REVIEW

Cell biology of molybdenum in plants

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Abstract The transition element molybdenum (Mo) is of essential importance for (nearly) all biological systems as it is required by enzymes catalyzing important reactions within the cell. The metal itself is biologically inactive unless it is complexed by a special cofactor. With the exception of bacterial nitrogenase, where Mo is a constituent of the FeMo-cofactor, Mo is bound to a pterin, thus forming the molybdenum cofactor (Moco) which is the active compound at the catalytic site of all other Mo-enzymes. In plants, the most prominent Mo-enzymes are nitrate reductase, sulfite oxidase, xanthine dehydrogenase, aldehyde oxidase, and the mitochondrial amidoxime reductase. The biosynthesis of Moco involves the complex interaction of six proteins and is a process of four steps, which also includes iron as well as copper in an indispensable way. After its synthesis, Moco is distributed to the apoproteins of Mo-enzymes by Moco-carrier/binding proteins that also participate in Moco-insertion into the cognate apoproteins. Xanthine dehydrogenase and aldehyde oxidase, but not the other Mo-enzymes, require a final step of posttranslational activation of their catalytic Mo-center for becoming active.

Keywords Molybdenum cofactor · Sulfite oxidase · Nitrate reductase · Xanthine dehydrogenase · Aldehyde oxidase · Molybdenum cofactor deficiency

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Introduction

The transition element molybdenum (Mo) occurs in a wide range of metalloenzymes in bacteria, fungi, algae, plants and animals where it forms part of the active centers of these enzymes (for reviews see Schwarz and Mendel 2006; Bittner and Mendel 2010). In order to gain biological activity, Mo has to be complexed by a pterin compound thereby forming the prosthetic group named molybdenum cofactor (Moco) (for reviews see Schwarz et al. 2009; Mendel and Schwarz 2011). It has been long known that Mo (Bortels 1930) is an essential nutrient for plants, animals and microorganisms. Mo is very abundant in the oceans in the form of the MoO_4^{2-} anion. Furthermore, in soils too, the molybdate anion is the only form of Mo that is available for plants and bacteria. Biologically, Mo belongs to the group of trace elements, i.e. the organism needs it only in minute amounts. If, however, an organism takes up too high amounts of Mo, toxicity symptoms are observed (Turnlund 2002). On the other side, unavailability of Mo, is lethal for the organism. More than 50 Mo-containing enzymes are known, most of them of bacterial origin, while only a handful Mo-enzymes were found among eukaryotes. In plants, five Mo-enzymes are known to this end: nitrate reductase (NR), sulfite oxidase (SO), xanthine dehydrogenase (XDH), aldehyde oxidase (AO) and mitochondrial amidoxime reductase (mARC). In addition to the pterintype of cofactor, there is another type of Mo-containing cofactor, which is found only once in nature, namely in bacterial nitrogenase, forming the so-called iron-molybdenum cofactor, FeMoco. Nitrogenase is required for biological nitrogen fixation, which is an essential step in the nitrogen cycle in the biosphere. In contrast to nitrogenase, all other Mo-containing enzymes characterized to this end contain the pterin-type cofactor (Table 1).

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Mo-enzyme	Number of genes	Subcellular location	Function	ROS/RNS side product
Nitrate reductase	2	Cytosol	Nitrate assimilation	NO
Sulfite oxidase	1	Peroxisome	Sulfite detoxification	H_2O_2
Xanthine dehydrogenase	2	Cytosol (peroxisome)	Purine degradation, NADH oxidase	Superoxide anions
Aldehyde oxidase	4	Cytosol	Synthesis of ABA (auxins)	H_2O_2 , superoxide anions
Mitochondrial amidoxime reductase	2	Mitochondria	Detoxification (?)	n.d.

Table 1 Mo-enzymes in plants (Arabidopsis thaliana)

This article will first present those plant enzymes that need Mo as catalytic metal, followed by a more cell biological part that will describe the way that Mo takes from uptake into the cell, via formation of the Moco and its storage, to its final insertion into apo-metalloenzymes. Most of our knowledge about eukaryotic Mo metabolism derives from studies in plants. This may be surprising, but in pregenomic times the detailed characterization of NR-deficient mutants (using callus cultures and in vitro-plants) contributed substantially to our understanding of the genetics and biochemistry of Moco biosynthesis in plants (summarized by Müller and Mendel 1989). These NR-deficient mutants fell into two groups: those with a defect in the NR-apoprotein and others which were defective in Moco. The identification of several genetic complementation groups among Moco-deficient plant mutants (using crossing methods but also protoplast fusions) and the conserved structure of Moco provided a basis to propose an evolutionary old multi-step pathway for the biosynthesis of Moco as prosthetic group needed by all Mo-enzymes (Mendel 1992). The molecular, biochemical and genetic analysis of Moco mutants in higher plants formed the basis to decipher Moco biosynthesis also in humans where Moco deficiency is a severe genetic disease with fatal consequences for the affected individuals (Reiss and Johnson 2003). All higher plants possess Mo-enzymes, however, genome-wide database analyses revealed a significant number of bacteria and unicellular lower eukaryotes that do not need Mo. Loss of Mo-utilization is obviously connected to a host-associated life-style that makes Mo-enzymes unnecessary, while all multi-cellular eukaryotes are dependent on Mo (Zhang and Gladyshev 2008). Therefore model yeasts like Saccharomyces cereviseae and Schizosaccharomyces pombe play no role in Mo research as they belong to those organisms that do not contain Mo-enzymes while other yeasts such as Pichia pastoris do need Moco.

What enzymes need Mo?

In general, all Mo-enzymes catalyze the transfer of an oxygen atom to or from a substrate (Hille 2002). Each

reaction, either reduction or oxidation, is also characterized by transfer of two electrons, which enforce the Mo atom to vary its oxidation state between IV and VI. Mo-enzymes present in eukaryotes are homodimeric proteins functioning only as a dimer, but not as a monomer. Some harbour an electron transport chain involving different prosthetic groups (FAD, heme or Fe-S, Moco) that are bound to separate domains identified on the enzyme's monomer. Fig. 1 shows the domain structure of the five plant Mo-enzymes where the domains are connected by regions highly variable in sequence thus obviously serving as interdomain hinge regions. It is obvious that these Mo-enzymes can be subdivided into two classes: the SO-class of Mo-enzymes is formed by SO and NR, and the XO oxidase (XO) family is represented by XDH, and AO, both classes sharing domains. We will later see that mARC forms a separate group.

Sulfite oxidase

SO (EC 1.8.3.1), the name-giving enzyme for the members of the SO-family of Mo-enzymes, catalyzes the oxidation from sulfite to sulfate (Eilers et al. 2001). The SO gene is widespread and highly conserved within the plant kingdom. SO possesses only Moco as redox center (Fig. 1) and-together with mARC-is the simplest Mo-enzyme found in eukaryotes. Oxygen serves as the terminal electron acceptor for plant SO and is reduced to hydrogen peroxide (Hänsch et al. 2006). The latter finding explains the peroxisomal localization of plant SO (Nowak et al. 2004) since excess hydrogen peroxide generated during sulfite oxidation can easily be eliminated by catalase. The physiological role of plant SO has been clarified only recently. As sulfite is a strong nucleophile that can react with a wide variety of cellular components, it was assumed that SO has a sulfite-detoxifying function and is required for removing excess sulfite from the cell. In support of this, Brychkova et al. (2007) and Lang et al. (2007) independently found that in comparison to wild-type plants SOdeficient plants are more susceptible to high concentrations of sulfite while SO-overexpressing plants are more tolerant



Fig. 1 Organization of biosynthesis, distribution, and maturation of Moco in plants. The basic steps of Moco biosynthesis are shown starting from the conversion of GTP to cPMP in the mitochondria, all subsequent steps proceed in the cytosol. Moco biosynthesis enzymes (named Cnx) and Moco-binding proteins (MoBP), are shown in *blue*. MPT-synthase, consisting of Cnx6 and Cnx7, is sulfurated by Cnx5, with the primary sulfur donor (X-S) mobilized by the rhodanese-like domain of Cnx5 (RLD) being unknown. The adenylation domain of Cnx5 (AD) is required for adenylation and activation of the small MPT synthase subunit Cnx7. It is assumed that copper (Cu) is inserted directly after dithiolene formation. The individual reactions of Cnx1 and its products (Moco, pyrophosphate PP_i, AMP, copper) are

indicated. Moco can be either bound to a Moco-binding protein (MoBP), to the five Mo-enzymes NR, SO, XDH, AO and mARC, or to the Moco-binding C-terminal domain (MBD) of the Mocosulfurase ABA3. The ABA3 C-terminus is the sulfuration platform for Moco. The NifS-like domain of ABA3 generates a protein-bound persulfide, which is transferred to Moco bound to its C-terminus which in turn exchanges non-sulfurated for sulfurated Moco. The domain structure of the Mo-enzyme monomers is given in the figure. It is obvious that the eukaryotic Mo-enzymes are evolutionary related to each other: XDH and AO form a pair, SO forms a pair with NR, and mARC arose from ABA3-MBD (modified after Bittner and Mendel 2010) to excess sulfite. Under normal conditions however, the lack of SO activity in plants is not related to an obvious phenotype, suggesting that SO represents a salvage enzyme rather than a metabolic housekeeping enzyme (Hänsch et al. 2007).

Nitrate reductase

NR (EC 1.7.1.1., formerly EC 1.6.6.1) is a another member of the SO family and a key enzyme of nitrate assimilation, where it catalyzes the reduction of nitrate to nitrite in the cytosol. As NR thereby provides essential nitrogen metabolites to the plant, it is obvious that plants with a deficiency in NR are no longer nitrogen autotroph and depend on alternative nitrogen sources such as ammonium. The monomer of NR consists of three distinct domains (Fig. 1): the N-terminal domain associated with Moco, the central heme-binding cytochrome b_5 domain, and the C-terminal FAD-binding domain, whereby two such monomers form the active homodimeric enzyme. The domains are separated by solvent-exposed and proteasesensitive linker regions, called hinge I and hinge II. In plants hinge I, the linker between the cytochrome b_5 domain and the Moco domain, contains a conserved serine residue which mediates contact with a 14-3-3 protein when phosphorylated, subsequently leading to inhibition of enzyme activity (Kaiser and Huber 2001). The three domains form three redox centers catalyzing the transfer of electrons from the reductant NAD(P)H via FAD, heme, and Moco to nitrate.

In contrast to the reaction catalyzed by SO the process of nitrate reduction consumes rather than produces electrons deriving either from NADH or from NADPH. The crystal structures of SO (Schrader et al. 2003) and of the Moco domain of NR (Fischer et al. 2005) revealed a high degree of structural conservation between both proteins with only few differences in the boundaries of the enzymes. Substrate binding is proposed to be very similar among SO and NR due to three overall conserved residues in the actives sites. Yet, there are also important changes within the active site that are believed to determine substrate specificity and catalytic properties. These structures revealed that the Mo-domain can be further subdivided into a Moco-binding catalytic domain and a C-terminal domain representing the dimerization interface of holo-NR. It is remarkable that the Moco is completely buried inside the protein and a funnel-like structure leads from the surface of the protein to the active site.

While in terms of nitrogen assimilation nitrite is further reduced in the chloroplasts to ammonium by nitrite reductase, it can also be reduced to the signalling molecule nitric oxide (NO) by NR itself (Yamasaki and Sakihama 2000) As post-translational modification of NR also modulated the NO-production rates, it was concluded that NR is indeed a producer of active nitrogen species also in vivo (Wang et al. 2011).

Xanthine dehydrogenase

The members of the XO family are molybdo-flavoenzymes that catalyze the oxidative hydroxylation of a wide range of aldehydes and aromatic heterocycles. Plant XDH (EC 1.17.1.4., formerly EC 1.1.1.204) is a key enzyme of purine degradation where it oxidizes hypoxanthine to xanthine and xanthine to uric acid. The monomer can be subdivided into three distinct domains (Fig. 1): an N-terminal domain for binding of two iron-sulfur clusters of the [2Fe-2S]-type, a domain harbouring a FAD-binding site, and a C-terminal domain required for Moco-binding and dimerization. Electrons derived from substrate hydroxylation at the Mocenter go via the Fe-S clusters to the flavin cofactor. At the FAD site, the electrons are transferred either to NAD⁺ to form NADH, or they are transferred to molecular oxygen to yield superoxide anions.

In addition to the hypoxanthine/xanthine-dependent production of reactive oxygen species (ROS) also NADH oxidase activity with simultaneous production of superoxide has been shown for plant XDH (Zarepour et al. 2010). Since this activity is very pronounced, this implies that XDH is an efficient producer of superoxide also in vivo, and that the enzyme may be involved in the regulation of the cellular NADH/NAD⁺ balance. The ROS production of plant XDH might also be of physiological importance because increasing XDH activities and simultaneous ROS production were observed upon plant-pathogen interactions (Montalbini 1992a, b), hypersensitive response (Montalbini and Della Torre 1996), drought stress (Yesbergenova et al. 2005), virus infection (Silvestri et al. 2008) and natural senescence (Pastori and Rio 1997; Hesberg, et al. 2004). The subcellular localization of XDH is still not absolutely clear. Both, a cytosolic (Datta et al. 1991) and peroxisomal localization (Sandalio et al. 1988) was reported, a recent article localizes XDH to both the cytosol and peroxisomes (Corpas et al. 2008).

Aldehyde oxidase

AO proteins are cytoplasmic enzymes (EC 1.2.3.1) that catalyze the oxidation of a variety of aromatic and nonaromatic heterocycles and aldehydes, thereby converting them to the respective carboxylic acid. AO enzymes are very similar to XDH (Fig. 1) as they share a high degree of sequence homology, so that it is presumed that during evolution AO has derived from XDH by gene duplication and neo-functionalization (Terao et al. 2001; Rodriguez-Trelles et al. 2003). The most prominent characteristics which distinguish AO from XDH enzymes have been found to concern the substrate binding at the Mo-center and binding of the physiological electron acceptor (Hille 2005). AO enzymes are strict oxidases that are unable to bind NAD⁺ and exclusively use molecular oxygen as electron acceptor. Upon transfer of substrate-derived electrons to molecular oxygen, plant AO generates hydrogen peroxide (Yesbergenova et al. 2005).

The Arabidopsis genome harbors four AO genes, AAO1-AAO4, whose products form homodimers as well as heterodimers, thereby leading to altered substrate specificities of the respective isoenzymes. In 6-day-old seedlings, the gene products of AAO1 and AAO2 form three homo- and heterodimeric AO isoenzymes capable of producing indole-3-acetic acid (Akaba et al. 1999). The fact that IAA belongs to the auxin family of plant hormones suggests a possible physiological role of these AO enzymes in auxin biosynthesis during early stages of plant development. The AAO3 homodimer is characterized by a high preference for abscisic aldehyde as substrate (Seo et al. 2000a, b), which is the ultimate precursor of abscisic acid (ABA) that is involved in many aspects of plant growth and development, including seed maturation, dormancy, leaf senescence as well as adaptation to a variety of environmental stresses (Seo and Koshiba 2002; Verslues and Zhu 2005). Arabidopsis mutants with a deficiency in AAO3 therefore are characterized by reduced ABA levels accompanied by excessive water loss and a wilty phenotype, and also by retarded vegetative growth and reduced stress tolerance (Seo et al. 2000a, b). Recently, AAO3 protein levels have been shown to be regulated by ubiquitin-dependent degradation via the 26S-proteasome to prevent premature senescence by accumulation of ABA (Raab et al. 2009). This suggests that ABA-synthesizing AO proteins play a critical role also during the onset of senescence, which requires a tight control of AO and ABA levels.

Mitochondrial amidoxime reductase

In mammals, the mitochondrial amidoxime reducing component mARC was shown to catalyze the reduction of a variety of *N*-hydroxylated compounds, some of which are widely used as prodrugs and thus, are of pharmaceutical relevance (Havemeyer et al. 2006); Gruenewald, et al. 2008). All eukaryotic genomes known to encode proteins for Moco biosynthesis and Mo-enzymes, likewise encode two mARC proteins, suggesting that mARC proteins form an own small protein family (Wahl et al. 2010). All eukaryotic mARC proteins including the plant counterparts, are characterized by the presence of N-terminal extensions which predict a mitochondrial localization of these proteins. With an average molecular mass of ~ 35 kDa and due to the fact that they bind Moco as the only prosthetic group, mARC protein are the smallest Mo-enzymes identified as yet. It is assumed that they play a detoxifying role in metabolism. In contrast to all other Mo-enzymes, eukaryotic mARC proteins do not exhibit enzymatic activity on their own but require other proteins like cytochrome b_5 and NADH/cytochrome b_5 reductase as electron transmitters and electron donors, respectively.

Molybdenum uptake into cells

How do Mo-enzymes acquire their catalytically important metal? Organisms take up Mo in the form of its molybdate anion. It requires specific uptake systems to scavenge molybdate in the presence of competing anions. In bacteria, high-affinity ABC-type transporters are described and in some bacteria specific molybdate-binding proteins are known with a capacity of up to eight anions (Pau and Lawson 2002) that store molybdate until further use by the cell. In contrast to bacterial molybdate homeostasis, eukaryotic molybdate transport is less well understood. Algae and higher plants are the only eukaryotes of which the molybdate uptake mechanisms have been unveiled recently. The protein MOT1 belonging to the large sulfate carrier family was shown to transport molybdate with ultrahigh affinity (nanomolar k_M value) across cellular membranes (Baxter et al. 2008; Tejada-Jimenez et al. 2007; Tomatsu et al. 2007). Surprisingly, it was not found to reside in the plasma membrane. Contradictory reports localized it to the endomembrane system (Tomatsu et al. 2007) or to the mitochondrial envelope (Baxter et al. 2008). Both suggested subcellular locations are questionable as the insertion of Mo into the Moco-backbone takes place in the cytosol. A recent paper described the vacuole as an important molybdate store and provided functional evidence for a second molybdate transporter (MOT2) which was localized to the tonoplast (Gasber et al. 2011). This result is supported by a paper analyzing the proteome of the tonoplast where MOT2 was reported to be a tonoplast protein (Carter et al. 2004). Still the cellular importer for Mo is missing, but it is likely that additional transporters, not only in autotrophs but also in animals, will carry out this task. Here the novel transporter CrMOT2 found both in Chlamydomonas reinhardtii and humans (Tejada-Jimenez et al. 2011) is a good candidate to play this role. Further, it can be assumed that in addition to a possible high-affinity system molybdate could also non-specifically enter the cell through the sulfate uptake system which was recently shown for a sulfate transporter (Fitzpatrick et al. 2008).

The Molybdenum cofactor

Once having entered the cell, Mo has to be attached to its cofactor scaffold thereby converting to Moco. Early work with mutants of Nicotiana tabacum (Mendel and Müller 1976) and the filamentous fungus Aspergillus nidulans (Cove and Pateman 1963) had postulated the existence of Moco. These genetic experiments revealed a novel mutant phenotype, namely the simultaneous loss of the two Mo-enzymes NR and XDH. Since Mo was the only common link between these two-otherwise very different-enzymes, it was suggested that both enzymes should share a common Mo-related cofactor, named Moco. Later, the Rajagopalan-group identified this cofactor as a tricyclic pteridine that coordinates the metal via a dithiolene group (Fig. 1, right upper part). Because of the unique nature of the pterin in Moco, the metal-free form of the cofactor is called molybdopterin or metal-containing pterin (MPT). The pterin structure of Moco is unique in nature and has probably been evolved in order to control and maintain the special redox properties of Mo. The task of the cofactor is to position the catalytic metal Mo correctly within the active center, to control its redox behaviour and to participate with its pterin ring system in the electron transfer to or from the Mo atom. The pterin with its several possible reduction states as well as different structural conformations could also be important for channeling electrons to other prosthetic groups (Kisker et al. 1997a, b). X-ray crystallographic analyses of Mo-enzymes revealed that the cofactor is not located on the surface of the protein, but it is buried deeply within the interior of the enzyme and a tunnel-like structure makes it accessible to the appropriate substrates (Kisker et al. 1997a, b; Fischer et al. 2005). During its life time, the Mo-enzyme does not liberate Moco. In vitro, however, one can remove Moco from the holoenzyme whereafter Moco loses Mo and undergoes rapid and irreversible loss of function due to oxidation (Rajagopalan 1996). The demolybdo-forms of Moenzymes are catalytically inactive. In bacteria, the Moco can also occur with a nucleotide attached via its phosphate group or in the form of two pterins coordinating one Mo (Schwarz et al. 2009).

Molybdenum cofactor biosynthesis

A mutational block of Moco biosynthesis leads to the loss of essential metabolic functions because all enzymes depending on Mo lose their activity, which ultimately causes death of the organism. In all organisms studied so far, Moco is synthesized by a conserved biosynthetic pathway that can be divided into four steps, according to the biosynthetic intermediates cPMP, MPT, adenlyated MPT, and Moco (Fig. 1). In eukaryotes always six gene products catalyzing Moco biosynthesis have been identified in plants (Mendel and Schwarz 2002), fungi (Millar et al. 2001) and humans (Reiss et al. 1998; Stallmeyer et al. 1999b). These genes are homologous to their counterparts in bacteria, and some but not all of the eukaryotic Moco biosynthesis genes are able to functionally complement the matching bacterial mutants. Genes and gene products were named in plants according to the cnx nomenclature (cofactor for nitrate reductase and xanthine dehydrogenase).

Biosynthesis in four steps

Step1 Moco biosynthesis starts in the mitochondria. GTP is transformed into the sulfur-free pterin compound cyclic pyranopterin monophosphate (cPMP), also known as precursor Z (Fig. 1), which is catalyzed by the proteins Cnx2 and Cnx3 in the mitochondrial matrix (Teschner et al. 2010). For its functioning, Cnx2 needs Fe-S clusters which are readily available in the mitochondrion. As all subsequent steps of Moco biosynthesis have been demonstrated to be localized in the cytosol, export of cPMP from the mitochondria into the cytosol is required to allow further processing to Moco. This task is fulfilled by the transporter protein Atm3, which is localized in the inner membrane of mitochondria and belongs to the family of ATP-binding cassette (ABC) transporters (Teschner et al. 2010). Surprisingly, Atm3 functions also in exporting a precursor compound essential for Fe-S cluster synthesis in the cytosol (Bernard et al. 2009).

Step 2 In the second stage sulfur is transferred to cPMP in order to generate the intermediate MPT (Fig. 1). This reaction is catalyzed by the enzyme MPT synthase, a heterotetrameric complex of two small (Cnx7) and two large (Cnx6) subunits that stoichiometrically converts cPMP into MPT. After MPT synthase has transferred the two sulfurs to cPMP, it has to be re-sulfurated by MPT-synthase sulfurase Cnx5 (Fig. 1) in order to reactivate the enzyme for the next reaction cycle of cPMP conversion. Cnx5 is a twodomain-protein that activates Cnx7 by adenylation (carried out by the Cnx5 N-terminal domain) followed by a sulfur transfer reaction (carried out by the Cnx5 C-terminal domain which has a rhodanese function). Cysteine is a likely candidate as final S-donor for the formation of MPT.

Step 3 After synthesis of the MPT moiety, the chemical backbone is built for binding and coordination of the Mo

atom. In the third step, therefore, Mo has to be transferred to MPT in order to form Moco, thus linking the molybdate uptake system to the MPT pathway. Mutants defective in this step produce MPT and can be partially repaired by growing them on high-molybdate medium (Mendel et al. 1981). In bacteria, this step is catalyzed by two proteins while during evolution to higher organisms these two proteins were fused to the two-domain protein Cnx1. The C-terminal Cnx1 domain (=Cnx1-G) adenylates MPT and forms the intermediate MPT-AMP that remains bound to Cnx1-G (Fig. 1). Most remarkably, the crystal structure of the Cnx1-G revealed a copper bound to the MPT dithiolate sulfurs (Kuper et al. 2004). Up to now the function of copper is unknown but it might play a role in sulfur transfer to cPMP, in protecting the MPT dithiolate from oxidation, and/or presenting a suitable leaving group for Mo insertion. The origin of copper is also unclear but it is reasonable to assume that it is transferred in vivo via the action of cytoplasmic chaperones thereafter binding to the dithiolate group of MPT just after the latter has been formed, i.e. at the end of step 2 of Moco biosynthesis. The copper is released after Mo has been inserted.

Step 4 In the final step of Moco biosynthesis (Fig. 1) MPT-AMP has to be converted into mature Moco: MPT-AMP is transferred to the N-terminal domain of Cnx1 (=Cnx1-E) thereby building a product-substrate channel. Cnx1-E cleaves the adenylate, releases copper and inserts Mo, thus yielding active Moco (Llamas et al. 2005). In vitro-studies with Cnx1-G-bound MPT-AMP revealed an inhibition of Moco synthesis in the presence of 1 μ M CuCl₂, providing a link between Mo and copper metabolism (Llamas et al. 2006). Remarkably, the human homolog of Cnx1 is the protein Gephyrin (Stallmeyer et al. 1999b) which fulfills a dual role: it functions not only as Mo insertase but it serves also as anchor protein for inhibitory neuroreceptors in the postsynaptic membrane of neurons (Schwarz et al. 2009).

Distribution of molybdenum cofactor: storage and transfer

In higher organisms, the steps beyond Moco biosynthesis are less well known and, therefore, research focuses on Moco transport, allocation and insertion into apo-Moenzymes (Mendel and Schwarz 2011). Here one has to keep the cellular context in mind as Moco biosynthesis occurs to be micro-compartmentalized in a multiproteinbiosynthesis complex localized in the cytosol of the cell. After completion of biosynthesis, Moco either has to be allocated and inserted into the apo-Mo-enzymes, or Moco has to become bound to a carrier protein that protects and stores Moco until further use thus providing a way to buffer supply and demand of Moco. In bacteria, a complex of proteins synthesizing the last steps of Moco biosynthesis donates the mature cofactor to apo-enzymes assisted by enzyme-specific chaperones (Leimkuhler et al. 2011). Nearly each bacterial Mo-enzyme has a private chaperone available (Magalon et al. 2011); however, in higher organisms, no Mo-enzyme-specific chaperones have been found yet. Moco is highly unstable once liberated from proteins; it loses the Mo atom and undergoes rapid and irreversible loss of function due to oxidation. Therefore, it was assumed that Moco does not occur free but permanently protein-bound in the cell. A cellular Moco distribution system should meet two demands: (1) It should bind Moco subsequent to its synthesis, and (2) it should maintain a directed flow of Moco from the Moco donor Cnx1 to the Moco-dependent enzymes. The availability of sufficient amounts of Moco is essential for the cell to meet its changing demand of Mo enzymes.

A Moco-storage/carrier protein has been described for the green alga *Chlamydomonas reinhardtii* (a homotetramer that can hold four Moco molecules) donating Moco to NR. In *Arabidopsis thaliana*, recently a novel protein family was identified (Kruse et al. 2010) consisting of eight members (Fig. 1) that all can bind Moco and are therefore named Moco-binding proteins (MoBP). They seem to be involved in the cellular distribution of Moco. Obviously, land plants with differentiated organs need more MoBPs than a unicellular and motile alga. It is, however, still open whether these proteins represent the default-pathway for Moco-allocation to its users or whether it is only a buffer, and the default-way goes directly from Cnx1 to the appropriate apo-Mo-enzymes.

Insertion of molybdenum cofactor into Mo-enzymes

Among eukaryotes, nothing is known how FAD, iron-sulfur centers, and heme are bound to the Mo-enzymes. For Moco, however, the first crystallographic analyses of Moenzymes made evident that the cofactor is deeply buried within the holoenzyme so that Moco could only have been incorporated prior to or during completion of folding and dimerization of the apoprotein monomers. As there are three classes of Mo enzymes known in higher organisms (Hille et al. 2011) also differences in the insertion of Moco might be considered. For insertion of Moco into the target apo-enzymes either (still unknown) chaperone proteins would be needed or the Moco-carrier/binding proteins could become involved at this stage which has been shown for the Arabidopsis MoBP proteins. Once being introduced into the apoprotein, Mo is coordinated by additional ligands present in the active center of the respective Moenzyme. Like every other protein, also Mo-enzymes are

finally degraded after having fulfilled their tasks. One can assume that Moco is degraded as well because free Moco released from proteins is extremely labile and sensitive to oxidation.

Post-translational activation of AO and XDH

Eukaryotic Mo-enzymes are subdivided into two groups: NR and SO belong to the SO family of Mo-enzymes and are activated by insertion of Moco. XDH and AO, both members of the XO family of Mo-enzymes, require a final step of maturation during or after insertion of Moco, i.e. the addition of a terminal sulfido group to the Mo-center in order to gain enzymatic activity (Hille et al. 2011). This sulfur ligand is added by a separate enzymatic reaction catalyzed by the Moco-sulfurase ABA3 (Fig. 1). ABA3 is a homodimeric two-domain protein (Bittner et al. 2001) with its N-terminal domain sharing structural and functional homologies to bacterial cysteine desulfurases. In a pyridoxal phosphate-dependent manner, the N-terminal domain of ABA3 decomposes L-cysteine to yield alanine and elemental sulfur (Heidenreich et al. 2005), the latter being bound as a persulfide to a highly conserved cysteine residue of ABA3 (Fig. 1). The C-terminal domain of ABA3 shares a significant degree of similarity to the third class of Mo-enzymes in eukaryotes (=mARC proteins) and was shown to bind sulfurated Moco, which receives the terminal sulfur via intramolecular persulfide-transfer from the N-terminal domain (Wollers et al. 2008). It is likely that subsequent to Moco-sulfuration, ABA3 exchanges nonsulfurated for sulfurated Moco thus activating its target Mo-enzyme.

Under physiological aspects the terminal sulfuration step provides an efficient way of regulating the amount of active XDH and AO enzymes in the cell. The concentration of physiologically active compounds like hormones, as produced by plant AO, and ROS, as produced by the action of AO and XDH from many species, can rapidly be increased by changing the ratio of inactive and active XDH and AO molecules. In fact, a rapid induction of the *aba3* gene was found upon drought and salt stress in *A. thaliana* as well as upon ABA treatment (Xiong et al. 2001), thereby being consistent with the conditions required for induction of AO and XDH (Seo and Koshiba 2002; Hesberg et al. 2004; Yesbergenova et al. 2005).

Mo deficiency

The Mo content of plants is directly correlated to the bioavailability of Mo in the soil. The lower the soil pH, the less available is Mo thus causing Mo deficiency in plants (Mengel and Kirkby 2001). Mo-deficient plants develop a characteristic phenotype including lesions and altered morphology of leaves (referred to as "whiptail") that was first described by Arnon and Stout (1939) and later analysed in detail by Hewitt's group (Fido et al. 1977). In contrast, Mo toxicity in plants under most agricultural conditions is rare. There is an extensive literature about Mo nutrition in field-grown plants that was earlier reviewed (Kaiser et al. 2005).

Mo deficiency could also be caused by a mutation in the Mo-specific uptake system (Gasber et al. 2011). Knockout mutants in MOT1 showed a slightly altered growth pattern (Baxter et al. 2008; Tomatsu et al. 2007). As there are, however, several Mo transporters in a plant cell a severe phenotype has not yet been reported.

Finally, Mo deficiency can be caused by a defect in Moco biosynthesis which has dramatic consequences for the cell because pleiotropically all Mo-enzyme activities are lost or strongly reduced. The loss of which Mo-enzyme is most severe for the plant? (1) NR-mutants with complete loss of NR-activity are known since many years; the loss of NR activity is lethal if these mutants are cultured with nitrate as sole nitrogen source (Gabard et al. 1988; Müller and Mendel 1989). (2) The loss of SO has no phenotype provided the plants are not challenged to an atmosphere with high sulfur dioxide (Lang et al. 2007). (3) Also the loss of XDH1 has no lethal consequences although the phenotype of the plants is slightly, but not dramatically changed (F. Bittner and R. Mendel, unpublished). (4) For AO, only a mutation in AAO3 is communicated (Seo et al. 2000a, b). AAO3 is important for the conversion of abscisic aldehyde into ABA, hence its loss leads to a wilty phenotype which is severe for the survival of the plant. (5) A knockout in one of the two mARC proteins has no obvious phenotype (F. Bittner and R. Mendel, unpublished). (6) Moco-sulfurase (ABA3) deficiency was found to be basically ascribed to the reduction of ABA levels due to the lack of AO activities (Leon-Klooserziel et al. 1996; Xiong, et al. 2001) and hence a wilty phenotype. In summary, the complete loss of Moco as it occurs in cnxmutants is lethal and leads to the death of plants when they are grown in soil. In cell culture, however, these mutants can be kept alive when grown on media with reduced nitrogen as N-source.

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Conflict of interest The author declares that he has no conflict of interests.

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