ENVIRONMENTAL BIOTECHNOLOGY

Interspecies distances between propionic acid degraders and methanogens in syntrophic consortia for optimal hydrogen transfer

Monika Felchner-Zwirello · Josef Winter · Claudia Gallert

Received: 11 October 2012 / Revised: 21 November 2012 / Accepted: 22 November 2012 / Published online: 12 December 2012 © Springer-Verlag Berlin Heidelberg 2012

Abstract A mixed culture from an anaerobic biowaste digester was enriched on propionate and used to investigate interspecies hydrogen transfer in dependence of spatial distances between propionate degraders and methanogens. From 20.3 mM propionate, 20.8 mM acetate and 15.5 mM methane were formed. Maximum specific propionate oxidation and methane formation rates were 49 and 23 mmol $mg^{-1}day^{-1}$, respectively. Propionate oxidation was inhibited by only 20 mM acetate by about 50 %. Intermediate formate formation during inhibited methanogensis was observed. The spatial distribution and the biovolume fraction of propionate degraders and of methanogens in relation to the total population during aggregate formation were determined. Measurements of interbacterial distances were conducted with fluorescence in situ hybridization by application of group-specific 16S rRNA-targeted probes and 3D image analyses. With increasing incubation time, floc formation and growth up to 54 µm were observed. Propionate degraders

M. Felchner-Zwirello · J. Winter · C. Gallert Institut für Ingenieurbiologie und Biotechnologie des Abwassers, KIT—Karlsruher Institut für Technologie, Am Fasanengarten, 76131 Karlsruhe, Germany

M. Felchner-Zwirello e-mail: Monika.Felchner-Zwirello@kit.edu

J. Winter e-mail: Josef.Winter@kit.edu

M. Felchner-Zwirello Department of Analytical Chemistry, Gdańsk University of Technology, Gabriela Narutowicza 11/12 Str., 80-233 Gdańsk, Poland

C. Gallert (🖂)

FB Technik/Abteilung Mikrobiologie-Biotechnologie, Hochschule Emden-Leer, Constantiaplatz 4, 26723 Emden, Germany e-mail: Claudia.Gallert@hs-emden-leer.de and methanogens were distributed randomly in the flocs. The methanogenic biovolume fraction was high at the beginning and remained constant over 42 days, whereas the fraction of propionate degraders increased with time during propionate feeding. Interbacterial distances between propionate degraders and methanogens decreased with time from 5.30 to 0.29 μ m, causing an increase of the maximum possible hydrogen flux from 1.1 to 10.3 nmolml⁻¹ min⁻¹. The maximum possible hydrogen flux was always higher than the hydrogen formation and consumption rate, indicating that reducing the interspecies distance by aggregation is advantageous in complex ecosystems.

Keywords Propionate degrading enrichment \cdot Interspecies hydrogen transfer \cdot Syntrophy \cdot Cell aggregation \cdot Interspecies distance

Introduction

Complex organic matter is degraded in anoxic or anaerobic environment to methane and carbon dioxide by syntrophic interaction of bacteria and archaea. Fermentative bacteria degrade organic substances to volatile fatty acids (VFAs), mainly acetate, propionate, and butyrate or additionally to lactate and ethanol at increasing hydrogen partial pressure. Except for acetate, these intermediates are highly reduced and must be converted by acetogenic bacteria to the methane precursors acetate, formate, or hydrogen (Gujer and Zehnder 1983; Gallert and Winter 2005). Propionate plays a crucial role due to the unfavorable thermodynamics ($\Delta G^{0} > 0$) for its fermentation to acetate, CO₂, and hydrogen. Anaerobic cleavage of propionate is possible only if the reaction products hydrogen and also acetate (Eq. 1) were removed effectively by, e.g., methanogens (Eqs. 2 and 3) or sulfate reducers in obligate syntrophic association with proton-reducing acetogenic bacteria (Schink 1992).

$$\label{eq:ch3} \begin{split} CH_3CH_2COO^- + 3H_2O &\rightarrow CH_3COO^- + HCO_3^- + H^+ + 3H_2 \\ \Delta G^{0\prime} &= +76.1 \ \text{kJ/reaction} \end{split} \tag{1}$$

 $CH_3COO^- + H_2O \rightarrow CH_4 + HCO_3^- \quad \Delta G^{0'} = -31 \text{ kJ/reaction}$ (2)

$$4H_2 + HCO_3^- + H^+ \rightarrow CH_4 + 3H_2O$$
 $\Delta G^{0'} = -135.6 \text{ kJ/reaction}$
(3)

The problem of VFA accumulation, especially of propionate, in anaerobic digesters has been recorded repeatedly (e.g., Inanc et al. 1999; Wang et al. 2006; Gallert and Winter 2008). The thermodynamic disadvantage of propionate degradation (Eq. 1) makes it the substrate, which often persists in anaerobic reactors (Shin et al. 2010). Unfavorable thermodynamic conditions prevail also for n-butyrate (ΔG^{0} '=+48.1 kJ). Both VFAs require a very low hydrogen partial pressure for degradation and hence syntrophic partner organisms (Dong and Stams 1995).

Propionate is the "leading" volatile fatty acid during anaerobic digestion, being converted to acetic acid, CO₂, and H₂ (Gerardi 2003). Two metabolic pathways for anaerobic propionate oxidation were described: In bacteria that express the methyl-malonyl-CoA pathway, propionate is fermented to acetate and CO₂ (Plugge et al. 1993) with release of reducing equivalents for oxidation of succinate to fumarate, malate to oxaloacetate, or pyruvate to acetyl-CoA, whereas bacteria that express the C6-dismutation pathway (de Bok et al. 2001) convert 2 mol of propionate to 3 mol of acetate with *n*-butyrate as an intermediate. Hydrogen and/or formate utilization for methane production or sulfate reduction is the driving force of anaerobic propionate oxidation. In anaerobic digesters, coupled methane formation keeps the hydrogen partial pressure pH₂ low enough for propionate oxidation (Thauer et al. 1977). The maximal pH₂ which still allows anaerobic butyrate or propionate oxidation ($\Delta G' < 0$) is 186 or 6.5 Pa, respectively at 25 °C, pH of 7, and 1 mM propionate concentration. Therefore, the specific activity of hydrogen-oxidizing methanogens controls favorable bioenergetic conditions for methane formation from organic material (Dwyer et al. 1988). Bryant et al. (1967) and Wolin (1975) called this process interspecies hydrogen transfer, but formate apparently can also be used as a buffer for interspecies electron flow (Thiele and Zeikus 1988; Bleicher and Winter 1994; Stams and Dong 1995). Comparing the diffusion constant and the solubility of hydrogen and formate, it can be expected from Fick's diffusion law that formate transfer from the producer to the consumer is the favored mechanism when interspecies distances are high, while hydrogen transfer becomes more favorable when interspecies distances are low (de Bok et al. 2004). Recent biochemical studies and genome analyses of pure cultures of syntrophic volatile fatty acid degraders indicated that hydrogen and formate transfer may occur simultaneously (Müller et al. 2010).

Small interspecies distances can be maintained by growth in flocs, granules, or aggregates or by inoculation of additional hydrogen consumers, which leads to increased propionate degradation rates (Dwyer et al. 1988; Thiele et al. 1988; Schmidt and Ahring 1995; Imachi et al. 2000; Ishii et al. 2005). As diffusion is the major process of hydrogen transport between microbial species (McCarty and Smith 1986), it appears from Fick's diffusion law that the hydrogen flux depends, inter alia, on the distance between H₂-producing and H₂-consuming organisms. The same considerations would be valid for formate flux calculations (Stams 1994). The standard redox potential of the H^+/H_2 couple is -414 mV and is almost the same as for the CO₂/formate couple (-420 mV). Both electron transfer reactions are important in syntrophic interactions (Schink 1997). Formate formation and degradation governed by the hydrogen partial pressure are of importance for electron transfer in syntrophic cultures. Until now, interspecies distances for an optimal electron transfer have only been calculated theoretically. Applied Fick's diffusion law computations revealed interspecies distances of syntrophic propionate oxidizers from methanogens of less than 3 µm (Schmidt and Ahring 1995), 2 µm (Ishii et al. 2005), 1 µm (Stams et al. 1989; Imachi et al. 2000), or even only 0.08 µm (Shi-yi and Jian 1992).

The aim of our study was to determine the correlation between aggregate growth and resulting interspecies distances of a mixed culture, enriched from a full-scale digester with propionate as the sole carbon source. Oxidation rates of propionate at different stages of growth were balanced with acetate and methane production. Measurements of interspecies distances are reported for the first time and results are discussed in relation to data from literature.

Materials and methods

Preparation of an enrichment culture

Liquid samples of a full-scale anaerobic reactor treating a biowaste suspension from source sorted biowaste collection of the City of Karlsruhe, Germany (Gallert et al. 2003), were suspended in a mineral medium (Felchner-Zwirello et al. 2012) with addition of propionate as the sole carbon source. Enrichments on propionate were carried out over a 3-year

period by means of 10 % monthly transfers into fresh medium, resulting in a mixed culture that consisted mainly of non-filamentous hydrogentrophic methane bacteria and lemon-shaped Gram-negative propionate degraders forming aggregates. The enrichment was cultivated in serum bottles that were closed with a rubber stopper and secured with an aluminum cap under a nitrogen gas atmosphere at 37 °C with gentle shaking in darkness. The pH of the mineral medium was 7. Propionic acid (Fluka Germany) from a 1-M stock solution in distilled water was repeatedly fed to give 5 mM starting concentration. After each feeding, the slight drop of the pH was corrected with 0.1 ml NaOH (1 M) to 7. Acetate, CO_2 , and methane were the main products of the enrichment culture. Methanosarcina-like methanogens were originally present in the inoculum from the full-scale digester, forming big aggregates, but these were lost (or diluted out) during 3 years of enrichment.

Sample preparation and experimental setup

The propionate-degrading enrichment of bacteria was cultured in 250-ml Schott bottles. One hundred ml of mineral medium was inoculated with 10 ml digester effluent or pregrown enrichment culture. Directly after inoculation and every day later on, 1-ml samples were withdrawn for analyses of propionic acid, acetic acid and the protein content (as a measure of cell growth). Methane, carbon dioxide, and hydrogen were withdrawn from the gas phase and determined every day by gas chromatography. Whenever propionic acid was degraded, biomass was collected for fluorescence in situ hybridization (FISH) in order to differentiate between propionate degraders and hydrogenotrophic methane bacteria. Organisms were identified after reaction with 16S rRNA-based oligonucleotide probes specific for Bacteria, propionate degraders, and Archaea (probe sequences are provided in "Microscopy"). The experiment was divided into five phases, namely, phase 0 (directly after inoculation), growth phase 1 for degradation of the first portion of propionate (until day 13), growth phase 2 after the second propionate feeding (day 22), growth phase 3 after the third propionate feeding (day 29), and growth phase 4 after the fourth propionate feeding (day 42). The experiment was carried out in duplicate and repeated tree times. Since the results of each experiment were similar, only one representative experiment is shown in this paper.

To determine the influence of acetate on propionate oxidation rates, 5, 10, 15, and 20 mM acetate was added together with the propionate feeding to 40-ml assays. Propionate oxidation rates were then determined in the presence of acetate.

Since formate might be an alternative electron carrier during syntrophic propionate degradation, methane generation of the enrichment culture from H_2/CO_2 (80:20 %; vol/vol) and formate (15 mM) in non-inhibited and in inhibited assays (by 10 mM bromoethane sulfonic acid, BESA) was compared. Ten-milliliter cultures in 22-ml glass vials (closed with a rubber stopper and sealed with an aluminum cap) were fed with H_2/CO_2 (2 atm) or 15 mM formate in the absence of propionate. Methane production in the presence or absence of 10 mM BESA was measured daily and corrected against the blank.

Analytical measurements

Methane, hydrogen, and carbon dioxide concentrations in the gas phase were analyzed with a Chrompack CP 9001 gas chromatograph (GC) equipped with a 50-m CP-Sil 8CB column (0.32 mm internal diameter (ID)) and a thermal conductivity detector. The detection limit of hydrogen was 128 Pa, which was not sensitive enough to measure hydrogen concentrations during interspecies H₂ transfer. Propionate and acetate were measured with a combination of a model 3950-Headspace sampling unit coupled to a Varian 431C GC and a Varian 210 mass spectrometer. A 30-m CP-Wax FFAP column of 0.25 ID was used for VFA separation (Felchner-Zwirello et al. 2012). The precision of measurements (%-RSD) was for acetic acid 6.8 %, for propionic acid 7.4 % and for methane 3 %, expressed as maximal coefficient of variation obtained for the concentration range measured during the experiment.

The protein concentration was measured photometrically with a Hach Lange DR 5000 spectrophotometer according to Lowry et al. (1951) after extraction of protein from single-cell or cell aggregate suspensions with 1 ml 1 M NaOH at 95 °C for 10 min. Triplicate analyses of each parameter for each sample were carried out and mean values were calculated.

Formic acid was quantified with R-Biopharm No. 10979732035 enzymatic assay (Boehringer, Mannheim, Germany). NADH/H⁺ formation by formate dehydrogenase was determined spectrophotometrically at 340 nm. The detection limit of formate was 1 mgl⁻¹ corresponding to 0.02 mM.

Microscopy

Fluorescence in situ hybridization was performed with 16S rRNA-based oligonucleotide probes specific for *Archaea* (Arc915: 5'GTG CTC CCC CGC CAA TTC CT; Stahl and Amann 1991) for most of the *Bacteria* (Eub338: 5'GCT GCC TCC CGT AGG AGT; Amann et al. 1990) and for two propionic acid degraders: MPOB1: 5'ACG CAG GCC CAT CCC CGA A, labeled at the 5' end with 6-carboxyfluorescein (FAM) for *Syntrophobacter fumaroxidans*-related organisms and KOP1: 5'TCA AGT CCC CAG TCT CTT CGA, labeled with cyanine 3 (Cy3) for *Syntrophobacter pfennigii*-related organisms (Harmsen et al.

1996). The probes were obtained from Eurofins MWG Operon, Ebersberg, Germany. The MPOB1 probe gave a signal for both, MPOB and KOP organisms (Harmsen et al. 1995), so the results might denote the presence of both groups of propionate degraders, whereas KOP1 reacted only with S. pfennigii-related organisms. After sample hybridization, 10 µl of biomass suspension was distributed onto each hole of a Teflon-coated microscope slide, air-dried, dehydrated in ethanol, and further processed as described by Amann et al. (1990). Oligonucleotide probes were prepared at a concentration of 5 pmol μ l⁻¹, and 4 μ l of each was spread onto the specimen, together with 8 µl of hybridization buffer with 30 % formamide. Before microscopic observation, counterstaining with 0.1 µM 4,6-diamidino-2phenylindol (DAPI) on ice for 15 min in darkness was done. Each sample from a certain growth phase (0-4) was prepared in octets and ten representative microscopic images were taken with an ApoTome (Axio Imager Z1, Zeiss, Jena, Germany) per "hole", giving altogether 80 images for further analysis. The epifluorescence microscope with Apo-Tome technique was equipped with several filter sets suitable for the used dyes Cy3, FITC (or FAM), and DAPI.

A series of microphotographs were taken in order to get a 3D aggregate image, with a z-stack dimension set at 0.3 μ m. The AxioVision software (Carl Zeiss Micro Imaging GmbH, Jena, Germany) was applied for analysis of obtained microphotographs.

The interspecies distances between differently stained organisms were measured manually with all relevant z-stack images. Measurements had an accuracy of 0.01 µm. For each growth phase, the interspecies distance between MPOB1hybridized propionate degraders (S. fumaroxidans-related organisms) and Arc915-hybridized methanogens in the closest vicinity was measured, requiring three up to 15 measurements per single image. Dispersed bacteria (single cells) were only considered in phase 0 directly after inoculation. During all further growth phases, aggregates were examined. Per growth phase, 80 microscopic images were used for analysis, meaning that 240 up to 1,200 single measurements of interspecies distances were performed to calculate this value. Additionally, the size of the formed aggregates was measured: broadness (x), height (y), and thickness (z). The obtained values were averaged for each image.

Population changes during the different growth phases of the enrichment culture were evaluated with the *daime* software for digital image analysis in microbial ecology (Daims et al. 2006). The same microscopic images as described above were used for biovolume fraction evaluation, relying on the comparison of the total biovolume (biomass stained with unspecific fluorescent dye DAPI) and the biovolume fraction of the bacterial and archaeal population that reacted with 16S rRNA-targeted specific oligonucleotide probes labeled with different fluorescent dyes. Interspecies hydrogen transfer versus distances between organisms: hydrogen fluxes

Hydrogen that was released by the propionate-degrading organisms (PA) must be transferred to the hydrogenconsuming methanogens by diffusion. According to Fick's diffusion law (Eq. 4), the hydrogen flux $(J_{\rm H_2})$ depends on the surface area of the H₂-producing, propionate-degrading organisms $(A_{\rm PA})$, the hydrogen diffusion constant in water $(D_{\rm H_2})$, the difference of the hydrogen concentration outside the hydrogen producer $(C_{\rm PA})$ and the hydrogen consumer $(C_{\rm H_2})$, and the interspecies distance between the two considered organisms $(d_{\rm PAH_2})$.

$$J_{H_2} = A_{PA} \cdot D_{H_2} \frac{C_{PA} - C_{H_2}}{d_{PAH_2}}$$
(4)

The difference of the hydrogen concentration at the cell surface of the hydrogen producer (C_{PA}) and the hydrogen consumer (C_{H_2}) was calculated according to Eq. 5, giving the maximum allowed H₂ concentration for propionate oxidation ($\Delta G < 0$) of 9.34 Pa and the minimum H₂ concentration required for methane production ($\Delta G > 0$) of 0.32 Pa. The calculated difference, $C_{PA} - C_{H_2}$, was set as a constant value during the different growth phases since it was not possible to measure hydrogen in the gas or liquid phase with the necessary low detection limit. The calculated H₂ flux after measuring the interspecies distances was therefore the "maximum possible H₂ flux".

$$\Delta G = \Delta G^{\circ} + RT \ln K \tag{5}$$

Statistical analysis

Statistical analysis (one- or two-way ANOVA) and curve fittings were performed with the statistic program Graph Pad Prism 4.0 (GraphPad Software, Inc., USA).

Results

Propionate degradation

The propionate-degrading enrichment culture was cultivated for 42 days under mesophilic conditions with four propionate feedings. Degradation of the four portions of propionate during the four growth phases, cumulative acetate and methane production, as well as the increasing protein concentration are shown in Fig. 1. Whenever 5 mM propionate was anaerobically oxidized by the mixed culture, propionate was replenished. The capacity of the carbonate buffer was sufficient to stabilize the pH at 7 during the experiment. pH adjustment with 0.1 ml NaOH (1 M) was only required after Fig. 1 Syntrophic oxidation of 5-mM portions of propionate by the enrichment culture and formation of acetate and methane during growth. Cell growth was measured as increasing concentration of protein after cell or aggregate harvest and hydrolysis with NaOH



every feeding with propionic acid. According to the stoichiometries depicted in Eqs. (1) and (3), from 1 mol of propionate maximally 1 mol of acetate and 0.75 mol of methane could be produced, which was in good agreement with the analytical results shown in Fig. 1. After four feedings. 20.8 ± 0.9 mM acetate and 15.5 ± 0.4 mM methane were formed from 20.3±0.5 mM propionate. Due to cell aggregation, it was not possible to measure optical densities precisely. Therefore, protein concentration was determined as an alternative parameter for growth. However, in the cultures with aggregate formation, even withdrawal of homogeneous samples was difficult. Thus, the protein curve was not as smooth as the curves for methane and acetate (Fig. 1). Nevertheless, the increase of protein concentration revealed a typical exponential behavior over time, resulting in a doubling time of the enrichment culture of 11 days, which corresponded to a growth rate μ of 0.062 day⁻¹.

Propionate degradation and methane formation rates—influence of acetate

Anaerobic propionate oxidation was coupled with hydrogen formation but proceeded only at low hydrogen partial pressure and thus was dependent on hydrogenotrophic methanogens in our enrichment culture. Figure 2a shows propionate oxidation and methane formation rates during the successive growth phases. Both rates were highest in the third growth phase with maximum values of 0.76 and 0.57 mmoll⁻¹ day⁻¹, respectively, and matched exactly with propionate degradation and methane production rates as deductable from Eqs. (1) and (3) (0.75 mol methane produced from 1 mol propionate oxidized). Specific propionate degradation and methane formation rates are shown in Fig. 2b. To obtain the specific rates, total protein concentration was attributed to propionate degraders and methanogens by hybridizing the mixed culture with 16S rRNA-based oligonucloetide probes targeting MPOB1-related propionate degraders and Arc915-related

methanogens. As a reference, an unspecific staining of all organisms with DAPI was performed and the results were expressed as "biovolume fraction" (Fig. 6). As can be seen



Fig. 2 Propionate oxidation and methane formation rates (Fig. 2a) as well as specific propionate oxidation and methane formation rates (Fig. 2b) during the four growth phases after repeated propionate feeding. Specific rates were obtained by taking the respective total protein concentration of all bacteria from Fig. 1 and calculating the protein content of the propionate degraders and of the methanogens according to their biovolume fraction in phases 1–4 after MPOB1- and Arc915-specific labeling (Fig. 6)

in Fig. 2b. the specific propionate oxidation and methane formation rates decreased linearly with incubation time. The maximum specific propionate oxidation and methane formation rates resulted during the first feeding and were 49 and 23 mmolmg⁻¹day⁻¹, respectively. Decreasing specific propionate oxidation and methane formation rates after repeated propionate feedings may indicate an inhibition of syntrophic propionate degradation in the enrichment culture. To test whether the accumulating acetate inhibited propionate oxidation, 5, 10, 15, and 20 mM acetate were supplemented and the propionate degradation rates with initial propionate concentrations of 5 mM were measured (Fig. 3), simulating the conditions during the four successive feeding cycles of Fig. 1. Without external acetate addition, $0.32 \text{ mmoll}^{-1} \text{day}^{-1}$ propionate was oxidized, which corresponded well with the results shown in Fig. 2a, where the propionate oxidation rate in phase 1 was $0.35 \text{ mmoll}^{-1} \text{day}^{-1}$ (= 100 %). When 5 mM acetate was supplied together with 5 mM propionate, the propionate oxidation rate decreased to 77.7 % and in the presence of 10, 15, and 20 mM acetate to 60.7, 57.9, and 52.9 %, respectively (Fig. 3), indicating an increasing endproduct inhibition.

Formate as an alternative electron carrier

Hydrogen and/or formate may serve for interspecies electron flow between syntrophic propionate degraders and methanogens. Since both partners might express formate dehydrogenase, formate might be produced from NADH + H⁺ at a concentration of less than 0.01 mM and a redox potential of < -290 mV (Sieber et al. 2012) or from H₂ and CO₂. Vice versa it might be degraded by both organisms. To test whether formate was released by the propionate oxidizers and consumed by the methanogens in the syntrophic enrichment culture, batch tests with H₂/CO₂, formate, and propionate in the absence and presence of the methanogenic inhibitor BESA were performed. Methane formation (Fig. 4a) and intermediary-released formate (Fig. 4b) were analyzed. When



Fig. 3 Inhibiting effect of acetic acid on propionate degradation. The initial propionate concentration was 5 mM in each assay. Different acetate concentrations were added to each assay

H₂/CO₂ was supplied for the mixed culture, methane and significant amounts of formate were the reaction products. Whereas methane must have been generated by the methanogens, formate might have been generated by both, methanogens and/or propionate degraders. In the presence of H_2/CO_2 and propionate, BESA-inhibited assays did not produce methane but released some formate produced by the methanogens, propionate degraders, or both. In non-inhibited cultures, hydrogen and CO₂ from cleavage of externally added formate was converted to methane by the methanogens at a rate of $0.042 \text{ mmoll}^{-1} \text{day}^{-1}$, which was six times lower than the methane production rate from H_2/CO_2 (0.26 mmoll⁻¹ day⁻¹). In an assay with propionate, the methane formation rate from the degradation products H₂ and CO₂ (aceticlastic methanogens were not present in the mixed culture) was only $0.037 \text{ mmoll}^{-1} \text{day}^{-1}$ due to slow propionate degradation. No formate was detectable as a product or intermediate under these conditions. Some formate was, however, produced from propionate when methanogenesis was inhibited completely by BESA (Fig. 4a, b). Whether the propionate degraders generated the formate already or whether they excreted H₂/CO₂ and the formate was formed by the methanogens via formate dehydrogenase could not clearly be distinguished. However, it is probable that formate was generated exclusively or predominantly by the methanogens since they were able to form a similar amount of formate from H₂/CO₂ during complete



Fig. 4 Methane (a) and formate (b) production by the syntrophic culture after addition of respective substrates \pm BESA as an inhibiting agent for methanogens; *For* formate, *Prop* propionate, *BESA* bromoethane sulfonic acid

inhibition of methanogenesis with BESA (Fig. 4a, b). A direct proof of interspecies hydrogen transfer during propionate oxidation with and without BESA failed due to a too-high detection limit of hydrogen by the available gas chromatograph of 128 Pa and presumably its conversion to formate only at > 128 Pa hydrogen partial pressure when methanogenesis was inhibited by BESA. No methane formation was observed after addition of BESA to all assays, indicating the successful inhibition of methanogenesis (Fig. 4a).

Species distribution in aggregates

To visualize syntrophic propionate degraders and methanogens in naturally occurring aggregates during propionate oxidation, specific 16S rRNA-based nucleotide probes with Cv3 label (orange fluorescence) for Arc915-related methanogens and FAM label (green fluorescence) for MPOB1-related propionate oxidizers were hybridized with the respective parent DNA, and microphotographs were taken during successive growth phases (Fig. 5). The total bacterial population was visualized by counterstaining with DAPI (Fig. 5, left row). Directly after inoculation in phase 0 and after only 13 days of incubation in growth phase 1, single cells could be seen, which aggregated and formed flocs in growth phase 2 to 4. The aggregates were visualized by "3D microscopic images" obtained from a series of 2D microphotographs representing different levels of cross-section through the aggregates after conversion with the AxioVision software and "extended image definition". The growth of aggregates during the different growth phases can be seen in Fig. 5a-j. Initially, single-cell suspensions were dominant, which were succeeded by small aggregates and finally by big floc-like aggregates. Aggregate disintegration in growth phases 3 and 4 is visible in Fig. 5h, j. Flocs contained propionate degraders aside from methanogens. Some micro-colonies that only contained propionate degraders were also present (data not shown), but their distribution in aggregates was not homogenous or predictable.

From DAPI and FISH images of aggregates in samples taken during different growth phases, the proportion of each syntrophic partner group was calculated by using the *daime* software for digital image analysis. The biovolume fraction of propionate degraders that hybridized with KOP1- and MPOB1specific probes, of methanogenic bacteria that hybridized with the Arc915-specific probe, and of prokaryotic organisms that hybridized with the Bacteria-specific Eub338 probe was determined and expressed as percentage of the total population (Fig. 6). During the consecutive feeding phases, the proportion of methanogenic bacteria remained relatively constant, representing approximately 52 % of the total population. The numbers of both examined propionate-degrading groups of acetogens were increasing within successive feedings. The KOP1-related biovolume fraction increased from initially 7 to 29 % after the fourth feeding with propionate, whereas the MPOB1-related biovolume fraction that includes also other propionate degraders (*S. fumaroxidans*) aside of KOP1 organisms increased with time from 12 to 50 % of the total biovolume.

Interspecies distances and aggregates growth

The dimensions of flocs containing MPOB1-related propionate degraders and Arc915-related methanogens after successive propionate feedings and the distances between organisms were manually determined from the "3D" microscopic images (Fig. 7). It can clearly be seen that interspecies distances between propionate degraders and methanogens were decreasing with incubation time. Initially, the average distance of a single bacterium from each other was 5.3 μ m (phase 0). The interspecies distance decreased to 1.56 µm after the first propionate feeding (phase 1), presumably due to aggregation. After the second, third, and fourth feeding, the interspecies distance decreased further to 0.44, 0.53, and 0.29 µm, respectively. Decreasing interspecies distances were reversely correlated with an increasing dimension of the aggregates in x, y, and z direction. The x parameter (broadness) was taken as a representative size value of the multidimensional aggregate and was plotted against growth (Fig. 7). The size of the aggregates increased from 18.8 to 54.3 µm after four feedings. There was almost no increase of the aggregate size during incubation after the fourth feeding, indicating a decreasing stability of the aggregates with a high tendency for disintegration (e.g., Fig. 5h, j).

With Fick's diffusion law (Eq. 4), the H_2 flux (J_{H_2}) from producers to consumers could be calculated. The surface area of the H₂-producing, propionate-degrading organisms (A_{PA}) was calculated from the surface of single organisms $(1.58 \,\mu\text{m}^2)$ and multiplied by the cell number per milliliter, the hydrogen diffusion constant in water $(D_{\rm H_2})$ at 37 °C was $8.27 \times$ 10^{-5} cm² s⁻¹, and the difference of the hydrogen concentration between hydrogen producer (CPA of 9.34 Pa) and hydrogen consumer ($C_{\rm H_2}$ of 0.32 Pa) was set constant at a value of 9.02 Pa. The Henry constant for solubility of H₂ in water at 37 °C was 9.57 nmoll⁻¹Pa⁻¹. The interspecies distance (μ m) between the two considered organisms (d_{PAH_2}) was experimentally determined during the respective growth phases. Table 1 shows the calculated maximum possible hydrogen fluxes J_{H_2} during increasing incubation time. The hydrogen flux depended on the hydrogen concentration gradient between producer and consumer and on the interspecies distance. In phase 1, the maximum possible hydrogen flux was 1.1 nmolml⁻¹min⁻¹. It increased with decreasing interspecies distances to a maximal value in phase 4 of 10.3 nmolml⁻¹min⁻¹ (Table 1). The hydrogen formation rates from propionate oxidation and the hydrogen consumption rates for methane production were considerably lower than the calculated maximum possible Fig. 5 Microphotographs of the propionate-degrading enrichment culture. Left row-DAPI staining of all bacteria and right row-hybridization with Cy3-labeled Arc915 and FAM-labeled MPOB1 oligonucleotide probe of the same images showing single bacteria after inoculation in phase 0 (**a**, **b**), loose floc formation in phase 1 (c, d), aggregate growth in phases 2 and 3, respectively (e, f and g, h, respectively), and beginning aggregate disintegration in phase 4 (i, j)



hydrogen flux. Hydrogen production and hydrogen consumption rates were in the same range, first increasing with floc formation and then decreasing when flocs began to decay (Table 1).

Discussion

The propionate-degrading enrichment culture multiplied with a growth rate of μ =0.062 day⁻¹ (Fig. 1) as



Fig. 6 Population analysis of the enriched propionate-degrading culture. The biovolume fractions were estimated as percentage of organisms that hybridized with the mentioned 16S rRNA-targeted oligonucleotide probes compared to the total biovolume determined with DAPI staining. Whereas the probe KOP1 detected only *S. pfennigii*-related organisms the probe MPOB1 gave a signal for *S. fumaroxidans*-related and *S. pfennigii*-related propionate oxidizers. The probe Eub338 detected almost all

microorganisms belonging to the domain *Bacteria* and the probe Arc915 all microorganisms belonging to the domain *Archaea*. According to twoway ANOVA, the biovolume fraction determined for a certain organism group with results determined for all other species was p=0.03 for KOP1, p=0.007 for MPOB1, p=0.08 for Eub338, p=0.01 for Arc915; the same test—all measurements vs. Eub338 (as a control data set); *p<0.05; ***p<0.001; ****p<0.001

determined from the increase of the total protein during exponential growth. According to data from literature (Table 2), the growth rates obtained from different pure cultures of mesophilic syntrophic propionate degraders in co-culture with hydrogenotrophic methanogens were in the range from 0.066 to 0.19 day⁻¹. The growth rates of syntrophic propionate degraders in axenic cultures in the presence of sulfate were still much lower than those of the sulfate reducers when grown in pure culture in the presence of sulfate. This might mean that syntrophic acetogenic bacteria were possibly suppressed by sulfate reducers in sulfate-rich ecosystems.

Our propionate-degrading enrichment culture could oxidize propionate with a rate of $0.76 \text{ mmoll}^{-1} \text{day}^{-1}$ (Fig. 2a). This rate was higher than that for pure cultures (Table 2) but one order of magnitude lower than the rate of propionate



Fig. 7 Increasing aggregate dimensions versus decreasing distances between propionate-degrading and methane-producing bacteria during growth phases 0–4. **p<0.01, one-way ANOVA (distances between organisms in phases 2–4 vs. distances in phase 0); the one-way analysis of variance for the distance measurements within several growth phases gave the value of p=0.0009

oxidation in the biowaste CSTR full-scale mesophilic digester for treatment of a biowaste suspension from the same origin (8.10 mmoll⁻¹ day⁻¹; Gallert and Winter 2008). It was two orders of magnitude lower than the rate of propionate oxidation by mesophilic sludge of a lab-scale digester, also treating a biowaste suspension (83.2 mmoll⁻¹ day⁻¹; Gallert et al. 2003). Since conversion rates in general depend on biomass concentration, it is necessary to relate degradation to active biomass, volatile suspended solids (VSS), or protein. To allow comparison of our results with data from literature, we calculated specific degradation rates from the total protein content under the assumption that 50 % of the dry matter of bacteria would be protein. Thus, the specific propionate oxidation rate of 49 mmol (mg protein)⁻¹ day⁻¹ (Fig. 2b) of our enrichment culture corresponding to 24.8 mmol (mg VSS)⁻¹ day⁻¹. This propionate oxidation rate is much higher than the propionate oxidation rate of 1.1 mmol $(g \text{ VSS})^{-1} \text{day}^{-1}$ measured in the sludge from an anaerobic hybrid reactor (Lens et al. 1996) or of 9.14 mmol (mg VSS)⁻¹day⁻¹ measured in granular sludge (Schmidt and Ahring 1995).

The degradation of propionate was inhibited by more than 50 % if acetate accumulated to 20 mM (Fig. 3). Similar results were obtained by Mawson et al. (1991) who reported a 50-% inhibition of propionate oxidation in the presence of 33 mM acetate. This was not caused by a pH change since the pH of the medium was sufficiently buffered.

Hydrogen or formate could be electron carriers from producers (acetogenic bacteria) to consumers (methanogenic bacteria), and the efficiency of electron transfer depends on interspecies distances. Possibilities to reduce transfer distances in liquid suspensions include either the use of artificially concentrated, dense cultures, or autoaggregated flocs, granules, or pellets. Floc formation of our enrichment culture from a biowaste digester during syntrophic propionate degradation

 Table 1 Comparison of hydrogen formation rates, hydrogen consumption rates, and the hydrogen flux from the propionate oxidizers to the methanogens in the syntrophic propionate-degrading culture

Growth phase (culture age)	Hydrogen flux $J_{H_2}^{a}$ [nmol ml ⁻¹ min ⁻¹]	Hydrogen formation rate ^b [nmolml ⁻¹ min ⁻¹]	Hydrogen consumption rate ^c [nmolml ⁻¹ min ⁻¹]
1 (13 days)	1.10	0.746±0.070	$0.741 {\pm} 0.030$
2 (22 days)	4.48	$1.375 {\pm} 0.021$	$1.486 {\pm} 0.280$
3 (29 days)	5.06	$1.572 {\pm} 0.017$	$1.583 {\pm} 0.237$
4 (42 days)	10.30	$1.239 {\pm} 0.112$	$1.111 {\pm} 0.208$

 $^aJ_{H_2}$ was calculated according to Fick's diffusion law; cell numbers of MPOB1-related propionate degraders in phase 1 were $5.05 \times 10^7 \ ml^{-1}$, in phase 2 $5.83 \times 10^7 \ ml^{-1}$, in phase 3 $7.91 \times 10^7 \ ml^{-1}$, and in phase 4 $8.82 \times x10^7 \ ml^{-1}$; cell numbers were calculated from the total cell number and percentage of MPOB1-related propionate degraders in each growth phase from Fig. 5

^b The hydrogen formation rate was calculated from the propionate oxidation rate assuming that 3 mol of H_2 were produced per mole of propionate (data from Fig. 2a)

^c The hydrogen consumption rate was calculated from the methane formation rate assuming that 4 mol of H_2 were consumed per mol of methane (data from Fig. 2a)

was observed (Fig. 5c-j), and an increasing size of the flocs up to 54 µm was determined (Fig. 7). Cluster formation of fatty acid-oxidizing acetogenic bacteria and H2-oxidizing methanogenic bacteria seems to facilitate hydrogen transfer (Conrad et al. 1986). Microscopic observations of flocs in syntrophic methanogenic cultures revealed the close neighborhood of acetogens and methanogens, allowing conclusions on the kinetics of interspecies electron transfer within the flocs (Thiele et al. 1988). The close neighborhood of propionate degraders and of methanogens in our floc-forming enrichment culture can be seen in Fig. 5c-j. The advantage of aggregation in syntrophic populations that degraded volatile fatty acids via interspecies hydrogen transfer was also investigated in sludge granules. With propionate as substrate, a 50-% higher H₂ consumption, measured as specific methanogenic activity, and a 20-% higher propionate oxidation were reported for intact granular sludge as compared to bacterial suspensions after disintegration of the granules (Shi-yi and Jian 1992; Schmidt and Ahring 1995). In experiments with co-cultures of thermophilic Pelotomaculum thermopropionicum and Methanobacterium thermoautotrophicum, co-aggregation apparently depended on the type of substrate. With thermodynamically unfavorable propionate, a large fraction of cells coaggregated, whereas a relatively small fraction of cells coaggregated with ethanol or 1-propanol as substrates (Ishii et al. 2005). In suspended propionate-degrading syntrophic cultures, one option to decrease interspecies distances is to increase the proportion of hydrogenotrophic methanogens for enhanced mass transfer of metabolites (Amani et al. 2012).

The methanogens in our propionate-degrading enrichment culture were able to produce methane from H_2/CO_2 and at a much lower rate from formate (Fig. 4a). Vice versa, formate was generated by methanogens when H₂/CO₂ was introduced into the gas phase and thus surplus hydrogen was available (Fig. 4b). Some formate (maximally 0.09 mM) was also excreted by this culture in the presence of propionate when hydrogenothrophic methanogenesis was completely inhibited. Formate formation from H₂/CO₂ would only require an active formate dehydrogenase and might stem from the methanogens, the propionate degraders, or both. With fumarate as an electron acceptor, Dong and Stams (1995) reported a minimal concentration of 0.024 mM formate that was generated by the MPOBs from propionate. In our non-BESA-inhibited batch assay with propionate, no formate was detected. In the literature, a controversial discussion is documented about the requirement or participation of formate, hydrogen, or both for electron interspecies transfer. Some authors state that both are equally important for syntrophic bacteria (Batstone et al. 2006; Schink 1997) or that formate may be more significant (de Bok et al. 2004) with fluxes 100 times higher than those for hydrogen in a co-culture of MPOBs and Methanospirillum hungatei (Dong and Stams 1995). Formate as an electron shuttle component may not be significant in suspensions of disintegrated thermophilic granules (Schmidt and Ahring 1993) or for mesophilic granular sludge (Schmidt and Ahring 1995),

The spatial distribution of propionate degraders and methanogens in our syntrophic enrichment culture was visualized with fluorescence in situ hybridization using specific 16S rRNA-targeted oligonucleotides labeled with FAM (propionate degraders) or Cy3 (methanogenic bacteria). The species of both trophic groups of bacteria in the flocs were almost randomly distributed with the formation of only very little micro-colonies. In order to quantify population changes during the different growth phases, the daime software for digital image analysis in microbial ecology was implemented (Daims et al. 2006) and biovolume fractions were determined. The total biovolume was obtained from DAPI-stained preparations, whereas the biovolume fractions of the propionatedegrading acetogens and the methane-forming archaea were determined with group-specific 16S rRNA-based oligonucleotide probes. Although DAPI has its limitations for such kind of evaluation (Nielsen et al. 2009), using it as a basis for the description of population fluctuations at each growth phase gave a sufficiently reliable "external" marker for changes of the ratio of propionate degraders and methanogens. The relative abundance of methanogens that hybridized with the specific 16S rRNA-targeted oligonucleotide probe Arc915 remained almost constant over the whole period of 42 days with four propionate feedings and accounted for approximately 52 % of the total biovolume in relation to

Table 2 Comparison of growth rates of pure cultures of propionate-degrading microorganisms

Propionate degrader	Syntrophic partner/co-substrate	Growth rate, $\mu \text{ [day}^{-1}\text{]}$	Reference
Pure cultures of syntrophic propionate-	degrading bacteria		
Syntrophobacter fumaroxidans	<i>Methanospirillum hungatei</i> Fumarate (no propionate)	0.17 0.09	Harmsen et al. 1998
	Sulfate	0.024	Van Kuijk and Stams 1995; Harmsen et al. 1998
Syntrophobacter pfennigii	<i>Methanospirillum hungatei</i> Sulfate	$0.066^{\rm a}$ $0.069^{\rm a}$	Wallrabenstein et al. 1995
Syntrophobacter sulphatireducens	Methanospirillum hungatei	0.12	Chen et al. 2005
Syntrophobacter wolinii	Methanospirillum hungatei and Desulfovibrio Sulfate and Desulfovibrio	$0.10^{\rm a}$ $0.19^{\rm a}$	Boone and Bryant 1980
	Sulfate	0.062	Wallrabenstein et al. 1994
Smithella propionica	Methanospirillum hungatei	n.d. ^b	Liu et al. 1999
Pelotomaculum schinkii	Methanospirillum hungatei; Methanobacterium formicium	0.099 ^a	de Bok et al. 2005
Pelotomaculum thermopropionicum	Methanothermobacter thermoautotrophicus	0.19	Imachi et al. 2000; Imachi et al. 2002
	Pyruvate (no propionate)	1.65	
Desulfotomaculum thermocisternum	Hydrogenotrophic methane bacterium; sulfate	n.d. ^b	Nilsen et al. 1996
Desulfotomaculum thermobenzoicum subsp. thermosyntrophicum	Methanobacterium thermoautotrophicum Pyruvate (no propionate)	0.12 0.33	Plugge et al. 2002
	Sulfate	0.099	
Pure cultures of selected sulfate-reduci	ng propionate-degrading bacteria		
Desulfobulbus propionicus	Sulfate	0.89	Stams et al. 1984
	Sulfate	1.66 ^a	Widdel and Pfennig 1982
Desulfobulbus elongatus	Sulfate	1.39	Samain et al. 1984
Strain PW	<i>Methanobacterium formicium</i> Sulfate	0.14 ^a 0.23 ^a	Wu et al. 1992

^a Conversion of t_d into μ using $\mu = \ln 2/t_d$

^b Not determined

unspecific DAPI staining (Fig. 6). Comparable results were reported for granular sludge fed with propionate, where the archaeal population remained constant at 55 % over 12 weeks (Harmsen et al. 1996). The biovolume fraction of propionate degraders in our enrichment culture, which hybridized with the 16S rRNA-targeted oligonucleotide probe KOP1 (S. pfennigii-related organisms) and MPOB1 (S. fumaroxidans-related organisms, including also S. pfennigii-related organisms), increased with time during the different growth phases. The biovolume fraction of MPOB-like propionate degraders accounted for 50 % after 42 days and was almost identical with the percentage value for the biovolume fraction that was obtained after hybridization with the universal EUB-probe (Fig. 6). Resilience and resistance of both methanogens and syntrophic propionate degraders must guarantee a stable metabolic function over time for survival despite disturbances.

Floc or aggregate formation was a possibility to decrease the interbacterial distance, which was necessary for the fast transfer of metabolites. In the literature, interspecies distances of 12.4 μ m in co-cultures of *S. fumaroxidans* and M. hungatei (Stams and Dong 1995), of 4 µm in mesophilic granular sludge between hydrogen-producing and hydrogen-consuming bacteria (Stams et al. 1989), of 2-3 µm in intact mesophilic sludge granules (Schmidt and Ahring 1995), or of 2.3 µm in aggregates of thermophilic P. thermopropionicum and M. thermoautotrophicum (Ishii et al. 2005) were reported. In our experiments, the interspecies distances decreased from 5.3 µm in the cell suspension directly after inoculation to 1.56 µm in phase 1, where small flocs were dominant, to 0.29 µm in phase 4 with mainly large flocs (Fig. 7). According to Fick's diffusion law (Eq. 4), an inverse relationship between syntrophic distance and hydrogen flux can be seen. A low spatial distance with a high diffusion rate would allow the maintenance of a low hydrogen partial pressure and enhance syntrophic propionate degradation. Calculation of the hydrogen concentrations according to Eq. 5 gave minimum H₂ concentrations of 0.32 Pa for the methanogens and maximum H₂ concentrations of 9.34 Pa for the propionate oxidizers delivering a maximum possible $C_{PA} - C_{H_2}$ value. According to de Bok et al. (2004), the highest H₂ level formed

by S. fumaroxidans was 6.8 Pa and the lowest H₂ level reached by M. hungatei was 2 Pa. In similar experiments of Scholten and Conrad (2000) with a syntrophic co-culture of S. fumaroxidans and M. hungatei, maximum H2 concentrations of 9 Pa and minimum H₂ concentrations of 0.8 Pa, respectively, were measured. These values were in accordance with those calculated for our co-culture. The maximum possible hydrogen flux $J_{\rm H_2}$ in the successive growth phases increased with decreasing interspecies distances. In phase 1, the maximum possible $J_{\rm H_2}$ was 1.1 nmolH₂ml⁻¹min⁻¹, which increased to 10.3 nmolH₂ $ml^{-1}min^{-1}$ in phase 4 (Table 1). A H₂ flux of 1.1 $molH_2ml^{-1}$ min⁻¹ in co-cultures of the propionate-degrading strain MPOB and M. hungatei was reported (Stams and Dong 1995). A H₂ flux of 5.38 nmolH₂ml⁻¹min⁻¹ in co-cultures of *P. thermo*propionicum and M. thermoautotrophicus can be calculated from the data of Ishii et al. (2005).

Simulation experiments for syntrophic degradation with hydrogen transfer resulted in the co-location of hydrogen producers and hydrogen consumers caused by the high uptake rates of hydrogen consumers rather than the sensitivity of hydrogen producers to hydrogen inhibition (Batstone et al. 2006). Results with our enrichment culture indicated that methanogens and syntrophic propionate degraders "find each other" after inoculation and form aggregates. By lowering the interspecies distance, the hydrogen flux from the hydrogen producer to the hydrogen consumer apparently did not limit the overall syntrophic interaction.

Acknowledgments This work has been conducted within the project on "Propionic acid metabolism during biowaste treatment using different digester regimes" (GA 546/4-2) sponsored by Deutsche Forschungsgemeinschaft, Bonn, Germany. Financial support came also from Karlsruhe Institute of Technology (KIT) within the STUB excellence initiative project "Anaerobic propionic acid degradation in syntrophic consortiamicrobial community distribution", conducted together with Water Chemistry at Engler Bunte Institut of KIT. Special thanks go to Heiko Schwegmann for help with the ApoTom.

References

- Amani T, Nosrati M, Mousavi SM (2012) Response surface methodology analysis of anaerobic syntrophic degradation of volatile fatty acids in an upflow anaerobic sludge bed reactor inoculated with enriched cultures. Biotechnol Bioproc Eng 17:133–144
- Amann RI, Binder BJ, Olson RJ, Chisholm SW, Devereux R, Stahl DA (1990) Combination of 16S rRNA-targeted oligonucleotide probes with flow cytometry for analyzing mixed microbial population. Appl Environ Microbiol 56:1919–1925
- Batstone DJ, Picioreanu C, van Loosdrecht MCM (2006) Multidimensional modeling to investigate interspecies hydrogen transfer in anaerobic biofilms. Wat Res 40:3099–3108
- Bleicher K, Winter J (1994) Formate production and utilization by methanogens and by sewage sludge consortia—interference with

the concept of interspecies formate transfer. Appl Microbiol Biotechnol 40:910–915

- Boone DR, Bryant MP (1980) Propionate-degrading bacterium, Syntrophobacter wolinii sp. nov. gen. nov., from methanogenic ecosystems. Appl Environ Microbiol 40:626–632
- Bryant MP, Wolin EA, Wolin MJ, Wolfe RS (1967) "Methanobacillus omelanskii", a symbiotic association of two species of bacteria. Arch Microbiol 59:20–31
- Chen S, Liu X, Dong X (2005) Syntrophobacter sulfatireducens sp. nov., a novel syntrophic, propionate-oxidizing bacterium isolated from UASB reactors. Int J Syst Evol Microbiol 55:1319–1324
- Conrad R, Schink B, Phelps TJ (1986) Thermodynamics of H₂-consuming and H₂-producing metabolic reactions in diverse methanogenic environments under in situ conditions. FEMS Microbiol Ecol 38:353–360
- Daims H, Lücker S, Wagner M (2006) daime, a novel image analysis program for microbial ecology and biofilm research. Environ Microbiol 8:200–213
- De Bok FAM, Stams AJM, Dijkema C, Boone DR (2001) Pathway of propionate oxidation by syntrophic culture of *Smithella propionica* and *Methanospirillum hungatei*. Appl Environ Microbiol 67:1800–1804
- De Bok FAM, Plugge CM, Stams AJM (2004) Interspecies electron transfer in methanogenic propionate degrading consortia. Water Res 38:1368–1375
- de Bok FAM, Harmsen HJM, Plugge CM, de Vries MC, Akkermans DL, de Vos WM, Stams AJM (2005) The first true obligately syntrophic propionate-oxidizing bacterium, *Pelotomaculum* schinkii sp. nov., co-cultured with *Methanospirillum hungatei*, and emended description of the genus *Pelotomaculum*. Int J Syst Evol Microbiol 55:1697–1703
- Dong X, Stams AJM (1995) Evidence for H₂ and formate formation during syntrophic butyrate and propionate degradation. Anaerobe 1:35–39
- Dwyer DF, Weeg-Aerssens E, Shelton DR, Tiedje JM (1988) Bioenergetic conditions of butyrate metabolism by a syntrophic, anaerobic bacterium in coculture with hydrogen-oxidizing methanogenic and sulfidogenic bacteria. Appl Environ Microbiol 54:1354–4359
- Felchner-Zwirello M, Winter J, Gallert C (2012) Mass spectrometric identification of ¹³C- labeled metabolites during anaerobic propionic acid oxidation. Chem Biodiv 9:376–384
- Gallert C, Winter J (2005) Bacterial metabolism in wastewater treatment systems. In: Jördening HJ, Winter J (eds) Environmental biotechnology—concept and applications. Wiley, Weinheim, pp 1–48
- Gallert C, Winter J (2008) Propionic acid accumulation and degradation during restart of a full-scale anaerobic biowaste digester. Bioresour Technol 99:170–178
- Gallert C, Henning A, Winter J (2003) Scale-up of anaerobic digestion of the biowaste fraction from domestic wastes. Wat Res 37:1433– 1441
- Gerardi MH (2003) The microbiology of anaerobic digesters. Wiley, New Jersey
- Gujer W, Zehnder AJB (1983) Conversion processes in anaerobic digestion. Wat Sci Technol 15:127–167
- Harmsen HJM, Kengen HMP, Akkermans ADL, Stams AJM (1995) Phylogenetic analysis of two syntrophic propionate-oxidizing bacteria in enrichment cultures. Syst Appl Microbiol 18:67–73
- Harmsen HJM, Akkermans ADL, Stams AJM, De Vos WM (1996) Population dynamics of propionate-oxidizing bacteria under methanogenic and sulfidogenic conditions in anaerobic granular sludge. Appl Environ Microbiol 62:2163–2168
- Harmsen HJM, Van Kuijk BLM, Plugge CM, Akkermans ADL, De Vos WM, Stams AJM (1998) Syntrophobacter fumaroxidans sp. nov., a syntrophic propionate-degrading sulfate-reducing bacterium. Int J Syst Bacteriol 48:1383–1387

- Imachi H, Sekiguchi Y, Kamagata Y, Ohashi A, Harada H (2000) Cultivation and in situ detection of a thermophilic bacterium capable of oxidizing propionate in syntrophic association with hydrogenotrophic methanogens in a thermophilic methanogenic granular sludge. Appl Environ Microbiol 66:3608–3615
- Imachi H, Sekiguchi Y, Kagamata Y, Hanada S, Ohashi A, Harada H (2002) *Pelotomaculum thermopropionicum* gen. nov., sp. nov., an anaerobic, thermophilic, syntrophic propionate- oxidizing bacterium. Int J Syst Evol Microbiol 52:1729–1735
- Inanc B, Matsui S, Ide S (1999) Propionic acid accumulation in anaerobic digestion of carbohydrates: an investigation on the role of hydrogen gas. Wat Sci Tech 40:93–100
- Ishii S, Kosaka T, Hori K, Hotta Y, Watanabe K (2005) Coaggregation facilitates interspecies hydrogen transfer between *Pelotomaculum* thermopropionicum and Methanothermobacter thermoatotrophicus. Appl Environ Microbiol 71:7838–7845
- Lens PNL, O'Flaherty V, Dijkema C, Colleran E, Stams AJM (1996) Propionate degradation by mesophilic anaerobic sludge: degradation pathways and effects of other volatile fatty acids. J Ferment Bioeng 82:387–391
- Liu Y, Balkwill DL, Aldrich HC, Drake GR, Boone RD (1999) Characterization of the anaerobic propionate degrading syntrophs *Smithella propionica* gen. nov., sp. nov. and *Syntrophobacter wolinii*. Int J Syst Bacteriol 49:545–556
- Lowry OH, Rosenbrough NJ, Farr AL, Randall RJ (1951) Protein measurement with the folin phenol reagent. J Biol Chem 193:265–275
- Mawson AJ, Earle RL, Larsen VF (1991) Degradation of acetic and propionic acids in the methane fermentation. Water Res 12:1549– 1554
- McCarty PL, Smith DP (1986) Anaerobic wastewater treatment. Environ Sci Technol 20:1200–1206
- Müller N, Worm P, Schink B, Stams AJM, Plugge CM (2010) Syntrophic butyrate and propionate oxidation processes: from genomes to reaction mechanisms. Environ Microbiol Rep 2:489–499
- Nielsen H, Daims H, Lemmer H (2009) FISH handbook for biological wastewater treatment: Identification and quantification of microorganisms in activated sludge and biofilms by FISH. IWA, London
- Nilsen RK, Torsvik T, Lien T (1996) Desulfotomaculum thermocisternum sp. nov., a sulfate reducer isolated from a hot North Sea oil reservoir. Int J Syst Bacteriol 46:397–402
- Plugge CM, Dijkema C, Stams AJM (1993) Acetyl–CoA cleavage pathway in a syntrophic propionate oxidizing bacterium growing on fumarate in the absence of methanogens. Microbiol Lett 110:71–76
- Plugge CM, Balk M, Stams AJM (2002) Desulfotomaculum thermobenzoicum subsp. thermosyntrophicum subsp. nov., a thermophilic, syntrophic, propionate-oxidizing, spore-forming bacterium. Int J Syst Evol Microbiol 52:391–399
- Samain E, Dubourguier HC, Albagnac G (1984) Isolation and characterization of *Desulfobulbus elongatus* sp. nov. from a mesophilic industrial digester. Syst Appl Microbiol 5:391–401
- Schink B (1992) Syntrophism among prokaryotes. In: Balows A, Trüper HG, Dworkin M, Harder W, Schleifer K-H (eds) The prokaryotes, 3rd edn. Springer, New York, pp 276–299
- Schink B (1997) Energetics of syntrophic cooperation in methanogenic degradation. Microbiol Mol Biol Rev 61:262–280
- Schmidt JE, Ahring BK (1993) Effects of hydrogen and formate on the degradation of propionate and butyrate in thermophilic granules from an upflow anaerobic sludge blanket reactor. Appl Environ Microbiol 59:2546–2551
- Schmidt JE, Ahring BK (1995) Interspecies electron transfer during propionate and butyrate degradation in mesophilic, granular sludge. Appl Environ Microbiol 61:2765–2767

- Scholten JCM, Conrad R (2000) Energetics of syntrophic propionate oxidation in defined batch and chemostat cocultures. Appl Environ Microbiol 66:2934–2942
- Shin SG, Lee S, Lee C, Hwang K, Hwang S (2010) Qualitative and quantitative assessment of microbial community in batch anaerobic digestion of secondary sludge. Biores Technol 101:9461–9470
- Shi-yi L, Jian C (1992) The contribution of interspecies hydrogen transfer to the substrate removal in methanogenesis. Process Biochem 27:285–289
- Sieber JR, McInerney MJ, Gunsalus RP (2012) Genomic insights into syntrophy: the paradigm for anaerobic metabolic cooperation. Annu Rev Microbiol 66:429–452
- Stahl DA, Amann R (1991) Development and application of nucleic acid probes. In: Stackebrandt E, Goodfellow M (eds) Nucleic acid techniques in bacterial systematics. Wiley, Chichester, pp 205– 248
- Stams AJM (1994) Metabolic interactions between anaerobic bacteria in methanogenic environments. Antonie Van Leeuwenhoek 66:271–294
- Stams AJM, Dong X (1995) Role of formate and hydrogen in the degradation of propionate and butyrate by defined suspended cocultures of acetogenic and methanogenic bacteria. Antonie Van Leeuwenhoek 68:281–284
- Stams AJM, Kremer DR, Nicolay K, Weenk G, Hansen AT (1984) Pathway of propionate formation in *Desulfobulbus propionicus*. Arch Microbiol 139:167–173
- Stams AJM, Grothenius JTC, Zehnder AJB (1989) Structure–function relationship in granular sludge. In: Hattori T, Ishida Y, Maruyama R, Morita R, Uchida A (eds) Recent advances in microbial ecology. Japan Scientific Societies, Tokyo, pp 440–445
- Thauer RK, Jungermann K, Decker K (1977) Energy conservation in chemotrophic anaerobic bacteria. Bacteriol Rev 41:100–180
- Thiele JH, Zeikus JG (1988) Control of interspecies electron flow during anaerobic digestion: significance of formate transfer during syntrophic methanogenesis in flocs. Appl Environ Microbiol 54:20–29
- Thiele JH, Chartrain M, Zeikus JG (1988) Control of interspecies electron flow during anaerobic digestion: role of floc formation in syntrophic methanogenesis. Appl Environ Microbiol 54:10–19
- Van Kuijk BLM, Stams AJM (1995) Sulfate reduction by a syntrophic propionate-oxidizing bacterium. In: Proceedings of the International Meeting on Anaerobic Processes for Bioenergy and Environment, part 5, Copenhagen, 25–27 January 1995
- Wallrabenstein C, Hauschild E, Schink B (1994) Pure culture and cytological properties of *Syntrophobacter wolini*. FEMS Microbiol Lett 123:249–254
- Wallrabenstein C, Hauschild E, Schink B (1995) Syntrophobacter pfennigii sp. nov., new syntrophically propionate-oxidizing anaerobe growing in pure culture with propionate and sulfate. Arch Microbiol 164:346–352
- Wang L, Zhou Q, Li FT (2006) Avoiding propionic acid accumulation in the anaerobic process for biohydrogen production. Biomass Bioenergy 30:177–182
- Widdel F, Pfennig N (1982) Studies on dissimilatory sulfate-reducing bacteria that decompose fatty acids. II. Incomplete oxidation of propionate by *Desulfobulbus propionicus* gen. nov., sp. nov. Arch Microbiol 131:360–365
- Wolin MJ (1975) Interactions between H₂-producing and methaneproducing species. In: Schlegel HG, Gottschalk G, Pfennig N (eds) Symposium on microbial production and utilization of gases. Goltz, Göttingen, pp 141–150
- Wu WM, Jain MK, de Macario EC, Thiele JH, Zeikus JG (1992) Microbial composition and characterization of prevalent methanogens and acetogens isolated from syntrophic methanogenic granules. Appl Microbiol Biotechnol 38:282–290