


Nitrogen fertiliser-induced changes in N₂O emissions are attributed more to ammonia-oxidising bacteria rather than archaea as revealed using 1-octyne and acetylene inhibitors in two arable soils

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Abstract Nitrification is believed to be one of the major sources of N₂O production emitted from soil. Previous studies showed that both ammonia-oxidising bacteria (AOB) and archaea (AOA) can produce N₂O via nitrification but their relative contributions are still poorly defined. Here, we used acetylene, an inhibitor of AOB and AOA ammonia monooxygenase (AMO), and 1-octyne, a selective inhibitor that specifically inhibits AOB AMO, to investigate how AOB versus AOA contribute to N₂O emissions in two distinct arable soils. Soil amended with ammonium (NH₄⁺) increased N₂O emissions to a greater extent than nitrate (NO₃⁻), and acetylene had a greater impact on N₂O emissions in NH₄⁺-treated soils than that in NO₃⁻-amended soils, which indicated that nitrification was the dominant N₂O emitting process in these two arable soils. In the alluvial and red soil, the percentage of evolved N₂O after application of NH₄⁺ by AOB were 70.5 ~ 78.1 % and 18.7 ~ 19.7 % by AOA, respectively. Quantitative PCR revealed that NH₄⁺ addition stimulated AOB growth, and the growth could be significantly inhibited by acetylene or 1-octyne in the two soils. The stimulation of N₂O emissions by NH₄⁺ and the relative suppression by inhibitors paralleled fluctuations in the AOB growth. In

addition, cumulative N₂O emissions were not correlated with AOA abundance in the two soils. Our results revealed that AOB could contribute more to soil N₂O production than AOA in the NH₄⁺-amended arable soils.

Keywords Nitrous oxide (N₂O) emission · Nitrification inhibitors · Quantitative PCR · *amoA* gene · Arable soil

Introduction

Nitrous oxide (N₂O) is a powerful greenhouse gas that results in global warming and stratospheric ozone depletion. Increasing application of mineral N fertiliser and the expansion of arable soils contributes to over half of anthropogenic N₂O production over the past few decades (Davidson 2009; Smith et al. 2012; Shcherbak et al. 2014). Therefore, it is necessary to understand the microbial mechanism of N₂O emission from soil in order to develop future mitigation strategies and achieve sustainable agriculture practices.

Nitrification is an important N cycling process in agricultural ecosystems, which oxidises ammonia (NH₃) to nitrate (NO₃⁻) via intermediate products, i.e. hydroxylamine (NH₂OH) and nitrite (NO₂⁻) in a two-step process, and can consequently emit N₂O as a by-product under certain soil conditions (Wrage et al. 2001). Ammonia oxidation, the oxidation of NH₃ to NO₂⁻ via NH₂OH, is the rate-limiting step in nitrification and can be carried out by both ammonia-oxidising bacteria (AOB) and ammonia-oxidising archaea (AOA). Nitrification-related processes (the oxidation of NH₂OH and NO₂⁻ reduction) mediated by AOB are recognised to be a main pathway of N₂O emission from arable soils (Gdde and Conrad 1999; Zhu et al. 2013; Huang et al. 2014a). The traditional viewpoint that soil ammonia oxidation and associated N₂O emission is exclusively carried out by

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AOB has been challenged by the discovery of *amoA* and *nirK* genes in AOA strains (Venter et al. 2004; Lund et al. 2012) and by the demonstration of the N₂O production capacity of AOA enriched or isolated from arable soils and marine ecosystem (Jung et al. 2011, 2014; Santoro et al. 2011; Loscher et al. 2012). In most soils, AOA outnumbers AOB abundance, a common feature in multiple ecosystems (Leininger et al. 2006; He et al. 2007; Hu et al. 2013), and a high AOA ammonia oxidation activity has been assessed in certain soils (Yao et al. 2011; Zhang et al. 2012; Hu et al. 2015). These results suggested that AOA could play a greater role than AOB in ammonia oxidation and N₂O production in some soils (Nicol et al. 2008; Stieglmeier et al. 2014; Hink et al. 2016). However, by characterising the conserved *amoA* gene, which encodes a subunit of the ammonia monooxygenase (AMO), it has been observed that the populations of ammonia oxidisers are largely controlled by soil type, NH₄⁺ concentration, pH and water content (Nicol et al. 2008; Di et al. 2009; Chen et al. 2010; Norton and Stark 2011). In ammonium-rich soils, AOB abundance and activity increased whereas AOA abundance is unaffected or inhibited in response to a high concentration of ammonium fertilisers (Jia and Conrad 2009; Di et al. 2009; Di and Cameron 2011; Sterngren et al. 2015; Ouyang et al. 2016). In unfertilised or acidic soils, AOA abundance and metabolic activity are much higher than those of AOB (Offre et al. 2009; Gubry-Rangin et al. 2010; Zhang et al. 2012). Thus, the role of AOB and AOA in overall ammonia oxidation and N₂O emission is not very well understood across different soils.

A key functional enzyme in both bacterial and archaeal ammonia oxidisers is AMO. Acetylene (C₂H₂) is a suicide substrate for AMO (Hynes and Knowles 1978), and thus inhibits ammonia oxidation by autotrophic nitrifiers (Jia and Conrad 2009) at low concentration (10–100 Pa) and inhibits N₂O reductase during denitrification at high concentration (10 KPa) (Huang et al. 2014b; Cui et al. 2016). Inhibition of soil ammonia oxidation by C₂H₂ is also used in routine methods to distinguish N₂O production by nitrifiers and denitrifiers in soils (Bateman and Baggs 2005; Zhu et al. 2013). Recently, it has been shown that AOB is more sensitive to 1-octyne, a C₈ alkyne inhibitor, compared with AOA (Taylor et al. 2013, 2015). Moreover, Lu et al. (2015) and Ouyang et al. (2016) distinguished AOB and AOA contribution to nitrification by using 1-octyne in soil microcosm incubation and found that 1-octyne selectively inhibited AOB growth rather than AOA in forest and arable soils. Therefore, we could use 1-octyne to assess the relative contribution of AOB and AOA to N₂O emission from two different arable soils. We hypothesised that nitrification was the main process of N₂O emission in arable soils and AOB rather than AOA was the major contributor to soil N₂O production in ammonium-rich arable soils.

Material and methods

Soil samples

Two arable soils from Luancheng county (40°7'34"N, 119°11'27"E), Hebei Province in northern China and Qiyang county (26°24'26"N, 112°00'45"E), Hunan Province in southern China were collected from the 0 to 20 cm layer: an upland alluvial soil and an upland red soil, respectively. The alluvial soil from Luancheng is classified as an aquic inceptisol according to US soil taxonomy (USDA 1994); its main properties are as follows: total N, 0.42 g kg⁻¹; dissolved organic C (DOC) 26 mg kg⁻¹; pH 8.0 (1:2.5 H₂O); exchangeable NH₄⁺-N 2.4 mg kg⁻¹; exchangeable NO₃⁻-N 117.8 mg kg⁻¹. The red soil from Qiyang is classified as paleudults in the USDA Soil Taxonomy; its main properties are as follows: total N, 1.51 g kg⁻¹; dissolved organic C (DOC) 33 mg kg⁻¹; pH 6.0 (1:2.5 H₂O); exchangeable NH₄⁺-N and NO₃⁻-N concentrations were 3.0 and 27.8 mg kg⁻¹, respectively.

Microcosm incubation

To evaluate the contribution of different microbial groups to N₂O emission, soils were subjected to microcosms incubation with acetylene and 1-octyne as selective inhibitors. The soil samples were sieved (<2 mm) and stored at 4 °C. The incubation experiments were conducted in 120-ml serum bottles containing 20 g of dry weight soil. The soil samples were pre-incubated for 7 days at 28 °C to stabilize the microbial population at 40 % water filled pore space (WFPS). After pre-incubation, soil was adjusted to 55 % WFPS following amendment with sterilised water (control, no N fertiliser or inhibitors) or a solution of 150 mg N as (NH₄)₂SO₄ or KNO₃ kg⁻¹ soil in the presence or absence of 1-octyne (Oct) (5 μM aqueous) as described by Taylor et al. (2013) or acetylene (Ace) (0.1 % v/v). In total, nine treatments were described as follows: control, Oct, Ace, NH₄⁺, NH₄⁺ + oct, NH₄⁺ + ace, NO₃⁻, NO₃⁻ + oct and NO₃⁻ + ace. Each treatment was replicated three times. As acetylene was able to completely inhibit ammonia oxidation by autotrophic nitrifiers (AOA and AOB), N₂O emission from AOA and AOB (AOA + AOB) was calculated by subtracting N₂O emission in the NH₄⁺ + ace treatment from values measured in the NH₄⁺ only treatment. N₂O production by AOA was determined by subtracting N₂O production in the NH₄⁺ + ace treatment from that detected in the NH₄⁺ + oct treatment, since 1-Octyne specifically inhibits AOB activity but not AOA. N₂O production from AOA plus AOB (AOA + AOB) minus N₂O emission by AOA was attributed to AOB. The relative contribution of N₂O emission by AOA and AOB was therefore deduced.

N₂O and soil sampling

Gas samples (10 ml) were collected from the headspace with syringe at 1, 2, 3, 4, 7, 10, 14 and 21 days after the start of incubation. N₂O concentration was analysed by a gas chromatograph (HP7890A, Agilent Technologies, CA, USA). Cumulative N₂O emission during the incubation period was estimated according to Ma et al. (2009). Aerobic conditions were maintained by opening bottles for 20 min at intervals of 2–3 days when water content and 1-octyne or acetylene were supplemented (Offre et al. 2009; Taylor et al. 2013; Hink et al. 2016). Three bottles per treatment were destructively sampled after incubation for 0, 7, 14 and 21 days. Exchangeable NH₄⁺-N and NO₃⁻-N were extracted with 0.01 M CaCl₂, under shaking for 1 h, and then, soil suspensions were filtered through Whatman no. 42 filter papers. Extracts were analysed colorimetrically by flow injection analyser (Skalar SAN++). The soil samples for molecular analysis were immediately stored at -80 °C.

DNA extraction and quantitative PCR (qPCR) analyses

Soil DNA was extracted from 0.4 g dry soil with MoBio Powersoil™ DNA isolation kit (MoBio Laboratories Inc, Carlsbad, CA) according to the manufacturer's protocol. Concentration and quality of the extracted DNA were measured using a Nanodrop ND-1000 spectrophotometer (NanoDrop Technologies, Wilmington, DE, USA). Soil DNA extracts were stored at -80 °C.

Quantification of *amoA* for AOB and AOA were determined using *amoA1f/amoA2r* (Rotthauwe et al. 1997) and *crenamoA23f/crenamoA616r* (Hu et al. 2015), respectively. PCR reactions were conducted on a CFX96 (Bio-rad, USA). Each 25- μ l reaction mixture included 12.5 μ l 2 \times Supermix (Bio-rad, USA), 0.5 μ l BSA, 0.5 μ l of each primer (10 mM) and 2 μ l 10-fold diluted DNA and 9 μ l sterilised deionised water. The thermocycling conditions for AOB and AOA were as follows: denaturation at 95 °C for 3 min, 40 cycles of denaturation at 95 °C for 30 s, annealing at 56 °C for 40 s, extension at 72 °C for 50 s and data collection at 83 °C for 15 s. Products' specificity was checked by agarose gel electrophoresis and melting curve analysis. Plasmids as prepared as previously (He et al. 2007) were used as standards for AOB and AOA qPCR, respectively. The efficiencies for AOB and AOA *amoA* amplification ranged from 90 to 98 %, and the correlation coefficient (R^2) was between 0.993 and 0.999 for both genes.

Statistical analysis

The statistical analysis was performed using SPSS 17.0 software (IBM Co., Armonk, NY, USA). ANOVA was performed to assess the significance of different treatments on N₂O

production, *amoA* gene abundance and exchangeable NH₄⁺-N and NO₃⁻-N concentrations. Correlations of cumulative N₂O emissions with AOB or AOA *amoA* gene abundance were assessed by Pearson's correlation procedure.

Results

N₂O emission

In the two soils, there was no clear difference in N₂O production among control, Oct and Ace treatments after 21 days of incubation (Fig. 1). In the alluvial soil, N₂O emission significantly increased with NH₄⁺ addition and reached 287.9 ng N g⁻¹ soil ($P < 0.01$). When NH₄⁺ was applied with 1-octyne or acetylene, cumulative N₂O production was dramatically reduced by 78.1 and 97 %, respectively. Nitrate-associated treatments did not significantly affect N₂O emission (relative to the control) ($P > 0.05$). In the red soil, N₂O emission was stimulated with the application of NH₄⁺, but to a lesser extent, i.e. reaching 84.9 ng N g⁻¹ soil ($P < 0.01$). N₂O emissions in the NH₄⁺ + oct or NH₄⁺ + ace treatments were reduced by 70.5 and 90 %, respectively. No significant difference in N₂O emissions in acetylene-treated soil samples with or without NH₄⁺ addition ($P > 0.05$) were observed. Amendment with NO₃⁻ slightly stimulated N₂O emission compared to the control. In the NO₃⁻-amended soil, acetylene addition inhibited N₂O emission, but 1-octyne had no significant inhibitory effect on it. In addition, the cumulative N₂O emission from the NO₃⁻ treatment was 36.4 and 4.7 times lower than that from the NH₄⁺ treatment in the alluvial and red soils, respectively. These results indicated that nitrification rather than heterotrophic denitrification might be the main sources of N₂O emission in the two soils.

Exchangeable NH₄⁺-N and NO₃⁻-N concentration

In the two soils, exchangeable NH₄⁺-N concentration was consistently low in the control, and 1-octyne alone had no significant effect on it ($P > 0.05$) (Fig. 2). In all acetylene-treated soils, nitrification was completely blocked and exchangeable NH₄⁺-N concentration strongly increased due to accumulation of mineralisation-derived NH₃ after 21 days of incubation. In the alluvial soil, exchangeable NH₄⁺-N concentration in the NH₄⁺ treatment decreased from 143.7 to 2.7 mg N kg⁻¹ soil and was almost completely consumed due to continued oxidation by the ammonia oxidisers in the first 7 days. After that, the exchangeable NH₄⁺-N concentration showed no significant difference from the control ($P > 0.05$). In the red soil, the exchangeable NH₄⁺-N concentration gradually decreased to approach that of the control at day 21. After application of 1-octyne with NH₄⁺, exchangeable NH₄⁺-N concentrations declined more slowly in the two

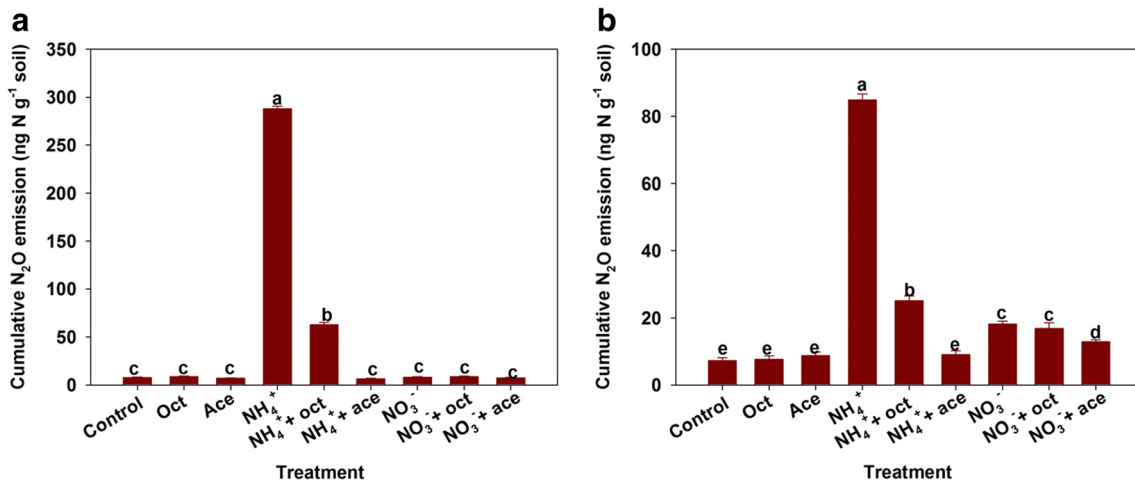


Fig. 1 Cumulative N₂O emissions after 21 days of laboratory incubation in an alluvial soil (a) and a red soil (b) which were amended with distilled sterilised water, NH₄⁺ or NO₃⁻ (control, NH₄⁺ or NO₃⁻), with 1-octyne (Oct, NH₄⁺ + oct or NO₃⁻ + oct) or with acetylene (Ace, NH₄⁺ + ace or

NO₃⁻ + ace). Data are means with standard deviations ($n=3$). Different letters above the bars denote significant difference and the same letters denote no significant difference

soils. Amendment with NO₃⁻ alone did not significantly affect the change in exchangeable NH₄⁺-N concentration. No obvious difference in exchangeable NH₄⁺-N concentration was observed between NO₃⁻ and NO₃⁻ + oct treatments ($P > 0.05$).

In the two soils, the exchangeable NO₃⁻-N concentration in the control slightly increased due to nitrification of NH₃ derived from organic N mineralisation during the incubation (Fig. 2). 1-Octyne addition alone had no significant effect on soil exchangeable NO₃⁻-N concentration, but acetylene

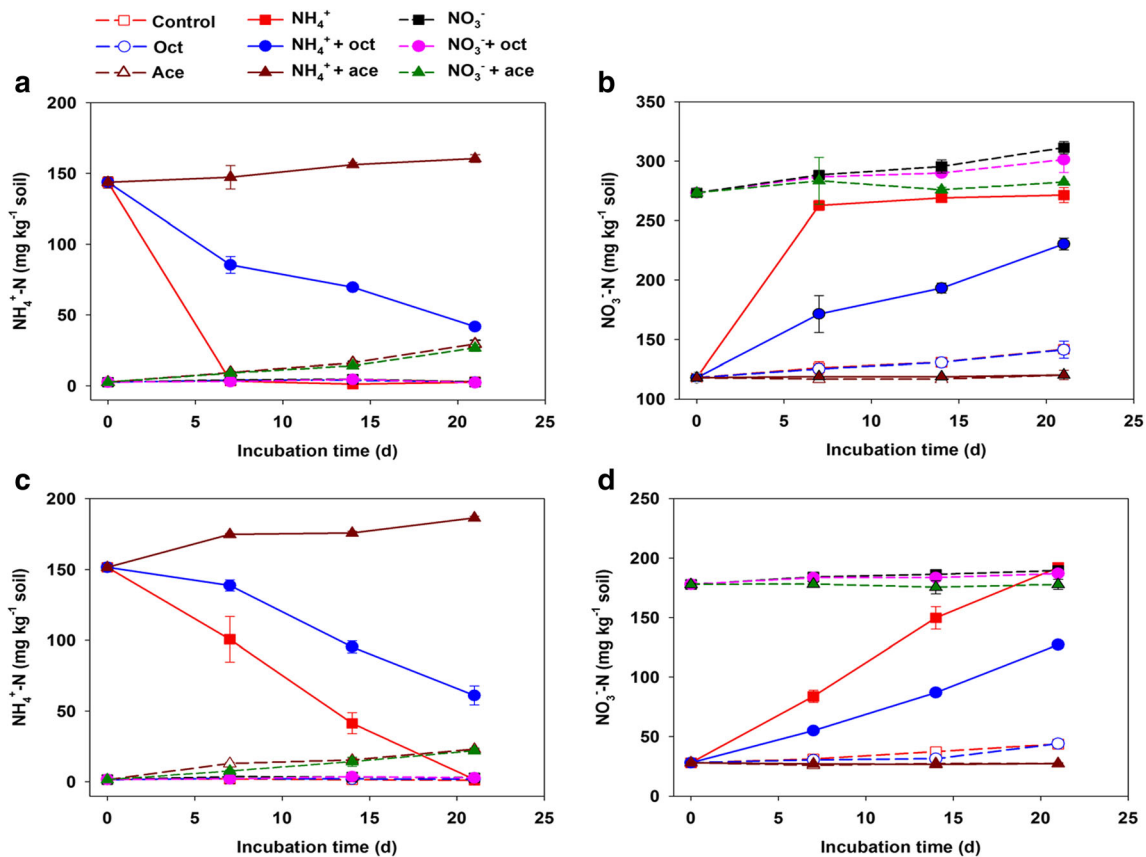


Fig. 2 The dynamics of NH₄⁺ and NO₃⁻ during incubation for 21 days in an alluvial soil (a, b) and a red soil (c, d) which were amended with distilled sterilised water, NH₄⁺ or NO₃⁻ (control, NH₄⁺ or NO₃⁻), with

1-octyne (Oct, NH₄⁺ + oct or NO₃⁻ + oct) or with acetylene (Ace, NH₄⁺ + ace or NO₃⁻ + ace). Data are means with standard deviations ($n=3$)

completely suppressed exchangeable NO_3^- -N production. Ammonium amendment substantially increased exchangeable NO_3^- -N concentrations ($P < 0.01$), whereas 1-octyne partly slowed exchangeable NO_3^- formation in the NH_4^+ + oct treatment (Fig. 2). Exchangeable NO_3^- -N concentrations in the NO_3^- and NO_3^- + oct treatments were similar after 21 days of incubation.

Abundance of AOB and AOA *amoA* genes

After 21 days, the AOB *amoA* gene abundance in the control was about 1.1×10^7 copies g^{-1} soil in the alluvial soil and 2.8×10^6 copies g^{-1} soil in the red soil, respectively (Fig. 3). No significant difference of AOB *amoA* gene abundance was observed among the control, Oct, Ace, NO_3^- , NO_3^- + oct and NO_3^- + ace treatments in the two soils ($P > 0.05$). The application of NH_4^+ significantly

increased AOB *amoA* gene abundance, reaching 3.1×10^7 copies g^{-1} soil in the alluvial soil and 7.2×10^6 copies g^{-1} soil in the red soil, respectively ($P < 0.01$). However, AOB *amoA* gene abundance in the NH_4^+ + oct and NH_4^+ + ace treatments was obviously lower than those in the NH_4^+ treatment. No significant variation of AOB *amoA* gene abundance was observed between NH_4^+ + oct and NH_4^+ + ace treatments ($P > 0.05$).

In the alluvial soil, AOA *amoA* gene abundance ranged from 1.7×10^8 to 1.9×10^8 copies g^{-1} soil, showing no significant difference among all the treatment groups ($P > 0.05$). In the red soil, AOA *amoA* gene abundance between the control and Oct treatments was similar, at around 3.1×10^7 copies g^{-1} dry soil. However, NH_4^+ or NO_3^- addition significantly reduced AOA *amoA* gene abundance, showing a possible inhibitory effect of inorganic N on the AOA populations in the acidic red soil.

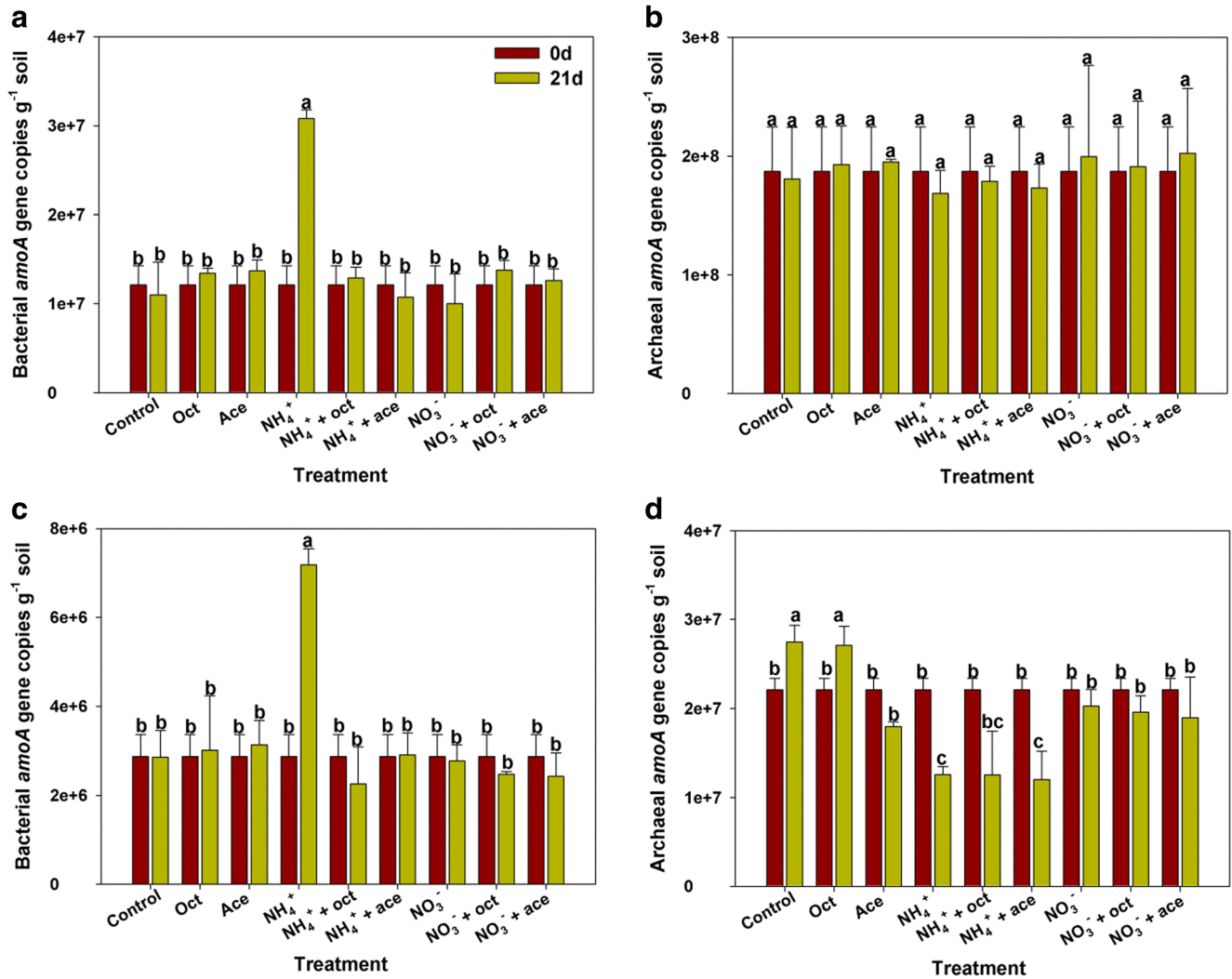


Fig. 3 AOB and AOA *amoA* genes abundance after 0 and 21 days of laboratory incubation in an alluvial soil (a, b) and a red soil (c, d) which were amended with distilled sterilised water, NH_4^+ or NO_3^- (control, NH_4^+ or NO_3^-), with 1-octyne (Oct, NH_4^+ + oct or NO_3^- + oct) or with

acetylene (Ace, NH_4^+ + ace or NO_3^- + ace). Data are means with standard deviations ($n = 3$). Different letters above the bars denote significant difference and the same letters denote no significant difference

The regression analysis between the AOB and AOA *amoA* gene abundance and cumulative N₂O emission from different treatments is illustrated in Fig. 4. Cumulative N₂O emissions were positively correlated with AOB abundance in the alluvial soil ($y = 1.46e^{0.17x}$, $R^2 = 0.98$, $P < 0.001$) and the red soil ($y = 3.06e^{0.048x}$, $R^2 = 0.85$, $P < 0.001$). However, AOA abundance was not correlated with cumulative N₂O emissions in either of the two soils.

The contribution of AOB and AOA to N₂O production in two arable soils

Octyne-sensitive (AOB) N₂O emissions varied significantly between soil types and NH₄⁺ amendment (Fig. 5). In the NH₄⁺ treatment, the fraction of octyne-sensitive (AOB) N₂O emission was much higher than that of octyne-resistant (AOA) N₂O emission in the two soils, and this effect depended on soil type. For example, octyne-sensitive (AOB) N₂O emission in the alluvial soil was 78.1 %, higher than in the red soil (70.5 %). The percentage of octyne-resistant (AOA) N₂O emission was 19.7 % in the alluvial soil and 18.7 % in the red soil, respectively. In unamended soils, no significant difference between the percentage of octyne-resistant (AOA)

N₂O emission and the fraction of octyne-sensitive (AOB) N₂O emission was observed (data not shown).

Discussion

Nitrification and denitrification are considered to be the two primary processes of N₂O production in arable soils (Wrage et al. 2001). It is commonly suggested that nitrification was the major process of N₂O emission in soils below 60–70 % WFPS, whereas denitrification would be the main source N₂O production above 70 % WFPS (Bateman and Baggs 2005; Braker and Conrad 2011; Ma et al. 2015). Here, we postulated that nitrification was the major source of N₂O emission because: (1) soil treated NH₄⁺ resulted in much higher N₂O production compared to that in NO₃⁻-treated soils; (2) applications of acetylene, which inhibits autotrophic ammonia oxidisers (AOB or AOA), reduced N₂O emission by 90–97 % in our soils, highlighting the importance of nitrifiers to N₂O emission; (3) all experiments were performed under aerobic conditions, and soil water content was kept at 55 % WFPS, which was more conducive to nitrification than denitrification (Bateman and Baggs 2005). Hence, these

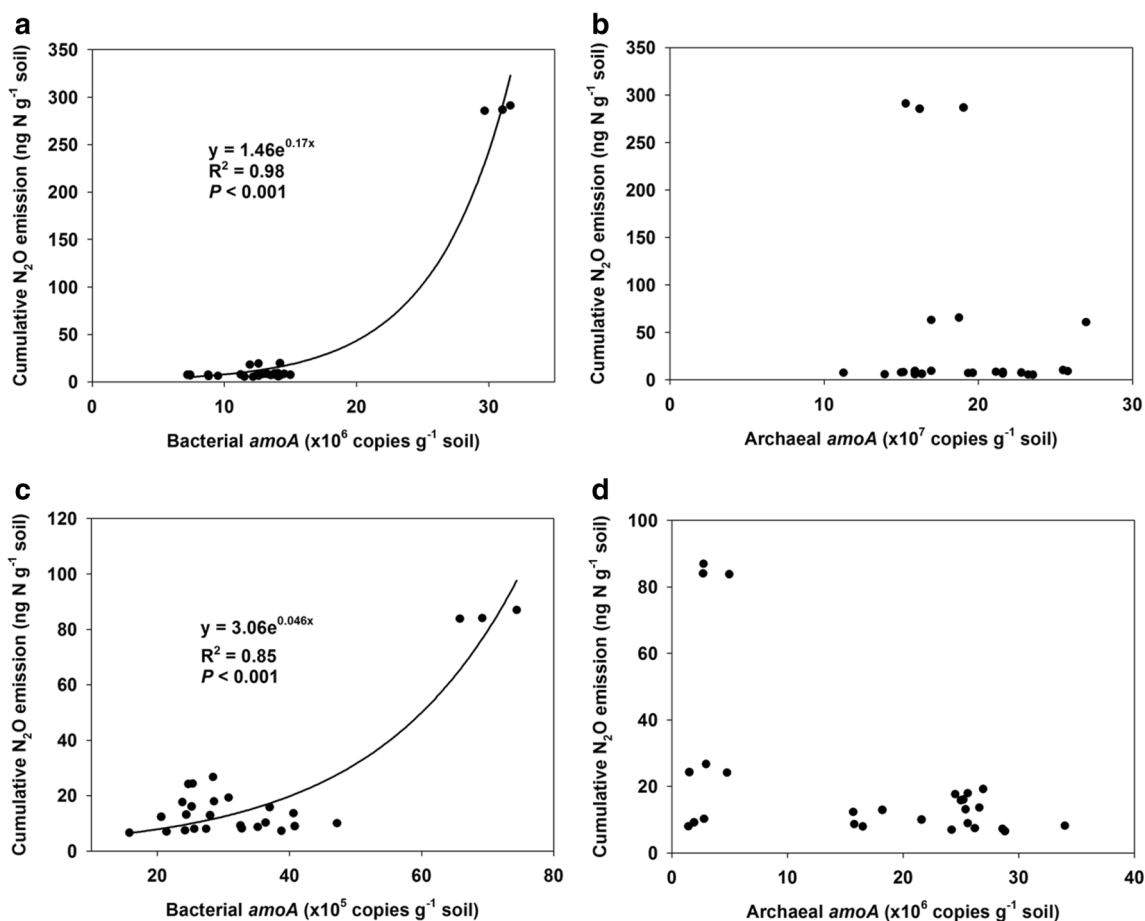
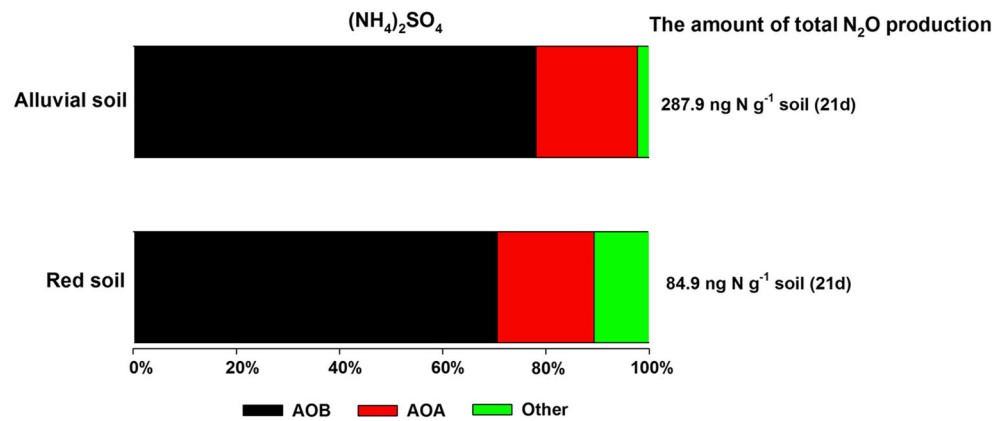


Fig. 4 Relationship of cumulative N₂O emissions with AOB (a, c) and AOA (b, d) *amoA* genes abundance in an alluvial soil (a, b) and a red soil (c, d)

Fig. 5 Relative contributions of AOB and AOA to N_2O emissions in the NH_4^+ treatment after 21 days of incubation in an alluvial soil and a red soil. The contributions related to AOB (black), AOA (red) and other microbes and processes (green) are shown as percentages of total N_2O production



results indicated that N_2O was mainly produced from nitrification in these two agricultural soils.

Clearly, soil treated with NH_4^+ strongly stimulated AOB growth rather than AOA in the two soils. AOA did not respond to the application of NH_4^+ in the alluvial soil, while it was inhibited with respect to the control in the red soil, thus confirming what it was obtained in agricultural soils (Wu et al. 2011; Ai et al. 2013; Habteselassie et al. 2013; Di et al. 2014, 2014; Chen et al. 2015). Generally, AOB growth is favored by high NH_4^+ -N soil conditions (Jia and Conrad 2009; Verhamme et al. 2011; Taylor et al. 2012), whereas AOA growth is not affected or suppressed (Di et al. 2009, 2010; Di and Cameron 2011; Di et al. 2014). Actually, the AOA population only grew in the control in the red soil, where NH_4^+ was generated from mineralised organic N, suggesting that AOA growth is favored in low fertility status soils (Nicol et al. 2008; Di et al. 2009, 2010; Zhang et al. 2012). Here, the stimulation and inhibition of AOB growth by NH_4^+ and inhibitors, respectively, occurred. Consequently, AOB *amoA* gene abundance was significantly correlated with cumulative N_2O emissions in the two soils. Similar studies also showed that AOB must play a critical role in N_2O production in high urea-N soils (Cui et al. 2013; Dai et al. 2013).

1-Octyne is a potentially selective inhibitor of AOB activity in agricultural soils (Taylor et al. 2013), and by using this selective inhibition, we showed that AOB had a great potential of contributing to soil N_2O production in high NH_4^+ -N agricultural soils despite the presence of abundant AOA in these soils. Two mainly N_2O -yielding can explain the reason for the higher N_2O production by AOB than AOA, NH_2OH oxidation and nitrifier denitrification (Shaw et al. 2006; Stein 2011; Schreiber et al. 2012). It was reported that N_2O emission from the oxidation of NH_2OH contributed very little to total N_2O production and nitrifier denitrification was believed to be the predominant process for N_2O production in soils with $50\% < WFPS < 70\%$ (Kool et al. 2010, 2011). Recently, Vajjala et al. (2013) found that NH_2OH was an intermediate of ammonia oxidation in AOA, but they would not be able to produce N_2O like AOB probably due to lack of

genes for a homologue of hydroxylamine oxidoreductase (HAO) known to be responsible for N_2O formation in AOB (Hooper and Terry 1979). In addition, AOA also lack genes encoding a potential NO-reductase (NOR), which is involved in nitrifier-denitrification and thus N_2O production in AOB (Stein 2011; Tourna et al. 2011). Therefore, we suggested that N_2O emission from the two tested soil probably depended on denitrification by AOB.

In addition, NI addition has limitations and the effectiveness of inhibitors in N_2O emission, and associated microbe studies may depend on soil conditions (Hatch et al. 2005). Stable isotope enrichment approaches should be applied to identify N_2O production during different processes following application of ^{15}N -labelled NH_4^+ or NO_3^- and ^{18}O -labelled H_2O and NO_3^- in short-term experiments (Cheng et al. 2014; Li and Lang 2014; Zhu et al. 2013). Zhu et al. (2013) carried out a dual labelling approach that employed an ^{18}O -, ^{15}N -enrichment method for distinguishing among nitrous oxide (N_2O) production from nitrifier nitrification, nitrifier denitrification and heterotrophic denitrification. A dual labelling method may improve our ability to discriminate between the contributions of nitrification and denitrification to total N_2O production. A combination of the stable isotopic enrichment approach coupled with molecular method may be required to estimate the contributions of all known N_2O -genic processes in the future.

In conclusion, we compared relative bacterial and archaeal contribution to soil N_2O production from two contrasting arable soils. By using inhibitors for specific microbial groups and metabolic pathways, we demonstrated that nitrification was the predominant driving process of N_2O emission and AOB dominated in N_2O production when NH_4^+ was applied in the alluvial and red soils. The relative contribution of N_2O emission from AOA is lower than that from AOB under this condition. Moreover, we also observed that the changes of N_2O emissions among all treatments were correlated with variation of AOB abundance but not with AOA abundance in the two tested arable soils. Therefore, AOB may play a more important role than AOA in N_2O emission in a high NH_4^+ -N concentration arable soils.

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References

- Ai C, Liang GQ, Sun JW, Wang XB, He P, Zhou W (2013) Different roles of rhizosphere effect and long-term fertilization in the activity and community structure of ammonia oxidizers in a calcareous fluvo-aquic soil. *Soil Biol Biochem* 57:30–42
- Bateman EJ, Baggs EM (2005) Contributions of nitrification and denitrification to N₂O emissions from soils at different water-filled pore space. *Biol Fertil Soils* 41:379–388
- Braker G, Conrad R (2011) Diversity, structure, and size of N₂O-producing microbial communities in soils—what matters for their functioning? *Adv Appl Microbiol* 75(75):33–70
- Chen X, Zhang LM, Shen JP, Xu ZH, He JZ (2010) Soil type determines the abundance and community structure of ammonia-oxidizing bacteria and archaea in flooded paddy soils. *J Soils Sediments* 10:1510–1516
- Chen QH, Qi LY, Bi QF, Dai PB, Sun DS, Sun CL, Liu WJ, Lu LL, Ni WZ, Lin XY (2015) Comparative effects of 3,4-dimethylpyrazole phosphate (DMPP) and dicyandiamide (DCD) on ammonia-oxidizing bacteria and archaea in a vegetable soil. *Appl Microbiol Biotechnol* 99:477–487
- Cheng Y, Wang J, Wang SQ, Zhang JB, Cai ZC (2014) Effects of soil moisture on gross N transformations and N₂O emission in a subtropical forest soils. *Bio Fertil Soils* 50:1099–1108
- Cui PY, Fan FL, Yin C, Li Z, Song AL, Wan Y, Liang YC (2013) Urea- and nitrapyrin-affected N₂O emission is coupled mainly with ammonia oxidizing bacteria growth in microcosms of three typical Chinese arable soils. *Soil Biol Biochem* 66:214–221
- Cui PY, Fan FL, Yin C, Song AL, Huang PR, Tang YJ, Zhu P, Peng C, Li T, Wakelin SA, Liang YC (2016) Long-term organic and inorganic fertilization alters temperature sensitivity of potential N₂O emissions and associated microbes. *Soil Biol Biochem* 93:131–141
- Dai Y, Di HJ, Cameron KC, He JZ (2013) Effects of nitrogen application rate and a nitrification inhibitor dicyandiamide on ammonia oxidizers and N₂O emissions in a grazed pasture soil. *Sci Total Environ* 465:125–135
- Davidson EA (2009) The contribution of manure and fertilizer nitrogen to atmospheric nitrous oxide since 1860. *Nat Geosci* 2:659–662
- Di HJ, Cameron KC (2011) Inhibition of ammonium oxidation by a liquid formulation of 3,4-Dimethylpyrazole phosphate (DMPP) compared with a dicyandiamide (DCD) solution in six new Zealand grazed grassland soils. *J Soils Sediments* 11:1032–1039
- Di HJ, Cameron KC, Shen JP, Winefield CS, O’Callaghan M, Bowatte S, He JZ (2009) Nitrification driven by bacteria and not archaea in nitrogen-rich grassland soils. *Nat Geosci* 2:621–624
- Di HJ, Cameron KC, Sherlock RR, Shen JP, He JZ, Winefield CS (2010) Nitrous oxide emissions from grazed grassland as affected by a nitrification inhibitor, dicyandiamide, and relationships with ammonia-oxidizing bacteria and archaea. *J Soils Sediments* 10:943–954
- Di HJ, Cameron KC, Podolyan A, Robinson A (2014) Effect of soil moisture status and a nitrification inhibitor, dicyandiamide, on ammonia oxidizer and denitrifier growth and nitrous oxide emissions in a grassland soil. *Soil Biol Biochem* 73:59–68
- Gdde M, Conrad R (1999) Immediate and adaptational temperature effects on nitric oxide production and nitrous oxide release from nitrification and denitrification in two soils. *Biol Fertil Soils* 30:33–40
- Gubry-Rangin C, Nicol GW, Prosser JI (2010) Archaea rather than bacteria control nitrification in two agricultural acidic soils. *FEMS Microbiol Ecol* 74:566–574
- Habteselassie MY, Xu L, Norton JM (2013) Ammonia-oxidizer communities in an agricultural soil treated with contrasting nitrogen sources. *Front Microbiol* 4:326
- Hatch D, Trindade H, Cardenas L, Carneiro J, Hawkins J, Scholefield D, Chadwick D (2005) Laboratory study of the effects of two nitrification inhibitors on greenhouse gas emissions from a slurry-treated arable soil: impact of diurnal temperature cycle. *Biol Fertil Soils* 41:225–232
- He JZ, Shen JP, Zhang LM, Zhu YG, Zheng YM, Xu MG, Di H (2007) Quantitative analyses of the abundance and composition of ammonia-oxidizing bacteria and ammonia-oxidizing archaea of a Chinese upland red soil under long-term fertilization practices. *Environ Microbiol* 9:2364–2374
- Hink L, Nicol GW, Prosser JI (2016) Archaea produce lower yields of N₂O than bacteria during aerobic ammonia oxidation in soil. *Environ Microbiol*. doi:10.1111/1462-2920.13282
- Hooper AB, Terry KR (1979) Hydroxylamine oxidoreductase of *Nitrosomonas* production of nitric-oxide from hydroxylamine. *Bioch Et Bioph Act* 571:12–20
- Hu HW, Zhang LM, Yuan CL, He JZ (2013) Contrasting Euryarchaeota communities between upland and paddy soils exhibited similar pH-impacted biogeographic patterns. *Soil Biol Biochem* 64:18–27
- Hu HW, Macdonald CA, Trivedi P, Holmes B, Bodrossy L, He JZ, Singh BK (2015) Water addition regulates the metabolic activity of ammonia oxidizers responding to environmental perturbations in dry sub-humid ecosystems. *Environ Microbiol* 17:444–461
- Huang T, Gao B, Hu XK, Lu X, Well R, Christie P, Bakken LR, Ju XT (2014a) Ammonia-oxidation as an engine to generate nitrous oxide in an intensively managed calcareous fluvo-aquic soil. *Sci Rep* 4:3950
- Huang Y, Li YY, Yao HY (2014b) Nitrate enhances N₂O emission more than ammonium in a highly acidic soil. *J Soils Sediments* 14:146–154
- Hynes RK, Knowles R (1978) Inhibition by acetylene of ammonia oxidation in *Nitrosomonas-Europaea*. *FEMS Microbiol Lett* 4:319–321
- Jia ZJ, Conrad R (2009) Bacteria rather than archaea dominate microbial ammonia oxidation in an agricultural soil. *Environ Microbiol* 11:1658–1671
- Jung MY, Park SJ, Min D, Kim JS, Rijpstra WIC, Damste JSS, Kim GJ, Madsen EL, Rhee SK (2011) Enrichment and characterization of an autotrophic ammonia-oxidizing archaeon of mesophilic crenarchaeal group I.1a from an agricultural Soil. *Appl Environ Microbiol* 77:8635–8647
- Jung MY, Well R, Min D, Giesemann A, Park SJ, Kim JG, Kim SJ, Rhee SK (2014) Isotopic signatures of N₂O produced by ammonia-oxidizing archaea from soils. *ISME J* 8:1115–1125
- Kool DM, Wrage N, Zechmeister-Boltenstern S, Pfeffer M, Brus D, Oenema O, Van Groenigen JW (2010) Nitrifier denitrification can be a source of N₂O from soil: a revised approach to the dual-isotope labelling method. *Eur J Soil Sci* 61:759–772
- Kool DM, Dolfing J, Wrage N, Van Groenigen JW (2011) Nitrifier denitrification as a distinct and significant source of nitrous oxide from soil. *Soil Biol Biochem* 43:174–178
- Leininger S, Urich T, Schloter M, Schwark L, Qi J, Nicol GW, Prosser JI, Schuster SC, Schleper C (2006) Archaea predominate among ammonia-oxidizing prokaryotes in soils. *Nature* 442:806–809
- Li P, Lang M (2014) Gross nitrogen transformations and related N₂O emissions in uncultivated and cultivated black soil. *Biol Fertil Soils* 50:197–20
- Loscher CR, Kock A, Konneke M, LaRoche J, Bange HW, Schmitz RA (2012) Production of oceanic nitrous oxide by ammonia-oxidizing archaea. *Biogeosciences* 9:2419–2429

- Lu XD, Bottomley PJ, Myrold DD (2015) Contributions of ammonia-oxidizing archaea and bacteria to nitrification in Oregon forest soils. *Soil Biol Biochem* 85:54–62
- Lund MB, Smith JM, Francis CA (2012) Diversity, abundance and expression of nitrite reductase (*nirK*)-like genes in marine thaumarchaea. *ISME J* 6:1966–1977
- Ma J, Ma E, Xu H, Yagi K, Cai ZC (2009) Wheat straw management affects CH₄ and N₂O emissions from rice fields. *Soil Biol Biochem* 41:1022–1028
- Ma L, Shan J, Yan XY (2015) Nitrite behavior accounts for the nitrous oxide peaks following fertilization in a fluvo-aquic soil. *Biol Fert soils* 51:563–572
- Nicol GW, Leininger S, Schleper C, Prosser JI (2008) The influence of soil pH on the diversity, abundance and transcriptional activity of ammonia oxidizing archaea and bacteria. *Environ Microbiol* 10:2966–2978
- Norton JM, Stark JM (2011) Regulation and measurement of nitrification in terrestrial systems. *Methods Enzymol* 486:343–368
- Offre P, Prosser JI, Nicol GW (2009) Growth of ammonia-oxidizing archaea in soil microcosms is inhibited by acetylene. *FEMS Microbiol Ecol* 70:99–108
- Ouyang Y, Norton JM, Stark JM, Reeve JR, Habteselassie MY (2016) Ammonia-oxidizing bacteria are more responsive than archaea to nitrogen source in an agricultural soil. *Soil Biol Biochem* 96:4–15
- Rothauwe JH, Witzel KP, Liesack W (1997) The ammonia monooxygenase structural gene *amoA* as a functional marker: Molecular fine-scale analysis of natural ammonia-oxidizing populations. *Appl Environ Microbiol* 63:4704–4712
- Santoro AE, Buchwald C, McIlvin MR, Casciotti KL (2011) Isotopic signature of N₂O produced by marine ammonia-oxidizing archaea. *Science* 333:1282–1285
- Schreiber F, Wunderlin P, Udert KM, Wells GF (2012) Nitric oxide and nitrous oxide turnover in natural and engineered microbial communities: biological pathways, chemical reactions, and novel technologies. *Front Microbiol* 3:372
- Shaw LJ, Nicol GW, Smith Z, Fear J, Prosser JI, Baggs EM (2006) *Nitrosospora* spp. can produce nitrous oxide via a nitrifier denitrification pathway. *Environ Microbiol* 8:214–222
- Shcherbak I, Millar N, Robertson GP (2014) Global meta-analysis of the nonlinear response of soil nitrous oxide (N₂O) emissions to fertilizer nitrogen. *Proc Natl Acad Sci U S A* 111:9199–9204
- Smith KA, Mosier AR, Crutzen PJ, Winiwarter W (2012) The role of N₂O derived from crop-based biofuels, and from agriculture in general, in Earth's climate. *Phil Trans R Soc B* 367:1169–1174
- Stein LY (2011) Surveying N₂O-producing pathways in bacteria. In: Klotz MG (ed) *Methods in enzymology*, vol 486. Academic Press, CA, pp 131–152
- Stergren AE, Hallin S, Bengtson P (2015) Archaeal ammonia oxidizers dominate in numbers, but bacteria drive gross nitrification in N-amended grassland soil. *Front Microbiol* 6:1350
- Stieglmeier M, Mooshammer M, Kitzler B, Wanek W, Zechmeister-Boltenstern S, Richter A, Schleper C (2014) Aerobic nitrous oxide production through N-nitrosating hybrid formation in ammonia-oxidizing archaea. *ISME J* 8:1135–1146
- Taylor AE, Zeglin LH, Wanzek TA, Myrold DD, Bottomley PJ (2012) Dynamics of ammonia-oxidizing archaea and bacteria populations and contributions to soil nitrification potentials. *ISME J* 6:2024–2032
- Taylor AE, Vajrala N, Giguere AT, Gitelman AI, Arp DJ, Myrold DD, Sayavedra-Soto L, Bottomley PJ (2013) Use of aliphatic n-Alkynes to discriminate soil nitrification activities of ammonia-oxidizing thaumarchaea and bacteria. *Appl Environ Microbiol* 79:6544–6551
- Taylor AE, Taylor K, Tennigkeit B, Palatinszky M, Stieglmeier M, Myrold DD, Schleper C, Wagner M, Bottomley PJ (2015) Inhibitory effects of C-2 to C-10 1-Alkynes on ammonia oxidation in two *Nitrososphaera* Species. *Appl Environ Microbiol* 81:1942–1948
- Tourna M, Stieglmeier M, Spang A, Konneke M, Schintlmeister A, Ulrich T, Engel M, Schloter M, Wagner M, Richter A, Schleper C (2011) *Nitrososphaera viennensis*, an ammonia oxidizing archaeon from soil. *Proc Natl Acad Sci U S A* 108:8420–8425
- USDA (1994) Keys to soil taxonomy. United States Department of Agriculture, Soil Conservation Service, 6th edn., p 306
- Vajrala N, Martens-Habbena W, Sayavedra-Soto LA, Schauer A, Bottomley PJ, Stahl DA, Arp DJ (2013) Hydroxylamine as an intermediate in ammonia oxidation by globally abundant marine archaea. *Proc Natl Acad Sci U S A* 110:1006–1011
- Venter JC, Remington K, Heidelberg JF, Halpern AL, Rusch D, Eisen JA, Wu DY, Paulsen I, Nelson KE, Nelson W, Fouts DE, Levy S, Knap AH, Lomas MW, Nealson K, White O, Peterson J, Hoffman J, Parsons R, Baden-Tillson H, Pfannkoch C, Rogers YH, Smith HO (2004) Environmental genome shotgun sequencing of the Sargasso Sea. *Science* 304:66–74
- Verhamme DT, Prosser JI, Nicol GW (2011) Ammonia concentration determines differential growth of ammonia-oxidising archaea and bacteria in soil microcosms. *ISME J* 5:1067–1071
- Wrage N, Velthof GL, van Beusichem ML, Oenema O (2001) Role of nitrifier denitrification in the production of nitrous oxide. *Soil Biol Biochem* 33:1723–1732
- Wu YC, Lu L, Wang BZ, Lin XG, Zhu JG, Cai ZC, Yan XY, Jia ZJ (2011) Long-term field fertilization significantly alters community structure of ammonia-oxidizing bacteria rather than archaea in a paddy soil. *Soil Sci Soc Am J* 75:1431–1439
- Yao H, Gao Y, Nicol GW, Campbell CD, Prosser JI, Zhang L, Han W, Singh BK (2011) Links between ammonia oxidizer community structure, abundance, and nitrification potential in acidic soils. *Appl Environ Microbiol* 77:4618–4625
- Zhang LM, Hu HW, Shen JP, He JZ (2012) Ammonia-oxidizing archaea have more important role than ammonia-oxidizing bacteria in ammonia oxidation of strongly acidic soils. *ISME J* 6:1032–1045
- Zhu X, Burger M, Doane TA, Horwath WR (2013) Ammonia oxidation pathways and nitrifier denitrification are significant sources of N₂O and NO under low oxygen availability. *Proc Natl Acad Sci U S A* 110:6328–6333