#### **ORIGINAL PAPER**



# Characterization of major properties and aroma profile of kiwi wine co-cultured by *Saccharomyces* yeast (*S. cerevisiae*, *S. bayanus*, *S. uvarum*) and *T. delbrueckii*

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#### Abstract

The effect of different yeast strains on the quality of kiwi wine was investigated by polyphase determine approaches in the present research. The influence of co-culture and inoculation sequence on the quality was also explored simultaneously. Results suggested that the characteristics of the kiwi wine were affected by the metabolic characteristic of strains. The flavor content and their flavor profile of samples fermented by co-culturing of strain among species, genus, and families. When *Saccharomyces bayanus* (Y5 or Y6) co-cultured with *Torulaspora delbruecki* Y7, the ratio of phenethyl alcohol increased, but that of octanoic acid and ethyl octanoate decreased significantly. The odor activity value (OAV) of ethyl octanoate and ethyl hexanoate was increased by co-culturing *Saccharomyces* with *T. delbrueckii*, and that of decanal and terpinen-4-ol was enhanced by co-culturing of different strains of *Saccharomyces*. It was an excepting process to obtain high quality of kiwi wine by co-culturing technology of yeasts, and was very effective to optimize the process by polyphase analysis approaches.

Keywords Kiwi wine · Flavor compounds · Co-culture · Saccharomyces · T. delbrueckii

# Introduction

In general, *Saccharomyces cerevisiae* is considered as an important starter in the wine fermentation process, whether it is applied as a commercial starter or indigenous yeasts, to regulate the microbial community diversity and their metabolism. But it is homogenizing for the quality feature of the final wine. In fact, the contribution to grape

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wine characteristics was not only depended on Saccharomyces, but also was closely related to non-Saccharomyces whether the wine belonged to Old World or the New World. Therefore, it is one of the focuses on co-culturing S. cerevisiae with non-Saccharomyces cerevisiae or non-Saccharomyces to enhance the quality of wine. In this way, the role of dominant flora originated from raw material and winery environment can be simulated effectively, and the wine fermentation process was regulation whether quality enhancing or avoiding safety risk. Different species of Saccharomyces involved in S. cerevisiae, S. bayanus, and S. uvarum, etc. have their own unique feature. S. cerevisiae is mainly used to produce ethanol and S. bayanus and S. uvarum are characterized by low acetic acid production and high yields of glycerin and lactic acid. The latter two are used to produce high-quality wines [1, 2]. For example, the characteristics of Malvasia delle Lipari wine fermented by S. uvarum differ compared with the wine fermented by S. cerevisiae, in which volatile acidity and ethanol were lower [3]. Similarly, it was observed that significant difference of flavor profile among two kinds of Chardonnay wines, which one was fermented by S. bayanus, and another was fermented by S. cerevisiae [4].

Non-Saccharomyces were originated from raw material and wine brewing environment. Of these non-Saccharomyces, grape, longan, lychee, and cherry wine were produced by coculturing *S. cerevisiae* with *T. delbrueckii*, and the content of acetate and ethyl ester were higher than that of ones only brewed by *S. cerevisiae* [5–8]. The contents of ethanol, glycerin, volatile acids, and organic acids were also different when cocoa beans, bilberry wine, and mango wine were fermented by co-cultured *S. cerevisiae* with *T. delbrueckii*, compared with the result obtained by pure fermentation pattern, especially ethanol and acetic acid content decreased [9–12]. Besides, Lu et al. reported that coculture of *T. delbrueckii* and *Pichia kluyveri* could complete alcoholic fermentation of durian wine [13].

Kiwifruit (*Actinidia chinensis*) is an edible body of kiwi woody vine and rich in nutrients. It is widely distributed in China, New Zealand, Italy, and Chile [14]. However, the shelf life is short and easy to over-ripe, threatening the further development of the kiwifruit chain [15]. Most nutrients and bioactive substances can be transferred to wine if it brewed wine, so that kiwifruit can be used to brew fruit wine to increase the added value [16]. *S. cerevisiae* has been used to brew kiwi wine. It suggested that the characteristics of the strain also affect the concentration of phenolics and volatiles [17]. However, the intensity of flavor was generally weaker than that of wine [18]. It may be caused by the difference of two raw material components. These shortcomings may be overcome by co-culturing *S cerevisiae* with non-*Saccharomyces cerevisiae* or non- *Saccharomyces*.

In the present, we investigated the impact of co-culturing *S. cerevisiae* with non-*Saccharomyces cerevisiae* (S. *bayanus* and *S. uvarum*), as well as inoculation mode on the kiwi wines quality compared with that of ones fermented by single strain. Additionally, the difference of volatiles profiles in kiwi wine brewed by co-culturing *Saccharomyces* strains and *T. delbrueckii* was studied. To the best of our knowledge, it was the first report on the influence of coculturing *Saccharomyces* and *T. delbrueckii* on the quality of kiwi wine.

# **Materials and methods**

#### **Materials and strains**

#### Chemicals

The standards, including oxalic acid, citric acid, tartaric acid, L-malic acid, succinic acid, lactic acid, acetic acid, propionic acid, methyl octanoate, and 2-octanol were purchased from Sigma-Aldrich. *Ltd. Co* (Shanghai, China). Other chemicals were purchased from Chengdu Jinshan (Chengdu, China).

#### Microorganism

Saccharomyces cerevisiae: S. cerevisiae Y1 and S. cerevisiae Y2, S. bayanus: S. bayanus Y3, S. bayanus Y4, and S. bayanus Y5, S. uvarum Y6, T. delbrueckii Y7 were isolated from the soil located at kiwifruit wine factory and identified according to the results of physiological and biochemical experiments, cell and colony morphology, as well as ITS sequence. These strains mentioned above were all preserved in our Lab, and S. cerevisiae Y1 (CCTCCM2019521), S. bayanus Y4 (CCTCCM2019522), and T. delbrueckii Y7 (CCTCC M2019523) were also preserved in the China Center for Type Culture Collection.

#### **Kiwi wine fermentation**

#### Kiwifruit juice preparation

Kiwifruit (var. Hayward) were harvested in 2019 with a total sugar of 8.68% and a total acid of 12.61 g/L (tartaric acid). It was purchased from the local farm product market beaten into pulp after selection and removed of impurities. 100 mg/L of sodium hydrogen sulfite was immediately added into the pulp to control harmful bacteria and yeasts. 20 mg/L of pectinase (Lallzyme EX-V, Lallemand, France) was added, and maintained at  $37 \pm 2$  °C for 60 min. The total soluble solids were adjusted to 19 °Brix by adding sucrose. 2 L of the pulp was spilt into a 2.5 L of the widemouth reagent bottle.

#### Starter suspension preparation

Pre-cultures were carried out initially by inoculating from the agar slant test tube of each strain into 5 mL of YPD broth (1% yeast extract, 2% peptone, 2% glucose) growing the cells at  $30 \pm 1$  °C on a test tube rotator for approximately 24 h. Thereafter, 50 µL of this pre-culture was then re-inoculated into 5 mL fresh YPD broth and grown for approximately 24 h at  $30 \pm 1$  °C on the test tube rotator. Cells were then inoculated from this pre-culture into 200 mL YPD broth at OD600 nm 0.1 and grown at  $30 \pm 1$  °C with shaking at 120 rpm until the cells reach mid-exponential phase (ca. 9 h). To obtain yeast cells, the broth was centrifuged at  $4500 \times g$  for 10 min to remove supernatant. Afterwards, the biomass was washed by resuspension in 0.9% sterile sodium chloride solution, followed by centrifugation at  $4500 \times g$  for 10 min. The washing was repeated three times, after which the pellet was collected and resuspended in the treated kiwifruit juice.

#### Fermentation procedure and condition

Prior to inoculation, the yeast cell population was determined by a hematocytometer. Four types of fermentation were conducted: (1) pure fermentations by inoculation with a single S. cerevisiae or non-Saccharomyces cerevisiae yeast strain for selecting the strains to fit kiwifruit wine-making, and the number of samples were Sample No 1, Sample No 2, and Sample No 3, respectively; (2) kiwifruit wine fermenting by co-culturing S. cerevisiae Y2 and with non-Saccharomyces cerevisiae, including S. bayanus Y5 and S. uvarum Y6 (sample No 4 and sample No 5); (3) kiwifruit wine fermenting by co-culturing S. cerevisiae Y2, S. bayanus Y5 and S. uvarum Y6 with T. delbrueckii Y7, respectively, via simultaneous inoculation, and the number of samples were Sample No 6, Sample No 7 and Sample No 8; (4) the co-culturing mode was the same as that of mode described above, but T. delbrueckii was first inoculated into kiwifruit juice, and then two days later, Saccharomyces was inoculated, and the number of samples were Sample No 9, Sample No 10 and Sample No 11, respectively.

Pre-treated kiwifruit juice was inoculated with reconstituted starter resuspension by the kiwifruit juice, and the initial concentration of starter reached the level of  $2.4 \times 10^6$  CFU/mL. The co-culture of fermentation was inoculated at a rate of 1:1 for two different stains starter. The fermentation was carried out under the static condition at 15° C. The residual sugar content, ethanol content, and pH were evaluated to monitor the fermentation. The fresh kiwi wine was filtered when the fermentation finished, and then stored at 4 °C for further analyze.

## Determining of physicochemical properties and antioxidant activity

The content of residual sugar, titration acidity, and ethanol was analyzed according to GB/T15038-2006 [19]. The color was determined with the method described by Bimpilas et al. [20], and absorbance was measured by UV–visible spectrophotometer (TU-1901, Beijing Purkinje General Instrument, Beijing, China), where color intensity  $= A_{420} + A_{520} + A_{620}$ , lustre  $= A_{420}/A_{520}$ . The polyphenol content, DPPH and ABTS free radical scavenging activity were determined by the method reported by Wang et al. [21]. The polyphenol content was expressed as gallic acid equivalent (mg GAE/L), and DPPH and ABTS free radical scavenging activity were expressed as trolox equivalent antioxidant capacity (TEAC).

#### **Organic acids analysis**

10.0 mL of sample was centrifuged at 12,000 r/min for 10 min at 4 °C, and the supernatant was purified by SPE column (Swell scientific instruments Co., Ltd. Chengdu,

China), subsequently filtered through a 0.22 µm filter (Micron Separation Inc., Westborough, MA). The filtered samples were injected into the Agilent 1260 HPLC (Agilent Technologies, Santa Clara, USA) system equipped with an Alltech OA-1000 organic acid column (300 mm×6.5 mm, Grace, Columbia, USA) maintained at 75 °C according to the method in Liang et al. [22]. Degassed H<sub>2</sub>SO<sub>4</sub> (9 mM) was used as mobile phase and the organic acids, including oxalic, citric, tartaric, L-malic, succinic, lactic and acetic were detected using UV detector (215 nm). A 10 µL injection volume was used for samples and standards. Organic acids were quantified by external standards. The result was expressed as g/L.

#### Volatile compounds analysis

The analysis of volatile compounds was performed on Trace GC Ultra-DSQ II GC–MS (Thermo Electron Corporation, Waltham, USA), in combination with HS-SPME and DVB/CAR/PDMS fiber (Supelco, Bellafonte, USA), according to Niu et al. [23]. The chromatographic column for GC–MS analysis was HP-Innowax ( $30 \text{ m} \times 0.25 \text{ mm} \times 0.25 \text{ µm}$ , Agilent J&W, Santa Clara, USA). Volatile compounds were identified by comparing MS with the standard library in NIST05 and verified by Kováts retention indices (RI) with that reported in literatures, which was calculated by using C<sub>8</sub>–C<sub>20</sub> n-alkanes. At the same time, semi-quantitation of volatile compounds can be obtained by comparing the area of internal standards (methyl octanoate and 2-octanol) with the total ion chromatogram.

#### **Statistical analysis**

All analysis were performed in triplicate and the data were described as means  $\pm$  standard deviation. Difference between samples was evaluated by analysis of variance (ANOVA) and Duncan's tests using IBM SPSS Statistics 25 (SPSS Inc., Chicago, Illinois, USA), and P < 0.05 indicated statistical significance. Partial least square-discrimination analysis (PLS-DA) was employed to correlate aroma compounds with different inoculation modes using Simca 14.1 (Umetrics, Umeå, Sweden). The cluster analysis and heatmap were carried out by the R language.

# **Results and discussion**

# Characterizing major properties of kiwi wine brewed by different strains

The major properties of kiwi wine brewed by six different strains, which fall into *Saccharomyces*, were investigated. Significant difference of residual sugar (RS express as glucose) and the color was observed, ranged from 25.08 to 40.82 g/L. It lied on the brewing characteristics of strain used. For example, the RS content for S. cerevisiae Y1 was not only higher than that of S. cerevisiae Y2, but also higher than other species, such as S. bayanus, S. uvarum. Contrast to S. cerevisiae, S. bayanus, and S. uvarum were mainly in lower acetic acid production and lower ethanol yield. However, no significant difference in ethanol content was founded among these kiwi wines, it may be caused by raw material feature as well as operation parameter, such as aeration conditions [24]. It was reported previously that S. bayanus and S. uvarum were not differed significantly from S. cerevisiae for these indexes as mentioned above. For example, Hayasaka et al. reported that there was no difference in ethanol content in Cabernet Sauvignon red wine fermented by S. bayanus and S. cerevisiae, respectively [25]. Similarly, Furmint wine fermented by S. uvarum and S. cerevisiae, respectively, had similar volatile acid content [26]. The color of the wine is an important sensory characteristic, and closely related to consumers' acceptance. Compared to other strains, the color intensity of kiwi wine fermented by S. bavanus Y5 was lower; it may be due to anthocyanin absorbed and reaction with some chromogenic substances [27], resulting in the color loss of the wine which can be avoided by selecting the appropriate yeast. Lustre of kiwi wine brewed by S. uvarum was significantly lower contrasted to other samples, which means it's reddish.

As shown in Fig. 1, no difference of total organic acid content, ranged from 27.02 to 28.34 g/L, was observed, while the profile was a little divergent among the six samples. Citric acid content from 12.63 g/L in raw material raised to 21.65–22.28 g/L constituting ca 80% of total



Fig. 1 Difference of organic acid profile in kiwi wine brewed by different strains

organic acid content by fermented, while quinine acid content was decreased significantly, from 7.52 g/L to undetectable. Total organic acid contents were higher than the grape wine content [28], even though that in kiwi wine [29]. It was related to variety, origin, as well as maturity in addition to the fermentation process [30], which resulted in the increase in content of L-malic acid and succinic acid.. Compared with *S. cerevisiae* Y1, the content of L-malic acid and succinic acid in kiwi wine fermented by *S bayanus* except for Y5 was higher. The acetic acid content in the sample fermented by Y1 and Y5, respectively, was lower. However, the contents of succinic and acetic acid in the samples fermented by *S. uvarum* was inconsistent with the results reported previously [2], which may be also resulted by similar causes as described above.

## Difference of volatiles in the kiwi wine fermented by various yeast strains

As shown in Table S2 (Online Resource), 46 volatile compounds were identified, in which 24 components were detected in all samples. These volatiles were divided into six classes according to their chemical structure (25 esters, 9 alcohols, 5 acids, 3 terpenes, 2 aldehydes, 2 phenols), and were reported in the previous document [17, 18]. Six components were dominant which were ethyl octanoate, ethyl decanoate, ethyl 9-decenoate, phenylethyl alcohol, octanoic acid, and decanoic acid, respectively. The total volatile content ranged from 5.43 to 9.40 mg/L in the six kiwi wines, and that in kiwi wine by Y5 was highest and the content of esters, acids, aldehydes, phenols were also highest, while that in kiwi wine by S. uvarum was lowest due to the lower content of esters, acids, phenols, and terpenes than that of others samples. The difference of volatile species among kiwi wine samples fermented by different yeast strains was observed, which 32, 32, 34, 37, 38 and 34 of volatile species in kiwi wine fermented by Y1, Y2, Y3, Y4, Y5, and Y6, respectively. Nine volatiles were not produced by S. cerevisiae, involved in decanal, isooctyl alcohol, 1-octanol, cis-hex-3-en-1-ol, isobutyl decanoate, pentyl decanoate, methyl decanoate, ethyl linoleate, and ethyl tetradecanoate and terpenes were not produced by Y1.

The volatiles profile among these samples were different, as shown in Fig. 2. Ethyl octanoate, ethyl decanoate, and ethyl 9-decenoate were the dominant in all samples endowed a fruity and floral scent to kiwi wine. However, their sum proportion was different, which were 72.23% (Y1), 73.05% (Y2), 81.97% (Y3), 76.67% (Y4) and 72.30% (Y5) and 71.50% (Y6), respectively. Compared to *S. cerevisiae* and *S. uvarum*, the ability of ethyl esters-producing for *S. bayanus* was stronger, which was also affected by fermentation temperature, aeration, and sugar contents [31]. Of alcohols, phenylethyl alcohol and isoamyl alcohol were dominant,



Fig. 2 Difference of volatile profile in kiwi wine brewed by different strains

ranged from 85.24 to 90.69%. The proportion of phenylethyl alcohol and isoamyl alcohol in other wines were also high [32]. Acids were one of the dominant volatiles ranging from 23.68 to 29.74%. The content in the samples fermented by Y4 and Y5, contributed by octanoic acid, decanoic acid, and dodecanoic acid, were higher than others which was no significant difference. Terpenes play an important role in the characteristic flavor of wine which endowed floral, rose, and bell orchid odor to wines, originated from the cell wall of grape skin. The content in wine relied on strains characteristics. In our present experiment, similar results were obtained using kiwifruit juice as raw material. Decanal was detected in the samples fermented by Y4 and Y6, which contributed an orange peel odor even though the content was very low. The content of 4-vinyl guaiacol (4-VG) in the samples fermented by Y5 was higher.

Of these volatiles, changes of eight volatiles content (OAV > 1) affected significantly the flavor characteristics of kiwi wine. In fact, the numbers of volatile with OAV > 1were closely related with the characteristic of species and strains. For example, seven and five components were detected out in the samples fermented by Y1, Y6, and Y2, respectively, in which isoamyl acetate was only detected in Y1 sample. Although, all Y3, Y4, and Y5 belonged to S. bayanus, 6 volatiles were detected in Y3 and Y5's, while 7 volatiles were only examined in Y4's. Among them, S. bayanus fermented kiwi wine has a higher OAV of ethyl octanoate (487.91-588.82), ethyl decanoate (5.09-6.85), and octanoic acid (1.53-2.31), which contribute to the fruity, floral, and cheese aroma of the kiwi wine, S. uvarum has a higher OAV of decanal (2.22) with aroma of orange peel, whereas Y1 has a higher OAV of isoamyl acetate (1.38) with aroma of banana. As shown in Fig. 3, decanal was not detected in S. cerevisiae, and Y2 has a lower OAV of 4-VG (0.20) resulted in an absence of spices scent.

Physiochemical indexes and volatile compounds (OAV > 0.1) of six yeasts were used for PLS-DA analysis, which 54.40% of the total variance was explained. As shown in Fig. 4, the results showed that all samples were divided into three groups, namely the *S. cerevisiae* group





Fig. 4 Biplot for PLS-DA of physiochemical indexes and volatile compounds (OAV > 0.1) in kiwi wine brewed by different strains

(Y1 and Y2), the *S. bayanus* group (Y3, Y4 and Y5), and the *S. uvarum* group (Y6). Among them, *S. cerevisiae* Y2 was related to L-malic acid, succinic acid, ethyl hexanoate, phenylethyl alcohol, and citronellol. In addition, Y5 has a higher yield of ethyl esters when compared with Y3 and Y4, such as ethyl decanoate, ethyl 9–decenoate, ethyl octanoate, ethyl dodecanoate, ethyl hexadecanoate. And the OAV of the aroma compounds of Y5 (OAV > 1) was larger than that of Y3 and Y4. Therefore, *S. cerevisiae* Y2, *S. bayanus* Y5, and *S. uvarum* Y6 were selected for the subsequent experiment.

# Effect of co-cultured on major properties and flavor profile in the kiwi wine

As shown in Table S3 (Online Resource), difference of major physicochemical properties, such as RS, ethanol, acidity (expressed as tartaric acid g/L), organic acid, and free radical removal capacity was no significant among these samples, which involved in different strains belonging to *Saccharomyces*, i.e., co-culturing *S. cerevisiae* with *S. bayanus* and *S. uvarum*, respectively, as well as *Saccharomyces* (*S. cerevisiae* Y2 with *S. bayanus* Y5 and *S. uvarum* Y6) co-cultured with non-*Saccharomyces* (*T. delbrueckii*) by simultaneous or sequential inoculating.

As shown in Table 1, sixty-one volatile compounds were identified among these samples. These components were divided into seven classes which involved in esters (24), alcohols (12), acids (9), aldehydes (3), ketones (3), phenols (4), and terpenes (6) according to their chemical structure. There are 37, 40, and 34 components identified in the samples fermented by Y2, Y5, and Y6, respectively. Among these samples, undetected components in Y2 and Y5 were all four but were ethyl 9-decenoate, hexanoic acid, linalool and citronellol for the former, and were isoamyl acetate, decyl alcohol, decanoic acid and decanal for the latter. However, four undetected components in Y6's were (E, E)-farnesol, 2,4-di-*tert*-butylphenol, 4-VG and (E)-nerolidol, respectively. 35 and 38 components were identified when Y2 co-cultured with Y5 and Y6, respectively. 11 and 9 volatiles were undetected in Y2 + Y5 and Y2 + Y6, respectively, which were once identified in the respective single strain, while 3 and 5 volatiles were newly detected. Six volatiles were undetected in both samples which involved 2-hydroxyethyl hexanoate, 1-hexadecanol, 2,6-di-*tert*-buty-4-sec-butyphenol, decyl alcohol, isoamyl acetate, and (E, E)-farnesol. Newly detected volatile was only diethyl succinate in both of the samples.

When three different strains belonged to Saccharomyces co-cultured with T. delbrueckii (Y7), volatiles species was unchanged for Y2 + Y7 and Y6 + Y7, undetected, and newly examined volatiles were all nine and ten contrast to the samples of Y2 and Y6, respectively. While undetected volatiles were eight, but new examined volatiles were five in Y5+Y7 compared with Y5's. New-detected components were 9,12-octadecadienoic acid ethyl ester, ethyl oleate, and tetradecanoic acid among three kinds of co-cultured samples, as well as nonanoic acid in Y5 + Y7 and Y6 + Y7. Only  $\alpha$ -terpineol was undetected in all co-cultured samples, decyl alcohol was unexamined in addition to Y5 + Y7, while diisobutyl adipate and ethyl decanoate were undetected in Y5 + Y7 and Y6 + Y7. Compared with the samples inoculated simultaneously, the volatiles species were changed in respective sequential inoculated ones. For example, the undetected species were 10, 8, and 13, while new examined

Table 1	Mean concentration of aroma of kiwi wine 1	ermented by	pure and co-	cultured amo	ng Saccharon	<i>nyces</i> , as well	as Sacchar	omyces with	I. delbrueckii	(hg/L)		
No.	Compounds	Pure			S. cerevisiae	and Saccha-	Saccharom	<i>tyces</i> and non	-Saccharomy	ces		
					romyces Sun	nultaneous	Simultaneo	sno		Sequential		
		Y2	Y5	Y6	Y2+Y5	Y2+Y6	Y2 + Y7	Y5+Y7	Y6+Y7	Y2 + Y7	Y5+Y7	Y6+Y7
Esters												
Acetate	SS											
E1	Isoamyl acetate	9.48 <sup>b</sup>	pu	12.92 <sup>b</sup>	pu	nd	$20.22^{\circ}$	$3.20^{a}$	pu	$2.73^{a}$	$9.64^{\mathrm{b}}$	12.37 <sup>b</sup>
E2	Hexyl acetate	nd	pu	nd	pu	pu	$4.90^{a}$	pu	pu	pu	pu	pu
E3	Phenethyl acetate	119.12 <sup>c</sup>	$265.37^{a}$	117.13 <sup>c</sup>	$105.82^{\circ}$	98.59°	$100.76^{\circ}$	$208.98^{\rm b}$	$114.86^{c}$	$115.01^{\circ}$	124.62°	$123.98^{\circ}$
	Total	$128.60^{\mathrm{ab}}$	265.37 <sup>d</sup>	$130.04^{ab}$	$105.82^{ab}$	98.59 <sup>a</sup>	125.89 <sup>ab</sup>	212.19 <sup>c</sup>	$114.86^{ab}$	117.74 <sup>ab</sup>	134.26 <sup>b</sup>	136.35 <sup>b</sup>
Ethyl e:	sters											
E4	Ethyl hexanoate	33.46 <sup>cd</sup>	$24.10^{abcd}$	12.45 <sup>ab</sup>	$22.74^{\mathrm{abc}}$	$15.63^{\rm abc}$	40.76 <sup>d</sup>	$22.03^{abc}$	6.45 <sup>a</sup>	$10.79^{ab}$	24.59 <sup>abcd</sup>	$26.94^{bcd}$
E5	Ethyl lactate	nd	nd	nd	$2.70^{a}$	nd	nd	nd	2.23 <sup>a</sup>	nd	nd	nd
E6	Ethyl octanoate	236.17 <sup>ef</sup>	$113.00^{b}$	$149.60^{\circ}$	233.45 <sup>ef</sup>	214.14 <sup>de</sup>	369.31 <sup>g</sup>	67.23 <sup>a</sup>	85.19 <sup>a</sup>	92.19 <sup>ab</sup>	$198.80^{d}$	252.56 <sup>f</sup>
Ε7	Ethyl 2-furoate	$7.87^{bcde}$	$7.14^{bcd}$	8.27 <sup>cde</sup>	9.25 <sup>e</sup>	8.57 <sup>de</sup>	$7.53^{bcd}$	$6.51^{ab}$	8.21 <sup>cde</sup>	7.14 <sup>bcd</sup>	$6.92^{bc}$	5.31 <sup>a</sup>
E8	Ethyl decanoate	$55.06^{\mathrm{b}}$	24.89 <sup>a</sup>	65.19 <sup>b</sup>	58.17 <sup>b</sup>	53.21 <sup>b</sup>	98.29 <sup>c</sup>	pu	pu	24.15 <sup>a</sup>	52.16 <sup>b</sup>	84.26 <sup>c</sup>
E9	Ethyl benzoate	$15.93^{\circ}$	14.64 <sup>bc</sup>	13.22 <sup>abc</sup>	$13.51^{abc}$	$13.30^{abc}$	11.71 <sup>ab</sup>	13.65 <sup>abc</sup>	11.62 <sup>ab</sup>	13.69 <sup>abc</sup>	$13.09^{\rm abc}$	$10.87^{\mathrm{a}}$
E10	Ethyl 9-decenoate	nd	$3.93^{a}$	12.15 <sup>b</sup>	11.29 <sup>b</sup>	8.22 <sup>b</sup>	$20.60^{\circ}$	pu	12.23 <sup>b</sup>	$9.50^{\mathrm{b}}$	12.44 <sup>b</sup>	17.23 <sup>c</sup>
E11	Ethyl phenylacetate	pu	11.25 <sup>a</sup>	pu	pu	pu	pu	$9.13^{\mathrm{b}}$	pu	$6.03^{\circ}$	nd	pu
E12	Ethyl dodecanoate	pu	pu	nd	pu	nd	pu	pu	nd	pu	$0.97^{a}$	pu
E13	Ethyl tetradecanoate	pu	pu	$6.76^{a}$	nd	nd	nd	pu	nd	nd	nd	pu
E14	Ethyl hexadecanoate	13.95 <sup>ab</sup>	7.56 <sup>a</sup>	$16.26^{ab}$	$12.37^{\mathrm{ab}}$	$10.96^{ab}$	12.56 <sup>ab</sup>	$14.88^{ab}$	$10.81^{ab}$	38.53 <sup>b</sup>	12.24 <sup>ab</sup>	pu
E15	Ethyl oleate	pu	pu	nd	$2.40^{a}$	nd	2.58 <sup>a</sup>	$3.65^{a}$	2.48 <sup>a</sup>	2.63 <sup>a</sup>	nd	pu
	Total	362.44 <sup>de</sup>	206.51 <sup>b</sup>	283.89 <sup>c</sup>	365.87 <sup>de</sup>	324.02 <sup>cd</sup>	$563.34^{\mathrm{f}}$	137.08 <sup>a</sup>	139.22 <sup>a</sup>	204.66 <sup>b</sup>	321.20 <sup>cd</sup>	397.17 <sup>e</sup>
Other e	sters											
E16	Methyl hexadecanoate	$7.53^{ab}$	$8.69^{\mathrm{b}}$	11.05°	nd	$6.58^{ab}$	7.82 <sup>ab</sup>	6.42 <sup>ab</sup>	$6.03^{a}$	pu	$6.53^{\mathrm{ab}}$	nd
E17	Octyl formate	pu	$2.67^{a}$	pu	$3.88^{\mathrm{b}}$	$3.69^{\mathrm{b}}$	$3.97^{\mathrm{b}}$	pu	pu	pu	pu	pu
E18	2-Hydroxyethyl hexanoate	$14.58^{a}$	pu	nd	nd	nd	nd	pu	nd	nd	nd	pu
E19	Diethyl succinate	pu	pu	pu	44.54 <sup>c</sup>	50.98 <sup>d</sup>	pu	pu	37.38 <sup>b</sup>	$39.40^{bc}$	43.62 <sup>c</sup>	27.97ª
E20	Diisobutyl adipate	22.39 <sup>b</sup>	$9.77^{\mathrm{ab}}$	$10.29^{ab}$	pu	4.23 <sup>a</sup>	$3.10^{a}$	pu	nd	$2.95^{a}$	$2.97^{a}$	pu
E21	Homosalate	pu	$5.05^{a}$	nd	pu	$2.99^{b}$	pu	pu	nd	pu	pu	pu
E22	9,12-Octadecadienoic acid ethyl ester	nd	pu	nd	nd	nd	$3.62^{\mathrm{ab}}$	$5.50^{\mathrm{b}}$	$2.86^{a}$	pu	nd	pu
E23	Diisobutyl phthalate	21.23°	10.41 <sup>abc</sup>	16.71 <sup>bc</sup>	15.48 <sup>bc</sup>	11.52 <sup>abc</sup>	$9.09^{ m abc}$	$9.68^{\rm abc}$	7.54 <sup>ab</sup>	pu	pu	pu
E24	Dibutyl phthalate	35.63°	23.42 <sup>bc</sup>	23.01 <sup>bc</sup>	$16.90^{b}$	$18.04^{b}$	14.58 <sup>ab</sup>	15.37 <sup>ab</sup>	pu	nd	nd	pu
	Total	$101.36^{\circ}$	$60.03^{\rm abc}$	$61.06^{abc}$	80.79 <sup>bc</sup>	$98.02^{\circ}$	42.17 <sup>ab</sup>	$36.96^{a}$	53.81 <sup>ab</sup>	42.35 <sup>ab</sup>	53.12 <sup>ab</sup>	$27.97^{a}$
	Total	592.40 <sup>e</sup>	531.91 <sup>cde</sup>	474.99°	552.49 <sup>cde</sup>	520.64 <sup>cde</sup>	731.39 <sup>f</sup>	386.23 <sup>b</sup>	$307.90^{a}$	364.75 <sup>ab</sup>	508.58 <sup>cd</sup>	561.49 <sup>de</sup>

Table 1	(continued)											
No.	Compounds	Pure			S. cerevisia	e and Saccha-	Saccharom	yces and non	1-Saccharomy	ces		
					romyces Sur	nultaneous	Simultanec	sno		Sequential		
		Y2	Y5	Y6	Y2+Y5	Y2+Y6	Y2 + Y7	Y5+Y7	Y6+Y7	Y2 + Y7	Y5+Y7	Y6 + Y7
Alcohc	sic											
A1	Isobutyl alcohol	$29.10^{d}$	$10.92^{\rm abc}$	22.88 <sup>d</sup>	nd	24.63 <sup>d</sup>	21.52 <sup>cd</sup>	$8.93^{\mathrm{ab}}$	$23.90^{d}$	21.74 <sup>cd</sup>	$19.62^{bcd}$	6.62 <sup>a</sup>
A2	Isoamyl alcohol	358.63 <sup>bc</sup>	252.76 <sup>ab</sup>	$399.54^{\circ}$	$334.36^{bc}$	384.13°	311.34 <sup>abc</sup>	$210.17^{a}$	373.54°	333.71 <sup>bc</sup>	319.91 <sup>abc</sup>	252.52 <sup>ab</sup>
A3	n-Hexanol	52.05 <sup>bc</sup>	$46.88^{\rm abc}$	52.11 <sup>bc</sup>	$42.72^{ab}$	57.18 <sup>c</sup>	44.54 <sup>ab</sup>	37.83 <sup>a</sup>	49.34 <sup>abc</sup>	$39.38^{\mathrm{ab}}$	$40.32^{ab}$	$37.10^{a}$
A4	1-Heptanol	5.90 <sup>cd</sup>	9.41 <sup>e</sup>	$2.36^{a}$	$5.32^{bcd}$	4.57 <sup>b</sup>	4.87 <sup>bc</sup>	6.05 <sup>d</sup>	2.24 <sup>a</sup>	$2.57^{\mathrm{a}}$	$3.21^{a}$	pu
A5	Menthol	nd	$4.89^{a}$	pu	nd	nd	nd	pu	pu	pu	nd	pu
$\mathbf{A6}$	Cyclododecanol	nd	pu	pu	nd	nd	nd	pu	4.61a	pu	pu	pu
A7	Decyl alcohol	$2.83^{a}$	pu	2.34 <sup>b</sup>	nd	nd	nd	nd	pu	pu	nd	pu
$\mathbf{A8}$	Phenethyl alcohol	887.07 <sup>ab</sup>	$1873.86^{e}$	1192.66 <sup>c</sup>	957.24 <sup>ab</sup>	930.05 <sup>ab</sup>	768.05 <sup>a</sup>	1467.99 <sup>d</sup>	$1070.06^{bc}$	$969.91^{\rm b}$	$874.93^{ab}$	926.67 <sup>ab</sup>
A9	1-Dodecanol	5.75 <sup>bcd</sup>	$8.00^{d}$	$6.16^{\rm cd}$	$5.04^{bcd}$	$4.78^{\rm bc}$	$4.55^{bc}$	$4.80^{\mathrm{bc}}$	$3.12^{bc}$	$3.64^{\rm bc}$	$2.66^{\mathrm{ab}}$	pu
A10	(E)-Nerolidol	$15.16^{d}$	$15.40^{d}$	nd	$10.09^{ab}$	$11.26^{bc}$	7.08 <sup>a</sup>	14.1 <sup>6 cd</sup>	$7.86^{a}$	$15.00^{d}$	$8.49^{\mathrm{ab}}$	nd
A11	1-Hexadecanol	$9.46^{a}$	pu	nd	nd	nd	nd	nd	nd	nd	nd	pu
A12	3,7,11-Trimethyldodeca-6,10-dien-1-ol	nd	14.19 <sup>b</sup>	nd	3.11 <sup>a</sup>	$3.33^{a}$	nd	$11.59^{\circ}$	$2.80^{a}$	$5.48^{d}$	3.01 <sup>a</sup>	nd
	Total	$1365.94^{ab}$	2236.31 <sup>e</sup>	1678.04 <sup>cd</sup>	1357.88 <sup>ab</sup>	1419.92 <sup>abc</sup>	$1161.96^{a}$	1761.53 <sup>d</sup>	1537.47 <sup>bcd</sup>	1391.44 <sup>ab</sup>	$1272.14^{ab}$	1222.91 <sup>a</sup>
Acids												
C1	Acetic acid	$9.09^{ m abc}$	$7.73^{ab}$	$6.49^{a}$	$8.93^{\rm abc}$	$10.57^{\rm bc}$	$9.25^{\rm abc}$	11.24 <sup>c</sup>	pu	pu	pu	pu
C2	Hexanoic acid	nd	42.01 <sup>b</sup>	$20.93^{a}$	45.09 <sup>b</sup>	44.31 <sup>b</sup>	41.33 <sup>b</sup>	36.60 <sup>b</sup>	21.96 <sup>a</sup>	$24.70^{a}$	$36.48^{\rm b}$	$34.94^{b}$
C3	Octanoic acid	322.06 <sup>de</sup>	159.02 <sup>ab</sup>	202.37 <sup>abc</sup>	335.55 <sup>de</sup>	309.63 <sup>cde</sup>	392.01 <sup>de</sup>	88.27 <sup>a</sup>	128.79 <sup>a</sup>	115.91 <sup>a</sup>	271.56 <sup>bcd</sup>	$409.87^{e}$
C4	Nonanoic acid	24.35 <sup>a</sup>	pu	pu	nd	7.56 <sup>b</sup>	$6.87^{\rm b}$	$6.09^{\mathrm{b}}$	5.58 <sup>b</sup>	6.58 <sup>b</sup>	6.59 <sup>b</sup>	5.53 <sup>b</sup>
C5	Decanoic acid	$91.69^{b}$	pu	75.34 <sup>b</sup>	97.00 <sup>b</sup>	93.78 <sup>b</sup>	126.67°	pu	40.76 <sup>a</sup>	$34.71^{a}$	71.69 <sup>b</sup>	101.41 <sup>bc</sup>
C6	9-Decenoic acid	nd	pu	11.19 <sup>ab</sup>	pu	nd	13.88 <sup>bc</sup>	pu	14.47 <sup>c</sup>	pu	$9.66^{a}$	pu
C7	Dodecanoic acid	$7.74^{bcd}$	$6.42^{abc}$	9.41 <sup>cd</sup>	$8.86^{bcd}$	$10.29^{d}$	8.13 <sup>bcd</sup>	6.64 <sup>abc</sup>	pu	5.75 <sup>ab</sup>	7.34 <sup>abcd</sup>	$4.10^{a}$
C8	Benzoic acid	pu	pu	pu	pu	pu	pu	pu	pu	$1.69^{a}$	pu	pu
C9	Tetradecanoic acid	nd	pu	pu	nd	nd	$6.08^{b}$	$4.77^{a}$	$5.26^{ab}$	pu	pu	pu
	Total	454.93 <sup>cde</sup>	215.19 <sup>ab</sup>	325.74 <sup>bc</sup>	495.43 <sup>def</sup>	476.13 <sup>def</sup>	$604.23^{f}$	153.61 <sup>a</sup>	216.81 <sup>ab</sup>	189.34 <sup>ab</sup>	403.31 <sup>cd</sup>	555.85 <sup>ef</sup>
Aldehy	ydes											
L1	Decanal	$12.46^{a}$	pu	$15.08^{a}$	15.79 <sup>a</sup>	pu	pu	pu	12.26 <sup>a</sup>	nd	pu	14.85 <sup>a</sup>
L2	Pentadecanal	nd	nd	pu	nd	5.41 <sup>a</sup>	pu	pu	pu	nd	pu	pu
L3	3,4-	$21.38^{a}$	20.05 <sup>a</sup>	pu	$36.86^{\circ}$	$23.31^{\mathrm{ab}}$	$23.98^{ab}$	$24.00^{\mathrm{ab}}$	25.98 <sup>abc</sup>	27.89 <sup>abc</sup>	$31.28^{\rm abc}$	34.56 <sup>bc</sup>
	Dimethylbenzaldenyde		40-0-0	0			odana	oquo o		odooo — e	odo	
	Total	33.84 <sup>w</sup>	20.05	15.08ª	52.65	28.72 <sup>auc</sup>	23.98	24.00 <sup>404</sup>	38.23 <sup>ul</sup>	27.89 <sup>auc</sup>	31.28	49.41 <sup>uc</sup>

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Table (	1 (continued)											
No.	Compounds	Pure			S. cerevisiae	and Saccha-	Saccharom	yces and non	-Saccharomy	ces		
					romyces SIL	nultaneous	Simultaneo	ns		Sequential		
		Y2	Y5	Y6	Y2 + Y5	Y2+Y6	Y2 + Y7	Y5+Y7	Y6+Y7	Y2 + Y7	Y5+Y7	Y6+Y7
Ketone	Se											
<b>K</b> 1	Acetoin	nd	$3.14^{a}$	pu	pu	nd	nd	nd	nd	nd	nd	nd
K2	Methyl isohexenyl ketone	nd	pu	pu	pu	nd	nd	pu	$2.27^{\mathrm{b}}$	$2.27^{\mathrm{b}}$	$2.38^{\mathrm{b}}$	pu
K3	Geranyl acetane	$12.99^{a}$	$18.32^{ab}$	23.44 <sup>abc</sup>	43.61 <sup>d</sup>	32.84 <sup>bcd</sup>	$22.61^{\text{abc}}$	$31.92^{bcd}$	$40.90^{d}$	41.25 <sup>d</sup>	37.88 <sup>cd</sup>	62.57 <sup>e</sup>
	Total	$12.99^{a}$	21.46 <sup>ab</sup>	23.44 <sup>ab</sup>	43.61 <sup>c</sup>	32.84 <sup>bc</sup>	22.61 <sup>ab</sup>	$31.92^{bc}$	43.17 <sup>c</sup>	43.52 <sup>c</sup>	$40.26^{\circ}$	62.57 <sup>d</sup>
Pheno	ls											
P1	2,6-Di- <i>tert</i> -butyl-4-sec-butylphenol	16.15 <sup>a</sup>	nd	pu	nd	nd	nd	nd	nd	nd	nd	pu
P2	Phenol, 2, 4-bis (1, 1-dimethylethyl)-6-nitro-	nd	pu	pu	pu	nd	nd	nd	nd	nd	5.55 <sup>a</sup>	pu
P3	4-Vinyl guaiacol	$13.63^{\rm bc}$	$6.58^{a}$	pu	17.47 <sup>cd</sup>	14.57 <sup>cd</sup>	$24.02^{e}$	20.37 <sup>de</sup>	pu	$7.72^{ab}$	$12.98^{abc}$	$31.63^{f}$
P4	2,4-Di-tert-butylphenol	509.43 <sup>bc</sup>	$493.80^{\mathrm{bc}}$	pu	$443.06^{\mathrm{bc}}$	$563.20^{\circ}$	$445.91^{\text{bc}}$	468.65 <sup>bc</sup>	pu	387.17 <sup>ab</sup>	$420.59^{ab}$	$311.56^{a}$
	Total	$539.20^{\circ}$	$500.37^{bc}$	453.62 <sup>abc</sup>	460.53 <sup>abc</sup>	577.78°	469.93 <sup>abc</sup>	489.02 <sup>bc</sup>	485.16 <sup>bc</sup>	394.89 <sup>ab</sup>	439.11 <sup>abc</sup>	$343.20^{a}$
Terper	les											
T1	Linalool	nd	$10.80^{\circ}$	$8.88^{a}$	$9.37^{ab}$	$10.20^{bc}$	$9.40^{ab}$	$9.88^{\rm abc}$	pu	nd	pu	pu
T2	Terpinen-4-ol	$47.13^{a}$	$46.93^{a}$	44.62 <sup>a</sup>	$46.58^{a}$	$50.37^{a}$	$39.59^{a}$	41.29 <sup>a</sup>	43.98 <sup>a</sup>	$39.79^{a}$	$40.77^{\mathrm{a}}$	47.62 <sup>a</sup>
T3	α-Terpineol	5.01 <sup>a</sup>	$4.85^{a}$	$5.16^{a}$	5.64 <sup>a</sup>	$6.28^{a}$	pu	nd	nd	nd	nd	pu
T4	Citronellol	pu	$11.98^{c}$	$10.93^{\rm bc}$	$9.02^{ab}$	$10.89^{bc}$	7.06 <sup>a</sup>	11.12 <sup>bc</sup>	$11.95^{\circ}$	$9.70^{\mathrm{bc}}$	$6.94^{a}$	8.61 <sup>ab</sup>
T5	Cedrol	$3.09^{a}$		pu	pu	nd	pu	pu	pu	nd	pu	pu
T6	(E, E)-Farnesol	15.82 <sup>b</sup>	$21.20^{a}$	pu	nd	nd	pu	15.05 <sup>b</sup>	nd	12.67 <sup>b</sup>	nd	pu
	Total	71.05 <sup>ab</sup>	95.75 <sup>c</sup>	69.59 <sup>ab</sup>	70.61 <sup>ab</sup>	77.74 <sup>b</sup>	$56.06^{a}$	77.35 <sup>b</sup>	65.45 <sup>ab</sup>	$71.24^{ab}$	$57.28^{a}$	56.23 <sup>a</sup>
	Total	$3070.36^{b}$	$3621.04^{\circ}$	3040.51 <sup>b</sup>	3033.20 <sup>b</sup>	3133.77 <sup>b</sup>	3070.16 <sup>b</sup>	2923.66 <sup>ab</sup>	2694.19 <sup>ab</sup>	2483.07 <sup>a</sup>	2751.97 <sup>ab</sup>	2851.66 <sup>ab</sup>

Data are the means (n=3)*nd* not detected

Values in the same row with different letter indicate significant difference (P < 0.05) according to Duncan's tests

species were 6, 9, and 5, respectively, in Y2 + Y7, Y5 + Y7and Y6 + Y7, which were fermented by sequential inoculated pattern. And then, 9,12-octadecadienoic acid ethyl ester, diisobutyl phthalate, and tetradecanoic acid were unexamined in sequential inoculated samples, while 9-decenoic acid and methyl hexadecanoate were unexamined, and diethyl succinate was newly analyzed in Y2 + Y7 and Y6 + Y7, acetic acid, dibutyl phthalate, and linalool were undetectable, diethyl succinate and methyl isohexenyl ketone were newly examined in Y2 + Y7 and Y5 + Y7's.

The volatiles quality of Y5 was slightly higher than the other two strain's. Phenethyl alcohol, 2,4-di-tert-butylphenol were dominant. When Y2 co-cultured with Y5 and Y6, the volatiles content was almost unchanged, but lower than Y5's. The abundance of phenethyl alcohol also decreased than Y5's while was a little increased than the other two respective strains. When Saccharomyces co-cultured with Y7, the volatiles content in Y5 + Y7 and Y6 + Y7 decreased. The contents of volatiles in Y2 + Y7 and Y5 + Y7, both were inoculated sequentially, were lower than the samples inoculated simultaneously. As shown in Fig. 5, dominant groups involved in alcohols, esters, acids, and phenols, accounted for 94.10% to 96.66%. The results suggested that the proportion of four dominant volatiles class depended on the interactive relationship between two strains and their characteristics. When Y2 co-cultured with Y5 or Y6, the result was different, the proportion of esters and phenol in the former sample was closer to Y2's, which was closer to that of Y6's; although, the proportion of acids increased in both co-cultured samples. Due to the simultaneous inoculated sample for Y2 co-cultured with Y7, the proportion of esters and acids increased compared with the respective strains.



Fig. 5 Difference of volatile profile in kiwi wine fermented by pure and co-cultured among *Saccharomyces*, as well as *Saccharomyces* with *T. delbrueckii* 

However, sequential inoculated sample was the opposite, especially, the proportion of acids induced to 7.63%. When Y5 or Y6 was co-cultured with Y7, respectively, the proportion of esters, acids, and phenols in the sequential inoculated samples increased, while the proportion of esters in the simultaneous inoculated samples decreased. It was reported that the contents of esters were increased by co-culture [33].

By coculturing, among Saccharomyces strains, the dominant components (abundance  $\geq 0.05$ ) have five of the same components which were ethyl octanoate, isoamyl alcohol, phenethyl alcohol, octanoic acid and 2,4-di-tert-butylphenol. Besides, Y5 + Y7 still have phenethyl acetate. Of them, the proportion of phenethyl alcohol accounted for 25.02% to 50.21%. Compared with pure culture, the content of these components was not changed significantly. Among them, the proportion of phenethyl alcohol increased, but the proportion of octanoic acid and ethyl octanoate decreased notably. For Y5, the proportion of phenethyl alcohol and phenethyl acetate decreased and others increased, resulting in the sum proportion of decrease of six dominant components. In addition, the sum proportion of five dominant components significantly increased by forming of 2,4-di-tert-butyphenol newly although ones of phenethyl alcohol decreased in sequential inoculated samples. The dominant components species were the same among these samples fermented by co-cultured among Saccharomyces strains, and their proportion was similar to Y2's. These results were consistent with the previous document on brewing Riesling wines by coculturing of yeast strain [34]. The content of phenylethyl alcohol in the wines sequentially inoculated with S. cerevisiae and T. delbrueckii was higher than that of the control [35]. Nonconventional yeast has been reported to have a desirable and negative impact on flavor profile [36]. In addition to simultaneous (Y2 + Y7) inoculation, the ketone content increased significantly after co-culture, and the improvement in ketone was mainly due to the increase of geranyl acetane which contribute the fruity and fresh aroma to the flavor profile of the wine. Besides, co-culture has little effect on the content of phenols and terpenes.

#### Characterizing the flavor of the co-cultured samples

The detected volatile compounds were used for PLS-DA analysis to better understand the relationship between different inoculation modes of kiwi wine and corresponding flavor compounds, which 33.40% of the total variance was explained. As the results showed in Fig. 6, kiwi wines were distributed in different locations depending on the pattern of inoculation. Heptyl alcohol, decyl alcohol, (E)-nerolidol, 1-hexadecanol, dodecanol, isobutanol, phenylethyl alcohol in alcohol were closely related to single fermentation. Meanwhile, fatty acids of hexanoic acid, octanoic acid, decanoic acid, and dodecanoic acid, ethyl esters of ethyl decanoate,



Fig. 6 Biplot for PLS-DA of aroma compounds in kiwi wine fermented by pure and co-cultured among *Saccharomyces*, as well as *Saccharomyces* with *T. delbrueckii*. Aroma compounds used for analysis were listed in Table 1

ethyl octanoate, ethyl lactate, and ethyl 2-furoate have a greater contribution to co-culture among *Saccharomyces*. Ethyl esters of ethyl 9-decenoate, ethyl dodecanoate, ethyl hexadecanoate, ethyl oleate, 9,12-octadecadienoic acid ethyl ester and ethyl phenylacetate, acetate esters of isoamyl acetate, hexyl acetate and phenylethyl acetate, ketones of geranyl acetane and methyl isohexenyl ketone, acids of 9-decenoic acid, tetradecanoic acid and benzoic acid have a greater contribution to the flavor of co-culture with *T. delbrueckii*.

As shown in Fig. 7, heatmap analysis of volatiles indicated that Y2, Y5, and simultaneous inoculated co-culture of Y5 and Y7 were grouped into a cluster due to higher concentrations of heptanol, 1-dodecanol, (E)-nerolidol, (E, E)-farnesol, and acetic acid. Another cluster which were composed of single strains Y6, sequence inoculated coculture samples as well as the rest simultaneous inoculated co-culture samples of them, Y6, Y2 co-cultured with Y5 or Y6, were grouped into one sub cluster since these sample, the concentrations of  $\alpha$ -terpineol, terpinen-4-ol, linalool,



Fig. 7 Heatmap represents for volatile profile of kiwi wine fermented by pure and co-cultured among *Saccharomyces*, as well as *Saccharomyces* with *T. delbrueckii*. Aroma compounds used for heatmap analysis were listed in Table 1

**Fig. 8** OAV profile analysis of kiwi wine fermented by pure and co-cultured among *Saccharomyces*, as well as *Saccharomyces* with *T. delbrueckii*. The OAV profile was expressed as the log of OAV from main volatiles (OAV > 1)



and citronellol were higher. When Y7 co-cultured with Y2 by simultaneous inoculated, as well as co-cultured Y5 or Y6 by sequence inoculated, which have a higher content of isoamyl acetate, ethyl hexanoate, ethyl octanoate, ethyl decanoate, ethyl 9-decenoate and 4-VG, so that fell into the same sub cluster. In addition, the rest of the samples inoculated were grouped in another sub cluster. As shown in Fig. 8, there were six volatile compounds of OAV > 1 in kiwi wine, including ethyl octanoate, decanal, terpinen-4-ol, ethyl hexanoate, linalool, and phenylethyl acetate. The difference of component' OAV resulted in very different contributing to kiwi wine flavor feature. The co-culture of different strains among Saccharomyces endowed to orange peel and menthol aroma since OAV's values of decanal (12.64) and terpinen-4-ol (10.07) were higher. Different strain of Saccharomyces co-cultured with T. delbrueckii endued a fruity and floral aroma as that of ethyl octanoate (184.65) and ethyl hexanoate (8.15) were higher.

# Conclusions

The contribution of yeast to physicochemical properties and flavor of kiwi wine varied with the strains, either species or family. It was mainly changed by co-culturing among strains that the flavor characteristics of kiwi wine, while no significant difference of their physiochemical properties was observed in present research. Co-culture pattern not only affected the volatile content, but also affected identified species and the flavor profile. The volatile content in these samples fermented by co-culture was slightly decreased compared with that in the respective single ferment samples. It was interesting that the contribution of unique aroma component to kiwi wine was enhanced by co-culture fermentation. The OAV of ethyl octanoate and ethyl hexanoate was increased by co-culturing strains in Saccharomyces with T. delbrueckii, while that of decanal and terpinen-4-ol was enhanced by coculturing of strains among Saccharomyces. Meanwhile, the effect was affected by the inoculation sequence. Results suggested that it was an excepting process to obtain high quality of kiwi wine by co-culturing technology of yeasts, and was very effective to optimize the process by polyphase analysis approaches.

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#### **Compliance with ethical standards**

**Conflict of interest** The authors declare that they have no conflict of interests.

**Compliance with ethics requirements** This article does not contain any studies with human or animal subjects.

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