

## **Inter-annual Variation of Ambient Ammonia and Related Trace Gases in Delhi, India**

**S. K. Sharma1,2 · Saraswati1,2 · T. K. Mandal1,2 · Mohit Saxena1**

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**Abstract** In this study, ambient  $NH_3$ , NO, NO<sub>2</sub>, CO and  $SO<sub>2</sub>$  were measured continuously from February, 2008, to December, 2016. The annual average mixing ratios (mole/ mole) of NH<sub>3</sub>, NO, NO<sub>2</sub>, CO and SO<sub>2</sub> were  $17.8 \pm 3.4$  ppb; 21.2 $\pm$ 2.3 ppb, 18.1 $\pm$ 3.2 ppb; 1.7 $\pm$ 0.3 ppm and  $2.0 \pm 0.3$  ppb, respectively. All the trace gases (NH<sub>3</sub>, NO,  $NO<sub>2</sub>$ , CO and  $SO<sub>2</sub>$ ) showed significant annual variation during the study. A significant increasing trend in mixing ratios of ambient  $NH_3$  and  $NO_2$  were observed at the observational site of Delhi, whereas, increasing trend were recorded in case of NO, CO and  $NO<sub>2</sub>$  mixing ratios. The results emphasized that traffic could be one of the significant sources of ambient  $NH<sub>3</sub>$  at the urban site of Delhi, as illustrated by positive correlations of  $NH<sub>3</sub>$  with traffic related co-pollutants (NO and CO).

**Keywords** Trace gases · Ammonia · Meteorological parameters · Annual variation

Since the gaseous toxic air pollutants are normally associated with acute short-term and chronic long-term human health problems, the study of these pollutants is essential to understand their mechanism of occurrence (Behera et al. [2015](#page-3-19); Saraswati et al. [2017](#page-3-20)). Ammonia  $(NH_3)$  is the third most abundant nitrogen containing gas in the atmosphere

 $\boxtimes$  S. K. Sharma sudhir.npl@nic.in; sudhircsir@gmail.com after  $N_2$  and  $N_2O$  (Seinfeld and Pandis [1998](#page-3-0); Aneja et al.  $2000$ ). As a primary alkaline gas in ambient air, NH<sub>3</sub> can neutralize nitric acid and sulphuric acid gases to form ammonium nitrate and sulphate (Pinder et al. [2007](#page-3-2); Sharma et al. [2012c](#page-3-3), [2014](#page-3-4), Saxena et al. [2017](#page-3-5)), which are important constituents of airborne fine particles or  $PM<sub>2.5</sub>$  (Chow et al. [2004](#page-3-6); Kean et al. [2000;](#page-3-7) Aneja et al. [2001](#page-3-8); Ram et al. [2010](#page-3-9); Huang et al.  $2011$ ). The most recent consideration for NH<sub>3</sub> emissions on the global scale is linked to climate change based on its ability to form  $PM<sub>2.5</sub>$ , specifically ammonium sulphate (Aneja et al. [2001](#page-3-8)). These aerosols can possibly increase the Earth's albedo (Sutton et al. [2000\)](#page-4-0). The need to better understand the role of this particular air pollutant has been underscored in recent years as its atmospheric concentration increases. Effective management strategies are essential to mitigate the pollutants over public health concern and to enhance visibility (Brunekreef and Holgate [2002](#page-3-11); Wang et al. [2012](#page-4-1)).

Major anthropogenic sources of ambient  $NH<sub>3</sub>$  are agricultural practices, livestock, transport and industrial activities (Sutton et al. [2000;](#page-4-0) Li et al. [2006;](#page-3-12) Sharma et al. [2010](#page-3-13)), along with natural sources like forest fires and emissions from soil (Olivier et al. [1998](#page-3-14); Lee et al. [2005](#page-3-15)). Recent studies (Sutton et al. [1995](#page-4-2), [2000](#page-4-0); Wang et al. [2012](#page-4-1)) have established that reduction of NO compounds in three-way catalytic converters of automobile exhaust and industrial power stations emissions are significant contributors to atmospheric  $NH<sub>3</sub>$  in urban environments  $(2NO + 2CO + 3H<sub>2</sub> \rightarrow 2NH<sub>3</sub> + 2CO<sub>2</sub>$  or  $2NO + 5H<sub>2</sub>$  $\rightarrow$  2NH<sub>3</sub> + 2H<sub>2</sub>O) (Gandhi and Shelef [1991](#page-3-16)). Correlation matrix of ambient  $NH<sub>3</sub>$  with CO also suggested the dependency of  $NH<sub>3</sub>$  mixing ratio on the traffic intensity and air temperature (Perrino et al. [2002](#page-3-17); Meng et al. [2011](#page-3-18); Sharma et al.  $2014$ ). Li et al. ([2006\)](#page-3-12) observed spikes in the NH<sub>3</sub> time-series that was correlated with the spikes in the CO,

<sup>1</sup> Environmental Sciences and Biomedical Metrology Division, CSIR-National Physical Laboratory, Dr. K S Krishnan Road, New Delhi 110 012, India

<sup>2</sup> Academy of Scientific and Innovative Research (AcSIR), CSIR-National Physical Laboratory campus, New Delhi 110 012, India

and inferred that the  $NH_3$  came from nearby traffic. In the previous study, we have reported the diurnal, seasonal and temporal variations in mixing ratios of ambient  $NH<sub>3</sub>$  and its interaction with other trace gases and reference theirein (Sharma et al. [2010a](#page-3-13), [b,](#page-3-21) [2012a,](#page-3-22) [b](#page-3-23), [c](#page-3-3), [2014;](#page-3-4) Saraswati et al. [2017](#page-3-20)). In this paper, we present the inter-annual variation in mixing ratios of ambient  $NH<sub>3</sub>$  and other related trace gases (NO, NO<sub>2</sub>, CO and SO<sub>2</sub>) in an urban area of Delhi.

## **Materials and Methods**

Delhi, the capital of India is surrounded by four different climatic zones (Himalyas in the north, central hot plains in the south, the Thar desert in the west and the Indo Gangetic plain in the east), which influence its semi-arid climate consided as one of the most polluted megacites of the world. The mixing ratios (mole/mole or mole fraction) of ambient  $NH_3$ , NO, NO<sub>2</sub>, CO and SO<sub>2</sub> were measured, along with meteorological parameters (temperature, relative humidity (RH), wind speed and wind direction) at the CSIR-National Physical Laboratory, New Delhi (28°38′N, 77°10′E; 218 m amsl) from February, 2008, to December, 2016. The observational site is amenable to free wind flow from all directions and represents a typical urban atmosphere, surrounded by huge roadside traffic  $(\sim 100 \text{ m})$  and agricultural fields in the southwest direction (~500 m). The observational area is under the influence of air mass flow from north-east to north-west in winter and from southeast to south-west in the summer. The temperature of Delhi varied from maximum in summer to minimum in winter. Detailed descriptions of the sampling site including meteorology are available in Sharma et al. [\(2016a](#page-3-24)).

Ambient  $NH<sub>3</sub>$  mixing ratio was measured initially using an  $NH_{3}$ -analyzer (CLD88CYp, ECO Physics AG, Duernten, CH) operating (accuracy:  $\pm 0.5$  ppb) on a chemiluminescence method (February, 2008, to December, 2012). From January, 2012, to December, 2015, the measurement of ambient  $NH_3$  was carried out using an  $NH_3$ -analyzer (AC32M&CNH3, Environment SA, Paris, FR), which also

operates (accuracy:  $\pm 1.0$  ppb) on the chemiluminescence method. NO and  $NO_2$  were measured using a  $NO_x$ -analyzer (CLD 88p, ECO Physics AG, Duernten, CH) with a photocatalytic converter (Model: PLC 860 M/s. ECO Physics AG; accuracy:  $\pm 0.5$  ppb). Carbon monoxide (CO) was observed continuously using a non-dispersive infrared (NDIR) gas filter correlation analyzer (Model 48 C; Thermo Fisher Scientific, Waltham, MA, USA). Sulphur dioxide was measured using a  $SO_2$ -analyser (APSA 360 A, Horiba Ltd, Japan). The detailed principle of operations, detection limits, calibration procedures, calibration standards used (traceable to NIST-USA)and repeatability etc., of all the respective analyzers are available in Saraswati et al.  $(2017)$  $(2017)$ . All the analytical instruments showed <2.0% repeatability error during the span calibration. The meteorological parameters (temperature, RH, wind speed and wind directions) were measured simultaneously using sensors at a meteorological tower, which was 100 m away from the observational site within the same campus (Sharma et al. [2010a](#page-3-13), [b](#page-3-21), [2014](#page-3-4)).

## **Results and Discussion**

The annual average mixing ratios of ambient  $NH<sub>3</sub>$ , NO, NO<sub>2</sub>, CO and SO<sub>2</sub> were  $17.8 \pm 3.4$  ppb;  $21.2 \pm 2.3$  ppb and  $18.1 \pm 3.2$  ppb;  $1.7 \pm 0.3$  ppm and  $2.0 \pm 0.3$  ppb, respectively from February, 2008 to December, 2016 (Table [1](#page-1-0)). The highest annual average mixing ratio of ambient  $NH<sub>3</sub>$ was observed in 2014 (24.3 $\pm$ 3.6 ppb), whereas the minimum mixing ratio ambient  $NH<sub>3</sub>$  was recorded in 2010  $(13.4 \pm 4.2 \text{ pb})$ . Similarly, higher annual average mixing ratios of ambient NO and  $NO<sub>2</sub>$  were recorded in 2016  $(25.6 \pm 7.5 \text{ pb})$  and  $2015 (23.9 \pm 6.6 \text{ pb})$ , respectively. Figure [1](#page-2-0) showed the annual variation alongwith trend in mixing ratios of ambient  $NH<sub>3</sub>$ , NO, NO<sub>2</sub>, CO and SO<sub>2</sub> during 2008–2016. Mixing ratios of all the trace gases varied non-significantly, except for ambient  $NH_3$  and  $NO_2$  during the study. This may have been due to source strength of

<span id="page-1-0"></span>



± Standard deviation

a Significantly different (*p*<*0.05*)

<sup>b</sup>Not significantly different  $(p > 0.05)$ 



<span id="page-2-0"></span>**Fig. 1** Annual variation in mixing ratios of ambient  $NH_3$ , NO, NO<sub>2</sub>,  $CO$  and  $SO<sub>2</sub>$  in Delhi, India

respective trace gases and prevailing meteorological conditions at the observational site in Delhi.

Kapoor et al.  $(1992)$  $(1992)$  reported the average NH<sub>3</sub> mixing ratio of NH<sub>3</sub> as  $47.3 \pm 13.6$  ppb in Delhi, whereas, Sharma et al. [\(2010](#page-3-13)) reported the average mixing ratio of ambient NH<sub>3</sub> as  $20.23 \pm 2.71$  ppb. Zutshi et al. ([1970\)](#page-4-3) reported the  $NH_3$  mixing ratio as 50.7 ppb at Mumbai, whereas, Sharma et al.  $(2016b)$  $(2016b)$  reported the NH<sub>3</sub> mixing ratio as  $43.4 \pm 7.0$  ppb at Kolkata, India. The diurnal, seasonal and

Annual average mixing ratios and inter-annual variations in mixing ratios of  $NH_3$ , NO, NO<sub>2</sub>, CO and SO<sub>2</sub> (from the average value of the study period) are summarized in Table [1.](#page-1-0) The maximum annual variation in mixing ratio of ambient  $NH<sub>3</sub>$  was in 2014 (6.58 ppb), whereas minumum in 2015 (0.72 ppb). The minimum and maximum inter-annual variations in mixing ratio of ambient NO were −0.35 ppb in 2012 and 4.35 ppb in 2016, respectively. Similarly, maximum (5.78 ppb) significant inter-annual variation in mixing ratio of  $NO<sub>2</sub>$  was observed during 2015. The mixing ratios of CO and  $SO_2$  were also varying inter-annually during the study period (Table [1\)](#page-1-0). The inter-annual variation in mixing ratios of  $NH_3$ , NO, NO<sub>2</sub>, CO and SO<sub>2</sub> were observed may be due to source strength of respective traces and meteorological condition of the observational site.

Although agriculture has historically been the major source of ambient  $NH<sub>3</sub>$ , however, the contribution of vehicles equipped with three-way catalytic converters has increased considerably at the urban sites (Sutton et al. [2000](#page-4-0); Heeb et al. [2008\)](#page-3-27). Therefore, in order to examine the contribution of traffic to  $NH<sub>3</sub>$ , it may be useful to compare/ correlate ambient  $NH<sub>3</sub>$  with those of primary pollutants mainly emitted by motor vehicle exhausts, such as NO and CO. Table [2](#page-2-1) summarized the correlations among  $NH<sub>3</sub>$  versus NO,  $NH_3$  versus  $NO_2$ ,  $NH_3$  versus CO and  $NH_3$  versus  $SO<sub>2</sub>$  during 2008–2016, along with meteorological parameters at the study site.

The correlation matrix of hourly annual average mixing ratios of  $NH<sub>3</sub>$  with NO and CO during 2008–2016 showed positive correlation ( $r=0.53$  NH<sub>3</sub> vs. NO;  $r=0.66$  NH<sub>3</sub> vs. CO), supporting the hypothesis that traffic is also an important source of  $NH<sub>3</sub>$  at the observational site (Table [2\)](#page-2-1). Meng et al. [\(2011](#page-3-18)) also reported a strong positive correlation  $(r=0.78)$  between NH<sub>3</sub> and NO during winter at Beijing,

<span id="page-2-1"></span>



\*Significant at *p*<0.05

China, and considered traffic to be one of the sources of ambient  $NH<sub>3</sub>$  at there. The correlation coefficient of  $NH<sub>3</sub>$ versus CO was also reported as  $r=0.71$  during winter in Beijing, China (Meng et al. [2011](#page-3-18)), further supporting the present study.  $SO<sub>2</sub>$  also reacts with the hydroxyl radical (OH) to form sulphuric acid (Seinfeld and Pandis [1998](#page-3-0)). Table [2](#page-2-1) shows a positive relationship of between  $NH_3$  and  $NO<sub>2</sub>$  ( $r=0.41$ ), which indicates the formation of  $HNO<sub>3</sub>$  and simultaneously  $NH<sub>4</sub>NO<sub>3</sub>$  in the atmosphere (Stockwell et al. [2000\)](#page-4-4). During night time, the increased humidity and lowered temperature, along with a higher mixing ratio of NH<sub>3</sub>, SO<sub>2</sub> and NO<sub>2</sub> may facilitate the formation of NH<sup>+</sup><sub>4</sub>

aerosol in the atmosphere. Table [2](#page-2-1) also shows the positive relationship between NH<sub>3</sub> and SO<sub>2</sub> during study ( $r=0.32$ ) indicationg the possible formation of  $(NH_4)$ <sub>2</sub>SO<sub>4</sub>.

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