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Effects of nitrification inhibitor and herbicides on nitrification, nitrite and nitrate consumptions and nitrous oxide emission in an Australian sugarcane soil

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Abstract

This study evaluated the impacts of a nitrification inhibitor (3,4-dimethylpyrazole phosphate, DMPP) and herbicides (atrazine and glyphosate) on nitrification, gross nitrite, and nitrate $(NO_2^{-}-N + NO_3^{-}-N)$ consumption rate, nitrous oxide (N_2O) emission, and abundances of microbial functional genes related to nitrogen (N) cycling in an Australian sugarcane soil. The experiment included four treatments: blank control (CK), DMPP application (NI), atrazine application (ATR), and glyphosate application (GLY). All treatments received (NH_4)₂SO₄ at 50 mg N kg⁻¹ dry soil and KNO₃ at 50 mg N kg⁻¹ dry soil and were incubated initially at 55% of water holding capacity (WHC) for 7 days and subsequently at 75% WHC for another 7 days (K¹⁵NO₃ with 5 atom% ¹⁵N added at the beginning of each stage). Compared with the CK treatment, DMPP application significantly decreased N₂O emissions throughout the incubation, while atrazine or glyphosate application significantly inhibited N₂O emissions only during the 4-7-day period. DMPP application also decreased ammoniumoxidizing bacteria (AOB) amoA gene abundances, gross $NO_2^{-}N + NO_3^{-}N$ consumption rates at 55 and 75% WHC, and nirS and nirK gene abundances of denitrifiers at 75% WHC. The atrazine and glyphosate applications decreased the gross nitrification and $NO_2^-N + NO_3^-N$ consumption rates, abundances of both ammonium-oxidizing archaea (AOA) and AOB amoA genes at 55 and 75% WHC, and abundances of functional genes related to different reactions of the denitrification during the incubation. These results suggested that DMPP, atrazine, and glyphosate could decrease soil gross nitrification and denitrification rates perhaps by inhibiting microbial functional gene abundances and that application of DMPP could effectively reduce N₂O emissions in the sugarcane cropping soil.

Keywords 3,4-dimethylpyrazole phosphate (DMPP) \cdot Atrazine \cdot Glyphosate \cdot Gross nitrification \cdot N₂O emission \cdot Functional gene abundance

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Introduction

As one of the largest sources of energy for human consumption and biofuel, sugarcane (*Saccharum* spp.) farming has significant economic and environmental implications (Thorburn et al. 2010). Sugarcane productivity largely relies on nitrogen (N) fertilizer input, with annual N application rates ranging from 100 to 300 kg N ha⁻¹ (Nachimuthu et al. 2016; Wang et al. 2016b). Sugarcane is mainly cropped in subtropical and tropical areas. The local wet and warm climate can stimulate N loss from the cropping systems via leaching, runoff, and denitrification, which can have detrimental impacts on the environment through water pollution and nitrous oxide (N₂O) emission (Kroon et al. 2016; Wang et al. 2016b).

Nitrification is a key process in N biogeochemistry and is mainly conducted by nitrifiers (Martens-Habbena et al. 2009; Sabba et al. 2015). During this process, ammonium (NH_4^+-N) is firstly oxidized to hydroxylamine (NH2OH), then to nitrite (NO_2^--N) and eventually to nitrate (NO_3^--N) . The NO_3^--N is also the substrate for denitrification in soils and can be easily washed into waterways. Excessive NO₃-N in surface water could lead to eutrophication, and high NO₃⁻-N content in drinking water is harmful to human health (Soares et al. 2012). Moreover, nitrification and denitrification result in nitrous oxide (N₂O) emissions from soils (Menéndez et al. 2012; Scheer et al. 2014). Nitrification inhibitors have been investigated extensively as potential means to reduce N2O emissions and enhance utilization efficiencies of applied N fertilizers (Decock 2014; Scheer et al. 2014). As one of the most popular nitrification inhibitors, 3,4-dimethylpyrazole phosphate (DMPP) has several distinct advantages: (1) DMPP is often more effective, relative to other products, and an application rate of 0.5-1.5 kg ha⁻¹ is sufficient to obtain an optimal inhibition effect (Zerulla et al. 2001; Scheer et al. 2014); (2) DMPP application could reduce NO₃⁻-N accumulations in crops and thus improve qualities of agricultural products (Zerulla et al. 2001; Chaves et al. 2006); (3) DMPP is less mobile in soils and may stay in the fertilized spot where NH_4^+ -N is adsorbed (Yu et al. 2007; Kong et al. 2016); and (4) DMPP is mineralized slowly in soils, and its effect could last for a relatively long time (Weiske et al. 2001).

In sugarcane production, weeds are serious threats to yield (Kaur et al. 2016; Nachimuthu et al. 2016). Generally, weeds have well-developed root systems and higher efficiencies of nutrient transportation, and thus they have competitive advantages for nutrient uptake over the sugarcane crops (Anjum and Bajwa 2007; Fahad et al. 2014). Weeds also compete for space, water, and light with the sugarcane crops and release some allelochemicals into the rhizosphere (Fahad et al. 2014). Therefore, it is often essential to apply herbicides, such as atrazine and glyphosate, to control the weeds in sugarcane farming. However, once the herbicides enter into soils, they might generate non-target effects on soil microorganisms, including those involved in N reactions (Haney et al. 2002; Mahía et al. 2011; Nguyen et al. 2016).

Soil moisture is a key factor influencing the abundances of functional genes related to N cycling and the effects of applied agrochemicals (Jiang et al. 2015; Martins et al. 2015). Therefore, responses of nitrification, denitrification, and N₂O emission to applied agrochemical could be affected by soil moisture (Martins et al. 2015; Bento et al. 2016; Cai et al. 2016). To the best of our knowledge, little information is available with respect to the effects of DMPP or herbicides on N cycling microorganisms in tropical or subtropical sugarcane cropping soils. In this study, we employed the isotope labeling method to quantify the effects of different

agrochemicals (DMPP, atrazine, and glyphosate) on gross nitrification and $NO_2^-N + NO_3^-N$ consumption rates at different moisture levels. In the meantime, soil N₂O emissions and functional gene abundances were also determined. This study aimed to improve our understanding of the potential impacts of DMPP and herbicides on N transformations and N₂O emissions and their relationships with biotic and abiotic factors in sugarcane cropping soils.

Materials and methods

Soil samples and chemicals

Soils were collected from a sugarcane field (18° 37' S, 146° 07' E) near the township of Ingham in Queensland, Australia. Mean annual temperature in this region is 24.0 °C, and mean annual rainfall is 2110 mm. The site had a fourth ratoon sugarcane crop (NQ239), initially planted in August 2012 on raised beds with a row spacing of 165 cm. The cropping field received about 150 kg N ha⁻¹ in October 2016. Green cane trash (residue) blanketing had been carried out in this field since 1987, with the crop residues (approximately 10 t of dry matter-equivalent ha^{-1}) being retained on the ground after harvest each year. Surface soil samples (0-10 cm) were randomly taken from the cropping beds, homogenized, and divided into two parts. One part of the soil sample was air-dried and sieved to < 2 mm for determinations of soil physical and chemical properties, and the other part was used for the incubation experiment. Main physical and chemical properties of the soil were clay content, 19%; sand content, 62%; pH (in water), 5.0; electrical conductivity, 3.4 ms m^{-1} ; total C content, 10.8 g kg⁻¹ dry soil; and total N content, 0.84 g kg⁻¹ dry soil (Wang 2016). The DMPP (ChemCruz[™]®, Santa Cruz Biotechnology, USA) was a water-soluble chemical agent grade powder containing 98% of the active ingredient. Atrazine (Gesaprim Granules®, Syngenta Crop Protection Pty Ltd., Australia) and glyphosate (Yates Zero®, Yates, Australia) were water-soluble commercial herbicides with active ingredient contents of 900 g kg⁻¹ and 440 g L⁻¹, respectively.

Experimental design

Four treatments were included in this study: (1) nitrification inhibitor DMPP application (NI) at 1% of the NH_4^+ -N input, (2) atrazine application (ATR) at 2.25 mg active ingredient kg^{-1} soil (on dry mass basis, equivalent to 3.00 kg active ingredient ha^{-1}), (3) glyphosate application (GLY) at 2.16 mg active ingredient kg^{-1} soil (equivalent to 2.88 kg active ingredient ha^{-1}); and (4) the blank control without any agrochemical application (CK). Each treatment had three replicates. The DMPP, atrazine, and glyphosate application rates were based on the respective recommended rates (Bonfleur et al. 2015; Zhang et al. 2018). Soil samples (50 g dry weight) were placed into 500 mL glass jars (6.8 cm in external diameter \times 14.5 cm in height). The agrochemicals were dissolved in double distilled H₂O (ddH₂O) and then added to the soils. Nine soil samples for each treatment were treated with $(NH_4)_2SO_4$ at 50 mg N kg⁻¹ soil, K¹⁵NO₃ (5 atom% ¹⁵N) at 10 mg N kg⁻¹ soil, and KNO₃ (unlabeled) at 40 mg N kg⁻¹ soil, and six soil samples for each treatment were initially treated with $(NH_4)_2SO_4$ at 50 mg N kg⁻¹ soil and KNO₃ (unlabeled) at 50 mg N kg⁻¹ soil. The soil samples were first adjusted to 55% water holding capacity (WHC). Three replicates of the nine soil samples treated with K15NO3 (5 atom% ¹⁵N) were destructively sampled on days 0, 3, and 7. To minimize possible impacts of ¹⁵NO₃-N recycling between different N forms, K¹⁵NO₃ (5 atom% ¹⁵N) solutions were added into the remaining six soil samples at 1.0 mg N kg⁻¹ dry soil on day 8 so as to adjust the soil moistures to 75% WHC. Test soils were further incubated and sampled (three replicates) on days 10 and 14. During the incubation, the glass jars were covered with parafilm and incubated in the dark at 28 °C, and an amount of ddH₂O equal to the weight loss was added to maintain soil moisture, when necessary. At each destructive sampling, the soils in each jar were homogenized and randomly taken to determine soil mineral N contents and microbial properties.

Determinations of mineral N contents and ^{15}N abundances in NO₂⁻-N + NO₃⁻-N

Soil exchangeable NH_4^+ -N and NO_2^- -N + NO_3^- -N were extracted by 2 M KCl (1:5 *w/v*). The extract was then filtered, and the filtrate (7–8 mL) was analyzed for mineral N (NH_4^+ -N and NO_2^- -N + NO_3^- -N) content using a colorimetric method (Rayment and Lyons 2010). The same amount of KCl solution was also filtered and analyzed as the blank.

The ^{15}N abundance in NO₂⁻-N + NO₃⁻-N was determined using the method described by Mary et al. (1998) with minor modification. Briefly, each filtrate (10 mL) was transferred into a 70-mL plastic jar. After adding 0.4 g of MgO into the filtrate, the jar was immediately sealed and placed on a rotary shaker for 7 days. The NH4⁺-N was transformed into NH3 and then the jar was opened to allow the NH₃ to volatilize. A filter paper disk (0.5 cm in diameter) was treated with 5 µL of 2.5 M KHSO₄, enclosed in Teflon tape and stuck to the lid of the jar. After addition of 0.2 g of reducing agent (Devarda's alloy), the jar was immediately closed and placed in the rotary shaker at 25 °C for 7 days. The NO₂⁻-N + NO₃⁻-N was converted to the NH₃ which was then absorbed by the KHSO₄-impregnated filter paper disk. After drying in a desiccator with concentrated H₂SO₄, the filter paper disk was encapsulated into a tin cup, and the ¹⁵N atom% in NO₂⁻-N + NO₃⁻-N was determined with a mass spectrometer (Isoprime-EuroEA 3000, Milan,

Italy). The $(NH_4)_2SO_4$ was used as an elemental standard to calculate the total N content. Primary standards (IAEA-N1 and IAEA-N2) were employed to calibrate the instrument for the quantification of ¹⁵N abundance.

Gas sampling and determination

The incubation jar was flushed with compressed air and sealed for 24 h prior to gas sampling, with the exception of the first gas sampling (0–6-h period). At the end of the enclosure period, a gas sample was taken with a 25-mL syringe from the headspace of each jar and was injected into a pre-evacuated 12-mL vial. The gas sample was analyzed with a gas chromatograph (Varian CP-3800, Middelburgh, Netherlands) having an electron capture detector (ECD). High purity N₂ was used as the carrier gas, and the column and ECD temperatures were 45 and 310 °C, respectively (Wang et al. 2011). A series of standard gas samples with different concentrations of N₂O (0, 0.4, 5, 10, and 20 μ L L⁻¹) were used for calibrating the gas chromatograph. The cumulative N₂O emissions were calculated assuming linear changes in N₂O emission rates during the period between two consecutive gas sampling events.

Soil DNA extraction and real-time quantitative PCR (qPCR)

Soils (approximately 0.25 g) were extracted for the total genomic DNA by a DNeasy® PowerSoil® Kit (QIAGEN, Gmbh, Germany). After extraction, DNA quality and concentration were evaluated with a Nanodrop 2000 spectrophotometer (Thermo Scientific, USA). The DNA samples were kept at -20 °C prior to determinations of the microbial properties.

Abundances of N cycling functional genes, including ammonium-oxidizing archaea and bacteria (AOA and AOB) amoA genes and denitrifying NO₃⁻-N reductase (narG), NO₂⁻-N reductase (nirK and nirS), and N₂O reductase (nosZ) genes, were quantified by the qPCR with a Bio-Rad Real-Time Detection System. The reaction system was 20.0 µL containing 0.5 µL of forward primer, 0.5 µL of reverse primer, 2.0 µL of DNA, 7.0 µL of sterile ddH₂O, and 10.0 µL of SYBR® Premix Ex Taq[™] (TaKaRa Biotech). The primers of functional genes with the related references and qPCR reaction conditions were shown in Table S1. The negative control with sterile ddH₂O as amplification template was also used in each gene determination. The standard was obtained by performing serial dilutions of the plasmid with the target gene. After the determination of fluorescence, melting curve analysis was also conducted to confirm PCR product specificity (Zhang et al. 2018). In this study, the amplification efficiencies ranged from 95.1 to 118.9%, with R^2 values > 98.5% for the calibration curves of functional genes.

Calculations and statistical analysis

Based on $NO_2^-N + NO_3^-N$ contents and ¹⁵N atom% of $NO_2^-N + NO_3^-N$, net nitrification, gross nitrification, and gross $NO_2^-N + NO_3^-N$ consumption (denitrification and biological immobilization) rates were calculated as follows (Mary et al. 1998; Gómez-Rey and González-Prieto 2015):

Net nitrification rate = $\frac{N2-N1}{T2-T1}$

Gross nitrification rate $= -\frac{N2-N1}{T2-T1} * \frac{\ln(E_{N2}/E_{N1})}{\ln(N2/N1)}$

Gross $NO_2^{-}N + NO_3^{-}N$ consumption rate

= gross nitrification rate-net nitrification rate

where N1 and N2 were the NO₂⁻-N + NO₃⁻-N contents at time T1 and T2, respectively. The E_{N1} and E_{N2} were the ¹⁵N atom % excess of NO₂⁻-N + NO₃⁻-N at time T1 and T2, respectively.

Two-way analysis of variance was conducted to determine significant differences among treatments, sampling time, and their interactions using SPSS v. 21.0, followed by Duncan's multiple range test to determine significant differences (P <0.05) among the different treatments. The MATLAB clustergram function was employed to generate a heat map for visual representation of gene abundances in different treatments and incubation times. The functional gene abundances, soil moistures, and mineral N contents were the candidate variables in the path analyses for gross nitrification and gross $NO_2^{-}N + NO_3^{-}N$ consumption rates. Stepwise regression analyses were employed for fitting the best prediction models of daily N₂O emissions with soil moistures, mineral N contents, and functional gene abundances. Moreover, the relationships among all functional gene abundances and daily N2O emissions under different soil moisture conditions were evaluated by the principal component analysis (PCA).

Results

Effects of DMPP and herbicides on gross nitrification and gross NO_2^{-} -N + NO_3^{-} -N consumption rates

Soil NO₂⁻-N + NO₃⁻-N contents in the CK treatment changed little during the first 7 days of incubation, but increased from $53.5 \pm 1.2 \text{ mg N kg}^{-1}$ soil on day 7 to $74.2 \pm 1.7 \text{ mg N kg}^{-1}$ soil on day 14 (Fig. S1A). The exchangeable NH₄⁺-N content decreased from days 7 to 14 (Fig. S1B). The four treatments had approximately similar net nitrification rates at 55 or 75% WHC (the slopes of changes in NO₂⁻-N + NO₃⁻-N contents with time under each treatment in Fig. S1A). Compared with the CK treatment, DMPP, and atrazine applications significantly decreased gross nitrification rates from day 3 onwards (P < 0.05; Table 1). Glyphosate application decreased gross nitrification rates during the 0–3 and 10–14-day period to 34.0 and 45.2% of those in the CK treatment, respectively, but had negligible effect on gross nitrification rate during the 4–7-day period. Gross NO₂⁻-N + NO₃⁻-N consumption rates exhibited similar treatment effects to the gross nitrification rates (Table 1). The agrochemical applications generally decreased gross NO₂⁻-N + NO₃⁻-N consumption rates, except the ATR treatment during the 0–3-day period.

Effects of DMPP and herbicides on N₂O emissions

As shown in Table 2, the NI treatment always had the lowest cumulative N_2O emissions among the four treatments throughout the incubation. The N_2O emissions in the ATR and GLY treatments were significantly lower than that in the CK treatment during the 4–7-day period (P < 0.05), but not during other incubation periods. In the whole incubation, compared with the CK treatment, DMPP application significantly decreased total N_2O emissions (P < 0.05), and applications of atrazine and glyphosate also decreased total N_2O emissions, with the differences between the CK and ATR or GLY treatments being not significant.

Effects of DMPP and herbicides on abundances of AOA amoA and AOB amoA genes

During the first 7 days at 55% WHC, the AOA *amo*A gene abundances increased with time in the CK treatment (Fig. 1a). The AOA *amo*A gene abundances in the NI treatment did not significantly differ with respect to those in the CK treatment during the 3–14-day period. However, DMPP application significantly decreased the AOB *amo*A gene abundances from day 3 onwards (P < 0.05). Atrazine or glyphosate application resulted in significantly lower abundances of both AOA and AOB *amo*A genes, relative to the CK treatment (Fig. 1). The atrazine and glyphosate had similarly negative impacts on the abundances of AOA and AOB *amo*A genes at 55% WHC, but the inhibitory effects of glyphosate diminished considerably at 75% WHC.

Effects of DMPP and herbicides on abundances of functional genes related to denitrification

The DMPP and herbicide applications generated various negative effects on abundances of functional genes related to denitrification (Fig. 2). In the CK treatment, the *nar*G gene abundance increased with time in the first 7 days of incubation and then changed slightly at 75% WHC (Fig. 2a). Compared with the CK treatment, the *nar*G gene abundance was not

Treatments	Gross nitrification rates (mg N kg^{-1} soil day ⁻¹)			Gross NO_2^{-} -N + NO_3^{-} -N consumption rates (mg N kg ⁻¹ soil day ⁻¹)			
	Days 0–3	Days 4–7	Days 10–14	Days 0–3	Days 4–7	Days 10–14	
СК	10.12 ± 1.84 a	10.73 ± 1.38 a	17.84 ± 6.35 a	9.56 ± 1.79 ab	15.07±0.91 a	7.75±1.82 a	
NI	7.84 ± 2.72 a	$5.05\pm1.19~b$	7.74 ± 2.91 b	$7.53\pm1.60\ b$	7.94 ± 1.64 b	-1.20 ± 0.44 c	
ATR	11.38 ± 0.18 a	$4.37 \pm 0.57 \ b$	5.58 ± 2.50 b	10.70 ± 0.87 a	$5.00 \pm 1.51 \text{ b}$	$2.23\pm0.63~b$	
GLY	$3.45\pm1.24\ b$	10.82 ± 3.05 a	$8.07\pm2.80~b$	2.38 ± 1.63 c	$6.27 \pm 3.67 \text{ b}$	-1.69 ± 0.29 c	

Table 1 Gross nitrification and gross NO₂⁻·N + NO₃⁻·N consumption rates in different treatments

CK blank control, NI nitrification inhibitor DMPP, ATR atrazine, GLY glyphosate

Significant differences among the different treatments were shown by different letters

significantly affected by the DMPP application. According to the results of Fig. 2b, atrazine application consistently reduced *nir*K gene abundances from day 3 onwards, and the *nir*K gene abundance in the ATR treatment was 29.6% of that in the CK treatment at the end of incubation. Compared with the CK treatment, DMPP application had negligible effects on *nir*K and *nir*S gene abundances at 55% WHC, but significantly decreased *nir*K and *nir*S gene abundances at 75% WHC (P < 0.05). Atrazine and glyphosate applications also decreased *nir*S gene abundances at 75% WHC (Fig. 2c). The *nosZ* gene abundances were decreased by the herbicides on day 3, but recovered to comparable levels to the CK treatment from day 7 onwards (Fig. 2d). At the end of incubation, there was no significant difference in *nosZ* gene abundances among the four treatments.

Relationships among functional gene abundances, mineral N contents, N₂O emissions, and N transformation rates

The abundances of functional genes revealed treatmentinduced effects from day 7 onwards (Fig. S2), and the samples of CK treatment on days 7, 10, and 14 were distinctly clustered, while earlier CK samples were mixed among the other treatments. The path analysis of factors potentially affecting gross nitrification rates (CMIN/df = 7.937, GFI = 0.891, and CFI = 0.911) indicated that soil moisture, exchangeable NH₄⁺-

 Table 2
 Cumulative N₂O

 emissions in different treatments

N, NO₂⁻-N + NO₃⁻-N, AOA *amo*A, and AOB *amo*A explained 46.0% of the variance in gross nitrification rates (Fig. 3a). The increases in abundances of both AOA *amo*A and AOB *amo*A genes probably enhanced gross nitrification rates, and the effect of AOB *amo*A gene was stronger than that of AOA *amo*A gene. For gross NO₂⁻-N + NO₃⁻-N consumption rates, the path analysis also had an acceptable model (CMIN/df = 1.223, GFI = 0.975, and CFI = 0.997). Among the four functional genes, the *nir*S gene had the strongest impact on the gross NO₂⁻-N + NO₃⁻-N consumption rates (Fig. 3b).

The stepwise regression models for predicting daily N2O emissions differed for the different incubation periods (Table 3). Throughout the 14-day incubation, daily N₂O emissions were positively correlated with AOB amoA abundances and soil moistures. At 55% WHC, daily N2O emissions were best described by the AOB amoA and nosZ gene abundances, with a positive and a negative relationship, respectively. However, when soil moisture was adjusted to 75% WHC, daily N₂O emissions were best described by the nirS gene abundances. An overall PCA also demonstrated the relationships among functional gene abundances and daily N2O emissions at different soil moistures (Fig. 4). At 55% WHC, the two-dimensional PCA plot explained 80.85% of the total variance, with PCA1 accounting for a greater percentage (67.46%). The N₂O emissions at 55% WHC were positively related to AOB amoA and AOA amoA gene abundances, but

Treatments	Cumulative N ₂ O emissions (μ g N kg ⁻¹ soil)						
	Days 0–3	Days 4–7	Days 8–10	Days 11–14			
СК	0.039 ± 0.029 a	0.247 ± 0.076 a	0.100 ± 0.042 a	0.422 ± 0.069 a			
NI	$0.002 \pm 0.0001 \text{ b}$	$0.074 \pm 0.042 \ b$	$0.018 \pm 0.007 \; b$	$0.086 \pm 0.033 \ b$			
ATR	0.057 ± 0.022 a	$0.084 \pm 0.054 \ b$	0.097 ± 0.044 a	0.338 ± 0.102 a			
GLY	0.087 ± 0.034 a	$0.078 \pm 0.018 \ b$	0.110 ± 0.034 a	0.441 ± 0.129 a			

CK blank control, NI nitrification inhibitor DMPP, ATR atrazine, GLY glyphosate

Significant differences among the different treatments were shown by different letters



Fig. 1 Effects of nitrification inhibitor and herbicides on abundances of soil **a** AOA *amoA* and **b** AOB *amoA* genes related to nitrification. CK blank control, NI nitrification inhibitor DMPP, ATR atrazine, and GLY glyphosate. Significant differences (P < 0.05) among different treatments for the same incubation time were shown with different lower case letters

had negative relationship with *nosZ* gene abundances (Fig. 4a). At 75% WHC, the PCA1 and PCA2 explained 50.45 and 18.74% of the total variance, respectively, and the N₂O emissions had close correlations with *nirS*, *nosZ*, and AOB *amoA* gene abundances (Fig. 4b).

Discussion

Ammonia oxidation is the rate-limiting step of soil nitrification, and previous studies have demonstrated that nitrification rates were positively correlated with AOA or AOB *amoA* gene abundances (Caffrey et al. 2007; He et al. 2007; Prosser and Nicol 2012). Relative to the CK treatment, DMPP application inhibited gross nitrification rates (Table 1), similar to what has already been reported (Florio et al. 2014; Liu et al. 2015). However, previous studies revealed different mechanisms of DMPP in controlling soil nitrification. Liu et al. (2015) suggested that DMPP application could inhibit soil nitrification by decreasing AOA amoA gene abundances rather than AOB amoA gene abundances in acid soils. Florio et al. (2014) also demonstrated that the transcription of AOA amoA gene was more sensitive to the DMPP than its bacterial counterpart. However, DMPP application was found to have negligible effects on AOA or AOB growth by Kong et al. (2016). Our study demonstrated that DMPP application significantly decreased AOB amoA gene abundances, whereas it had negligible effects on AOA amoA gene abundances (Fig. 1a). The decline in AOB amoA gene abundances confirmed the finding by Kleineidam et al. (2011) and coincided with the decline in gross nitrification rates. Therefore, DMPP application inhibited gross nitrification perhaps by decreasing the abundances of AOB amoA gene in this sugarcane cropping soil.

Atrazine or glyphosate application also decreased gross nitrification rates (Table 1). Previous studies demonstrated that atrazine application could generate non-target effects on soil microorganisms and decrease soil microbial biomass (Mahía et al. 2008, 2011). Indeed, both soil AOA and AOB amoA gene abundances decreased after atrazine application in the present study. The AOB amoA gene abundances also significantly decreased in response to glyphosate application (Fig. 1). The non-target suppressive effects of atrazine and glyphosate on the functional genes related to N cycling might be responsible for the inhibitions of gross nitrification and gross $NO_2^{-}N + NO_3^{-}N$ consumption and thus the decrease in N₂O emission. Jiang et al. (2015) also found that application of the herbicide butachlor reduced N2O emission by decreasing denitrifying bacteria abundance in a rice-wheat cropping system. Moreover, atrazine application could decrease soil urease activity and N mineralization rate, and glyphosate application could inhibit soil N fixation (Santos and Flores 1995; Mahía et al. 2011). These observations and our results suggest that most soil N transformations could be suppressed by atrazine and glyphosate applications. However, soil characteristics, such as pH value and moisture, could affect the longevity and impact of the applied herbicide (Bento et al. 2016; Nguyen et al. 2016). For example, in contrast to the above findings, Allegrini et al. (2017) found that in the Argentine soils, abundances of AOB amoA gene (copies ng⁻¹ DNA) were not affected by three repeated glyphosate applications at 49.0 mg kg⁻¹ soil per application. Mahía et al. (2011) showed that soil N availability was a key factor determining the degradation rate of atrazine. Future study should examine the effects of herbicides on other N cycling processes in relation to soil properties, and longer incubation periods are also needed to determine the longevities of the effects after herbicide applications.

The ratios of AOA to AOB *amo*A gene abundances in this study ranged from 5.79 to 47.49, confirming previous reports that the AOA *amo*A gene was more abundant than AOB





Fig. 2 Effects of nitrification inhibitor and herbicides on abundances of soil **a** *nar*G, **b** *nir*K, **c** *nir*S, and **d** *nos*Z genes related to denitrification. CK blank control, NI nitrification inhibitor DMPP, ATR atrazine, and

GLY glyphosate. Significant differences (P < 0.05) among different treatments for the same incubation time were shown with different lower case letters

Fig. 3 Path analyses of the factors potentially affecting **a** gross nitrification and **b** gross NO₂⁻-N + NO₃⁻-N consumption rates during the incubation. Solid arrows showed positive effects, and dash arrows showed negative effects. Data on the arrows were standardized direct effects, and * and ** showed statistically significant at P < 0.05 and P < 0.01, respectively



Periods	Stepwise regression models	R^2	P value
Days 0–14	Daily N ₂ O emission = $6.53 \times 10^{-10} \times AOB$ <i>amo</i> A abundance + $1.45 \times 10^{-2} \times soil$	0.395	< 0.01
Days 0-7 (55% WHC)	moisture -7.18×10^{-3} Daily N ₂ O emission = $7.89 \times 10^{-10} \times AOB$ amoA abundance $-2.87 \times 10^{-11} \times nosZ$	0.437	< 0.01
Days 8-14 (75% WHC)	abundance + 4.68×10^{-9} Daily N ₂ O emission = $1.04 \times 10^{-11} \times nir$ S abundance + 1.90×10^{-3}	0.229	< 0.05

Table 3 Quantitative relationships of daily N_2O emissions to functional gene abundances, soil moistures, and exchangeable NH_4^+ -N and NO_2^- -N + NO_3^- -N contents during different incubation periods

amoA gene in some soils (Hayden et al. 2010; Martins et al. 2015). Generally, AOA could inhabit soils under a broad range of environmental conditions and are less susceptible to soil disturbances than AOB (Ouyang et al. 2016; Zhang et al. 2016), due to the differences in cellular compositions, energy metabolisms, and tolerances to soil xenobiotics (French et al. 2012; Hatzenpichler 2012). However, there are controversial views on the roles of AOA and AOB in soil nitrification. Prosser and Nicol (2012) found that AOA amoA gene abundance controlled soil nitrification, whereas we suggested that in this soil, AOB amoA gene was more important in determining gross nitrification rate than AOA *amoA* gene (Fig. 3), confirming the finding by Jia and Conrad (2009). Moreover, others showed that both AOA amoA and AOB amoA genes could influence soil nitrification (He et al. 2007; Wang et al. 2016a). Selective inhibitors, such as 1-octyne specifically inhibiting AOB ammonia monooxygenase, may be used to investigate the contributions of AOB and AOA to soil nitrification and N₂O emission in future studies (Wang et al. 2016a).

As a key greenhouse gas, N_2O can be emitted from both soil nitrification and denitrification. There has been no consensus on the relationships between soil N_2O emissions and abundances of functional genes in soil N cycling (Liu et al. 2013; Wu et al. 2017). According to Liu et al. (2013), N_2O emissions from

abundances of functional genes related to denitrification. In contrast, Wu et al. (2017) demonstrated that nirK and nirS gene abundances were explanatory variables in estimating N₂O emissions from different agricultural soils. Considering that denitrification occurs under anaerobic conditions, nitrification should be the main process of N₂O production at 55% WHC, since this moisture was not favorable for soil denitrification (MacGregor 1972; Lan et al. 2013). This hypothesis was supported by the positive correlations between N2O emissions and AOB amoA gene abundances (Table 3; Fig. 4a). However, at 75% WHC, denitrification activity could increase, and soil denitrification was probably the primary source of N₂O emission under this wet condition (Duan et al. 2017). Consequently, the N₂O emission was best described by the nirS gene abundance (Table 3). Similarly, Yang et al. (2017) revealed that there were significant correlations between soil N2O emissions and nirS or nirK gene abundances under the conditions favorable for denitrification. Duan et al. (2017) also observed a good correlation between nirS gene abundances and denitrification activities. Based on these observations and the lower gross nitrification and gross $NO_2^{-}N + NO_3^{-}N$ consumption rates in the NI treatment (Table 1), we suggest that DMPP probably decreased N₂O emissions by inhibiting AOB amoA gene abundances at

forest soils were regulated by environmental factors rather than

Fig. 4 Principal component analyses of the relationships among functional gene abundances and daily N_2O emissions at **a** 55% WHC and **b** 75% WHC



55% WHC and *nir*S gene abundances at 75% WHC. It should be mentioned that the primers used for the determinations of *nir*K and *nir*S gene abundances mainly detected the *nir*K- or *nir*S-denitrifiers of the phylum *Proteobacteria* (Wei et al. 2015). Recent studies showed that some denitrifiers belonging to *Actinobacteria*, *Chloroflexi*, or *Spirochetes* could not be quantified with the primers used in this study. Thus, the determined gene abundance might be lower than that of denitrifiers inhabiting the test soil (Wei et al. 2015; Bonilla-Rosso et al. 2016). Use of novel primers may provide a more comprehensive understanding of the impacts of agrochemicals on *nir*K- or *nir*S-denitrifiers and their roles in soil N₂O emissions.

Conclusion

Compared with the CK treatment, DMPP application significantly inhibited gross nitrification rates and decreased AOB amoA gene abundances, but had no significant effects on AOA amoA gene abundances. Therefore, DMPP application inhibited gross nitrification in the sugarcane soil probably by reducing AOB amoA gene abundances. DMPP also decreased N₂O emissions by inhibiting soil nitrification at 55% WHC and perhaps denitrification at 75% WHC. Both atrazine and glyphosate applications decreased gross nitrification rates perhaps by inhibiting abundances of both AOA and AOB amoA genes and decreased gross $NO_2^{-}N + NO_3^{-}N$ consumption rates by inhibiting various denitrifying genes, resulting in lower cumulative N2O emissions. Further studies on herbicides should take into account other N transformation processes including gross N mineralization and N immobilization, and longer incubation periods are also needed to determine the longevities of the effects after herbicide applications in different soil types.

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