



# Synthesis of novel curcumin-based aqueous polyurethane dispersions for medical textile diligences with potential of antibacterial activities

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## Abstract

In this work, a biologically active curcumin molecule is used as an antibacterial agent, and the insertion of this naturally occurring biomolecule into the backbone of water-dispersible polyurethane has been successfully achieved to synthesize bio-based antibacterial textile finishes. These curcumin-based water-dispersible polyurethane (CUR-WDPU) dispersions were prepared by utilizing isophorone diisocyanate (IPDI), polyethylene glycol (PEG), dimethylolpropionic acid (DMPA) and triethylamine (TEA) following the prepolymer mixing process by incorporating variable molar quantities of curcumin (CUR). Structure elucidation of synthesized CUR-WDPU dispersions was obtained through Fourier transformed infrared spectroscopy (FTIR) which confirmed the insertion of CUR into the WDPU backbone. Using the pad-dry-cure procedure, the varying varieties of plain weave polyester/cotton blended dyed and printed textile samples were treated with synthesized CUR-WDPU finishes. The antibacterial activities of these treated textiles have been assessed, and the outcomes revealed that the insertion of curcumin into the PU polymer chain has significantly boosted the antibacterial activities of PU dispersions. These newly prepared CUR-WDPUs dispersions are proved to be eco-friendly antimicrobial finishes because these are containing natural bioactive agents such as curcumin, showing potential antibacterial applications on polyester/cotton textiles. Predominantly, this research work is an attempt toward the greener approach of novel bio-based finishing materials preferably useful for textile diligences. Future investigations of these finishes will explore the other textile assets of poly-cotton textiles without adversely influencing their color fastness and mechanical properties.

**Keywords** Curcumin · Polyurethane · Textile finish · Antimicrobial · Functional textiles

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## Introduction

Nowadays, water-dispersible polyurethanes (WDPU)s coatings have captured the attentiveness of manufacturers due to their superior properties such as good stability at low temperature conditions, excellent flexibility, very less or zero volatile organic solvents (VOS), high level of weather and water resistance, the stability of pH, excellent resistant to solvents and very striking chemical and mechanical properties [1–4].

Aqueous PU dispersions (WDPU)s are frequently consumed for coating extensive varieties of materials such as textile, leather products, plastics, furniture and flooring, in addition to being used as base coats for automotive, topcoats for vinyl upholstery, footwear adhesives and printing inks [5–7]. Due to unique features of polyurethanes (PU)s, many researchers synthesized different types of PU)s based on a number of diols and diisocyanates and studied for various applications [8–11]. To provide ease, care and comfort to consumers in this current exotic environment, the role of textile goes beyond and changes the traditional rationale of being used as clothing to the textiles with multifunctional assets [12]. WDPU)s are also tremendously used as textile finishing agents to upgrade the durability, quality and fleeting appearance of the fabric and for the development of breathable and non-formaldehyde coatings [13, 14]. The antistatic, hydrophilic and anti-soiling finishing properties of water-dispersible polyurethanes are incredibly appropriate for synthetic fibers [15–18]. Several functional properties like: antimicrobial [19–28], wrinkle resistance [19, 22, 23, 29–31], softness [21], stain resistant [24, 25], antistatic charges [26], hydrophilicity and UV resistance/blocking properties are pioneered in textile goods [27, 28, 32]. All these potential properties can be acquired separately or combined, to obtain multifunctional textiles using chemical finishing materials [12]. In textile manufacturing, utilization, transportation and storage, the goods can be stricken by microbes like bacteria, fungi and algae. These microbes have the ability to infect textiles and causing fiber damage, imparting unpleasant odors, slick and slimy feel [12].

In the quest to protect human beings and the textile products from all expected bacterial infections, antibacterial finishing of textiles was considered the best solution. Copious chemical finishing agents have been utilized for antimicrobial finishing of textile products such as silver,  $\text{TiO}_2$ , chitosan, salts containing metal atoms, triclosan, ZnO materials and quaternary ammonium salts [19, 33–37]. Moreover, curcumin [38–41], vanillin [42, 43] and tertiary butyl hydroquinone (TBHQ) [44–48] are the polyphenolic compounds which can be used to achieve antimicrobial properties. Curcumin is obtained from *Curcuma longa* plant that imparts numerous pharmacological effects. This naturally occurring compound manifests remarkable antibacterial [47], antifungal [48], antioxidant [49], anti-inflammatory [50], anticoagulant [51] and antitumor [52] activities. Bioavailability of curcumin is severely restricted due to its very low solubility in aqueous medium, degradation at alkaline pH, inadequate tissue absorption and rapid systemic elimination [53–56]. Curcumin has been used as a natural healing and antimicrobial agent for wounds [57]. Traditionally, it is also used as a dyeing or

coloring agent in textiles [40, 41]. In current research work, we used curcumin as an antibacterial agent in water-dispersible polyurethane to achieve bio-based antibacterial textile finishes. The insertion of naturally occurring bioactive polymers as chain extenders into the polymeric chain of WDPUs has become the source of attraction for polymer scientists [7, 58, 59]. This addition played a considerable part in enhancing the antibacterial properties of the finished textiles. To our best knowledge, we report that curcumin is employed to extend the polymeric chain of water-dispersible polyurethane for the first time. In this research project, curcumin-based water-dispersible polyurethanes are formed by using isophorone diisocyanate (IPDI), polyethylene glycol (PEG), dimethylolpropionic acid (DMPA), dibutyltin dilaurate (DBTDL), triethylamine (TEA) and curcumin (CUR). Furthermore, physical, structural and antibacterial investigations were carried out in order to estimate the influence of varying molar ratios of curcumin on final WDPU properties.

## Experimental section

### Materials

Curcumin (99%), dimethylolpropionic acid (DMPA, 99.9%), isophorone diisocyanate (IPDI, 99.9%) and dibutyltin dilaurate (DBTDL, 99.9%) were bought from Sigma Aldrich Chemical Co, USA. Polyethylene glycol (PEG, 99%) ( $M_n=600$  g/mole), triethylamine (TEA, 99%), acetic acid (99%), acetone (99%) and methyl ethyl ketone (MEK, 99%) were of analytical grade and purchased from Merk Chemical Co, UK. Deionized water, emulsifiers (WN with HLB=14 and NP-6 having HLB=10) were purchased from a local textile chemical market. DMPA and PEG were dried to ensure the removal of water vapor and air at 60 °C for 4 h in the oven before usage. To dry MEK, anhydrous  $\text{CaSO}_4$  was used. All the other chemicals were utilized as received. Amounts of the reactants for reactions were calculated by the formula: amount=no. of moles  $\times$  molecular mass.

### Pretreatment of textile substrates (cotton/polyester blended)

Dyed and printed poly-cotton plain weave fabric was provided by Kamal Textile Mills Ltd., Khurarrarianwala, Faisalabad, Pakistan and its specifications are given in Table 1. All the fabric samples were washed before application of CUR-WDPU-IPDI finishes, at 100 °C for 30 min with detergent in the laboratory. Afterward, the fabric samples were rinsed and dried at room temperature, and pH of the fabrics was maintained at about 6.5–7.5.

**Table 1** Specifications of the used fabric

Sr. no	Quality	Construction/count	Blend ratio cotton/poly-ester	GSM (g/m <sup>2</sup> )	PPI <sup>a</sup>	EPI <sup>b</sup>	Processed application
01	Plain weave; poly-cotton	(40×40/96×76)	44/56	106	76	96	Reactive dyes
02	Plain weave; poly-cotton	(40×40/96×76)	44/56	111	76	96	Pigment dyes

<sup>a</sup> Picks Per Inch; number of weft threads per inch of woven fabric

<sup>b</sup> Ends Per Inch; number of warp threads per inch of woven fabric

## Experimental procedure

Preparation of PU prepolymer with NCO terminals and extension of this polymeric chain were carried out according to the suggested synthetic route [60]. Following this synthetic path, NCO end capped PU prepolymer was formed in a round bottom glass reactor fixed up with a reflux condenser, nitrogen inlet, mechanical stirrer, thermometer and a temperature regulator. First, PEG (1.0 mol) and DMPA (0.8 mol) were added into the reactor. Reaction was carried out at 80–90 °C for 30 min. Then, a single drop of the catalyst (DBTDL) was added into the reaction mixture followed by the addition of isophorone diisocyanate (IPDI) (2 mol) under vigorous stirring. At this moment, the reaction mixture was allowed to react for further 2 h at 70–80 °C, which resulted into a hydrophilic PU prepolymer with NCO terminals. To check the progression of PU prepolymer formation, the FTIR spectrum of this synthesized NCO end capped PU prepolymer was acquired (Fig. 2 (d)).

TEA (0.9 mol) was added to NCO terminated PU prepolymer as neutralizing agent to neutralize the carboxylic (–COOH) groups that exist in PU polymeric chain, and this neutralization was accomplished for the next 45 min at 55 °C. Neutralizing agent was added in slight excess to that of DMPA to validate the neutralization process [61]. To decrease the viscosity of polymer solution, very little amount of methyl ethyl ketone (MEK) was added into the reaction mixture. The formation of neutralized NCO terminated PU prepolymer was confirmed through FTIR analysis (Fig. 1).

Chain extension of the neutralized PU prepolymer was performed by curcumin (dissolved in the appropriate amount of MEK). Chain extension step was carried out for next 30 min followed by dropwise addition of an estimated volume of deionized water under vigorous stirring. The dispersion step was further preceded at room temperature for next 2 h. A stable CUR-WDPU dispersion having 35% solid content was formed. The schematic illustration of the synthesis of curcumin-based WDPU is shown in Fig. 1. By the above described process, a total of 5 samples of curcumin-based water-dispersible polyurethanes using isophorone diisocyanates (IPDI) were prepared. The detailed formulation and sample code designation of all these samples is given in Table 2.



**Fig. 1** A schematic illustration of the synthesis of curcumin-based water-dispersible polyurethane

**Table 2** Formulation of curcumin-based WDPU dispersion

Sample codes	IPDI <sup>a</sup> (moles)	PEG <sup>b</sup> (moles)	DMPA <sup>c</sup> (moles)	TEA <sup>d</sup> (moles)	Curcumin (moles)
CUR-WDPU-01	2.0	1	0.8	0.9	0.01
CUR-WDPU-02	2.0	1	0.8	0.9	0.02
CUR-WDPU-03	2.0	1	0.8	0.9	0.03
CUR-WDPU-04	2.0	1	0.8	0.9	0.04
CUR-WDPU-05	2.0	1	0.8	0.9	0.05

<sup>a</sup> Isophorone diisocyanates

<sup>b</sup> Polyethylene glycol

<sup>c</sup> Dimethylolpropionic acid

<sup>d</sup> Triethylamine

## Finish application

Two different emulsifiers (WN having HLB value 14 and NP-6 having HLB value 10) were dissolved in 1 L of distilled water to which 20 g or 40 g of CUR-WDPU dispersion was added. This mixture was homogenized by a mechanical stirrer for 5–10 min to prepare a 2% or 4% solution of the polymeric dispersion. Dyed and printed textile swatches were immersed into the prepared dispersions. Pad-dry-cure method was used for dispersion applications, and the wet textile swatches were squeezed after some minutes between two stainless steel rollers of the pad-der machine. The padding pressure of rollers was adjusted to allow a pick-up of 75%, and padding speed was kept at 3 m/min. After that, the textile swatches were dried and cured at 150 °C for 1 min in an electric oven.

## Investigations of physicochemical parameters

### Physical characteristic measurements

Physical characterizations of CUR-WDPU emulsions have been done to measure a number of variables such as emulsion stability, emulsion appearance, film appearance and tackiness. All these parameters were measured at 25 °C.

### Solid content measurements

The dry weight contents or solid contents of CUR-WDPU's have been measured by taking a weighed quantity of formulated dispersion in an aluminum cup that was put into an electric oven for drying at 80 °C till the weight of dry contents was constant. Solid contents were then estimated using the following equation.

$$\text{Solid contents (\%)} = \frac{A - B}{D} \times 100$$

where “B” is the mass of empty aluminum cup, “D” is mass of the cup and CUR-WDPU before drying and “A” is the mass of cup and CUR-WDPU after drying.

### Molecular characterization

In order to validate the incorporation of curcumin into the PU polymeric chain, the molecular characterization of the synthesized samples was achieved in ATR mode by a Bruker-IFS 48 FTIR spectrometer (Ettlingen, Germany). All the FTIR spectral analysis of monomers and synthesized CUR-WDPU emulsion samples were recorded at 400–4000 cm<sup>-1</sup> wavelength region.

## Evaluation of antibacterial activities

Disk diffusion assay has been used to evaluate the antibacterial activity of synthesized CUR-WDPUs dispersions [62]. For inhibition studies, three different strains of bacteria were taken. 1000 mL of medium based on nutrient agar was prepared from which 150 mL of nutrient agar was poured in three flasks and autoclaved at 120 °C for 15 min and then cooled to room temperature. Afterward, the actively growing bacterial strains (*Escherichia coli*, *Staphylococcus aureus* and *Bacillus subtilis*) were added to the cooled mediums. Each sterilized petri plate was filled with around 20 mL of nutrient agar medium and allowed to set at ambient temperature. The untreated and treated textile swatches were cut into about  $1 \pm 0.5$  mm disks and placed on the prepared petri plates. All the plates were put in the incubator for 24 h at 37 °C [62]. After incubation, the zones of inhibition were measured in millimeters (mm) around the untreated and treated textile disks.

## Results and discussion

### Physical characterization of curcumin/water-dispersible polyurethane based on IPDI (CUR-WDPU-IPDI)

The outcomes of physical characterization of CUR-WDPUs based on IPDI by varying mole ratio of curcumin are shown in Table 3. Solid contents of the prepared series of CUR-WDPU-IPDI emulsions range from 35.60 to 39.89%. The gradual enhancement in the dry weight contents may be due to the continuing rise in the molar amount of curcumin. The appearance was approximately identical in all the synthesized emulsions, i.e., mustard transparent, and the diluted solution of all the emulsions was yellow. This yellow color was due to curcumin which is a yellow-colored bioactive compound, incorporated into the WDPU backbone as a chain extender. Tackiness is another considerable aspect of the coated material. To observe the tackiness, each one of the synthesized samples was tack-free. The consequences associated to the stability of the CUR-WDPU-01 to CUR-WDPU-05 emulsions are

**Table 3** Physical characteristics of curcumin-based water-dispersible polyurethanes based on IPDI

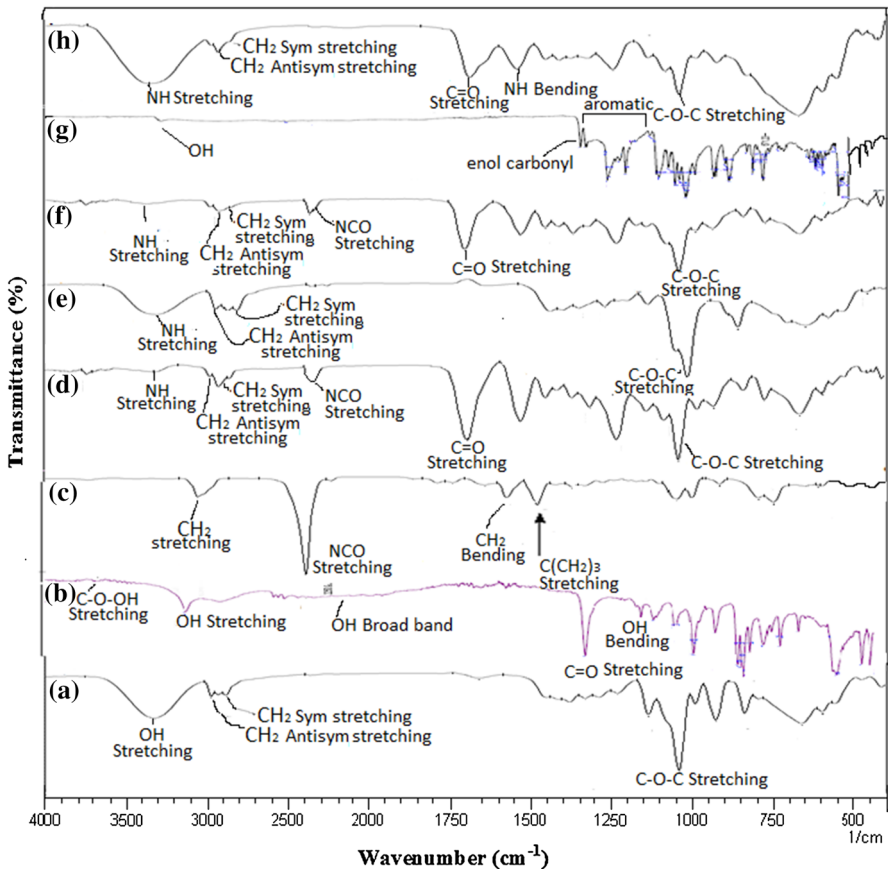
Sample code	Solid contents (%)	Emulsion appearance	Emulsion stability	Tackiness and film appearance
CUR-WDPU-01	35.60	Mustard transparent	> 1 year	Tack free bronzed
CUR-WDPU-02	37.20	Mustard transparent	> 1 year	Tack free bronzed
CUR-WDPU-03	37.89	Mustard transparent	> 1 year	Tack free bronzed
CUR-WDPU-04	38.07	Mustard transparent	> 1 year	Tack free bronzed
CUR-WDPU-05	39.89	Mustard transparent	> 1 year	Tack free bronzed

displayed in Table 3 which shows that stability of all the emulsions was approximately the same, i.e., greater than one year.

### Structural characterization

FTIR spectra of all the monomers such as IPDI, PEG (Mn = 600), TEA, DMPA, curcumin and reaction intermediate products such as NCO terminated PU prepolymer, neutralized NCO terminated PU prepolymers, and final product of CUR-WDPUs dispersions are shown in Fig. 2. All the other FTIR spectral series of five CUR-WDPUs are displayed in Fig. 3.

FTIR spectrum of polyethylene glycol (PEG) (Mn = 600) is shown in Fig. 2a. It shows a broad band at  $3439.08\text{ cm}^{-1}$  representing OH stretching, and the peaks observed at  $2879.72\text{ cm}^{-1}$  and  $2972.31\text{ cm}^{-1}$  are ascribed to  $\text{CH}_2$  antisymmetric



**Fig. 2** FTIR spectra: **a** PEG (Mn=600) **b** DMPA **c** IPDI **d** NCO terminated PU prepolymer **e** TEA **f** Neutralized NCO terminated PU prepolymer **g** Curcumin **h** Final CUR-WDPU-01



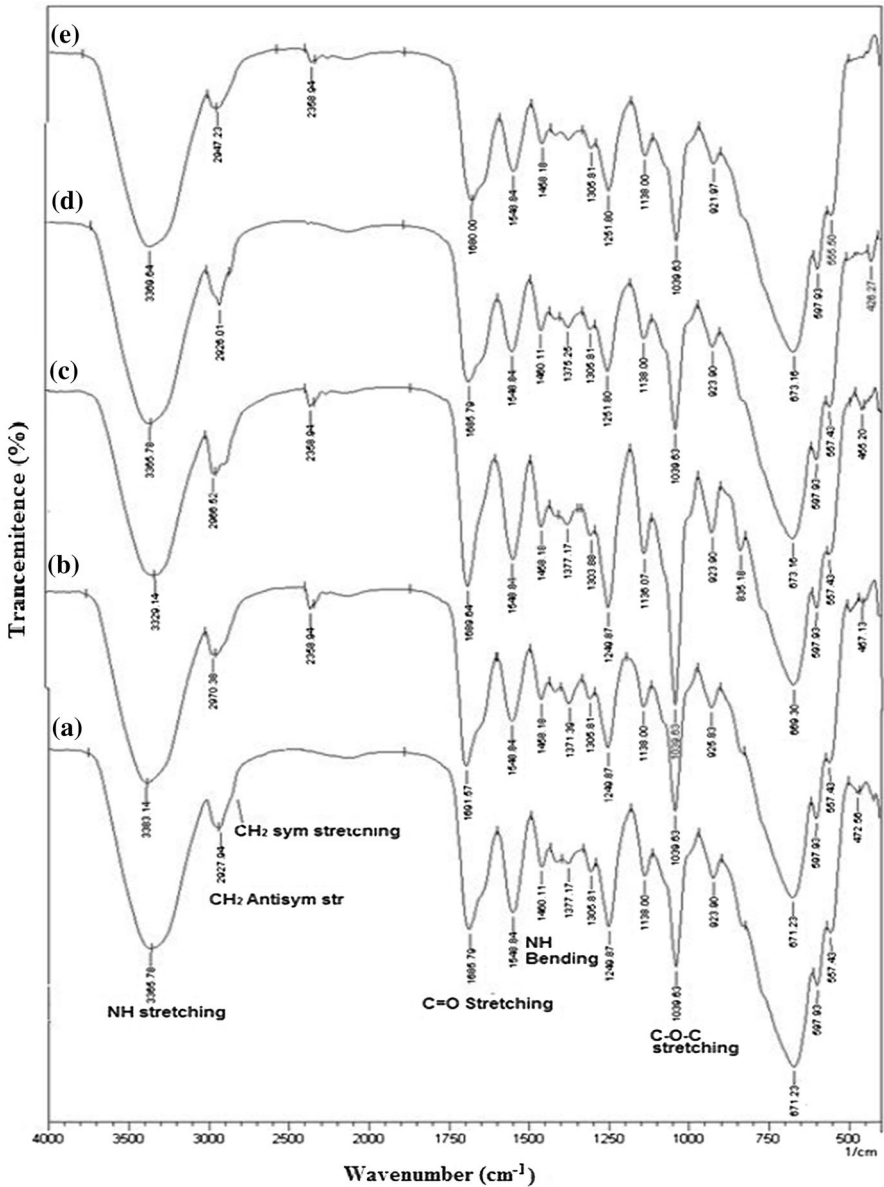


Fig. 3 FTIR spectra: a CUR-WDPU-01 b CUR-WDPU-02 c CUR-WDPU-03 d CUR-WDPU-04 e CUR-WDPU-05

and symmetric stretchings. Vibrational stretching of C–O–C was observed at  $1041.56\text{ cm}^{-1}$ .

Dimethylolpropionic acid (DMPA) spectrum (Fig. 2b) displayed a broad band at  $3560.59\text{ cm}^{-1}$  due to carboxylic group (–COOH) stretching. The other prominent

bands at  $3360.65\text{ cm}^{-1}$  and  $2588.47\text{ cm}^{-1}$  are attributed to the hydroxyl ( $-\text{OH}$ ) group stretching vibrations. The bending vibrational peak related to  $\text{CH}_2$  was observed at  $1456.05\text{ cm}^{-1}$  [44].

FTIR spectrum of IPDI (Fig. 2c) showed an intense and very sharp peak at  $2249.00\text{ cm}^{-1}$  which is due to the isocyanate ( $-\text{NCO}$ ) stretching vibrations. The signal associated with  $\text{CH}_2$  symmetric vibration is appeared at  $2951.09\text{ cm}^{-1}$ , while the signal at  $1463.97\text{ cm}^{-1}$  displayed  $\text{CH}_2$  bending vibrations. The peak at  $1359.82\text{ cm}^{-1}$  is attributed to  $\text{C}(\text{CH}_3)_2$  present on the carbocyclic ring of IPDI.

PEG, DMPA and IPDI were reacted in a flask in the presence of a catalyst DBTDL which resulted to the formation of PU prepolymer with NCO terminals. FTIR spectrum of NCO end capped PU prepolymer (Fig. 2d) exhibited a broad signal at  $3323.35\text{ cm}^{-1}$  which is attributed to  $-\text{NH}$  stretchings [59]. Other significant peaks have been observed at  $1699.29\text{ cm}^{-1}$  entitled to  $-\text{C}=\text{O}$  stretchings; peaks seemed at  $2889.37\text{ cm}^{-1}$  and  $2962.66\text{ cm}^{-1}$  are associated to  $-\text{CH}$  symmetric and  $-\text{CH}$  asymmetric stretchings of  $-\text{CH}_2$  group. The NCO end capped PU prepolymer was further allowed to react with triethylamine that leads to the formation of neutralized PU prepolymer with NCO terminals. The FTIR spectrum of neutralized PU prepolymer with NCO terminals (Fig. 2f) showed a prominent broad peak at  $3323.35\text{ cm}^{-1}$  which is attributed to  $-\text{NH}$  stretching [63, 64]. The  $-\text{CH}_2$  asymmetric stretching was detected at  $2960.73\text{ cm}^{-1}$  and symmetric stretching at  $2881.65\text{ cm}^{-1}$ . The other sharp signals that appeared at  $1697.36\text{ cm}^{-1}$  and  $1529.55\text{ cm}^{-1}$  were associated to  $-\text{C}=\text{O}$  and  $-\text{C}=\text{C}$  stretchings, respectively. From the FTIR spectrum of neutralized polyurethane prepolymer with NCO terminals, it was clearly observed that the peak associated with isocyanate (NCO) groups has been insignificant in appearance. Afterward, neutralized NCO terminated PU prepolymer extended with curcumin which resulted in the formation of proposed curcumin-based water-dispersible polyurethane (CUR-WDPU). FTIR spectrum of curcumin, presented in Fig. 2g, showed a prominent peak at  $3515\text{ cm}^{-1}$  associated with phenolic  $-\text{OH}$  stretching vibration, sharp absorption peaks appeared at  $1627$  and  $1602\text{ cm}^{-1}$  due to enol-carbonyl stretchings [63, 64]. Similar stretchings of benzene ring appeared at  $1600\text{--}1400\text{ cm}^{-1}$ . The peak at  $1597\text{ cm}^{-1}$  is for  $-\text{C}=\text{O}$ , at  $1508\text{ cm}^{-1}$  for  $-\text{C}=\text{C}$  vibrations and at  $1426\text{ cm}^{-1}$  for olefinic  $-\text{CH}$  bending vibrations, and the absorbance peaks at  $1274\text{ cm}^{-1}$ ,  $1197\text{ cm}^{-1}$ ,  $1153\text{ cm}^{-1}$  and  $1024\text{ cm}^{-1}$  are related to  $-\text{C}=\text{O}$  stretching vibrations. FTIR spectrum of CUR-WDPU-1 (Fig. 2h) showed typical signals for  $-\text{NH}$  stretchings at  $3365.78\text{ cm}^{-1}$  and  $-\text{CH}$  symmetric stretching associated to  $-\text{CH}_2$  groups was seemed at  $2927.94\text{ cm}^{-1}$ . The absorption peaks at  $1685.79\text{ cm}^{-1}$  and  $1548.84\text{ cm}^{-1}$  were correlated to  $-\text{C}=\text{O}$  stretching and  $-\text{NH}$  deformations. The peaks at  $1454.33\text{ cm}^{-1}$ ,  $1409.96\text{ cm}^{-1}$  and  $1323.17\text{ cm}^{-1}$  are associated to  $\text{CH}_2$  stretching, bending and wagging vibrations, respectively.  $\text{C}-\text{O}-\text{C}$  stretching has been observed at  $1041.56\text{ cm}^{-1}\text{--}1244.09\text{ cm}^{-1}$ . On extending the neutralized NCO end capped PU prepolymer with varying mole ratios of curcumin, the FTIR spectral series of CUR-WDPU 01 to CUR-WDPU-05 is presented in Fig. 3(a–e). All the five spectra have shown identical peaks of  $-\text{NH}$  stretching vibrations,  $-\text{CH}$  symmetric and asymmetric stretching vibrations of  $\text{CH}_2$  groups, hydrogen-bonded  $-\text{C}=\text{O}$  stretching,  $-\text{NH}$  bending and  $\text{C}-\text{O}-\text{C}$  vibrations of curcumin-ether-type absorption.

## Antimicrobial activity of curcumin water-dispersible polyurethanes based on IPDI (CUR-WDPU-IPDI)

Curcumin is being considered the safest, effective antimicrobial agent for ages. Besides, it is also used on cotton, wool and other textile materials as a potential inhibitor for microbial growth [40, 65]. For the advancement of antimicrobial skin ointments and suspensions with enhanced wound dressing and skin protection properties, curcumin amalgamation with diverse antimicrobial agents is utilized [66]. Previous research works confirmed the biocompatibility of polyurethanes [67, 68], and the current study assessed the antibacterial potential of CUR-WDPUs based on IPDI after applying on dyed/printed poly-cotton fabrics. The outcomes reported in Table 4 presented the higher antimicrobial activity of finishes against gram-positive bacterial strains such as *Bacillus subtilis* and *Staphylococcus aureus* than gram-negative bacterial strains such as *Escherichia coli*. Gul and Bakht [69] investigated that curcumin excellently inhibited the growth of certain gram-positive and gram-negative bacterial strains. It was also observed that an increase in the concentration of curcumin results in increased antibacterial activity as inferred from the previous studies [70, 71]. The enhanced antibacterial properties of the fabric are very important for usage in textile diligence to prevent or delay the growth of microbes on the surface of the textile.

Curcumin has suppressed the gram-positive and gram-negative bacterial cytokinesis through induction of filamentation. The investigations on *B. subtilis*, *S. aureus* and *E. coli* reported that curcumin has the ability to inhibit FtsZ polymerization due to which prokaryotic cell division is disrupted [72–74]. The zone of inhibition generated around the fabric disks was measured which ranges from 10 to 15 mm (Fig. 4 and Table 4). The outcomes showed that the dyed and printed fabrics, after application of curcumin-based water-dispersible polyurethanes, were found to be more effective against all the tested microbes.

## Conclusion

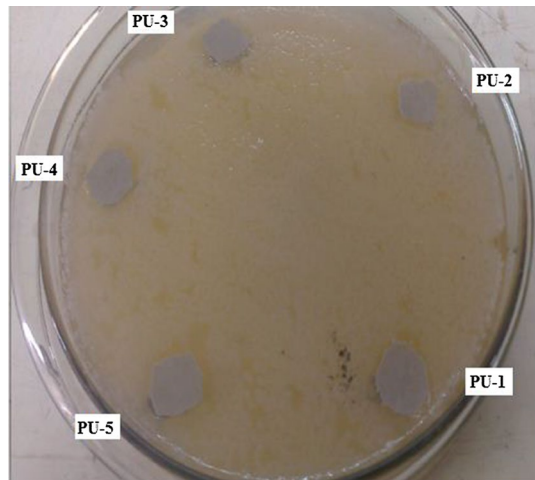
The incorporation of biologically active molecules into the water-dispersible polyurethanes (WDPU) has been comprehended to be an advanced technique to introduce bio-based finishing materials for various textile products. Novel curcumin-based water-dispersible polyurethane series containing CUR-WDPU-01 to CUR-WDPU-05 having varying mole ratios of curcumin were synthesized. By the prepolymer mixing process, the preparation of neutralized NCO end capped PU prepolymer has been accomplished using isophorone diisocyanate (IPDI), polyethylene glycol (PEG of  $M_n=600$ ), dimethylolpropionic acid (DMPA) and triethylamine (TEA) with molar ratios of 2:1:0.9:0.8. In the chain extension step, the neutralized NCO terminated PU prepolymer was extended with varying molar ratios of pristine curcumin. The formation of aqueous dispersion was achieved by the addition of estimated amount of deionized water. Structural characterization of all the synthesized CUR-WDPUs samples was done using the FTIR technique which confirmed the suggested structure of finally synthesized CUR-WDPU emulsions. The physical parameters, for example appearance of emulsion, the stability

**Table 4** Antibacterial activity of treated (dyed, printed) with CUR-WDPU-IPDI and untreated textiles

Bacterial species	Inhibition zone (mm)*											
	Dyed			Printed								
	<i>Escherichia coli</i>	<i>Bacillus subtilis</i>	<i>S. aureus</i>	<i>Escherichia coli</i>	<i>Bacillus subtilis</i>	<i>S. aureus</i>						
Fabric sample	2% 4%	2% 4%	2% 4%	2% 4%	2% 4%	2% 4%						
CUR-WDPU-01	10 ± .43	10 ± .42	11 ± .67	10 ± .20	8 ± .11	8 ± .36	10 ± .34	10 ± .93	12 ± .26	11 ± .03	07 ± .56	08 ± .56
CUR-WDPU-02	10 ± .24	12 ± .56	12 ± .39	11 ± .53	8 ± .23	8 ± .27	11 ± .89	09 ± .73	13 ± .04	11 ± .46	08 ± .65	07 ± .23
CUR-WDPU-03	11 ± .90	10 ± .54	11 ± .27	10 ± .20	9 ± .10	8 ± .74	10 ± .24	11 ± .02	12 ± .83	12 ± .56	09 ± .04	09 ± .10
CUR-WDPU-04	13 ± .01	12 ± .83	12 ± .04	13 ± .10	10 ± .09	10 ± .67	12 ± .28	12 ± .03	13 ± .20	13 ± .29	09 ± .40	09 ± .20
CUR-WDPU-05	15 ± .20	13 ± .04	15 ± .90	15 ± .01	11 ± .73	10 ± .47	12 ± .23	11 ± .08	14 ± .22	13 ± .12	10 ± .12	10 ± .07
Untreated	—	—	—	—	—	—	—	—	—	—	—	—

\*Each zone of inhibition is expressed as mean ± standard error (S.E.) (n = 5)

**Fig. 4** Photographs displaying the antimicrobial assessment of the treated textiles via disk diffusion assay



of the emulsion, film appearance, tackiness and solid contents (%), were investigated. The dyed and printed poly-cotton textile swatches were treated with 2% or 4% dilutions of polymeric dispersion solution by the pad-dry-cure procedure. The investigations of antibacterial activities of all the untreated and treated textile swatches have been done by disk diffusion assay. The outcomes of antibacterial assessment revealed that the post-treatment of textiles with synthesized CUR-WDPUs results in a remarkable rise in antibacterial activity. Overall, the best outcomes were achieved by 2% dilution of CUR-WDPUs as compared to 4% dilution. This could be ascribed to better penetration and excellent compatibility between both fabric and CUR-WDPUs. Moreover, it was observed from the results that by increasing the molar quantity of curcumin into the PU polymeric chain, the significant enhancements in antibacterial activity were exhibited. These newly synthesized CUR-WDPUs are eco-friendly, and bio-based antibacterial textile finishes with many other potential applications for polyester/cotton textiles. Future investigations of these finishes will explore the other textile assets for polyester/cotton textiles without adversely affecting their color fastness and mechanical properties. The entire study is a pioneer step on the way to the greener approach and bio-based finishing materials that can almost certainly be utilized for perspective textile usages.

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**Data availability** Data regarding all the experiments have been included in the manuscript; therefore, no further data are available for sharing.

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