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Discovery and frequency measurement of far-infrared laser emissions generated by optically pumped CH2DOH

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ABSTRACT A recently improved three-laser heterodyne system was used to frequency measure ten previously observed optically pumped far-infrared (FIR) laser emissions from the partially deuterated methanol isotopologue CH₂DOH. Also, a 64.0 μ m FIR emission generated by the $9P32$ line of the carbon dioxide ($CO₂$) laser was discovered and frequency measured. These newly measured frequencies have fractional uncertainties on the order of $\pm 2 \times 10^{-7}$ and correspond to laser wavelengths ranging from 42.6 to 152.7 μ m. The offset frequency for the CO₂ pump laser was measured for twenty-two CH₂DOH FIR laser emissions.

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1 Introduction

2 Experimental details

The $CH₂DOH$ isotopic species of methanol has emerged as an effective medium for generating optically pumped far-infrared (FIR) laser emissions. Along with the first observation of FIR laser action from this molecule in 1978 [1], subsequent investigations have detailed the discovery and frequency measurement of $CH₂DOH$ FIR laser emissions [2–6]. Currently, there are over eighty known FIR laser emissions having wavelengths that range from 42.6 to $762.5 \,\mu m$. Of these, 23 lines are in the short-wavelength $(25 \mu m < \lambda < 150 \mu m)$ portion of the FIR region and fifty have been frequency measured. Utilizing improvements recently made to the three-laser heterodyne system [7, 8], the frequencies of ten previously discovered FIR laser emissions have been measured for the first time. Along with the discovery and frequency measurement of a new short-wavelength FIR laser emission, 22 CO_2 offset frequencies for FIR laser lines are reported.

The optically pumped molecular laser used to generate FIR emissions consisted of a tunable Fabry–Pérot cavity that was pumped in a X–V geometry with infrared radiation from a 2-mlong CO₂ laser, discussed in detail in [9]. The FIR cavity utilizes a nearly confocal mirror system with one mirror mounted on a calibrated micrometer used to tune the cavity into resonance with the FIR laser radiation. Initial FIR laser wavelengths were determined by measuring the micrometer movement between twenty adjacent longitudinal modes, corresponding to ten full wavelengths. The CH₂DOH, 98% D enriched, sample was obtained from Cambridge Isotope Laboratories.

The frequencies of the FIR laser emissions were measured using the three-laser heterodyne technique [7, 8, 10, 11]. Here, reference frequencies were used from two continuous-wave $CO₂$ lasers, selected such that their difference frequency was in the FIR region, close to that of the unknown laser fre-

quency. The $CO₂$ reference emissions were stabilized by locking each laser to the saturation dip in the $4.3 \mu m$ fluorescence signal from an external reference cell [12]. These signals, along with the unknown FIR laser emission, were then combined in a metal–insulator–metal (MIM) point contact diode consisting of a sharpened tungsten whisker in contact with a polished nickel rod. The unknown FIR laser emission and the difference frequency created from the two CO₂ reference emissions produced a beat frequency in the microwave region that was amplified (using one of two amplifiers) and observed on a spectrum analyzer. The amplifiers used in this work were either an Avantek amplifier (operating between 0.1 and 1200 MHz) or a Miteq AFS44 amplifier (operating between 0.1 to 26.5 GHz). Once the beat was observed, it was measured by placing standardized frequency markers at half the maximum amplitude, symmetrically about the center of the peak. The average of these frequencies was taken as the center of the beat frequency. A typical display on the spectrum analyzer can be seen in Fig. 1. Due to the increased spectral range (up to 25 GHz) and sensitivity (up to a factor of 30) of the heterodyne system [8], several previously observed FIR emissions, such as the $49.0 \,\mu m$ FIR emission shown in Fig. 1, were able to

be frequency measured. To calculate the unknown FIR laser frequency, v_{FIR} , the relation

$$
\nu_{\text{FIR}} = |n_1 \nu_{\text{CO}_2(I)} - n_2 \nu_{\text{CO}_2(II)}| \pm \nu_{\text{beat}} ,
$$
\n(1)

was used, where $|n_1v_{CO_2(I)} - n_2v_{CO_2(I)}|$ is the difference frequency synthesized

by the two CO_2 reference lasers and v_{beat} is the measured beat frequency. The integers n_1 and n_2 correspond to the respective harmonics used to generate the difference frequency.

The one-sigma fractional uncertainty, $\Delta v/v$, of the FIR laser frequency measurements in this work was $\pm 2 \times 10^{-7}$. This uncertainty was derived mainly from the broadened gain curve of the observed FIR laser emission and the reproducibility of the measurements. To minimize this uncertainty, the FIR cavity was tuned across the gain curve and the center frequency was measured using a peak hold feature. Irises internal to the laser cavity were used to eliminate higher-order modes and help shape the gain curve to a symmetric pattern. In all cases, two or more sets of $CO₂$ reference lines were used and at least fifteen sets of measurements were recorded and averaged to obtain the FIR laser frequencies. In general, the fundamentals $(n_1 = 1, n_2 = 1)$ and first harmonics $(n_1 = 2, n_2 = 2)$ were used to generate the beat frequency; however, for wavelengths shorter than $60.0 \,\mathrm{\upmu m}$, only the first harmonics could be used.

The reported offset was the difference measured between the CH2DOH absorption frequency and the center frequency of the $CO₂$ pump emission. The $CO₂$ pump line was set to obtain optimal FIR power and was then mixed (in the MIM diode) with the same $CO₂$ emission, generated by the $CO₂$ reference laser, locked to its center frequency. As with FIR laser frequency measurements, the difference between these laser emissions creates a beat frequency in the microwave region. The beat was observed and measured on the spectrum analyzer with an uncertainty

of either ± 10 , ± 15 or ± 20 MHz. This uncertainty was mainly derived from the reproducibility of these measurements.

3 Results and discussion

The measured FIR laser frequencies and their offsets, along with new offset measurements for previously measured FIR laser frequencies, are listed in Tables 1 and 2, respectively, arranged in order of the $CO₂$ pump line. All new frequency measurements are reported with their corresponding wavelength and wavenumber, calculated from the average frequency using $1 \text{ cm}^{-1} = 29979.2458 \text{ MHz}$, as well as with their offset and respective reference. All FIR laser frequencies were measured under optimal operating conditions.

4 Conclusions

This work reports the discovery of the $64.0 \,\mu m$ FIR laser line from CH2DOH when optically pumped by the $9P32$ CO₂ emission. The frequency of this line, along with the frequencies of ten previously observed FIR laser emissions generated by optically pumped $CH₂DOH$, has been meas-

^a Measured with four sets of reference lasers

^b Measured with two sets of reference lasers

^c Measured with three sets of reference lasers

^d Observed in the parallel polarization at 13 Pa as a weak (0.01 to 0.001 mW) line

TABLE 1 Measured FIR laser frequencies generated by optically pumped CH₂DOH

^a All FIR laser frequencies were previously reported in [2]

TABLE 2 New offset measurements for previously measured FIR laser frequencies generated by optically pumped CH2DOH

FIGURE 1 Observed beat frequency at −23 782.5 MHz between the $49.0 \,\mu m$ FIR emission of $CH₂DOH$ (generated by the 10*P*46 pump line) and the 9*R*40 and 10*R*40 CO₂ reference emissions

ured with fractional uncertainties on the order of $\pm 2 \times 10^{-7}$. In addition, the offset frequencies for 22 CH₂DOH laser emissions have been measured. These measurements will be useful for future assignments of FIR laser emissions by calculation of combination loops from high-resolution Fourier transform data [13]. Not only will these data be helpful in the theoretical modeling of $CH₂DOH$, these emissions can serve as sources of coherent radiation for a variety of spectroscopic investigations, including laser magnetic resonance spectroscopy.

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