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# Effects of moisture and temperature on greenhouse gas emissions and C and N leaching losses in soil treated with biogas slurry

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Abstract The objective of this study was to examine the effects of soil moisture, irrigation pattern, and temperature on gaseous and leaching losses of carbon (C) and nitrogen (N) from soils amended with biogas slurry (BS). Undisturbed soil cores were amended with BS (33 kg N  $ha^{-1}$ ) and incubated at 13.5°C and 23.5°C under continuous irrigation (2 mm day<sup>-1</sup>) or cycles of strong irrigation and partial drying (every 6 weeks, 1 week with 12 mm day<sup>-1</sup>). During the 6 weeks after BS application, on average, 30% and 3.8% of the C and N applied with BS were emitted as carbon dioxide  $(CO_2)$ and nitrous oxide (N<sub>2</sub>O), respectively. Across all treatments, a temperature increase of 10°C increased N<sub>2</sub>O and CO<sub>2</sub> emissions by a factor of 3.7 and 1.7, respectively. The irrigation pattern strongly affected the temporal production of CO<sub>2</sub> and N<sub>2</sub>O but had no significant effect on the cumulative production. Nitrogen was predominantly lost in the form of nitrate (NO<sub>3</sub>). On average, 16% of the N applied was lost as NO<sub>3</sub><sup>-</sup>. Nitrate leaching was significantly increased at the higher temperature (P < 0.01), while the irrigation pattern had no effect (P=0.63). Our results show that the C and N turnovers were strongly affected by BS application and soil temperature whereas irrigation pattern had only minor effects. A considerable proportion of the C and N in BS were readily available for soil microorganisms.

**Keywords** Biogas slurry · Nitrous oxide emissions · Methane oxidation · Nitrate leaching · Temperature · Soil moisture

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#### Introduction

Agriculture accounted for 10-12% of total global anthropogenic emissions of greenhouse gases in 2005 (Smith et al. 2007). The most important greenhouse gases released from agricultural land are methane (CH<sub>4</sub>), nitrous oxide  $(N_2O)$ , and carbon dioxide  $(CO_2)$  which contribute 6.5%, 5.5%, and 0.1% to total global anthropogenic greenhouse gas emissions, respectively. The net flux of CO<sub>2</sub> related to agriculture is estimated to be approximately balanced whereas agriculture contributes about 60% and 50% of the global anthropogenic N<sub>2</sub>O and CH<sub>4</sub> emissions, respectively (Smith et al. 1998, 2007). Due to concerns about the greenhouse effect, renewable energy sources, such as biogas production from energy crops, are becoming increasingly important. As maize has one of the highest CH<sub>4</sub> yields per hectare, its use for biogas production is likely to increase in the future (Amon et al. 2007) and therefore the availability of maize-derived biogas slurry (BS) as an organic fertilizer. However, the use of BS from energy crops as fertilizer and the effect of field applications on greenhouse gas emissions have not received much attention yet. Moeller and Stinner (2009) compared the N<sub>2</sub>O emissions of clover/grass-ley when mulched and incorporated as green manure with the N2O emissions resulting from the field application of BS made from the same clover/grass-ley. The field applications of BS resulted in a large increase in N<sub>2</sub>O emissions, but it was 38% less than that from the mulched and incorporated clover/grass-ley. Senbayram et al. (2009) reported that soils treated with BS derived from maize emitted more N<sub>2</sub>O than soils treated with mineral nitrogen (N) fertilizer. In a laboratory study, they found that 2.6% of the applied BS-N was emitted as N<sub>2</sub>O.

In soil, N<sub>2</sub>O is produced through denitrification and nitrification, both of which are mediated by microorganisms. While nitrification takes place under aerobic conditions, denitrification predominantly occurs under anaerobic conditions (Dittert and Mühling 2009). Nitrification and denitrification rates have been found to be closely related to the water-filled pore space (WFPS) of a soil. Nitrification rates are generally highest when soil moisture is below field capacity and nearly stops in saturated soils due to lack of oxygen  $(O_2)$ , whereas denitrification rates generally increase with increasing soil moisture content (Davidson 1992; Maag and Vinther 1996). Bateman and Baggs (2005) found that all of the  $N_2O$  emitted from a silt loam at 70% WFPS was produced during denitrification, while at 35-60% WFPS nitrification was the main process producing N<sub>2</sub>O. Besides soil moisture content and temperature, the availability of substrates, ammonium  $(NH_4^+)$  for nitrification and nitrate (NO<sub>3</sub><sup>-</sup>) as well as available carbon (C) for denitrification, affects the rates of N<sub>2</sub>O emissions (Smith et al. 1998). In general, anaerobic fermentation increases the  $NH_4^+$  content in the substrate as well as the stability of organic matter but decreases the C to N ratio considerably (Gutser et al. 2005), resulting in a product with a high content of directly available N. However, during anaerobic digestion for CH<sub>4</sub> production, volatile fatty acids and other labile organic compounds are formed as intermediates (Cysneiros et al. 2008; Jacobi et al. 2009). These compounds, when still present in the BS, may be readily available C sources for soil microorganisms.

The aim of this study was to investigate the effects of soil moisture, irrigation pattern, and temperature on gaseous emissions (CO<sub>2</sub>, N<sub>2</sub>O, and CH<sub>4</sub>) and leaching losses of C and N compounds from undisturbed soil cores in an incubation experiment.

# Materials and methods

#### Soil and biogas slurry

Soil was sampled in March 2009 at the long-term experimental site "Hohes Feld" near Göttingen in Lower Saxony, Germany. The preceding crop was winter wheat, harvested in July 2008. In August 2008, the field was tilled with a rotary harrow. During winter, the field lay fallow. The mean annual precipitation at the site is 645 mm, and the mean annual temperature is 8.7°C (Deutscher Wetterdienst 2009). The soil type is a Haplic Luvisol (FAO 1998) derived from loess (Ehlers et al. 2000). The soil sampled had sand, silt, and clay contents of 15%, 67%, and 18%, respectively, a bulk density of 1.42 gcm<sup>-3</sup>, and a pH of 6.9. The total carbon (C<sub>t</sub>) content gradually decreased with depth in the plow layer (26 cm) from 13.3 g kg<sup>-1</sup> in the top 9 cm to 8.2 g kg<sup>-1</sup> dry soil in the lowest 9 cm of the profile. The C to N ratio ranged from 9.5 to 10.0 and was not affected by depth. When the incubation started (see below), the microbial biomass C was  $328 \text{ mg kg}^{-1}$  in the top 9 cm of the profile and  $125 \text{ mg kg}^{-1}$  in the lower part of the plow layer (the methods used are described below).

Undisturbed soil cores were sampled using plexiglass cylinders with an inner diameter of 15 cm. The cylinders were inserted into the soil to a depth of 26 cm and dug out with a spade. The soil cores remained in the cylinders for the duration of the incubation.

BS was obtained from the Institute for Anaerobic Technology at the University of Applied Sciences of Giessen. The BS is a by-product of the anaerobic digestion (4 months at 55°C) of corn silage for CH<sub>4</sub> production in a laboratory-scale reactor. The BS had a dry matter content of 8.4% and a pH of 8.9. The C and N contents in the dry matter were 41% and 5.1%, respectively, resulting in a C to N ratio of 8. About 24% of the N was in the form of  $NH_4^+$ , while no  $NO_3^-$  was detectable.

#### Incubation

The study was carried out in a climate chamber equipped with an automated microcosm system (Hantschel et al. 1994). The cores were placed on a ceramic plate with a 1-µm pore diameter to which a constant suction of 100 hPa was applied. Prior to the incubation, the cores were conditioned at 13.5°C for 6 months and irrigated with 2 mm  $day^{-1}$  of a 0.01 M CaCl<sub>2</sub> solution adjusted to a pH of 5.5. After the conditioning phase, the cores were randomly assigned to one of the treatments (see below). All cores were incubated for 30 weeks. During the first 16 weeks of the experiment, the temperature was maintained at 13.5°C, while it was increased to 23.5°C for weeks 17 to 30. Three factors were investigated, namely (1) BS application (cores amended with BS and unamended controls), (2) temperature (13.5°C and 23.5°C), and (3) irrigation treatment. Three irrigation treatments were included:

- 1. Continuous irrigation (2 mm day $^{-1}$ ),
- Cycles of strong irrigation and partial drying (1 week with an irrigation rate of 12 mm day<sup>-1</sup> was followed by 5 weeks without irrigation) with BS application immediately before a period of strong irrigation, and
- 3. Cycles of strong irrigation and partial drying with BS application 1 week after the end of a period of strong irrigation.

In all treatments, the cumulative irrigation over any 6-week period was 84 mm. Separate cores were used for BS applications at the two temperatures. To the cores fertilized at 13.5°C, BS was applied during week 5 when applied before the period of strong irrigation or

week 8 when applied after the period of strong irrigation. To the cores fertilized after the temperature rise, BS was applied during week 17 or 20. Four cores per treatment were incubated. BS was applied at a rate of 33 kg total N per hectare. For this, a 3–4 cm-thick layer of the topsoil was removed, the slurry, suspended in 50 mL of 0.01 M CaCl<sub>2</sub> solution, added to the cores, and the topsoil placed back on top. The control cores were treated identically, except that 50 mL of 0.01 M CaCl<sub>2</sub> solution without BS was added. The irrigation was reduced by the amount of liquid added in the fertilized cores. During the incubation, the soil cores were hermetically sealed, and fresh air flow through the headspace was 20 ml min<sup>-1</sup>.

Leachates were collected weekly, weighed, and analyzed for concentrations of  $NO_3^-$ ,  $NH_4^+$ , total organic C (TOC), and total dissolved N (TDN). Every 3.7 h, a gas sample was analyzed for CO<sub>2</sub>, CH<sub>4</sub>, and N<sub>2</sub>O. Soil moisture content was monitored in two cores per treatment with soil moisture probes (ECH2O EC-5; Decagon Devises, Inc., Pullman, WA, USA). Measurements were taken every 30 min. The soil moisture probes were placed at a depth of 5–6 cm. During week 18 and at the end of the incubation, soil moisture content in the topsoil of all cores was determined using a ThetaProbe Soil Moisture Sensor-ML2x (Delta-T Devices Ltd., Cambridge, UK). To calibrate the soil moisture probes, gravimetric water content was determined in the top 5 cm of each core at the end of the experiment by drying a sample at 105°C for 24 h.

### Chemical analyses

Two soil cores were used for determination of soil characteristics after the conditioning phase. Soil moisture content was determined by drying at 105°C for 24 h. The pH was measured in a 0.01 M CaCl<sub>2</sub> solution (2.5 ml solution per gram of fresh soil) and particle size distribution with the pipet method according to DIN ISO 11277 (2002). Soil microbial biomass was determined with the chloroform fumigation extraction method (Vance et al. 1987). Total contents of C and N were determined by dry combustion on a CN Elemental Analyzer (Elementar Vario El, Heraeus, Hanau, Germany). The same methods were used for BS analyses, except that total N was analyzed by the Kjeldahl method on a Büchi 323 (Büchi Labortechnik, Essen, Germany). Ammonium and NO<sub>3</sub><sup>-</sup> were extracted from the BS with 0.5 M K<sub>2</sub>SO<sub>4</sub> (20 ml per gram of fresh BS) and determined on a continuous flow analyzer (Evolution II auto-analyzer, Alliance Instruments, Salzburg, Austria).

Gaseous emissions of  $CO_2$ ,  $N_2O$ , and  $CH_4$  were measured by a gas chromatograph (Shimadzu GC-14, Shimadzu Scientific Instruments, Columbia, USA). Nitrate and  $NH_4^+$  concentrations in the leachates were determined on a continuous flow analyzer. TDN and TOC in the leachates were analyzed using a DIMATOC<sup>®</sup> 2000 (Dimatec Analysentechnik, Essen, Germany). Several times during the incubation, subsamples of the leachate were filtered through a 0.45- $\mu$ m polyamide filter before analysis to determine the dissolved organic C content.

#### Statistical analysis

Statistical analyses were performed using SPSS 14.0.1 (SPSS 2005). The data were analyzed as a three-way ANOVA with temperature, fertilizer treatment, and irrigation pattern as factors. Mean comparisons were performed using the Tukey test. Effects were considered significant for P < 0.05. In order to avoid confounding effects due to differences in irrigation water application rates, comparisons of gaseous emissions from the different treatments were made for the 6-week period following the application of BS, as the amount of irrigation water applied was the same for all soil cores during any 6-week period. Effects of BS application on leaching losses became first evident after 4 weeks. Leaching losses from the different treatment were therefore compared for the 6-week period between weeks 4 and 10 after BS applications.

#### Results

### Soil moisture

The gravimetric moisture content in the upper 6 cm of the cores with continuous irrigation averaged 27.5%, which corresponds to an average WFPS of 84% in the soil core (data not shown). In the treatments with cycles of strong irrigation and partial drying, the gravimetric moisture content in the upper 6 cm reached 29.5% during periods of strong irrigation and dropped to 26.5% during the 5 weeks without irrigation. These gravimetric moisture contents correspond to 90% and 81% WFPS in the soil core during periods of strong irrigation and partial drying, respectively. As the topsoil was loosened to incorporate the BS, the WFPS may have been lower in the topsoil after the BS application than in the rest of the core. Neither the increase in temperature after 16 weeks nor the application of BS had a significant effect on the gravimetric moisture content (data not shown).

# Carbon dioxide production

At 13.5°C, CO<sub>2</sub> emission rates from the continuously irrigated control soils averaged 20 mg CO<sub>2</sub>-C m<sup>-2</sup> h<sup>-1</sup> and decreased slightly over time (Fig. 1a). The temperature rise resulted in a steep and immediate increase in CO<sub>2</sub>-C

Fig. 1 Average CO<sub>2</sub> emissions from soil cores amended with BS and unfertilized cores (control). The treatments included continuous irrigation (a, d), cycles of strong irrigation and partial drying with BS application before a period of strong irrigation (b, e), and cycles of strong irrigation and partial drying with BS application after a period of strong irrigation (c, f). The undisturbed soil cores were incubated at 13.5°C and 23.5°C. Arrows indicate the time of BS application. Each line represents the average of four cores. The standard error of the mean (n=4)is shown for the first measurement each week



production rates up to 49 mg m<sup>-2</sup> h<sup>-1</sup> for 1 week, and average emission rates at 23.5°C were increased to 33 mg  $CO_2$ -C m<sup>-2</sup> h<sup>-1</sup> (Fig. 1d). Therefore, the  $Q_{10}$  factor for  $CO_2$ emissions was 1.7. The application of BS increased  $CO_2$ emission rates at both temperatures for a few weeks. However, after 3 weeks (when BS was applied at 13.5°C) to 7 weeks (at 23.5°C), the  $CO_2$  emissions had dropped to the level of the control. The amount of applied C, which evolved as  $CO_2$  during the 6 weeks following BS application, was 13% and 34% at 13.5°C and 23.5°C, respectively (Fig. 2).

At 13.5°C, the average  $CO_2$  production of control soils in the treatments with cycles of strong irrigation and partial drying was 23 mg  $CO_2$ -C m<sup>-2</sup> h<sup>-1</sup>, with emissions averaging 25 mg CO<sub>2</sub>-C m<sup>-2</sup> h<sup>-1</sup> during partial drying periods and 14 mg CO<sub>2</sub>-C m<sup>-2</sup> h<sup>-1</sup> during wetter periods (Fig. 1b–c). Similar to the soils under continuous irrigation, the temperature rise led to an increase in average CO<sub>2</sub> production rates to 34 mg m<sup>-2</sup> h<sup>-1</sup> (Fig. 1e–f), corresponding to a  $Q_{10}$  factor of 1.5.

When BS was applied directly before a week of stronger irrigation, the CO<sub>2</sub>-C emissions slightly decreased during the period of stronger irrigation but increased considerably afterwards compared to the control (Fig. 1b, e). During subsequent periods of stronger irrigation, fertilization had a positive effect on CO<sub>2</sub> emissions only at the higher temperature (Fig. 1e). The amount of applied C which evolved as CO<sub>2</sub> within 6 weeks after BS application was



**Fig. 2** Proportion of BS C and N lost from soil cores incubated under *continuous* irrigation, and cycles of strong irrigation and partial drying with slurry applied before (*periodic strong irrigation-1*) and after a period of strong irrigation (*periodic strong irrigation-2*). The undisturbed soil cores were incubated at 13.5°C and 23.5°C. Gaseous losses include CO<sub>2</sub> and N<sub>2</sub>O, and losses in leachate include TOC, NO<sub>3</sub><sup>-</sup>-N, and N<sub>org</sub>. The difference between TDN and NO<sub>3</sub><sup>-</sup>-N corresponded to N<sub>org</sub>. *Negative values* for N<sub>org</sub> indicate that the losses in the BS-amended cores were lower than in the control

17% and 58% at 13.5°C and 23.5°C, respectively, when BS was applied directly before a week of stronger irrigation. BS applications 1 week after a period of strong irrigation also increased CO<sub>2</sub> emission rates during periods without irrigation and had little effect during periods of stronger irrigation (Fig. 1c, f). The CO<sub>2</sub>, which evolved within 6 weeks after BS application, accounted for 21% and 38% of the C applied with the BS at the lower and higher temperature, respectively (Fig. 2).

Across all treatments, differences due to BS application and temperature were highly significant (P<0.01 for both factors), while no significant differences were induced by the irrigation treatments (P=0.34; Table 2).

## Nitrous oxide emissions

On average, control soils under continuous irrigation emitted 22  $\mu$ g N<sub>2</sub>O-N m<sup>-2</sup> h<sup>-1</sup> at 13.5°C (Fig. 3a). With the rise in temperature, N<sub>2</sub>O production reached a peak of 408  $\mu$ g m<sup>-2</sup> h<sup>-1</sup> within a day. The average emission rate at the higher temperature was 194  $\mu$ g N<sub>2</sub>O-N m<sup>-2</sup> h<sup>-1</sup> (Fig. 3d), which resulted in a  $Q_{10}$  factor of 8.8. BS application at the lower temperature in the treatment with continuous irrigation resulted in an instant increase in N<sub>2</sub>O emissions, followed by a decrease over the next 3 weeks to the level of the control (Fig. 3a). When soils were fertilized at the higher temperature, maximum N<sub>2</sub>O emissions of 1,250  $\mu$ g N<sub>2</sub>O-N m<sup>-2</sup> h<sup>-1</sup> were measured, and the emission rates remained above the level of the control for about 6 weeks (Fig. 3d). During the 6 weeks following the BS application, 2.2% and 6.8% of the N applied were lost in the form of N<sub>2</sub>O at 13.5°C and 23.5°C, respectively (Fig. 2).

In the treatments with cycles of strong irrigation and partial drying, N<sub>2</sub>O emissions were highest 1 or 2 days after the beginning of the period of stronger irrigation. A second smaller peak was observed 1 day after the end of the period of stronger irrigation. This trend was independent of temperature and BS application time. During drier periods, N<sub>2</sub>O production was generally low, except directly after the application of BS (Fig. 3b-c, e-f). Most of the BS-related N<sub>2</sub>O emissions took place within a few weeks after the BS application. However, during periods of stronger irrigation, increased N2O emissions could still be observed after several irrigation cycles. The increase in temperature had little effect on the proportion of the N applied lost as N<sub>2</sub>O because N<sub>2</sub>O emissions increased in the BS-amended and in the control soil. When BS was applied before a period of stronger irrigation, 3.5% and 3.7% of the applied N were lost as N<sub>2</sub>O at 13.5°C and 23.5°C, respectively. When the application took place during the drier period, 3.9% and 3.0% of the N applied were lost at 13.5°C and 23.5°C, respectively (Fig. 2).

As with CO<sub>2</sub>, no significant differences were induced by irrigation pattern (P=0.71), as higher N<sub>2</sub>O emissions during periods of stronger irrigation were compensated for by lower emissions during drier periods in treatments with cycles of strong irrigation and partial drying (Table 2). In contrast, the increase in temperature resulted in significantly higher N<sub>2</sub>O emissions (P<0.01), with the  $Q_{10}$  factor across all treatments reaching 3.7. Slurry application also significantly increased N<sub>2</sub>O emissions compared to the unamended control (P<0.01).

# Methane consumption

Methane emission rates were constantly negative in all treatments, indicating that net  $CH_4$  consumption took place (data not shown). The average  $CH_4$  consumption in the treatments with continuous irrigation was 4 µg  $CH_4$ -C m<sup>-2</sup> h<sup>-1</sup>, which corresponds to 0.13 mg  $CH_4$  m<sup>-2</sup> day<sup>-1</sup>. The same rate was measured during periods of stronger irrigation in the treatFig. 3 Average N<sub>2</sub>O emissions from soil cores amended with BS and unfertilized cores (control). The treatments included continuous irrigation (a, d), cycles of strong irrigation and partial drying with BS application before a period of strong irrigation (b, e), and cycles of strong irrigation and partial drying with BS application after a period of strong irrigation (c, f). The undisturbed soil cores were incubated at 13.5°C and 23.5°C. Arrows refer to time of BS application. Each line represents the average of four cores. The standard error of the mean (n=4) is shown for the first measurement each week



ments with cycles of strong irrigation and partial drying, while CH<sub>4</sub> consumption reached 8 µg CH<sub>4</sub>-C m<sup>-2</sup> h<sup>-1</sup> during drier periods (Table 1). No significant differences in cumulative CH<sub>4</sub> consumption were caused by BS application (P=0.96) or temperature (P=0.33; Table 2). In contrast, CH<sub>4</sub> consumption was significantly increased in the treatments with cycles of strong irrigation and partial drying compared to the continuously irrigated cores (P<0.01).

# Leaching losses

Nitrate leaching from the cores under continuous irrigation without BS application was relatively constant averaging 37 mg  $NO_3^{-}N$  m<sup>-2</sup> day<sup>-1</sup> at 13.5°C and of 51 mg NO<sub>3</sub><sup>-</sup>-N m<sup>-2</sup> day<sup>-1</sup> at 23.5°C, which corresponds to a  $Q_{10}$  factor of 1.38 (Fig. 4). Between 60% and 70% of the NO<sub>3</sub><sup>-</sup>-N leached from the control soils of the treatments with cycles of strong irrigation and partial drying was lost during periods of stronger irrigation. While only 22 to 28 mg NO<sub>3</sub><sup>-</sup>-N m<sup>-2</sup> day<sup>-1</sup> was lost during drier periods, the average losses reached 257 to 368 mg NO<sub>3</sub><sup>-</sup>-N m<sup>-2</sup> day<sup>-1</sup> during periods of stronger irrigation, with the higher rates occurring at the higher temperature. Across all treatments, BS application and temperature both had highly significant effects on NO<sub>3</sub><sup>-</sup> leaching (*P*<0.01), while the irrigation treatments had no significant effect (*P*=0.63). Effects of BS application on NO<sub>3</sub><sup>-</sup> leaching were most pronounced between weeks 4 and 10 after the application date. During

Table 1 Cumulative gas emissions and leaching losses from soils amended with BS or without BS application (control)

Emissions and leaching losses (g $m^{-2}$ )	Temperature (°C)	Continuous irrigation		Periodic strong irrigation-1		Periodic strong irrigation-2	
		Control	BS	Control	BS	Control	BS
N <sub>2</sub> O-N	13.5	0.021 (0.010)	0.091 (0.015)	0.034 (0.009)	0.147 (0.036)	0.028 (0.009)	0.155 (0.063)
	23.5	0.197 (0.089)	0.418 (0.107)	0.094 (0.021)	0.214 (0.048)	0.078 (0.021)	0.177 (0.028)
CO <sub>2</sub> -C	13.5	19.62 (2.37)	22.95 (1.91)	22.53 (1.61)	26.97 (2.13)	22.01 (1.51)	27.49 (1.46)
	23.5	34.99 (4.58)	44.06 (2.63)	35.09 (3.74)	50.30 (8.29)	33.78 (4.76)	43.89 (1.81)
CH <sub>4</sub> -C	13.5	-3.25 (1.01)	-3.47 (0.60)	-6.00 (0.67)	-5.97 (0.46)	-6.29 (0.66)	-6.08 (0.34)
	23.5	-4.34 (1.27)	-3.27 (0.72)	-7.24 (1.40)	-6.93 (0.43)	-7.05 (1.34)	-7.31 (1.19)
NO <sub>3</sub> <sup>-</sup> -N	13.5	1.47 (0.17)	2.24 (0.13)	1.51 (0.13)	1.77 (0.33)	1.59 (0.13)	2.31 (0.25)
	23.5	2.47 (0.45)	2.98 (0.52)	2.72 (0.17)	3.14 (0.24)	2.70 (0.15)	3.18 (0.25)
N <sub>org</sub>	13.5	0.29 (0.06)	0.36 (0.08)	0.57 (0.06)	0.38 (0.05)	0.62 (0.06)	0.62 (0.08)
	23.5	0.79 (0.11)	1.16 (0.22)	0.91 (0.05)	0.72 (0.32)	0.98 (0.08)	1.20 (0.15)
TOC	13.5	1.11 (0.21)	0.95 (0.11)	0.86 (0.03)	0.81 (0.05)	0.88 (0.04)	0.81 (0.05)
	23.5	1.54 (0.31)	1.69 (0.19)	0.97 (0.02)	1.14 (0.09)	0.99 (0.02)	1.11 (0.09)

Undisturbed soil cores were incubated at two different temperatures and subjected to continuous or periodic irrigation patterns (periodic strong irrigation-1, BS applied immediately before 1 week of strong irrigation; periodic strong irrigation-2, BS applied 1 week after period of strong irrigation). Cumulative gas emissions were calculated for 6 weeks following BS application. Cumulative leaching losses were calculated between weeks 5 and 10 after applications. Values shown are means (n=4) and standard errors

this 6-week period, cumulative  $NO_3^-$ -N losses across all irrigation treatments corresponded to 18% and 14% of the N applied in the BS at 13.5°C and 23.5°C, respectively.

Losses of TDN closely followed  $NO_3^-$ -N losses. In fact, about 80% of the N in TDN was generally in the form of  $NO_3^-$ , while the concentration of  $NH_4^+$  in the leachate was always below the detection limit (data not shown). Thus, the difference between TDN and  $NO_3^-$ -N corresponded to organic N (N<sub>org</sub>). In contrast to  $NO_3^-$ , the application of BS before a period of stronger irrigation decreased leaching of organic N compared to the unamended control (Fig. 2).

Leaching losses of TOC were significantly increased at the higher temperature (P<0.01; Table 2), with losses averaging 24 and 36 mg C m<sup>-2</sup> day<sup>-1</sup> at the lower and higher temperature, respectively. In contrast, the application

Table 2 Statistical analysis of the data presented in Table 1

of BS had no significant effect on TOC leaching (P=0.73). However, TOC losses were significantly higher under continuous irrigation than under cycles of strong irrigation and partial drying. Dissolved organic C (<0.45 µm) accounted for about 94% of the TOC leached.

# Discussion

Under the conditions of our experiment, C and N turnovers in undisturbed soil cores were strongly affected by BS application and soil temperature. In contrast, the irrigation pattern and time of BS application had minor effects. In addition, only a few weak interactions were observed between these factors (Table 2).

	IRR P values	BS	Т	IRR×BS	IRR×T	BS×T				
N <sub>2</sub> O emissions	0.71	< 0.01	< 0.01	0.85	0.02	0.12				
CO <sub>2</sub> emissions	0.34	< 0.01	< 0.01	0.90	0.49	0.27				
CH <sub>4</sub> emissions	< 0.01	0.96	0.33	0.93	0.97	0.75				
NO <sub>3</sub> <sup>-</sup> leaching	0.63	< 0.01	< 0.01	0.68	0.56	0.70				
Norg leaching	0.03	0.47	< 0.01	0.06	0.22	0.21				
TOC leaching	< 0.01	0.73	< 0.01	0.94	0.08	0.12				

The datasets were analyzed as a three-factor analysis of variance. The factor IRR includes the levels continuous irrigation, cycles of strong irrigation and partial drying with BS application before a period of strong irrigation, and cycles of strong irrigation and partial drying with BS application after a period of strong irrigation. The factor BS includes cores amended with BS and unamended controls, while the factor T includes  $13.5^{\circ}$ C and  $23.5^{\circ}$ C

IRR irrigation, T temperature

Fig. 4 Average leaching losses of NO<sub>3</sub><sup>-</sup>-N from soil cores amended with BS and unfertilized cores (control). The treatments included continuous irrigation (a, d), cycles of strong irrigation and partial drying with BS application before a period of strong irrigation (b, e), and cycles of strong irrigation and partial drying with BS application after a period of strong irrigation (c, f). The undisturbed soil cores were incubated at 13.5°C and 23.5°C. Arrows refer to time of BS application. Sym*bols* represent means (n=4) with standard error of the mean



Biogas slurry application

With the application of BS,  $CO_2$  production increased immediately. Within 6 weeks, on average, 30% of the C applied with BS was respired as  $CO_2$ . These results indicate that a considerable proportion of the BS was readily available. In some microsites,  $O_2$  demand likely exceeded supply by diffusion, resulting in anaerobic conditions favorable for denitrification to occur.

Across all treatments, 3.8% of the N applied with the BS was lost as N<sub>2</sub>O during the 6 weeks following its application. In a study carried out at  $13.5^{\circ}$ C with a similar soil type and BS from the same fermenter, Sänger et al. (2010) found that about 1.2% of the added N was lost as

 $N_2O$ . Surprisingly, this proportion is lower than the average of 3.2% found in the present study at the lower temperature even though the application rate in the present study was three times lower. This suggests that the application rate had only a minor effect on the proportion of N lost as  $N_2O$ . In fact, in a pot experiment over 35 days with soil held at 85% WFPS, Senbayram et al. (2009) found that about 1.7% and 1.5% of the applied BS were lost as  $N_2O$  with application rates of 90 and 360 kg N ha<sup>-1</sup>, respectively. When applied to grassland at a rate of 480 kg N ha<sup>-1</sup>, less than 0.5% of the BS-N was lost in the form of  $N_2O$ (Senbayram et al. 2009).

In a review of the literature, de Klein et al. (2001) found that the N<sub>2</sub>O emissions reported for mineral

fertilizers generally ranged from 0.1% to about 2% of the N applied; however, a few short-term studies using nitratebased fertilizers reported emissions of up to 12%. For a wide variety of animal manures, emissions ranged from 0% to about 5% of the manure N applied (de Klein et al. 2001). This comparison suggests that the N<sub>2</sub>O emissions from BS in our study are similar to those reported for animal waste and slightly higher than emissions from mineral fertilizers. However, in the studies reviewed by de Klein et al. (2001), N fertilizer was applied to crops in the field. The fact that no N uptake by plants took place in our study most likely resulted in higher N<sub>2</sub>O emissions compared to field studies.

The application of BS increased leaching losses of  $NO_3^-$  but not of TOC, indicating that the microorganisms were limited by the availability of C while N availability exceeded demand. This result is not surprising given the narrow C to N ratio of BS, which was 8, and the high content of  $NH_4^+$ . On average, about 16% of the N applied with the BS was leached from the cores between weeks 5 and 10 after application.

In summary, a large proportion of the C and N were readily available to soil microorganisms. The excess N, which leached as  $NO_3^-$  in our study, could have been taken up by crops in a planted field. In contrast, as with slurry from animal husbandry, the application of BS to fallow fields may result in high  $NO_3^-$  leaching losses.

#### Soil moisture effects

In general, soil moisture has a strong effect on the decomposition of organic material. While at low moisture water is the limiting factor, excess moisture inhibits aerobic metabolism because of O<sub>2</sub> diffusion limitations. In an incubation study with different soils, Franzluebbers (1999) found that C mineralization was highest when the WFPS was between 53% and 66%. In the present study, WFPS ranged between 80% and 90%, which was likely at the upper limit for optimal aerobic respiration. This is evident from the fact that CO<sub>2</sub> evolution strongly decreased during rewetting periods. However, while the irrigation treatment affected the temporal pattern of CO<sub>2</sub> evolution, it had no significant effect on the cumulative  $CO_2$  production. This is likely due to the fact that the variation in soil moisture observed in our study was relatively small.

While soil moisture had a strong effect on the temporal pattern of  $N_2O$  emissions, there was no significant effect of the different irrigation treatments on cumulative  $N_2O$  emissions. These results were unexpected and differ from Clayton et al. (1997) and Smith et al. (1998) who found, in a field study, that  $N_2O$  emissions depend particularly on WFPS and rainfall at the time of fertilization. However, the

WFPS in these two studies ranged from 40% to 90%, while the fluctuations in WFPS in our study were much smaller. This is most likely the reason for the small effects the irrigation treatments had in our study.

Uptake of atmospheric CH<sub>4</sub> through biological oxidation has been reported in a variety of upland soil ecosystems. Methane oxidation generally occurs in wellaerated soils, although an anaerobic pathway for CH<sub>4</sub> oxidation has also been described (Segers 1998). Boeckx and van Cleemput (2001) summarized experimentally determined CH<sub>4</sub> uptake rates from different ecosystems and from different parts of the world. For arable soils, the calculated average of the minimum and maximum values from these data was 0.25–0.83 mg  $CH_4$  m<sup>-2</sup> day<sup>-1</sup>. With 0.13 to 0.26 mg  $CH_4$  m<sup>-2</sup> day<sup>-1</sup>, the  $CH_4$  consumption measured in our study was therefore at the lower end of this range. In contrast to N<sub>2</sub>O, soil moisture significantly affected CH<sub>4</sub> consumption. Methane consumption rates were increased during drier periods which may be due to the fact that CH<sub>4</sub> consumption in aerated soils is mainly controlled by soil moisture, as CH<sub>4</sub> diffusion is much slower in water than in air (Dörr et al. 1993). However, as soils contain both methanotrophic and methanogenic soil microorganisms, it is not possible to determine with the data available whether this difference was due to increased CH<sub>4</sub> consumption or decreased CH<sub>4</sub> production.

### Temperature

While the irrigation treatment and BS application time had only minor effects on the proportion of BS-N lost in the form of N<sub>2</sub>O, the effect of temperature was much stronger. Across all treatments, the 10°C increase in temperature increased N<sub>2</sub>O losses by a factor of 3.7, which is more than twice as high as the factor for  $CO_2$  (1.7). The  $Q_{10}$  factor for CO<sub>2</sub> found in our study is within the range of values reported in a number of laboratory studies summarized by von Lützow and Kögel-Knabner (2009). High  $Q_{10}$ factors for N<sub>2</sub>O have been reported in several studies. Abdalla et al. (2009) found  $Q_{10}$  factors ranging from 4.4 to 6.2 for a temperature range of 10-25°C in an anaerobic laboratory incubation with pasture soil amended with mineral fertilizer. In an incubation with soil cores of an imperfectly drained Gleysol fertilized with NH4NO3, Dobbie and Smith (2001) reported  $Q_{10}$  factors for an arable soil of 8.9 for the 12-18°C interval. These results indicate that N<sub>2</sub>O emissions are very sensitive to increasing temperature.

# Aerobic and anaerobic processes

The high  $N_2O$  emission rates indicate that anaerobic conditions were dominant in the soil cores, especially after

the addition of BS and at the higher temperature. However, the high  $CO_2$  production rate,  $CH_4$  consumption, and nitrification suggest aerobic conditions.

The greatly increased N<sub>2</sub>O emissions observed after the application of BS suggest that microorganisms used NO<sub>3</sub> as an alternative electron acceptor when O<sub>2</sub> became limiting. As the BS contained almost no  $NO_3^-$ , denitrifying bacteria must have either used NO<sub>3</sub><sup>-</sup> from the soil solution or NH<sub>4</sub><sup>+</sup> from the BS must have been nitrified in aerobic microsites and diffused to anaerobic sites where denitrification took place. However, nitrification may also result in the production of N<sub>2</sub>O. In a summary of published data, Mathieu et al. (2006) found that the proportion of nitrified N lost as N<sub>2</sub>O generally ranged between 0.05% and 0.5% for well-aerated soils while it may increase to values higher than 1% under O<sub>2</sub> limiting conditions. In the present study, about 37 and 51 mg  $NO_3^{-}N$  m<sup>-2</sup> day<sup>-1</sup> were leached from the unamended cores under continuous irrigation at 13.5°C and 23.5°C, respectively. Under the assumption that most of the  $NO_3^{-}$  produced was leached during the course of the experiment and that between 0.05% and 1% of the N was lost as N<sub>2</sub>O during nitrification, the production of the NO<sub>3</sub> leached would have caused N2O-N emissions of 19 to  $371 \ \mu g \ m^{-2} \ day^{-1}$  at  $13.5^{\circ}C$  and of 26 to  $510 \ \mu g \ m^{-2} \ day^{-1}$ at 23.5°C. Therefore, at the lower temperature, between 4% and 76% of the N<sub>2</sub>O emissions may have been produced through nitrification, while the contribution of nitrification was below 11% at the higher temperature. Even though these calculations are rough estimates, they indicate that nitrification may have considerably contributed to the N2O emissions even though the soils were relatively moist. However, denitrification was most likely the predominant source of N<sub>2</sub>O in our study, especially at the higher temperature. Applying the same calculations for the other treatments, some general patterns emerge. The contribution of nitrification was likely higher at the lower temperature than at the higher temperature. In addition, nitrification likely contributed less to N<sub>2</sub>O emissions in the BS-amended soils compared to the control. In both cases, these findings can be explained by the increased O<sub>2</sub> demand at the higher temperature and in BS-amended soil, which resulted in more anaerobic microsites.

Our finding that net  $CH_4$  consumption took place throughout the experiment in all treatments also indicates that aerobic microsites existed, as aerobic  $CH_4$  oxidation requires both oxygen and  $CH_4$ . Anaerobic methane oxidation is also possible; however, little is known about this process in the soil environment. Its contribution to  $CH_4$ consumption seems low (Segers 1998). These results highlight the complexity of biochemical reactions in soil, which is a heterogeneous environment. During the incubation, aerobic and anaerobic microsites likely existed on a temporal and on a spatial scale. The application of an organic C and N source, such as BS, may strongly affect the  $O_2$  balance in the soil and, with it, the rates of the biochemical processes taking place.

#### Conclusion

The use of maize and other crops for  $CH_4$  production is likely to increase in the future, and increasing quantities of maize-derived BS will become available as fertilizer. So far, only a few studies have dealt with the effects of BS applications on C and N turnovers in soil.

Our results indicate that  $N_2O$  emissions may be relatively large when BS is applied to moist soil. However, the proportion of N lost as  $N_2O$  is within the range reported for slurries from animal husbandry. The small effect of soil moisture at the time of application was at least partly due to the high initial moisture contents and subsequent small fluctuations in soil moisture content observed in our study and cannot be generalized.

Our results also indicate that a relatively large proportion of the N is mineralized and nitrified during the first few weeks after BS application and could be susceptible to N leaching in fallow fields. However, our study was carried out under controlled conditions, and there is still little known about the proportion of N which becomes available during the cropping season. In addition, we used BS from only one fermenter in our study, and the properties of BS from other fermenters and production processes may be quite variable.

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