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Cold pasteurisation of red wines with high hydrostatic pressure to control *Dekkera/Brettanomyces*: effect on both aromatic and chromatic quality of wine

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Abstract This paper reports the effect of high hydrostatic pressure (HHP) on wines contaminated with *Dekkeral Brettanomyces* populations of 10^4 and 10^6 cfu mL⁻¹ growing at either pH 3.2 or 3.6 (both normal pHs for red wine) and at room temperature (25 °C). HHP (100 MPa for 24 h) was highly effective at controlling the growth of all combinations of starting yeast population and pH, yet caused scant modification of thermosensitive wine molecules such as pigments and volatile compounds (molecules that greatly influence wine quality). The present results support the potential use of HHP as a means of cold-pasteurising wines to control *Dekkera/Brettanomyces*.

Keywords Brettanomyces · Dekkera · High hydrostatic pressures · Ethylphenols · GC–FID · GC–MS · HPLC–DAD–ESI/MS · Red wines

Introduction

The yeast genera *Dekkera* and *Brettanomyces* (in fact sporulating and non-sporulating forms of the same organism) cause problems for winemakers owing to their ability to transform grape hydroxycinnamic acids (HCAs) into vinylphenols, and then the latter into ethylphenols. The metabolic pathway leading to the formation of ethylphenols involves the enzymes hydroxycinnamate decarboxylase [10] and vinylphenol reductase [11, 29]. *Dekkera*

members catabolise ferulic, caffeic and *p*-coumaric acids and possess inducible enzymes with similar pH and temperature optima [12]. *Dekkera/Brettanomyces* can transform some 90 % of HCA into ethylphenols [8] under oenological conditions, especially those encountered during ageing. For a review of the main physiological, morphological and metabolic characteristics of *Dekkera/ Brettanomyces*, see [28].

Ethylphenols are molecules with low sensorial thresholds; for example, 4-ethylphenol has a sensorial threshold of just 230 µg L⁻¹ [4]. Unfortunately, they have a negative impact on wine quality. The perception of a 'Brett character' (i.e. 'animal', 'old leather' or 'stable' aromas that reduce the value of aged wines), however, depends on the type of grape used. Higher sensorial thresholds have been reported in monovarietal Cabernet Sauvignon wines than Tempranillo wines [24]. Other authors report wine 4-ethylphenol concentrations of 3.0, 1.74 and 0.68 mg L⁻¹ to be associated with strong, medium and absent Brett character, respectively [17]. Although wineries do not like to admit *Dekkera/Brettanomyces* problems, the presence of these yeasts (or their ethylphenol product) has been detected in the cellars of many winemaking regions [2, 23, 26].

Dekkera/Brettanomyces can be controlled by flash pasteurisation and membrane filtration [3], but these techniques also have negative impacts on the sensorial quality of wines. The addition of SO₂ can also control these yeasts [9], but it is difficult to guarantees a stable molecular SO₂ concentration in wines ageing in wooden barrels. Other additives, such as dimethyl dicarbonate [6], are also effective, but deciding on an adequate dose can be complicated.

The food industry is currently tending towards the use of non-thermal food preservation techniques. One of these, subjection to high hydrostatic pressure (HHP), can be

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regarded as a form of cold pasteurisation. Pressures of 100–700 MPa are commonly used, which damage the cell walls and membranes of spoilage organisms [20]. High hydrostatic pressure has the advantage that it produces fewer sensorial modifications to foods than do physical techniques such as conventional pasteurisation or filtration. Neither does it have any of the disadvantages of adding chemical agents. Subjecting wines to high pressures has been proposed as a means of microbiologically and biochemically stabilising wines [7, 25] with minimum sensorial repercussions [19].

The aim of the present work was to test the use of HHP as a means of controlling *Dekkera/Brettanomyces* in red wines and to determine the effects of such treatment on wine colour and aromatic profile.

Materials and methods

Yeast strain used and the preparation of contaminated wines

The wine used was DO *Mancha*, which is made from Tempranillo, Merlot and Syrah grapes (*V. vinifera* L.) (alcohol content, 13 % v/v; pH, 3.6; total SO₂, 35 mg L⁻¹; free SO₂, 11 mg L⁻¹). To provide a wine sample with a second pH, a fraction of the original was acidified with tartaric acid (Panreac, Barcelona, Spain) to pH 3.2.

The spoilage yeast used was *Dekkera bruxellensis* D37 (a member of the *Instituto de Fermentaciones Industriales* collection, CSIC, Madrid, Spain), regarded here as a model member of *Dekkera/Brettanomyces*.

Wines with *D. bruxellensis* D37 populations of either 10^6 or 10^4 cfu mL⁻¹ were prepared by adding, to 50 mL of wine, either 312 or 3 µL of a yeast suspension (population 1.6×10^8 cfu mL⁻¹) prepared in YEPD medium [15]. Four types of contaminated wines were finally produced, that is, with populations of either 10^6 or 10^4 cfu mL⁻¹ and

at pH 3.2 or 3.6. Non-contaminated wine samples were also prepared at both pHs.

High hydrostatic pressure

The contaminated wines were pressurised to eliminate *D. bruxellensis* D37 cells using a TFS-2L HHP apparatus (Mayekawa, Tokyo, Japan) (treatment vessel volume 2 L; pressurisation range 0–100 MPa). This was performed at the maximum pressure produced by this apparatus (100 MPa) for 24 h at 25 °C in Falcon tubes in a vacuum-sealed plastic bag. All four contaminated wines (i.e. both population sizes at both pHs) were subjected to HHP. Control wines with populations of 10^6 cfu mL⁻¹ at pH 3.2 and 3.6, not subjected to HHP, were used to confirm normal yeast development. All assays performed in triplicate.

Non-contaminated wine at pH 3.6 was also subjected to HHP at 100 MPa to determine whether this treatment had any effect on wine sensorial features.

Selective/differential media for D. bruxellensis D37 detection

The growth of *D. bruxellensis* D37 in the wines was confirmed using DBDM selective/differential medium [27].

Determination of anthocyanins

The following anthocyanins and pyranoanthocyanins were analysed using an Agilent Technologies (Palo Alto, CA) HPLC chromatograph series 1100, equipped with a diode array detector and a quadrupole mass spectrometer with an electrospray interface: delphinidin-3-O-glucoside (D3G), petunidin-3-O-glucoside (Pt3G), peonidin-3-O-glucoside (Pn3G), malvidin-3-O-glucoside (M3G), malvidin-3-O-glucoside (Pn3G), malvidin-3-O-glucoside (M3G), malvidin-3-O-glucoside -pyruvate adduct (vitisin A; VITA), malvidin-3-O-(6"-*p*-acetylglucoside)-pyruvate adduct (CmVITA), malvidin-3-O-(6"-*p*-coumaroylglucoside)-pyruvate adduct (CmVITA),

Table 1 Growth and 4-ethylphenol content of the differently prepared and treated wines

Treatment (performed in triplicate)	Population size (cfu mL ⁻¹)	pН	Pressure (MPa; 24 h)	Growth (cfu mL^{-1} after 30 days on DBDM medium)	4-ethylphenol production $(\mu g \ L^{-1})$
3.2 C	_	3.2	_	-	0.0 ± 0.0
3.6 C	_	3.6	_	_	0.0 ± 0.0
3.2 C 10 ⁶ DB	10 ⁶	3.2	_	$>10^{4}$	405.0 ± 110.0
3.6 C 10 ⁶ DB	10^{6}	3.6	-	>10 ⁴	682.0 ± 55.0
3.2 10 ⁶ DB	10^{6}	3.2	100	-	0.0 ± 0.0
3.6 10 ⁶ DB	10^{6}	3.6	100	-	0.0 ± 0.0
3.2 10 ⁴ DB	10^{4}	3.2	100	-	0.0 ± 0.0
3.6 10 ⁴ DB	10^{4}	3.6	100	-	0.0 ± 0.0

Values for 4-ethylphenols are means \pm SD for three replicates. C control, DB contaminated with D. bruxellensis D37

petunidin-3-O-(6"-acetylglucoside) (Pt3G6Ac), peonidin-3-O-(6"-acetylglucoside) (Pn3G6Ac), malvidin-3-O-(6"-acetylglucoside) (M3G6Ac), peonidin -3-O-(6"-p-coumaroylglucoside) (Pn3GCm), malvidin-3-O-(6"-p-coumaroylglucoside) (M3 GCm), malvidin-3-O-glucoside-4-vinylcatechol (M3GVCath), malvidin-3-O-glucoside-4-vinylepicatequine (M3GVEpiCat), malvidin-3-O-glucoside-4-vinylphenol (M3GVPh), malvidin-3-O-glucoside-4-vinvlguaiacol (M3GVG) and malvidin-3-O-(6"-p-coumaroylglucoside)-vinylphenol (M3GCmVPh). Gradients of solvent A (water/formic acid, 90:10, v/v) and B (methanol/formic acid, 90:10, v/v) were used in a reversephase Kinetex C18 column (Phenomenex, Torrance, CA, USA) (100 \times 4.6 mm; particle size 2.6 µm) as follows: 10 % B (0.8 ml min⁻¹) from 0 to 3 min, 10–50 % B linear $(0.8 \text{ ml min}^{-1})$ from 3 to 18 min, 50 % B from 18 to 20 min, 50–10 % B linear (0.8 ml min⁻¹) from 20 to 21 min, and reequilibration of the column from 21 to 23 min. Detection was performed by scanning in the 500-600 nm range. Quantification was performed by comparison against an external standard at 525 nm and expressed as a function of the concentration of malvidin-3-O-glucoside (Extrasynthèse, Genay, France). The different anthocyanins were identified by their retention times with respect to the majority anthocyanin malvidin-3-O-glucoside, and by comparing the UV-visible and mass spectra with data in the literature [13, 22]. The ESI parameters were drying gas (N2) flow, 10 ml min⁻¹; temperature, 350 °C; nebuliser pressure, 380 Pa (55 psi); and capillary voltage, 4,000 V. Mass spectrometry was performed in positive mode scanning, from 100 to m/z 1,000, using as a fragmenter voltage of 150 V from 0 to 23 min. One-hundred-microlitre samples of previously filtered (0.45-µm membrane filters made of cellulose methylic esters) (Tecknokroma, Barcelona, Spain) wines were injected into the HPLC apparatus. The detection limit was 0.1 mg L^{-1} .

Analysis of volatile compounds by gas chromatography with flame ionisation detection

Volatile compounds were determined using an Agilent Technologies 6850 gas chromatograph (Network GC System) equipped with an integrated flame ionisation detector, as described by [1]. A DB-624 column ($60 \text{ m} \times 250 \text{ }\mu\text{m} \times 1.40 \text{ }\mu\text{m}$) was employed, calibrated with the following compounds as external standards: acetaldehyde, methanol, 1-propanol, 1-butanol, 2-butanol, isobutanol, 2-methyl-1-butanol, 3-methyl-1-butanol, 2-phenylethyl acetate, 2-phenylethyl alcohol, diacetyl, ethyl acetate, isoamyl acetate, isobutyl acetate, ethyl butyrate, ethyl lactate and hexanol. 4-Methyl-2-pentanol was used as an internal standard (all compounds from Fluka, Sigma–Aldrich Corp., Buchs SG, Switzerland). The injector temperature was 250 °C, and the detector temperature was 300 °C. The column temperature

Treatment	Hq	Pressure (MPa)	D3G	Pt3G	Pn3G	M3G	Pt3GAc	Pn3GAc	M3GAc	Pn3GCm	M3GCm	Totals
3.2 C	3.2	I	$1.2\pm0.04^{ m d}$	$1.9\pm0.06^{\mathrm{c}}$	$1.0\pm0.07^{ m d}$	$10.2 \pm 0.10^{\mathrm{e}}$	0.3 ± 0.03^{c}	$0.5\pm0.02^{ m c}$	$1.4\pm0.08^{ m d}$	$0.3\pm0.01^{ m cd}$	$0.9\pm0.01^{ m bc}$	$23.8\pm0.35^{\mathrm{e}}$
3.6 C	3.6	I	$1.1\pm0.09^{\mathrm{c}}$	$1.7\pm0.12^{ m b}$	$0.9\pm0.02^{ m dc}$	$9.5\pm0.15^{ m d}$	$0.3\pm0.01^{ m bc}$	$0.5\pm0.01^{ m b}$	$1.3\pm0.07^{ m c}$	$0.3\pm0.03^{ m d}$	$1.0\pm0.12^{ m c}$	$22.2\pm0.28^{ m d}$
3.6P C	3.6	100	$0.8\pm0.11^{\rm a}$	$1.4\pm0.05^{\mathrm{a}}$	$0.8\pm0.01^{\rm a}$	$8.4\pm0.15^{\rm b}$	$0.3\pm0.02^{\mathrm{ab}}$	$0.4\pm0.02^{\mathrm{a}}$	$1.2\pm0.03^{ m b}$	$0.3\pm0.01^{ m ab}$	$0.8\pm0.01^{\rm ab}$	$20.4\pm0.21^{\mathrm{b}}$
3.2.10 ⁶ DB	3.2	100	$1.0\pm0.06^{ m b}$	$1.6\pm0.07^{ m b}$	$0.9\pm0.01^{ m b}$	$8.6\pm0.01^{\rm bc}$	$0.3\pm0.01^{\rm ab}$	$0.5\pm0.02^{ m b}$	$1.2\pm0.01^{ m b}$	$0.3\pm0.01^{ m abc}$	$0.8\pm0.00^{\rm ab}$	$21.3\pm0.03^{\rm c}$
3.6.10 ⁶ DB	3.6	100	$0.8\pm0.06^{\mathrm{a}}$	$1.3 \pm 0.11^{\mathrm{a}}$	$0.8\pm0.03^{\mathrm{a}}$	$7.7\pm0.54^{\mathrm{a}}$	$0.3\pm0.02^{\mathrm{a}}$	$0.4\pm0.02^{\mathrm{a}}$	1.0 ± 0.06^{a}	$0.3 \pm 0.01^{\mathrm{a}}$	$0.7\pm0.04^{\mathrm{a}}$	$19.0\pm1.01^{\mathrm{a}}$

Table 2 Anthocyanin contents (mg L^{-1}) of the differently prepared and treated wines

 21.9 ± 0.07 ^{cd}

 $0.9 \pm 0.03^{\rm b}$

 $0.3 \pm 0.01^{
m bcd}$

 $1.3 \pm 0.02^{\rm c}$

 $0.5\pm0.01^{
m bc}$

 $\pm 0.01^{ab}$

0.3

 $8.8\pm0.04^{\circ}$

 $0.9 \pm 0.01^{\circ}$

 $1.6\pm0.04^{
m b}$

 $1.0\pm0.01^{
m bc}$

100

3.2

3.2.10⁴ DB

bruxellensis D37

= contaminated with D.

Values are the means \pm SD for three replicates. Means with the same lower case letter in the same column are not significantly different (p < 0.05). C control, DB

was 40 °C for the first 5 min, rising linearly by 10 °C min⁻¹ until reaching 250 °C; this temperature was then maintained for 5 min. Hydrogen was used as the carrier gas; this was provided by a hydrogen generator (LNI Schmidlin SA, Geneva, Switzerland). The flow rate was 22.1 L min⁻¹, the injection split ratio was 1:10, and the detection limit was 0.1 mg L⁻¹. One hundred microlitres of internal standard (concentration 500 mg L⁻¹) were added to 1mL test samples and filtered through syringe membrane filters (0.45 μ m pore size) (Teknokroma, Barcelona, Spain). They were then placed in 2-mL glass vials sealed using a PTFE/silicon septum. One microlitre of this filtrate was injected into the GC apparatus.



Fig. 1 Principal components analyses for different samples treatments (pH, HHP, *Dekkera* inoculation and amount of cells inoculated) regarding with the monomeric anthocyanin concentration

Analysis of methylphenol by gas chromatography-mass spectrometry

Ethylphenol was analysed using an Agilent Technologies 6890 N-MSD-5973 N gas chromatography-mass spectrometer (GC-MS) in single-ion mode (SIM) detection after liquid extraction with dichloromethane (Merck, Hohenbrunn, Germany). Ten millilitres of wine was mixed with 1 mL of dichloromethane, 100 μ L of a 10 mg L⁻¹ solution of 3.4dimethylphenol (Merck, Hohenbrunn, Germany) as an internal standard, and 1.5 g of NaCl to produce a salting out effect. The mixture was vortex-stirred for 5 min and then centrifuged at 7,000 rpm for 10 min at 4 °C. The dichloromethane phase was filtered through 0.45-um syringe membrane filters, and 1 uL injected into the GC-MS apparatus. A DB-WAX column (30 m \times 0.25 mm internal diameter ×0.25 µm film thickness) (J & W Scientific, Folsom, CA, USA) was used for chromatographic separation; the split ratio used was 20:1. The oven temperature programme was 60 °C for 1 min, followed by a 3 °C min⁻¹ ramp until 150 °C, followed by a 10 °C min⁻¹ ramp to 210 °C, which was maintained for 2 min. The helium flow rate was 1 mL min $^{-1}$. Detection was performed using SIM mode MS (target ion for 4-ethylphenol = 107). Calibration patterns were produced using reference compounds (Alfa Aesar, Karlsruhe, Germany).

Statistical analysis

Means and standard deviations were calculated, and ANOVA and least significant difference (LSD) tests performed using PC Statgraphics v.5 software (Graphics Software Systems, Rockville, MD, USA). Significance was set at p < 0.05 for the ANOVA matrix F value. The LSD test was used to detect significant differences between

Table 3 Pyranoanthocyanin and dimer contents (mg L^{-1}) of the differently prepared and treated wines

Treatment	pН	Pressure (MPa)	VITA	AcVITA	CmVITA	M3GvinCat	M3GvinEpic	M3GvinPhe	M3GvGua	M3GCm4vPh
3.2 C	3.2	_	$1.9\pm0.0^{\rm d}$	$0.6\pm0.0^{\mathrm{a}}$	$0.6\pm0.0^{\rm b}$	$0.6\pm0.0^{\rm b}$	$0.7\pm0.0^{\mathrm{b}}$	$0.6\pm0.0^{\mathrm{b}}$	0.3 ± 0.0^{a}	$0.3\pm0.0^{\rm a}$
3.6 C	3.6	-	1.8 ± 0.1 cd	$0.7\pm0.0^{\rm b}$	$0.6\pm0.0^{\rm b}$	0.6 ± 0.0^{ab}	$0.7\pm0.0^{\mathrm{b}}$	$0.6\pm0.0^{\rm b}$	0.3 ± 0.0^a	0.3 ± 0.0^{a}
3.6P C	3.6	100	1.8 ± 0.1^{ab}	0.6 ± 0.0^a	$0.6\pm0.0^{\rm b}$	$0.6\pm0.0^{\rm b}$	$0.7\pm0.0^{\mathrm{b}}$	$0.6\pm0.0^{\rm b}$	0.3 ± 0.0^a	0.3 ± 0.0^{a}
3.2.10 ⁶ DB	3.2	100	$1.8\pm0.0^{\mathrm{bc}}$	$0.6\pm0.0^{\mathrm{a}}$	$0.6\pm0.0^{\mathrm{b}}$	0.6 ± 0.0^{b}	$0.7\pm0.0^{\mathrm{b}}$	0.6 ± 0.0^{b}	$0.3\pm0.0^{\mathrm{a}}$	0.3 ± 0.0^a
3.6.10 ⁶ DB	3.6	100	$1.7 \pm 0.1^{\mathrm{a}}$	$0.6\pm0.0^{\mathrm{a}}$	0.5 ± 0.0^{a}	0.5 ± 0.0^a	0.6 ± 0.0^{a}	0.5 ± 0.0^a	$0.3\pm0.0^{\mathrm{a}}$	0.3 ± 0.0^a
3.2.10 ⁴ DB	3.2	100	$1.8\pm0.0^{\mathrm{bc}}$	0.6 ± 0.0^{a}	$0.6\pm0.0^{\mathrm{b}}$	$0.6 \pm 0.0^{\mathrm{b}}$	$0.7\pm0.0^{\mathrm{b}}$	$0.6\pm0.0^{\mathrm{b}}$	0.3 ± 0.0^{a}	0.3 ± 0.0^a

The values are the means \pm SD of 3 replicates. Means with the same lower case letter in the same column are not significantly different (p < 0.05). C control, DB = contaminated with D. bruxellensis D37

Table 4 Voi	latile m	olecule content (m	g L^{-1}) of the diff.	erently prepared an	d treated wines						
Treatment	Hq	Pressure (MPa)	Acetaldehyde	Methanol	1-propanol	Diacetyl	Ethyl acetate	2-butanol	Isobutanol	1-butanol	2 m-1-butanol
3.2 C	3.2	I	$5.7\pm4.0^{\mathrm{a}}$	$193.5 \pm 2.2^{\rm bc}$	$47.4 \pm 1.2^{\rm b}$	$0.7\pm0.0^{ m ab}$	64.1 ± 2.2^{bc}	1.3 ± 0.6^{a}	55.0 ± 1.5^{bcd}	$0.7\pm0.1^{ m b}$	283.2 ± 11.6^{cde}
3.6 C	3.6	I	$9.4\pm0.6^{ m b}$	$192.9 \pm 3.6^{\mathrm{bc}}$	$46.6\pm0.9^{\mathrm{b}}$	$0.7\pm0.1^{\mathrm{ab}}$	$62.5\pm1.3^{ m bc}$	$1.0\pm0.1^{ m a}$	$53.9\pm1.1^{ m b}$	$0.5\pm0.1^{\mathrm{a}}$	$272.4 \pm 5.3^{\mathrm{bc}}$
3.2P C*	3.2	100	11.1	200.9	47.4	0.7	62.8	0.9	54.4	0.6	276.3
3.6P C	3.6	100	$10.3\pm0.3^{ m b}$	$195.9\pm5.3~\mathrm{cd}$	$47.4\pm0.7^{ m b}$	$0.7\pm0.1^{\mathrm{ab}}$	$63.4\pm0.9^{ m bc}$	1.0 ± 0.1^{a}	$54.9\pm0.5^{ m bc}$	$0.9 \pm 0.0^{\rm c}$	$276.4 \pm 3.0^{\mathrm{bcd}}$
3.2.10 ⁶ DB	3.2	100	$9.0\pm0.5^{ m b}$	$187.4\pm2.8^{\rm ab}$	$45.6\pm0.7^{\rm ab}$	$0.6\pm0.1^{\mathrm{a}}$	$61.5\pm1.5^{ m b}$	$1.0\pm0.1^{ m a}$	$53.2\pm0.8^{\mathrm{ab}}$	0.4 ± 0.0^{a}	267.8 ± 3.0^{ab}
3.6.10 ⁶ DB	3.6	100	$11.3 \pm 2.1^{\mathrm{b}}$	$185.4\pm1.5^{\rm a}$	$44.7\pm0.7^{\mathrm{a}}$	$0.6\pm0.1^{\mathrm{ab}}$	$58.7\pm0.8^{\mathrm{a}}$	1.0 ± 0.0^{a}	$51.6\pm0.9^{\mathrm{a}}$	0.4 ± 0.0^{a}	$261.0\pm3.7^{\mathrm{a}}$
3.2.10 ⁴ DB	3.2	100	$9.4\pm0.7^{ m b}$	$201.8 \pm 4.1^{\mathrm{d}}$	$49.2 \pm 1.6^{\rm c}$	$0.7\pm0.1^{\mathrm{ab}}$	$62.4\pm2.2^{\mathrm{bc}}$	1.1 ± 0.3^{a}	$56.2\pm2.0~\mathrm{cd}$	$0.7\pm0.1^{ m b}$	$286.3 \pm 7.1^{\mathrm{de}}$
3.6.10 ⁴ DB	3.6	100	$11.6\pm0.6^{\mathrm{b}}$	$209.9 \pm 6.9^{\mathrm{e}}$	$49.6\pm1.0^{\rm c}$	$0.8\pm0.2^{ m b}$	$64.9\pm0.5^{\mathrm{c}}$	1.0 ± 0.1^{a}	$57.1\pm0.8^{ m d}$	$0.8\pm0.1^{\rm bc}$	$289.0\pm5.0^{\mathrm{e}}$
Treatment	Ηd	Pressure (MPa)	3 m-1-butanol	Isobutyl acetate	Ethyl butyrate	Ethyl lactate	Isoamyl acet	ate Hexanol	2-phenyle	thanol 2-phe	nylethanol acetate
3.2 C	3.2	I	$70.7 \pm 1.0^{ m bc}$	I	I	204.0 ± 7.8^{a}	I	2.2 ± 0.3	$a 54.6 \pm 16$	8 ^{ab} –	
3.6 C	3.6	I	$71.9 \pm 0.6^{ m bcd}$	1	1	$202.9\pm2.9^{\mathrm{a}}$	I	4.3 ± 0.1	° 43.5 ± 3.	5 ^a –	
3.2P C*	3.2	100	74.8	1	1	210.7	I	2.6	42.7	I	
3.6P C	3.6	100	$72.0\pm1.4~\mathrm{cd}$	1	1	205.6 ± 6.0^{ab}	I	4.9 ± 0.5	c^{cd} 46.3 ± 1.	^{ab}	
3.2.10 ⁶ DB	3.2	100	$68.0\pm2.1^{\rm a}$	I	I	200.0 ± 4.1^{a}	I	5.5 ± 0.6	5^{d} 42.8 ± 1.0) ^a –	
3.6.10 ⁶ DB	3.6	100	$69.5\pm0.5^{\mathrm{ab}}$	1	1	$196.8\pm3.2^{\mathrm{a}}$	I	4.9 ± 0.5	$5 cd 56.0 \pm 1.2$	5 ^b -	
3.2.10 ⁴ DB	3.2	100	$74.2 \pm 1.8^{ m d}$	0.2 ± 0.2	I	$220.5 \pm 13.6^{\circ}$	I	3.5 ± 0.2	b^{b} 45.7 ± 3.0) ^{ab} –	
3.6.10 ⁴ DB	3.6	100	$76.9 \pm 1.6^{\mathrm{e}}$	I	I	$217.3\pm8.6^{\mathrm{bc}}$	Ι	2.5 ± 0.5	5^{a} 44.2 ± 2.1	5 ^a –	
Values are th D. bruxellens	ie meau vis D37	$ns \pm SD$ of three final of the final of the final of the field of th	ermentations. Mea not detected. Det	ns with the same list	ower case letter	in the same col	umn are not sig	nificantly differ	ent $(p < 0.05)$.	C control, DB	contaminated with

* Sample 3.2P C was run without replication

means. Multivariate statistical analyses of data were performed using principal components analysis (PCA).

Results and discussion

Viability of D. bruxellensis D37 following HHP treatment

Following HHP treatment, viable *D. bruxellensis* D37 cells were sought by plating on DBDM medium (one of the most selective media available for these yeasts [27]. None of the inoculated wines (i.e. irrespective of the inoculum size or pH) subjected to HHP showed any viable cells. Further, 4-ethylphenol was only detected in the non-HHP-treated inoculated controls (at pH 3.2 and 3.6). In these non-pressurised control wines, populations of $>10^4$ cfu mL⁻¹ were



Fig. 2 Principal components analyses for different samples treatments (pH, HHP, *Dekkera* inoculation and amount of cells inoculated) regarding with the concentration of aromatic compounds

detected and 4-ethylphenol concentrations of >400 μ g L⁻¹ (Table 1).

Effect of pressurisation on wine pigment content

Irrespective of pH, the HHP-treated wines, contaminated or not, showed a smaller total anthocyanin content (by 15 %) than the non-HHP-treated about controls (Table 2). Indeed, these controls showed larger contents of all anthocyanin monomers, especially of malvidin-3-O-glucoside (Table 2). Thus, the final anthocyanin content was affected by HHP treatment. These results agree with those reported by [5]. According to these ANOVA results, the anthocyanin content is not affected by the presence of D. bruxellensis D37. However, when a PCA is performed considering all treatments (depending on: pH, HHP, inoculation and number of cells) regarding the contents of monomeric anthocyanins, the treatments can be separated in isolated groups (Fig. 1). Therefore, the inoculation of D. bruxellensis D37 influences the anthocyanin profile of the wine. The two principal components explain 83.9 % of the variance in this analysis.

Some authors have reported *Dekkera/Brettanomyces* to reduce strongly wine anthocyanin content (for a review see [28], probably the manifestation of β -glucosidase activity [18]. This discrepancy with the present results may be explained in that the present work was performed with wines recently contaminated with *D. bruxellensis* D37, while these other authors made observations on wines that had been contaminated for much longer.

Irrespective of pH, the HHP-treated wines, contaminated or not, also showed a smaller vitisin A content than the non-HHP-treated controls (Table 3). Corrales et al. [5] reported the formation of certain types of pyranoanthocyanin (vitisin A-type) via condensation reactions during

Table 5 Colour measurements of the differently prepared and treated wines

Treatment	pН	Pressure (MPa)	420	520	620	Colour intensity	Tonality
3.2 C	3.2	-	$0.449 \pm 0.002^{\rm a}$	$0.469 \pm 0.002^{\rm d}$	$0.092 \pm 0.001^{\circ}$	$1.010 \pm 0.004^{\rm bc}$	0.958 ± 0.001^{a}
3.6 C	3.6	_	0.457 ± 0.003^{ab}	$0.447 \pm 0.003^{\rm bc}$	0.103 ± 0.001^{d}	$1.007 \pm 0.006^{\rm bc}$	$1.023 \pm 0.000^{\circ}$
3.2P C*	3.2	100	0.466	0.463	0.064	0.993	1.005
3.6P C	3.6	100	0.492 ± 0.003^{d}	0.459 ± 0.005^{bcd}	0.106 ± 0.004^{d}	1.058 ± 0.012^{d}	1.073 ± 0.006^{d}
3.2.10 ⁶ DB	3.2	100	$0.468 \pm 0.002^{\rm bc}$	0.461 ± 0.003^{cd}	$0.092 \pm 0.002^{\rm c}$	$1.021 \pm 0.007c$	$1.014 \pm 0.002^{\circ}$
3.6.10 ⁶ DB	3.6	100	0.450 ± 0.013^{a}	$0.410\pm0.018^{\rm a}$	$0.088 \pm 0.005^{\rm c}$	0.948 ± 0.036^{a}	1.099 ± 0.021^{e}
3.2.10 ⁴ DB	3.2	100	0.458 ± 0.013^{ab}	0.463 ± 0.013^{cd}	0.065 ± 0.004^{a}	$0.985\pm0.030\mathrm{b}$	$0.989 \pm 0.004^{\rm b}$
3.6.10 ⁴ DB	3.6	100	$0.471 \pm 0.001^{\circ}$	0.445 ± 0.001^{b}	$0.075\pm0.000^{\rm b}$	$0.991 \pm 0.001^{\rm bc}$	1.059 ± 0.003^{d}

Values are the means \pm SD of three fermentations. Means with the same lower case letter in the same column are not significantly different (p < 0.05). C control, DB contaminated with D. bruxellensis D37

* Sample 3.2P C was run without replication

pressurisation but depending on temperature. In the present work, no such effect was seen probably because of the much less extreme pressure and temperature conditions employed.

Effect of pressurisation on wine volatile compound content

The main molecules studied were higher alcohols and esters. Only small differences (perhaps unnoticeable to consumers even though in some cases significant) were seen between the differently treated wines and controls (Table 4).

GC has not been used before to study the modification of the volatile compound profile of wines treated by HHP. However, the present results confirm those of other authors who pressurised wine at 400 MPa at 20 °C for 2 min [5], at 400–500 MPa at 4–20 °C for 5–15 min [25], and at 350 MPa at 25 °C for 20 min [19], and who used wine tasters to determine their effects.

Wine ester contents are very sensitive to thermal stabilisation treatments and freezing [16]. These molecules, which have fruity aromas and strongly influence the olfactory quality of wines, generally have low sensorial thresholds and are normally found in equilibrium with their hydrolysis products. They can therefore be understood as indicators of the 'gentleness' of a microbiological stabilisation treatment. In the present work, independent of the pH, no significant differences were seen in the ethyl acetate content of the HHP-treated controls and the HHP-treated wines inoculated with 10^4 cfu mL⁻¹; in addition, the 10^6 cfu mL⁻¹-inoculated HHP-treated wines showed only a <10 % lower ethyl acetate concentration than the same controls (Table 4). The results for ethyl lactate behaved in the same way. A PCA statistical analysis was performed to evaluate using a multivariate technique the repercussion of HHP treatment in aromatic quality. This analysis shows that 66 % of variance can be expressed by two components and also that all controls either pressurised or not and at the two studied pH are grouped in the centre of the figure (Fig. 2). Only the treatments inoculated with Dekkera were separated in the left and right of the figure depending on inoculation size. HHP treatment can therefore be considered gentle.

Influence of HHP treatment on wine colour

Although HHP treatment (irrespective of pH or contamination with *D. bruxellensis* D37) led to some significant differences in colour intensity compared to the appropriate controls, no clear trend was seen with regard to the effects of inoculation or HHP treatment (Table 5). This agrees with that reported by other authors [7, 25]. However, lower pH (3.2) was associated with a reduction in the tonality (degree of 'yellow') of the controls (irrespective of HHP treatment or contamination with *D. bruxellensis* D37). Increasing tonality has been associated both with oxidation and ageing. Thus, tonality protection looks more related with low pH and not dependent on HHP treatment.

Industrial scale-up

HHP technique is enough mature to be used at industrial level in fact it is applied by many companies working in juice, meat, ham, dairy and other products. The cost of an industrial device to process by HHP ranges between 500,000 \notin and 1.5 million \notin [21]. The operational costs including amortisation in HHP processing can range between 0.045 \notin /kg to 0.087 \notin /kg depending on pressure, apparatus and production [14].

Conclusions

Traditional techniques for microbiologically stabilising wines, such as filtration, pasteurisation and flash pasteurisation at 60-70 °C all have a negative effect on their sensorial quality and reduce their polyphenolic compound, pigment and volatile compound contents. HHP treatment, however, seems to control *D. bruxellensis* D37 very well while largely avoiding these negative effects. The results therefore suggest that HHP might be a good alternative for controlling *Dekkera/Brettanomyces* in red wines.

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