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$(Bi_{0.5}Na_{0.5})_{0.94}Ba_{0.06}TiO_3$ lead-free ceramics with simultaneous addition of CeO₂ and La₂O₃

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ABSTRACT (Bi_{0.5}Na_{0.5})_{0.94}Ba_{0.06}TiO₃ (BNBT-6) with simultaneous addition of 0.5 mol% CeO₂ and 0.25, 0.5 and 0.75 mol% La₂O₃, respectively, was prepared by a conventional ceramic fabrication technique. An addition of 0.5 mol% CeO₂ together with 0.5 mol% La₂O₃ enhanced the piezoelectric and dielectric constant of BNBT-6 ceramics significantly. At room temperature, this composition exhibits a high piezoelectric constant ($d_{33} = 162 \times 10^{-12}$ C/N) and relatively low dielectric loss (tan $\delta = 2.0\%$) at 1 kHz. X-ray diffraction pattern shows that the coexistence of tetragonal and rhombohedral phases in the BNBT-6 composition was not changed by adding CeO₂+La₂O₃, and they diffused into the BNBT lattice during sintering. SEM observation indicates that a small amount of CeO₂+La₂O₃ almost does not affect the microstructure.

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1 Introduction

As we know, $Pb(Zr, Ti)O_3$ (PZT) ceramics play a dominant role in ongoing piezoelectric applications because of their superior piezoelectric properties. However, the high volatilization of PbO, a main composition of PZT ceramics, during firing and the disposal of PbO-contaminated materials cause crucial environmental pollution [1]. As one of the important eco-materials, lead-free piezoelectric ceramics are attracting considerable attention because of their outstanding advantages in free control atmosphere and no lead contamination.

The (Bi_{0.5}Na_{0.5})TiO₃ composition (abbreviated to BNT) is a perovskite ferroelectric with good potential to replace PZT piezoelectric ceramics because of its strong ferroelectricity. It has a high Curie temperature, $T_c = 320$ °C, a relatively large remanent polarization of $P_r = 38 \,\mu\text{C/cm}^2$ and a coercive field of $E_c = 73 \,\text{kV/cm}$ at room temperature [2–4]. A low-temperature phase transition from the ferroelectric to the anti-ferroelectric phase at about 200 °C in the BNT composition leads to a very interesting anomaly in the dielectric properties. However, the high conductivity of pure BNT ceramics causes difficulty in the poling process. Furthermore, its piezoelectric properties are too low for practical use. To improve the poling process and the electrical properties, some modification of the BNT system has been carried out. It has been reported that BNT base compositions modified with BaTiO₃ (abbreviated as BT) [5], NaNbO₃ [6], BiFeO₃ [7], Bi₂O₃Sc₂O₃ [8], La₂O₃ [9] and $Ba(W_{1/2}Cu_{1/2})O_3$ [10], showed improved piezoelectric properties and were easier to pole compared with pure BNT ceramics. Among them, BNT-BT (abbreviated to BNBT) was more interesting because there is a rhombohedral (F_R) tetragonal (F_T) morphotropic phase boundary (abbreviated to MPB) between BNT and BT. It has been reported that the $(Bi_{0.5}Na_{0.5})_{0.94}Ba_{0.06}TiO_3$ (abbreviated to BNBT-6) composition near the MPB has relatively large piezoelectric properties [5]. Some further enhancements in the piezoelectric properties of BNBT ceramics were reported by adding Nb₂O₅ [10] and CeO₂ [11].

In this work, both CeO_2 and La_2O_3 are added to further enhance the piezoelectric properties of BNBT ceramics. The effects of simultaneous doping of $CeO_2+La_2O_3$ on the electrical properties and microstructure of the BNBT-6 ceramics were examined. The dielectric properties of the samples were also measured as a function of temperature.

2 Experimental

The conventional mixed-oxide technique was used to prepare $(Bi_{0.5}Na_{0.5})_{0.94}Ba_{0.06}TiO_3$ (BNBT-6) ceramics. The starting materials were Bi_2O_3 (99.9%, Acros Organics), Na_2CO_3 (99.5%, Wako Pure Chemical Industries Ltd.), $BaCO_3$ (99.9%, Wako Pure Chemical Industries Ltd.) and TiO_2 (99.9%, Aldrich Chem. Co.). The powders were milled in alcohol using Y_2O_3 -stabilized zirconia balls for 6 h. Calcination was conducted at 800 °C for 2 h. Then, 0.5 mol% CeO_2 (99.9%, Aldrich Chem. Co.) together with 0.25, 0.5 and 0.75 mol% La₂O₃ (99.9%, Aldrich Chem. Co.) were added to the calcinated BNBT-6 base powders, respectively. The powders were milled again for 6 h, dried and granulated by adding PVA (Wako Pure Chemical Industries Ltd.) as a binder. The

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granulated powders were pressed into discs with diameter of 22 mm and thickness of 2 mm. The compacted discs were sintered at 1180 °C for 2 h in air. Silver paste (Guangzhou Luxing Tech. Ltd.) was fired on both faces of the discs at 650 °C as electrodes. The samples for the measurement of the piezo-electric properties were poled in silicone oil at 50 °C under 4.5 kV/mm for 10 min.

The density of the sintered samples was measured using the Archimedes method. After poling, a d_{33} meter (ZJ-30 Piezo d_{33} Meter) was used to measure the d_{33} coefficient at 100 Hz. The dielectric properties of the samples were determined at room temperature using an impedance analyzer (Agilent 4294A precision impedance analyzer) at 1 kHz. The temperature dependence of the dielectric constant and the dissipation factor of the unpoled samples were measured at a step of 5 °C using a Hewlett–Packard 4192A impedance analyzer in a temperature-programmable tube furnace at different frequencies. The XRD pattern was obtained, to identify the crystal phase of the samples, using an X-ray diffractometer (XRD–Philips Expert System). A scanning electron microscope (SEM-Leica Stereoscan 440) was used to observe the microstructure of the sintered samples.

3 Results and discussion

Table 1 shows the electrical properties of BNBT-6 ceramics and BNBT-6 ceramics with CeO₂+La₂O₃. The results indicate that simultaneous addition of an optimal amount of CeO₂ and La₂O₃ into BNBT-6 enhances the piezoelectric and dielectric constants significantly at room temperature. With 0.5 mol% CeO₂ and 0.5 mol% La₂O₃, the piezoelectric constant d_{33} increases from 129 to 162×10^{-12} C/N. The dielectric constant also increases from 625 to 831. The dissipation factor slightly varies with the relative amount of CeO₂ and La₂O₃ within the tested range. It has been reported that CeO₂ has the dual effect of improving the piezoelectric properties of BNBT-6 ceramics and decreasing the dissipation factor at the same time, as it does in lead-based Pb(Zr, Ti)O₃ ceramics [12]. The mechanism for the effect of CeO_2 is considered to be concerned with its two valence states: Ce⁴⁺, with a radius of 0.94 Å, and Ce^{3+} , with a radius of 1.18 Å. Ce^{3+} occupies the A site as Ba^{2+} in the BNBT lattice and functions as a donor, leading to some A-site vacancies, which facilitates the movement of the domain wall so as to improve the piezoelectric properties. Meanwhile, Ce⁴⁺ may go into the Ti⁴⁺ site and change the space charge, which suppresses the domain movement, resulting in a decrease in the dissipation factor. La_2O_3 is a typical soft additive for PZT ceramics [12] and is also effective for enhancing the piezoelectric properties of BNT ceramics [9]. La₂O₃ goes into A sites in the BNBT lattice by occupying a Bi^{3+} site or a Ba^{2+} site. The former results in deformation of the lattice and the latter leads to some vacancies in A-sites of the lattice, which makes the movement of the domain walls easier, so as to increase the piezoelectric and dielectric constants. The P-E loop measurement, as shown in Fig. 1, reveals an increase in the remanent polarization $P_{\rm r}$ and a decrease in the coercive field E_c of BNBT-6 ceramics by the addition of 0.5 mol% CeO₂ and 0.5 mol% La₂O₃ simultaneously, which implies the enhancement of the ferroelectric and piezoelectric effects and the facilitation of domain movement, leading to an increase in piezoelectric properties. However, a suitable addition amount is important. Addition of 0.75 mol% La₂O₃ may cause too many defects to result in a decrease in the piezoelectric properties. We have reported recently that BNBT-6 ceramics with CeO₂ have a piezoelectric constant d_{33} of 152×10^{-12} C/N [11]. It is thus seen that simultaneous addition of CeO2+La2O3 has a complex effect of modification.

Figures 2 and 3 show the temperature dependence of dielectric constant ε_r and dissipation factor tan δ of the unpoled BNBT-6 ceramics and BNBT-6 ceramics containing 0.5 mol% CeO₂ and 0.5 mol% La₂O₃ at different frequencies, respectively. The temperature-dependent dielectric constant shows relaxor-like behavior, probably due to cation disorder, as Na⁺ and Bi³⁺ are randomly distributed in A-sites of the lattice [13]. Comparing with the results reported by Takenaka [5], both samples are characterized by the existence of three phases of ferroelectric, anti-ferroelectric and paraelectric states in different temperature ranges. Here, T_d refers to the temperature of the transition between the ferroelectric

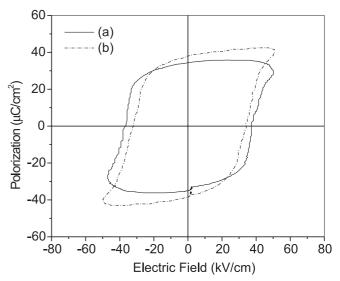


FIGURE 1 P-E hysteresis loops of (a) BNBT-6 ceramics and (b) BNBT-6 ceramics containing 0.5 mol% CeO₂ and 0.5 mol% La₂O₃ at room temperature

Composition	d_{33} (10 ⁻¹² C/N)	<i>ɛ</i> r	tan δ (%)	ρ (g/cm ³)
BNBT-6	129	625	1.9	5.69
BNBT-6+0.5 mol% CeO ₂ +0.25 mol% La ₂ O ₃	158	787	1.7	5.67
BNBT-6+0.5 mol% CeO ₂ +0.5 mol% La ₂ O ₃	162	831	2.0	5.68
BNBT-6+0.5 mol% CeO ₂ +0.75 mol% La ₂ O ₃	65	310	2.4	5.69

TABLE 1 The electric properties and density of BNBT-6 with simultaneous addition of $CeO_2+La_2O_3$

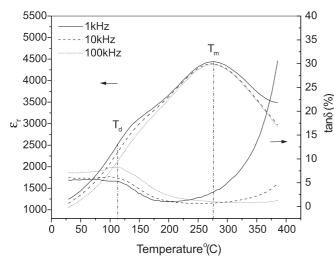


FIGURE 2 Temperature dependence of dielectric constant and dissipation factor of BNBT-6 ceramics at different frequencies

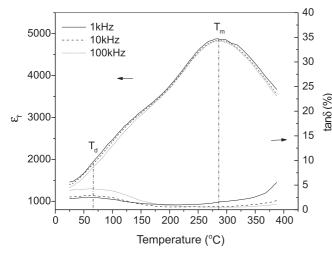


FIGURE 3 Temperature dependence of dielectric constant and dissipation factor of BNBT-6 ceramics containing 0.5 mol% CeO₂ and 0.5 mol% La₂O₃ at different frequencies

phase and anti-ferroelectric phase and $T_{\rm m}$ stands for the temperature corresponding to the maximum value of dielectric constant. An obvious decrease in $T_{\rm d}$ and a slight increase in $T_{\rm m}$ are observed by adding CeO₂+La₂O₃. Furthermore, the dielectric constant and dissipation factor at $T_{\rm m}$ increase with the addition of CeO₂+La₂O₃. At 1 kHz and at a temperature

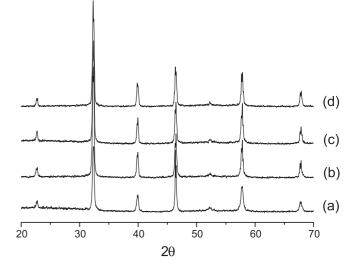


FIGURE 4 X-ray diffraction patterns of BNBT-6 ceramics with: $0 \mod \%$ CeO₂+La₂O₃ (a); 0.5 mol% CeO₂ and 0.25 mol% La₂O₃ (b); 0.5 mol% CeO₂ and 0.5 mol% La₂O₃ (c); and 0.5 mol% CeO₂ and 0.75 mol% La₂O₃ (d)

higher than T_m , the dissipation factor of BNBT-6 compositions increases abruptly with increasing temperature (Fig. 2), which may be attributed to high conduction. It is evident that the addition of CeO₂+La₂O₃ can decrease the conductivity of BNBT-6 ceramics at high temperature to reduce the dissipation factor (Fig. 3). Also, it should be noted that the T_d and T_m of both samples are almost independent of the measurement frequencies. Below T_d and above T_m , the dielectric constant of BNBT-6 ceramics varies with the measuring frequencies. However, the addition of CeO₂+La₂O₃ results in the dielectric constant of BNBT-6 ceramics being almost independent of the measurement frequencies over the whole temperature range.

XRD analysis was performed to examine the influence of $0.5 \text{ mol}\% \text{ CeO}_2 + 0.5 \text{ mol}\% \text{ La}_2\text{O}_3$ addition on the structure of BNBT-6 ceramics. All samples showed a similar Xray diffraction pattern (Fig. 4) and no second phase was observed. It is known that the BNT system is rhombohedral and that BT is tetragonal at room temperature. A rhombohedral– tetragonal morphotropic phase boundary (MPB) exists in their solid solution near the BNBT-6 composition. For the composition near to the MPB, the X-ray diffraction pattern is characterized by the presence of two separated peaks to (003)

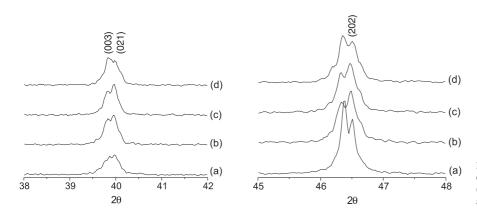


FIGURE 5 Enlarged X-ray diffraction patterns of BNBT-6 compositions corresponding to (a), (b), (c), (d) shown in Fig. 4 at 2θ of 38° to 42° and 45° to 48°

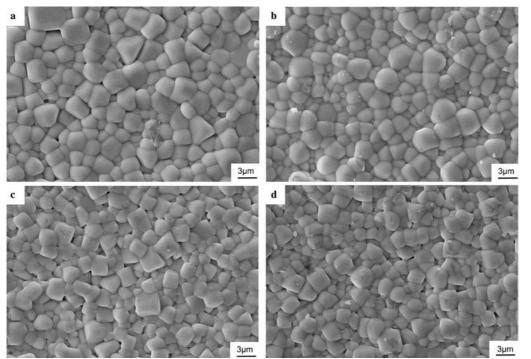


FIGURE 6 SEM micrographs of BNBT-6 ceramics sintered at 1180 °C for 2 h with: 0 mol% CeO₂+La₂O₃ (a); 0.5 mol% CeO₂ and 0.25 mol% La₂O₃ (b); 0.5 mol% CeO₂ and 0.5 mol% La₂O₃ (c); and 0.5 mol% CeO₂ and 0.75 mol% La₂O₃ (d)

and (021) at about 39.8°, and the splitting of the peak to (202) planes at around 46.5° [10]. As the BNBT-6 composition used in this test was at the MPB, its X-ray pattern exhibited the feature of peak splitting at these diffraction angles, as shown in Fig. 5. This result indicates that tetragonal and rhombohedral phases coexist in both BNBT-6 and $CeO_2+La_2O_3$ -added BNBT-6 ceramics. Therefore, it can be concluded that the addition of CeO_2 and La_2O_3 does not cause an obvious change to the phase structure.

The microstructures of the sintered BNBT-6 ceramics containing 0.5 mol% CeO₂ and 0.25, 0.5 and 0.75 mol% La₂O₃ were observed using SEM, and micrographs of the samples are shown in Fig. 6. All these samples are very dense. No obvious change in grain size is caused by adding CeO₂+La₂O₃ in the tested range. That is to say, the modification of CeO₂+La₂O₃, other than its effect on microstructure, plays a major part in affecting the piezoelectric properties of the BNBT-6 ceramics.

4 Conclusion

 $(Bi_{0.5}Na_{0.5})_{0.94}Ba_{0.06}TiO_3$ (BNBT-6) ceramics with simultaneous addition of CeO₂ and La₂O₃ have been investigated. It was found that at 0.5 mol% CeO₂+0.5 mol% La₂O₃, both the piezoelectric and dielectric constants of BNBT-6 ceramics were increased significantly, but the dissipation factor was unaffected. The addition of CeO₂+La₂O₃ did not cause an obvious change in microstructure. BNBT-6 ceramics containing $CeO_2+La_2O_3$ still showed the coexistence of tetragonal and rhombohedral phases similar to the BNBT-6 ceramics and no second phase was observed.

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