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Oxidative effects of the human antifungal drug clotrimazole on the eucaryotic model organism *Saccharomyces cerevisiae*

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Abstract

Clotrimazole is a type of antifungal medication developed from azole compounds. It exhibits several biological actions linked to oxidative stress. This study focuses on the oxidative effects of clotrimazole on the eukaryotic model yeast, *Saccharomyces cerevisiae*. Our results showed that although initial nitric oxide levels were above control in clotrimazole exposed cells, they showed decreasing tendencies from the beginning of incubation and dropped below control at 125 μ M from the 60th min. The highest superoxide anion and hydrogen peroxide levels were 1.95- and 2.85-folds of controls at 125 μ M after 15 and 60 min, respectively. Hydroxyl radical levels slightly increased throughout the incubation period in all concentrations and reached 1.3-fold of control, similarly at 110 and 125 μ M in the 90th min. The highest level of reactive oxygen species was observed at 110 μ M, 2.31-fold of control. Although NADH/NADPH oxidase activities showed similar tendencies for all conditions, the highest activities were found as 3.07- and 2.27-folds of control at 125 and 110 μ M in the 15th and 30th min, respectively. The highest superoxide dismutase and catalase activities were 1.59- and 1.21-folds of controls at 110 μ M clotrimazole in 30 and 90 min, respectively. While the drug generally induced glutathione-related enzyme activities, the ratios of glutathione to oxidized glutathione were above the control only at low concentrations of the drug. The levels of lipid peroxidation in all treated cells were significantly higher than the controls. The findings crucially demonstrate that this medicine can generate serious oxidative stress in organisms.

Highlights

1. Clotrimazole treatment enhanced NO• and ROS levels in S. cerevisiae cells.

2.NADH and NADPH oxidases both cause and contribute to ROS generation.

3. The activities of two major antioxidant enzymes, SOD and CAT, were also highly increased.

4. The antioxidant system's response was insufficient to keep LPO levels under control.

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Graphical Abstract



Keywords Clotrimazole · Saccharomyces cerevisiae · Antioxidant system · Oxidative stress · Radical production

Introduction

Clotrimazole (1-[(2-chlorophenyl)-diphenylmethyl]imidazole; CLT) is an azole derivative antifungal drug that is effective against a wide range of fungal pathogens (Crowley and Gallagher 2014; Yassin et al. 2023; Feng et al. 2023; Ghahartars et al. 2024). As has been well known for years, the fungistatic mechanism of CLT is associated with the blocking of the synthesis of ergosterol by the inhibition of lanosterol 14 α -demethylase, which belongs to the cytochrome P450 monooxygenase superfamily (Van Den Bossche et al. 1983). Apart from this activity, the studies show that CLT has various biological effects, including disruption of the glycolytic pathway, G1-phase arrest, induction of apoptosis, inhibition of transient receptor potential and calcium-dependent potassium channel, heme-induced membrane damage, and disruption, suppression, and modulation of PI3K/AKT/ mTOR, NF-kB, and ERK-p65 pathways, respectively (Jensen et al. 1999; Khanna et al. 1999; Tiffert et al. 2000; Ito et al. 2002; Liu et al. 2023; Kadavakollu et al. 2014; do Nascimento Júnior et al. 2024; Paul et al. 2024). Considering that many of these biological activities are associated with oxidative stress, it becomes vital to understand the effects of CLT on cellular redox status, which can be basically defined as the balance between reactive species (RS) and antioxidant defense mechanisms. A review of the literature shows that CLT induces oxidative stress in different cell types. Trivedi et al. (2005) stated that CLT inhibits the hemoperoxidase of the unicellular protozoan parasite Plasmodium falciparum, and this inhibition is associated with the development of oxidative stress. In a study conducted on breast cancer cells, CLT in combination with the antineoplastic drug paclitaxel increased the level of RS and oxidative stress (Sharma et al. 2017). In a current study, it was reported that CLT treatment leads to actin aggregation, mitochondrial alteration, production of total reactive oxygen species (tROS) and oxidative stress which causes to activation of MAPK signalling pathways in Saccharomyces cerevisiae (Sellers-Moya et al. 2021). Melefa and Nwani (2021) showed that CLT exposure led to an increase in lipid peroxidation (LPO) levels in non-target aquatic species African catfish, such as the Clarias gariepinus. In the same research, it was also demonstrated that while CLT treatment increased catalase (CAT) and superoxide dismutase (SOD) activities, it reduced the glutathione (GSH) level, glutathione reductase (GR), and glutathione peroxidase (GSHPx) activities. Another study showed that CLT influenced the cell membrane potential and permeability in Leishmania donovani, an intracellular parasite of the genus Leishmania that causes leishmaniasis, by producing mitochondrial superoxide (Paul et al. 2024). On the other hand, Pulathan and collegues (2024) stated that CLT demonstrated a protective impact on lung injury in an experimental model of ruptured abdominal aortic aneurysm. This antifungal has been found to be effective for the regression of endometriotic implants through the downregulation of inducible nitric oxide synthase, modulation of the antioxidant defense system, and oxidative stress biomarkers including SOD, GSH and glutathione S-transferase (GST), membrane LPO, and protein carbonylation (Machado et al. 2023). Additionally, a similar disruption effect of CLT therapy on oxidative status

has been determined under other kinds of undesirable pathological conditions (Rodrigues Henriques and Gamboa de Domínguez 2012; Osmanağaoğlu et al. 2012; Gao 2024).

In this paper, we aimed to investigate the effects of CLT on the levels of the main RS, namely nitric oxide (NO•), superoxide anions (O₂•⁻), hydroxyl radicals (•OH), hydrogen peroxide (H_2O_2) , and tROS. The changes in the activities of NADH and NADPH oxidase enzymes responsible for the generation of $O_2^{\bullet-}$ were detected. In addition, the levels of the main components of the antioxidant defense system, including SOD, CAT, GR, GST, and GSHPx, and the ratio of GSH/GSSG (GSSG is the oxidized form of GSH) were evaluated. Finally, we determined the LPO levels, one of the known markers of oxidative stress status, depend on CLT treatment. We used the wild-type strain of S. cerevisiae as a model organism. This yeast is a non-pathogenic but opportunistic strain, also known as baker's yeast. The cultivation of S. cerevisiae, whose genome has been fully elucidated, is very simple, rapid, and economical. Furthermore, it has been well known for years that several yeast proteins are functionally interchangeable with highly homologous mammalian proteins, and numerous metabolic pathways, including lipid, protein and energy metabolism, unfolded protein response, prion development, mitochondrial metabolism, and oxidative stress adaptive response, are also conserved in S. cerevisiae yeast (Rinaldi et al. 2010; Barros et al. 2010; Duina et al. 2014; Rouchidane Eyitayo et al. 2021; Zhu et al. 2023). Therefore, the data obtained from the study contribute to the existing literature on the understanding of the CLT-induced oxidative stress mechanism.

Materials and methods

Materials

Unless otherwise stated, all of the chemicals used were analytical or higher purity, obtained from Sigma-Aldrich, Inc. (St. Louis, MO, USA). *S. cerevisiae* BY4742 wild type strain (MATalpha his $3\Delta 1 \ leu 2\Delta 0 \ lys 2\Delta 0 \ ura 3\Delta 0$) was obtained from GE Healthcare Dharmacon, Inc., USA. The measurements were done using Varioskan Flash Spectral Scanning Multimode Reader (Thermo Fisher, USA) and the UV-1700 UV Pharmaspec Spectrophotometer (Shimadzu, Japan).

Cell culture and drug treatment conditions

The yeast cells were routinely grown on yeast-peptonedextrose-agar medium (YPDA) at 30 0 C for 2 days. 1 1 of medium consists of 10 g of yeast extract, 20 g of peptone, 20 g of glucose, and 20 g of bactoagar. For the drug treatment, the cells were cultured up to their mid-exponential phase in YPD medium without agar and sub-cultured in 100 ml of the same medium to achieve an optical density (OD) at 660 nm of 0.240, which corresponds to approximately 3.0×10^6 cells per 1 ml. Then, CLT in the concentration range of 50–125 μ M was added, and the incubation was carried out for 15–90 min at 30 °C with 180 rpm agitation.

Biochemical assays

At the end of the above-mentioned incubation period, CLTtreated (test groups) and untreated (control groups) cells were harvested after two washing steps with PBS solution at 4 °C and stored in -20 °C until analysis. If the assay needs a supernatant, the cells were homogenized in a Mixer Mill MM 400 (Retsch) for 14 min at a frequency of 30 Hz with glass beads (Carl Roth, 0,25-0,5 mm) at a ratio of 1:1, and the samples were kept in an ice bath for 5 min intervals. The obtained samples were stored in -80 °C until analysis.

Determination of the changes in reactive species levels

The levels of $O_2^{\bullet-}$ in the test and control groups were determined by using the chemiluminogenic di-acridinium compound bis-N-methylacridinium nitrate (lucigenin), which emits light upon a reaction with $O_2^{\bullet-}$ (Skatchkov et al. 1999). In the assay, approximately 3.0×10^5 cells were mixed with 5 µM lucigenin prepared in YPD, and the mixture was incubated at 37 °C for 1 h. The results were expressed as relative luminescence units (RLU).

The procedure described by Barja (1999) was followed with little modification for the determination of H₂O₂ levels in test and control groups, which is based on the generation of fluorescent biphenyl homovanillic acid (b-HVA) upon oxidation of HVA through the reaction between H_2O_2 and horseradish peroxidase (HRP). In the assay, 150 µl of yeast cell supernatant was mixed with 1.5 ml of potassium phosphate buffer, which contains 3.0 mM MgCl₂, 145 mM KCl, 30 mM HEPES, 0.1 mM EGTA, 0.1 mM HVA, 2.5 mM pyruvate/malate, and 850 U HRP (pH 7.4), and the mixture was incubated at 30 °C for 15 min. Then, 0.5 ml of cold stop solution, which contains 0.1 M glycine and 25 mM EDTA-NaOH, was added to the mixture, and the measurement was carried out at $\lambda_{ex}\!=\!312$ nm and $\lambda_{em}\!=\!420$ nm. The calculations were made using the calibration curve of H₂O₂, and the results were expressed as nmol H₂O₂ per gram of wet weight (nmol/gww).

The procedure described by Kirkland et al. (2007) was followed for the determination of tROS levels in test and control groups by using the fluorescent probe 5-(and-6)chloromethyl-2',7'-dichlorodihydrofluorescein diacetate (CM-H2DCFDA, C6827, Invitrogen, Grand Island, NY). In the assay, mid-exponential cells were suspended in YPD medium with 10 μ M CM-H2DCFDA (loading medium) and incubated at 28 °C for 1 h. Then, the loading medium was removed, and fresh YPD medium with or without CLT was added. At the end of the drug treatment periods, the measurement was carried out at λ_{ex} = 485 nm and λ_{em} = 528 nm, and the results were expressed as relative fluorescence units (RFU).

The procedure described by Ruenroengklin et al. (2009) was followed with little modification for the determination of •OH levels in the test and control groups, which is based on the generation of a pink color of the malondialdehyde (MDA)-thiobarbituric acid (TBA) product through the release of MDA after the oxidative cleavage of 2-deoxy-ribose. In the assay, 750 µl of yeast cell supernatant was treated with 750 µl of 10 mM deoxyribose and incubated at 37 °C for 60 min. Then, 500 µl of 1% (w/v) aqueous TBA-trichloroacetic acid (TCA) (pH 7.4) was added to the mixture, and additional incubation was carried out for 30 min at 100 °C. The measurement was carried out at $\lambda_{ex} = 532$ nm and $\lambda_{em} = 553$ nm against the blank, and the results were expressed as a percentage of the control group.

To determine the levels of NO• in the test and control groups, nitrite (NO₂⁻) levels, which is one of the primary, stable, and nonvolatile breakdown products of NO•, were determined by using Griess reagent, the system based on which a diazotization reaction (Giustarini et al. 2008). In the assay, yeast cell supernatant was mixed with Griess reagent in a ratio of 1:1 and incubated at room temperature for 30 min. The measurement was carried out at 540 nm against the blank, and the calculations were made using the calibration curve of sodium nitrite. The results were expressed as μ mol sodium nitrite per gram wet weight (μ mol/gww).

Determination of the changes in enzyme activities

The activities of NADH and NADPH oxidases were measured at 340 nm based on the disappearances of NADH and NADPH substrates, respectively (Anders et al. 1970). The calculations were made using the extinction coefficient of 0.00622 l/ μ mol/cm which is the extinction coefficient of the substrates. One IU was defined as the amount of substrate oxidized per ml of enzyme in a minute.

The SOD activities were measured at 490 nm based on inhibition of the spontaneous autoxidation of 6-hydroxydopamine (6-OHDA) by the action of the enzyme (Crost et al. 1987). One IU was defined as the enzyme amount that required 50% inhibition of 6-OHDA autoxidation.

The CAT activities were measured at 240 nm based on the disappearance of H_2O_2 by the action of the enzyme (Aebi 1974). The calculations were made using the extinction coefficient of 43.6 $M^{-1}cm^{-1}$ which is the extinction coefficient of H_2O_2 . One IU was defined as the enzyme amount required to decompose one µmole of H_2O_2 .

The GR activities were measured at 340 nm based on the disappearance of NADPH by the action of the enzyme (Massey and Williams 1965). The enzyme activity was calculated using 0.00622 l/µmol/cm, which is the extinction coefficient of NADPH. One IU was defined as the enzyme amount required to oxidize 1 nmol of NADPH per minute at 25 0 C.

The GST activities were measured at 340 nm based on the formation of a conjugate between GSH and 1-chloro-2,4-dinitrobenzene (CDNB) by the action of the enzyme (Habig et al. 1974). The enzyme activity was calculated using 0.0096 l/µmol/cm, which is the extinction coefficient of the product. One IU was defined as the enzyme amount required to conjugate 10.0 nmol of CDNB with GSH per minute at 25 $^{\circ}$ C.

The GSHPx activities were measured at 340 nm based on the disappearance of NADPH by the action of the enzyme (Paglia and Valentine 1967). The enzyme activity was calculated using 0.00622 l/µmol/cm, which is the extinction coefficient of NADPH. One IU was defined as the enzyme amount required to oxidize GSH per minute at pH 7.0 at 25 °C.

Determination of the changes in GSH/GSSG ratios

The GSH/GSSG ratios were calculated by measuring total glutathione (GSH+GSSG) and reduced glutathione (GSH) levels separately and dividing the levels of GSH to GSSG. To measure GSH+GSSG and GSH levels, the procedure reported by Teare et al. (1993) was followed, which is based on the conversion of GSSG to GSH via the action of GR and then monitoring the formation of yellow product through the reaction of GSH with 5,5'-dithiobis-(2-nitrobenzoic acid) (DTNB) at 412 nm. To determine only GSH levels, the GR enzyme solution was removed from the reaction mixture. The calculations were made using the calibration curve of GSH.

Determination of the total protein concentrations

The total protein concentrations were measured using the Bradford method based on the binding of basic proteins with aromatic side chains to Coomassie Brilliant Blue G-250 dye (Bradford 1976).

Determination of the changes in malondialdehyde levels

The MDA levels were measured based on the pink chromogen formation upon reaction with TBA (Schemedes and Holmer 1989). In the assay, 0.5 ml of yeast cell supernatant was mixed with 1.5 ml of 10% TCA, and the mixture was incubated at 90 0 C for 15 min. Then, the mixture was centrifuged, and the upper phase was added to the TBA solution in a ratio of 1:1. After additional incubation at 90 0 C for 15 min, the measurement was carried out at 532 nm against the blank, and the results were expressed as µmol MDA per gram wet weight (µmol/gww) by using the extinction coefficient of the MDA-TBA product, 1.56×10^{5} l/mol/cm.

Statistical analysis

The data are reported as the mean \pm SEM of three independent experiments. The differences in variance were analyzed statistically using a two-way analysis of variance (ANOVA), and Tukey's test was used as a *post hoc*. Graph-Pad Prism version 9.5.1 was used for all statistical analyses.

Results and discussion

Determination of the changes in reactive species levels

In this paper, we have firstly examined the changes in the levels of RS, including NO•, $O_2^{\bullet-}$, and •OH radicals, besides tROS and H₂O₂, in yeast cells depending on the treatment with CLT at 50-125 µM for 15-90 min, the conditions of which the yeast cell death was induced (Kavakcioglu and Tarhan 2018). As can be seen from Fig. 1A, CLT treatment generally caused statistically significant increases in the NO• levels compared to the control for all incubation periods. On the other hand, a time dependent decreasing trend in the initial levels of NO• was observed for all the studied concentrations of the drug. The levels of this radical were found to be under control only for higher concentrations of the CLT after longer exposure periods. The decreases at 125 µM drug treatment for 60 and 90 min were found to be statistically significant (p < 0.05). In a study conducted by Motawi et al. (2015), it was shown that 20 µM CLT alone

Fig. 1 The effects of CLT treatments at different concentrations for 15–90 min on NO• (A), O_2 •⁻ (B), H_2O_2 (C), •OH (D) and tROS (E) levels of *S. cerevisiae* cells. Data with error bars show the mean \pm S.E.M of three experiments. **p* < 0.05 denotes significant differences between control and other studied group by Tukey's multiple range tests



and its combination with 15 µM imatinib mesylate, a tyrosine kinase inhibitor, statistically significantly increased NO• production compared to the control group after 48 h of treatment in the human breast tumor cell line T47D. In another study, it was similarly reported that the treatment with 25 μ M of CLT alone or 25 μ M of CLT plus 12.5 nM of the antineoplastic drug paclitaxel for 24 h statistically significantly increased the levels of NO• in the human breast cancer cell lines MCF-7 and MDA-MB-231 (Sharma et al. 2017). Although the treatment periods are longer, it can be stated that these literature data are compatible with our findings by considering the differences in the cell models studied. All these increases in NO• levels can be evaluated as cellular oxidative stress response because of its fast reaction with O₂•⁻ to form peroxynitrite (Squadrito and William 1998).

 $O_2^{\bullet-}$ is mainly generated during the autoxidation of glucose to an enediol radical anion, which is then converted into ketoaldehydes and O₂•⁻ radicals (Bozinovski et al. 2015). Our results showed that the levels of this radical almost always statistically significantly increased with the increasing concentration of the drug (p < 0.05), and it reached the highest value of 29.65 ± 0.208 RLU at 125 μ M after 15 min. Similarly to the changes in NO• levels, O₂•levels also showed a decreasing trend with an increasing treatment period (Fig. 1B). Additionally, there were positive correlations between the levels of $O_2^{\bullet-}$ and NO• for all studied drug concentrations ($r_{50\mu M}$ =0.995; $r_{75\mu M}$ =0.984; $r_{110\mu M} = 0.932$; $r_{125\mu M} = 0.772$), which supports the idea that the increases in NO• levels may part of the oxidative stress response system. We did not come across any study reporting the effects of CLT on NO• and O₂•- levels together to better understand the roles of these RS in CLT-induced toxicity. Nevertheless, we encountered a paper that demonstrates a significant increase in the fluorescence signal with the O₂•⁻ indicator dihydroethdium in yeast after treatment with CLT (Sellers-Moya et al. 2021). The production of O₂•⁻ after treatment with another azole antifungal drug, fluconazole, in combination with *Lippia origanoides* essential oil or thymol was also reported (Zapata-Zapata et al. 2023).

 H_2O_2 is another RS whose levels were investigated in yeast cells depending on the application of the CLT. Although H_2O_2 is a non-radical RS with a relatively long half-life, it is responsible for the generation of the •OH radical through the Fenton and Haber-Weiss reactions (Halliwell 1991, 2006). In our study, the levels of H_2O_2 in yeast cells were statistically significantly induced by the increasing concentrations of CLT for all individual treatment periods (p < 0.05) (Fig. 1C). Also, significant changes were recorded for almost all studied concentrations depending on the extension of time. The highest induction was observed at 125 µM treatment for 60 min as 25.58 ± 0.013 nmol.gww⁻¹, which is approximately 2.9-fold of the related control. Similar inductions were reported in the literature for the fungal strains *Phanerochaete chrysosporium*, *Schizosaccharomyces pombe*, and *Penicillium chrysogenum* treated with menadione, a natural organic compound well known to induce ROS (Emri et al. 1999; Le et al. 1995; Tongul and Tarhan 2014; Criddle et al. 2006; Loor et al. 2010; Sun et al. 2021).

The changes in intracellular •OH levels that depend on CLT treatment are presented in Fig. 1D. As is known, •OH radical is highly reactive and can trigger an overwhelming variety of oxidative reactions. We observed that CLT treatments generally caused statistically significant inductions in •OH levels compared to related controls. Additionally, generally non-significant increases were recorded with prolonged incubation times for each studied dose. As stated above, •OH radicals can be generated via Fenton and Haber-Weiss reactions. As a matter of fact, we found strong negative correlations between •OH and O₂•-levels in yeast cells exposed to CLT ($r_{50\mu M}$ =-0.899; $r_{75\mu M}$ =-0.941; $r_{110\mu M}$ =-0.835; $r_{125\mu M}$ = -0.159). On the other hand, there was no correlation between •OH and H₂O₂ levels.

The levels of tROS as well as specific radical species were also measured in the study. As can be seen from Fig. 1E, tROS levels were statistically significantly induced after 50-125 µM CLT treatment at each treatment period compared to the related controls (p < 0.05). We also observed significant increases at almost all time points of the incubation period for each studied drug dose. The highest increase was recorded at approximately 2.68-fold of the control at 125 µM CLT after 45 min of treatment. It was determined that while there were generally strong positive correlations between tROS and H₂O₂ levels dependent on both the drug dose and treatment time, these correlations were only found with the increasing concentration of the drug between tROS and other oxygen radicals, namely $O_2^{\bullet-}$ and $\bullet OH$. So, it can be stated that especially non-radical H₂O₂ is of significance in terms of the intracellular tROS levels in S. cerevisiae exposed to CLT. Similar to our results, 10 µM amphotericin B, which belongs to the polyene class of antifungals, 12.5 µg/ml miconazole, and 104 µM fluconazole, the other two members of azole antifungal agents, led to increases in tROS levels in the range of 2-4 folds after approximately 2-3 h treatment of Candida albicans and Cryptococcus gattii cells, respectively (Ferreira et al. 2013). Moreover, recent studies have proposed a molecular mechanism that is dependent on ROS as the basis for the azole-induced cytotoxicity (de Andrade Neto et al. 2020; Petricca et al. 2023; Lu et al. 2024).

Determination of the changes in NADH and NADPH oxidase activities

The enzymes of NAD(P)H oxidases are known as endogenous ROS producers with their O2.0- producing activity from oxygen. The changes in the activities of NADH and NADPH oxidases in S. cerevisiae cells exposed to CLT are presented in Fig. 2A and B, respectively. According to the obtained results, except for 50 µM CLT, both enzyme activities in drug treated groups were generally found to be statistically significantly higher than related control activities throughout the treatment period (p < 0.05). On the contrary, the enzyme activities in the cells treated with 50 μ M of the antifungal were lower than control values for all treatment times studied. While the highest NADH oxidase activity was found as $177,91 \pm 3.53$ U/mg at 125 μ M CLT after 15 min of exposure, NADPH oxidase activity reached its maximum at 110 μ M CLT after 30 min with a value of 45.87 + 0.67U/mg. Furthermore, NADH and NADPH oxidase activity changes showed positive correlations with O2.-levels at the concentration of 125 µM (NADH oxidase-O₂•-: r_{125µM}= 0.89 and NADPH oxidase- $O_2^{\bullet-}$: $r_{125uM} = 0.68$). Our results indicate that fungal NADPH oxidases contribute to the formation of radicals, despite the lack of research on their involvement in antifungal drug-induced ROS (Gonzalez-Jimenez et al. 2023).

Determination of the changes in SOD and CAT enzyme activities

We have analyzed the SOD and CAT enzyme activities as the main antioxidant enzymes depending on CLT administration in the present study. While SOD catalyzes the conversion of O₂•⁻ to H₂O₂, CAT catalyzes the decomposition of H₂O₂ to H₂O and O₂ (Mandell 1975). In CLT treated yeast cells, SOD enzyme activities were induced compared to the control group, with generally statistically significant differences (Fig. 3A). It was observed that the inductions in SOD activity were insignificant but gradual at 50 and 75 µM CLT treatments as the incubation period increased. The highest value reached 21.37 ± 0.99 U/mg, approximately 1.64-fold of the control, at 110 µM after 45 min of treatment. In a study using green algae cells, it was similarly reported that the antifungals propiconazole, tebuconazole, and myclobutanil significantly increased SOD activity (Nong et al. 2021). As expected, there were strong negative correlations between SOD activities and O2. I levels during the entire incubation period for 50 μ M (r_{50uM}=-0.96) and up to 45th min for 75 and 110 μ M treatments ($r_{75\mu M}$ =-0.99; $r_{110\mu M}$ =-0.88). SOD activities also showed a positive correlation with H_2O_2 levels at concentrations of 50 and 75 μM $(r_{50uM} = 0.87; r_{75uM} = 0.77)$. On the other hand, we did not determine any of these correlations at 125 µM, which can be interpreted as a serious deterioration of cellular redox status at higher concentrations of the antifungal.

Similar to the activities of SOD, generally statistically significant increases compared to the control group were recorded in the activities of the CAT enzyme (Fig. 3B). Although slight increases were observed between 0 and 110 μ M, approximately similar activities were measured at 125 μ M of CLT throughout the treatment time. The highest increase in CAT enzyme activities was obtained approximately 1.22-fold in the sample exposed to 110 M of the drug for 90 min. Because positive correlations were found between CAT activities and H₂O₂ levels at all studied concentrations (r_{50 μ M}=0.93; r_{75 μ M}=0.24; r_{110 μ M}=0.92; r_{125 μ M}



Fig. 2 The effects of CLT treatments at different concentrations for 15-90 min on NADH (A) and NADPH (B) oxidase enzyme activities of *S. cerevisiae* cells. Data with error bars show the mean \pm S.E.M of

three experiments. *p < 0.05 denotes significant differences between control and other studied group by Tukey's multiple range tests



Fig. 3 The effects of CLT treatments at different concentrations for 15–90 min on SOD (**A**) and CAT (**B**) enzyme activities of *S. cerevisiae* cells. Data with error bars show the mean \pm S.E.M of three experi-

ments. *p < 0.05 denotes significant differences between control and other studied group by Tukey's multiple range tests

= 0.66), it is considered that the inductions in the activities of CAT depending on CLT treatment are not sufficient to scavenge H_2O_2 .

Determination of the changes in GSH related antioxidant systems

In the presented paper, the changes in the levels of GSH related antioxidant systems, namely GR, GST, GSHPx activities, and GSH/GSSG ratios, were also determined. The GR enzyme plays a key role in maintaining a reducing environment in the cell by reducing GSSG to GSH. Overall, all investigated exposure times showed statistically significant induction of GR activities in test groups compared to their controls (Fig. 4A). The highest GR activity was found as 252.00 ± 3.89 U/mg at 125 µM drug concentration after 60 min of treatment. On the other hand, a very sharp decrease was observed after an additional 15 min treatment up to approximately 100 U/mg. The negative correlations $(r_{50\mu M} = -0.694; r_{75\mu M} = -0.872; r_{110\mu M} = -0.689)$ between GR enzyme activities and GSH/GSSG ratios reflect that the inductions in GR activity were not sufficient for the regeneration of GSH. GST enzymes catalyze the conjugation reaction of electrophilic xenobiotics with GSH and generate non-toxic polar metabolites from these toxic chemicals. These enzymes also have peroxidase activity, which can reduce the toxicity of hydroxyl-peroxide (Marrs 1996). We found that the enzyme activities were statistically significantly induced in all treatment groups compared to the related controls (p < 0.05) (Fig. 4B). The gradual increases were recorded in the enzyme activities with the increasing concentrations of the drug, and almost all these increases were statistically significant (p < 0.05). GST activity statistically

significantly increased at 110 µM throughout all incubation times, and these increases showed strong negative correlations (r=-0,994) with GSH values (data not shown). GSHPx is responsible for the reduction of lipid hydroperoxides and H₂O₂ to corresponding alcohols and water, respectively, and it is thought to protect the cell against mild oxidative stress (Mates 2000). Similar to the other two enzymes, the GSHPx activities in CLT treated yeast cells were generally higher than those in the control group for all time points, with generally statistically significant differences, and they increased as the drug concentration increases (Fig. 4C). The enzyme activity increased up to approximately 2.56-fold of the control in cells exposed to 125 µM CLT for 30 min. The typical sharp decrease observed in GR and GST enzyme activities was also recorded in GSHPx enzyme activity after 60 min of treatment with 125 µM CLT. Unfortunately, we did not find any significant correlation between GSHPx activity and H₂O₂ levels.

These results clearly show that CLT at the studied concentration range induces the activities of these GSH-related enzymes, as observed in SOD and CAT activities. On the other hand, we found that while GSH: GSSG ratios were generally above the control at 50 and 75 μ M, they fell below control values in a statistically significant manner at the higher doses of the drug (p < 0.0001) (Fig. 4D). As is well known, the ratio of this redox couple is an important indicator of redox status and oxidative stress. Therefore, it is clear that oxidative stress is dominant in the cell despite the increased antioxidant enzyme activities, especially at relatively high concentrations of the drug. To put it simply, clotrimazole-induced toxicity seems to be linked to a higher level of GSSG. Comparable findings were observed





Control

50 µM

75 µM

110 µM

125 µM

Fig. 4 The effects of CLT treatments at different concentrations for 15–90 min on GR (A), GST (B), GSHPx and GSH/GSSG (D) levels of *S. cerevisiae* cells. Data with error bars show the mean \pm S.E.M of

for the antifungals tebuconazole and econazole (Petricca et al. 2023).

Determination of the changes in malondialdehyde levels

The MDA levels in yeast cells treated with CLT are presented in Fig. 5 compared to the untreated control group. As can be clearly seen, these levels statistically significantly increased in CLT exposed yeast cells compared to the untreated control for all studied incubation periods (p < 0.05). The levels reached their maxima by 7.431 ±0.219 µM MDA/gww in the treatment with 125 µM CLT for 90 min. Hence, it can be clearly stated that the damage to membrane lipids cannot be prevented despite the significant increases in the activities of the main antioxidant enzymes.

A recent study on the juvenile catfish *Clarias gariepinus* found similar results (Melefa and Nwani 2021). In contrast to our findings, Usul et al. (2006) demonstrated the protective effect of CLT on spinal cord lipid peroxidation. Finally, we came across a study on the effects of fluconazole and

three experiments. *p < 0.05 denotes significant differences between control and other studied group by Tukey's multiple range tests

itraconazole, other azole-class antifungals. While 0.7 μ M itraconazole caused 3-fold increases in LPO levels of the yeast-like fungus *Cryptococcus gattii*, 104 μ M fluconazole did not change the levels after 1 h of treatment (Ferreira et al. 2013).

As a conclusion, the results clearly demonstrated that although increases were observed in the activities of the main antioxidant enzymes compared to their control values, the inductions in radical production could not be prevented and showed their adverse effects on cells with the increases in membrane LPO levels. The drug potentially reduces the defenses of cells, thereby creating opportunities for other oxidative stress-related conditions, such as cellular apoptosis. **Fig. 5** The effects of CLT treatments at different concentrations for 15–90 min on LPO levels of *S. cerevisiae* cells. Data with error bars show the mean \pm S.E.M of three experiments. *p < 0.05denotes significant differences between control and other studied group by Tukey's multiple range tests



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Data availability The data that support the findings of this study are available on request from the corresponding author. The data are not publicly available due to privacy or ethical restrictions.

Declarations

Ethics approval and consent to participate Not applicable.

Consent for publication Not applicable.

Competing interests Berna Kavakc?o?lu Yard?mc? and Leman Tarhan declare that they have no conflict of interest.

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