

# Attenuation of Sulfur Dioxide Damage to Wheat Seedlings by Co-exposure to Nitric Oxide

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Received: 30 June 2016 / Accepted: 2 May 2017 / Published online: 11 May 2017  
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**Abstract** The protective function of nitric oxide (NO) has been extensively clarified in plant responses to abiotic stresses. However, little is known about the regulation of NO in plants exposed to sulfur dioxide (SO<sub>2</sub>). In the present study, we found that co-exposure to NO significantly attenuated SO<sub>2</sub>-induced wheat seedling growth inhibition. Data showed that NO efficiently prevented SO<sub>2</sub>-triggered oxidative stress, as indicated by decreasing reactive oxygen species production, lipid peroxidation, and electrolyte leakage. This might be attributed to the regulatory role of NO in antioxidative defense, such as increasing the activities of antioxidative enzymes and the contents of non-enzymatic antioxidants. The SO<sub>2</sub>-caused declines in soluble protein and chlorophyll content were efficiently recovered by NO application. Photosynthetic parameters, such as net photosynthetic rate, maximum photochemical efficiency, and actual photochemical efficiency, were protected by NO. In conclusion, this study demonstrated that during SO<sub>2</sub> exposure, co-application of NO can efficiently alleviate plant damage probably by regulating the antioxidative defense, and protecting plant photosynthesis-related process.

**Keywords** Sulfur dioxide · Nitric oxide · Wheat · Antioxidative defense · Photosynthesis

Sulfur dioxide (SO<sub>2</sub>) is a major air pollutant produced by burning and processing of sulfur-containing fossil fuels.

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Although the release of SO<sub>2</sub> has been strictly controlled, its atmospheric concentration has increased in many areas of the world during the past few decades due to increased SO<sub>2</sub> emissions (Smith et al. 2011). Exposure to high doses of SO<sub>2</sub> can induce plant leaf chlorosis and necrosis, growth inhibition, even death (Agrawal et al. 2003). Several mechanisms have been proposed to account for the SO<sub>2</sub> phytotoxicity, but the essential mechanism is still unclear. SO<sub>2</sub> enters plant leaves mainly through stomata, and fast dissolves in cellular fluid to form sulfite (SO<sub>3</sub><sup>2-</sup>) and bisulfite (HSO<sub>3</sub><sup>-</sup>), which have been verified to be toxic to many plant biochemical and physiological processes (Pfan and Heber 1986). For instance, bisulfite may reduce photosynthesis rate and thereby crop yield through inducing reactive oxygen species (ROS) production probably by plasma membrane NADPH (reduced form of nicotinamide adenine dinucleotide phosphate) oxidase (Li et al. 2007). The inhibition of photosynthesis and the formation of ROS are generally recognized as the first effects of SO<sub>2</sub> on plants (Bowler et al. 1992; Nouchi 2002). ROS in excess causes cell membrane and DNA damage, protein and enzyme inactivation (Valko et al. 2007). Therefore, in the present study, the photosynthesis-related performance and the oxidation-antioxidation contest are analyzed to evaluate the NO attenuation in SO<sub>2</sub>-induced damage to wheat seedlings.

Nitric oxide (NO) is a redox signaling molecule with versatile biological functions. Since the end of the twentieth century when the plant function of NO was suggested, more and more researches have linked NO with a large number of plant phenomena (Wendehenne and Hancock 2011). In plant response to stress conditions, NO can exert a positive or negative regulatory role depending on its local concentration and spatial generation patterns (Mur et al. 2012). Although NO is an uncharged free radical with limited reactivity, it can interact rapidly with other radicals (for

a review see Fancy et al. 2017). For example, NO can react with superoxide anion ( $O_2^{\cdot-}$ ) to produce a more powerful oxidant ONOO<sup>-</sup>. Also,  $NO_2^{\cdot-}$  and  $NO_3^{\cdot-}$  are generated following the  $O_2^{\cdot-}$  oxidation of NO. Nitric oxide also interacts with  $O_2$  to form  $NO_2$ . These resultant molecules are much stronger oxidants compared with NO. However, at low concentrations, NO provides a protective role in plant response to abiotic stress such as high salt, heavy metals, drought, excess light, cold, heat, ozone, UV-B, and so on (Fancy et al. 2017). Many mechanisms have been proposed to decipher the plant protection of NO, among which the scavenging of ROS and the regulation of stomatal aperture are generally accepted. Based on the fact that little is known about the NO regulatory role in plant response to  $SO_2$ , we carried out this study. The aim is to explore plant response mechanisms to  $SO_2$  and NO alone or in combination.

## Materials and Methods

Wheat (*Triticum aestivum* L.) seeds were purchased from a local market. Healthy seeds were surface sterilized with sodium hypochlorite (0.1%, w/v) for 5 min, then washed by rinsing in distilled water, and sown in pots containing a mixture of peat + perlite (3:1, v/v) in a growth chamber under 14-h photoperiod (irradiance of  $200 \mu\text{mol m}^{-2} \text{s}^{-1}$ ), day/night temperature of 22/18°C and relative humidity of 75%. Plants were randomly divided into control group (control plants),  $SO_2$  treatment group ( $SO_2$  plants), NO treatment group (NO plants), and  $SO_2$ +NO treatment group ( $SO_2$ +NO plants). For the exposure of  $SO_2$  and NO alone or in combination, 2-week-old plants were transferred to an air tight glass chamber ( $0.8 \times 0.8 \times 0.8$  m), and  $SO_2$  and/or NO was supplied directly from cylinders into a dilution reservoir into which charcoal filtered air was drawn. The  $SO_2$  and NO gases are purchased from Dalian Special Gases Co., LTD (Dalian, China). The original concentrations of  $SO_2$  and NO gases in cylinders are 2% and 3%, respectively, prepared by charcoal filtered compressed air. The diluted gas mixture was delivered to fumigation chamber equipped with inlet (at the upper portion) and outlet (at the lower portion) tubings. The flow rate was regulated by a mass flow controller (D07-7C; Sevenstar Electronics, Beijing, China). The concentrations of the chamber  $SO_2$  and/or NO were continuously monitored using a gas analyzer (SWG 300-1; MRU, Germany), and the detection limit is 0.1 ppm. The outlet was connected to three glass bottles in tandem containing NaOH solution trap for exhaust gases. To ensure the effective concentrations of the tested gases in the fumigation chamber, we measured the gas recovery rate using standard concentrations of gases, with recovery rates of 94.6% and 97.4% for  $SO_2$  and NO, respectively. For the control, charcoal-filtered air was supplied. Plants were

fumigated during the light period for 6 h  $d^{-1}$  for 3 days. The fumigation concentrations of  $SO_2$  and NO (if any) were  $20 \pm 1$  and  $10 \pm 1$  ppm, respectively. Although concentrations greater 2 ppm  $SO_2$  are not generally obtained in the environment, higher concentrations of  $SO_2$  are the usual doses in an acute exposure experiment such as in the previous studies (Li and Yi 2012; Li et al. 2016) where  $30 \text{ mg m}^{-3}$   $SO_2$  (about 10.5 ppm) was used. Except where mentioned in the text, all physiological and biochemical analyses were carried out immediately following the end of fumigation. Each experiment was performed separately three times.

At the end of fumigation, aerial plant parts were harvested, and washed to remove surface residues by rinsing in tap water, then blotted moisture with filter papers and weighed as fresh weight. For dry weight determination, the plant was dried at 80°C in an oven for 48 h. To observe plant recovery after fumigation, the plant growth was prolonged for 7 days at ambient air conditions, and the fresh weight and dry weight were respectively determined with above-mentioned methods. Pictures of aerial plant parts were taken before harvesting.

Total chlorophyll contents were determined using an 80% (v/v) acetone extract of leaves, and soluble protein contents were measured using the Coomassie Brilliant Blue colorimetric method, for a detail description see Zhang and Qu (2003). The leaf levels of hydrogen peroxide were estimated as described by He et al. (2014). Briefly, collected fresh leaves (1.0 g) were homogenized with 10 mL cooled acetone at 10°C using a mortar and pestle. The homogenate was filtered with Whatman No. 1 filter paper, then mixed with 4 mL titanium reagent and 5 mL ammonium solution, and centrifuged at 10,000g for 10 min. The precipitate was dissolved in 10 mL 2 M  $H_2SO_4$ , and recentrifuged. The supernatant was analyzed at 415 nm in a spectrophotometer (UV762, INESA, Shanghai, China). The content of malondialdehyde was assayed according to Shalata and Tal (1998). Electrolyte leakage was determined using conductivity meter (SA29-DDB-11A, Midwest Group, Beijing, China) as described by Hao et al. (2012).

For the determination of antioxidative enzymes, leaves were collected and homogenized using a mortar and pestle under liquid nitrogen. The powder was suspended in 5 mL of ice-cold extract solution consisting of 50 mM phosphate-buffered saline (pH 7.8), 0.1 mM EDTA, 1% (v/v) Triton X-100, and 4% (w/v) polyvinyl pyrrolidone. The homogenate was incubated for 10 min on ice, and centrifuged at 12,000g at 4°C for 15 min. The supernatant was used to measure (1) superoxide dismutase (SOD; EC 1.15.1.1) by the nitroblue tetrazolium (NBT) method (Beyer and Fridovich 1987), and one unit was defined as the amount of enzyme required to inhibit the photoreduction of NBT by 50%; (2) peroxidase (POD; EC 1.11.1.7)

according to Hemeda and Klein (1990). An increase per minute of 0.01 absorbance value at 470 nm was defined as one unit of POD; (3) catalase (CAT; EC 1.11.1.6) by measuring the decomposition of  $\text{H}_2\text{O}_2$  at 240 nm for 3 min as described by Aebi (1983), where 0.04% (v/v) of  $\text{H}_2\text{O}_2$  was added as the initial concentration. A decrease per minute of 0.01 absorbance value at 240 nm was defined as one unit of CAT. For all these measurements, a spectrophotometer UV762 (INESA, Shanghai, China) was used. For the expression of specific activity, protein content in the crude enzyme extract was quantified by the method of Bradford (1976) using bovine serum albumin as standard.

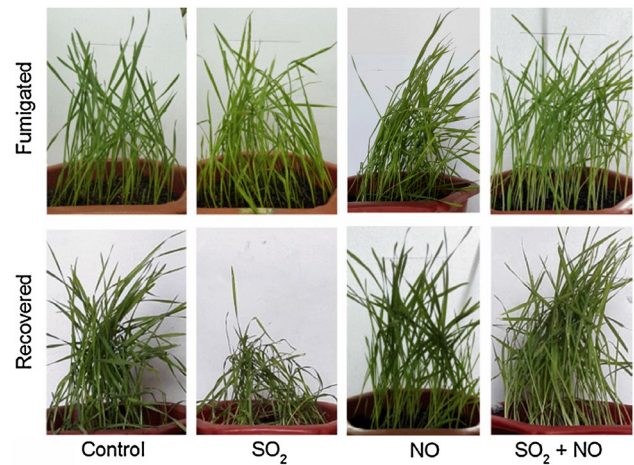
For the determination of ascorbic acid (AA) content, fresh leaves were ground under liquid nitrogen, and the powder was added to a test tube containing 5 mL of 10% trichloroacetic acid. The mixed solution was incubated for 10 min on ice, centrifuged at 12,000g for 15 min at 4°C, and supernatant was used to assay AA content based on the method of Keller and Schwager (1977). The reduced glutathione (GSH) content was detected as described by Griffith and Meister (1979). Fresh leaves were ground under liquid nitrogen, and the powder was placed in a test tube containing 5 mL of 2% metaphosphoric acid, and incubated for 10 min on ice. After centrifuged at 12,000g for 15 min at 4°C, the supernatant was neutralized by adding 0.6 mL 10% sodium citrate.

Net photosynthetic rate and stomatal conductance were measured using a portable photosynthetic system (LI-6200, LI-COR, Lincoln, NE, USA) at ambient conditions, irradiance of  $200 \mu\text{mol m}^{-2} \text{s}^{-1}$ , and 25°C. Maximum photochemical efficiency ( $F_v/F_m$ ) and actual photochemical efficiency of PS2 ( $\Phi\text{PS2}$ ) were analyzed using a fluorometer (Handy-PEA, Hansatech, Norfolk, UK). Briefly, plants were dark-adapted for 15 min before measurement. Minimal fluorescence yield ( $F_0$ ) was detected with a weak irradiance ( $0.5 \mu\text{mol m}^{-2} \text{s}^{-1}$ ) from a light-emitting diode. Maximum fluorescence ( $F_m$ ) was determined by a 1.0-s saturating pulse ( $1500 \mu\text{mol m}^{-2} \text{s}^{-1}$ ), and steady state fluorescence ( $F_s$ ) was detected by 15-min actinic light ( $30 \mu\text{mol m}^{-2} \text{s}^{-1}$ ).  $F_v/F_m = (F_m - F_0)/F_m$ , and  $\Phi\text{PS2} = (F_m - F_s)/F_m$ .

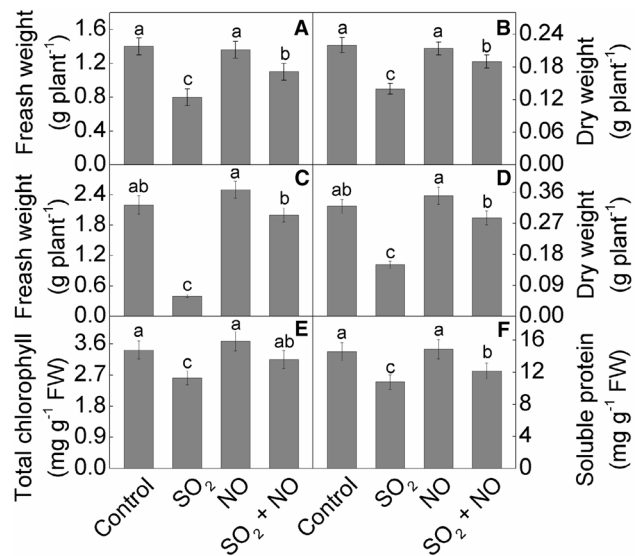
Experiment data are expressed as means  $\pm$  SD. Difference among treatments were analyzed by one-way analysis of variance (ANOVA) using SAS software (SAS Institute, Cary, NC, USA), taking  $p < 0.05$  as significant according to Duncan's multiple range test.

## Results and Discussion

Sulfur dioxide exposure caused leaf wilt and chlorosis, and leaf apex necrosis, especially severe in 7-day-recovery plants (Fig. 1). The phenotype of plants received only NO



**Fig. 1** Representative pictures of wheat seedlings. Fumigated: 14-day-old plants exposed to 20 ppm of  $\text{SO}_2$ , 10 ppm of NO or both  $6 \text{ h d}^{-1}$  for 3 days. Recovered: plants prolonged growth 7 days after the fumigation



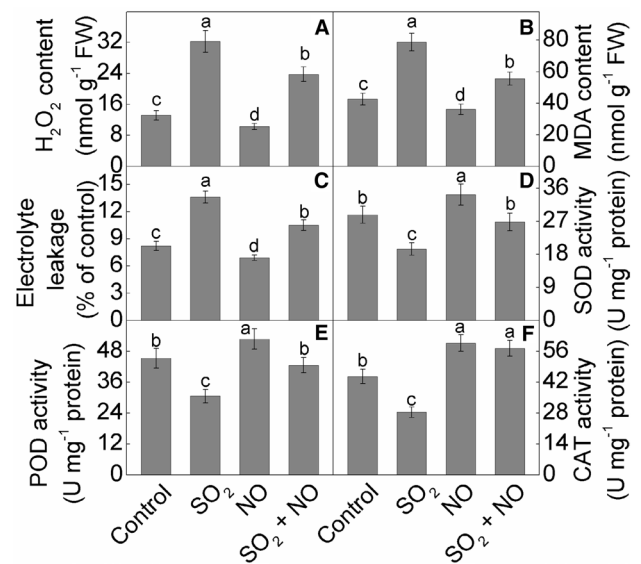
**Fig. 2** Effect of  $\text{SO}_2$  and NO on plant growth, total chlorophyll and soluble protein content. **A** and **B**, fresh and dry weight of aerial plant parts at the end of fumigation, respectively; **C** and **D**, fresh and dry weight of aerial plant parts after 7 days recovery, respectively; **E**, total chlorophyll content; and **F**, soluble protein content. Means  $\pm$  SD,  $n=3$  (different letters indicate significant difference at  $p < 0.05$ )

treatment was similar to control plants. As expected, the  $\text{SO}_2$ -induced damages were dramatically alleviated by co-exposure to NO (Fig. 1). The fresh and dry weight of aerial plant parts at the end of fumigation and after the recovery period also properly reflected  $\text{SO}_2$ -induced plant damage, as well as the repairing function of NO (Fig. 2a–d). Interestingly, although NO did not alter the fresh or dry weight of plants at the end of fumigation (Fig. 2a, b), it moderately

promoted plant growth during the recovery period, as compared with control plants (Fig. 2c, d).

In the previous SO<sub>2</sub>-related studies, the mean concentration of fumigation chamber SO<sub>2</sub> was monitored by colorimetric measurement such as using pararosaniline hydrochloride spectrophotometry (Hao et al. 2005, 2011; Li et al. 2016), or SO<sub>2</sub> was supplied by SO<sub>2</sub> donors (Wei et al. 2013). In the present study, a gas analyzer was used to continuously monitor the fumigation concentration, thereby greatly improved the experimental accuracy. Our data clearly showed that the effect of SO<sub>2</sub> on wheat seedlings, as indicated by leaf morphology, dry weight, especially the fresh weight, was much more prominent during the recovery period than the fumigation period. This suggested that SO<sub>2</sub>-induced subsequent reactions may be more toxic to plants. The co-exposure to NO almost completely mitigated the SO<sub>2</sub> phytotoxicity in the recovery plants (Figs. 1, 2c, d), implying that NO-induced subsequent changes may provide a protective effect even long after the NO is removed. Additionally, NO treatment alone promoted plant growth compared with control plants during the 7-day-recovery period, although to a lesser extent. This further confirmed the NO function inducing a primed state, which benefits plant growth, or renders plants to respond more efficiently to subsequent stress conditions (Conrath 2011). The prolonged effect of exogenous NO might be related to the induction of endogenous NO synthesis (Groß et al. 2013). In addition, NO gas was used in this study instead of NO donors. Although many NO-releasing compounds can be used conveniently in most of the NO-related studies, they have undesired side-effects on plants probably by the remaining products when all NO has been produced. Also, the *in planta* kinetics of NO generation by various NO donors are different, including NO releasing amount, timing and persistence (Mur et al. 2013). Therefore, NO gas fumigation is an alternative method for the application of NO (Bai et al. 2011; Kovacs et al. 2015).

Because of the severe damage of SO<sub>2</sub> plants after recovery period, physiological and biochemical parameters were measured only on the plants at the immediate end of fumigation. When compared with the control level, total chlorophyll content in SO<sub>2</sub> plants decreased significantly, whereas it was maintained at the same level in NO plants, and just slightly declined in SO<sub>2</sub>+NO plants (Fig. 2e). Similarly, the content of soluble protein was comparable to the control level in NO plants, but obviously lowered in SO<sub>2</sub> plants, which was largely resumed by NO co-exposure (Fig. 2f). The level of H<sub>2</sub>O<sub>2</sub> dramatically increased in SO<sub>2</sub> plants (Fig. 3a), correspondingly, the content of malondialdehyde (MDA) and electrolyte leakage were greatly elevated (Fig. 3b, c), indicating that the SO<sub>2</sub> exposure in this study had caused oxidative damage to plants. However, the SO<sub>2</sub>-induced increases in H<sub>2</sub>O<sub>2</sub>, as well as in the



**Fig. 3** Effect of SO<sub>2</sub> and NO on oxidative stress and antioxidative defense. A, H<sub>2</sub>O<sub>2</sub> content; B, malondialdehyde content; C, electrolyte leakage; D, SOD activity; E, POD activity; and F, CAT activity. Means ± SD, n=3 (different letters indicate significant difference at  $p < 0.05$ )

oxidative stress as shown by MDA production and electrolyte leakage, were efficiently prevented by NO co-exposure (Fig. 3a–c). The values of all these oxidative stress-related parameters were at lower levels in NO plants than those in control plants (Fig. 3a–c).

The protective role of NO in plant chlorophyll against various stress factors have been repeatedly reported (e.g. Fatma et al. 2016 and references therein). This might be one of the mechanisms responsible for NO-mediated protection in photosynthetic rate under stress conditions. Also, many publications documented that exogenous NO can efficiently reverse stress-induced soluble protein decrease (e.g. Zeng et al. 2011). Several lines of evidence have demonstrated that SO<sub>2</sub> can promote ROS production and cause oxidative stress (Dat et al. 2000; Hao et al. 2011). In this study, the SO<sub>2</sub>-induced oxidative damage may be a major reason causing plant chlorosis, necrosis, and protein degradation. Exogenous NO always improved abiotic stress tolerance associated with a decrease in the levels of H<sub>2</sub>O<sub>2</sub> and MDA (Groß et al. 2013). This held true in the case of SO<sub>2</sub> stress. As a direct scavenger of ROS, NO is not just a down-stream signal of H<sub>2</sub>O<sub>2</sub> but also affect ROS generation and degradation (Groß et al. 2013). For instance, NO suppresses ROS accumulation by inhibiting the activity of ROS-producing enzyme NADPH oxidase (Yun et al. 2011).

The activities of antioxidative enzyme SOD, POD and CAT, and the contents of non-enzymatic antioxidant glutathione and ascorbic acid were markedly decreased under SO<sub>2</sub> exposure, whereas they were at a comparable level,

even further enhanced in CAT activity in  $\text{SO}_2 + \text{NO}$  plants relative to control plants. Without exception, the above-mentioned antioxidants increased significantly in NO plants (Figs. 3d–f, 4a, b). This correlated well with the change trends in  $\text{H}_2\text{O}_2$ , MDA and electrolyte leakage. A large body of evidence has shown that exogenous NO-mediated plant tolerance to stress conditions is associated with NO as a signal molecule stimulating the antioxidant system and/or a primed state of stress defense. For example, exogenous application of NO (as NO donors or NO gas) can efficiently up-regulate the activities of SOD, POD and CAT (for a review see Groß et al. 2013). Apart from promoting the expression level of enzymatic proteins, increasing evidence also showed that NO improved antioxidant enzyme activities via the regulation of antioxidant protein carbonylation and S-nitrosylation statuses (e.g. Bai et al. 2011). Also, many studies have found that exogenous NO might promote the contents of glutathione and ascorbic acid under stress conditions (such as Keyster et al. 2012; Hasanuzzaman and Fujita 2013). The study by Innocenti et al. (2007) demonstrated that exogenous NO stimulated transcription levels of genes involved in glutathione synthesis, thereby resulting in an increase of glutathione.

The exposure to  $\text{SO}_2$  obviously lowered stomatal conductance regardless of whether NO was present, whereas NO treatment alone markedly increased stomatal conductance (Fig. 4c). Photosynthetic rate (Pn), maximum photochemical efficiency (Fv/Fm) and actual photochemical efficiency ( $\Phi\text{PS2}$ ) were significantly decreased by  $\text{SO}_2$  (Fig. 4d–f), but the reduced magnitudes of Pn and  $\Phi\text{PS2}$

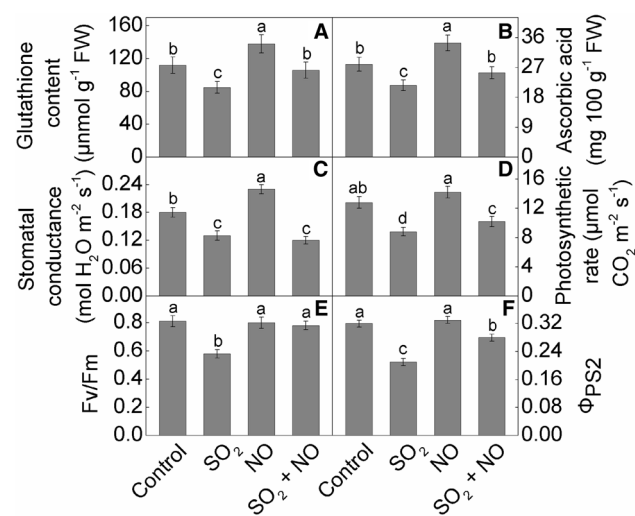
obviously narrowed in  $\text{SO}_2 + \text{NO}$  plants (Fig. 4d, f), and the ratio of Fv/Fm was even at a comparable level to the control (Fig. 4e). Nitric oxide treatment alone did not affect the levels of Fv/Fm and  $\Phi\text{PS2}$ , but enhanced Pn to a lesser extent (Fig. 4d). The inhibited effect of  $\text{SO}_2$  on stomatal conductance, as well as the photosynthesis-related parameters was also found in Arabidopsis plants (Hao et al. 2011). Many studies have demonstrated that NO is involved in plant stomata movements, either opening or closing depending on NO concentrations tested (for a review see Mur et al. 2013). However, our data showed that NO used alone increased stomatal conductance, but simultaneous use of NO did not influence the  $\text{SO}_2$ -induced inhibitory extent on stomatal conductance, implying that the alleviation effect of NO on  $\text{SO}_2$  plants was related to other factors rather than to stomatal factors, even though  $\text{SO}_2$  enters mesophyll mainly via stomata. Many studies have shown that exogenous NO can efficiently prevent stress-induced decrease in chlorophyll fluorescence parameters such as  $\Phi\text{PS2}$  and Fv/Fm, thereby protecting photosynthesis (e.g. Zhang et al. 2006). In the present study, co-exposure to NO largely counteracted the  $\text{SO}_2$  damage to  $\Phi\text{PS2}$  and Fv/Fm, even though NO used alone did not affect these two parameters (Fig. 4e, f).

In conclusion, this short paper demonstrated that co-exposure to NO gas can efficiently alleviate  $\text{SO}_2$ -induced plant damage by enhancing antioxidative capability, reducing oxidative stress, protecting photosynthesis system, therefore, maintaining a moderate photosynthetic rate and plant growth. This extends the list of NO involved in plant response to abiotic stresses.

**Acknowledgements** This research was supported by the National Natural Science Foundation of China (Grant No. 31572213, 31570502, 31270446).

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**Fig. 4** Effect of  $\text{SO}_2$  and NO on non-enzymatic antioxidant contents, and photosynthesis-related parameters. A, glutathione content; B, ascorbic acid content; C, stomatal conductance; D, net photosynthetic rate; E, maximum photochemical efficiency; and F, actual photochemical efficiency. Means  $\pm$  SD,  $n=3$  (different letters indicate significant difference at  $p < 0.05$ )

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