



# Mass spectrometry-based shotgun lipidomics – a critical review from the technical point of view

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## Abstract

Over the past decade, mass spectrometry (MS)-based “shotgun lipidomics” has emerged as a powerful tool for quantitative and qualitative analysis of the complex lipids in the biological system. The aim of this critical review is to give the interested reader a concise overview of the current state of the technology, focused on lipidomic analysis by mass spectrometry. The pros and cons, and pitfalls associated with each available “shotgun lipidomics” method are discussed; and the new strategies for improving the current methods are described. A list of important papers and reviews that are sufficient rather than comprehensive, covering all the aspects of lipidomics including the workflow, methodology, and fundamentals is also compiled for readers to follow.

**Keywords** Lipidomics · Mass spectrometry · Electrospray ionization · MALDI · Collision induced dissociation · Lipid

## Abbreviations

Cer	Ceramide
Cer[AdS]	$\alpha$ -hydroxy fatty acyl containing dihydrosphingosine ceramide
Cer[AS]	$\alpha$ -hydroxy fatty acyl containing ceramide
Cer[OH]	$\omega$ -hydroxy fatty acyl containing ceramide
CID	Collision induced dissociation
CL	Cardiolipin
DAG	Diacylglycerol
DGDG	Diglycosyldiacylglycerol
ESI	Electrospray ionization
FAB	Fast atom bombardment
FFA	Free fatty acid
FT ICR	Fourier transform ion cyclotron resonance
GalCer	Galactosylceramide
GlcCer	Glucosylceramide
HR	High resolution
IM	Ion mobility
LC	Liquid chromatography
MALDI	Matrix assisted laser desorption ionization

MS	Mass spectrometry
NLS	Neutral loss scan
PA	Phosphatidic acid
PC	Phosphatidylcholine
PE	Phosphatidylethanolamine
PG	Phosphatidylglycerol
PI	Phosphatidylinositol
PIS	Precursor ion scan
pPS	Plasmalogen phosphatidylserine
PS	Phosphatidylserine
SFC	Supercritical fluid chromatography
SM	Sphingomyelin
Sulf	Sulfatide
TAG	Triacylglycerol
TLC	Thin layer chromatography
TOF	Time-of-flight
TSQ	Triple stage quadrupole

## Introduction

The emergence of the electrospray ionization (ESI) technique [1, 2] coupled with mass spectrometry (MS) that permits direct analysis of intact phospholipids with 2 to 3 orders of magnitude more sensitivity than that achieved by FAB/MS [3–5], prompts the development of “shotgun lipidomics”, a term first coined by Han and Gross in their seminal work applying mass spectrometric techniques without prior

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chromatographic separation for global analysis of lipids from crude lipid extracts of biological origin [6, 7]. The superb sensitivity and ease of continuous sample introduction with ESI, coupled with tandem mass spectrometry provide opportunities to explore the structure and the fragmentation process of complex lipids in greater detail [8, 9]. The ions generated by ESI minimized some of the problems such as the complication of the spectra associated with matrixes employing FAB/MS, or MALDI-TOF, the other sensitive MS tool for lipid analysis. On the other hand, MALDI-TOF has emerged as another MS-based lipidomic tool for, e.g., imaging lipid in the tissue, due to its sensitivity and the abundance of lipids in the tissue that can be released by MALDI [10, 11].

Lipids are present in all prokaryotic and eukaryotic cells and are essential to maintain the life. Lipids contain a range of chemical functional groups, which can be divided into many classes and subclasses. To completely analyze their structures in a short period of time is a daunting task. Up to now, numerous articles and reviews regarding lipidomic analysis, including shotgun lipidomics, have been published [12–54]. Over the years since the application of MS in lipid analysis, the sensitivity, fidelity, and speed have been improved significantly. The driving force behind the improvement is a constant advance in mass spectrometry and related technologies, and more importantly, the desire to gain insight into the biological processes related to lipids. The ultimate goal for these continuous efforts is to develop a reliable, high throughput, sensitive, and automatic method for global analysis of lipids in the cellular level, leading to understand the biological function of lipids related to human health and diseases.

The aim of this review is to provide readers with critical evaluation of the current state of “shotgun lipidomics”. This article is devoted to the application of MS to the quantitation and elucidation of the chemical structure of the major lipids of biological origins deriving mainly from eukaryotic systems. The “LC/MS lipidomics” approach and the lipids outside the major lipid classes will not be reviewed. Of importance, pertinent examples that highlight the key role of the state-of-the-art MS methods employing ESI high resolution multiple stage mass spectrometry, and the traditional triple quadrupole and MALDI-TOF-TOF mass spectrometry to analyze lipids will be discussed. I will provide an overview and critique of the current understanding of lipid analysis, taking into account the various methodologies without employment of chromatographic separation.

The non-MS related techniques such as the development of the sample preparation methods before MS analysis and of the data processing methods after MS-data acquisition will not be reviewed here. However, a list of important papers and reviews that cover all the aspects of lipidomics, including the workflow, methodology, and fundamentals is compiled for readers to follow. It is also noteworthy that various nomenclature systems have been used, and the abbreviations,

nomenclatures, and lipid classification published by LIPID MAPS consortium [55] with modification will be adopted in this review.

## Analytical methods toward shotgun lipidomics

Table 1 is a collection of some of the most representative literatures of the MS-based lipidomics, including those involving chromatographic separation by liquid chromatography (LC), thin layer chromatography (TLC), ion mobility (IM), and supercritical fluid chromatography (SFC) combined with MS for lipid analysis for interested readers to follow.

### Continuous infusion “shotgun lipidomics”

Dependent on the specimen, lipids are often present as a big family, consisting of numerous structures in various subclasses (Fig. 1), with a wide range of concentrations. Lipids are “sticky”, i.e., they can aggregate, interact, and are difficult to remove. Additionally, the physicochemical properties such as the basicity, polarity, vapor pressure, and chargeability among various lipid classes are different. As such, the physical states including the salt content and the pH of the lipid specimen when ionization takes place in the ion source are also different. Thus, the ionization efficiencies of various lipid families upon subjected to different ionization methods can vary significantly. Shotgun lipidomics tries to minimize these differences and maximize the ionization efficiency. On the other hand, it also exploits these differences; thereby, the various lipid classes can be formed by charge separation (e.g., deprotonation) or adduct formation (e.g., protonation, alkaline metal adduct ion, halogen adduct ion) in the solution or gas phase.

Han and coworkers describe a work flow applying direct infusion in combination with tandem quadrupole mass spectrometry for sensitive detection of various lipid classes. Anionic lipids, including cardiolipins (CLs), phosphatidylglycerols (PGs), phosphatidylinositols (PIs), phosphatidylserines (PSs), phosphatidic acids (PAs), and sulfatides (Sulf) of cellular lipid extract of biological origins were analyzed as  $[M - H]^-$  ions by negative-ion ESI-MS, while LiOH in methanol (50 nmol/mg of protein) were added to each individual cellular extract to supply  $Li^+$  ions prior to analyses of triacylglycerols (TAGs), galactocerebrosides (GalCers), phosphatidylcholines (PCs), and sphingomyelins (SMs) as the  $[M + Li]^+$  adduct ions. In the meantime, phosphatidylethanolamine (PEs) molecular species was turned into anionic phospholipids and was analyzed as  $[M - H]^-$  ions. Thereby, separation of various lipid classes can be established by electrospray ionization [13]. This approach to achieve lipid

**Table 1** Summary of current literatures related to MS-based shotgun lipidomic

Type	Sample inlet		Article type	Reference #	Comments	
Shotgun	Direct infusion (including Nanomate)	Flow injection				
		x			12	One of the earliest papers on global lipid analysis by combined linked scans using TSO with flow injection
	x		Review	13, 14, 15		Multiple dimensional shotgun using combined linked scans with TSQ for global lipid analysis
	x		Review	15		Global plant lipid analysis using combined linked scans with TSQ
					16	Removal of non-sphingolipid by lithium methoxide treatment to enhance shotgun sphingomyelin analysis
	x				17	Database software development based on multiple dimensional linked scans for automation in quantitative and qualitative lipid analysis
	x		Review	18, 21, 24, 28, 33, 43, 49		HR MS/MS-shotgun lipid profiling with a Q-TOF or Orbitrap FT and database search
			Review	19, 23, 26, 41		General review on MS-based lipidomics, including fundamentals, workflow, and methodology
					40, 45	Tips and methodology for shotgun lipidomics
	x		Review	6, 34, 52		General review on MS-based lipidomics
			Review	37, 39		A must-read very comprehensive review covering all the aspects of shotgun lipidomics
	x				42	Shotgun on lysophospholipids
	x				38	Applying photochemical reaction before ESI followed by MS/MS to locate fatty acid double bond
	x				35	A novel approach based on MS <sup>1</sup> to MS <sup>n&gt;3</sup> DDA/DIA for targeted and untargeted lipid analysis using ultra-HR Orbitrap/Fusion instrument and homebuilt database (ALEX) analysis
	x		Review	50		Shotgun on aminophospholipids and plasmalogen lipid via specific reaction with amino group and O-alk-1'-enyl substituent
					54	Applying ozonolysis to ESI generated fatty acid-containing lipids followed by MS analysis to locate double bond position
Chromatographic separation	HPLC	HPLC/ion mobility	SFC	Review	22	LC/MS analysis on neutral glycerolipid and cholesterol ester
					30, 47	SFC/ESI-HRMS with packed column for lipidomic analysis
			Review	46		General review on lipid analysis by SFC-MS
		-/x			32	Drift-tube ion mobility (IM)-MS analysis of glycerophospholipids
	x				20	ESI/LC/MS analysis of DAG
	x				29	Nano LC/MS on oxysterol analysis
		-/x	Review	31		Comprehensive review on qualitative and quantitative analysis of lipid by IM-MS and IM MS/MS.
			Review	36		General review of FTICR/Orbitrap instrument method on lipidomics including metabolomics
x				44	LC/MS based sphingolipid analysis	
	-/x			27	Differential ion mobility spectrometry (DMS)-driven shotgun lipidomics demonstrating the application of DM separation for distinction of isobaric and isomeric phospholipids, ether-acyl and diacyl PC and PE, and of PC and SM lipid families	

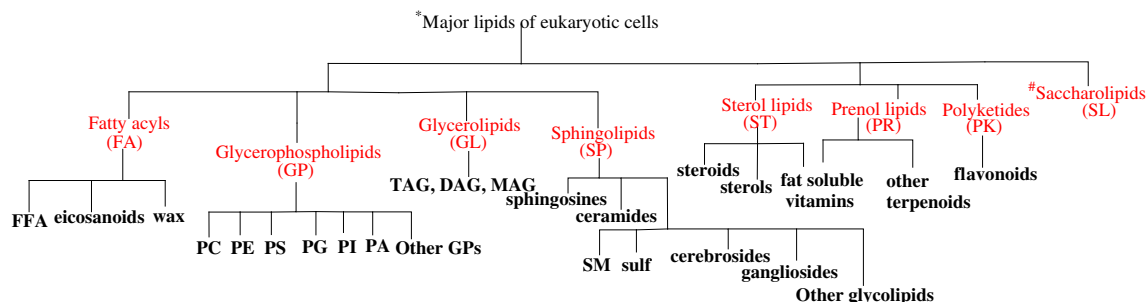
**Table 1** (continued)

Type	Sample inlet	Article type	Reference #	Comments
Shotgun	Direct infusion Flow injection (including Nanomate)			
		x/x	51	UPLC/DMS with Q-Trap for qualitative and quantitative analysis of phospholipids
		x/x	53	Online ozonolysis combined with ion mobility-mass spectrometry for lipid double bond recognition and lipid isomer differentiation
		Review	48	Review on TLC-blot MALDI-TOF lipidomic with informative references
		Review	102	A very comprehensive review covering all aspects of MALDI TOF MS analysis of lipids

analysis without prior chromatographic separation via manipulation of the physical state of the various lipid (e.g., basicity) and the ionization method (i.e., positive- and negative-ion ionization), was termed as “intrasource” separation [6, 56]. However, this approach to accomplish lipid separation is in fact via the manipulation of ionization efficiency by which the signals of different lipid classes were selectively suppressed or preferentially formed, and physical separation of lipid classes in the ion source has not been evidenced.

The concept that quantitates PC, SM, and TAG as lithium adduct ions by adding LiOH provides discrete  $[M + Li]^+$  molecular species apart from other lipid classes for profiling. More importantly, the stable  $[M + Li]^+$  molecular species can be further subjected to CID tandem mass spectrometry for total structural characterization [9, 57–59]. In the same context, the discrete  $[M - H]^-$  ions of Sulf, PI, CL, GalCer, and Cer formed in the negative ion mode can be applied for quantitation; and structural identification and isomer differentiation can be achieved by further CID tandem mass spectrometry [60–66]. With the addition of internal standards, simultaneous semi-qualitative and quantitative methods via precursor ion scan (PIS) and neutral loss scan (NLS) features readily available for a tandem quadrupole instrument, can be easily implemented [61–68].

However, several drawbacks have been encountered in this approach: (1) the addition of LiOH rarely results in the exclusive formation of  $[M + Li]^+$  adduct ions, in particular for the lipid extract of biological origin where  $Na^+$  is ubiquitous. The presence of the companion  $[M + Na]^+$  ions (sometimes  $[M + K]^+$  ions also present), which are 16 Da heavier complicates the analysis. For example, the  $[M + Li]^+$  ion of 16:0/18:1-PC has the same nominal mass as the  $[M + Na]^+$  ion of 1-O-hexadecanyl-2- octadecadienoyl sn-glycero-3-phosphocholine (e16:0/18:2-PC) at  $m/z$  766.6 (unless a sufficient high resolution mass spectrometry can be implemented for their distinction); (2) In the presence of  $Li^+$ , PE can also form  $[M + Li]^+$  ions [69], which is indistinguishable from the PC  $[M + Li]^+$  species consisting of C2-chain length shorter (identical elemental composition), although these adduct ions are often of low abundance. (3) To facilitate the formation of ceramides as the  $[M - H]^-$  ions by adding LiOH to raise the pH of the solution cannot completely prevent the anionic adduct ion formation such as  $[M + Cl]^-$  ions, in particular, if  $CHCl_3$  was used as co-solvent to dissolve the sample for infusion ( $Cl^-$  can be generated from  $CHCl_3$  by ESI). To minimize  $[M + Cl]^-$  ion formation, skimmer CID (or termed as fragmentor voltage for Agilent instruments) can be applied to transform



**Fig. 1** The most commonly known lipids in eukaryotic system. \*Lipid classification adopted from LIPID MAPS GATEWAY (<http://www.lipidmaps.org/data/structure/index.html>; <https://doi.org/10.1194/jlr>.

E400004-JLR200 and <https://doi.org/10.1194/jlr.R800095-JLR200> (where structures are also available) with modification. #Lipid class rarely present in eukaryotic cells

them into  $[M - H]^-$  ions by removal of HCl. (4) LiOH is slightly soluble in methanol (9.76 mg/100mL methanol); and is a base that can saponify TAG, leading to artificial formation of diacylglycerols (DAGs) and fatty acids. (5)  $Li^+$  has two stable isotopes with masses of 6.01512 u (7.6%) and 7.01600 u (92.4%) that can complicate the analysis. To remedy, exclusively  $^7LiOH$  can be used.

To overcome these shortcomings, 0.5%  $NH_4OH$  (29%  $NH_3$  basis) in methanol solution (pH ~10) can be added to the lipid extract prior to ESI. In this manner, PC and SM can be exclusively detected as the  $[M + H]^+$  ions without compromising the sensitivity; and TAG in the fashion of  $[M + NH_4]^+$  can be formed with less complexity and more sensitivity compared with that seen as the  $[M + Li]^+$  ions, which are often complicated by simultaneous formation of  $[M + Na]^+$  ions, in particular for biological samples (unpublished results). The addition of  $NH_4OH$  also drastically increases the sensitivity of the anionic lipids that are to be detected as the  $[M - H]^-$  ions [70], including CL, PG, PI, PS, PA, and Sulf, and >2-fold sensitivity increase has been observed compared with that obtained from the original solution (i.e., no  $NH_4OH$  was added). More importantly, sphingonoids such as ceramide and GlcCer can be exclusively formed as the  $[M - H]^-$  ions in the negative-ion mode with a greater sensitivity, averting the formation of the undesired adduct ions such as  $[M + Cl]^-$  and  $[M + CH_3CO_2]^-$  completely. For lipid species such as phosphatidylinositol polyphosphate ( $PIP_n$ ,  $n = 1, 2, 3$ ), acylCoA and CL, more doubly charged ions can be formed. The readily recognizable MS pattern of the doubly charged ions (0.5 Da apart for isotopic peaks) provides extra information for structure identification; meanwhile, the resulting  $[M - 2H]^{2-}$  ions can also be detected by mass spectrometers designed with a limited mass range (<1000 Da).

### Loop injection versus continuous infusion

The shortcoming of using continuous infusion for lipid analysis is the cross-contamination among samples that were subjected to analysis sequentially. Another drawback of using continuous infusion as sample inlet is the accumulation of nonvolatile metal ion such as  $Na^+$  as infusion persists. This leads to the decline of the  $[M - H]^-$  ion formation for PPLs (such as PI, PE, and PG) in the negative ion mode, and the rise of the formation of  $Na^+$  adduct ions, particularly in the analysis of biological specimens, where  $Na^+$  is often abundant. To overcome the contamination problem from the carryover of previous samples, a chip-based nanospray ion source, for example the Advion TriVersa NanoMate (<https://advion.com/products/triversa-nanomate/>) for direct infusion was implemented in many laboratories. The setup has a minimal sample consumption due to the fabrication suitable for low

flow rate (low as to 20 nL/min) and because each sample uses its nanoelectrospray emitter, the sample-to-sample carry-over can be minimized. All nanospray ion sources provide good sensitivity and consume a very small amount of samples, but relatively more complicated to implement. The chip-based nanospray ion source has been well received in shotgun lipidomic analysis [18, 24, 35, 42, 71]. However, the device is rather expensive and the cost of operation is relatively much higher.

On the other hand, the conventional loop injection method using a syringe or a HPLC pump with isocratic flow combined with an autosampler affords background subtraction to minimize sample carryover from the previous injections. Loop injection without chromatographic separation also provides speed, automation, and is suitable for all the analysis [72–74]. Loop injection adds flexibility in the solvent selection, which can be further optimized to the conditions that maximum sensitivity (e.g.,  $NH_4OH$  or  $HCO_2H$  can be added before ESI analysis in negative- or positive-ion mode, respectively), or types of adduct ions (e.g., via adding adulterary ions such as  $Na^+$ ,  $Li^+$  to form  $[M + Na]^+$  and  $[M + Li]^+$  ions) can be reached. The flexibility in the sample amount and concentration to be introduced for analysis and in the flow rate can increase the linear range for quantitation. Loop injection in conjunction with high resolution or tandem quadrupole mass spectrometry thus provides a very simple setup that can be easily implemented for lipidomic analysis.

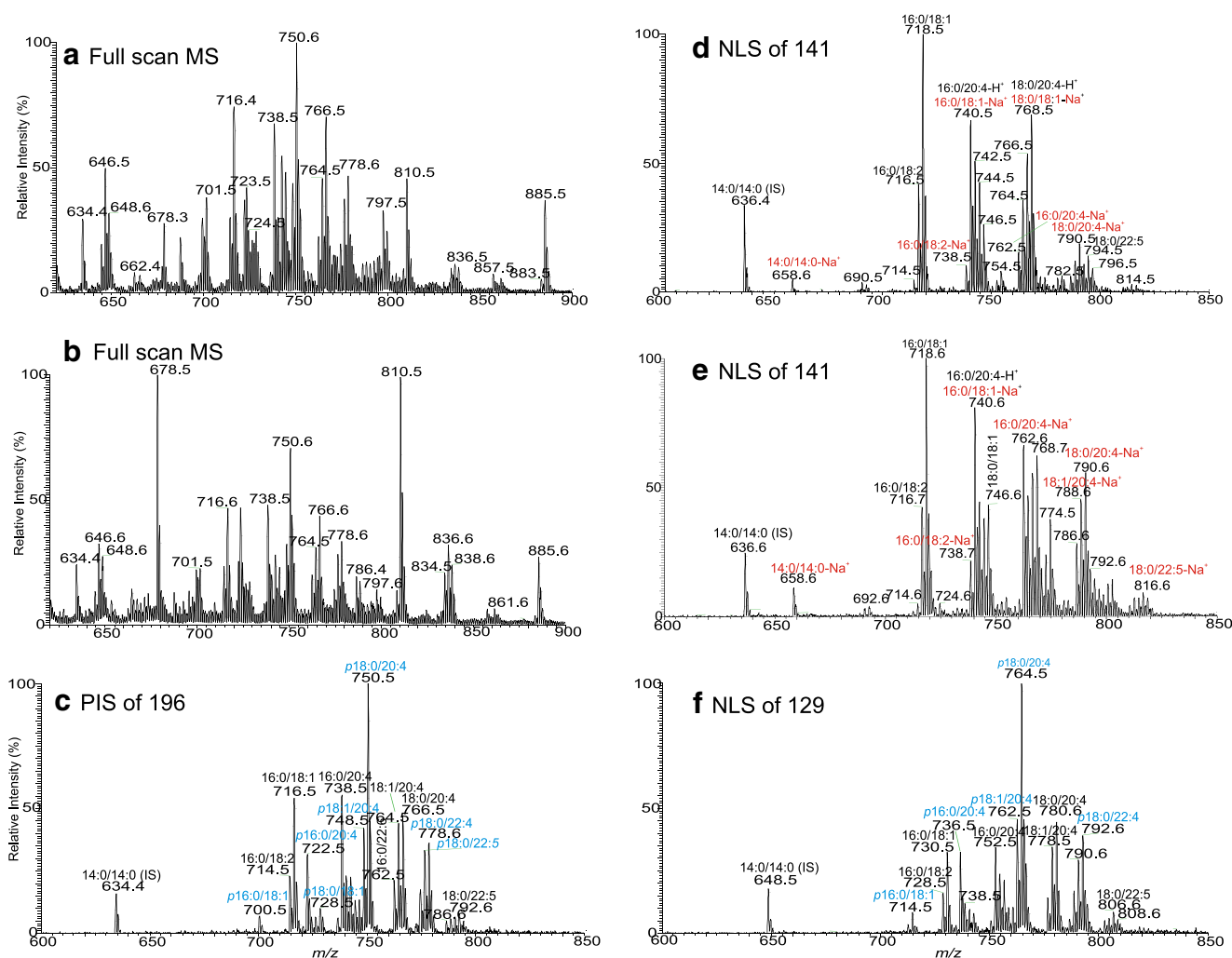
### Shotgun lipidomics with tandem quadrupole mass spectrometry

Insight into the fragmentation processes is crucial for optimal utilization of tandem quadrupole mass spectrometry toward shotgun lipidomics. Structural information from product-ion spectrum affords identification of lipid structures [9]. Accordingly, relevant ions can also be selected for precursor ion scan (PIS) and neutral loss scan (NLS) (Table 2) for quantitation of various intact lipid classes without chromatographic separation [12, 75]. Selectivity provided by linked scans using a tandem quadrupole instrument minimizes background ions, and  $s/n$  can be improved. However, drastic decline in the sensitivity and, in many cases, loss of the fidelity that reflects the entire specific lipid classes in the mixture were often seen, dependent on the ions selected for linked scan. For example, quantitation of PE species in a lipid extract from blood (Fig. 2, panels a and b) applying PIS of 196 (Fig. 2, panel c) to fish out PE species as  $[M - H]^-$  ions has very poor sensitivity because ion of  $m/z$  196 is of low abundance in the PE product ion spectrum [76], while applying NLS of 141 (Fig. 2, panel d) for profiling PE as  $[M + H]^+$  ions [12, 15] severely discriminates against plasmalogen-PE species. For example, ions of  $m/z$  700 (p16:0/18:1), 722 (p16:0/20:4), and 750 (p18:0/20:4)

**Table 2** Summary of the class-specific linked scan modes available in positive and negative ion ESI tandem quadrupole MS

Lipid class	Detected ion type	Scan mode	Fragment type	*Optimal CE	References	Comments
PC	[M + H] <sup>+</sup>	PIS 184	[phosphocholine + H] <sup>+</sup>	34 eV	12	Excellent sensitivity
SM	[M + H] <sup>+</sup>	PIS 184	[phosphocholine + H] <sup>+</sup>	38 eV	12	Excellent sensitivity
PS	[M + H] <sup>+</sup>	NLS 185	loss of phosphoserine	32 eV	12	Moderate sensitivity
PE	[M + H] <sup>+</sup>	NLS 141	loss of phosphoethanolamine	32 eV	12	Moderate sensitivity, discriminates against plasmalogen PE; overlaps with [M + Na] <sup>+</sup>
Cer	[M + H-H <sub>2</sub> O] <sup>+</sup>	PIS 264	[LCB + H - 2H <sub>2</sub> O] <sup>+</sup>	32 eV	79	Good sensitivity
Cer	[M + Li] <sup>+</sup>	NLS 48	loss of [H <sub>2</sub> O + HCHO]	45 eV	78	Good sensitivity for all LCB?nFA-Cer; not every specific; discriminates against LCB/αhFA-Cer
PC	[M + Li] <sup>+</sup>	NLS 183	loss of phosphocholine	35 eV	83	Good sensitivity; favors polyunsaturated-FA containing PC
SM	[M + Li] <sup>+</sup>	NLS 183	loss of phosphocholine	40 eV	58, 67	Good sensitivity
SM	[M + Li] <sup>+</sup>	NLS 59	loss of trimethylamine	30 eV	58, 67	Good sensitivity
SM	[M + Li] <sup>+</sup>	NLS 213	loss of [phosphocholine + HCHO]	52 eV	58, 67	Good sensitivity
PS	[M + Li] <sup>+</sup>	NLS 191	loss of [phosphoserine Li salt]	35 eV	9	Good sensitivity
PE	[M - Li+2Li] <sup>+</sup>	NLS 129	loss of [phosphoethanolamine Li salt - H <sub>2</sub> O]	32 eV	69	Good sensitivity and fidelity; dilithiated ions are less readily formed
PG	[M - Li+2Li] <sup>+</sup>	PIS 167	[phosphoglycerol - H + 2Li] <sup>+</sup>	45 eV	9	Good sensitivity and fidelity; dilithiated ions difficult to form
GlcCer	[M + Li] <sup>+</sup>	NLS 210	loss of [Glc + HCHO]	65 eV	59	Good sensitivity
PE	[M + Na] <sup>+</sup>	PIS 164	[phosphoethanolamine + Na] <sup>+</sup>	35 eV	75	Good sensitivity; discriminates against plasmalogen PE
PG	[M - H] <sup>-</sup>	PIS 153	[phosphoglycerol - H <sub>2</sub> O - H] <sup>-</sup>	35 eV	12	Poor sensitivity; not specific, co-present with all other phospholipids, and CL
PS	[M - H] <sup>-</sup>	NLS 87	loss of [serine - H <sub>2</sub> O]	25 eV	9, 12	Excellent sensitivity
PE	[M - H] <sup>-</sup>	PIS 196	[phosphoethanolamine - H + C <sub>3</sub> H <sub>4</sub> O]	55 eV	12, 76	Poor sensitivity
PE	[M + Cl] <sup>-</sup>	NLS 36	loss of H <sup>35</sup> Cl	25 eV	75	Excellent sensitivity; [M + Cl] <sup>-</sup> ions not readily formed
PI	[M - H] <sup>-</sup>	PIS 241	[phosphoinositol - H <sub>2</sub> O - H] <sup>-</sup>	45 eV	12, 65	Excellent sensitivity
Cer	[M - H] <sup>-</sup>	NLS 256	loss of [C <sub>13</sub> H <sub>27</sub> CH=CHCHO + H <sub>2</sub> O]	32 eV	62, this paper	Excellent sensitivity; sphingosine (d18:1) ceramide only; discriminates against d18:1/αhFA-Cer
Cer	[M - H] <sup>-</sup>	NLS 327	loss of [sphingosine + CO]	32 eV	62	Excellent sensitivity; specifically detect d18:1/αhFA-Cer
Sulf	[M - H] <sup>-</sup>	PIS 97	H <sub>2</sub> PO <sub>4</sub> <sup>-</sup>	65 eV	63,64	Good sensitivity
PIP	[M - H] <sup>-</sup>	PIS 321	[inositol diphosphoric acid - H <sub>2</sub> O - H] <sup>-</sup>	50 eV	65	Moderate sensitivity
PIP2	[M - H] <sup>-</sup>	PIS 241	[phosphoinositol - H <sub>2</sub> O - H] <sup>-</sup>	50 eV	65	Good sensitivity; co-present with PI, PIP <sub>2</sub>
PIP2	[M - H] <sup>-</sup>	PIS 401	[inositol triphosphoric acid - H <sub>2</sub> O - H] <sup>-</sup>	50 eV	65	Moderate sensitivity
IPC	[M - H] <sup>-</sup>	PIS 241	[phosphoinositol - H <sub>2</sub> O - H] <sup>-</sup>	50 eV	65	Good sensitivity; co-present with PI, PIP
IPC	[M - H] <sup>-</sup>	PIS 241	[phosphoinositol - H <sub>2</sub> O - H] <sup>-</sup>	45 eV	68	Good sensitivity
CL	[M - H] <sup>-</sup>	PIS 153	[phosphoglycerol - H <sub>2</sub> O - H] <sup>-</sup>	50 eV	66	Poor sensitivity; not specific, co-present with all other phospholipids
CL	[M - 2H] <sup>2-</sup>	PIS 153	[phosphoglycerol - H <sub>2</sub> O - H] <sup>-</sup>	25 eV	66	Poor sensitivity; unique doubly charged ion pattern easy to identify

\*Collision energy (CE) was optimized for overall fidelity and sensitivity



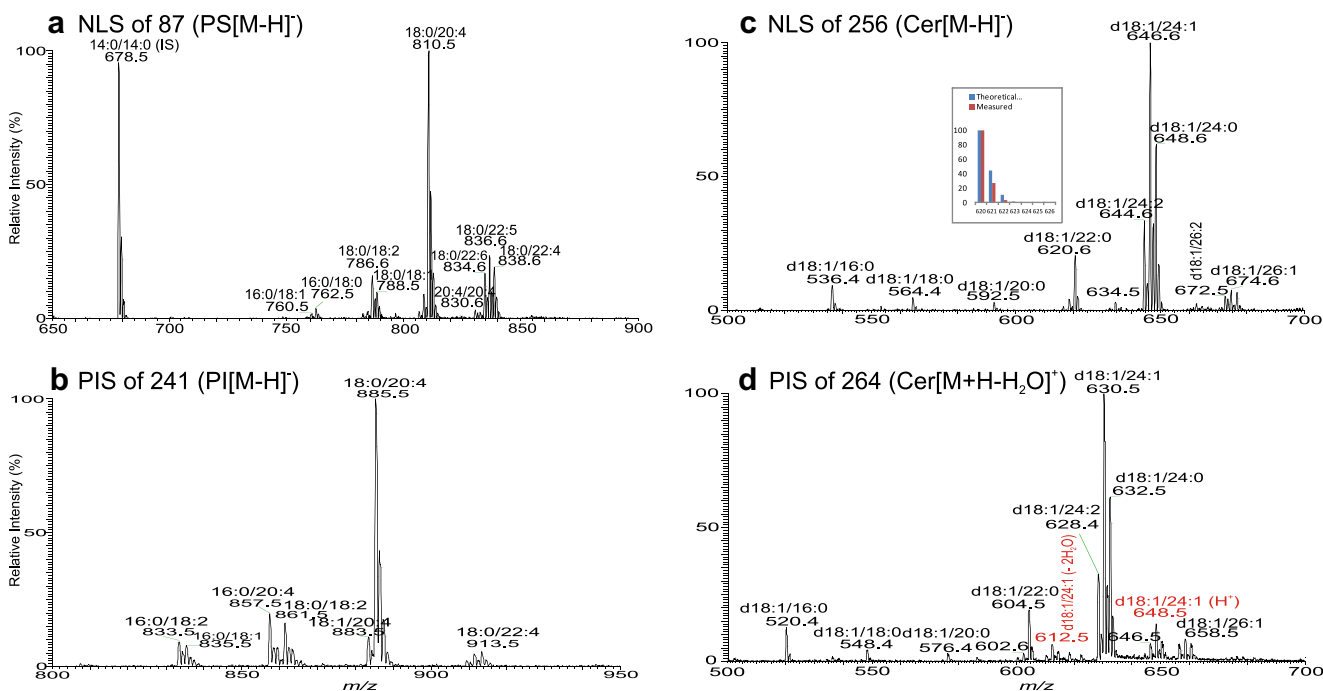
**Fig. 2** The negative ion ESI-MS spectrum of the total lipid extract from Blight & Dyer extraction of human blood via direct infusion of the undiluted solution (**a**), and the same solution diluted three times with methanol (**b**). The drastic difference in the profiles of panels a and b, due to sample dilution reveals the suppression effect, in particular, on the PS species by other lipid classes. Despite the suppression, the profiles of each lipid family remain largely unchanged. Panel **c** that illustrates the tandem MS (seen as  $[M - H]^-$  ions) obtained by use of PIS of 196 (collision energy 50 eV) in the negative ion mode; and the positive ion tandem MS (seen as

$[M + H]^+$  ions) acquired by NLS of 141 (CE 32 eV) accumulated at the first 5 min (**d**), and at the third 5 min (**e**) signal average, all represent the PE species in the sample. The  $Na^+$  build-up during infusion is seen by the drastic increase of the  $Na^+$  adduct ions such as ions at  $m/z$  658, 738, 740, 762, and 790 in Panel **e**. Panel **f** illustrates the tandem mass spectrum acquired by NLS of 129 (collision energy 32 eV) that identifies PE species as the  $[M - H + 2Li]^+$  ions. The profile is similar to that of Fig. 2c, demonstrating its utility in the lipidomic profiling of PE molecules in lipid extract using a TSQ instrument

in Fig. 2a and c are abundant, but the corresponding  $[M + H]^+$  ions expected at  $m/z$  702, 724, and 752 are absent in Fig. 2d and e. Meanwhile, the concurrence of the  $[M + Na]^+$  ions by NLS of 141 from species that are 22 Da lighter overlaps the  $[M + H]^+$  species that are 22 Da heavier. For example, the  $[M + Na]^+$  ions of 16:0/18:1-PE ( $m/z$  740) and 18:0/18:1-PE ( $m/z$  768) have the same nominal masses as the  $[M + H]^+$  ions of 16:0/20:4-PE and 18:0/20:4-PE, respectively, and are inseparable by a quadrupole instrument (Fig. 2d and e). The sodium adduct ions as seen at  $m/z$  658, 740, 762, and 790 in Fig. 2e, which were obtained in the consecutive 2 min scans, also increase, due to

rapid  $Na^+$  build-up as infusion persists. These problems make direct quantitation of PE species in the lipid extracts by NLS of 141, in particular for samples of biological origin where  $Na^+$  content is often abundant and unpredictable, not reliable. To overcome these shortcomings, strategies including derivatization of the amino head group to enhance sensitivity before shotgun lipidomic analysis have been recently described [50, 77]. Alternatively, NLS of 129 can be used to profile PE species as  $[M - H + 2Li]^+$  ions (Fig. 2f) with good sensitivity (Table 2) [69].

The profiles of the ESI-MS can vary significantly depending on the concentration of the sample introduced to the ESI



**Fig. 3** The tandem mass spectra obtained by NLS of 87 (collision energy 25 eV) (**a**), and by PIS of 241 (collision energy 40 eV) (**b**) that profile the  $[M - H]^-$  ions of PS and PI species, respectively, in the lipid extract from blood. Panel **c** and Panel **d** illustrate the MS/MS spectra obtained by NLS of 256 (collision energy 32 eV) that represent the  $[M - H]^-$  ions of d18:1-ceramide family and by PIS of 264 (collision energy 32 eV) that detect the same ceramide family as the  $[MH^+ - H_2O]^+$  ions in positive ion mode (please note: in the same scan, ceramides in the forms of  $[M + H]^+$  (e.g.,  $m/z$  648) and  $[M + H - 2H_2O]^+$  (e.g.,  $m/z$  612) were also seen as the minor

species), respectively. Tandem mass spectra from NLS (**a** and **c**) and PIS (**b** and **d**) readily fish out the entire specific lipid families; and with the addition of internal standard, semi-quantitation of PS (**a**), PI (**b**), and Cer (**c** and **d**) is feasible. Another notion is that spectra from NLS (**a** and **c**) are more resolved than those from PIS (**b** and **d**), and the isotope cluster pattern of all NLS and PIS spectra follows the elemental composition of which the fragments are subtracted (an example is shown for the ion of  $m/z$  620 in panel **c**, inset)

source for ionization, regardless of the sample inlet via infusion or loop injection. For example, the relative intensities of the ions of  $m/z$  678 (14:0/14:0), 810 (18:0/20:4), and 836 (18:0/22:5) in the PS family acquired with the original extract are significantly lower (Fig. 2a) compared with the intensities of the same ions acquired with the same extract that was diluted two times with methanol (Fig. 2b). The relative abundance of PS rises even higher in the ESI-MS acquired with further diluted samples (data not shown). These results may be attributable to the suppression effect on PS from other major lipid classes such as PC. Therefore, signal response in ESI-MS is not necessarily a good reflection of the concentration of various lipid families in a mixture but is rather a good reflection of the concentration in the same lipid family. For example, spectra from NLS of 87 (Fig. 3a) and from PIS of 241 (Fig. 3b) that respectively profile the  $[M - H]^-$  ions of PS and PI species in the sample are very similar to their ESI-MS profiles (Fig. 2a and b), regardless of the sample concentration (e.g., Fig. 2b is obtained from the less concentrated sample). The MS/MS spectrum profiles from NLS of 256 (Fig. 3c) that represent sphingosine (d18:1-LCB) ceramide species observed as the  $[M - H]^-$  ions, from NLS of 48 (data not shown), and from PIS of 264 (Fig. 3d) that

respectively detects the same ceramide family as the  $[M + Li]^+$  [78] and  $[M + H - H_2O]^+$  ions [79] in the same sample, are similar. This feature permits simultaneous semi-qualitative and semi-quantitative analysis of these lipids in mixtures with addition of internal standards. This is the underlying principle of shotgun lipidomics employing a tandem quadrupole instrument [76]. However, variation in sensitivity among the various species in the same families dependent on the structures, such as the type of the radyl group (e.g., PE as described earlier), number of double bond, chain length, etc., makes accurate quantitation impossible (see later for more discussion), despite several approaches such as the addition of multiple internal standard for each lipid class, and implementation of collision energy ramp have been used for more accurate quantitation [25, 39, 40, 80] with the tradeoff of simplicity and speed.

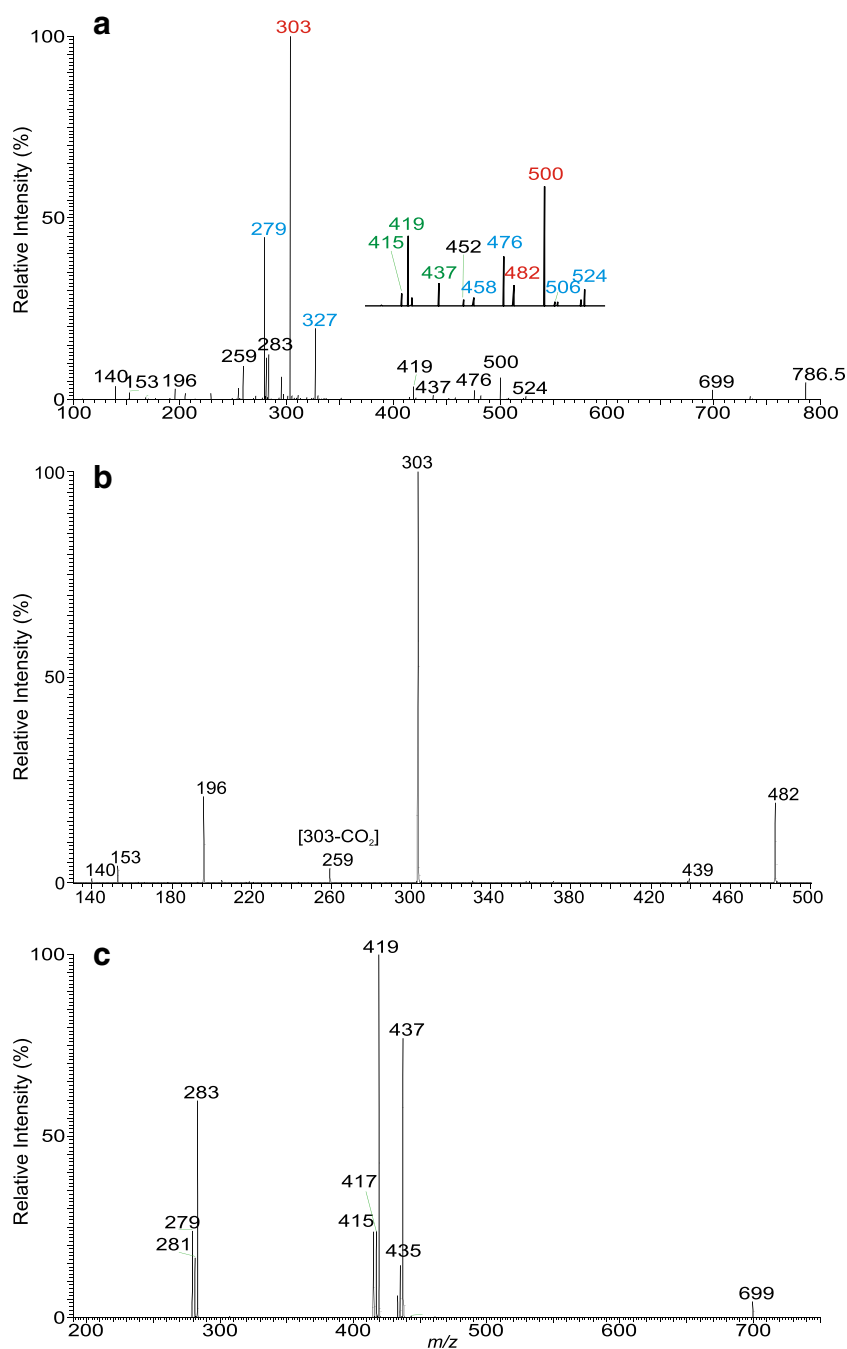
With the combination of PIS and NLS, isobaric isomers from various lipid classes can be extracted. For example, ions of  $m/z$  786 are present in the spectrum of NLS of 87 that recognizes PS species (Fig. 3a) and in the PIS spectrum of 196 (Fig. 2c) that detects PE molecules in the same lipid extract from plasma (Fig. 2a). High resolution Orbitrap MS spectrum showed two separated peaks that match 40:8-PE



(measured  $m/z$ : 786.5080; calculated C<sub>45</sub>H<sub>73</sub>O<sub>8</sub>NP: 786.5079) and 36:2-PS (measured  $m/z$ : 786.5293; calculated C<sub>42</sub>H<sub>77</sub>O<sub>10</sub>NP: 786.5291), respectively. The HCD product-ion spectrum of  $m/z$  786 obtained by Orbitrap (Fig. 4) contains the ion pairs at 303/303, representing 20:4-FA carboxylate anions, and ions at  $m/z$  500 and 482, arising from loss of 20:4-FA as ketene and acid, respectively, together with ions at  $m/z$  196 and 140 that are signature ions of PE [76, 81]. The presence of these ions led to the identification of the major 20:4/20:4-PE isomer. The

spectrum also contains the ion pairs of  $m/z$  524/506 and 476/458, arising from losses of 18:2- and 22:6-FA substituents, respectively, along with the ion pairs at  $m/z$  279/327, representing 18:2- and 22:6-FA carboxylate anions, respectively, indicating the presence of 18:2/22:6-PE isomer. The confirmation of the 20:4/20:4-PE isomer is further conducted by MS<sup>3</sup> on the ion of  $m/z$  482 (786 → 482) (Fig. 4b), which yielded a predominant ion at  $m/z$  303 (20:4-FA carboxylate), along with ions of  $m/z$  439 (loss of aziridine), 196 and 140 that signify a PE molecule [76, 81].

**Fig. 4** (a) The HCD product-ion spectrum of the ion of  $m/z$  786 obtained by an Orbitrap. The ion mainly consists of both a 20:4/20:4-PE and 18:2/22:6-PE, together with minor 18:0/18:2- and 18:1/18:1-PS isomers. The presence of 20:4/20:4-PE is further supported by the MS<sup>3</sup> spectrum of the ion of  $m/z$  482 (786 → 482; panel b), which contains signature ions of PE at  $m/z$  196, 153, and 140 (also seen in Fig. 4a), and  $m/z$  439 (loss of aziridine). The presence of minor (18:0/18:2 + 18:1/18:1)-PS isomers was further confirmed by the MS<sup>3</sup> spectrum of the ion of  $m/z$  699 (786 → 699; panel c). See text for the detail leading to the above structural assignments



In the same spectrum (Fig. 4a), the ion at  $m/z$  699 arose from elimination of a serine residue (loss of [serine – H<sub>2</sub>O]; 87 Da), indicating the presence of PS species [82]. Further dissociation of the ion of  $m/z$  699 (786 → 699) (Fig. 4c) gave rise to ion pairs of  $m/z$  437/419 and 433/415 arising from losses of 18:2- and 18:0-FA substituents as ketene and acid, respectively, consistent with the observation of the ions of  $m/z$  279 and 283, representing 18:2- and 18:0-FA carboxylate anions, respectively. The results confirm the presence of 18:0/18:2-PS, consistent with the presence of the ions of  $m/z$  437 and 419 in Fig. 4a. Meanwhile, the spectrum (Fig. 4c) also contained the minor ions of  $m/z$  281 (18:1-carboxylate anion) and 435/417 pairs, arising from losses of 18:1-FA substituent as a ketene and an acid, respectively, indicating the presence of a minor 18:1/18:1-PS isomer. The above results demonstrate the utility of multiple stage mass spectrometry, in particular combined with high resolution mass spectrometry in the shotgun lipidomic approach towards identification of complex lipid structures.

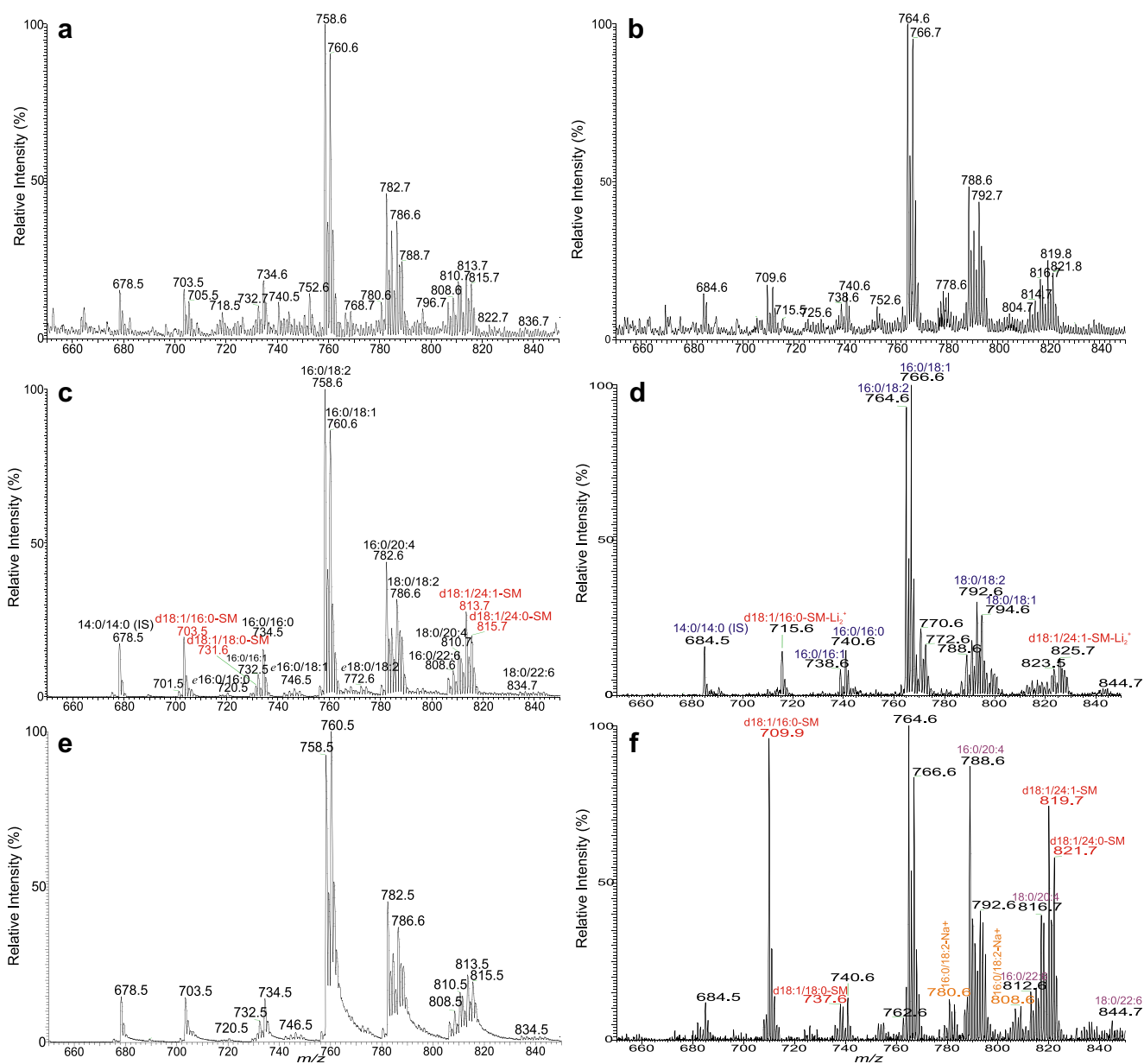
The ESI-MS profiles of [M + H]<sup>+</sup> ions (Fig. 5a) and [M + Li]<sup>+</sup> ions (Fig. 5b) of PC from the same extract are nearly identical. Tandem MS employing PIS of 184 (Fig. 5c) is also nearly identical to the ESI-MS, and the method has been used for quantitating and profiling PC molecules in lipid extracts [12]. However, NLS of 189 to profile PC as [M + Li]<sup>+</sup> ions (Fig. 5d) discriminates against polyunsaturated fatty acid containing PC species such as 16:0/20:4-PC, while NLS of 183 favors polyunsaturated fatty acid containing PC molecules (Fig. 5f) [83]. Information from combination of these scans provides structural recognition of PC species [84], which is the concept behind the multi-dimensional MS-based shotgun lipidomics [25]. NLS and PIS can be useful in the profiling of specific and minor lipid classes due to removal of background ions. For instance, minor sphingomyelin subclasses with various long-chain base such as d18:2-LCB containing SM in bovine erythrocytes have been identified by NLS of 427, a specific scan to detect d18:2/FA-SM [58, 67].

Lipid profiling using a triple quadrupole instrument by use of NLS resulted in a well resolved tandem mass spectrum because the simultaneous scan of Q1 and Q3 is similar to double focus acquisition [85]. For example, NLS of 256 [61, 62] with Q1 and Q3 mass window setting at 0.7 to 1 Da (half-height width) to profile sphingosine (d18:1) ceramide species in the same lipid extract (Fig. 3c) yields a unit mass resolved mass spectrum with very low background. In contrast, the spectrum from PIS of 264 (Fig. 3d), an approach to detect the same ceramide family as the [MH<sup>+</sup> – H<sub>2</sub>O]<sup>+</sup> ions in the positive ion mode [79] is less resolved, and the background ions are higher. Similarly, NLS spectra with baseline resolved peaks were seen in Fig. 5d and f; while Fig. 5c obtained by PIS is less resolved.

It should be noted that to obtain a near baseline resolved PIS spectrum, e.g., the spectrum shown in Fig. 5c using, for

example, a Thermo TSQ instrument under optimal collision energy (e.g., 32 eV for PC), a much lower Ar collision gas pressure (ca 0.6 mTorr) [6] must be applied to reduce multiple collisions that may deteriorate the resolution due to energy spread, while full width at half maximum (FWHM) of Q1 was set at 0.7 Da and Q3 can be set between 0.7 to 1.5 Da to maximize the sensitivity. The spectrum is in sharp contrast to Fig. 5e, which is obtained using the same condition but with Ar collision gas pressure of 1.2 mTorr, the target gas pressure optimized to maximize the CID fragmentations for acquiring a NLS spectrum or a product-ion spectrum. The spectrum (Fig. 5e) is poorly resolved, resulting in loss of sensitivity and mass accuracy in the analysis. It is also important to note that the isotope cluster pattern of all NLS and PIS spectra follows the theoretical plot of the elemental composition of the fragment ions (i.e., the mass of the scanned residue is to be subtracted) rather than that of the elemental composition of the precursor because fragment ions were scanned during data acquisition [86, 87]. For example, the observed isotopic abundance of the ion of  $m/z$  678.5 (Fig. 5c) representing a [M + H]<sup>+</sup> ion of 14:0/14:0-PC (elemental composition: C<sub>36</sub>H<sub>73</sub>O<sub>8</sub>NP) arising from PIS of 184 (a loss of phosphocholine residue; C<sub>5</sub>H<sub>15</sub>NPO<sub>4</sub>) matches the formula of C<sub>31</sub>H<sub>58</sub>O<sub>4</sub>, where a C<sub>5</sub>H<sub>15</sub>NPO<sub>4</sub> residue was subtracted from C<sub>36</sub>H<sub>73</sub>O<sub>8</sub>NP rather than its molecular formula of C<sub>36</sub>H<sub>73</sub>O<sub>8</sub>NP. The isotopic abundance of ceramide species detected as the [M – H]<sup>–</sup> ions derived from NLS of 256 (loss of C<sub>16</sub>H<sub>34</sub>NO) [62] follows an elemental composition of [M – H – C<sub>16</sub>H<sub>34</sub>NO] (Fig. 3c). Thus, for example, d18:1/22:0-Cer at  $m/z$  620 (elemental composition: C<sub>40</sub>H<sub>78</sub>NO<sub>3</sub>) in the NLS of 256 spectrum follows the isotopic pattern of C<sub>24</sub>H<sub>46</sub>NO (C<sub>40</sub>H<sub>78</sub>NO<sub>3</sub> – 256), of which the first isotopic [M – H + 1]<sup>–</sup> ion is 27.03% (relative intensity), which is significantly different from 44.92%, a plot from the molecular formula of C<sub>40</sub>H<sub>78</sub>NO<sub>3</sub> (see Fig. 3c, inset for comparison). Therefore, to perform de-isotoping to achieve accurate quantitation using data from PIS and NLS, care should be taken to assure that correct elemental formula is used for de-isotoping [88].

To quantitate and characterize TAG as [M + Li]<sup>+</sup> ions, Han and Gross described a multi-dimensional ESI tandem quadrupole mass spectrometric approach to identify the fatty acid substituents utilizing a combination of NLSs of various fatty acid substituents in the molecules [89]. The method assigned the major structures, but not the minor species. Another drawback is that the scanning method is nonspecific. For example, cleavages of 18:0-fatty acid (C<sub>17</sub>H<sub>35</sub>CO<sub>2</sub>H) substituent and of 18:3-fatty acid substituent as Li salt (i.e., loss of C<sub>17</sub>H<sub>29</sub>CO<sub>2</sub>Li) all resulted in loss of 284, and thus NLS of 284 fails to distinguish a R<sub>x</sub>COOH loss from a loss of R<sub>x</sub>COOLi residue, which is nevertheless not a major fragmentation process observed for the [M + Li]<sup>+</sup> ions of TAG [57, 90]. An alternative approach to fingerprint TAG molecules in a mixture is that TAGs are to be desorbed in the presence of



**Fig. 5** The positive ion ESI-MS spectrum of the  $[M + H]^+$  (a), and  $[M + Li]^+$  (b) ions of the total lipid extract from Blight & Dyer extraction of human blood via direct infusion, and the tandem mass spectra obtained from PIS of 184 (collision energy 32 eV) with a target gas (Ar) pressure of 0.6 Torr (c), and of 1.2 Torr (e). Panel c bears a nearly identical profile as the ESI-MS (a), indicating the utility of PIS of 184 in the semi-quantitation analysis of PC as  $[M + H]^+$  ions. However, severe peak tailing was seen in Panel e, deteriorating the accuracy in the PC analysis. The profile of the ESI-MS spectrum of the  $[M + Li]^+$  ions of the same lipid extract (b) after addition of LiOH is similar to Panel a, but the tandem MS obtained from NLS of 189 (d) is discriminating against the

PC species consisting of polyunsaturated fatty acid substituents, as species of 18:0/20:4 (788), 16:0/22:6 (812), 18:0/20:4 (816), and 18:0/22:6 (844) are of low abundance; while these ions are abundant in the NLS spectrum of 183 (f) due to the preferential neutral loss of 183. Similar results were also observed for the SM family. Other notions include observation of the dilithiated species of SM as shown in panel d, and of the sodiated species of, e.g., 16:0/18:2 (780), 18:0/18:2 (808) in panel f revealing the pitfalls of this method in the PC analysis. It should be noted that in the structural assignments, only the major isomer is shown, and the identified structures as marked have been verified by product ion spectra and high resolution mass spectrometry

$NH_4^+$  as described earlier. In this manner, exclusive  $[M + NH_4]^+$  adduct ions with a much greater sensitivity than that seen as the  $[M + Li]^+$  ions can be formed. The identification of

the structures by multi-dimensional mass spectrometry can be established by NLSs of  $R_xCOONH_4$ . Because  $[M + NH_4]^+ - R_xCOONH_4$  ions are nearly the exclusive and most

prominent fragment ions in the product-ion spectrum obtained by a tandem quadrupole instrument [91], multi-dimensional ESI tandem quadrupole mass spectrometric analysis of TAG as  $[M + \text{NH}_4]^+$  ions [92], therefore, affords a more explicit and more sensitive method than that desorbed and analyzed as the  $[M + \text{Li}]^+$  ions. However, CID on  $[M + \text{Li}]^+$  ions applying multiple stage tandem mass spectrometry provides near complete structural information, including the location of double bond of the fatty acid substituents and their regio-specificity on the glycerol backbone [57]. The structural complexity of TAGs in biological specimen is often a subject of discussion and the approaches toward their structural identification and quantitation have been reviewed [93–95].

Owing to its relative low cost and ease of operation, shotgun lipidomics with ESI-TSQ mass spectrometric approach, irrespective of the method (direct infusion or loop injection with a syringe pump) for sample inlet, provides the simplest method that can be easily implemented in a lab with limited resources, even without an HPLC or other more sophisticated logistics.

### Shotgun lipidomics with ESI-high resolution mass spectrometry (HRMS)

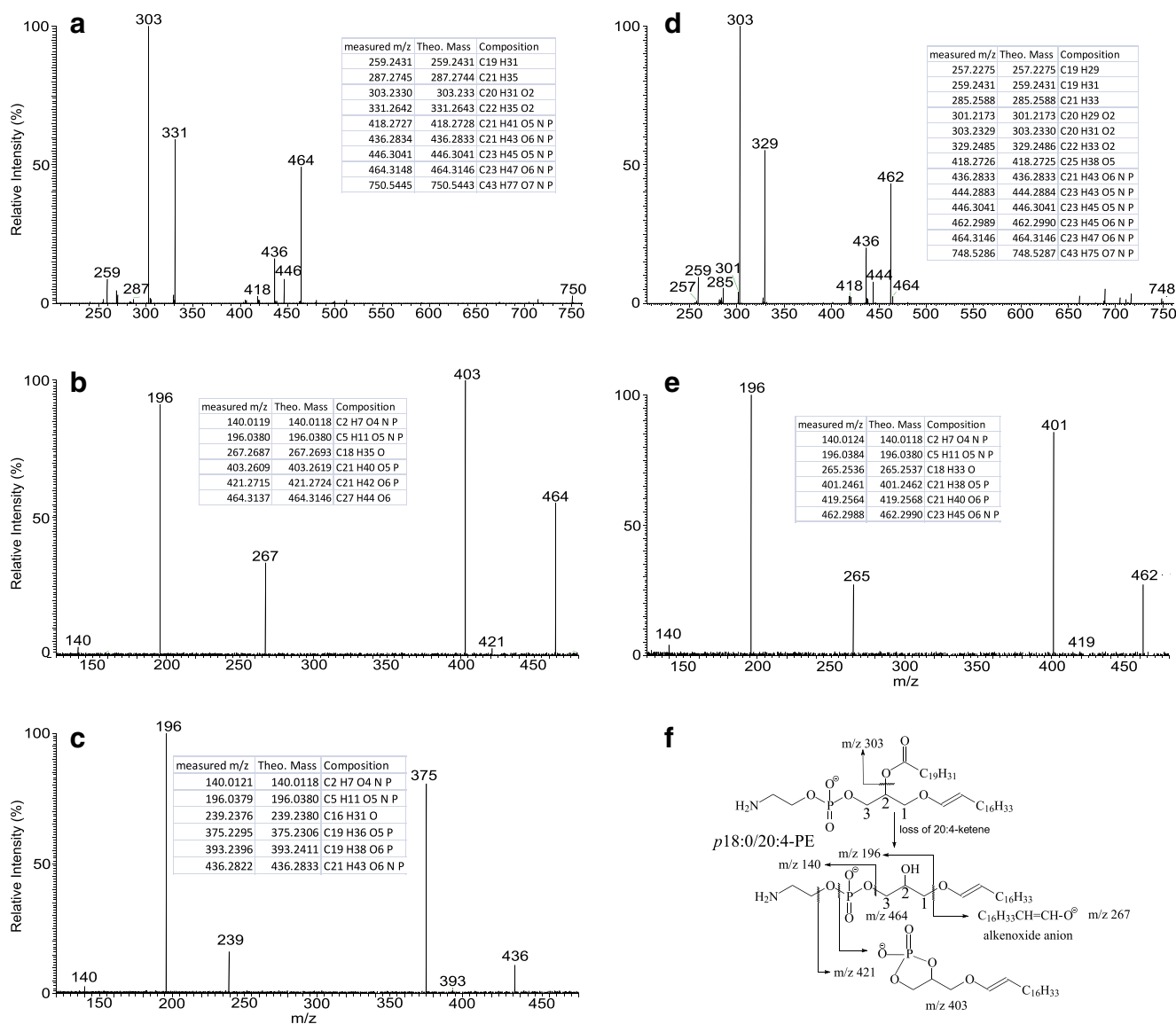
Nowadays high resolution (HR) mass spectrometry based on FT (such as FT ICR and Orbitrap) or quadrupole time of flight (Q-TOF) instrumentation has emerged as another shotgun lipidomic tool [21, 24, 39, 43, 49, 71, 96], providing an additional gain in selectivity. The trade-off of high resolution is at the cost of sensitivity and speed. Another downside of a high resolution instrument is the hefty cost.

High field FT ICR and the most recent advances in Orbitrap technology can achieve over 1,000,000 resolution [97], providing the resolution necessary for separating the isotopic isomers. A Q-TOF instrument, on the other hand, provides speed but a typically much lower resolution (<40,000). The insufficient resolution for “high resolution shotgun lipidomics (e.g., <100,000 resolution at  $m/z$  400)” causes the partial overlap between the  $^{13}\text{C}_2$ -containing isotopologue species and the monoisotopic ion of the similar lipid species with one less double bond, resulting in significant peak broadening and the shift of the mass measurement, thereby hampering accurate quantitative analysis if the peak intensity is used to compare to that of an internal standard for quantification, as well as the accuracy of the measured  $m/z$  that can be used for database search [98].

Direct measurement of intact masses does not allow their unequivocal assignment to individual species or even to the lipid classes. However, HRMS is rather effective in the realization of adduct ions [99], leading to recognize artificial ion formation upon ESI on lipid mixture (also see related discussion on Fig. 6); and also very efficient in the differentiation, for example, of ether glycerophospholipids (i.e.,

plasmalogen and plasmanyl lipids) from odd-chain containing diacyl glycerophospholipids [81, 100]. High resolution  $\text{MS}^3$  spectra further recognize the formation of alkenoxide anions that can be used to identify plasmalogen and distinguish it from plasmanyl lipid. For example, the high resolution ( $R = 100,000$  at  $m/z$  400)  $\text{MS}^2$  spectrum of the ions of  $m/z$  750.5 (Fig. 6a) from the same blood lipid extract seen in Fig. 3a gave rise to 464/446 and 436/418, arising from losses of 20:4- and 22:4-FA substituents as a ketene and an acid, respectively (in Fig. 6, all insets show the accurate mass measurements of the fragment ions). The spectrum also contains the ions of  $m/z$  303 and 331, arising from cleavage of 20:4- and 22:4-FA substituents, along with ions of  $m/z$  287 ( $\text{C}_{21}\text{H}_{35}^-$ ) and 259 ( $\text{C}_{19}\text{H}_{31}^-$ ), arising from further loss of  $\text{CO}_2$  from 20:4- and 22:4-carboxylate anions, respectively. The high resolution  $\text{MS}^3$  spectrum of the ions of  $m/z$  464 ( $750.5 \rightarrow 464$ ; Fig. 6b) contains the ions of  $m/z$  403 (loss of ethanolamine), 196 and 140 confirming the presence of the phosphoethanolamine head group, together with the ion of  $m/z$  267 (both measured and calculated  $\text{C}_{16}\text{H}_{33}\text{CH}=\text{CH}-\text{O}^-$  were seen at  $m/z$  267.2693), an octadecenoxide anion arising from cleavage of the 1-O-octadecenoyl substituent (see Fig. 6f for the fragmentation scheme). The results confirm the presence of the major p18:0/20:4-PE isomer. Similarly, a  $\text{C}_{14}\text{H}_{29}\text{CH}=\text{CH}-\text{O}^-$  alkenoxide anion at  $m/z$  239 was observed in the HR  $\text{MS}^3$  spectrum of  $m/z$  436 ( $750.5 \rightarrow 436$ ; Fig. 6c), along with ions of  $m/z$  375, 196, and 140, pointing to the presence of a minor p16:0/22:4-PE isomer. The  $\text{MS}^2$  spectrum of the ions of  $m/z$  748.5 (Fig. 6d) is similar, consisting of the  $m/z$  462/444 and 436/418 ion pairs along with ions of  $m/z$  303 and 329, consistent with the presence of the 20:4, and 22:5-FA substituents at sn-2. The high resolution  $\text{MS}^3$  spectra of the ions of  $m/z$  462 ( $748.5 \rightarrow 462$ ; Fig. 6e) contains the alkenoxide ion of  $m/z$  265 ( $\text{C}_{16}\text{H}_{31}\text{CH}=\text{CH}-\text{O}^-$ ), leading to the assignment of p18:1/20:4-PE structure; while the  $\text{MS}^3$  spectrum of  $m/z$  436 ( $748.5 \rightarrow 436$ ) (data not shown) is identical to Fig. 6c, revealing the presence of the alkenoxide ion of  $m/z$  239 that identifies the p16:0/22:5-PE isomer. A minor  $m/z$  464/446 ion pair arising from loss of 20:5-FA substituent is present in Fig. 6d, consistent with the observation of  $m/z$  301 and 257 ( $301 - \text{CO}_2$ ).  $\text{MS}^3$  on the ion of  $m/z$  464 ( $748.5 \rightarrow 464$ ) gave rise to  $m/z$  267, and the spectrum is similar to Fig. 6b (data not shown). These results led to assignment of a minor p18:0/20:5-PE isomer. The above results demonstrate that implementing HR linear ion-trap multiple stage mass spectrometry (LIT  $\text{MS}^n$ ) for lipidomic analysis readily led to confident structure assignments, revealing the presence of many isomeric structures, including very minor isomers.

High resolution shotgun lipidomics using data-dependent acquisition with a QqTOF instrument performing multiple precursor ion scanning or a LIT Orbitrap applying HCD tandem MS [24, 49, 71], and data-independent acquisition scans with a LIT Orbitrap performing CID/HCD  $\text{MS}^n$  [35] all



**Fig. 6** (a) The high resolution ( $R=100,000$ , at  $m/z$  400) CID product-ion spectrum of the ion of  $m/z$  750 obtained by an Orbitrap instrument, and its  $MS^3$  spectra of the ions of (b)  $m/z$  464 ( $750 \rightarrow 464$ ) and (c) 436 ( $750 \rightarrow 436$ ). Panel d shows the high resolution CID product-ion spectrum of the ion of  $m/z$  748, and (e) its  $MS^3$  spectrum of the ion of  $m/z$  462 (748

$\rightarrow 462$ ). The proposed fragmentation processes from  $MS^2$  and  $MS^3$  leading to assignment of p18:0/20:4-PE structure are shown in Panel f. The inset Tables are the HR mass measurements of the fragment ions, the theoretical masses and their deduced elemental compositions

generate huge quantities of data, which require vast logistics support to make the analysis possible, in addition to the hefty price tag of the high-end mass spectrometers and the chip-based infusion device (e.g., the Advion TriVersa NanoMate); therefore the approach is very difficult to implement. Another limitation is that the genuine linked PIS and NLS features that are readily available to a TSQ instrument is not applicable to today's HRMS instruments. Recently, Synder et al. developed a single analyzer neutral loss scan and parent ion scan in a linear quadrupole ion trap that extends the application of IT technology to another dimension in MS analysis. However, whether the technology can be extended to the FT-based HRMS (e.g., Orbitrap) and to lipidomic analysis remains to be seen [87, 101].

## Shotgun lipidomics using MALDI-TOF mass spectrometry

MALDI-TOF has been used for analysis of various lipid classes [102–105]. The ease of instrument operation and sample preparation, the speed of the analysis, and freedom from crossover sample contamination, make MALDI-TOF an ideal complementary tool for shotgun lipidomic analysis. In this context, MALDI-TOF has allowed researchers to engage in imaging lipids, mapping the distribution of various lipids in different tissues successfully [10, 106–109]. MALDI-TOF equipped with a reflectron and delayed extraction achieves >20,000 resolution and offers good mass accuracy for

assigning intact molecular ions of lipid species. The downside of this technology, at present, is lack of reproducibility and lack of universal matrix for lipid analysis. Although 2,5-dihydroxy benzoic acid,  $\alpha$ -cyano-4-hydroxy-cinnamic acid, 9-amino-acridine, and 5-chloro-2-mercaptobenzothiazole have been the most commonly used [102], the selection of matrixes so far has mainly relied on trial and error. The interference of matrix ions in the mass range <600 Da, the unpredictable adduct ion formation (i.e., alkali metal and matrix-lipid adduct ions), and ion suppression of low abundance molecules by high abundance or easily ionizable species also deter its application in the analysis of crude lipid extracts [110]. Because of the drastic differences in ionization efficiency among different lipid classes due to their physical states such as the polarity and salt content, the ion intensity in the MALDI-TOF spectra is also not always a good reflection of the actual relative abundances of the molecules in a mixture. MALDI-TOF also generates artifacts, such as DAGs from prompt fragmentation of TAGs. Nevertheless, MALDI-TOF provides a fast, easy, and useful tool for profiling complex lipid mixture, in particular microbial lipids such as lipid A [111] and phosphatidylinositol mannosides [112]. More recently, an innovative laser-based post-ionization approach coined as MALDI-2, which utilizes a second post-ionization laser beam for secondary MALDI-like ionization of the plume of the matrix, and the analyte molecules generated by the standard MALDI laser was introduced [113]. This innovation was found to increase up to 2 orders of magnitude sensitivity for many lipid classes and minimize ion suppression from MALDI-1. However, the application is mainly focused on imaging, and the use for shotgun lipidomic analysis has not been reported [113, 114]. Another recent innovation to overcome ion suppression of targeted lipid analytes and low mass ions by compounds that exhibit more efficient ionization and/or are present at high abundance is a nanoparticle-assisted laser desorption/ionization (LDI) MS approach using, e.g., dopamine-modified TiO<sub>2</sub> monolith, as an alternative to traditional MALDI matrixes. This approach has been successfully applied to localize small metabolites (molecular mass below 500 Da) and lipids in the mouse brain for MS imaging. Again, whether this method can be used for shotgun lipidomic fingerprinting remains to be seen [109].

MALDI coupled to TOF/TOF instruments has limited success in the structural analysis of complex lipid structures [115–117]. Fragment ions seen in MALDI-TOF and MALDI-TOF/TOF spectra are often from post-source decay (PSD) of the precursor ions [118]. Fragment ions from MALDI-TOF PSD are broad and the low  $m/z$  fragment ions are less detectable. To obtain product ions with a wide range of energies produced by PSD, the curved-field reflectron time-of-flight/time of flight (MALDI TOF/ReTOF) mass spectrometer was invented [119, 120], and structural analysis of a variety of lipids have been reported [115, 116, 121]. Similarly,

the MALDI TOF/TOF instrument developed by Bruker Daltonics used “LIFT” technology, which allows all the fragment ions from laser-induced dissociation followed by CID with 8 KeV energy, to be detected in a single run [122]. All the above designs allow unit mass resolution of fragment ions, providing structural information complimentary to that obtained by low energy CID. For example, tandem TOF mass spectra contain rich structural information for locating the methyl side chain, functional group(s), and the alkyl tail of long chain fatty acids [123]. However, the application in the structural analysis of lipid in a mixture is constrained by the fact that the precursor ion selection window of today’s instruments (MALDI TOF/TOF instruments from Bruker and Shimadzu alike) fails to isolate two ions that are 2 Da apart for further dissociation. Thus, product-ion spectra from two different ion species that are 2 Da apart in a mixture, such as two similar lipid species differed by a double bond, are indistinguishable [123].

Recently, a spiral MALDI-TOF/TOF instrument with 60 K resolving power was developed [124] to allow monoisotopic isolation of the precursor ions that can be further subjected to CID with 20 keV energy to obtain tandem mass spectrum for near complete characterization of several lipid compounds [125–128]. This innovation may revive the interest in employing high energy CID tandem mass spectrometric approach for structural identification of complex lipids. However, the conversion efficiency (precursor to product ion) remains low, which is typical of a high energy CID process.

In comparison with MALDI TOF shotgun lipidomics, ambient ionization techniques that do not require matrix such as DESI (desorption electrospray ionization) [129, 130] and REIMS (rapid evaporative ionization mass spectrometry) [131] have recently entered the rather crowded shotgun lipidomic arena, indicating the increasing awareness of the importance of this field. All these approaches target lipid analysis or lipid imaging using the similar ESI method with modification, with the mission that lipid profiling can be completed within a short period of time with the simplest sample preparation to the norm of shotgun lipidomics, rather than the comprehensive analysis conducted by the conventional approach. All these methods focus mainly on the clinical application [132–134] and seldom have been seen in the advance of the knowledge of lipidomics.

## Long-standing issues in shotgun lipidomics

### Artificial lipid ion species generated by electrospray ionization

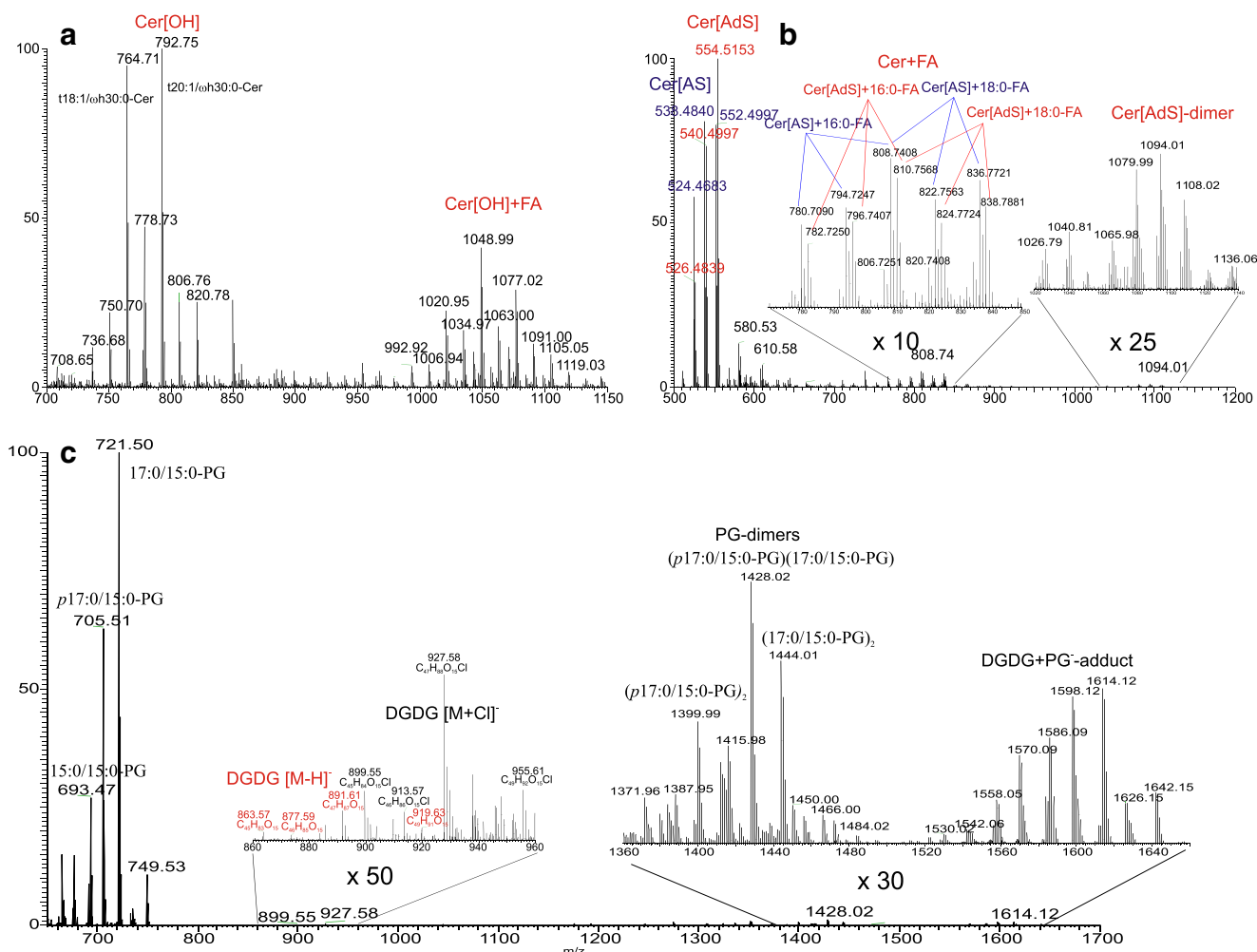
In shotgun lipidomics, it should be noticed that artifacts generated by the ESI process are inevitable, resulting in faulty

structural identification. For examples, PA, lyso-PPL (including lyso-CL), DAG, MAG, and fatty acids can be generated from glycerophospholipids, and extensive dissociation has been observed for e.g., ceramide in the positive-ion mode (loss of H<sub>2</sub>O) and acyl-CoA. Meanwhile, dimer formation and adduct ion formation across different lipid species in the same family and different lipid families also commonly occur, adding complexity and uncertainty to the ESI MS-based shotgun lipidomic analysis. Such formations have been seen for nearly all the lipid families, in particular, when ionization occurred in a more concentrated lipid condition. The degree of formation also depends on the lipid classes and the content of the solvents that dissolve the lipid. For example, PC forms various adducts with various anionic ions [135], and these adduct ions have been exploited for structural characterization, applying CID tandem mass spectrometry [100, 135]. In the presence of Cl<sup>-</sup>, the d18:1/hFA-Cer subclass forms adduct ion with Cl<sup>-</sup> (formation of [M + Cl]<sup>-</sup>) much more preferentially than does the d18:1/nFA-Cer subclass [62] (Cl<sup>-</sup> can arise from CHCl<sub>3</sub> as solvent that dissolves lipids when ionization takes place). Skin ceramides in the Cer[OH] family such as t18:1/ωhFA-Cer at *m/z* 764 and t20:1/ωhFA at *m/z* 792 [136] form unusual adduct ions with long chain fatty acid (RCO<sub>2</sub><sup>-</sup>) such as palmitic anion (C<sub>15</sub>H<sub>31</sub>CO<sub>2</sub><sup>-</sup>) to form [M + C<sub>15</sub>H<sub>31</sub>CO<sub>2</sub>]<sup>-</sup> ions at *m/z* 1020 and 1048 (Fig. 7a). Cer[AS] (e.g., d16:1/αh16:0-Cer, d17:1/αh16:0-Cer and d18:1/αh16:0-Cer at *m/z* 524, 538, and 552, respectively) forms Cer[AS]-FA adduct ions at *m/z* 780, 794, 808 with 16:0-FA and at *m/z* 808, 822, and 836 with 18:0-FA (Fig. 7b, inset at left), and Cer[AdS] (e.g., d16:0/αh16:0-Cer, d17:0/αh16:0-Cer, and d18:0/αh16:0-Cer at *m/z* 526, 540, and 554, respectively) forms the similar Cer[AdS]-FA adduct ions (ions seen at *m/z* 782, 796, 810, and at 810, 824, and 838), but Cer[AdS] also form dimers such as ions at *m/z* 1080, 1094, and 1108, while dimer formation for Cer[AS] family (e.g., ions at *m/z* 1048, 1062, 1076, 1090, and 1104) were not observed (Fig. 7b, inset at right). These adduct ions are readily identified by HRMS (Fig. 7b). In a lipid extract from *Listeria monocytogenes* cells containing high content of PGs [137], the aggregation between PG and diglucosyl diacylglycerol (DGDG) to form PG-DGDG dimer ions of *m/z* 1614, 1598, etc. (Fig. 7c, inset) is much more favorable than formation of the [M + Cl]<sup>-</sup> ions seen at *m/z* 927, 913, and 899, which were also more abundant than the corresponding [M - H]<sup>-</sup> ions of DGDG seen at *m/z* 891, 877, 863, etc. Thus, these adduct formations severely compromise the sensitivity of DGDG, if it is to be detected as the [M - H]<sup>-</sup> ions in the negative ion mode. Meanwhile, formation of PG dimers (in the same sample) not only occurred across the various species but also is species-dependent. For example, the ion of *m/z* 1428 representing a (17:0/15:0-PG)-(p17:0/15:0-PG) dimer is more abundant than the same species dimer ion of *m/z* 1444 ([17:0/15:0-PG]<sub>2</sub>), which is more abundant than *m/z* 1400 ([p17:0/15:0-PG]<sub>2</sub>) (Fig. 7c,

inset at right). In the current shotgun lipidomic analysis, such formations are often left unaccounted for because of the complexity of the ion formation, and the high mass cutoff in a MS scan that the high *m/z* covering dimer ions is not scanned. These problems underscore the limitation of shotgun lipidomics towards a very precise lipid analysis.

### The structural details of the long-chain fatty acid substituents and the stereochemistry of lipid

Another long-standing issue in shotgun lipidomic analysis is that the structural details of the fatty acyl substituents, including the location of functional groups such as methyl, hydroxyl, double bond(s), and/or cyclo side group(s) along the chain (if present), as well as the cis/trans geometry and the stereochemistry such as region-specificity are often left unreported because they are very difficult to define. The cis/trans geometry cannot be addressed by any MS-based shotgun lipidomics means. One study reported a very slight difference in the spectra generated by electron impact excitation of the ion pairs of a cis and trans monounsaturated fatty acyl formed by ESI for their distinction. However, the difference is within the instrument variation during acquisition [138]. A speculation is that possible differences in the reaction rate and branching (i.e., the generated fragment ion pairs induced by ozone) ratio between cis/trans monounsaturated fatty acid pairs upon subjected to ozone-induced dissociation (OzID) mass spectrometry may exist, and these differences may be potentially used for their differentiation [139]. Both high energy (KeV) and low energy CID tandem mass spectrometry have been successfully applied to decipher the structural details of long chain fatty acids and the fatty acyl substituents inside complex lipids with or without derivatization [57, 123, 140–142]. However, tremendous effort is required, increasing the throughput time. Infusion/OzID mass spectrometry is useful and effective in the location of double bond(s), but the approach is difficult to implement (e.g., modification of instrument is required) [54]. Although infusion/OzID mass spectrometry is applicable to locate unsaturated bond(s) in the lipid molecules with simplest structures, application to lipids possessing multiple acyl groups with heterogeneous unsaturated fatty acyl chains, or present in multiple isomeric forms that are commonly seen in the lipid samples of biological origins, is yet to be demonstrated. The implementation of OzID ion trap mass spectrometry has shown that ions from MS<sup>2</sup> can be isolated and further subjected to OzID, resulting in location of double bond on another alkyl chain, but with drastic decrease in sensitivity [54]. Most recently, an HPLC/ion mobility separation has been implemented to the device to improve the technique, and different lipid classes and isomers in a mixture of standard lipids can be analyzed in a single run [53]. However, whether this method is applicable to the more complex lipid mixture remains to be seen. Another innovation implementing on-line



**Fig. 7** The high resolution (HR) ESI MS of the [M - H]<sup>-</sup> ions of **(a)** skin Cer[OH] family (e.g., ions at  $m/z$  764 and 792), and the adduct ions formed with 16:0-FA (e.g., ions at  $m/z$  1020 and 1048); and of **(b)** skin Cer[AdS] (e.g., ions at  $m/z$  524, 538, and 552) and Cer[AdS] (e.g., ions at  $m/z$  526, 540, and 554) subclasses and the adduct ions formed with FA (panel **b**, **inset** at left). Cer[AdS] also forms dimer anions (panel **b**, **inset** at right). The formation of these adduct ions is consistent with HR mass measurements that gave the elemental composition of the sum of the two species ((FA [M - H]<sup>-</sup> ion + Cer) or (Cer [M - H]<sup>-</sup> ion + FA) alike). For example, aggregation of ions of  $m/z$  524.4683 (formula: C<sub>32</sub>H<sub>62</sub>O<sub>4</sub>N), 538.4840 (C<sub>33</sub>H<sub>64</sub>O<sub>4</sub>N), and 552.4997 (C<sub>34</sub>H<sub>66</sub>O<sub>4</sub>N) in the Cer[AdS] family with

16:0-FA at 256.2402 (C<sub>16</sub>H<sub>32</sub>O<sub>2</sub>), gave ions at  $m/z$  780.7090 (C<sub>48</sub>H<sub>94</sub>O<sub>6</sub>N), 794.7247 (C<sub>49</sub>H<sub>96</sub>O<sub>6</sub>N), and 808.7496 (C<sub>50</sub>H<sub>98</sub>O<sub>6</sub>N). The measured deviations by an Orbitrap Velos are within 1 ppm. All adduct ion formations are readily revealed by HRMS (**inset** at left). Panel **c** shows the ESI MS of the [M - H]<sup>-</sup> ions of PG and diglucosyl diacylglycerol (DGDG), and its Cl<sup>-</sup> adduct ions (**inset** at left) in a lipid extract from *L. monocytogenes* cells. Formation of PG-DGDG dimers (**inset** at right) is also present in higher abundance than DGDG desorbed as the [M - H]<sup>-</sup> ions (label in red). Panel **c** (**inset** at right) shows the ions of PG dimeric anions formed from across species (PG-pPG), or same species (PG-PG) aggregations

photochemical derivatization (Paternò-Büchi reaction) with tandem mass spectrometry into a shotgun lipidomics workflow has demonstrated the promise in which the double bond positions of the long chain fatty acids and of the fatty acid substituents inside the glycerophospholipids can be elucidated and quantitation of isomeric structures in a mixture can be performed [38]. This approach first applied acetone as the Paternò-Büchi (PB) reagent for 250 nm UV irradiation to add acetone to the intact unsaturated lipids to form PB products (add 58 Da), which was subsequently subjected to low energy CID to yield fragment ions by cleavages of the original C=C bonds, permitting location of double bond(s). This innovation requires no instrument modification and can

be easily implemented and is suitable in a shotgun lipidomics setting, and data interpretation is relatively simple [38]. The method appears to be most applicable to glycerophospholipids with homoconjugated unsaturated fatty acid moieties. Polyunsaturated long chain fatty acids such as 20:4-FA desorbed as [M - H]<sup>-</sup> ions are not applicable but can be achieved in the less well formed [M + Li]<sup>+</sup> form [38]. In this regard, this approach may not be that much more superior than the other tandem mass spectrometric approaches that characterize the same molecules as metal adduct ions or as derivatives using high energy/low energy CID [117, 141–143].

Tandem mass spectrometry is effective in determining the regio-specificity, e.g., the sn-1/sn-2 position of the fatty acyl



substituents on the glycerol backbone of glycerophospholipids [9]. However, it cannot address the purity of the molecule if both isomers (e.g., both A/B and B/A in sn-1/sn-2 configuration) coexist.

### Quantitation challenges in shotgun lipidomics

The ever-increasing sensitivity and improvement in resolution and the linear range of modern instruments make today's mass spectrometry instrument a powerful tool for lipid quantitation. However, to achieve accuracy and precision in shotgun lipidomic quantitation requires not only the specific details in execution of the experiments but also a long process time that is against the norm of shotgun lipidomics. A complete coverage of shotgun lipidomic quantitation is another subject matter; and will not be addressed in detail in this review. An insightful article covering all aspects of lipidomic quantitation by Wang and coworkers is recommended to readers [80].

To obtain absolute or relative quantitation, internal standard(s) has to be added before lipid extraction. Ideally, a stable isotope labeled analogous molecule, in particular  $^{13}\text{C}$ -labeled counterpart, can be used because of its unique property that can behave in the same manner as the analyte during sample preparation and MS analysis (this quantitation strategy is coined as “stable isotope dilution”). For global lipid quantitative analysis, however, it is not possible that all the stable isotope standards are available and can be added simultaneously. Hence, the stable isotope dilution strategy can only be suitable for very few targeted lipid quantitations, and can only be fulfilled with a higher cost (isotope labeled standard is usually expensive), if available.

A compromised approach without using stable isotope internal standards to achieve a more accurate quantitation is the use of non-isotope labeled internal standards such as homologous lipids, which are either chain shortened species or odd carbon number fatty acyl containing species that normally are either rare or absent in the biological samples. The short-chain homolog internal standards, for example, 1,2-dimyristoyl (14:0) or dimyristoleoyl (14:1) [13, 84, 144] containing phospholipid, is typically chosen so that the molecular weight of the internal standard is away from but close to that of the endogenous species. Thereby, a narrower mass scan range (more sensitivity due to the increased duty cycle) and less discriminatory scan (less chain length dependency) can be performed.

Not all lipid molecular species can be detected with equal efficiency, i.e., the response of mass spectrometer to the lipid is dependent on the lipid class, the chain length, the lipid concentration, and the unsaturation state of the acyl chain of the molecule [80, 145]. To address this problem, a minimum of two internal standards or three with one (di)unsaturated fatty acyl chain that cover the entire mass range have been suggested to be included for each lipid class to correct the

dependency problems [80, 145]. However, the inclusion of so many internal standards adds to the complexity, and significantly increases the data analysis time. Additionally, it is also time-consuming to prepare a cocktail including optimal amounts of all the internal standards to be added (i.e., trial and error workup procedure is required).

Recently, shotgun lipidomics with ion-mobility adopting the standard shotgun lipidomics workflows was implemented, enabling to separate and identify isomeric and isobaric lipids independent of the resolution settings of the MS analysis to improve the accuracy in quantitation [27]. This approach leads to highly comprehensive and quantitative lipidomic analysis via Q1 or MRM scan in a triple quadrupole instrument for rapid lipid profiling. The MRM approach is complimentary to the high resolution MS approach without adding much analysis time and significantly increases the sensitivity and specificity. Despite the improvement by addition of the separation step, the inclusion of internal standard(s) for precise quantitation is inevitable.

Absolute quantitation remains a significant challenge even if reference compounds are available and thus the response factors for all the lipid species to be measured can be precisely estimated. In contrast, relative quantitation, e.g., assessing fold change in the biological study, where the signal of the target lipid relative to the internal standard is to be evaluated, even a single internal standard could possibly be used for correction of the variations caused by sample preparation, ionization process, or CID ion chemistry. This is by far the most reachable goal for modern shotgun lipidomic quantitation.

### Conclusions

“Shotgun lipidomics” intends to achieve quantitative and qualitative analysis of the heterogeneous lipids of biological origin within a short period of time using mass spectrometric techniques without chromatographic separation. The power of this techniques can be envisioned by the completion of the entire lipidome of various cells [13, 43, 146–149]. The technique combined with other “-omic” techniques such as genomics, has also led to the realization of many important roles that lipids play in the biological system [15, 37, 39, 150–154].

As described, each shotgun lipidomic method has its advantages, disadvantages, and limitations. “Shotgun lipidomics” mainly addresses all the major lipid content and the targeted lipids; minor species and the unexpected new lipids are often left unreported. LC/MS lipidomic approach, on the other hand, adds another dimension in the analysis, thus, cross-over lipid class interference, dimer formation across different lipid classes, and adduct ion formations that often confound structural analysis can be minimized and more structural coverage and more accurate structural assignment

can be achieved. The advance of liquid chromatography-ion mobility-mass spectrometry (LC-IM-MS) offers an additional dimension to separate lipid species by molecular size, shape, and charge. Thereby, more structural information and more confident structural assignments can be accomplished [31]. Nevertheless, the addition of HPLC or other separation methods for more complete lipidomic analysis can only come with a trade-off of the “speed”, which is the hallmark of “shotgun lipidomics”.

Nowadays many Web resources such as cyberlipid center (<http://www.cyberlipid.org/>) and AOCS lipid library (<http://lipidlibrary.aocs.org/>) provide invaluable information for lipid studies. To fulfill high throughput mass spectrometry lipidomic analysis, database search to enable quick retrieval of relevant information from thousands of records in a lipid database becomes increasingly important. Tools that have been developed for analysis of MS-based lipidomic data include MS and MS/MS data by various labs [88, 155–158], and lipid consortium such as LIPID MAPS Lipidomics Gateway ([www.lipidmaps.org/resources/tutorials/databases.html](http://www.lipidmaps.org/resources/tutorials/databases.html)), and National Institute of Standards and Technology (NIST) (<http://chemdata.nist.gov/mass-spc/ms> search); and commercialized software such as Lipidizer™ (<https://sciex.com/applications/life-science-research/lipidomics/lipidizer-platform>), LipidSearch™ (<https://www.thermofisher.com/order/catalog/product>), and SimLipid® ([www.premierbiosoft.com](http://www.premierbiosoft.com)) also become available. However, user discretion is advised to avoid data misinterpretation.

The continuous advance in mass spectrometry techniques and development in bioinformatics, and more importantly, the ceaseless efforts of many elite researchers in this field make shotgun lipidomics a reality. However, caution should be exercised to avoid false-positive and false-negative identification in the shotgun lipidomic analysis (As a reviewer for many prestigious journals in the field of lipidomics for many years, this author has received many manuscripts that contained errors in the assignment of lipid structures, utilizing database tools including both free online search such as LIPIDMAP and commercial database. The errors mainly stemmed from authors who lack basic knowledge in either MS or lipid biochemistry, and some errors are inconceivable). After all, insight into the exact biochemical events and complex biochemical interactions taking place at the cellular level is more than lipidomic analysis, which is far beyond being a layman’s term.

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## Compliance with ethical standards

**Conflict of interest** The author declares no conflict of interest.

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