic sandstones in eastern North America which support the interpretation of zircon Ce(IV)/Ce(III) and $\text{Eu}_N/\text{Eu}_N^*$ ratios above 300 and 0.4, respectively, representing anomalous magma oxidation states

although becoming increasingly oxidized, apparently remained in the field of sulfide dominance throughout their magmatic evolution.

Implications for mineral exploration

This study has shown that seven calc-alkaline intrusions directly associated with porphyry copper mineralization have higher zircon Ce(IV)/Ce(III) ratios than 14 related but barren intrusions. This result implies that the ore-bearing porphyries were more oxidized than their barren counterparts. If these results are confirmed by further work, then the measurement of the Ce(IV)/Ce(III) ratio in zircon may provide a useful tool in exploration for porphyry copper deposits. This method has two applications. First, to discriminate between ore-bearing and barren intrusions at an early stage of exploration. Second, by measuring the Ce(IV)/Ce(III) ratio in detrital zircons from stream sediments, it should be possible to trace anomalously oxidized zircons back to their prospective igneous source.

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