

In vitro evaluation of selenium genotoxic, cytotoxic, and protective effects: a review

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Abstract Selenium is an oligoelement with essential biological functions. Diet is the most important selenium source, and intake of this element depends on its concentration in food and amount of food consumed. Among the essential human micronutrients, selenium is peculiar due to its beneficial physiological activity and toxicity. It may have anticarcinogenic effects at low concentrations, whereas at concentrations higher than those necessary for nutrition, it can be genotoxic and carcinogenic. Because of that, selenium is probably the most widely investigated of all the oligonutrients. In the last decades, there has been increasing interest in several nutritional Se compounds because of their environmental, biological, and toxicological properties, particularly for their cancer- and disease-preventing activities. This article gives an overview of the results of in vitro studies on mutagenicity, genotoxicity, cytotoxicity, and DNA repair conducted within the last decades with different organic and inorganic selenium compounds. Results from these studies provide a better knowledge on the selenium activity and help to elucidate the reasons underlying its duality in order to regulate its correct use in nutrition and clinic.

Keywords Cytotoxicity · DNA repair · Genotoxicity · Mutagenicity · Selenium

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Introduction

Selenium

Selenium (Se) is an essential trace element for humans, animals, and some bacteria. It is important for many cellular processes, because it is a component of several selenoproteins with essential biological functions (Letavayová et al. 2008a). There are at least 25 human selenoproteins and 24 in the mouse, each characterized by the incorporation of selenium into the primary sequence as the amino acid selenocysteine (SeCys; Kryukov et al. 2003; Foster et al. 2006). Some of these selenoproteins are selenoenzymes, such as thioredoxin reductase and glutathione peroxidase, which represent fundamental antioxidative systems for the maintenance of cellular redox homeostasis (Rayman 2000). Thus, Se functions in the body as an antioxidant, in thyroid hormone metabolism, redox reactions, reproduction, and immune function (Rayman 2000; Combs et al. 2009).

Nevertheless, Se is also toxic, and chronic exposure in humans or animals results in selenosis (Goldhaber 2003). Other related toxic effects are a disruption of endocrine function, synthesis of thyroid hormones and growth hormones, and an insulin-like growth factor metabolism (Navarro-Alarcon and Cabrera-Vique 2008). The mechanism of Se toxicity has not been clarified but mostly attributed to its ability to induce oxidative stress both in vitro and in vivo (Kitahara et al. 1993; Yan and Spallholz 1993).

Diet is the most important Se source, and intake of this essential element depends on its concentration in food and amount of food consumed (Navarro-Alarcon et al. 2005). Se bioavailability varies according to the Se source and nutritional status of the subject, being significantly higher for organic Se forms (Navarro-Alarcon and Cabrera-Vique 2008). Se contents in several foods are gathered in Table 1.

Table 1 Se content ($\mu\text{g}/\text{kg}$, range or mean) in main food from different countries. Source: adapted from Combs (1988), Amodio-Cochieri et al. (1995), Marro (1996), Murphy and Cashman (2001), McNaughton and Marks (2002), Sirichakwal et al. (2005), Pappa et al. (2006), and Panigati et al. (2007)

	Beef	Bread	Cereals	Cheese	Chicken	Eggs	Fish	Fruits	Milk	Pork	Rice	Vegetables
Australia	72–121	92.6–125	62.9	70–78.9	116–280	190–414	20–632	4.5–76	2.5–25.9	94–205	25	0.5–32
Canada	30–310 ^a	–	10–1,350	60	150	60	46–1,570	1–23	10	–	–	10–119
Finland	10–70 ^a	–	5–115	10–40	50–100	110–180	180–980	2–30	2–20	–	–	1–2
Germany	130–280 ^a	–	30–880	100	150	180	240–530	10–41	10	–	–	4–98
Greece	33.5–63.1	37.9–150.2	19.1–20.2	14.3–127.9	76.3–82.4	56.4–181.1	28.7–519.9	1.1–7.9	10.7–22.2	90–98.2	17.7–20.5	1.2–15.9
Ireland	61–105	15–158	–	9.5–11.5	86–147	56–282	268–298	–	14–22	82–129	10–17	10–38
Italy	15–446 ^a	12	0–43	30–140	15–416	29–89	118–293	1–13	10–35	–	20.1	1–25
New Zealand	22.3–83	31.6–59.4	–	23	137–145	157–161	195–512	–	1–14	19.3–150	0	0–2.5
Thailand	72–226	–	–	–	156–271	145–420	196–1,137	–	19–36	142–250	29–65	1–127
UK	30–76	43–92	20–530	7.4–12	60–70	90–120	200–500	5	10–15	140	4–13	3–22
USA	134–190	282–366	300–560	13.9	190–276	225–308	126–502	1–13	20–21	144–450	75	1–1,180

^a Se content in red muscle meats

Se content in food is influenced by geographical location, seasonal changes, protein content, and food processing. As a result, Se levels in foods can vary manyfold not only between countries but also between regions in a country. A food may have more than tenfold difference in Se content, depending on where it was produced (Reilly 2006).

Se intake is mainly in the form of organic compounds ingested in grains, meat, yeast, and vegetables (Cao et al. 2004). The US Food and Nutrition Board (1980) considered to be the Estimated Safe and Adequate Daily Intake for Se of 50–200 μg , being 55 $\mu\text{g}/\text{day}$ the Recommended Dietary Allowance (RDA) for Se for both men and women. Nevertheless, per capita intakes of Se can vary widely between countries. A major reason for this is the difference in food consumption patterns and, especially, in the types of staple foods consumed (Reilly 2006). The no observed adverse effect level (NOAEL) of dietary Se was estimated to be 1,540–1,600 $\mu\text{g}/\text{day}$ (Whanger 2004). At these doses, Se has the potential to induce toxic side effects such as induction of DNA damage (Reid et al. 2004; Wycherly et al. 2004).

The major chemical forms of Se are organic, as selenomethionine (SeMet), SeCys, and methylselenocysteine (MeSeCys), and inorganic as selenite and selenate (Letavayová et al. 2006). Studies on the short-term effects of Se showed that inorganic Se (selenate, selenite) is more toxic and less bioavailable than organic forms (Letavayová et al. 2008a), and one potential reason for differences in genotoxicity observed among selenocompounds is their distinct metabolism (Suzuki et al. 2006a, b). However, Se toxicity depends not only on the Se compound and dose but also on the method of administration, animal species, exposure time, idiosyncrasy, physiological status, and interaction with other metals, nutrients, etc. (Burk and Levander 2002).

In the last decade, there has been increasing interest in several nutritional Se compounds because of their environmental, biological, and toxicological properties, particularly for their cancer- and disease-preventing activities. Among the essential trace mineral nutrients, Se is unique due to its catalytic activity and toxicity (Letavayová et al. 2008a). Hamilton (2004) reported the existence of three Se levels of biological activity: (1) trace concentrations are required for normal growth and development; (2) moderate concentrations can be stored, and homeostatic functions maintained; and (3) elevated concentrations can result in toxic effects. As a general rule, Se may have anticarcinogenic effects at low concentrations, whereas at concentrations higher than those necessary for nutrition, it can be genotoxic and carcinogenic (Bronzetti et al. 2001), and its toxic level is relatively close to that required for normal health (FAO 2001). Nevertheless, the mechanisms of action of Se compounds, either via a prooxidant pathway, as seen in cytotoxicity and apoptosis, or via an antioxidant pathway, as proposed in cancer chemoprevention, are still unclear but intriguing (Shen et al. 2001; Hurst et al. 2008).

Selenium controversy

In the last two decades, there has been much progress in our knowledge and understanding of the biological roles of Se and its importance in human nutrition (Navarro-Alarcon and Cabrera-Vique 2008). Different chemical forms of this element produced genotoxic effects in a great variety of in vitro and in vivo studies (Ammar and Couri 1981; Biswas et al. 2000; Cemeli et al. 2006); nevertheless, Se genotoxicity still generates controversy, and IARC (1987) concluded that there were not sufficient data to consider Se as carcinogen for humans. On the other hand, several organic

and inorganic Se compounds have been reported to be effective chemopreventive agents against multiple models of animal tumorigenesis (Letavayová et al. 2008a). However, despite its antioxidant properties and requirement for human and animal nutrition, the appropriate form of Se for supplementation continues to be debated, as well as the optimal concentrations of Se that provide protection against genetic damage with the least toxicity (Santos and Takahashi 2008).

Results of the two great trials performed with Se to date are really representative of this controversy. A randomized controlled trial, the Nutritional Prevention of Cancer study (Clark et al. 1996), demonstrated substantial reduction in the risk of several cancers, most notably cancer of the prostate (Duffield-Lillico et al. 2002), among *subjects* supplemented with Se in the form of selenized yeast. These clinical data, supported by other epidemiological and pre-clinical data, led to intense interest in the potential of the Se as a non-toxic mean of preventing prostate and other cancers. On this basis, the largest cancer prevention trial ever performed (Selenium and Vitamin E Cancer Prevention Trial, SELECT) was designed to test the hypothesis that daily use of Se (as SeMet) or vitamin E, alone or in combination, could prevent prostate cancer in more than 32,000 men (Klein et al. 2000). However, against all predictions, SELECT recently reported that neither Se nor vitamin E had any beneficial effect on major health outcomes (Lippman et al. 2009).

Given this considerable controversy generated by the results of these and many other Se studies, a more detailed characterization of the effects of this element is required to define the conditions in which they appear and to be able to establish proper standards of Se use. This is especially imperative nowadays, since many people consume Se supplements. On the other hand, there is no international consensus on how to evaluate the potential risk of genotoxic carcinogens in food; moreover, oligoelements present at low levels in food, as Se, frequently give rise to difficulties to evaluate the potential risk of genotoxicity (O'Brien et al. 2006). In this regard, *in vitro* studies have become increasingly important in the last decades, since they can complement and provide more specific information than those performed *in vivo*.

Evaluation of selenium activity: *in vitro* vs. *in vivo* tests

As the human being is continuously exposed to several chemical, physical, and biological agents, there is a need to evaluate diverse types of biological alterations in order to thoroughly assess the genotoxic/mutagenic potential of a substance, and this requires the use of a battery of *in vivo* and *in vitro* assays (Maurici et al. 2005). Data obtained from animal experiments yield information pertaining to

the dose for lethal or sublethal toxicity which corresponds to many different general toxic mechanisms and effects. The information derived from *in vivo* studies is essential for determining the potential toxicity of a chemical to humans and other life forms (Barile 2008). The *in vitro* systems must try to imitate the real organism conditions, but many times, *in vitro* metabolic activation systems cannot mimic entirely the mammalian *in vivo* conditions. The test, therefore, does not provide direct information on the mutagenic and carcinogenic potency of a substance in mammals.

Thus, the use of animals in mutagenicity testing is primarily required when it is necessary to investigate whether mutagenic activity detected *in vitro* is reproduced *in vivo*. However, except in those cases in which high, or moderate and prolonged human exposure is expected (e.g., many human medicines), there is no justification for the routine use of animals for mutagenicity tests when there is no evidence for activity at *in vitro* assays (COM 2000). Moreover, there are compounds for which standard *in vivo* tests do not provide additional useful information. This includes compounds for which data from studies on toxicokinetics or pharmacokinetics indicate that they are not systemically absorbed and therefore are not available for the target tissues in standard *in vivo* genotoxicity tests. In all these cases, it may be appropriate to base the evaluation only on *in vitro* testing (FDA 2007).

For that reason, although current *in vitro* methods are not considered sufficient to serve as full animal replacements at this time, different institutes are currently developing *in vitro* tests able to predict compound effects *in vivo*. Standardized and validated *in vitro* methods have replaced or reduced some human and animal studies (Carfi et al. 2007). In case of Se, for example, *in vitro* bioaccessibility methods of simulated digestion are an alternative to *in vivo* bioavailability procedures for calculating the percentage of an element which is transformed into absorbable forms in the digestive tract. *In vitro* bioaccessibility analytical procedures are often useful, because they are simple, rapid, inexpensive, and allow individual experimental variables to be easily controlled (Cabrera et al. 1996).

In general terms, the *in vivo* and *in vitro* tests are equally necessary, since both provide relevant information on the characterization of the action mechanisms of an agent. However, there are certain situations where it is preferable to use one kind of test and not another. *In vivo* tests are especially important to evaluate the toxicokinetic effects or the metabolism alterations in the organism, for example. Nevertheless, *in vitro* methods are significantly faster and less expensive than *in vivo* assays; moreover, animal studies require a high number of individuals and raise important ethical concern (Carfi et al. 2007). The advantages of *in vitro* studies include the rapid assessment of large numbers of chemicals, the suggestion of a mechanism

for carcinogenicity or mutagenicity, the reduction, replacement, and refinement of animal testing, and a contribution to human and animal risk assessments that correlates as well as the predictive ability of animal toxicology testing (Barile 2008).

Next sections of this review describe the results of the in vitro studies carried out in the last years to evaluate and characterize Se molecular activity.

Toxic effects of selenium

Several in vivo, in vitro, and epidemiological studies describe adverse effects of Se. It was found that Se can induce DNA damage (Biswas et al. 2000; Machado Mda et al. 2009), produce oxidative stress (Wycherly et al. 2004), and increase lipid peroxidation (Colado-Megía et al. 2004), generate neurotoxicity in mice (Ammar and Couri 1981), provide no protection against adverse actions induced by other compounds as arsenic (Hasegkar et al. 2006) or sodium metavanadate (Zwolak and Zaporowska 2009), be ineffective at preventing basal cell carcinoma, and increase, in a non-significant way, the incidence of squamous cell carcinoma and total non-melanoma skin cancer in people supplemented with Se (Duffield-Lillico et al. 2003). However, many of these effects depend on the Se level and chemical form, and results of in vitro studies could help to delimitate those conditions in which toxicity becomes evident.

Selenium mutagenicity and genotoxicity

An effective strategy in mutagenicity and genotoxicity assessment uses tests that produce reproducible and biologically relevant data based upon three stages (Barile 2008). Stage 1 uses bacterial gene mutation assays, stage 2 assays are cytogenetic tests that monitor clastogenicity and aneugenicity, and stage 3 assays record the induction of gene mutations in cultured mammalian cells.

Bacterial mutagenesis assays are the most widely used short-term tests for screening for potential mutagens and carcinogens. They are highly sensitive for genotoxic agents, technically easy, fast, and inexpensive (Barile 2008). Several studies aimed at evaluating the genotoxicity of Se by means of bacterial mutagenesis assays are collected in the literature. Inorganic forms of Se, such as selenites, do not give any indication of being mutagenic in the *Salmonella*/microsome assay despite of producing positive results in the *Bacillus subtilis* rec-assay (Nakamuro et al. 1976; Lofroth and Ames 1978). However, in other studies, selenate and selenite were shown to be weakly mutagenic, giving rise to base-pair substitution (Noda et al. 1979), and high concentrations of selenite induced mutagenicity in the *S. typhimurium* strain TA104 (Kramer and Ames 1988).

Besides a few Se studies using aquatic invertebrates (Tran et al. 2007), fish cell lines (Al-Sabti 1994) or plants (Yi and Si 2007), the mainly eukaryotic non-mammalian cell system used to test mutagenicity and genotoxicity is the yeast. Sodium selenite mutagenicity and genotoxicity was early detected in different strains of *Saccharomyces cerevisiae* (Rosin 1981; Anjaria and Madhavanath 1989). Rosa et al. (2004) combined the use of the *Salmonella*/microsome assay and the yeast *S. cerevisiae* to test for putative mutagenicity, genotoxicity, and recombinogenicity of diphenyl diselenide and to determine whether DNA damage produced is repairable. They showed that this Se compound is a weak mutagen which probably generates DNA strand breaks through both an intercalating action and a prooxidant effect. In a more recent study, the effects of Se (sodium selenite and SeMet) at the genetic level were analyzed by means of a *S. cerevisiae*-based assay (Seitomer et al. 2008). They determined which genes are involved in responding to high environmental Se using a collection of viable haploid null allele strains representing the major stress pathways. Results suggested that both selenite and SeMet are likely inducing DNA damage by generating reactive species. Letavayová et al. (2008a) characterized three different nutritionally available Se compounds (sodium selenite, SeMet and MeSeCys) for their toxicity and mutagenicity as well as potential detrimental effects on DNA using the budding yeast *S. cerevisiae*. Only sodium selenite manifested significant toxic effect in yeast, and this effect was accompanied by a promutagenic activity observed only in the stationary phase of growth. The data also suggested that, in inducing oxidative DNA damage, sodium selenite may generate double-strand breaks in replicating yeast cells (Letavayová et al. 2008a).

Despite its usefulness, it is not always possible to extrapolate the results of bacterial or yeast assays to the mammalian system. In such case, or in order to complete these results, mammalian cell systems should be employed. Machado Mda et al. (2009), for example, observed that DFDD (3',3'-ditrifluoromethyldiphenyl diselenide) is not mutagenic for bacteria or yeast; however, it may induce weak genotoxic effects in V79 cells. Se mutagenicity and genotoxicity has been tested in a great variety of in vitro assays with mammalian cell systems. Results of all of them emphasize the importance of the chemical form (Nakamuro et al. 1976; Sirianni and Huang 1983; Smith et al. 2004) and level (Biswas et al. 2000; Weitberg et al. 1985; Abul-Hassan et al. 2004) in the Se effects.

Nakamuro et al. (1976) tested five Se compounds for their ability to induce chromosome aberrations (CA) in cultured human leukocytes. They all showed chromosome breaking activity, but it was significantly higher for the compounds with four-valent than with six-valent Se, the efficiency being in the decreasing order selenious

acid > sodium selenite > Se dioxide \gg selenic acid > sodium selenate. These results seem to be represented in other subsequent Se studies. Inorganic Se compounds, as sodium selenite, sodium selenate, and sodium selenide, were reported to increase CA rates in different cell lines. Selenite induced CA in human fibroblasts (Lo et al. 1978) and lymphocytes (Biswas et al. 2000; Abul-Hassan et al. 2004; Whiting et al. 1980; Khalil 1989), and DNA damage induced by sodium selenate (Biswas et al. 2000; Whiting et al. 1980) and sodium selenide (Whiting et al. 1980) was also found in human cells. Furthermore, organic forms of Se, as SeMet (Khalil 1989) and other synthetic organo-Se compounds (Khalil and Maslat 1990), have shown their ability to induce CA in human lymphocytes.

The different capabilities of Se compounds (sodium selenide, Se dioxide, Se(0), sodium selenate, and sodium selenite) to induce sister chromatid exchanges (SCE) were clearly demonstrated in an early study of Ray and Altenburg (1980). The SCE-inducing abilities in decreasing order of their effectiveness were Se(0) > Se dioxide > sodium selenide > sodium selenite > sodium selenate. Increases in SCE rates induced by Se, mainly as sodium selenite, were also found in other *in vitro* studies (Sirianni and Huang 1983; Ray et al. 1978; Morimoto et al. 1982; Ray and Altenburg 1982).

Other studies using micronucleus (MN) test also reported genotoxic effects of several Se compounds in different cell lines. Treatment with diphenyl diselenide, for example, induced an increase in the number of MN in V79 Chinese hamster cells, showing mutagenic risk by this molecule at high concentrations (Rosa et al. 2007a). Selenous acid increased MN formation in mouse bone marrow cells (Itoh and Shimada 1996), in human lymphocytes, and in TK6 lymphoblastoid cell line (Cemeli et al. 2006); sodium selenate and sodium selenite also showed genotoxicity in TK6 cells (Cemeli et al. 2006). Nevertheless, some studies have shown that Se does not produce considerable increase in MN frequency (Berces et al. 1993). Moreover, the work of Ebert et al. (2006) on bone marrow stromal cells with low antioxidative capacity concluded that selenite supplementation of cultures appears to be an important countermeasure to restore their antioxidative capacity and to reduce cell damage in the context of tissue engineering and transplantation procedures.

Prooxidant responses of Se compounds have also been reported. DNA damage induced by sodium selenate, sodium selenite, and selenous acid on their own was detected with the single cell gel electrophoresis (comet) assay in human lymphocytes (Cemeli et al. 2003). Results obtained with this test also showed that selenite induced oxidative stress and apoptosis, and these effects were significantly attenuated by superoxide dismutase, catalase and deferoxamine (Shen et al. 1999). Prooxidant activity exhib-

ited by organoselenium compounds when used in relatively high concentrations was suggested to be linked to genotoxicity observed in human leukocytes by the comet assay (Santos et al. 2009). In this study, the organoselenium amino acid derivatives were more genotoxic than the aromatic derivatives. Methylseleninic acid induced apoptosis without induction of reactive oxygen species (ROS) into two prostate cancer cell lines, whereas selenite generated strand breaks in DNA of LNCaP cells and induced apoptosis by producing superoxide to activate p53 (Li et al. 2007). At high doses, diphenyl diselenide also generated DNA strand breaks, as detected using the comet assay (Rosa et al. 2007a). Lu et al. (1995, 1996) also observed by means of filter elution analyses that sodium selenite and sodium selenide induce single and double DNA strand breaks in a mouse mammary epithelial cell line, whereas MeSeCys and Se-garlic extract only induce single-strand breaks and in lesser degree in the same cells. However, no significant genotoxic effect was found for selenite, selenate, SeMet, or Se-MeSeCys in C6 rat glial cells (Yeh et al. 2006), for SeMet in human lymphocytes (Laffon et al. 2009) and human fibroblasts (Seo et al. 2002), and for ebselen in HepG2 (Yang et al. 1999) and V79 cells (Miorelli et al. 2008).

Selenium cytotoxicity: effects on cell cycle and apoptosis

The effect of Se alone or in combination with other compounds on the growth and proliferation of different mammalian cells has been investigated mainly by means of flow cytometry techniques. In an early study, Se, as sodium selenite, was shown to decrease the growth of fibroblasts and hepatoma cells in a dose-dependent manner, and this inhibition was reversible upon removal of Se from the growth medium (LeBoeuf et al. 1985).

Later, it was reported that selenite inhibited cell growth by G₂/M arrest in a mammary tumor cell line (Lu et al. 1995), in human esophageal cancer cells when combined with zinc (Xiao et al. 2008), and in lymphoblastic leukemia MT-4 cells (Philchenkov et al. 2007); however, it promoted cell proliferation at high concentrations (Xiao et al. 2008). An increase in the S-phase fraction in the presence of Se was found in a human maxillary cancer cell line (Yamamoto et al. 1996). The effect of Se-garlic extract and Se-MeSeCys on cell morphology, cell growth, and cell cycle progression was also studied in mammary epithelial cells, both agents inducing growth inhibition by G₁-phase cell cycle arrest (Lu et al. 1996). SeMet also induced G₂/M arrest in certain prostate and colon cancer cell lines (Goel et al. 2006; Zhao and Brooks 2007), methylseleninic acid caused G₀/G₁ arrest in prostate cancer cells (Zhao et al. 2004), and ebselen interfered with both the proton-translocating function and the ATPase activity of the plasma

membrane H^+ -ATPase, inhibiting yeast growth in a concentration- and time-dependent manner (Chan et al. 2007). Another recent work investigated the variability of the effects on cell viability, redox modulation, and disruption of subcellular compartments by different selenocompounds (SeMet, methylseleninic acid, and selenazolidines) in several human lung cell lines (Poerschke et al. 2008). Results of this study demonstrated that all selenocompounds behave different, and that the chemical form of the organic selenocompound is a major determinant in the expected cellular response.

Results from apoptosis studies have shown that several selenocompounds (mainly sodium selenite but also SeMet, Se dioxide, and methylseleninic acid) induce cell death in different mammalian cell lines: human prostate cancer cells (Xiang et al. 2009), lymphoblastic leukemia MT-4 cells (Philchenkov et al. 2007), HepG2 cells (Zou et al. 2007), colon cancer cell lines (Goel et al. 2006), lymphoma cell lines and primary lymphoma cultures (Last et al. 2006), leukemia cell lines (Wang et al. 2004), human pulmonary adenocarcinoma cells (Chen et al. 2003), and brain tumor cell lines (Rooprai et al. 2007). The methylated Se compounds, such as methylselenocyanate or MeSeCys, also induced cell injury and death by apoptosis in a mouse leukemia cell line (Wilson et al. 1992). The precise mechanisms of apoptosis induced by the Se compounds are not well understood (Philchenkov et al. 2007); however, it is believed that ROS may play a crucial role in Se-decreased cell viability and Se-induced apoptosis (Zou et al. 2007).

Shen et al. (2001) designed a study to investigate the interaction effects of selenite and SeMet plus vitamin C, trolox (a water-soluble vitamin E), and copper sulfate, on cell viability and induction of 8-hydroxy-2'-deoxyguanosine (8-OHdG) adduct formation in DNA of primary normal human keratinocytes (NHK). The data showed that selenite, but not SeMet, induced oxidative DNA damage as 8-OHdG adducts, but coinubation with vitamin C or copper sulfate protected NHK cells against that selenite-induced cytotoxicity. However, synergistic effects were observed between selenite and trolox resulting in enhanced cytotoxicity. On the other hand, no effects on cell viability were observed when cells were treated with SeMet plus vitamin C, trolox, or copper sulfate. Previous findings had already shown that high doses of selenite, acting as a pro-oxidant, induced cytotoxicity and DNA adducts in mouse skin cells, whereas SeMet did not (Stewart et al. 1999). Furthermore, other studies reported that selenite and its metabolites at high doses resulted in cytotoxicity, DNA fragmentation (Garberg et al. 1988; Wilson et al. 1992), and cellular apoptosis (Stewart et al. 1997; Davis et al. 1998).

Zhong and Oberley (2001) employed several methodologies (western blot, structural evaluation of mitochondria,

cell growth analysis...) to investigate the effects of Se, as sodium selenite, in the LNCaP human prostate cell line. The data enabled the authors to conclude that the in vitro biological consequences of selenite exposure were different between acute and long-term exposure. In acute exposure, selenite caused cell death, mainly apoptosis attributable to oxidative stress; in chronic long-term exposure, selenite caused only minimal cell death but inhibited cell growth by modifying gene expression and cell cycle progression.

Morris et al. (2006) assayed the BrdU incorporation into DNA of primary epithelial prostate and LNCaP cells treated with SeMet or Se(0) to determine DNA synthesis. The results of the study demonstrated that both chemical Se forms can induce delay in DNA synthesis in a dose-dependent manner in both cell lines. Li et al. (2007) treated two human prostate cancer cell lines with selenite and methylseleninic acid, and results obtained showed that these Se forms induce ROS formation and apoptosis in both cell lines. In another work, the effects of methylseleninic acid on gene expression were evaluated by means of western blot and oligonucleotide array analysis in human prostate cancer cells (Dong et al. 2003). Data showed that Se alters the expression of different important genes inducing an increase in p21^{WAF1} and p19^{INK4d} protein synthesis and a down-regulation of CDK1, CDK2, and cyclin A. This agrees with previous studies which reported that Se can upregulate or downregulate certain genes (El-Bayoumy and Sinha 2005).

Effect of selenium on DNA repair and synthesis

A role for Se in DNA repair was first noticed when Se treatment was shown to enhance host cell reactivation of a UV-damaged reporter plasmid template by enhancing DNA repair protein complexes (Seo et al. 2002). Enhancement of DNA repair could be a mechanism of chemoprevention, and only very few compounds have been yet shown to act by this mechanism (Collins et al. 2003). Furthermore, Zhang et al. (2008) inferred that Se only enhances DNA repair of normal tissues as a consequence of the selective modulation of Se on Nrf2 in tumor and normal tissues (Kim et al. 2007).

Yeasts are very useful and powerful model systems for elucidating many DNA repair phenomena and pathways highly relevant to areas of investigation in human biology (Letavayová et al. 2008a). Human genetic defects associated with DNA repair can often be addressed directly in yeast because of evolutionary conservation of genes and systems (Resnick and Cox 2000). In a recent study, Letavayová et al. (2008b) used *Saccharomyces* to test DNA repair processes and concluded that the Rad52 protein is indispensable for repairing sodium selenite-induced

double-strand breaks, suggesting a fundamental role of homologous recombination in this repair process and providing the first evidence that this pathway may have a fundamental role in the repair of sodium selenite-induced toxic DNA lesions.

On the other hand, the unscheduled DNA synthesis (UDS) test is commonly used to *in vitro* assay the influence of different chemical and physical agents on DNA synthesis and repair processes of mammalian cells (Barile 2008). UDS studies were conducted to determine the effects of Se on cell proliferation and the stages of the cell cycle affected by this element (LeBoeuf et al. 1985). Despite of the fact that many of the *in vitro* studies concluded that Se (mainly as selenite form) induces an inhibition of DNA synthesis (reviewed by Frenkel and Falvey 1988), Se has also been suggested to be a DNA repair promoter (Russell et al. 1980). Whiting et al. (1980) studied the induction of UDS in cultured human cells by different inorganic and organic Se compounds. They found that inorganic compounds (sodium selenate, sodium selenite, and sodium selenide) induced low levels of UDS in absence of glutathione, but high levels of UDS were found in the presence of this peptide. Nevertheless, no UDS was detected in cells treated with organic compounds (selenocystamine or selenomethionine), with or without added glutathione, and only selenocystine induced a low level of UDS, being also enhanced by glutathione. In one recent study, different mammalian cells lines (rat gut epithelial cells, primary mouse bone marrow cells, and human squamous cell carcinoma of the head and neck cells) were treated with SeMet and with a variety of DNA-damaging agents, and then UDS was determined. Data showed that SeMet pretreatment caused a DNA repair response, which protected from subsequent challenge with DNA-damaging agents (Fischer et al. 2007).

The comet assay is another test usually employed in DNA repair studies because of its sensitivity for the measurement of radiation- or chemically induced DNA damage and repair in viable cells (McKelvey-Martin et al. 1993). In this regard, Seo et al. (2002) confirmed by means of this assay that SeMet induces DNA repair in normal human fibroblasts *in vitro* after a challenge with UV-radiation, and Laffon et al. (2009) reported that bleomycin-induced DNA damage in human lymphocytes was repaired better in the presence of SeMet.

DNA synthesis was also evaluated *in vitro* by measuring incorporation of ^3H -thymidine into rat lens following systemic delivery of a cataractogenic dose of selenite. UDS was found to be $\sim 10\%$ of the total DNA formed, but there was a 30 and 70% increase in this putative DNA repair in the lenses from selenite-treated animals at 6 and 24 h after the injection, respectively; ^3H -thymidine incorporation into DNA remained elevated compared to controls through 96 h (Huang et al. 1990). The effect of Se (as Se dioxide) on the

accuracy of DNA synthesis *in vitro* was also analyzed by means of the fidelity assay. Se did not alter fidelity under normal conditions of magnesium activation, nor affected the mutagenicity of manganese (Tkeshelashvili et al. 1980). However, several Se-derived compounds (dimethylselenone, diphenylselenone, sodium selenite, and MeSeCys) reversed the proangiogenesis effect of arsenic, which is initiated at the endothelial cell plasma membrane by activation of the ERK1/2 signal transduction pathway (Mousa et al. 2007).

Verma et al. (2004) demonstrated that gastric adenocarcinoma SNU-1 cells responded to SeMet with a biphasic proliferative curve: enhanced incorporation of ^3H -thymidine into DNA within a very narrow range of SeMet concentrations, followed by decreased ^3H -thymidine uptake at higher levels. This biphasic effect of Se on cell growth was also observed in another previous *in vitro* study (Medina and Oborn 1984): some Se concentrations stimulated cell growth, whereas others were cytotoxic, and the inhibition of cell growth by Se was reversed when these doses were removed from the growth medium. The increased cell growth was reflected by an increased cell number, increased uptake of ^3H -thymidine into DNA, increased DNA labeling index, and increased rate of DNA synthesis. The differential effects of Se were manifested by 48 h after the addition of Se to the cell culture medium.

As general conclusions from results of different assays to study the influence of Se on DNA synthesis, it seems that it depends mainly on the cell line employed (Webber et al. 1985; Vadgama et al. 2000), the chemical Se form (Whiting et al. 1980; Frenkel 1985; Bansal and Sood 1999), and the Se concentrations assayed (Medina and Oborn 1984; Morrison et al. 1988; Nano et al. 1989; Verma et al. 2004).

Selenium antigenotoxicity and protective effect

Several studies described important beneficial properties of Se as antioxidant agent (Roussyn et al. 1996; Hassan et al. 2009; Machado Mda et al. 2009), as protector element against UV light (Rafferty et al. 2003), lead (Aykin-Burns and Ercal 2006), mercury (Lemire et al. 2006; Kaur et al. 2009; Peterson et al. 2009), and cadmium (Frisk et al. 2002), as reducer of progression of HIV infection (Hurwitz et al. 2007), as enhancer of immune system (Kiremidjian-Schumacher et al. 1994), and as anticarcinogenic agent against different types of cancer (Clark et al. 1996; Reid et al. 2002; Cai et al. 2006).

As happened with Se genotoxicity, mutagenicity, and cytotoxicity, the results of most of Se *in vitro* studies indicate that the antigenotoxic properties of Se compounds are highly dependent upon the conditions under which they are evaluated (Cemeli et al. 2006), and that the protection

offered by Se compounds against damage induced in genetic material is time and dose dependent (An et al. 1988). A clear example of this is the study by Weitberg et al. (1985) who showed that sodium selenite had variable effects on the number of SCE induced by stimulated human phagocytes in mammalian cells depending on the concentration used. Low concentrations of sodium selenite protected target cells; however, intermediate concentrations had no effect on oxidant-induced SCE formation, and high concentrations increased the number of exchanges.

Using the Ames test, it was reported that sodium selenite was effective in the reduction of the mutagenicity induced by a variety of mutagens (Martin et al. 1981). Furthermore, co-incubation of sodium selenite and N-methyl-N-nitrosourea (MNU) or N-methyl-N'-nitro-N-nitrosoguanidine (MNNG) with bacterial cells (*S. typhimurium*) yielded an evident inhibition of the mutagenicity induced by these alkylating agents (Balansky 1992). It was also observed that this Se compound reduced only very slightly the genotoxic effect of nitrofurans (Gajewska et al. 1990), and that pretreatment of cells with a non-lethal dose of selenite induced the synthesis of proteins which protected the cells from killing by H₂O₂ or high doses of selenite (Kramer and Ames 1988). The genotoxic effects of three Se compounds (sodium selenate, sodium selenite, and selenous acid) were tested in the Ames test and also investigated for their interaction with potassium dichromate (Cemeli et al. 2003). None of them showed significant effect in the Ames test without metabolic activation, and moreover, sodium selenate showed antigenotoxic properties against potassium dichromate.

The possible antigenotoxic effects of Se were also investigated in yeasts by several authors using different Se compounds as ebselen (Chan et al. 2007; Miorelli et al. 2008) or SeMet (Longo et al. 1995; Bronzetti et al. 2001). The antioxidant, mutagenic, and antimutagenic effects of ebselen were evaluated in *S. cerevisiae* strains proficient and deficient in antioxidant defences. Ebselen showed strong activity against H₂O₂-induced oxidative damage in the antimutagenic assay using N123 strain and in the antioxidative assay using strains lacking antioxidant defences (Miorelli et al. 2008). In order to examine the antimutagenic effects of sodium selenite and SeMet, *S. cerevisiae* was treated with H₂O₂ (Bronzetti et al. 2001). D7 strain of *S. cerevisiae* was used, because it constitutes a rapid and inexpensive genetic model to investigate the toxic and mutagenic effect of various compounds. The antimutagenic effect was evident for both sodium selenite and SeMet, according to data previously described in the literature (Longo et al. 1995).

Many in vitro studies have proven that adequate levels of Se can reduce the CA induced by different mutagenic compounds. Se (as sodium selenite) was found to protect cells

against sodium arsenite by reducing the frequency of gaps and chromatid breaks induced by this compound (Sweins 1983; Beckman and Nordenson 1986). Also, Se-enriched green tea was both able to prevent the CA induced by mytomicin C in mouse spermatocytes and to enhance glutathione peroxidase and superoxide dismutase activity in blood serum and liver (Li et al. 2009). Another study concluded that sodium selenite under specific conditions reduces the percentage of cells with N-methyl-N'-nitro-N-nitrosoguanidine (MNNG)-induced CA (An et al. 1988), but this protective effect is clearly time and dose dependent, resulting in toxic effects at high concentrations. Se (sodium selenite and SeMet) also protected mammalian cells against lead acetate- and sodium sulfite- (Beckman and Nordenson 1986), carbon tetrachloride- (Siviková et al. 2001) and doxorubicin-induced damage (Santos and Takahashi 2008).

Se was also found to reduce the SCE levels induced by different compounds. Sodium selenite, for example, significantly reduced SCE frequencies induced by fluorescent light in human fibroblasts (Parshad et al. 1980), and also by arsenic (Hu et al. 1996), and by carbon tetrachloride (Siviková et al. 2001) in peripheral lymphocytes. Moreover, some studies demonstrated that sodium selenite can antagonize the ability of other compounds to cause DNA damage leading to the formation of SCE. This was the case of two mercury derivatives (Morimoto et al. 1982), and methyl methanesulfonate or N-hydroxy-2-acetylaminofluorene (Ray and Altenburg 1978), which cause an increase in SCE, but simultaneous addition of sodium selenite to the cultures resulted in SCE frequencies below the sum of the SCE frequencies produced by the individual compounds.

Several works designed to evaluate the antigenotoxic properties of different Se compounds by means of MN test are collected in the literature. Diphenyl diselenide at low concentrations showed antimutagenic properties against H₂O₂, methyl methanesulphonate, and UVC radiation in lung fibroblast cells (Rosa et al. 2007b); supplementation of human MCF-7 breast carcinoma cells or mouse fibroblasts with low levels of sodium selenite protected these cells from ultraviolet-induced chromosome damage (Baliga et al. 2007); selenous acid and sodium selenate reduced the DNA damage induced by potassium dichromate in human lymphocytes and TK6 cells, respectively (Cemeli et al. 2006); sodium selenite decreased the MN rate induced by MNNG in children's foreskin fibroblasts (An et al. 1988); V79 cells showed diminished cadmium-induced MN frequency when treated with sodium selenite (Hurná et al. 1997), and its protective effect was also demonstrated in ovine peripheral lymphocytes cultured with carbon tetrachloride (Siviková et al. 2001).

The literature agrees with the protective effect of Se evaluated with the comet assay against a variety of chemical or physical toxic agents. In vitro investigations with this

assay found that Se (sodium selenite and ebselen) prevented DNA damage from H_2O_2 in murine lymphoma cells (Bouzyk et al. 1997), in HepG2 cells (Yang et al. 1999), in Chinese hamster V79 cells (Miorelli et al. 2008), and in mouse hepatoma Hepa 1c1c7 cells (Keck and Findley 2006). Sodium selenite also inhibited the DNA damage caused by cadmium chloride in rat hepatic cells (Yu and Chen 2004). Sodium selenate avoided DNA strand breaks mediated by UVA radiation in human skin fibroblasts (Emonet-Piccardi et al. 1998) and by quenched potassium dichromate in human lymphocytes (Cemeli et al. 2003). Sodium selenite and SeMet protected keratinocytes against UV-induced oxidative damage (Rafferty et al. 2003), as well as SeMet protected against genotoxicity induced by doxorubicin in human lymphocytes (Santos and Takahashi 2008). SeMet was also found to protect against bleomycin-induced DNA damage on human lymphocytes (Laffon et al. 2009). Finally, low concentrations of diphenyl diselenide showed antimutagenic properties in V79 cells treated with H_2O_2 , methyl methanesulphonate, and UVC radiation, probably due to the antioxidant properties of diphenyl diselenide (Rosa et al. 2007a, b).

Regarding to the cell growth, the effect of two Se compounds and methyl mercury was also studied in cell cultures (Alexander et al. 1979). Selenite at low concentration and seleno-di-*N*-acetyl-glycine in 1,000-fold higher concentrations offered considerable protection against the growth inhibiting effect and the stimulation of glucose and lactate uptake caused by methyl mercury in rat Morris hepatoma cells. However, no protective effect of Se was observed in other cell types as human lymphocytes and human embryonic fibroblasts. The data obtained suggested that Se compounds exert their protective effect through cell-specific processes rather than by a direct chemical reaction between selenite and methyl mercury. In another study, Hurst et al. (2008) exposed two human prostate cell lines to nutritionally relevant doses of MeSeCys and selenite, ranging from deficient to the equivalent of Se supplementation in humans. Several Se-responsive genes were identified by means of two microarray platforms, many of which have been ascribed to cancer cell growth and progression. The study revealed that MeSeCys can alter the expression of several types of collagen and thus potentially modulate the extracellular matrix and stroma, which may at least partially explain the anticancer activity of MeSeCys.

Concluding remarks

Se is one of the oligoelements most studied because of its particular properties. Like some other trace elements, Se is bimodal in nature whereby its beneficial properties occur in a limited range of daily intake below which it cannot per-

form its essential functions, and above which it is toxic (Alaejos et al. 2000). This nutritional range between essentiality and toxicity in Se is fairly narrow in comparison with the other essential trace elements (Letavayová et al. 2008a), and it could explain, among other causes, the enormous variability in the results of Se studies. As a consequence of these properties, Se can be included in the class of “Janus compounds”, having two “faces” on the same head (Miorelli et al. 2008). In general, at low concentrations, Se compounds are antimutagenic and anticarcinogenic, whereas at high concentrations, they are mutagenic, toxic, and possibly carcinogenic (Letavayová et al. 2008a).

When the effects of different selenocompounds were evaluated by means of the different *in vitro* assays, results obtained varied highly showing a great controversy. As general conclusions, Se resulted in no mutagenic or weakly mutagenic effects in bacterial assays (Lofroth and Ames 1978; Noda et al. 1979; Morimoto et al. 1982), but mutagenicity and genotoxicity of this element, mainly as sodium selenite, was observed in numerous studies with yeasts (Rosin 1981; Anjaria and Madhavanath 1989; Letavayová et al. 2008a). On the other hand, antigenotoxic properties of Se against a great variety of mutagenic agents were also detected in both cell systems (Martin et al. 1981; Longo et al. 1995; Bronzetti et al. 2001). This agrees with the results of different *in vitro* studies performed in mammalian cell systems. Data showed that Se induces CA (Nakamuro et al. 1976; Biswas et al. 2000) and SCE (Ray and Altenburg 1980, 1982; Sirianni and Huang 1983), inhibits DNA synthesis (Frenkel and Falvey 1988) and cell growth (Lu et al. 1995; Goel et al. 2006; Philchenkov et al. 2007), and promotes apoptosis (Chen et al. 2003; Last et al. 2006; Xiang et al. 2009). But also antigenotoxic and antimutagenic properties of adequate doses of Se against many chemical and physical agents have been described (Parshad et al. 1980; Sweins 1983; Beckman and Nordenson 1986; An et al. 1988; Hu et al. 1996; Bouzyk et al. 1997; Siviková et al. 2001; Cemeli et al. 2006; Rosa et al. 2007b; Santos and Takahashi 2008).

Nevertheless, all these results are not constant in the literature and vary enormously even when the same *in vitro* tests are employed. Many factors contribute to this great variety of results, mainly its chemical form (Nakamuro et al. 1976; Whiting et al. 1980; Sirianni and Huang 1983) and the concentration used (Weitberg et al. 1985; Biswas et al. 2000; Verma et al. 2004), but also the exposure time (Ray and Altenburg 1978; An et al. 1988), the treatment conditions (Cemeli et al. 2006; Ray et al. 1978), the cell type or the target tissue (Webber et al. 1985; Vadgama et al. 2000), and other previous factors as method of administration, animal species, physiological status, interaction with other compounds, etc. (Burk and Levander 2002). So, although it is common to speak of Se in the universal term

of the element, just Se, the dose and form of the Se species actually determine its biological activity, be it the dietary essential nutrient, the cancer-preventing agent, or the toxicant (Letavayová et al. 2008a).

Se is an important element with beneficial properties as nutrient, and its dietary deficiency is linked to some diseases, e.g., Keshan disease and Kashin-Beck disease (Thomson 2004). Moreover, solid evidence based on epidemiological studies conducted in the last 50 years shows an inverse relationship between Se intake and cancer incidence (Alaejos et al. 2000; Surai 2006). For these reasons, today many people consume Se supplements on a regular basis to increase their intake and improve their nutritional status. They do this in the belief either that Se levels in the diet are inadequate or that the additional intake will provide protection against a variety of health problems. Much of current interest in Se as a supplement was triggered by the report by Clark et al. (1996) (Nutritional Prevention of Cancer study). The use of dietary supplements is considerable in many countries and appears to be increasing. These products are tested in a battery of genotoxicity assays (Griffiths and Matulka 2006), as those described in this paper, before being commercially available, normally in tablet form, in quantities up to 200 µg, and sometimes more, per tablet (Reilly 2006).

Despite the recent results of SELECT (Lippman et al. 2009), supplemental Se has been shown to have cancer-protective effects in a variety of experimental settings and clinical studies (reviewed by Whanger 2004) and to reduce the incidence and mortality of total cancer (Clark et al. 1996), prostate cancer (Duffield-Lillico et al. 2002), liver cancer (Yu et al. 1997), and stomach cancer (Blot et al. 1993) in human interventional trials. In general, the anticarcinogenic effect of Se against leukemia and cancers of the colon, rectum, pancreas, breast, ovaries, prostate, bladder, lung, and skin seems clear at least under some conditions (reviewed by Sunde 2000) and is closely related to its role in selenoproteins-reducing oxidative stress, to its ability to enhance the immune response or, more likely, to its ability to produce antitumorigenic metabolites (e.g., methylselenol or its precursors) that can perturb tumor-cell metabolism, inhibit angiogenesis and induce apoptosis in cancer cells (Rayman 2000; Whanger 2004). The source of the Se supplement (SeMet) in SELECT and the relatively high initial levels of Se in the enrolled men have been suggested to contribute to the negative results obtained in this trial (Hatfield and Gladyshev 2009).

But in spite of the extensive literature describing the antimutagenic and anticarcinogenic effects of Se compounds, little is known on their mode of action (Miorelli et al. 2008), although the anticancer activity of Se seems to be also dose dependent and species specific (Hurst et al. 2008). The bulk of our knowledge on the mechanisms of

cancer prevention by Se is based on animal data and from studies conducted in in vitro systems (El-Bayoumy and Sinha 2005), and the modulation of certain in vitro markers may also be of value in predicting the effectiveness of novel forms of Se for cancer prevention. Thus, there is a plausible correlation between the relevance of these in vitro markers and the consequence of in vivo cancer protection. Whether these markers apply only to the biology of Se chemoprevention or could be extended to other classes of anticancer agents remains to be investigated (Lu et al. 1996).

In short, nowadays, besides the beneficial properties that Se has as nutrient and the fact that it seems to be effective in cancer prevention, the genotoxic effects of Se are currently being demonstrated in present studies. In this sense, the enormous variety of in vitro assays are allowing to describe, characterize, and delimit these effects in order to provide important information on the correct use of Se supplements in human health and chemoprevention. These assays show several advantages, as allowing controlling the features of the exposure and employing human cell lines that can provide a more real view of its effects on the human organism, what make them a perfect complement to in vivo assays when these can be used or an appropriate substitute when not.

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