



Computational-Aided Discovery of Novel 1,3-Benzodioxole Plant Growth Retardants

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Abstract

Plant growth retardants can enhance the lodging resistance of crops and prevent yield losses. In this study, we used the crystal structure of the gibberellin receptor GIBBERELLIN INSENSITIVE DWARF1 (GID1) to aid the discovery of novel growth retardants. The Maybridge database was filtered based on multiple-screening analysis, including pharmacophore modeling, target structures, Lipinski's rules of 5, and physicochemical properties. Finally, the 1,3-benzodioxole compounds were screened and their potential interaction mechanisms, synthesis, and biological activity on the root lengths of wild-type and mutant-type *Arabidopsis thaliana* were investigated. The results showed that the compounds inhibited root growth in WT *A. thaliana* at a range of concentrations, with optimal inhibition observed at 100 $\mu\text{mol/L}$. The inhibition by compound HTS05309 was comparable to that of uniconazole (100%). Quantitative experiments confirmed that the compounds inhibited *A. thaliana* growth. These results suggested that novel plant growth retardants have been identified based on computational-aided discovery.

Keywords Gibberellin · GID1A · 1,3-Benzodioxole · Plant growth retardant

Introduction

Gibberellins are a class of plant growth regulators that promote the longitudinal growth of stems and induce hydrolytic enzymes in germinating seeds (Matsuoka 2003; Petracek et al. 2013; Rademacher 2015). To date, 136 gibberellins have been identified in plants, fungi and bacteria (Garcia-Martinez and Gil 2002). Gibberellin acid (GA_3) has been widely used in agricultural production as a known promoter of plant growth (Ozga and Reinecke 2003) (Fig. S1).

Phthalimide compounds, including AC94377 (Suttle and Hultstr 1987), AC006, AC007, and AC015 (Yalpani et al. 1989; Suttle et al. 1992), or GA_3 derivatives including 10d and 10f (Tian et al. 2017) have been identified as functional analogs of gibberellins (Fig. S1). N-Substituted phthalimides (NSPs) compete for binding to GA_4 with gibberellin-binding proteins (Yalpani et al. 1989; Jiang et al. 2017). GIBBERELLIN INSENSITIVE DWARF1 (GID1) was discovered to be a soluble gibberellin receptor (Ueguchi-Tanaka et al. 2005; Nakajima et al. 2006; Hauvermale et al. 2014) which provides the foundation for target-based signaling transduction, metabolism, and molecular design.

Although gibberellin increases plant yields, it also causes conflicts between “high yield and lodging”. In the 1960s, the large-scale promotion of semi-dwarf rice and wheat varieties greatly improved crop yields and solved the food crisis (Peng et al. 1999). Furthermore, dwarf plants with gibberellin synthesis and signal transduction defects are largely resistant to lodging. Uniconazole and paclobutrazol, trinexapac-ethyl and prohexadione calcium, inhibitors of kauri oxidase, and gibberellin 3-oxidase ($\text{GA}3\text{ox}$) inhibit gibberellin biosynthesis and are broad-spectrum, high-efficiency, low-residue plant growth retardants that can dwarf plants and shorten root lengths (Adriansen and Odgaard 1997).

Jine Wang and Zhikun Yang contributed equally to this article.

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As such, plant growth retardants can increase plant yields through lodging resistance.

In this study, we combined plant growth regulators, computer-aided drug design and organic synthesis, based on multiple screening, gibberellin targets, pharmacodynamics, Lipinski's rules of five, physicochemical properties, toxicity and biological activity determination, to identify new and highly active plant growth retardants. Moreover, molecular binding modes and binding free energy studies were also performed.

Materials and Methods

Molecular Conformation and Energy Optimization

Ligands were constructed using the Build/Edit module. The lowest-energy conformations were optimized by SYBYL 7.3 software using the steepest gradient descent method (Hirano et al. 2010). The Tripos force-field was selected and the Gasteiger–Hückel charge was loaded with 10,000 iterations and an energy convergence of 0.005 kJ/mol. Other parameters were defined as default values unless otherwise indicated.

Molecular Docking

Molecular docking was performed using SYBYL 7.3 software using the Surflex-dock algorithm (Jain 2003; Clark et al. 2002) in which the crystal structure of GA₃-GID1A-DELLA from the RCSB protein data bank (PDB ID: 2ZSH) was used as the docking receptor. The receptor model was prepared with charges, atom types, protonation, and the addition of hydrogen atoms. The best protocol in the active domain of the receptor was obtained using a ligand docking mode.

Building Pharmacophore Models

Seven reported gibberellins including GA₃, GA₄, GA₇ and functional analog NSPs including AC94377, AC006, AC007, and AC015 were used to generate a 3D pharmacophore model using the GALAHAD method. This method encompassed five features including: hydrogen bond donor and acceptor atoms; positive nitrogen; and negative hydrophobic centers. Twenty pharmacophore models were obtained and evaluated, and representative models were selected. Parameters including population size, max generations, mols required to N hit, and keep best N models were set to 45, 45, 5 and 20, respectively. Other parameters used default values, unless stated otherwise.

Virtual Screening

The Maybridge database containing a collection of 58,698 molecules was used for virtual screening. To ensure accuracy of screening protocols, the energy of the molecules was minimized prior to the screening process. Initially, the Maybridge database was filtered using Lipinski's rules of five ($M < 450$) (Sander et al. 2015; Kelder et al. 1999). LogS and ADME/T for the molecules was then predicted using the open software DataWarrior in which low solubility and toxic compounds were filtered. The resultant molecules were finally screened against the pharmacophore models of gibberellins and the active pocket of GID1A using SYBYL 7.3 software. The screening molecules were ranked by score clustering and the top 15% were selected. Four of the selected were synthesized based on their solubility and available synthesis.

Molecular Dynamics Simulations

The complex crystal structure of GA₃-GID1A-DELLA was used to mimic the receptor–ligand interaction between through molecular dynamics simulations. During the preparation of the receptor structure, the protonation states of histidine residues were manually inspected to ensure consistency with the local chemical environment. All systems were optimized using Amber14 software. The force-field parameters for the proteins were generated using Amber ff99SB (Hornak et al. 2006), whilst those for ligands were generated using General Amber Force Field (GAFF) (Wang et al. 2004). A rectangular water box was used to fill the gaps between the receptor and ligands, which was generated with TIP3P water molecules with an edge of 12 Å. Na⁺ ions were added to neutralize the system and data collection of 10 ns of the production run was performed through molecular dynamics simulations (Chen et al. 2018). Finally, calculations of the free binding energy were performed using the MM/GBSA method available in Amber14 software (Zhu et al. 2014).

Biological Activity Assays

The bioactivity of the synthesized compounds was determined in vivo by measuring the root length of *A. thaliana*. 1/2 MS (Murashige–Skoog) culture media were prepared by mixing distilled water (1 L) with MS (1.08 g), sucrose (5 g), and agar (4 g), and pH was adjusted to 5.9 with Sodium hydroxide. The media was sterilized in a high-pressure steam sterilization pot and *A. thaliana* seeds were sterilized in 1% sodium hypochlorite for 15 min. Following sterilization with sodium hypochlorite, seeds were washed five times with

sterile distilled water. Seeds were planted in 1/2 MS media with four concentrations (0.1, 1, 10, and 100 $\mu\text{mol L}^{-1}$). GA_3 and uniconazole were included as controls. Each medium was planted with 12 seeds and maintained at 4 °C for two days. Then, seeds were incubated under consecutive light (400 $\mu\text{mol}^{-2} \text{s}^{-1}$ of photon flux density) for 7 days. Root lengths were measured using ImageJ software (Tian et al. 2017).

Results

Molecular Docking of GAs and Their Functional Analogs

Molecular simulations can be used in drug design, but their reliability is dependent on their correlation to experimental data. To verify the accuracy of our molecular simulations, GA_3 , GA_4 , and GA_7 , and four functional analogs, AC94377, AC006, AC007, and AC015 (Fig. S1) were docked onto the *GID1A* receptor to predict the correlation between binding scores and experimental protein-binding activity.

As shown in Fig. S2, the predicted docking scores correlated with the protein-binding activity, pIC_{50} , and high R^2 values (approximately equal to 0.71). The high-predicted docking scores indicated a high affinity between the ligand and receptor. The conformations of high-scoring ligands were then superimposed (Table S1). The NSPs were found to be highly similar to gibberellins and could fit into the binding pocket of the *GID1A* receptor, indicating their potential as competitive inhibitors to gibberellins.

Pharmacophore Models

Seven molecules with diverse structures, including GA_3 , GA_4 , GA_7 , AC94377, AC006, AC007, and AC015, could fit into the binding cavity of *GID1A* with good binding ability. These compounds were used as training sets for the construction of pharmacophore models using the GALAHAD method (Abrahamian et al. 2003; Richmond et al. 2006). The best pharmacophore model with the highest number of features was selected amongst the 20 generated pharmacophore models using the GALAHAD method (Fig. S3). The generated query consisted of four hydrophobic centers, one H-bond donor atom, and three H-bond acceptor atoms. Six ‘hit’ ligands amongst the 7 molecules were obtained through 3D searches. Furthermore, 100 molecules, including 7 in the training sets and 93 randomly selected from the Maybridge database were used to verify the reliability of the generated model. Seven molecules in the training sets achieved enrichment factors (EFs) (Huang et al. 2006) as an index of screening ability of 10. This indicated that the model was reliable,

and can accurately identify gibberellin-like molecules from pools of molecules from the Maybridge database.

Virtual Screening

Virtual screening against the Maybridge database was performed according to the workflow shown in Fig. 1. Prior to virtual screening, the 2D structures of all molecules were converted to 3D structures using SYBYL 7.3. Database screening was performed using the Lipinski’s rule of five ($M \leq 450$) (Huang et al. 2006) using SYBYL 7.3 software. DataWarrior was used to determine toxicity, LogS, and the polar surface area (PSA, $\text{PSA} < 120 \text{ \AA}^2$) of all molecules. All toxic molecules were removed from the database. A total of 3078 molecules resembling the best pharmacophore models were selected and used in the screening process. The molecules were docked onto the binding site of the *GID1A* receptor using SYBYL7.3. The results were ranked by total scores. To ensure screening accuracy, each conformation obtained from the docking was subjected to clustering analysis, in addition to visual judgments. The top 15% of molecules were selected, amongst which four 1,3-benzodioxole compounds were selected for mechanistic studies.

Binding Mode Analysis

Compounds HTS05467, HTS05309, HTS05461, NRB00180 and GA_3 were docked onto the *GID1A* receptor, and the binding modes were analyzed to elucidate potential molecular mechanisms. The interactions between GA_3 and the receptor indicated hydrogen bonds between the ligand and residues SER116 (1.85 \AA), TYR127 (1.87 \AA), and SER191 (2.74 \AA) (Fig. 2a). Water molecules are required for the GA_3 -*GID1A* interaction (Murase et al. 2008). In this study, water molecules $\text{H}_2\text{O}348$, $\text{H}_2\text{O}350$, and $\text{H}_2\text{O}364$ formed hydrogen bonds with GA_3 , whilst other water molecules including $\text{H}_2\text{O}380$, $\text{H}_2\text{O}400$, $\text{H}_2\text{O}401$ created aqueous

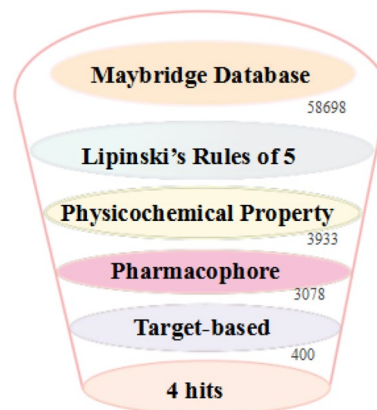


Fig. 1 Virtual screening workflow

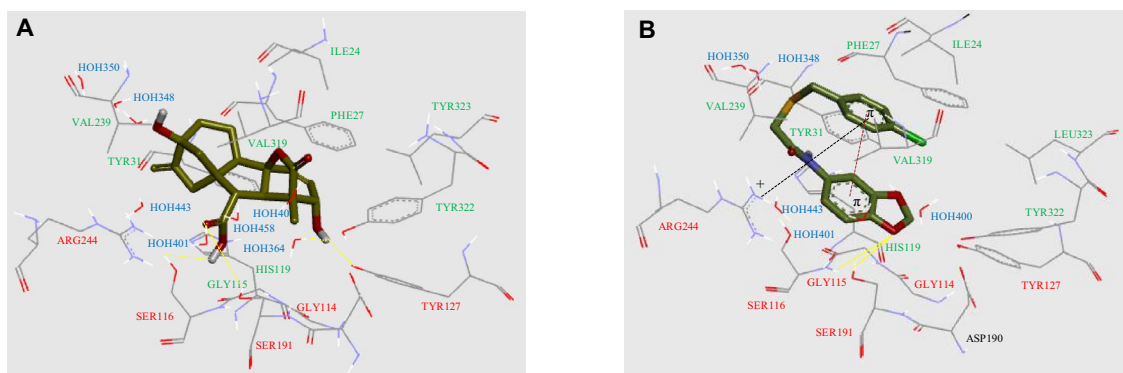


Fig. 2 Proposed binding modes of GA₃ (**a**) and compound HTS05309 (**b**) to GID1A (2ZSH). Hydrophobic residues are labeled in green, electrostatic interactions are labeled in red, water molecules are labeled in blue, and unique residues ASP190 were labeled in black.

environments for the complexation of GA₃-GID1A-DELLA. These conclusions were consistent with previous studies (Murase et al. 2008; Duan et al. 2013).

Compound HTS05309 could bind to the GID1A receptor (Fig. 2b) through a similar binding mode to GA₃. Moreover, the binding conformation of HTS05309 with a linker carrying two aromatic rings in a ‘U-type’ conformation was loaded onto the receptor-binding site. During compound HTS05309 binding, van der Waals interactions contributed to the hydrophobic effects, involving residues PHE27, TYR31 and VAL319. Whilst residues, GLY114, SER116, SER127, ASP190, SER191, and ARG244 contributed to the electrostatic interactions, water molecules from H₂O348, H₂O350, H₂O364, H₂O400, H₂O401, H₂O443, and H₂O458 played a major role during ligand–receptor binding. The intramolecular π – π interaction, which enhanced the structural stability and affinity, was formed between the two benzene rings of compound HTS05309. The head of the benzene ring of compound HTS05309 was situated in the hydrophobic environment, with the interaction between residues ILE24, PHE27, and TYR31 at the cap region of the DELLA protein stabilizing its structure (Hirano et al. 2007). The other end of compound HTS05309 interacted with residues GIY115, SER116, TYR127, and SER191 in a relatively hydrophilic environment, thereby enhancing the affinity towards the inhibitor, whilst stabilizing the binding site. The intermolecular π -cation interaction that enhances binding affinity, lipophilicity, specificity, and drug design (Yoon et al. 2013) was formed between the benzene ring in compound HTS05309 and an N atom of ARG244 in GID1A. Hydrogen bonding between GIY115 (2.61 Å) or SER191 (2.51 Å) of the GID1A receptor and HTS05309 was also observed. As shown in Table S2, the structure of HTS05309 is similar to that of GA₃ with an overlap score of 0.61. The binding score (5.36) to the receptor was comparable to the

The yellow-dotted lines indicate the hydrogen bonds, purple-dotted lines indicate π – π interactions, and black solid lines indicate π -cation (Color figure online)

binding score of GA₃ (5.92). It can be speculated that compound HTS05309 acts as a gibberellin functional analog. Compound HTS05309 showed a similar binding mode to GA₃ mainly due to the similarity in pharmacophores, competing with GA₃ for receptor binding. This finding requires confirmation in protein-binding activity assays, and physiological studies.

Verification of Binding-Free Energy

The free binding energy of a 10 ns molecular dynamic simulation of compound HTS05309 and GA₃ was assessed using AMBER14 software. As shown in Table 1, HTS05309 binds to the GID1A receptor with a total binding energy of – 47.53 kcal/mol, which is remarkably lower than that of GA₃ (– 39.13 kcal/mol). This indicates that HTS05309 has a stronger receptor-binding affinity than GA₃. Furthermore, energy decomposition assessments showed that van der Waals and electrostatic interactions were the major contributors to the binding affinities. The binding energy of compound HTS05309 was dictated by van der Waals interactions (– 48.20 kcal/mol) which were twofold higher than the electrostatic interactions (– 23.86 kcal/mol). As

Table 1 Calculated binding-free energy (kcal/mol) of the complex formed between compound HTS05309 and GID1A, and GA₃ and GID1A

	HTS05309	GA ₃
ΔE_{VDW}	-48.20 ± 0.30	-44.42 ± 0.30
ΔE_{ele}	-23.86 ± 0.53	-33.02 ± 0.66
ΔG_{sol}	24.54 ± 0.30	38.31 ± 0.35
ΔG_{tot}	-47.53 ± 0.48	-39.13 ± 0.46

ΔG_{VDW} Van der Waals interaction, ΔG_{ele} electrostatic interaction, ΔG_{sol} polar interaction

previously described (Chen et al. 2018), non-polar solvation exerts positive effects on the binding energy, while polar solvation has negative effects, in agreement with the results of this study (Table 1).

$$\nu G_{\text{tot}} = \nu G_{\text{VDW}} + \nu G_{\text{ele}} + \nu G_{\text{sol}}$$

To explore the contribution of each individual residue, the energy decomposition of ligand–residue pairs was determined. The decomposing energy of key-binding residues between GA₃ and compound HTS05309 are illustrated in Fig. 3. The contribution rates of the major residues for GA₃ and compound HTS05309 are listed in Table S3. In agreement with our binding mode analysis, GA₃-receptor binding was dominated by Van der Waals interactions involving LE24, PHE27, TYR31, and VAL319 (Duan et al. 2013), with a contribution rate of 50.76%, compared to the electrostatic interactions of SER116, SER191, and ARG244 (Murase et al. 2008) that contributed only 9.85% (Table S3).

ILE24, PHE27, TYR31, GLY115, HIS119, VAL239, VAL319, TYR322, and LEU323 interact with GA₃ and HTS05309 through van der Waals forces (Table S3) in agreement with our binding mode results. The contribution rate of ILE24 in the cap region to GA₃ binding was fivefold higher than HTS05309 (0.87%). However, the contribution of PHE27 to GA₃ binding (13.09%) was lower than HTS05309 (17.61%) as HTS05309 was closer to PHE27. TYR31 was found to interact with GA₃ and HTS05309 with almost equal van der Waals forces. GLY115 was unique and stabilized compound HTS05309 through H-bond interactions (Fig. 3). Van der Waals interactions were not involved in GA₃ binding, enhancing the binding affinity. The other 3 residues HIS119, VAL239, and TYR322 also interacted with GA₃ and HTS05309 with equal contribution rates of approximately 11%. It was noteworthy that the contribution of VAL239 to the binding was higher than that of HIS119 and TYR322. Residue VAL319 reported to be important for the formation of the GA₃-GID1A-DELLA complex (Murase

et al. 2008), also interacts with HTS05309 through van der Waals interactions. Compound HTS05309 binds to the GID1A receptor with a van der Waals force of − 48.20 kcal/mol, which is close to that of GA₃ (− 44.42 kcal/mol). This indicates that HTS05309 and GA₃ have equal binding affinities for van der Waals interactions.

Other residues including SER116, SER127, ASP190, SER191, and ARG244 interacted with GA₃ and HTS05309 through electrostatic interactions (Table S3), in agreement with our binding mode analysis. However, despite hydrogen bond formation between GA₃ and SER116 (1.85 Å), the contribution of SER116 to the GA₃-GID1A-DELLA (3.55%) interaction remained lower than that of HTS05309-GID1A-DELLA (5.52%), most likely due to a loss of polarity. ASP190 has a negative effect on the receptor interaction between GA₃ (− 10.92%) and HTS05309 (− 4.23%) due to effects of polarity that may offset the electrostatic interactions. ASP190 may thus be important in the ligand-target interaction, and should be considered during drug design. Hydrogen bonds between SER191 and the ligand were observed in the crystal structure of GA₃-GID1A-DELLA and the docking of HTS05309-GID1A-DELLA complexes. Its contribution to the GA₃-GID1A-DELLA complex was 1.64-fold higher than the HTS05309-GID1A-DELLA complex. ARG244 equally interacts with GA₃ and HTS05309 (8.19% and 9.01%, respectively) and has a smaller effect on the polarity interaction. Overall, although the interaction modes of HTS05309 and GA₃ were similar, polar interactions lead to differences in the total free binding energy. For GA₃ and compound HTS05309, the differences were reflected by the contribution of PHE27 and ASP190.

Biological Activity Assays

To evaluate the activity of the lead compound, HTS05467, HTS05309, HTS05461 and NRB00180, were synthesized in two-steps (Platzer et al. 2016; Yan et al. 2016) as shown in

Fig. 3 Binding-free energy decomposition for the main residues of compound GA₃ and HTS05309

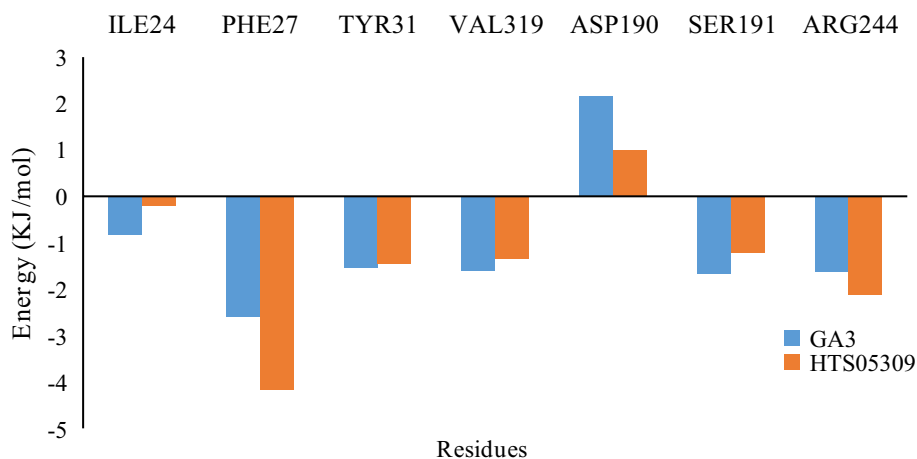


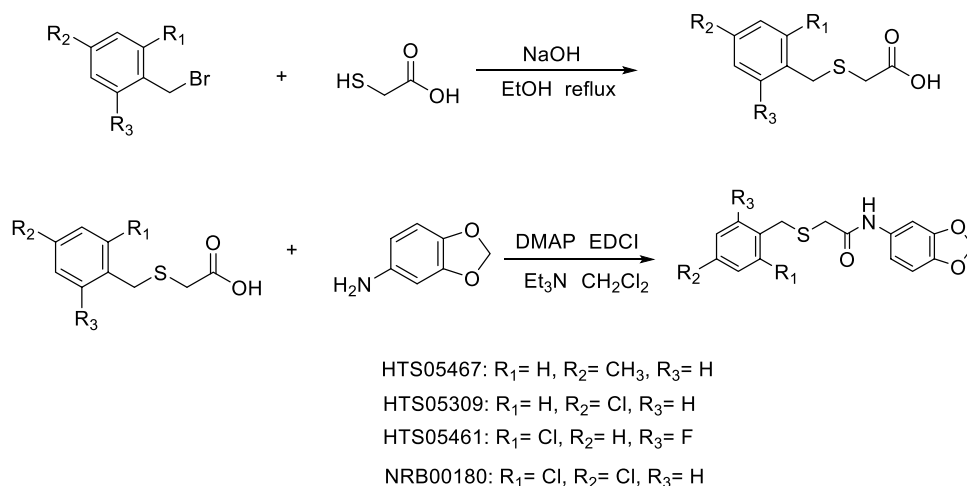
Fig. 4 Synthetic scheme of the 1,3-benzodioxole compounds

Fig. 4. Firstly, benzyl bromide with substituents was reacted with thioglycolic acid. Using Et₃N as an acid-binding agent, the thioether was then reacted with benzo[*d*] [1,3] dioxol-5-amine under the catalysis of DMAP and EDCI, producing the target products. The details in synthesis are shown in supplementary material.

Root lengths were used as indicators of HTS05467, HTS05309, HTS05461 and NRB00180 activity against the growth of WT and mutant *A. thaliana* (Table 2). In control group, GA₃ had no effect, whilst uniconazole led to significant inhibition. As indicated by the root length, HTS05467, HTS05309, HTS05461 and NRB00180 had a significant inhibition growth effects on WT plants, which

is similar with that of uniconazole. Compounds HTS05467, HTS05309, HTS05461 and NRB00180 had no significant effects on WT plants at low concentration, but exhibited considerable effects at high concentrations (Liang et al. 2018). At 0.1 and 10 μmol L⁻¹ (Table 2), the effects of compound HTS05461 on WT plants were similar with uniconazole, and higher than that of compounds HTS05309 (10% and 11%) and NRB00180 (15% and 16%). At 10 μmol L⁻¹, compounds HTS05309 and NRB00180 significantly inhibited WT plants (25% and 37%, respectively) whilst HTS05467 and HTS05461 had no significant effects. At 100 μmol L⁻¹, all compounds significantly inhibited WT plants with inhibition rates of 89%, 100%, 76%, and 73% for compounds

Table 2 Effects of 1,3-benzodioxole compounds on the root lengths of *Arabidopsis*

Comp		Wild type				Mutant type
		0.1 μmol L ⁻¹	1 μmol L ⁻¹	10 μmol L ⁻¹	100 μmol L ⁻¹	100 μmol L ⁻¹
GA ₃		1.94 ± 0.26	1.87 ± 0.17	1.78 ± 0.22	1.81 ± 0.32	0.68 ± 0.13
Uniconazole		1.46 ± 0.15	1.43 ± 0.18	1.45 ± 0.14	0 ± 0	–
HTS05467		1.81 ± 0.24	1.80 ± 0.29	1.72 ± 0.32	0.20 ± 0.05	0.35 ± 0.10
HTS05309		1.72 ± 0.18	1.70 ± 0.17	1.44 ± 0.19	0 ± 0	0.34 ± 0.09
HTS05461		1.58 ± 0.16	1.56 ± 0.24	1.67 ± 0.18	0.44 ± 0.11	0.28 ± 0.07
NRB00180		1.63 ± 0.28	1.61 ± 0.18	1.20 ± 0.16	0.50 ± 0.15	0.44 ± 0.08

Data are mean values ± SD obtained from three independent experiments

HTS05467, HTS05309, HTS05461 and NRB00180, respectively. Compounds HTS05467, HTS05309, HTS05461 and NRB00180 had strong inhibitory effects on mutant plants. In general, compounds HTS05467, HTS05309, HTS05461 and NRB00180 had significant inhibitory effects on both WT and mutant type. In particular, compound HTS05309 was similar with that of uniconazole, which exhibited 100% inhibition at $100 \mu\text{mol L}^{-1}$.

The results showed that the lead compounds inhibited plant growth, which may be due to the common pharmacophore characteristics between the promoters and inhibitors. To validate these findings, four commercially available growth retardants, namely uniconazole, paclobutrazol, prohexadione calcium and trinexapac-ethyl were collected to establish the pharmacophore model (Fig. 5). The compounds were screened through flexible screening. Upon comparison, the main difference was that the inhibitor model lacked a symmetric hydrogen bond acceptor, and the only hydrogen bond donor was hydrophobic. Inhibitors and accelerator models are more comparable with overlapping pharmacodynamics. Thus, the model can successfully screen growth retardants.

Discussion

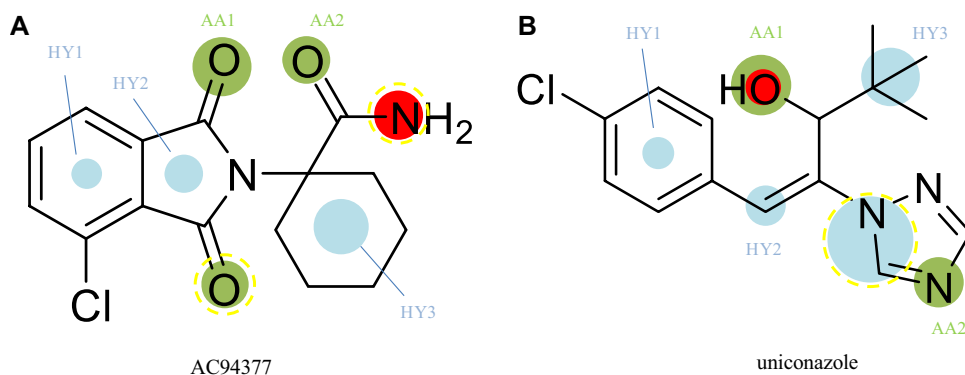
Phthalimides have been shown to act as competitive inhibitors of gibberellin receptors, of which AC94377 and GA_3 have comparable effects (Jiang et al. 2017) and bind to the active sites of GA_4 receptors. Our findings suggests that AC94377 competes with gibberellin for receptor binding due to its high-structural similarity, with both compounds entering the active pocket of the receptor, with similar modes of action. The crystal structure of GA_3 -GID1A-DELLA, docking models, dynamic simulations, and combined free decomposition indicated that residues TYR127, SER116, and SER191 contribute important electrostatic interactions and hydrogen bonding during GA_3 and GID1A binding, whilst non-polar residues ILE24, PHE27, TYR31 and VAL319 contribute hydrophobic effects. Important residues during

the binding of GA_3 to GID1A have been previously characterized (Murase et al. 2008; Duan et al. 2013). ASP190 negatively effects the interaction due to polarity effects offsetting the electrostatic interactions, which may be of importance during ligand-target drug design.

The GALAHAD method can accurately construct pharmacodynamics using external macro definition files to assign features, whilst allowing the donor and acceptor to overlap (Abrahamian et al. 2003). Richmond et al. (2006). showed that the ligand pharmacophore generated by GALAHAD resembles that of the crystal structure of its receptor. Moreover, the identification of new features can be beneficial for drug modifications, and so the GALAHAD method was used to establish accurate pharmacophore models for screening using gibberellins and its functional analogs. Virtual screening employs multiple pathways, including Lipinski's rule of five ($M \leq 450$) (Huang et al. 2006), physicochemical properties, toxicity, LogS, polar surface areas (PSA, $\text{PSA} < 120 \text{ \AA}$), pharmacodynamics of rifalazil and target-based drug design. A total of four 1,3-benzodioxole compounds were screened from the Maybridge database. The binding modes of the lead compounds were similar to GA_3 , whilst the binding activity of the compounds increased. The most significant difference between the binding mode of lead compounds and GA_3 was the contribution of ASP190. Compared with GA_3 , the polar interactions and collision between the ligands and target were enhanced, thus increasing binding affinity. Thus, on the premise of retaining interactions with key residues, reducing the polar interaction and collision of the ligands with ASP190 enhances the affinity between the ligand and its target, which is conducive to improved activity. Screening and bioactivity determinations based on the pharmacophore model and target revealed that the model was a valid screening method that could exclude molecules that did not match the structure and reduces the number of compounds entering the next stage. Screening based on these targets could lead to the identification of drugs based on the binding mode (hydrogen bonds, hydrophobicity and hydrophilicity).

In summary, the 1,3-benzodioxoles HTS05467, HTS05309, HTS05461 and NRB00180 were filtered out

Fig. 5 Promoter and retardant pharmacophore characteristics. **a** Promoter model using AC94377 as an example. **b** Characteristics of the inhibitor simulation using uniconazole as an example. Cyan spheres: hydrophobic centers (HY); red spheres: donor atoms (DA); green spheres: acceptor atoms (AA); yellow-dotted lines: differences between the models (Color figure online)



from the Maybridge database through a multiple-screening strategy. Their potential interaction mechanisms were studied and lead compounds were synthesized to determine their effects on WT and mutant *Arabidopsis* root lengths. The 1,3-benzodioxole compounds, and compounds HTS05467, HTS05309, HTS05461 and NRB00180 exhibited inhibitory effects on *A. thaliana* growth (inhibition rates of 89%, 100%, 76% and 73%, respectively). Compound HTS05309 was comparable to uniconazole in terms of activity. Quantitative experiments further demonstrated that the 1,3-benzodioxole compounds inhibited *A. thaliana* growth. The inhibitor model was constructed to reveal this phenomenon, and indicated that the retardant or inhibitor model and accelerator model have similar pharmacophore characteristics, ultimately leading to the production of an inhibitory lead compound. The discovery of gibberellin inhibitors is beneficial to improving plant resistance to lodging and stress stimuli, ultimately increasing the biological yield.

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Author Contributions Jine Wang carried out drug calculation, biological activity experiments and manuscript writing; Zhikun Yang carried out the synthesis of lead compounds, biological activity experiments and chapter writing; Hongxia Duan helped to carry out experimental analysis and control the writing progress and review of manuscripts; Liusheng Duan confirmed the research content, controlled the progress of the test and the quality of the manuscript writing; Weiming Tan conceived and designed experiments, and controlled the writing progress and review of manuscripts.

Compliance with Ethical Standards

Conflict of interest The authors declare no conflict of interest.

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