RESEARCH PAPER

Electrical potential-assisted DNA-RNA hybridization for rapid microRNA extraction

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Abstract

Analysis of microRNAs (miRNAs) is important in cancer diagnostics and therapy. Conventional methods used to extract miRNA for analysis are generally time-consuming. A novel approach for rapid and sensitive extraction of miRNAs is urgently need for clinical applications. Herein, a novel strategy based on electrical potential-assisted DNA-RNA hybridization was designed for miRNA extraction. The entire extraction process was accomplished in approximately 3 min, which is much shorter than the commercial adsorption column method, at more than 60 min, or the TRIzol method, at more than 90 min. Additionally, the method ofered the advantages of simplicity and specifcity during the extraction process by electrical potential-assisted hybridization of single-stranded DNA and RNA. Taking let-7a as an example, satisfactory results were achieved for miRNA extraction in serum, demonstrating the applicability in miRNA nucleic acid amplifcation.

Keywords miRNA extraction · Nucleic acid extraction · Rapid extraction · Electrical potential-assisted hybridization

Abbreviations

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Introduction

RNAs are associated with various physiological and pathological processes, including metabolism, apoptosis, necrocytosis, proliferation, diferentiation and development [[1](#page-8-0)[–3](#page-8-1)]. MicroRNAs (miRNAs) are noncoding RNAs with length less than 100 base pairs (bp). With the increasing interest in miRNA therapeutics and targeting of miRNAs to treat disease, there is a need for tools to enable the fast and sensitive detection of miRNAs to determine their function [\[4](#page-8-2)[–6](#page-8-3)]. Reverse transcription quantitative polymerase chain reaction (RT-qPCR) is a particularly attractive technology for detection of miRNA due to its specifcity, sensitivity and relatively rapid time-to-answer [\[7](#page-8-4)[–10](#page-8-5)]. A rapid approach for obtaining pure and abundant miRNA molecules is vital in nucleic acid-based analysis.

The most widely implemented approaches reported for miRNA extraction include conventional solvent methods or commercial kit reagents. TRIzol reagent is the most commonly used solvent due to good consistency in recovery [\[11](#page-9-0)]. However, the method relies on a set of time-consuming reagents and enrichment steps, with over 4 h of complex operation to collect and process miRNAs, hindering its application for rapid detection preparation. The adsorption column method has been extensively harnessed for numerous commercial kit reagents owing to the simplicity of direct physical adsorption. However, more than 60 min of time

consumption and low recovery have impeded widespread application of this method because of the difficulty in separation. Rapid extraction of miRNA remains a great challenge because of degradation and forming co-precipitation [\[12](#page-9-1), [13](#page-9-2)]. These challenges require the development of novel strategies for rapid, simple and efficient miRNA extraction.

Owing to operational simplicity, remarkable sensitivity and the potential for automation, electrochemical-based strategies have drawn tremendous interest, and have achieved signifcant progress in the early clinical analysis of diseases. Electrochemical detection of small nucleic acid molecules in particular has become a research hotspot in the feld of bioengineering [[14–](#page-9-3)[17](#page-9-4)]. In the utilization of oriented DNA immobilized on electrodes, diverse electrochemical devices and platforms have been employed to detect miRNAs in the development of electrochemical assay to achieve high sensitivity and selectivity in terms of the electrochemical signal transduction, which also provides the possibility for the extraction of miRNAs owing to the combination of specifc DNA-RNA recognition [\[18–](#page-9-5)[20\]](#page-9-6). However, the current process of the specifc hybridization between the DNA capture and the target miRNA all need hours of reaction [[21\]](#page-9-7), and the miRNA specifcally captured can only be detected by electrochemical methods, and cannot be isolated [[22\]](#page-9-8). The complex modifcations commonly used in electrochemical methods undoubtedly increase the total time of the detection process, and the reproducibility of the detection of low-abundance targets is low [\[23\]](#page-9-9). For the same low concentration of targets, RT-qPCR increases its concentration through exponential amplifcation, which greatly improves the sensitivity and accuracy of detection [[24](#page-9-10), [25\]](#page-9-11). However, the complicated sample extraction steps such as in the TRIzol method still make RT-qPCR detection of miRNA time-consuming [\[26](#page-9-12)]. Controlled electric felds can be used to regulate transport and hybridization of single-stranded oligonucleotides [[27](#page-9-13)]. Heller et al. [[28](#page-9-14)] frst used electric felds to enhance DNA hybridization. Conde et al. [\[29](#page-9-15)] subsequently demonstrated that the use of microsecond voltage pulses increased the speed of hybridization by at least $10⁸$ times compared with the passive control reaction without an electric feld. In addition, Gebala et al. [\[18\]](#page-9-5) proposed a method for assisting DNA hybridization by applying an external voltage. However, although much has been done for extension or adaptation of nucleic acids, the overall detection process for miRNA still takes several hours, and there is no report based on electrical potential-assisted hybridization for miRNA extraction.

The expression of let-7a is closely related to a variety of cancers [[30\]](#page-9-16), so it is considered an important biomarker for accurate tumor diagnosis and prognosis [[31](#page-9-17)]. Several methods have been developed for let-7a detection, including sequencing, electrochemical and amplification methods. Next-generation sequencing achieves high-throughput profiling, but cannot reliably quantify low-abundance analytes [[32\]](#page-9-18). Electrochemical methods and standard assays based on amplifcation such as PCR are sensitive but often require complex sensor fabrication and timeconsuming extraction [[32](#page-9-18)[–34](#page-9-19)]. At present, a combination electrochemical and amplifcation protocol has been proposed to achieve more sensitive and specifc detection of let-7a, but complex modifcations and rigorous protocol design are still required [[35](#page-9-20)]. Herein, a rapid, simple and efficient method for miRNA extraction was investigated based on electrical potential-assisted DNA-RNA hybridization. Taking miRNAs of let-7a as an example, the region of the single-stranded DNA responsible for let-7a hybridization was designed and immobilized on the electrode. With the assistance of the electric feld, let-7a quickly formed an enrichment on the electrode surface, thereby increasing the chance of collision, and greatly increasing the nucleic acid hybridization rate. The entire let-7a isolation process was completed in 150 s, including 90 s capture and 60 s rapid elution steps. The extraction time was signifcantly shortened. The performance characteristics of the method for RNA extraction were investigated by RT-qPCR and accelerated strand exchange amplifcation (ASEA) [\[36\]](#page-9-21), and its possible application in real samples was evaluated for early cancer screening.

Materials and methods

Materials and reagents

DNA oligonucleotides were purchased from Sangon Biotech (Shanghai, China). DNA capture probe (15-mer): 5′CAA CCT ACT ACC TCA–SH 3′; complementary let-7a target: 5' UGA GGU AGU AGG UUG UAU AGU U 3′ (The underlined portion was reverse complement with DNA capture probe). TaqMan™ MicroRNA Reverse Transcription Kit was purchased from Applied Biosystems (USA). PerfectStart™ II Probe qPCR SuperMix was purchased from TransGen Biotech Co., Ltd. (Beijing, China). The ASEA detection kit was provided by Navid Biotech Co., Ltd. (Qingdao, China). The miRcute miRNA extraction and isolation kit was provided by Tiangen Biotech Co., Ltd. (Beijing, China). TRIzol was provided by Thermo Fisher Scientific Co., Ltd. (Shanghai, China). Ethanol, potassium ferricyanide $(K_3Fe(CN)_6)$, potassium ferricyanide trihydrate $[K_4Fe(CN)_6.3H_2O]$, chloroform and isopropanol were obtained from Sinopharm Chemical Reagent Co., Ltd. (Jinan, China). Tris(2-carboxyethyl)phosphine hydrochloride (TCEP), 6-mercaptohexanol (MCH), RNase-free water, KCl and NaCl were purchased from Sigma. All reagents were of analytical grade and were used as received. Human serum was provided by Qingdao Central Hospital.

Preparation of electrodes and DNA immobilization

Gold electrodes (2 mm diameter, Wuhan Goslink Technology Co., Ltd., China) were polished with wet alumina slurries with particle size of 0.3 μm (Shanghai Chenhua Instrument Co., Ltd., China) with a motor sander (GU-100) for 1 min each. Then they were rinsed ultrasonically for 1 min with ethanol and deionized water, respectively, and finally dried with nitrogen. The electrodes were further electrochemically characterized by means of cyclic voltammetry (CV) in a real-time prepared 0.5 M H_2SO_4 solution in a potential range from 0 V to $+1.2$ V vs. Ag/AgCl (3.0 M KCl) until a stable voltammogram was obtained. Finally, the electrodes were rinsed with water and dried under a nitrogen stream.

Electrical potential‑assisted DNA‑RNA hybridization for miRNA extraction

The prepared electrode was soaked in 20 μL of immobilization solution for 8 h. The immobilization solution contains 1 μM capture DNA probe and 10 mM TCEP. In addition, the electrode incubated in the $10 \mu L$ 1 mM MCH for 30 min to block the sites not occupied by the ligated capture. The modifed electrode was put into 200 μL of hybridization solution containing the target strand, two potential pulses of 100 mV for 0.01 s and −200 mV for 0.01 s were applied for a total of 4500 cycles, and electrical potential-assisted DNA-RNA hybridization was used to hybridize the target miRNA to the capture probe. The hybridization solution contained 10 mM phosphate-buffered saline (PBS) and 1 M NaCl, pH 7.4. The electrode was then rinsed with RNase free water, and a potential pulse of −1 V was applied for 60 s to elute the DNA-miRNA hybrid into 50 μL of RNase-free water to complete the isolation of the target miRNA.

Electrochemical measurement

CV and electrochemical impedance spectroscopy (EIS) were performed with an electrochemical workstation (CHI600E, CH Instruments, Inc.). The three-electrode system consisting of a gold working electrode, a platinum wire auxiliary electrode and an Ag/AgCl (3 M KCl) reference electrode was used. The CV was carried out at a scan rate of 100 mV/s. The EIS was measured in potassium ferri/ferrocyanide solution (5.0 mM [Fe(CN)6]^{3–/4–}, 0.1 M KCl), the voltage parameter was equilibrium potential, and the frequency was from 0.1 Hz to 100,000 Hz.

RT‑qPCR and ASEA amplifcation

The miRNAs were reverse transcribed into complementary DNA (cDNA) according to the procedure described by Chen et al. $[34]$ $[34]$. Reactions of 7.5 μ L were incubated in the CFX96 Connect™ Real-Time PCR System (Bio-Rad, CA, USA) for 30 min at 16 °C, 30 min at 42 °C, 5 min at 85 °C and then held at 4 °C. Real-time PCR was performed using a PerfectStart™ II Probe qPCR SuperMix. The fuorescence signal was detected by the CFX96 Connect[™] Real-Time PCR System at 95 °C for 10 min, followed by 40 cycles of 95 °C for 15 s and 60 °C for 1 min. ASEA amplifcation for 30-bp target was performed using the ASEA detection kits. The fuorescence signal was detected by the CFX96 Connect™ Real-Time PCR System at 40 cycles of 76 °C for 1 s and 60 °C for 1 s. Sequences of the primers and probes of RT-qPCR and ASEA are shown in Table S1.

Results and discussion

Scheme of the electrical potential‑assisted hybridization to extract miRNA

The principle of the electrical potential-assisted hybridization for miRNA extraction is illustrated in Fig. [1.](#page-3-0) The construction of the miRNA extraction platform was based on electrical potential-assisted DNA/RNA hybridization. The three terminally modifed sulfhydryl DNA capture probes were immobilized on the surface of the gold electrode by forming gold−sulfur bonds consisting of a highly reproducible recognition interface. In the presence of target miRNAs, hybridization between the DNA capture and miRNA target occurs, and acceleration of hybridization is achieved by electrical potential assistance. The extraction process only includes a binding step and isolation step. In the binding step, the repetitive electric felds of the two potential pulses were applied within 90 s to hybridize the target miRNA with the DNA capture. In the isolation step, the target miRNA was separated from the surface of the gold electrode by applying a continuous potential pulse of 60 s. The potential pulse condition was optimized, and the results are shown in Fig. S1.

Feasibility of electrical potential‑assisted DNA‑RNA hybridization to extract miRNA

To obtain insight into the interface properties of the electrode surface of electrodes, EIS was employed to test the detailed information of surface-modified impedances changes before and after DNA modifcation. In the Nyquist diagram, semicircle diameters represent the charge transfer resistance (Rct). Figure [2A](#page-4-0) shows the Nyquist diagram of gradual modifcation processes measured in PBS bufer (pH 7) containing 5 mmol/L $[Fe(CN)₆]^{3-/4-}$ and 0.1 mol/L KCl. Compared with the bare Au electrode (curve a), the Rct of the Au electrode modifed with capture probe (b)

Fig. 1 Schematic representation of electrical potential-assisted hybridization to extract miRNA

was obviously increased, which proved the successful modifcation of the capture probe. After the hybridization with 100 nM of the target sequence, there was an obvious improvement in the charge transfer rate over the electronic surface, confrmed by a smaller semicircle (curve c). The decrease in Rct after hybridization was attributed to the new conformation of the secondary structure of the double-layer [[37,](#page-9-22) [38\]](#page-9-23). After applying a negative electric field, smaller semicircles were observed in the electrode surface (curve d) due to the breakage of the gold–sulfur bond for RNA recovery. The above EIS results suggest the feasibility of the proposed method for miRNA extraction. To further investigate the miRNA extraction process, RT-qPCR was chosen to determine the amount of miRNA (Fig. [2B](#page-4-0)). As expected, the fuorescence signal based on the electrical potentialassisted method for RNA extraction (a) was signifcantly increased compared with the no-template control (curve c). The samples eluted directly after 90 s passive hybridization did not exhibit any changes in fuorescence signal (curve b), indicating that the electrical potential-assisted DNA-RNA hybridization was able to efectively extract miRNA. In addition, this strategy relies on a specifc complementary nucleic acid base to achieve hybridization and extraction of specifc targets. Therefore, the specifcity of the proposed miRNA extraction strategy was evaluated by extracting various sequences contained in the let-7 families and detecting using RT-qPCR. As shown in Fig. [2C,](#page-4-0) only let-7a was amplifed, while non-target sequences had no signal during the amplifcation process, which proved that let-7a was successfully extracted, and indicated the high specifcity of the strategy for let-7a extraction.

The volume of the hybridization system for the extraction of miRNA

The volume of hybridization buffer is crucial for the electrochemical performance response to the binding target. During the hybridization step, a known potential pulse was applied to the electrode. The larger the hybridization volume, the farther the edge of the solution is from the position where the electric feld is applied, so that the infuence of the electric potential becomes smaller [[39\]](#page-9-24). Therefore, the parameter of reaction volume was optimized here. The different volumes of the hybridization system with the same number of 10^6 copies (Fig. [3A\)](#page-5-0) or same concentration of 10^5 copies/ μ L (Fig. [3B\)](#page-5-0) were prepared for electrical potentialassisted extraction. Subsequently, RT-qPCR was investigated for quantifcation. As shown in Fig. [3A](#page-5-0), a lower volume of the hybridization system decreased the cycle threshold (Ct) value, suggesting higher efficiency. The difference in Ct value between diferent volumes was within one cycle, when the sample concentration was constant (Fig. [3B](#page-5-0)). Therefore,

Fig. 2 The feasibility of potential-assisted hybridization to extract miRNA. (**A**) Electrochemical impedance spectroscopy for the gold electrode. $a = EIS$ of an unmodified gold electrode; $b = EIS$ of a single-stranded (ss)DNA/MCH-modified gold electrode; $c=60$ s electrical potential-assisted hybridization with let-7a; d = the gold electrode at a constant potential of −1 V for 60 s. (**B**) Amplifcation of

let-7a. a=let-7a extracted by electrical potential-assisted hybridization; $d = let$ -7a eluted directly after 90 s passive hybridization; c=notemplate control. (**C**) RT-qPCR result corresponding to some nucleic acid sequences of let-7 family. $a-d=$ let-7a, let-7b, let 7-c and let-7 i, respectively. e=no-template control

the diference in performance was related to the volume, and the extraction effect in 200 μ L was the best. Consequently, the volume of the hybridization system was determined in 200 μL.

Extraction of targets of diferent lengths using electrical potential‑assisted hybridization

In order to evaluate the universal capability of the developed strategy for miRNA extraction, simulated samples with three lengths of 22 bp (Fig. $4a$), 39 bp (Fig. $4b$) and 60 bp (Fig. [4c\)](#page-6-0) were extracted by the electrical potential-assisted method. The target concentrations were 10^3 copies/ μ L to 10^6 copies/ μ L, and they were quantified by RT-qPCR (22 bp) and ASEA (39 bp and 60 bp). Although there were some diferences in the extraction efects of diferent concentrations and diferent lengths of samples, all the simulated samples were successfully extracted, which provided evidence for the universal application of the proposed strategy.

The application under a biological environment

To access the ability of the proposed assay method for diagnostic purposes, miRNA samples of 10^8 copies/ μ L let-7a were extracted from different concentration of human serum based on the electrical potential-assisted method. The quality of miRNA was determined based on the Ct value obtained by RT-qPCR analysis. Although the Ct value gradually increased with the increase in serum concentration due to the interfering substances

Fig. 3 RT-qPCR of miRNA isolated from diferent sample volumes. (A) The different volumes of the hybrid system with $10⁶$ copies target. a–d=hybridization in 200 μ L, 500 μ L, 1 mL and 2 mL hybridization buffer, respectively, and $e = no$ -template control. (**B**) The dif-

(Fig. $5B$), the quantitative values of miRNA extracted by the electrical potential-assisted method were sufficient for RT-qPCR (Fig. [5C\)](#page-7-0). In particular, at a high concentration of serum, there was still detectable signal appearing for the RT-qPCR (Fig. $5A$), indicating that the potential assisted-based method provided a simple and effective tool for miRNA extraction in complex samples. The possible reasons for the inhibitory effect of serum were as follows: On the one hand, serum could not provide the most suitable ionic strength for the hybridization of capture DNA and target miRNA like a hybridization solution. On the other hand, complex biological components such as proteins in the serum migrated to the vicinity of the electrode under the influence of the electric field; the steric hindrance of nucleic acid hybridization was caused by these components. Therefore, the efficiency of nucleic acid hybridization was reduced.

Comparison of electrical potential‑assisted extraction method with adsorption column and TRIzol method

To assess the performance of the proposed method for miRNA extraction, different copies of miRNA were extracted using the electrical potential-assisted method, adsorption column method and TRIzol method, and RT-qPCR analysis was performed. The operation steps of the adsorption column method and TRIzol method are shown in the supplementary material. The quality of RNA samples was determined by the Ct value based on RT-qPCR. As shown

a–d=hybridization in 200 μ L, 500 μ L, 1 mL and 2 mL hybridization buffer, respectively, and $e = no$ -template control

ferent volumes with the same concentration of 10^5 copies/ μ L target.

in Fig. [6,](#page-7-1) the amount of RNA strongly affected the efficiency of RT-qPCR. The Ct value of RNA samples increased with the decreasing concentration of RNA samples from 10^5 copies/μL to 10 copies/μL for the three methods. The RNA sample with lower concentration extracted using the electrical potentialassisted method showed a faster Ct value than that of the adsorption column and TRIzol methods, indicating higher efficiency for low-abundance RNA extraction. In particular, the acceptable Ct value obtained in 10 copies/μL of RNA sample with the electrical potential-assisted method is important for low-abundance miRNA-based diagnostics. These results suggest the universality of the new strategy based on the electrical potential-assisted method focusing on RNA extraction, which enhances the sensitivity of miRNAbased pathogenic diagnostics.

As mentioned earlier, the TRIzol and adsorption column methods are the most commonly used for RNA extraction. Compared to time consumption of more than 90 min for the traditional TRIzol method and more than 60 min for the adsorption column method, the potential-assisted method needed only 3 min because of the electric field on the electrode-accelerated hybridization and desorption, which effectively protected RNA from enzymatic digestion (Table [1](#page-8-6)) [[40](#page-10-0), [41](#page-10-1)]. Moreover, the proposed method demonstrated good specificity and sensitivity due to the highly efficient hybridization reaction between the DNA recognition probe and RNA. These features will allow the method to be used for relative RNA quantification in diagnostics.

Fig. 4 Extraction of the targets with length of (**a**) 22 bp, (**b**) 39 bp and (**C**) 60 bp using electrical potential-assisted hybridization. Relationship between the Ct values of RT-qPCR and the logarithmic values of the target concentration. Black lines represent the synthetic

targets used directly for RT-qPCR; red lines represent the targets extracted by electrical potential-assisted hybridization before RT-PCR. Error bars show mean standard deviations of three determinations. The illustrations represent RT-qPCR quantitative values

Versatility of miRNA extraction

Based on specific base complementation, the capture probe sequence can be replaced according to the sequence of the detection target, in order to realize the extraction of various miRNAs. miRNA 21 is considered a diagnostic and prognostic biomarker of various human malignancies [\[42\]](#page-10-2). Therefore, the capture probe was designed according to the miRNA 21 sequence. A concentration of miRNA 21 with 10^6 copies/ μ L was diluted in artificial serum, and then the extraction was assisted by electric potential. As shown in Fig. [7](#page-8-7), miRNA 21 was successfully amplified by RT-qPCR, proving that miRNA 21 isolated by potential-assisted hybridization can be used for amplification

detection. This demonstrates the miRNA extraction versatility of the proposed electrical potential-assisted extraction method.

Conclusion

An ultrafast electrical potential-assisted miRNA extraction scheme is reported here for the frst time. This work has demonstrated that electrical potential-assisted DNA-RNA hybridization is capable of miRNA recovery within a shorter time than with commercial methods. miRNA extraction was achieved within 3 min, much shorter than that of traditional methods such as the adsorption

 $-100%$

Serum

75%

:50%

25%

Fig. 5 The anti-interference ability of electrical potential-assisted hybridization to extract miRNA. (**A**) RT-PCR of diferent volume percentages of hybridization solution and interference serum. a=100% hybridization buffer, b=75% hybridization buffer + 25% serum, c=50% hybridization bufer + 50% serum, d=25% hybridization buffer + 75% serum, and e=100% serum. f=no-template

control. (**B**) The Ct value of RT-qPCR corresponding to diferent percentages of hybridization solution and interfering serum in the hybridization system. (**C**) The quantitative values of RT-qPCR. $1=100\%$ hybridization buffer, $2=75\%$ hybridization buffer + 25% serum, $3=50\%$ hybridization buffer + 50% serum, $4=25\%$ hybridization buffer $+75\%$ serum, and $5=100\%$ serum

Fig. 6 Comparison of electrical potential-assisted hybridization with adsorption column method and TRIzol method. The concentration of RNA samples ranged from 10^5 copies/ μ L to 10 copies/ μ L. The

abscissa represents the logarithm of the target concentration, and the ordinate represents the average Ct value (**a**) and quantitative value (**b**) of RT-PCR. Each value represents 10 parallel experiments

Fig. 7 miRNA 21 detection using electrical potential-assisted hybridization protocol. $a = miRNA 21$ and $b = no$ -template control

column method (>60 min) or TRIzol method (>90 min). The method of electrical potential-assisted DNA-RNA hybridization for miRNA extraction was simple, rapid and efficient compared with traditional methods. The resulting preparation of miRNAs in serum was highly enriched in the target miRNA, supporting the applicability of the method in real samples, which is efectively ready for RTqPCR analysis. It is worth mentioning that this system was able to detect miRNA in concentrations as low as 10 copies/μL combined with RT-qPCR, which is much lower than the conventional concentration of miRNA in cancer cells. The high extraction efficiency of low-concentration miRNA in this work will aid in accurate quantitation of low-abundance miRNAs, providing an efective tool in the development of miRNA extraction for clinical diagnosis.

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Data availability All data generated or analyzed during this study are included in this manuscript and its supplementary information fles.

Declarations

The studies were approved by the appropriate ethics committee and were performed in accordance with ethical standards.

Competing interests The authors declare that there are no competing interests associated with the manuscript.

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