



# Enhanced ferroelectric properties of BiFeO<sub>3</sub> thin films utilizing four buffer layers: Nd<sub>2</sub>O<sub>3</sub>, Eu<sub>2</sub>O<sub>3</sub>, Ho<sub>2</sub>O<sub>3</sub>, and Er<sub>2</sub>O<sub>3</sub>

Tung-Ming Pan<sup>1,2</sup> · Zhong-Yi Chen<sup>1</sup> · Jim-Long Her<sup>3</sup>

Received: 11 May 2024 / Accepted: 5 July 2024 / Published online: 16 July 2024  
© The Author(s), under exclusive licence to Springer-Verlag GmbH Germany, part of Springer Nature 2024

## Abstract

This paper investigates the ferroelectric and structural properties of BiFeO<sub>3</sub> thin films with four different RE<sub>2</sub>O<sub>3</sub> (Nd<sub>2</sub>O<sub>3</sub>, Eu<sub>2</sub>O<sub>3</sub>, Ho<sub>2</sub>O<sub>3</sub>, and Er<sub>2</sub>O<sub>3</sub>) buffer layers fabricated on a SrRuO<sub>3</sub>/n<sup>+</sup>-Si substrate through spin-coating. To analyze the BiFeO<sub>3</sub> films with RE<sub>2</sub>O<sub>3</sub> buffer layers, various techniques, such as X-ray diffraction, secondary ion mass spectrometry, atomic force microscope, and X-ray photoelectron spectroscopy were employed to investigate the crystalline structures, depth profiles, surface topographies, and chemical compositions. It was found that the BiFeO<sub>3</sub> film with RE<sub>2</sub>O<sub>3</sub> buffer layers exhibited improved electrical properties such as leakage current, remnant polarization, and coercive field compared to the control BiFeO<sub>3</sub> film without a buffer layer. Moreover, the Eu<sub>2</sub>O<sub>3</sub> buffer layer exhibited the lowest leakage current of  $2.05 \times 10^{-6}$  A/cm<sup>2</sup>, the highest remnant polarization of 43.76 μC/cm<sup>2</sup>, and the smallest coercive field of 188 kV/cm among all the RE<sub>2</sub>O<sub>3</sub> buffer layers. The outcome is likely to have been caused by the introduction of Eu<sup>3+</sup> ion to the BiFeO<sub>3</sub> film, which resulted in a reduction in surface roughness, a significant preferred orientation of (110), and an increased concentration of Fe<sup>3+</sup> ion. Consequently, this inhibited the fluctuation of Fe<sup>3+</sup> to Fe<sup>2+</sup> ions and reduced the occurrence of oxygen vacancies.

**Keywords** BiFeO<sub>3</sub> · Nd<sub>2</sub>O<sub>3</sub> · Eu<sub>2</sub>O<sub>3</sub> · Ho<sub>2</sub>O<sub>3</sub> · Er<sub>2</sub>O<sub>3</sub> · Buffer layers · Sol–gel method

## 1 Introduction

In recent decades, BiFeO<sub>3</sub> films have garnered significant attention for their potential use in multifunctional electronic devices. This is largely due to their possession of both ferroelectricity, which occurs at a relatively high Curie temperature of approximately 830 °C, and ferromagnetism, which exhibits a high Néel temperature of around 370 °C [1–4]. Ferroelectric ABO<sub>3</sub> perovskites, represented by compounds such as BaTiO<sub>3</sub>, PbZr<sub>0.53</sub>Ti<sub>0.47</sub>O<sub>3</sub>, SrTiO<sub>3</sub>, and LaAlO<sub>3</sub> [5–7], are highly regarded materials recognized for their unique perovskite crystal structure. This structure is characterized by A-site cations at the corners, B-site cations at the center, and a surrounding network of oxygen anions. Typically

represented by the formula ABX<sub>3</sub>, where A and B are cations and X is an anion, commonly oxygen [8–10]. BiFeO<sub>3</sub> stands out among ferroelectric materials due to its exceptional combination of multiferroic properties, high Curie temperature, chemical durability, robust piezoelectricity, and tunable characteristics. These attributes make BiFeO<sub>3</sub> a highly promising material for a wide range of technological applications, including spintronics, sensors, actuators, data storage, and energy conversion devices [11–13].

The presence of active lone-pair electrons in BiFeO<sub>3</sub> is believed to generate ferroelectric polarization, as the 6s<sup>2</sup> orbitals electrons fill one of the resulting orbitals in the Bi<sup>3+</sup> ion (A site). In contrast to its magnetic property, the superexchange interactions of Fe<sup>3+</sup> (B site) and O<sup>2-</sup> ions result in antiferromagnetic behavior with G-type magnetic ordering [14]. However, the twisted rhombohedral perovskite structure of the BiFeO<sub>3</sub> film, which belongs to the R3c space group, exhibits a large leakage current and high coercive field [15, 16], limiting its potential applications. Researchers have pursued approaches such as ion substitution, process modification, and insertion of a buffer layer to overcome these limitations.

✉ Tung-Ming Pan  
tmpan@mail.cgu.edu.tw

<sup>1</sup> Department of Electronics Engineering, Chang Gung University, Taoyuan 33302, Taiwan, ROC

<sup>2</sup> Division of Urology, Linkou Medical Center, Chang Gung Memorial Hospital, Taoyuan 33305, Taiwan, ROC

<sup>3</sup> Division of Natural Science, Center for General Education, Chang Gung University, Taoyuan 33302, Taiwan, ROC

Incorporating transition and rare-earth (RE) metals into BiFeO<sub>3</sub> films has been suggested as a means to enhance leakage current performance. By substituting RE ions for Bi ions at the A site and transition ions for Fe ions at the B site, the loss of Bi<sup>3+</sup> ions can be prevented, and fluctuations in the valence state of Fe<sup>3+</sup> ions can be inhibited. Consequently, these actions may increase intrinsic polarization and decrease leakage current levels [1, 17–20]. Nevertheless, doping BiFeO<sub>3</sub> films with RE elements presents various challenges, including concerns regarding phase stability, fluctuations in charge carrier concentration, structural integrity, thermal behavior, uneven distribution of dopants, fabrication complexities, and compatibility issues during integration [21, 22]. Doping perovskites with RE ions results in the charge compensation, either through the formation of oxygen vacancies or alterations in the oxidation states of other cations in the compound. These vacancies significantly impact the structural, electronic, and functional properties of the film [23–25]. Additionally, the Bi layer-structured ferroelectrics play a crucial role in determining the electrical properties of these materials, significantly influencing ferroelectricity, dielectric behavior, piezoelectricity, conductivity, multiferroicity, thermal stability, and fatigue resistance [26, 27].

Several research teams have observed significant impacts on the characteristics of BiFeO<sub>3</sub> thin films based on the chosen buffer layer. Specifically, Zheng et al. noted that employing a SrRuO<sub>3</sub> buffer layer led to improved electrical properties of the BiFeO<sub>3</sub> thin film due to enhancements in surface morphology and crystallization [28]. Leu et al. demonstrated that the addition of a Bi<sub>2</sub>O<sub>3</sub> buffer layer effectively bolstered both the structural and electrical properties of the BiFeO<sub>3</sub> thin film [29]. Similarly, Tang et al. reported that the use of LaNiO<sub>3</sub> as a buffer layer enhanced the crystalline quality of BiFeO<sub>3</sub>, as it is structurally compatible with BiFeO<sub>3</sub> [30]. Finally, Cao et al. provided evidence that La<sup>3+</sup>-doped BiFeO<sub>3</sub> films deposited on Si substrates with LaNiO<sub>3</sub> as a buffer layer displayed a pronounced magneto-optical effect [31]. Additionally, research has explored the incorporation of RE ions into the CdSe or CdS films. The inclusion of Er in CdSe nanocrystals leads to notable changes in the lattice structure of Cd<sub>1-x</sub>Er<sub>x</sub>Se, thereby enhancing photoluminescence [32]. Furthermore, studies have examined the structural and optical properties of Cd<sub>1-x</sub>Eu<sub>x</sub>S thin films fabricated on glass through a chemical bath method [33].

Sol–gel spin-coating was selected for its versatility in depositing a range of materials, including oxides, nitrides, and composites. It is suitable for creating complex multilayer structures and heterostructures. This method produces uniform, high-quality thin films with precise compositional control, offering cost-effectiveness and scalability while allowing fine-tuning of film properties and morphology [34–36]. In this study, RE<sub>2</sub>O<sub>3</sub> film is selected as the buffer

layer for various reasons: (1) RE<sub>2</sub>O<sub>3</sub> film possesses large energy gap, high dielectric constant and good thermal stability; (2) RE cations can be substituted for the Bi<sup>3+</sup> ion in the BiFeO<sub>3</sub> film to modify the crystal structure; (3) RE<sub>2</sub>O<sub>3</sub> film can act like a sink for defects or oxygen vacancies in the buffer layer; and finally, (4) RE<sub>2</sub>O<sub>3</sub> film can hinder the electron injection from the bottom electrode. The present investigation delves into the production of BiFeO<sub>3</sub> thin films using a sol–gel spin-coating technique that operates at a low temperature. This process may be appropriate for Si process technology. The current literature lacks investigation of the ferroelectric behavior of BiFeO<sub>3</sub> thin film on different RE<sub>2</sub>O<sub>3</sub> buffer layers (Nd<sub>2</sub>O<sub>3</sub>, Eu<sub>2</sub>O<sub>3</sub>, Ho<sub>2</sub>O<sub>3</sub>, and Er<sub>2</sub>O<sub>3</sub>), and this research topic addresses this gap. The study focuses on the structural, surface morphological, depth profiles, film compositional, and ferroelectric properties of BiFeO<sub>3</sub> thin films with and without RE<sub>2</sub>O<sub>3</sub> buffer layer on SrRuO<sub>3</sub>. The results demonstrate that the incorporation of RE<sub>2</sub>O<sub>3</sub> films as a buffer layer can reduce leakage current and improve remanent polarization.

## 2 Experimental

Strontium nitrate Sr(NO<sub>3</sub>)<sub>2</sub> and ruthenium chloride hydrate RuCl<sub>3</sub>·xH<sub>2</sub>O were utilized as the primary raw materials for synthesizing SrRuO<sub>3</sub> film. Meanwhile, bismuth nitrate pentahydrate Bi(NO<sub>3</sub>)<sub>3</sub>·5H<sub>2</sub>O and iron nitrate Fe(NO<sub>3</sub>)<sub>3</sub>·9H<sub>2</sub>O were utilized for BiFeO<sub>3</sub> thin film synthesis. In addition, neodymium acetate hydrate Nd(CH<sub>3</sub>CO<sub>2</sub>)<sub>3</sub>·xH<sub>2</sub>O, europium acetate hydrate Eu(CH<sub>3</sub>CO<sub>2</sub>)<sub>3</sub>·xH<sub>2</sub>O, holmium acetate hydrate Ho(CH<sub>3</sub>CO<sub>2</sub>)<sub>3</sub>·xH<sub>2</sub>O, and erbium acetate hydrate Er(CH<sub>3</sub>CO<sub>2</sub>)<sub>3</sub>·xH<sub>2</sub>O were used for synthesizing different RE<sub>2</sub>O<sub>3</sub> buffer layers such as Nd<sub>2</sub>O<sub>3</sub>, Eu<sub>2</sub>O<sub>3</sub>, Ho<sub>2</sub>O<sub>3</sub>, and Er<sub>2</sub>O<sub>3</sub>. The chemical reagents were obtained from Sigma Aldrich and were combined in a specific ratio. For instance, BiFeO<sub>3</sub> was prepared using a 1:1 ratio of bismuth and iron nitrates with an additional 10% weight of bismuth nitrate to account for losses during the sol–gel process, while SrRuO<sub>3</sub> was prepared using a 1:1 ratio of strontium nitrate and ruthenium chloride hydrate.

To produce a high-quality precursor solution, a 1:1 mixture of 0.1 M strontium nitrate and 0.1 M ruthenium chloride hydrate was dissolved in 10 mL of ethanol. Separately, 0.25 M bismuth nitrate and 0.2 M iron nitrate were dissolved in 10 mL of ethylene glycol with constant stirring. The two solutions were then combined and stirred at 70 °C for 2 h. Citric acid was added as a stabilizer to adjust the solution's viscosity during gel formation. Next, 0.1 M neodymium acetate hydrate, 0.1 M europium acetate hydrate, 0.1 M holmium acetate hydrate, or 0.1 M erbium acetate hydrate was dissolved in 10 mL of nitric acid and stirred for 2 h. Citric

acid was used as a complexing agent to stabilize the metal-citrate complex in aqueous solutions.

First, the SrRuO<sub>3</sub> gel was spin-coated onto an n<sup>+</sup>-silicon (100) substrate at 200 rpm for 30 s, followed by 1000 rpm for 15 s. The substrate was heated on a hot plate at 200 °C for 3 min and annealed in N<sub>2</sub> gas at 550 °C for 1 h. Next, the RE<sub>2</sub>O<sub>3</sub> (Nd<sub>2</sub>O<sub>3</sub>, Eu<sub>2</sub>O<sub>3</sub>, Ho<sub>2</sub>O<sub>3</sub>, and Er<sub>2</sub>O<sub>3</sub>) chemical solutions were spin-coated on the SrRuO<sub>3</sub>/n<sup>+</sup>-Si substrate at 200 rpm for 10 s, followed by 2000 rpm for 20 s. The substrates were baked on a hot plate at 150 °C for 3 min and annealed in O<sub>2</sub> gas at 400 °C for 10 min. Subsequently, thin BiFeO<sub>3</sub> films were spin-coated onto the RE<sub>2</sub>O<sub>3</sub> buffer layers at 200 rpm for 10 s, followed by 3000 rpm for 20 s. These films were baked on a hot plate at 200 °C for 3 min and then annealed in O<sub>2</sub> gas at 400 °C for 10 min. Finally, all BiFeO<sub>3</sub>/RE<sub>2</sub>O<sub>3</sub>/SrRuO<sub>3</sub>/n<sup>+</sup>-Si substrate thin films were annealed at 600 °C for 1 h, and 50 nm-thick Pt top electrodes were deposited using a sputtering system with an area of 3.14 × 10<sup>-4</sup> cm<sup>2</sup> through a shadow mask. Figure 1a and b illustrate the schematic cross-sectional views of the BiFeO<sub>3</sub> film and the RE<sub>2</sub>O<sub>3</sub>-buffered BiFeO<sub>3</sub> film, respectively.

The BiFeO<sub>3</sub> thin films with four different RE<sub>2</sub>O<sub>3</sub> buffer layers were analyzed for their structural characteristics using various techniques, including X-ray diffraction (XRD), atomic force microscopy (AFM), X-ray photoelectron spectroscopy (XPS), and secondary ion mass spectrometry (SIMS). The crystallographic measurements were performed using a Rigaku D/MAX2000 XRD with a radiation wavelength (λ) of 1.5418 Å and a 2θ range of 10–90°, with a step size of 0.04°/s. The surface topography of the films was examined with a Solver P47-PRO SPM in tapping mode, and the surface roughness was determined from 3 × 3 μm<sup>2</sup> scan areas. The depth profiles of the films were analyzed using a ToF-SIMS IV/Ion-ToF system with an O<sub>2</sub><sup>+</sup> primary beam. The chemical bonding of the BiFeO<sub>3</sub> films with different RE<sub>2</sub>O<sub>3</sub> buffer layers was investigated using a Thermo Scientific ESCALAB XI<sup>+</sup> X-ray photoelectron spectrometer microscope. The leakage current density–electric field (J–E) characteristics and polarization–electric field (P–E) hysteresis loops of the BiFeO<sub>3</sub> thin films were measured using a Keysight B1500A semiconductor device analyzer and a

modified Sawyer–Tower circuit, respectively, for each of the four RE<sub>2</sub>O<sub>3</sub> buffer layers.

### 3 Results and discussion

#### 3.1 Structural properties of BiFeO<sub>3</sub> with four RE<sub>2</sub>O<sub>3</sub> buffer layers

The XRD patterns in Fig. 2a depict the BiFeO<sub>3</sub> thin films with and without various RE<sub>2</sub>O<sub>3</sub> buffer layers. The primary diffraction peak, indexed as a hexagonal BiFeO<sub>3</sub> structure with R3c space group [JCPDS: 71-2494], was observed in the control sample at (202). However, the formation of impurity and secondary phases is a common occurrence in the synthesis of BiFeO<sub>3</sub> films using solid-state and sol–gel methods, primarily due to imprecise temperature control

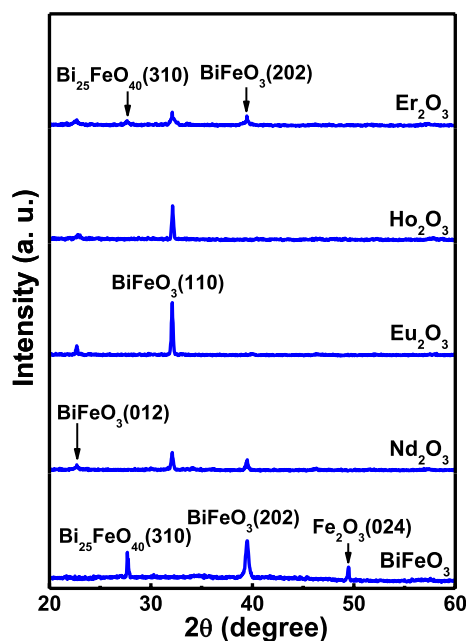
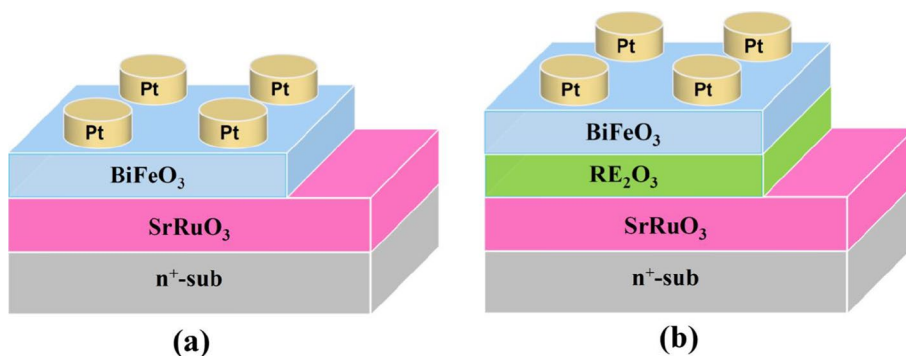


Fig. 2 XRD patterns of BiFeO<sub>3</sub> thin film and four different RE<sub>2</sub>O<sub>3</sub>-buffered BiFeO<sub>3</sub> films

Fig. 1 Schematic cross-section views of capacitive device structures using **a** BiFeO<sub>3</sub> and **b** RE<sub>2</sub>O<sub>3</sub>-buffered BiFeO<sub>3</sub> films



[37, 38]. The XRD analysis revealed the prominent BiFeO<sub>3</sub> (202) peak, accompanied by impurities (Fe<sub>2</sub>O<sub>3</sub>) and secondary phases (Bi<sub>25</sub>FeO<sub>40</sub>) [JCPDS: 33-0664 and 46-0416] present in the control sample. This contrasts with the findings of Zheng et al. [39] and Wu and Wang [40], who observed strong preferred BiFeO<sub>3</sub> (110) and notable (111) orientations, respectively, in their thin films. With the substrate temperature increasing from 450 to 620 °C, the overall crystallinity of the BiFeO<sub>3</sub> film significantly improved, as evidenced by the enhanced sharpness and peak intensity of the (111) peak in the XRD patterns. In the BiFeO<sub>3</sub> films with the RE<sub>2</sub>O<sub>3</sub> buffer layer, a more distinct BiFeO<sub>3</sub> (110) peak was observed, while it was absent in the film lacking the RE<sub>2</sub>O<sub>3</sub> buffer layer. The addition of the RE buffer layer prompted the formation of a single-phase BiFeO<sub>3</sub> structure during annealing at 600 °C, with RE ions occupying the A site within the crystal lattice, as verified by SIMS data. This process effectively eradicated the presence of multiple phases. This behavior can be ascribed to the smaller ionic radii of Nd<sup>3+</sup> (0.983 Å), Eu<sup>3+</sup> (0.947 Å), and Ho<sup>3+</sup> (0.901 Å) ions in comparison to Bi<sup>3+</sup> (1.03 Å) ions [41]. In the sample featuring an Er<sub>2</sub>O<sub>3</sub> buffer layer, three faint BiFeO<sub>3</sub> diffraction peaks were observed, suggesting a distorted perovskite-based hexagonal structure, accompanied by a relatively weak (310) peak corresponding to the Bi<sub>25</sub>FeO<sub>40</sub> secondary phase. However, these impurity and secondary phase peaks vanished in the Nd<sub>2</sub>O<sub>3</sub>, Eu<sub>2</sub>O<sub>3</sub>, and Ho<sub>2</sub>O<sub>3</sub>-buffered SrRuO<sub>3</sub> film, indicating the integration of trivalent Nd, Eu, and Ho ions into the BiFeO<sub>3</sub> crystal lattice. The BiFeO<sub>3</sub> film with the Nd<sub>2</sub>O<sub>3</sub> buffer layer exhibited three faint peak intensities, while those with the Eu<sub>2</sub>O<sub>3</sub> and Ho<sub>2</sub>O<sub>3</sub> buffer layers showed two distinct diffraction peaks at 22.74~22.82 and 32.12~32.20 corresponding to (012) and (110). Additionally, the BiFeO<sub>3</sub> films with the Nd<sub>2</sub>O<sub>3</sub>, Eu<sub>2</sub>O<sub>3</sub>, and Ho<sub>2</sub>O<sub>3</sub> buffer layers demonstrated characteristic peaks slightly shifted towards higher diffraction angles. The introduction of RE elements aided in reducing oxygen vacancies and suppressing the formation of impurity phases associated with the oxidational states of Fe<sup>3+</sup> [20], without inducing any structural transitions.

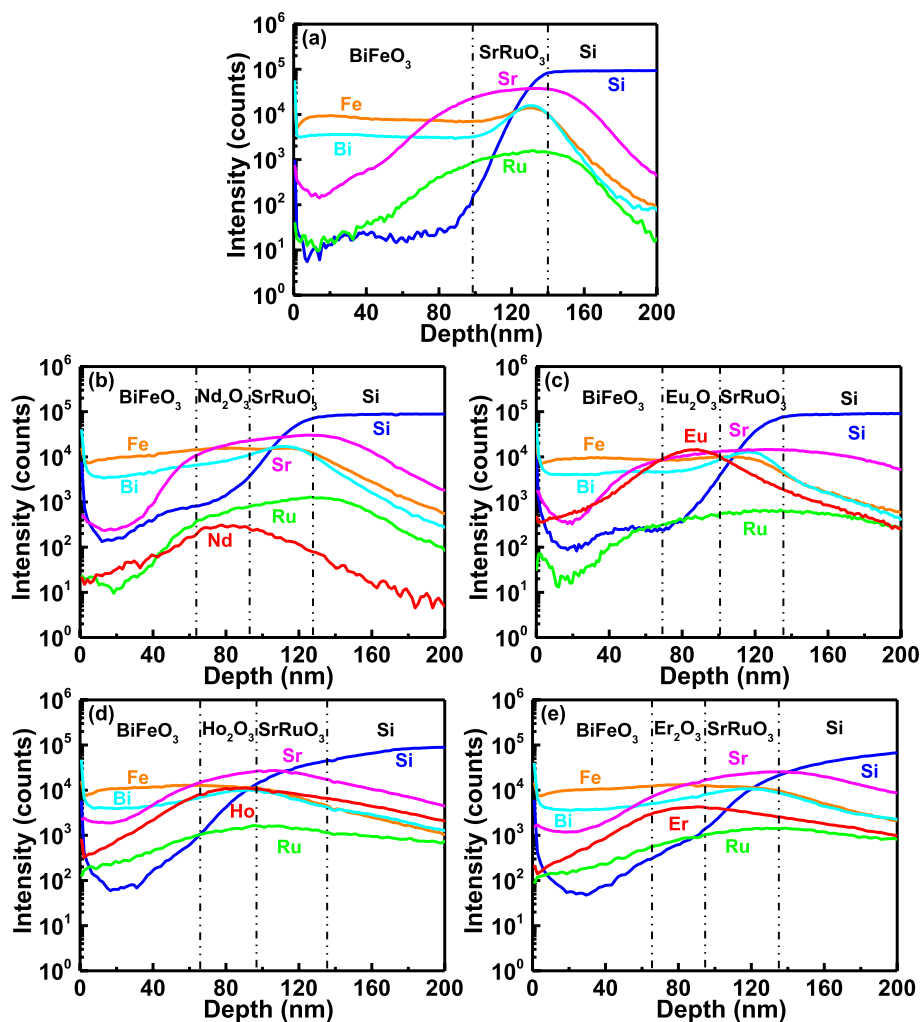
SIMS analysis was employed to examine the influence of the RE<sub>2</sub>O<sub>3</sub> buffer layer on the depth distributions of elements across the thickness of BiFeO<sub>3</sub> thin films. Figure 3a gives that the distribution of Bi and Fe in the BiFeO<sub>3</sub> film was uniform, except at the interface with the SrRuO<sub>3</sub> electrode, where a notable decrease in Fe and Bi concentrations occurred. This implies significant inward diffusion of Fe and Bi ions at the SrRuO<sub>3</sub>/Si interface, consistent with previous research [42] that emphasizes substantial inward diffusion of Bi and Fe into the Si substrate in SIMS profiles. The presence of an RE<sub>2</sub>O<sub>3</sub> buffer layer notably affects the distribution of elements in the BiFeO<sub>3</sub> thin film. Figure 3b–e show that the Bi ion intensities in samples with various RE<sub>2</sub>O<sub>3</sub> buffer

layers were higher than in the control sample, indicating different crystallinity. This outcome can be attributed to the elevated concentration of Bi, which accelerates volatile or out-diffusion of Bi from the unstable buffer layer, leading to the formation of a defective film structure. Moreover, a small amount of RE ions in the buffer layer diffused gradually and incorporated into the BiFeO<sub>3</sub>, resulting in a consistently high concentration of Bi. While the samples with different RE<sub>2</sub>O<sub>3</sub> buffer layers exhibited distinct properties, their distributions of Bi and Fe elements were similar. It was observed that the Bi ions accumulated in the SrRuO<sub>3</sub> film, except for the Ho<sub>2</sub>O<sub>3</sub> buffer layer. The RE buffer layers not only facilitated the formation of the (110) orientation but also hindered the crystallization of Bi-rich phases. Moreover, there was minor inter-diffusion between BiFeO<sub>3</sub> and RE<sub>2</sub>O<sub>3</sub> due to the low deposition temperature of the BiFeO<sub>3</sub> film. The sample featuring the Eu<sub>2</sub>O<sub>3</sub> buffer layer exhibited the highest Bi ion intensity, likely due to the incorporation of Eu ions into the BiFeO<sub>3</sub> film, which suppressed the out-diffusion of Bi ions. Conversely, the Er<sub>2</sub>O<sub>3</sub> buffer layer facilitated a localized high Bi concentration, enhancing BiFeO<sub>3</sub> crystallization; however, its instability resulted in the formation of a structurally defective film with a Bi-depleted surface. Thus, we suggest employing BiFeO<sub>3</sub> thin films with RE<sub>2</sub>O<sub>3</sub> buffer layers, as they facilitate the nucleation and growth of the (110) texture structure with reduced lattice distortions.

Figure 4 depicts the surface morphologies of BiFeO<sub>3</sub> thin films examined using AFM with and without various RE<sub>2</sub>O<sub>3</sub>. The control BiFeO<sub>3</sub> film had a surface roughness of 12.6 nm, while films with Nd<sub>2</sub>O<sub>3</sub>, Eu<sub>2</sub>O<sub>3</sub>, Ho<sub>2</sub>O<sub>3</sub>, and Er<sub>2</sub>O<sub>3</sub> buffer layers showed roughness values of 7.67, 5.77, 6.97, and 8.56 nm, respectively. In contrast, the BiFeO<sub>3</sub> film deposited on the SrRuO<sub>3</sub> electrode had an uneven, flaky, and micro-cracked surface. However, the use of RE<sub>2</sub>O<sub>3</sub> buffer layers resulted in notable differences in surface morphology, with flatter surfaces and smaller grain sizes observed in the RE<sub>2</sub>O<sub>3</sub> buffered-BiFeO<sub>3</sub> films. The presence of minor grains in the BiFeO<sub>3</sub> films on the RE<sub>2</sub>O<sub>3</sub> buffer layer is attributed to the high number of nucleation sites in the buffer layer, resulting in lower R<sub>rms</sub> values. It is noteworthy that the RE<sub>2</sub>O<sub>3</sub> buffer layer visibly improved the surface morphology by smoothing it out, with reduced size and number of hills when Bi is substituted by RE ions. The bond energy of RE–O is stronger than those of Fe–O and Bi–O, as per the Pauling electronegativity concept. The electronegativity values of Bi, Fe, Nd, Eu, Ho, Er, and O are 2.02, 1.83, 1.14, 1.2, 1.23, 1.24, and 3.44, respectively. Consequently, more formation heat is expelled during film growth, leading to a more stable structure.

To comprehend the ferroelectric characteristics of BiFeO<sub>3</sub> films, it is essential to assess the variable oxidation state of Fe ions between Fe<sup>2+</sup> and Fe<sup>3+</sup>, which significantly influences the properties of the film. To achieve this,

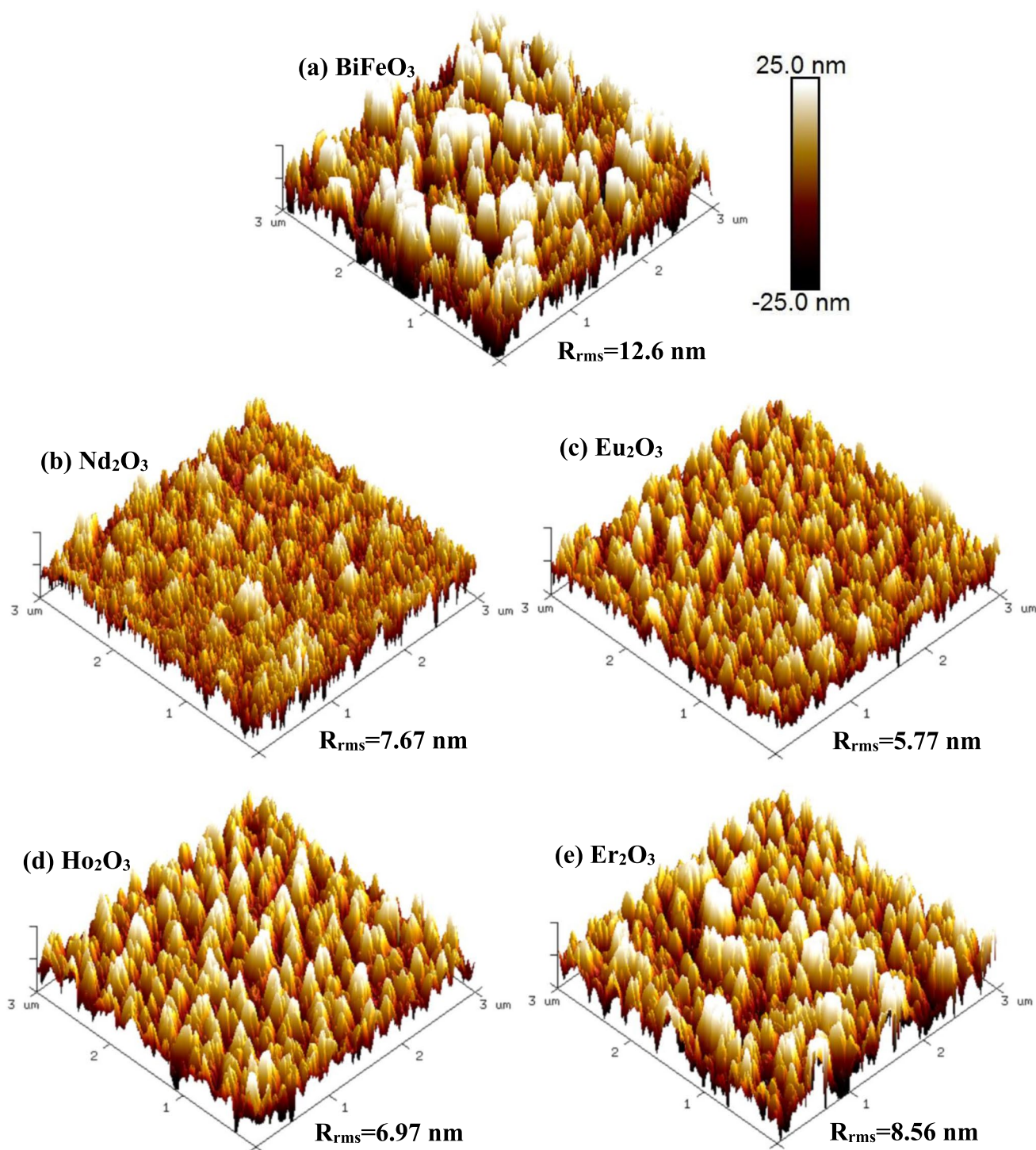
**Fig. 3** SIMS depth profile of **a** control BiFeO<sub>3</sub> film and that with **b** Nd<sub>2</sub>O<sub>3</sub>, **c** Eu<sub>2</sub>O<sub>3</sub>, **d** Ho<sub>2</sub>O<sub>3</sub>, and **e** Er<sub>2</sub>O<sub>3</sub> buffer layers



the oxidation states of Fe ions in control BiFeO<sub>3</sub> films and RE<sub>2</sub>O<sub>3</sub>-buffered BiFeO<sub>3</sub> films were assessed by analyzing the narrow scan XPS spectra of Bi 4f, Fe 2p, and O 1s lines, as shown in Fig. 5a–c. The calibration of the core-level photoelectron binding energies was achieved by using the binding energy of the C 1s photoelectron at 285 eV. Following the subtraction of the Shirley-type background, the spectra were deconvoluted using Gaussian-Lorentz functions. The Bi doublet in the spectrum of the control sample consisted of two peaks at 159.1 and 164.4 eV, with a spin–orbit splitting of 5.3 eV, attributed to the Bi–O bonds. The RE<sub>2</sub>O<sub>3</sub> buffer layer caused a slight shift in the Bi 4f<sub>7/2</sub> and Bi 4f<sub>5/2</sub> peaks towards higher binding energies (0.1–0.2 eV), indicating substitution of RE<sup>3+</sup> ions at Bi<sup>3+</sup> sites in the BiFeO<sub>3</sub> lattice. The chemical shift in Bi 4f two peaks may be due to the variation in electronegativity values of the elements Bi, Fe, Nd, Eu, Ho, Er, and O. The covalency/ionicity of Bi–O, Fe–O, Nd–O, Eu–O, Ho–O, and Er–O bonds were calculated for the samples with and without the RE<sub>2</sub>O<sub>3</sub> buffer layer. The fraction of covalency ( $F_c$ ) was defined as  $F_c = \exp(-(\Delta E_N)^2/4)$ , where  $\Delta E_N$  is the difference in

electronegativity value between the anion and cation, while the fraction of ionicity was estimated by  $F_i = (1 - F_c)$  [43]. Based on the electronegativity values of Bi, Fe, Nd, Eu, Ho, Er, and O elements mentioned above, the  $F_c$  values of Bi–O, Fe–O, Nd–O, Eu–O, Ho–O, and Er–O bonds were calculated to be 0.6, 0.52, 0.27, 0.29, 0.29, and 0.3, respectively, while those of  $F_i$  values were estimated to be 0.4, 0.48, 0.73, 0.71, 0.71, and 0.7. The ionicity value of the RE–O bond is much greater than that of the Bi–O bond, suggesting that the bonding energy of the RE–O bond in the oxygen octahedron is higher than that of the Bi–O bond, leading to a slight shift of the 4f<sub>7/2</sub> and Bi 4f<sub>5/2</sub> peaks towards higher binding energies.

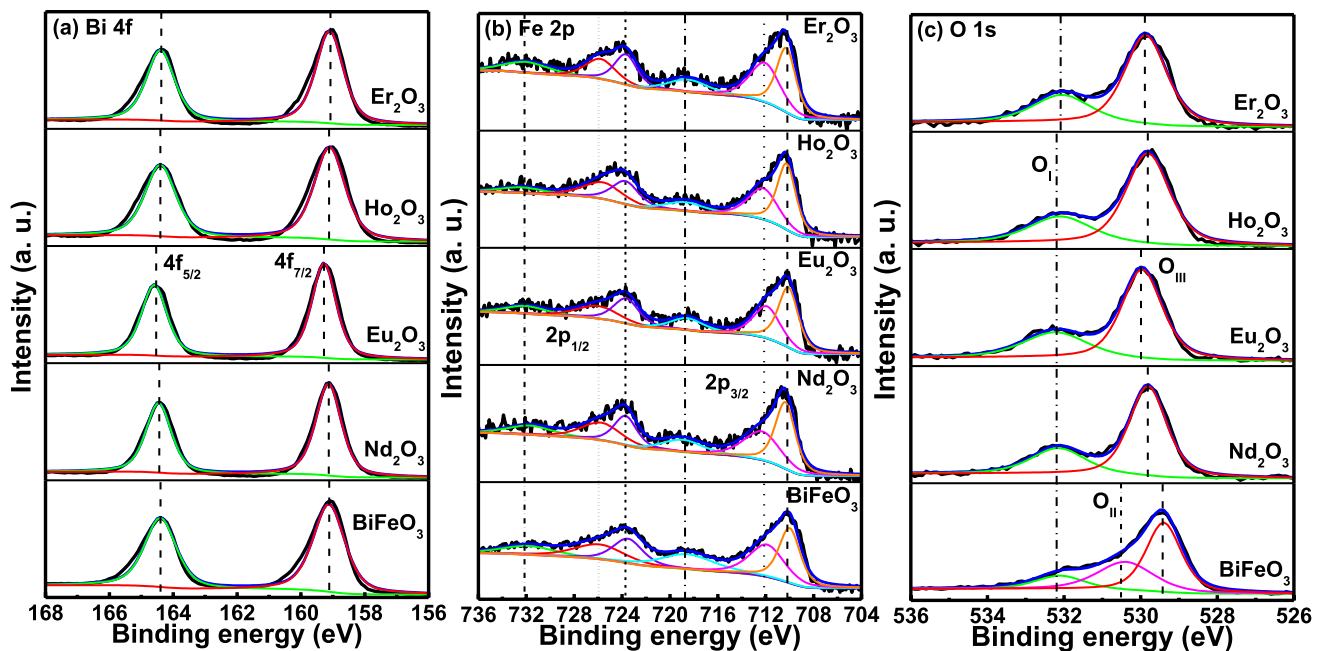
The Fe 2p XPS core spectra for the control BiFeO<sub>3</sub> sample and BiFeO<sub>3</sub> samples with RE<sub>2</sub>O<sub>3</sub> buffer layers are presented in Fig. 5b. The peaks at ~710.1 and ~723.8 eV were observed for Fe<sup>3+</sup> oxidation state, while those at ~712.1 and ~726 eV were assigned to Fe<sup>2+</sup> oxidation state [44]. Additionally, satellite peaks for 2p<sub>3/2</sub> and 2p<sub>1/2</sub> were observed at ~718.5 and ~732.1 eV, respectively, which are characteristic of the Fe oxidation state. The compositional ratio of Fe<sup>2+</sup>/Fe<sup>3+</sup> was calculated as 0.98, 0.87, 0.74, 0.84



**Fig. 4** AFM images of **a** control BiFeO<sub>3</sub> film and BiFeO<sub>3</sub> film with **b** Nd<sub>2</sub>O<sub>3</sub>, **c** Eu<sub>2</sub>O<sub>3</sub>, **d** Ho<sub>2</sub>O<sub>3</sub>, and **e** Er<sub>2</sub>O<sub>3</sub> buffer layers

and 0.91 for the control BiFeO<sub>3</sub> sample and those with Nd<sub>2</sub>O<sub>3</sub>, Eu<sub>2</sub>O<sub>3</sub>, Ho<sub>2</sub>O<sub>3</sub>, and Er<sub>2</sub>O<sub>3</sub> buffer layers, respectively, using curve fitting. The BiFeO<sub>3</sub> samples with RE<sub>2</sub>O<sub>3</sub> buffer layers showed a lower Fe<sup>2+</sup>/Fe<sup>3+</sup> compositional ratio compared to the control sample. The Eu<sub>2</sub>O<sub>3</sub>-buffered BiFeO<sub>3</sub>

film demonstrated the lowest compositional ratio among the buffer layers, possibly due to the incorporation of Eu<sup>3+</sup> ion, which enhances the crystallization behavior of BiFeO<sub>3</sub>, suppresses the volatilization of Bi<sup>3+</sup> ion, and reduces the oxygen vacancy. Figure 5c gives the de-convoluted O 1 s peak into



**Fig. 5** XPS spectra of **a** Bi 4f, **b** Fe 2p, and **c** O 1s for BiFeO<sub>3</sub> thin film and four different RE<sub>2</sub>O<sub>3</sub>-buffered BiFeO<sub>3</sub> films

three peaks for the BiFeO<sub>3</sub> thin film: higher binding (O<sub>I</sub>) energy, medium binding (O<sub>II</sub>) energy, and lower binding (O<sub>III</sub>) energy. The peaks at 532.2, 530.5, and 529.4 eV were assigned to the chemisorbed oxygen related to the hydroxyl group, oxygen vacancy, and lattice oxygen, respectively [44, 45]. The presence of absorbed water is relevant to the hydroxyl group. However, the BiFeO<sub>3</sub> thin films with four RE<sub>2</sub>O<sub>3</sub> buffer layers had only two O<sub>I</sub> and O<sub>III</sub> peaks at ~532.1 and ~529.5 eV, which were associated with lattice oxygen and the hydroxyl group, respectively. The RE<sub>2</sub>O<sub>3</sub>-buffered sample prevents the creation of oxygen vacancies.

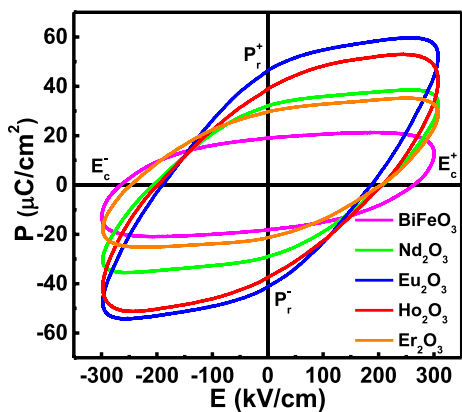
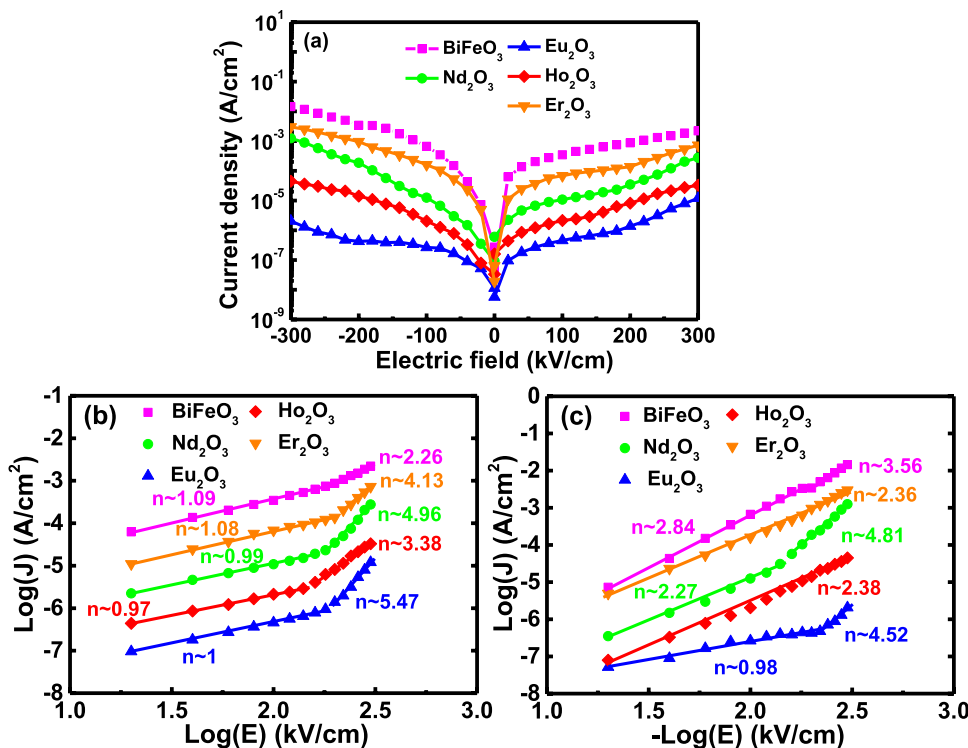
### 3.2 Ferroelectric characteristics of BiFeO<sub>3</sub> capacitive devices with and without RE<sub>2</sub>O<sub>3</sub> buffer layers

Figure 6a displays the J-E curves of BiFeO<sub>3</sub> capacitive devices with and without RE<sub>2</sub>O<sub>3</sub> buffer layers. The control BiFeO<sub>3</sub> sample exhibited a measured leakage current density of  $2.21 \times 10^{-3}$  A/cm<sup>2</sup> and  $1.45 \times 10^{-2}$  A/cm<sup>2</sup> at positive and negative applied electric fields of 300 kV/cm, respectively. At the same applied electric field, the RE<sub>2</sub>O<sub>3</sub>-buffered film showed a significantly reduced leakage current density, at least one order of magnitude lower than that of the control BiFeO<sub>3</sub> sample. The main origin of the high leakage current in the BiFeO<sub>3</sub> film is believed to be oxygen vacancies resulting from Bi deficiency, fluctuating chemical valence of Fe ion, and different defects including cracks, pores, and interstices [44]. The SIMS data reveals the detection of a depletion layer containing Bi and Fe elements near the surface of the SrRuO<sub>3</sub>/Si substrate. This occurrence might

be attributed to the presence of the capping layer, initially abundant in Bi. Such a layer, potentially comprising secondary phases or defective structures, has the potential to influence the ferroelectric properties of the film. The decrease in leakage current observed in the RE<sub>2</sub>O<sub>3</sub>-buffered films can be attributed to multiple factors, such as the decrease in Fe valence fluctuation as well as the oxygen vacancies and the inhibition of Bi volatilization caused by the substitution of Bi ions with RE ions. Moreover, the incorporation of RE ions into the BiFeO<sub>3</sub> film also contributed to the improvement in surface morphology, resulting in a denser microstructure and fewer interstices, which are favorable for the improvement in leakage current density. Additionally, among the buffer layers, the BiFeO<sub>3</sub> film with the Eu<sub>2</sub>O<sub>3</sub> buffer layer exhibited the lowest leakage current density of  $1.20 \times 10^{-5}$  A/cm<sup>2</sup> and  $2.05 \times 10^{-6}$  A/cm<sup>2</sup> at 300 kV/cm and -300 kV/cm, respectively. This outcome may be attributed to the higher Fe<sup>3+</sup> ion content of the Eu<sub>2</sub>O<sub>3</sub>-buffered layer, resulting in a smoother surface, reduced volatilization of Bi ions, and stabilization of the hexagonal structure of BiFeO<sub>3</sub>.

Understanding the leakage current behavior in all samples may require considering various conduction mechanisms such as Schottky emission at the interface, Fowler–Nordheim (FN) tunneling at the interface, space-charge-limited conduction (SCLC), and Poole–Frenkel (FP) emission limited to the bulk. When a positive electric field is applied, the control BiFeO<sub>3</sub> sample and RE<sub>2</sub>O<sub>3</sub>-buffered samples exhibit leakage currents, as shown in Fig. 6b where plotting log(J) versus log(E) suggests SCLC as the dominant leakage current process. At low electric fields, the leakage

**Fig. 6** **a** J-E curves of BiFeO<sub>3</sub> capacitive devices with and without four different RE<sub>2</sub>O<sub>3</sub> buffer layers. Plots of log(J) versus log(E) for capacitive devices using BiFeO<sub>3</sub> thin film and four different RE<sub>2</sub>O<sub>3</sub>-buffered BiFeO<sub>3</sub> films at **b** positive electric field and **c** negative electric field



**Fig. 7** P-E hysteresis loops of capacitive devices using BiFeO<sub>3</sub> thin film and four different RE<sub>2</sub>O<sub>3</sub>-buffered BiFeO<sub>3</sub> films

current behavior of all samples can be explained by Ohmic conduction, while at high electric fields, SCLC is believed to dominate the conduction. Figure 6c shows that SCLC is likely the leakage current mechanism in both BiFeO<sub>3</sub> films with and without the RE<sub>2</sub>O<sub>3</sub> buffer layer.

At room temperature and 1 kHz, the P-E hysteresis loops of the control BiFeO<sub>3</sub> film and RE<sub>2</sub>O<sub>3</sub>-buffered BiFeO<sub>3</sub> films were measured, and the results are presented in Fig. 7. The maximum electric field applied was approximately 300 kV/cm. The P-E loop of the control BiFeO<sub>3</sub> thin film had a roundish shape, indicating a high leakage current density. The remanent polarization (P<sub>r</sub>) values of the BiFeO<sub>3</sub> film

without and with the Nd<sub>2</sub>O<sub>3</sub>, Eu<sub>2</sub>O<sub>3</sub>, Ho<sub>2</sub>O<sub>3</sub>, and Er<sub>2</sub>O<sub>3</sub> buffer layers were about 18.61, 30.47, 43.76, 38.17, and 25.46 μC/cm<sup>2</sup>, respectively, while the coercive field (E<sub>c</sub>) values were approximately 264, 207, 188, 203, and 226 kV/cm, respectively. The RE<sub>2</sub>O<sub>3</sub> buffer layer significantly improved the ferroelectric behavior of the BiFeO<sub>3</sub> thin film by reducing the leakage current density. The Eu<sub>2</sub>O<sub>3</sub>-buffered layer showed the lowest coercive field and the largest remanent polarization among the RE<sub>2</sub>O<sub>3</sub> buffer layers. The increase in polarization for the Eu<sub>2</sub>O<sub>3</sub>-buffered layer was attributed to the lowest current density and the highest degree of (110)-preferred orientation.

### 4 Conclusion

A comprehensive series of experiments was undertaken to explore the characteristics of BiFeO<sub>3</sub> thin films, each featuring a unique RE<sub>2</sub>O<sub>3</sub> buffer layer. These films were fabricated through a simple spin-coating method on SrRuO<sub>3</sub>/n<sup>+</sup>-Si substrates. Examination of XRD data confirmed a distinct (110) preferred orientation within the RE<sub>2</sub>O<sub>3</sub> buffer layer, free from any impurities or secondary phases. This finding was reinforced by SIMS depth profiling, which revealed a substantial concentration of RE ions within the BiFeO<sub>3</sub> film. Additionally, AFM imaging revealed the smooth surface of the RE<sub>2</sub>O<sub>3</sub> buffer layer, adorned with small, flake-like structures free of any cracks. Further analysis using XPS demonstrated a notable Fe<sup>3+</sup>/Fe<sup>2+</sup> ratio and a suppression of oxygen



vacancies in the RE<sub>2</sub>O<sub>3</sub> buffer layers. These structural attributes were found to have a direct correlation with the ferroelectric properties of the BiFeO<sub>3</sub> thin films. Remarkably, the films featuring RE<sub>2</sub>O<sub>3</sub> buffer layers exhibited superior electrical characteristics compared to the control film. Notably, the Eu<sub>2</sub>O<sub>3</sub> buffer layer demonstrated exceptional performance, boasting the lowest leakage current of  $2.05 \times 10^{-6}$  A/cm<sup>2</sup>, the highest remnant polarization of 43.76 μC/cm<sup>2</sup>, and the smallest coercive field of 188 kV/cm. These outstanding attributes stem from a combination of factors, including low surface roughness, robust (110) orientation, increased Fe<sup>3+</sup> content, and minimized oxygen vacancies. Consequently, the Eu<sub>2</sub>O<sub>3</sub> buffer layer holds significant promise for enhancing the ferroelectric properties of BiFeO<sub>3</sub> thin films, making them highly suitable for a diverse range of multifunctional applications.

**Acknowledgements** The authors would like to thank their appreciation to the entire staff of the Thin Film Measurement Lab, Chang Gung University for their technical help. This work was supported by Ministry of Science and Technology (MOST) of Taiwan under contract of MOST 109-2221-E-182-028.

**Author contributions** All authors contributed to the study conception and design. Material preparation, data collection, and analysis were performed by TM, ZY and JL. TM and ZY contributed significantly to analysis and manuscript preparation. The first draft of the manuscript was written by TM. All authors read and approved the final manuscript.

**Data availability** The data presented in this study are available on request from the corresponding author.

## Declarations

**Conflict of interest** The authors declare that they have no conflict of interest.

## References

1. A. Billah, Y. Matsuno, A.N. Anju, K. Koike, S. Kubota, F. Hirose, B. Ahmad, Unusual behavior of magnetic coercive fields with temperature and applied field in La-doped BiFeO<sub>3</sub> ceramics. *ACS Appl. Electron. Mater.* **5**, 4261–4267 (2023)
2. R.S. Viswajit, K. Ashok, K.B. Jinesh, Tailoring of charge carriers with deposition temperature in pulsed laser deposited BiFeO<sub>3</sub> thin films. *Appl. Surf. Sci.* **661**, 160016 (2024)
3. S. Ratha, M. Kuppan, G. Egawa, S. Yoshimura, Excellent magnetic properties in multiferroic BiFeO<sub>3</sub> based thin films for magnetic devices application. *Nano-Struct. Nano-Objects* **35**, 101007 (2023)
4. X. Gao, L. Dai, Y. Liu, K. Wang, D.V. Karpinsky, L. Liu, Y. Wang, Reliable ferroelectricity in sol–gel-derived BiFeO<sub>3</sub> thin films below 200 nm. *J. Am. Ceram. Soc.* **107**, 3301–3312 (2024)
5. E. Hannachi, M.I. Sayyed, K.A. Mahmoud, Y. Slimani, S. Akhtar, B. Albarzan, A.H. Almuqrin, Impact of tin oxide on the structural features and radiation shielding response of some ABO<sub>3</sub> perovskites ceramics (A = Ca, Sr, Ba; B = Ti). *Appl. Phys. A* **127**, 970 (2021)
6. M. Reda, E.E. Ateia, S.I. El-Dek, M.M. Arman, New insights into optical properties, and applications of Zr-doped BaTiO<sub>3</sub>. *Appl. Phys. A* **130**, 240 (2024)
7. T.-M. Pan, H.-C. Wang, J.-L. Her, Structural and electrical characteristics of high-performance stacked YbTi<sub>x</sub>O<sub>y</sub>/PbZr<sub>0.53</sub>Ti<sub>0.47</sub>O<sub>3</sub> gate dielectrics for InGaZnO thin-film transistors. *J. Alloys Compd.* **842**, 155844 (2020)
8. M. Sk, Recent progress of lead-free halide double perovskites for green energy and other applications. *Appl. Phys. A* **128**, 462 (2022)
9. S. Supriya, Crystal structure engineered non-toxic Bi<sub>0.5</sub>Na<sub>0.5</sub>TiO<sub>3</sub> based thin films- fabrication process, enhanced electrical performance, challenges and recent reports. *J. Inorg. Organomet. Polym. Mater.* **33**, 3013–3026 (2023)
10. S. Supriya, Electric field assisted spark plasma sintering of ABO<sub>3</sub> perovskites: crystal structure, dielectric behavior and future challenges, open ceramics. *Open Ceramics* **18**, 100608 (2024)
11. L. You, A. Abdelsamie, Y. Zhou, L. Chang, Z.S. Lim, J. Wang, Revisiting the ferroelectric photovoltaic properties of vertical BiFeO<sub>3</sub> capacitors: a comprehensive study. *ACS Appl. Mater. Interfaces* **15**, 12070–12077 (2023)
12. X. Tan, X. Sun, J. Jiang, D. Chen, Improved polarization retention in epitaxial BiFeO<sub>3</sub> thin films induced by strain relaxation. *Appl. Surf. Sci.* **635**, 157703 (2023)
13. Z. Wang, X. Yang, X. He, H. Xue, X. Wang, H. Dong, J. Zhu, W. Mao, X. Xu, X. Li, Roles of oxygen vacancy and ferroelectric polarization in photovoltaic effects of BiFeO<sub>3</sub> based devices. *Solid State Commun.* **360**, 115042 (2023)
14. P. Ravindran, R. Vidya, A. Kjekshus, H. Fjellvåg, Theoretical investigation of magnetoelectric behavior in BiFeO<sub>3</sub>. *Phys. Rev. B* **74**, 224412 (2006)
15. F. Kubel, H. Schmid, Structure of a ferroelectric and ferroelastic monodomain crystal of the perovskite BiFeO<sub>3</sub>. *Acta Crystallogr. B* **46**, 698–702 (1990)
16. H.W. Shin, J.Y. Son, Leakage current characteristics of polycrystalline BiFeO<sub>3</sub> thin films affected by thickness-dependent domain wall currents. *J. Alloys Compd.* **968**, 172113 (2023)
17. T. Ahmad, K. Jindal, M. Tomar, P.K. Jha, Effect of codoping of rare earth elements and Cr on multiferroic, optical and photocatalytic properties of BiFeO<sub>3</sub>. *Mater. Today Commun.* **37**, 107516 (2023)
18. S. Kumari, K. Anand, M. Alam, L. Ghosh, S. Dixit, R. Singh, A.K. Jain, S.M. Yusuf, C. Gautam, A.K. Ghosh, A. Mohan, S. Chatterjee, Enhancement of multiferroic and optical properties in BiFeO<sub>3</sub> due to different exchange interactions between transition and rare earth ions. *Phys. Status Solidi B* **260**, 2300026 (2023)
19. Y. Xu, C. Deng, X. Wang, Bandgap modulation and phase boundary region of multiferroic Gd, Co co-doped BiFeO<sub>3</sub> thin film. *AIP Adv.* **13**, 115004 (2023)
20. W. Mao, Q. Yao, Y. Fan, Y. Wang, X. Wang, Y. Pu, X. Li, Combined experimental and theoretical investigation on modulation of multiferroic properties in BiFeO<sub>3</sub> ceramics induced by Dy and transition metals co-doping. *J. Alloys Compd.* **784**, 117–124 (2019)
21. D.V. Karpinsky, A. Pakalniškis, G. Niaura, D.V. Zhaludkevich, A.L. Zhaludkevich, S.I. Latushka, M. Silibin, M. Serdechnova, V.M. Garamus, A. Lukowiak, W. Stręk, M. Kaya, R. Skaudžius, A. Kareiva, Evolution of the crystal structure and magnetic properties of Sm-doped BiFeO<sub>3</sub> ceramics across the phase boundary region. *Ceram. Int.* **47**, 5399 (2021)
22. M.S. Bernardo, Synthesis, microstructure and properties of BiFeO<sub>3</sub>-based multiferroic materials: a review. *Bol. Soc. Esp. Ceram. Vidr.* **53**, 1 (2014)
23. J.A. Boukhari, A. Khalaf, R.S. Hassan, R. Awad, Structural, optical and magnetic properties of pure and rare earth-doped NiO nanoparticles. *Appl. Phys. A* **126**, 323 (2020)

24. S. Supriya, Effect of doping and enhanced microstructures of bismuth titanates as aurivillius perovskites. *Micron* **162**, 103344 (2022)
25. S. Supriya, Research progress, doping strategies and dielectric-ferroelectric anomalies of rare earth-based  $\text{Bi}_{0.5}\text{Na}_{0.5}\text{TiO}_3$  perovskites. *J. Rare Earths* (2024). <https://doi.org/10.1016/j.jre.2023.10.009>
26. S. Supriya, Tailoring layered structure of bismuth-based aurivillius perovskites: recent advances and future aspects. *Coord. Chem. Rev.* **479**, 215010 (2023)
27. K.S.K.R.C. Sekhar, T. Patri, A.M. Tighezza, D.S. Saini, P. Rosaliah, A. Ghosh, Dielectric relaxation and electrical conductivity property correlation in Gd-doped BBTO Aurivillius ceramics. *Appl. Phys. A* **130**, 150 (2024)
28. R.Y. Zheng, C.H. Sim, J. Wang, S. Ramakrishna, Effects of SRO buffer layer on multiferroic  $\text{BiFeO}_3$  thin films. *J. Am. Ceram. Soc.* **91**, 3240–3244 (2008)
29. C.-C. Leu, T.-J. Lin, S.-Y. Chen, C.-T. Hu, Effects of bismuth oxide buffer layer on  $\text{BiFeO}_3$  thin film. *J. Am. Ceram. Soc.* **98**, 724–731 (2014)
30. W. Tang, J. Yang, J. Zhang, Y. Jiang, J. Wang, L. Cao, Y. Fu, Write-once-read-many-times memory device based on  $\text{Pt}/\text{BiFeO}_3/\text{LaNiO}_3$  heterostructures. *Appl. Surf. Sci.* **618**, 156591 (2023)
31. Q. Cao, Y. Zhao, R. Ye, X. Chen, X. Hu, N. Zhuang, Piezoelectric and magneto-optical effect of lanthanum-doped bismuth ferrite films on silicon substrate. *J. Alloys Compd.* **967**, 171840 (2023)
32. Y.J. Acosta-Silva, L.A. Godínez, M. Toledano-Ayala, R. Lozada-Morales, O. Zelaya-Angel, A. Méndez-López, Study of the effects of Er doping on the physical properties of CdSe thin film. *Magnetochemistry* **9**, 107 (2013)
33. Y.J. Acosta-Silva, A. Méndez-López, F. de Moure-Flores, S. Tomás, R. Lozada-Morales, M. Meléndez-Lira, O. Zelaya-Angel, Characterization of substitutional and interstitial  $\text{Eu}^{+3}$ -positions in CdS lattice. *Mater. Chem. Phys.* **257**, 123763 (2021)
34. N.V. Giridharan, S. Supriya, Effect of processing on the properties of  $\text{Bi}_{3.15}\text{Nd}_{0.85}\text{Ti}_3\text{O}_{12}$  thin films. *Thin Solid Films* **516**, 5244–5247 (2008)
35. S. Supriya, A review on lead-free- $\text{Bi}_{0.5}\text{Na}_{0.5}\text{TiO}_3$  based ceramics and films: dielectric, piezoelectric, ferroelectric and energy storage performance. *J. Inorg. Organomet. Polym. Mater.* **32**, 3649–3676 (2022)
36. S.R.A. Ahmed, Structural and electrical behaviors of silicon nitride thin-films deposited using spin coating technique. *Appl. Phys. A* **129**, 504 (2023)
37. P. Suresh, S. Srinath, Effect of synthesis route on the multiferroic properties of  $\text{BiFeO}_3$ : a comparative study between solid state and sol-gel methods. *J. Alloys Compd.* **649**, 843–850 (2015)
38. S. Sharma, V. Singh, R.K. Kotnala, R.K. Dwivedi, Comparative studies of pure  $\text{BiFeO}_3$  prepared by sol-gel versus conventional solid-state-reaction method. *J. Mater. Sci. Mater. Electron.* **25**, 1915–1921 (2014)
39. R. Zheng, X. Gao, J. Wang, Multiferroic  $\text{BiFeO}_3$  thin films buffered by a  $\text{SrRuO}_3$  layer. *J. Am. Ceram. Soc.* **91**, 463–466 (2008)
40. J. Wu, J. Wang,  $\text{BiFeO}_3$  thin films of (111)-orientation deposited on  $\text{SrRuO}_3$  buffered  $\text{Pt}/\text{TiO}_2/\text{SiO}_2/\text{Si}(100)$  substrates. *Acta Mater.* **58**, 1688–1697 (2010)
41. R.D. Shannon, Revised effective ionic radii and systematic studies of interatomic distances in halides and chalcogenides. *Acta Crystallogr.* **32**, 751–767 (1976)
42. C.-C. Leu, T.-J. Lin, S.-Y. Chen, C.-T. Hu, Effects of bismuth oxide buffer layer on  $\text{BiFeO}_3$  thin film. *J. Am. Ceram. Soc.* **98**, 724–731 (2015)
43. S. Chauhan, M. Kumar, S. Chhoker, S.C. Katyal, M. Jewariya, B.N. Suma, G. Kunte, Structural modification and enhanced magnetic properties with two phonon modes in Ca-Co codoped  $\text{BiFeO}_3$  nanoparticles. *Ceram. Int.* **41**, 14306–14314 (2015)
44. F. Lin, Q. Yu, L. Deng, Z. Zhang, X. He, A. Liu, W. Shi, Effect of La/Cr codoping on structural transformation, leakage, dielectric and magnetic properties of  $\text{BiFeO}_3$  ceramics. *J. Mater. Sci.* **52**, 7118–7129 (2017)
45. Y.-H. Si, Y. Xia, S.-K. Shang, X.-B. Xiong, X.-R. Zeng, J. Zhou, Y.-Y. Li, Enhanced visible light driven photocatalytic behavior of  $\text{BiFeO}_3$ /reduced graphene oxide composites. *Nanomaterials* **8**, 526 (2018)

**Publisher's Note** Springer Nature remains neutral with regard to jurisdictional claims in published maps and institutional affiliations.

Springer Nature or its licensor (e.g. a society or other partner) holds exclusive rights to this article under a publishing agreement with the author(s) or other rightsholder(s); author self-archiving of the accepted manuscript version of this article is solely governed by the terms of such publishing agreement and applicable law.