

# Metabolic engineering of *Escherichia coli* for biotechnological production of high-value organic acids and alcohols

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**Abstract** Confronted with the gradual and inescapable exhaustion of the earth's fossil energy resources, the bio-based process to produce platform chemicals from renewable carbohydrates is attracting growing interest. *Escherichia coli* has been chosen as a workhouse for the production of many valuable chemicals due to its clear genetic background, convenient to be genetically modified and good growth properties with low nutrient requirements. Rational strain development of *E. coli* achieved by metabolic engineering strategies has provided new processes for efficiently biotechnological production of various high-value chemical building blocks. Compared to previous reviews, this review focuses on recent advances in metabolic engineering of the industrial model bacteria *E. coli* that lead to efficient recombinant biocatalysts for the production of high-value organic acids like succinic acid, lactic acid, 3-hydroxypropanoic acid and glucaric acid as well as alcohols like 1,3-propanediol, xylitol, mannitol, and glycerol with the discussion of the future research in this area. Besides, this review also discusses several platform chemicals, including fumaric acid, aspartic acid, glutamic acid, sorbitol, itaconic acid, and 2,5-furan dicarboxylic acid, which have not been produced by *E. coli* until now.

**Keywords** *Escherichia coli* · Metabolic pathway · Organic acid · Alcohol · Platform chemicals

## Introduction

Confronted with the gradual and inescapable exhaustion of the earth's fossil energy resources, industrial biotechnologies for the production of platform chemicals, which can be either directly used or further processed for the production of large-volume and high value-added products in the chemical industry, have recently gained tremendous interests and attentions (Hatti-Kaul et al. 2007).

Biosynthesis of chemicals and biomaterials from recombinant microorganisms has arisen as a competitive alternative to the traditional chemistry-based routes. The development of recombinant DNA technology promotes the advancement of metabolic engineering in which we can get genetically modified strains to improve the yields of end products (Vemuri and Aristidou 2005; Bianco et al. 2006).

Among the candidate biocatalysts, *Escherichia coli*, a gram-negative bacterium, has been widely used in various biotechnological processes. Compared with other micro-biocatalysts to produce organic acids or alcohols, *E. coli* has several advantages: (1) many feasible genetic tools have been developed in *E. coli* for over 30 years; (2) sensitive to many antibiotics which can be used as engineering tags to chase the genetically modified strains (Nelson and Cox 2000); (3) clear genomic, proteomic, and metabolic information which benefit the metabolic engineering efforts (Harrington et al. 2000); (4) a wealth of knowledge on its central carbon metabolism (Sauer and Eikmanns 2005) and physiology; (5) a specific advantage—it grows quickly in minimal media and maintains the ability to metabolize both 5 and 6 carbon sugars (Zaldivar et al. 2001); and (6) its high tolerance to organic acids. Due to these advantages, artificial metabolic pathways have been constructed in *E. coli* by recruiting and integrating recombinant enzymes from a

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variety of resources to produce compounds which do not have metabolic pathways in wild microorganisms.

In the previous reviews, only several high-value platform chemicals were summarized with different microcatalysts. These reviews focused on the development of metabolic engineering for one or several compounds production with many detailed description of products (Chotani et al. 2000; Wee et al. 2006; Wendisch et al. 2006; Park and Lee 2008; Shanmugam and Ingram 2008; Jiang et al. 2009; Okabe et al. 2009; Saxena et al. 2009; Song and Vieille 2009). However, scientists engineering *E. coli* strains were not particularly concerned with finding the relationship in all high-value platform chemical metabolic pathways including native and constructed.

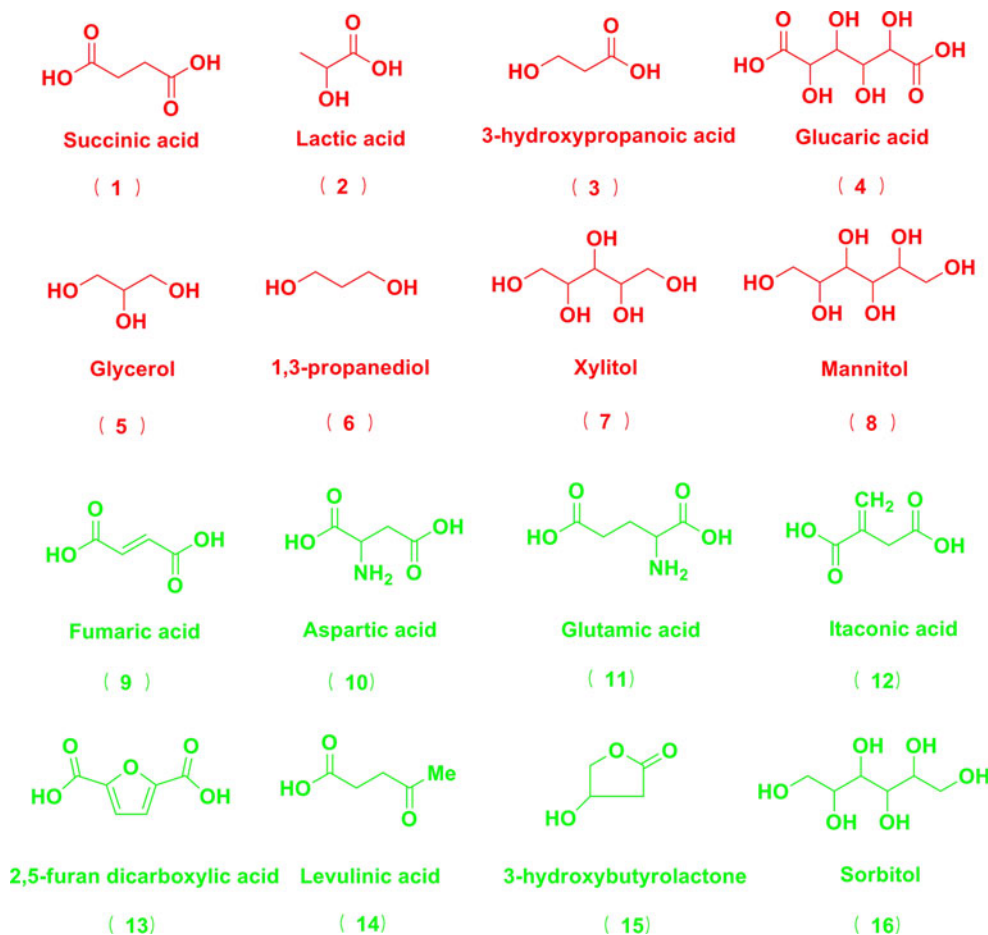
In this review, we summarize the application of *E. coli* strains for efficient production of a group of high-value organic acids or alcohols such as succinic acid, 3-hydroxypropanoic acid, glucaric acid, xylitol, and glycerol, which were identified as “top value-added chemicals from biomass” in a report of the Pacific Northwest National Laboratory and the National Renewable Energy Laboratory (Werpy and Petersen 2004). We also review the production of other important platform organic acid or alcohols like lactic acid, 1,3-propanediol, and mannitol. Meanwhile, the

most recent commercial information of all these compounds is given. In addition, the review proposes the possibility that several platform chemicals, including fumaric acid, aspartic acid, glutamic acid, sorbitol, and itaconic acid, are produced by *E. coli* in the near future. The pathway interrelation of all high-value chemicals is given in Fig. 2 except three chemicals: 13, 14, and 15 [see Fig. 1, chemicals (13), (14), and (15)].

### Succinic acid

Succinic acid [see Fig. 1, chemical (1)] is a platform chemical which can be converted to 1,4-butanediol and related products, tetrahydrofuran,  $\gamma$ -butyrolactone, *n*-methyl pyrrolidinone, and 2-pyrrolidinone, or other chemicals that are used to make a wide assortment of products. Succinic acid has the potential to replace important chemical intermediates (Kleff 2007). These chemicals are requested by American markets total almost 1 billion pounds or more than \$1.3B every year from the information of Pacific Northwest National Laboratory. The industrial potential for succinic acid fermentations was recognized as early as 1980 (Zeikus 1980).

**Fig. 1** The structures of high-value chemical building blocks. Chemicals (1)–(8) can be produced by *E. coli*; chemicals (9)–(16) have the possibility to be produced by *E. coli*



*E. coli* can utilize multiple pathways to form succinic acid (der Werf et al. 1997). Under anaerobic conditions, wild-type *E. coli* can produce a small amount of succinic acid (7.8%) by mixed-acid fermentation because of insufficient reducing power (Lin et al. 2005c; Wendisch et al. 2006). In order to improve succinic acid production, genetically modified metabolic pathways are first constructed to enhance the key enzyme activity. For example, succinic acid production from glucose by *E. coli* was significantly increased by overexpression of endogenous phosphoenolpyruvate carboxylase (*ppc*, PEPPc), while overexpression of endogenous phosphoenolpyruvate carboxykinase (*pck*, PEPCK) had no effect (Millard et al. 1996). However, overexpression of *Actinobacillus succinogenes* PEPCK increased the production of succinic acid as much as 6.5-fold (Kim et al. 2004). Using *E. coli* AFP111 as biocatalyst in which the *pyc* gene (pyruvate carboxylase) was overexpressed (AFP111/pTrc99A-*pyc*), a final succinic acid concentration and productivity reached 99.2 g/l and 1.3 g/l/h (Vemuri et al. 2002). Simultaneous overexpression of genes encoding PEPPc from *Sorghum vulgare* and pyruvate carboxylase (*pyc*) from *Lactococcus lactis* in *E. coli* increased the succinic acid yield, with a concomitant decrease of the lactate yield (Lin et al. 2005d). As fumarate reductase (*frdABCD*) is also a key enzyme in this metabolic pathway, two recombinant *E. coli* strains with amplified *frdABCD* activity have been constructed for the conversion of fumarate to succinic acid (Goldberg et al. 1983; Wang et al. 1998). In *E. coli* strains with recombinant *frdABCD*, when the concentration of glucose was 23.7 g/l, conversion rate of fumarate to succinic acid could reach 93%. However, malate could be accumulated when glucose was absent or cell density in the cultures was quite low. To improve the yield of succinic acid, further studies have been done, e.g., overexpressing of cyanobacterium *Anabaena* sp. 7120 *ecaA* (encoding carbonic anhydrase) in *E. coli*, to enhance the supply of  $\text{HCO}_3^-$  for the improvement of succinic acid production (Wang et al. 2009).

The other metabolic engineering strategy to increase the production of succinic acid was to knock out or inhibit the enzyme in succinic acid competition pathways. Chatterjee et al. obtained AFP111 by spontaneous chromosomal mutation of the *ptsG* (glucose phosphotransferase) gene in strain NZN111, which is unable to ferment glucose due to the inactivation of the genes encoding pyruvate: formate lyase and lactate dehydrogenase. In a batch reactor, this strain AFP111 could produce 36 g/l of succinic acid (Chatterjee et al. 2001). Sánchez et al. constructed *E. coli* strain, SBS550MG, that inhibited the central metabolic pathway by deactivating *adhE* (encoding aldehyde dehydrogenase), *ldhA* (encoding lactate dehydrogenase), and *ack-pta* (encoding acetate kinase-phosphate acetyltransferase) from the central metabolic pathway and activating the glyoxylate pathway through the inactivation of *iclR* (encoding isocitrate

lyase). In SBS550MG, the yield of succinic acid from glucose is about 1.6 mol/mol with an average anaerobic productivity rate of 10 mM/h (Sánchez et al. 2005). Because anaerobic fermentation of succinic acid has a number of disadvantages, Lin et al. constructed a mutant *E. coli* strain by deletion of *sdh* (encoding succinic acid dehydrogenase), *poxB* (encoding pyruvate oxidase), *pta-ack*, *iclR*, and *ptsG* genes, and overexpression of *ppc* gene. Fed-batch fermentation of Lin's strain resulted in a concentration of up to 58.3 g/l of succinic acid with an overall yield of 0.85 mol/mol glucose (Lin et al. 2005a, b). Considering the cost of construction, materials, purification, and waste disposal, novel strains of *E. coli* C were constructed by deletion of five genes, *ldhA*, *adhE*, *ackA*, *focA* (formate transporter), and *pflB* (encoding pyruvate-formate lyase), to produce high titers of succinic acid (600–700 mM) in simple batch fermentations (10% sugar) using mineral salts medium without any complex nutrients (Jantama et al. 2007). In addition, the effects of different carbon sources were investigated on the production of succinic acid in *E. coli* strain AFP184, which lacks functional genes coding for pyruvate formate lyase, fermentative lactate dehydrogenase, and the glucose phosphotransferase system. The result showed that carbon sources could significantly affect the yield of succinic acid with only small amounts of byproducts formed (Andersson et al. 2007).

Engineering *E. coli* strains were tested as biocatalysts for the scale-up production of succinic acid even though their strain information was limited. In 2008, DSM and France's Roquette Frères developed biotechnical methods for the production of kilo-scale succinic acid, using *E. coli* as catalyst and glucose as feedstock. Meanwhile, Myriant from the USA tested the scale-up production of succinic acid in the 20,000-l bioreactor, also using *E. coli* as catalyst but unrefined sugar as feedstock (<http://www.rsc.org/chemistryworld/News/2010/January/21011003.asp>).

The biological production processes need to be economically feasible including a yield around 0.88 g/g, a rate between 1.8 and 2.5 g/l/h and a titer around 80 g/l (Beauprez 2010). To date, none of the developed microbial strains which have been reported has reached all of these standards. However, developments in the metabolic engineering methods mentioned above showed great promise for further improvements in the near future. Next efforts should be done to optimize the current metabolic engineering towards succinic acid rather than to set up new metabolic routes.

## Hydroxy-propanoic acids

Lactic acid [see Fig. 1, chemical (2)] and 3-hydroxypropanoic acid [3-HP, see Fig. 1, chemical (3)] are industrially relevant microbial products. They receive significant attention due to

their application as platform chemicals and building blocks for “bio-based” polymers. Lactic acid, as the monomer for biodegradable polymer, will replace various petrochemical industry-based polymers in applications ranging from packaging to fibers (Tullo 2000) and be used in some new fields such as cosmetics, chemical industry, food, and pharmaceuticals (Wee et al. 2006). 3-HP has similar applications but without the side-chain methyl groups, which influence the strength of the materials (Zhu et al. 2010). Besides, it is a precursor for the synthesis of many high-volume commercial intermediates such as 1,3-propanediol (1,3-PDO) and acrylates (Brown 2003; Werpy and Petersen 2004).

### Production of lactic acid

The microorganisms selected for the biotechnological production of lactic acid include two groups: bacteria and fungi (Wee et al. 2006). *Lactobacilli* are commonly used for industrial production of lactic acid, but the fermentations rarely yield an optically pure product, while in *E. coli*, lactic acid production is one of the most successful large-scale fermentations to produce organic acids. The productivity for lactic acid differs upon various *E. coli* strains. Wild-type *E. coli* produces a mixture of organic acids including lactate. Engineering constructed *E. coli* for the production of lactic acid has several advantages compared to many other microorganisms available for lactic acid production. For example, *E. coli* can be metabolically engineered to produce optically pure lactic acid with less other fermentation byproducts, and Belgian scientists have reported that lactic acid has little effect on the proliferation of *E. coli* (<http://www.foodnavigator.com>).

Optical lactic acid production was greatly developed in the recent years. Chang et al. demonstrated that an *E. coli* RR1 *pta* (encoding phosphotransacetylase) mutant could be used as the host for the production of optically pure D- or L-lactic acid. They introduced L-lactate dehydrogenase genes from *Lactobacillus casei* into a *pta ldhA* (encoding lactate dehydrogenase) mutant strain, which lacked phosphotransacetylase and D-lactate dehydrogenase. The results suggested that the central fermentation metabolism of *E. coli* could be reoriented to the production of D(–)- or L(+)-lactic acid (Chang et al. 1999). Similarly, Dien et al. constructed recombinant *E. coli* for L(+)-lactic acid production from hexose and pentose sugars. They metabolically engineered *E. coli* including a *ptsG*<sup>–</sup> but glucose<sup>+</sup> mutant that carried mutations in the genes encoding PFL (pyruvate-formate lyase) and D-LDH (lactate dehydrogenase) and expressed a gene encoding L-LDH for the construction of carbon catabolite repression mutants (Dien et al. 2001, 2002). Zhou et al. constructed derivatives of *E. coli* W3110 as new biocatalysts for the production of D-lactic acid. D-Lactic acid production by these new strains (SZ40, SZ58, and

SZ63) approached the theoretical maximum yield of two molecules per glucose molecule. Competing pathways were eliminated by chromosomal inactivation of genes encoding fumarate reductase (*frdABCD*), alcohol/aldehyde dehydrogenase (*adhE*), and pyruvate formate lyase (*pflB*). The cell yield and lactate productivity were increased by a further mutation in the acetate kinase gene (*ackA*) (Zhou et al. 2003). Although capable of efficient fermentation of 5% (w/v) glucose or sucrose, higher sugar concentrations were incompletely metabolized by this biocatalyst and continuous antibiotic selection was required for plasmid maintenance. Hence, Zhou et al. constructed KO11-based biocatalysts (strain SZ132, produced by metabolic evolution from SZ110) that fermented glucose and sucrose to produce over 1 mol D(–)-lactate per liter of fermentation broth (10% w/v glucose or sucrose) (Zhou et al. 2005). Although SZ132 rapidly fermented 10% (w/v) sugars to completion, rich medium was required and performance was poor in mineral salts medium. Therefore, further improvements were done on SZ132. A non-recombinant mutant of *E. coli* B, strain SZ194 was developed from SZ132 that produced over 1 M D-lactate from glucose (or sucrose) in 72 h using mineral salts medium supplemented with 1 mM betaine in simple anaerobic fermentations. Rates and yields were highest at pH 7.5 with only trace amounts of co-products (Zhou et al. 2006). What is more, altering NADH utilization pathways had effects on the distribution of metabolic products. D-Lactate was the primary product in the *ndh nuo* (NADH-dehydrogenase-encoding genes) *adhE* inactivation strain (Yun et al. 2005).

Expanding the substrate range was also concerned in engineering *E. coli* strain improvements. A cluster of sucrose genes (three adjacent chromosomal operons, *cscR'*, *cscA*, and *cscKB*, encoding a repressor protein, invertase, fructokinase, and anion symport) were cloned and characterized from *E. coli* KO11. The resulting plasmid was functionally expressed in strain SZ63, which produced over 500 mM D(–)-lactate from sucrose (Shukla et al. 2004). By combining two substrate-selective strains of *E. coli*, knockout in *pflB* (encoding pyruvate-formate lyase), the xylose-selective strain with deletions of the *glk* (encoding glucokinase), *ptsG*, and *manZ* (encoding mannose PTS permease) genes and the glucose-selective strain with a *xyIA* (encoding xylose isomerase) deletion in a single process, xylose and glucose in a mixed sugar solution were simultaneously converted to lactate (Eiteman et al. 2008, 2009).

All engineered *E. coli* strains mentioned above have various problems for industrial applications, thus we need to optimize the lactic acid biocatalysts with increased fermentation rates, product titer, and yields to reduce costs. In a recent patent, *E. coli* was constructed by deleting the genes that encode competing pathways followed by a growth-based selection for mutants with improved performance for the

production of lactate, which could ferment 10% glucose or sucrose to produce over 1 mol D(–)-lactate of fermentation broth, with yields based on metabolized sugar ranging from about 88% to about 95%. Over 100 g/l in 48 h of chirally pure L(+)- and D(–)-lactate (>99.9% chiral purity) can be readily produced by recombinant *E. coli* B in mineral salts medium supplemented with 1 mM betaine (Shengde et al. 2010), showing great potential for scale-up production of lactic acid.

#### Production of 3-hydroxypropanoic acid

3-Hydroxypropionic acid (3-HP) production has the potential to be the next chemical intermediate produced by fermentation following lactic acid and succinic acid. Recently, Jiang et al. reviewed all 12 known synthetic pathways to produce 3-HP from sugars, glycerol, or carbon dioxide (Jiang et al. 2009). In this review, seven possible biosynthetic routes were summarized for the formation of 3-HP from glucose. Considering the thermodynamical advantage and the balance of reducing power and ATP, the pathway [see Fig. 2, enzymes used—*pgt* (encoding pyruvate-glutamate transaminase), *aam* (encoding alanine 2,3-aminomutase), *aoat* (encoding  $\beta$ -alanine-2-oxoglutarate aminotransferase), and *hid* (encoding 3-hydroxyisobutyrate dehydrogenase)] from pyruvate to 3-HP is most attractive. The point of this pathway is to increase the activity of the enzyme *aoat*. *E. coli* cannot metabolize glycerol into 3-HP because it lacks glycerol dehydratase (*dhaB*) and the native expression of aldehyde dehydrogenase (*aldH*) is also very weak (Jo et al. 2008). In 2001, formation of low concentrations of 3-HP (0.2 g/l) from glycerol, via 3-hydroxypropanal, was demonstrated with genetically modified *E. coli* expressing a *dhaB* from *Klebsiella pneumoniae* and a non-specific *aldH* from *Saccharomyces cerevisiae* (Suthers and Cameron 2001). A recombinant *E. coli* strain was developed by cloning *dhaB* and *aldH* genes, and the production of 3-HP from glycerol was demonstrated (Raj et al. 2008). After optimizing the physicochemical parameters, the recombinant *E. coli* strain, expressing *dhaB* and *aldH* genes, can produce 3-HP at 31 g/l with a yield of 35% when glycerol is used as the sole carbon and energy source (Mohan Raj et al. 2009). However, the productivity is far behind its commercial application. Hence, further work will remain to be done. Excitedly, Cargill and Novozymes predicted that 3-HP would be produced and sold within 5 years (<http://www.cargill.com/news-center/news-releases/2008/NA3007665.jsp>).

#### Glucaric acid

Glucaric acid [see Fig. 1, chemical (4)] has been studied for therapeutic purposes including cholesterol reduction

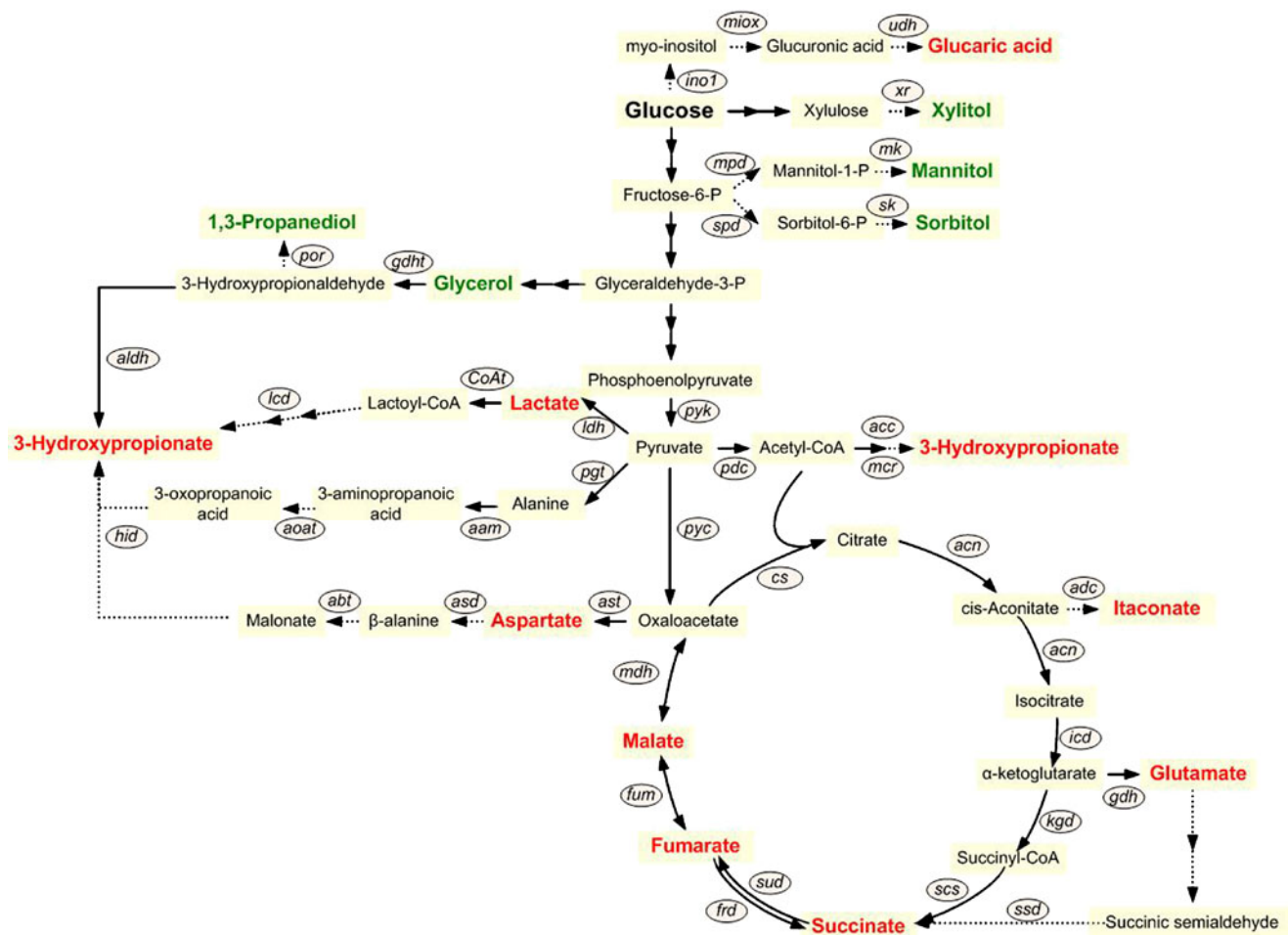
(Walaszek et al. 1996) and cancer chemotherapy (Singh and Gupta 2007). It also has potential applications as a building block for a number of polymers such as new nylons and hyperbranched polyesters.

Moon et al. constructed a synthetic pathway for the production of D-glucaric acid in *E. coli* by co-expression of the genes encoding *myo*-inositol-1-phosphate synthase (Ino1) from *S. cerevisiae*, *myo*-inositol oxygenase (MIOX) from mice, and uronate dehydrogenase (Udh) from *Pseudomonas syringae*. This is a simple and economical pathway for the production of D-glucaric acid compared with the long pathway of more than ten conversion steps from glucose in mammals. Glucaric acid productivity of over 1 g/l was observed (Moon et al. 2009). To further improve the productivity, they used modular, synthetic scaffolds for producing high-value compounds economically, efficiently, and cleanly, thus improving the maximum titers to 2.5 g/l (Moon et al. 2010). There is no industrial supply of glucaric acid by microbial fermentation so far. However, Rivertop Renewables is currently converting renewable glucose into glucaric acid. The company hopes to build a commercial plant and put out small production volumes by September 2011 (<http://cleantech.com/news/5613/startup-pursues-glucaric-acid-poten>).

#### Glycerol

Glycerol [see Fig. 1, chemical (5)], 1,2,3-propanetriol, is a commodity chemical used in cosmetics, liquid soaps, food, pharmaceuticals, lubricants, antifreeze solutions, and tobacco. The further chemical modification of glycerol is also carried out in industry, especially the synthesis of some esters, polyethers, and alkyd resins. Behr et al. reviewed the improved utilization of glycerol (Behr et al. 2008).

The most important microbial biocatalyst is *S. cerevisiae* but other yeast species (Taherzadeh et al. 2002). *E. coli* does not have a natural efficient pathway to produce glycerol. In early-stage research, a Genencorv International and Dupont team had demonstrated obvious glycerol production in *E. coli*. Expression of either glycerol-3-phosphate dehydrogenase (GPD1) or glycerol-3-phosphatase (GPP2) in *E. coli* resulted in a low level of glycerol production (Bulthuis et al. 2002). When GPD1 and GPP2 were co-expressed in *E. coli*, the yield of glycerol increased 10–20-fold. Because *E. coli* can utilize glycerol as carbon source through glycerol kinase (*glpK*) and glycerol dehydrogenase (*gldA*) pathways, Nair et al. removed these dissimilation pathways, which resulted in minimal glycerol consumption and increased carbon yield to glycerol with near theoretical yield and over 200 g/l of glycerol (Nair et al. 2005). Recently, in order to demonstrate, under appropriate “metabolic pressure”, the evolu-



**Fig. 2** The metabolic pathways for the production of the high-value chemical building blocks by using *E. coli*. The solid lines indicate *E. coli* native pathways while the dotted lines refer recombinant pathway by metabolic engineering strategies. Enzymes encoded by the genes shown are: *ino1*: myo-inositol-1-phosphate synthase; *miox*: myo-inositol oxygenase; *udh*: uronate dehydrogenase; *xr*: xylose reductases; *mpd*: mannitol-6-phosphate dehydrogenase; *mk*: mannitol kinase; *spd*: sorbitol-6-phosphate dehydrogenase; *sk*: sorbitol kinase; *gdht*: glycerol dehydrogenase; *por*: 1,3-propanediol oxidoreductase; *aldh*: aldehyde dehydrogenase; *pyk*: pyruvate kinase; *ldh*: lactate dehydrogenase; *CoAt*: Co-enzyme A transferase; *lcd*: lactyl-CoA dehydratase; *pgt*: pyruvateglutamate transaminase; *aam*: alanine 2,3-

aminomutase; *aoat*:  $\beta$ -alanine-2-oxoglutarate aminotransferase; *pdc*: pyruvate dehydrogenase complex; *acc*: acetyl-CoA carboxylase; *mcr*: malonyl-CoA reductase; *pyc*: pyruvate carboxylase; *cs*: citrate synthetase; *acn*: aconitase; *adc*: aconitate decarboxylase; *icd*: isocitrate dehydrogenase; *gdh*: glutamate dehydrogenase; *kgd*:  $\alpha$ -ketoglutarate dehydrogenase; *scs*: succinyl-CoA synthetase; *ssd*: succinic semialdehyde dehydrogenase; *sud*: succinic acid dehydrogenase; *frd*: fumarate reductase; *fum*: fumarase; *mdh*: malate dehydrogenase; *ast*: aspartate aminotransferase; *asd*: aspartate decarboxylase; *abt*: 4-aminobutyrate aminotransferase; *hid*: 3-hydroxyisobutyrate dehydrogenase

tionary adaptation of a heterogeneous pathway in a strain and validate this concept, the *S. cerevisiae* glycerol pathway (GPD1 and GPP2) was evolved *in vivo* in *E. coli*. Three key modifications were introduced in the central metabolism of *E. coli*. Without any optimization of the fermentation conditions, high yield (1 mol/mol), titer (130 g/l), and productivity (16 g/l/day) were obtained in glucose-limited fed-batch cultures (Meynial Salles et al. 2007).

Recently, the study on its microbial production is not very popular because of its rising inevitable formation as a by-product of biodiesel production. However, with growing interest of glycerol as a bulk platform chemical, microbial

production of glycerol still could be a concern based on the consideration of biomass from waste residual, and in that case, engineered *E. coli* due to its special features could be one alternative to *S. cerevisiae*.

### 1,3-Propanediol

1,3-Propanediol [1,3-PD; see Fig. 1, chemical (6)] is suitable for fiber and textile applications. It has the potential to replace the traditional polyethylene terephthalate and polybutylene terephthalate as well as being used in

solvents, adhesives, laminates, resins, detergents, and cosmetics as a bifunctional organic compound (Zeng and Biebl 2002).

1,3-Propanediol production needs two enzymes, glycerol dehydratase (encoded by the gene *dhaB1-3*) and 1,3-propanediol oxidoreductase (encoded by the gene *dhaT*) in natural organisms (Nakamura and Whited 2003). Experiments to overexpress these genes in *E. coli* have been successful, but the 1,3-PD concentrations obtained with these constructs could originally not be raised above 9 g/l (Daniel and Gottschalk 1992). By expressing a non-specific alcohol dehydratase instead of *dhaT* for the conversion of 3-hydroxypropionaldehyde, a final concentration of 129 g/l was obtained with transformed *E. coli* in a patent (Zeng and Biebl 2002).

Recently, Wang et al. (2007) co-expressed in *E. coli* three genes including *dhaB*, which encodes glycerol dehydratase; *dhaT*, which encodes 1,3-PD oxidoreductase; and *gdrAB*, which encodes glycerol dehydratase. In a fed-batch fermentation of glycerol and glucose, the recombinant *E. coli* consumed 14.3 g/l of glycerol and produced 8.6 g/l of 1,3-propanediol. In the substitution case of *yqhD* (encoding alcohol dehydrogenase from *E. coli*) for *dhaT*, the final 1,3-propanediol concentration of the recombinant *E. coli* could reach 13.2 g/l. To improve 1,3-PD yield, several efforts were done by expressing three genes (*dhaB1*, *dhaB2*, and *yqhD*) in a recombinant *E. coli* and constructing a novel 1,3-PD operon of these three genes randomly arrayed under the control of a constitutive, temperature-sensitive promoter in the vector pBV220 for heterologous expression in *E. coli*. Under a two-stage fermentation process, the overall 1,3-PD yield and productivity reached 104.4 g/l and 2.61 g/l/h (Tang et al. 2009), which show a good potential for the economic and effective production of 1,3-PD. For scale-up production, Dupont and Genencor have commercialized 1,3-PD by engineered *E. coli*, and 1,3-PD-based polyester which is estimated to be 1–2 billion pounds per year over the next 10 years can also be produced by Dupont and Genencor using glucose as the feedstock (Nakamura and Whited 2003).

## Xylitol

Xylitol [see Fig. 1, chemical (7)] is the first rare sugar that has global markets. Xylitol can be used as a raw material for a number of different bioconversions for other rare sugar production. It has beneficial health properties and is used as a nutritive sweetener and food additive (Granstrom et al. 2007). Its major use is for the prevention of dental caries as xylitol prevents the growth of microorganisms responsible for tooth decay. The xylitol market is increasing and is estimated to be \$340 million year<sup>-1</sup>

and priced at \$4–5 kg<sup>-1</sup> according to Prakasham et al. (2009).

*E. coli* has the ability to assimilate both hexose and pentose sugars besides the advantages mentioned above. In order to produce xylitol in an economical and eco-friendly manner, metabolically engineered *E. coli* has been studied as an alternative for industrial production of xylitol. Xylose reductases catalyze the initial reaction in the xylose utilization pathway, the NAD(P)H-dependent reduction of xylose to xylitol. Häcker et al. cloned and successfully expressed the xylose reductase gene from *Candida tenuis* in *E. coli* (Häcker et al. 1999). In 1999, the D-xylose reductase (XR) gene (*xylA*) of *Candida tropicalis* was also expressed in *E. coli*, which successfully converted D-xylose to xylitol. When D-xylose (50 g/l) and D-glucose (5 g/l) were added to IPTG-induced cells, 13.3 g/l of xylitol was produced during 20 h of cultivation (Suzuki et al. 1999).

Cirino et al. replace *E. coli*'s native cyclic AMP receptor protein (CRP) with a cyclic AMP-independent mutant (CRP\*) which facilitated xylose uptake and xylitol production from mixtures of glucose and xylose, with glucose serving as the growth substrate and source of reducing equivalents. Overexpression of NADPH-dependent CbXR (xylose reductases from *Candida boidinii*) produced the highest concentrations of xylitol in shake-flask cultures (Cirino et al. 2006). Subsequent deletion of the *xylB* gene (encoding xylulokinase) and expression of xylose reductase from *C. boidinii* resulted in a strain which produced xylitol from glucose–xylose mixtures. Khankal et al. engineered *E. coli* W3110 to produce xylitol from a mixture of glucose plus xylose by expressing xylose reductase and deleting xylulokinase (*DxylB*), combined with either plasmid-based expression of a xylose transporter (*XylE* or *XylFGH*) (Khankal et al. 2008). More researches were done by heterologous expression of D-xylulokinase from *Pichia stipitis* with high levels of xylitol production by engineered *E. coli* co-expressing xylose reductase during growth on xylose (Akinterinwa and Cirino 2009). In current research, Cheng et al. cloned the NAD-dependent D-xylulose-forming D-arabitol dehydrogenase gene (*aArDH*) from an acetic acid bacterium and heterogeneously expressed it in *E. coli* for the bio-conversion of D-arabitol to xylitol (Cheng et al. 2009).

Although other microorganisms such as yeasts were tested as xylitol producers, *E. coli* offers the increasing possibilities of economic production due to its special advantages. Furthermore, *E. coli* can reduce required energy comparing to the relative chemical methods. For the production of xylitol via biotechnological process at an economic industrial scale, focus should be maintained on a common platform of understanding of the hydrolysate material, hydrolysis procedure, microbial performance, and bioconversion environ-

ment, and downstream processing is one of the most essential aspects for development of integrated technological solution (Prakasham et al. 2009).

## Mannitol

Mannitol [see Fig. 1, chemical (8)] has a variety of applications in pharmaceutical products, food industry, and medicine (Song and Vieille 2009). *E. coli* was engineered to produce mannitol from fructose by constructing a recombinant oxidation/reduction cycle. The recombinant strain co-expressed *mdh* gene (encoding malate dehydrogenase) from *Leuconostoc pseudomesenteroides*, *fdh* gene (encoding formate dehydrogenase) from *Mycobacterium vaccae* N10, and *glf* gene (encoding the glucose facilitator protein) from *Zymomonas mobilis*. Under pH control mode (by the addition of formic acid), the concentration of mannitol was 362 mM within 8 h and the yield was 84 mol% (Kaup et al. 2004). Further research was done by additional expression of extracellular glucose isomerase in *E. coli*, leading to the formation of 800 mM mannitol from 1,000 mM glucose. Co-expression of the *xylA* gene of *E. coli* in this strain led to a mannitol concentration of 420 mM from 1,000 mM glucose (Kaup et al. 2005). Considering the restriction of long-term stability of recombinant *E. coli* which produced mannitol from fructose, *fupL* gene, encoding a putative mannitol permease, was cloned and expressed in *E. coli*. The productivity for the biotechnical production of mannitol was enhanced by 20% (Heuser et al. 2009). More researches still need to be done to optimize the *E. coli* strains for the commercial production of mannitol. At the same time, other biocatalysts like *L. intermedius* NRRL B-3693 was used to commercially produce mannitol by the American company zuChem. However, compared with *E. coli*, *L. intermedius*-based process has too many by-products in the final products (<http://www.foodnavigator-usa.com/Financial-Industry/ZuChem-gears-up-for-first-mannitol-sweetener>).

## Other top value-added platform chemicals

1. Fumaric acid [see Fig. 1, chemical (9)] has many potential industrial applications, e.g., precursors of unsaturated polyester resins and plasticizers, miscellaneous including lubricating oil, inks, and lacquers, carboxylating agent for styrenebutadiene rubber, personal care additives, and food and beverage additives (Roa Engel et al. 2008). Fungi have been used in fermentation processes for fumaric acid production.
2. Aspartic acid [see Fig. 1, chemical (10)] and glutamic acid [see Fig. 1, chemical (11)] are building blocks for active ingredients required in the pharmaceutical and

cosmetics industries. The biosynthesis of aspartic acid and glutamic acid in some bacteroids has been researched, such as *Rhizobium lupine* and *Rhizobium japonicum* (Lillich and Elkan 1971; Kretovich et al. 1981). The aspartate pathway has also been researched in plants. Single and often multiple genes have been cloned and expressed successfully in *E. coli* (Azevedo et al. 2006). The metabolic engineering of glutamic acid production by *Corynebacterium glutamicum* has been investigated in depth (Kimura 2002; Schultz et al. 2007).

3. Sorbitol [see Fig. 1, chemical (16)] has broad applications not only in the food industry but also in pharmaceutical production (Silveira and Jonas 2002). Two recombinant strains of *Lactobacillus plantarum* (Ladero et al. 2007) and *Lactobacillus casei* (Yebra and Perez-Martinez 2002; Nissen et al. 2005) were constructed to produce sorbitol.
4. Itaconic acid [IA; see Fig. 1, chemical (12)] has been used in synthetic resins, synthetic fibers, plastics, rubbers, surfactants, oil additives, and biomedical fields (Okabe et al. 2009). *Aspergillus terreus* is now the most frequently used commercial producer of IA, and the IA production yield from sugar is higher than 80 g/l (Willke and Vorlop 2001). However, the production costs are relatively high.
5. 2,5-Furan dicarboxylic acid [see Fig. 1, chemical (13)] can be used to produce numerous furan derivatives, succinic acid, esters, levulinic acid, furanoic polyamines, and polyethylene terephthalate analogs. Its biosynthesis has not been reported so far as well as levulinic acid [LA; see Fig. 1, chemical (14)] and 3-hydroxybutyrolactone [see Fig. 1, chemical (15)]. The researches still need to be further enhanced.

## Conclusions

Compared with conventional petrochemical production, bio-based platform chemicals, which can be either directly used or further processed for the production of large-volume and value-added products in the chemical industry, have gained more concern. *E. coli* strains could be engineered for overproduction of native and non-native small molecules with obvious advantages over other bacteria. Improved *E. coli* strains have been developed that are well adapted to fermentation processes and produce desired products with high yields. With recent successes of metabolic engineering of *E. coli*, several other platform chemicals [see Fig. 1, chemicals (3)–(8)] have the potential to be scaled up in the near future. However, the biotransformation route of some top value-added chemicals has not been researched such as chemicals (12) and (16) in



Fig. 1. New tools such as modulating expression of chromosomal genes or creating multiple chromosomal deletions will certainly improve the efficiency and the fine-tuning of metabolic engineering for *E. coli* to overcome its native limitation. Furthermore, all of genomic, proteomic, and metabolic modeling tools could be used in *E. coli* to enhance the production of the present organic acids/alcohols products, and on this base to further develop innovative organic acids/alcohols products through the creation and optimization of novel metabolic pathways.

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