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Steam explosion pretreatment used to remove hemicellulose to enhance the production of a eucalyptus organosolv dissolving pulp

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Abstract The steam explosion (SE) pretreatment associated with the organosolv process was investigated to produce dissolving pulp from eucalyptus. Prehydrolysis Kraft (PHK) pulping was also done to produce viscose and acetate grade pulps as reference. The organosolv pulps were delignified in two steps with sodium chlorite. Viscose and acetate grade PHK pulps were bleached by $OD_0(EH)D_1P$ and $OD_0(-EP)D_1PCCE$ sequences, respectively. Dilute acid-catalyzed (with acid addition) SE pretreatment dissolved more xylan than auto-catalyzed (no acid addition) SE pretreatment. Steam-exploded unbleached organosolv pulps showed lower residual lignin content and screened yield than unbleached organosolv pulps without SE pretreatment. Steam explosion pretreatment helped to decrease lignin content and damaged fiber length of unbleached organosolv pulps. The 1.0% H₂SO₄ organosolv pulp (organosolv dissolving pulp at bioconversion conditions) showed the highest reactivity. Even showing low viscosity for some applications, in general, the organosolv dissolving pulps produced in this study can be used for making lyocell fibers.

Introduction

Dissolving pulp is a high-grade cellulose pulp used to manufacture cellulosederived products that include regenerated fibers or films (viscose, lyocell), cellulose esters (e.g., acetates, nitrates) and cellulose ethers (e.g., carboxymethyl celluloses)

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(Sixta 2006). Dissolving pulps are mainly produced by the sulfite pulping and prehydrolysis Kraft (PHK) processes. Sulfite pulping can simultaneously remove lignin and hemicellulose. However, sulfite process produces pulps with weak strength properties, is sensitive to some wood species due to interfering substances such as resins, extractives and has an inefficient chemical recovery process. Kraft pulping process has well-established chemical recovery process, stronger fibers compared to sulfite pulps and high yields, but retains hemicellulose and preserves cellulose degree of polymerization (DP) which is not appropriate for dissolving pulps. Hence, additional chemical processing steps including prehydrolysis, cold caustic extraction (CCE) and hypochlorite (H) treatment must be implemented in the Kraft process in order to produce dissolving pulps.

Besides sulfite pulping, acidic organosolv process can simultaneously remove both the lignin and hemicellulose and has been used as a pretreatment step for the biomass-to-ethanol process (Pan et al. 2006). However, it may also be ideally suited to help make dissolving grade pulp. Moreover, there have only been a few studies investigating the potential of organosolv pulping for dissolving pulp production (Caraschi et al. 1996; Kirci and Akgul 2002; Sixta et al. 2004). Sixta et al. (2004) evaluated the potential of the Milox, Acetosolv and Formacell and acid Mg sulfite pulping processes from *Eucalyptus globulus* in the dissolving pulp production for conversion into viscose fibers. These researchers found that organosolv pulping processes proved to be more selective in terms of pentosan removal and cellulose degradation as compared to conventional sulfite pulping. Manfredi et al. (2012) evaluated the effects of ethanol pulping process of *Eucalyptus* sp. wood chips and the efficiency of phosphinic acid as catalyst in this process. They found that ethanol pulping process has shown potential for dissolving pulp industries. The phosphinic acid improved delignification and hydrolysis of cellulose chains.

Dissolving pulps are characterized as being pulps with high cellulose content and minimum amounts of non-cellulosic impurities (hemicelluloses, extractives and inorganics) (Sixta 2006). To try to reduce the extent of processing necessary to produce dissolving pulp, steam explosion pretreatment can be used for hemicellulose removal. Steam pretreatment involves treatment with steam under high pressure and temperature, followed by quick release of the pressure and explosion of the biomass (Laxman and Lachke 2009). The mechanical disruption of the pretreated material can be either by violent discharge into a collecting tank (explosion) or by mild blending after bleeding the steam pressure down to atmospheric (no explosion) (Ramos 2003). Steam explosion pretreatment requires low energy input, is fast, cheap and has negligible environmental impact (Laxman and Lachke 2009).

The aim of this work was to evaluate steam explosion pretreatment for hemicellulose removal prior to organosolv pulping to produce dissolving pulps.

Materials and methods

Working plan

The experimental plan showing all steps involved in the production of steam explosion (SE)—organosolv and prehydrolysis Kraft dissolving pulps—is presented in Fig. 1. Different routes were used to produce dissolving pulps. On the first route (1), the eucalyptus chips were pretreated with auto-catalyzed and dilute acid-catalyzed steam explosion (SE). Thereafter, the pretreated samples were cooked at 170 °C with ethanol/water (65%/35%) liquor, with 0, 0.5 and 1.0% of sulfuric acid as a catalyst and a liquor-to-biomass ratio of 16/1. The delignification was run in two stages with sodium chlorite. On the second route (2) dissolving pulps were produced without pretreatment of the biomass. The organosolv pulping was carried out under the same conditions as route 1, except for the liquor-to-biomass ratio which was 7/1. Sodium chlorite bleaching was also performed under the same conditions of route 1. Route 3 produced the dissolving pulp by Kraft pulping process using the prehydrolysis pretreatment. The prehydrolysis Kraft (PHK) pulps were produced as reference for the organosolv pulps. The prehydrolysis step was performed at 170 °C with retention time of 15 min for viscose grade and 30 min for

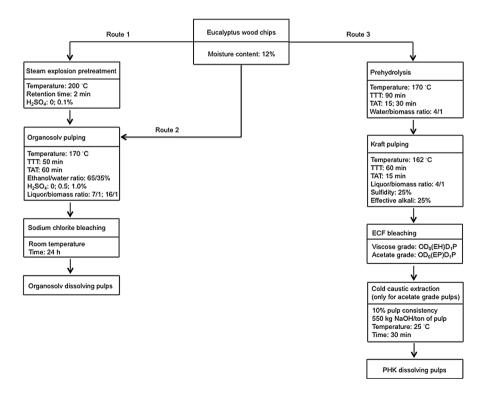


Fig. 1 Work plan showing all steps involved in the production of dissolving pulp by the SE—organosolv and PHK processes

acetate grade. Kraft pulping was conducted under the same conditions for viscose grade and acetate grade pulps, at a temperature of 162 °C, a retention time of 15 min and sulfidity and effective alkali of 25%. The bleaching sequences for viscose grade and acetate grade pulps were $OD_0(EH)D_1P$ and $OD_0(EP)D_1P$, respectively. After peroxide bleaching stage, a purification step (cold caustic extraction—CCE) was run for acetate grade pulp.

Preparation of the eucalypt chips for the steam explosion (SE) pretreatment

Industrial wood chips of a *Eucalyptus urophylla* x *Eucalyptus grandis* clonal hybrid, *Eucalyptus urograndis*, were screened through a $12.5 \times 12.5 \text{ mm}^2$ sieve, air-dried and collected as the raw material for SE pretreatment. The moisture content of the wood chips was about 12%. Samples of eucalypt chips were ground using a Wiley mill, and the fraction passing between 40 and 60 mesh was collected for chemical analysis. The chemical composition of eucalyptus chips is summarized in Table 1.

SE pretreatment

Five batches of 200 g (o.d.) eucalyptus chips were impregnated with water or dilute acid water (0.1% H_2SO_4 w/w) overnight at room temperature. The impregnated biomass samples were subsequently loaded into a 2-L Stake Tech II steam gun (Stake Tech II batch reactor, SunOpta (formerly Stake Technologies) of Norval, ON, Canada) and steam-exploded at 200 °C for 2 min (pressure 242 ψ) (Han et al. 2010). All batches were collected together to minimize sample variation. A single trial was performed at the same conditions without explosion to evaluate the effect of the explosion on the fiber length. The resulting slurry was collected, and the

Component	%, on wood wt.
Carbohydrate ^a	
Glucan	46.50 (0.45)
Mannan	0.60 (0.01)
Xylan	10.50 (0.00)
Arabinan	0.20 (0.01)
Galactan	0.90 (0.01)
Acid soluble lignin ^a	3.50 (0.11)
Acid insoluble lignin ^a	26.30 (0.20)
Extractives ^b (Toluene/ethanol; ethanol; hot water)	2.90 (0.02)
Ash ^b	0.20 (0.01)
Silica ^b	0.04 (0.01)

 Table 1 Chemical composition of eucalyptus chips

 $^a~\%$ of dry extractive-free wood, $^b~\%$ of dry received wood. Standard deviations from the mean are shown within parenthesis

water-soluble fraction was separated from the solid fraction with vacuum filtration. The water-soluble fraction was used for sugar analysis. The solid fraction yield was determined gravimetrically. The solid fraction was washed with water and used as the feedstock for all the organosolv pulping.

Prehydrolysis

A couple of PHK cooking trials were performed to simulate viscose and acetate grade pulps using eucalypt wood chips for comparison. About 1200 g wood chips (o.d.) and required amount of water were loaded into a PARR reactor model 4555-85 (19 L) to reach the ratio of water to wood chips (o.d.) of 4/1 (v/w). The prehydrolysis was performed according to Longue Junior and Colodette (2011). The conditions were: temperature 170 °C, 90 min to temperature; reaction time of 15 min (for viscose grade) and 30 min (for acetate grade) at temperature.

Organosolv and PHK pulping

The organosolv pulping was performed according to Pan et al. (2006) using a PARR reactor (2L). Pulping conditions used are summarized in Table 2. After cooking, vessels were cooled to room temperature in a water bath. Pulp and liquor were then separated with vacuum filtration. The pulp was washed three times (350 mL each) with 60 °C aqueous ethanol, which was at the same concentration of ethanol as the original pulping liquor. The pulp was then washed three times (350 mL each) with water at 60 °C, and the washes discarded. The washed pulp was homogenized in a standard British disintegrator for 10 min and passed through a laboratory flat screen with 0.15-mm slits to remove rejects.

PHK cooking trials using eucalyptus wood chips and commercial conditions were performed to produce viscose and acetate grade pulps as reference. The PHK cooking trials were carried out right after prehydrolysis in the same PARR reactor using the following conditions: liquor-to-biomass ratio 4/1 (v/w), temperature 162 °C, 60 min to temperature; reaction time of 15 min at temperature, sulfidity and effective alkali charges of 25%. After cooking, black liquor was collected and chips

Pretreatment	Untreated wood chips	Steam-exploded wood chips		
Dry weight (g)	200	100		
Temperature (°)	170	170		
Time to temperature (min)	50	50		
Time at temperature (min)	60	60		
Ethanol/water ratio (%)	65/35	65/35		
H ₂ SO ₄ , % (w/w)	0; 0.5; 1.0	0.5		
Liquor/biomass ratio	7/1	16/1		

 Table 2
 Organosolv pulping conditions for untreated and steam-exploded eucalyptus chips

were washed and disintegrated in a laboratorial "hydrapulper" of 25 L capacity. The pulps were screened in a "Voith" laboratorial screener equipped with 0.20-mm slits plate and centrifuged to a consistency of about 30%.

Bleaching

The organosoly pulps were delignified in two steps with sodium chlorite according to the procedure in the Pulp and Paper Technical Association of Canada (PAPTAC 1998) Useful method G10.U. Viscose and acetate grade pulps were bleached by $OD_0(EH)D_1P$ and $OD_0(EP)D_1PCCE$ sequences, respectively, where O = oxygen; D_0 and D_1 = chlorine dioxide; (EH) = alkaline extraction with hypochlorite; (EP) = alkaline extraction with hydrogen peroxide; P = hydrogen peroxide;CCE = cold caustic extraction. The general bleaching conditions applied to each bleaching stage are shown in Table 3. The O stage was carried out in a Mark V (Quantum Technologies Inc.) mixer/reactor. An additional CCE step was included after peroxide bleaching stage to maximize the non-cellulosic impurities removal. CCE treatment was at 30 °C, 25 min at 10% consistency and an initial NaOH concentration of 550 kg NaOH/t of pulp. EH stage was performed aiming to decrease cellulose DP. The other bleaching stages were carried out in polyethylene bags, which was heated in a microwave to the desired temperature and transferred to a temperature-controlled steam bath and kept for the required time. After each bleaching stage, pulp samples were washed with 9 m³ of distilled water.

Analytical procedures

Oven-dried weights were determined by drying to constant weight at 105 °C in a convection oven. The pulps screened yield was determined gravimetrically by

Parameters	Viscose grade PHK pulp				Acetate grade PHK pulp						
	0	D	(EH)	D	Р	0	D	(EP)	D	Р	CCE
Consistency (%)	10	10	10	10	10	10	10	10	10	10	10
Temperature (°C)	115	70	70	70	70	90	70	70	70	70	25
Time (min)	60	30	90	120	120	60	30	90	120	120	30
Final pH	-	3.0	7.0	4.5	_	_	3.0	-	4.5	_	_
O ₂ (kg/t)	20	_	_	_	_	20	_	_	_	_	_
ClO ₂ (kg/t)	-	5.0	_	10	-	_	5.0	-	2.0	-	-
NaClO, as Cl ₂ (kg/t)	-	_	4	_	_	_	_	-	-	_	_
MgSO ₄ (kg/t)	_	_	_	_	_	5	_	1.5	_	_	_
H_2O_2 (kg/t)	_	_	_	_	1.0	_	_	4.0	_	1.0	_
NaO (kg/t)	20	_	1.0	_	3.0	20	_	8.0	-	3.0	550
H_2SO_4 (kg/t)	-	5.0	-	0.25	-	-	16	-	0.25	-	-

 Table 3 General bleaching conditions used for viscose and acetate grade pulps

drying a small part of representative samples at 105 °C overnight. The concentration of dissolved xylan was determined by post-hydrolysis analysis of the liquid samples. Briefly, 0.7 mL of 72% H₂SO₄ was added to 15 mL of liquid samples and the volume was made up to 20 mL with water. Subsequently, the samples were autoclaved at 121 °C for 1 h and analyzed by HPLC. Brightness determination was performed according to TAPPI 452 om-08, kappa number according to TAPPI 236 om-06 (TAPPI 2010), intrinsic viscosity according to SCAM-CM 15:99 (1999). The degree of polymerization (DP_v) was calculated from the intrinsic viscosity by the following equation (Van Heiningen et al. 2004): $DP_v = (1.65\eta_{int})^{1.11}$, where η_{int} is the intrinsic viscosity of the substrate. Ash and metal ions were determined according to TAPPI 211 om-93 and TAPPI 266 om-94, respectively. Sugar analysis was determined according to Wallis et al. (1996). Alpha cellulose content was determined by TAPPI T203 om-93. Alkali solubility was expressed as alkali resistance and determined according to TAPPI 235 cm-00, by the following formula: R% = 100 - S. Pulps reactivity was measured according to Fock (1959). Fiber length of the substrates was measured using a Fiber Quality Analyzer (LDA02, OpTest Equipment, Inc., Hawkesbury, ON, Canada). Briefly, a dilute suspension of fibers with a fiber frequency of 25–40 EPS (events per second) was transported through a sheath flow cell where the fibers are oriented and positioned. The images of fibers were detected by a built-in CCD camera, and the length of the fibers was measured by circular polarized light.

Results and discussion

SE pretreatment

Conditions used in this study are summarized in Table 4. Initial trials were performed to define the acid load to be used in organosolv pulping, once we could not have a pulp using auto-catalyzed (no acid addition) condition (100% of reject). The minimum amount of sulfuric acid enough to produce a pulp was 0.5% H₂SO₄ (w/w). Bioconversion condition (organosolv pretreatment conditions used to purify cellulose for bioconversion to ethanol: 170 °C, 1% H₂SO₄, 65% ethanol) was also evaluated for making a dissolving grade pulp. Other conditions used were auto-catalyzed (no acid addition) or dilute acid-catalyzed (with acid addition) SE prior to

used in this	Steam explosion (SE)	Organosolv pulping	Organosolv pulp
	No pretreatment	0% H ₂ SO ₄	100% of reject
	No pretreatment	0.5% H ₂ SO ₄	Pulp
	No pretreatment	1.0% H ₂ SO ₄	Pulp
	0.1% H ₂ SO ₄	0% H ₂ SO ₄	>30% of reject
	$0\% H_2SO_4$	0.5% H ₂ SO ₄	Pulp
	0% H ₂ SO ₄	0% H ₂ SO ₄	100% of reject
	0.1% H ₂ SO ₄	0.5% H ₂ SO ₄	Pulp

 Table 4
 Conditions used in this study

Fig. 2 Screened yield (a), fiber length (b) and kappa number (c) of unbleached pulps. SE pretreatment \blacktriangleright steam explosion pretreatment, *NP* no SE pretreatment, *NE* steam pretreatment without explosion. *Error* bars represent one standard deviation from the mean

auto-catalyzed or dilute acid-catalyzed organosolv pulping. Besides auto-catalyzed organosolv pulping, the condition auto-catalyzed SE prior to auto-catalyzed organosolv also was not able to produce pulp (100% of reject). The condition 0.1% H_2SO_4 SE prior to auto-catalyzed organosolv gave a pulp with more than 30% of rejects, which was also not used for this study.

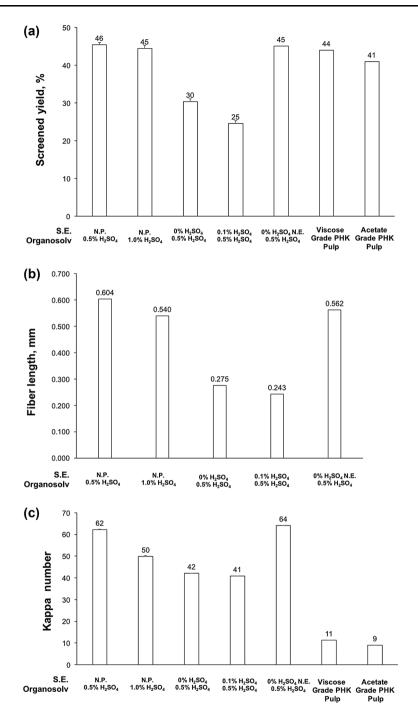
In SE process, high-pressure, high-temperature steam is introduced into a sealed chamber containing woody lignocellulosic material in the form of chips. After some minutes, 2 min in this study, the pressure is released, causing the steam to expand within the lignocellulosic matrix, separating individual fibers with minimal loss of material (Mabee et al. 2006). Steam-exploded chips showed yield higher than 90% regardless of the acid addition in the pretreatment.

Hemicelluloses are undesirable impurities in dissolving pulps. Thus, it is necessary to reduce the hemicellulose content and produce pulps with high cellulose content (Sixta 2006). In the SE process, the acetyl groups provide autocatalysis along with the hydronium ions. In addition, the hemicellulose–lignin bonds are cleaved, which all promote hemicellulose hydrolysis to soluble sugars (Chen and Liu 2007; Liu et al. 2013; Martín-Sampedro et al. 2012). In this study, the yield of dissolved xylan was 27 and 37% for auto-catalyzed and acid-catalyzed SE pretreatment, respectively. The addition of acid in the pretreatment increased the dissolution of xylan in the liquor.

Organosolv and PHK pulping

Screened yield, fiber length and kappa number of unbleached pulps are shown in Fig. 2. In general, organosolv pulps without SE pretreatment and PHK pulps (viscose and acetate grade pulps) showed screened yield higher than 40% (Fig. 2a). Organosolv pulps without SE pretreatment (0.5% H₂SO₄ organosolv and 1.0% H₂SO₄ organosolv pulps) showed almost the same yield (about 45%). Steam-exploded organosolv pulps showed low yields, about 30 and 25% for 0% H₂SO₄ SE 0.5% H₂SO₄ organosolv and 0.1% H₂SO₄ SE 0.5% H₂SO₄ organosolv pulps, respectively. To evaluate the effect of the explosion on fiber length and yield, an auto-catalyzed single trial was performed at the same conditions (200 °C with a retention time of 2 min) without explosion. As shown in Fig. 2a, auto-catalyzed non-exploded 0.5% H₂SO₄ organosolv pulp obtained screened yield of 45%.

The explosion effect can also be noted analyzing fiber length (Fig. 2b). Organosolv pulps without SE pretreatment (0.5% H₂SO₄ organosolv and 1.0% H₂SO₄ organosolv pulps) and auto-catalyzed non-exploded 0.5% H₂SO₄ organosolv pulp showed fiber length from 0.50 to 0.60 mm. However, steam-exploded organosolv pulps (0% H₂SO₄ SE 0.5% H₂SO₄ organosolv and 0.1% H₂SO₄ SE 0.5% H₂SO₄ organosolv and 0.1% H₂SO₄ SE 0.5% H₂SO₄ organosolv pulps) showed fiber length lower than 0.30 mm, noting that explosion breaks the fiber damaging fiber length and also pulp yield.



According to Martín-Sampedro et al. (2012), in the blasting process, rapid flashing to atmospheric pressure and turbulent flow of the material cause fragmentation of the material by mechanical shear and make the material fiberized. Lignin became plasticized, which made the steamed material easily fiberized. Therefore, steam explosion pretreatment weakened the lignocellulosic structure, opened up the cellulose fibers for further reactions and increased the extractability of lignin during the subsequent Kraft pulping process. In this paper, organosolv pulps without SE pretreatment (0.5% H₂SO₄ organosolv and 1.0% H₂SO₄ organosolv pulp showed residual lignin content higher than steam-exploded organosolv pulps (Fig. 2c), which is in accordance with Martín-Sampedro et al. (2012), once steam explosion weakens the lignocellulosic structure helping to increase the extraction of lignin. PHK pulps showed low residual lignin content, which was expected once delignification in the Kraft process is more efficient than organosolv process.

Bleaching and dissolving pulp characteristics

Use of dissolving wood pulp depends on its purity (cellulose content), and to obtain products of high quality, these pulps must fulfill certain requirements, such as high cellulose content, low hemicellulose content and high cellulose reactivity (Bajpai 2012). As can be noted in Table 5, the acetate grade pulp showed the highest purity

Steam explosion	_ ^a	_ ^a	0% H₂SO₄	0.1% H₂SO₄	Viscose grade PHK pulp	Acetate grade PHK pulp	
Organosolv	0.5% H ₂ SO ₄	1.0% H ₂ SO ₄	0.5% H ₂ SO ₄	0.5% H ₂ SO ₄			
Alpha cellulose (%)	89.4	83.9	86.9	87.9	95.4	97.7	
Xylan (%)	4.5	2.1	3.8	3.0	3.9	1.5	
Micro-kappa number	1.5	0.54	1.2	1.1	0.61	0.81	
Brightness, % ISO	88.6	90.2	88.7	89.3	90.4	89.4	
Viscosity (dm ³ / kg)	499	208	269	174	481	750	
DP _v ^b (Da)	1721	651	868	535	1654	2708	
R18 ^c (%)	91.4	87.8	89.2	92.4	95.8	97.7	
R10 ^d (%)	87.3	80	84.7	83.4	95.0	97.7	
Ash content (%)	0.17	0.08	0.09	0.07	0.08	0.25	
Fe (mg/kg)	91	97	103	97	23	9.4	
Ca (mg/kg)	69	46	69	69	35	38	

Table 5 Properties of organosolv and PHK (viscose and acetate grade) dissolving pulps

^a without steam explosion pretreatment, ^b viscosity average cellulose degree of polymerization, ^c alkali resistance in 18% NaOH, ^d alkali resistance in 10% NaOH

(97.4% of alpha cellulose), which fulfills acetate grade pulp requirements (US 2112999 A, 1938). A pulp useful for making lyocell fibers has a low degree of polymerization and alpha cellulose content less than about 90% (US 6706876 B2, 2004). Organosolv pulps produced in this study achieved alpha cellulose content between 83.9 and 89.4%, which qualifies them for making lyocell fibers (US 6706876 B2, 2004).

Hemicelluloses are undesirable impurities in dissolving pulps, and the undesirable effects of it in dissolving pulps are well established (Molin and Teder 2002; Page 1983, 1985). Organosolv pulps produced in this study achieved xylan content between 2.1 and 4.5%, which makes them acceptable for viscose grade applications, including viscose grade pulp produced in this study.

Organosolv and PHK pulps showed very low residual lignin content (between 0.07 and 0.20%) as can be seen from the micro-kappa number (between 0.54 and 1.5). All pulps achieved high brightness levels, about 88.6 and 90.4%.

The 1.0% H₂SO₄ organosolv pulp and steam-exploded organosolv pulps (0% H₂SO₄ SE 0.5% H₂SO₄ organosolv and 0.1% H₂SO₄ SE 0.5% H₂SO₄ organosolv pulps) showed considerably reduced DPs (i.e., low viscosities). Acid addition in the case of 1.0% H₂SO₄ organosolv pulp degraded polysaccharide chains reducing cellulose viscosity (208 dm³/kg). As mentioned before, steam explosion pretreatment opened up the structure, helping the protons to act on the cellulose, reducing its viscosity (and DP) giving low-viscosity pulps, 269 (DP of 868) and 174 (DP of 535) dm³/kg for 0% H₂SO₄ SE 0.5% H₂SO₄ organosolv and 0.1% H₂SO₄ SE 0.5% H₂SO₄ organosolv pulps, respectively. However, low-DP pulps are also suitable for some applications, e.g., lyocell fibers covering DP range from about 150 to 3000 (US 6706876 B2, 2004). PHK pulps showed suitable viscosity values for their respective applications and also the 0.5% H₂SO₄ organosolv pulp that achieved viscosity value (499 dm³/kg) suitable for viscose grade applications.

The acetate grade pulp was of high purity (R18 97.7%). Between organosolv pulps the 0.1% H₂SO₄ SE 0.5% H₂SO₄ pulp showed the most degraded cellulose content (R18 minus R10). This behavior can be explained by the action of mechanical disruption of steam explosion pretreatment breaking the cellulosic fibers and acid addition used to improve the separation of lignin from the carbohydrate components. In general, degraded cellulose content of organosolv pulps can be related to their respective viscosity values.

The presence of inorganic compounds can impair the filterability of a cellulose spinning dope and leads to a gradual clogging of the spinnerets, and controlling the inorganic compounds content to the lowest level is the best option (Page 1983; Sixta 2006). In general, all pulps showed low ash content (0.07–0.25%). According to some specifications, Fe and Ca content should be lower than 5 and 15 mg/kg, respectively (Lenzing 2014; Sixta 2006). In this study, all pulps showed Fe and Ca ions content higher than the specifications for dissolving grade pulps.

Reactivity is considered the most significant property of dissolving pulps and is related to the accessibility of chemicals to the cellulose (Krassig 1993), which means the relative ease in which the hydroxyl groups can be achieved by reagents (Sixta 2006). The highest reactivity (70%) was reached by 1.0% H₂SO₄ organosolv pulp (Fig. 3). The acid load added to the pulping contributed to cleaving the

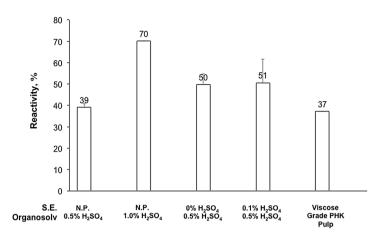


Fig. 3 Reactivity of organosolv and viscose grade dissolving pulps. SE pretreatment steam explosion pretreatment, *NP* no SE pretreatment. *Error bars* represent one standard deviation from the mean

glycosidic bonds by reducing the degree of polymerization of cellulose chains and increasing pulp reactivity. It seems that acid load was also more selective removing hemicellulose (2.1% of xylan), ensuring higher reactivity. Steam-exploded organosolv pulps showed similar reactivity (about 50%). These pulps showed lower reactivity than 1.0% H₂SO₄ organosolv pulp (70%) and higher xylan content (3.8% of xylan for 0% H₂SO₄ SE 0.5% H₂SO₄ organosolv pulp. The 0.5% H₂SO₄ organosolv pulp showed reactivity (39%) close to viscose grade pulp reactivity (37%).

Conclusion

Steam explosion pretreatment dissolved more xylan in the liquor for the dilute acidcatalyzed condition. Steam explosion pretreatment helped to decrease lignin content of unbleached organosolv pulps and damaged fiber length. The 1.0% H₂SO₄ organosolv pulp showed the highest reactivity due to higher acid load that reduced cellulose DP and increased pulp reactivity. In general, organosolv pulping process produced pulps with low DP values which can be used for making lyocell fibers.

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