

Photo-induced Toxicity of Four Polycyclic Aromatic Hydrocarbons (PAHs) to Embryos and Larvae of the Carpet Shell Clam *Ruditapes decussatus*

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Received: 16 December 2011 / Accepted: 10 March 2012 / Published online: 24 March 2012
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Abstract In this work, we assessed the photo-toxicity of four polycyclic aromatic hydrocarbons (PAHs) to embryos and larvae of the European clam *Ruditapes decussatus*. The exposure of *R. decussatus* embryos (24 h) and larvae (96 h) to anthracene, fluoranthene, pyrene and benzo[a]pyrene resulted in reduction of normal D-veliger percentages and high larval mortality, both in darkness and under sunlight conditions. Based on the calculated EC₅₀ and LC₅₀ values, the toxicity of the forementioned PAHs was respectively enhanced 72, 35, 60 and 23 times in the embryotoxicity test and 32, 31, 12 and 61 times in the larval mortality test when exposures were performed under sunlight conditions. Simultaneous exposure to sunlight and these PAHs enhanced their toxicity in comparison to dark conditions. The clam embryos and larvae appear to be environmentally relevant life-stages in assessing the toxic and photo-toxic risk of PAHs that enter the marine environment.

Keywords Bioassay · Embryos · Larvae · Clams · Polycyclic aromatic hydrocarbons · Photo-toxicity

Polycyclic aromatic hydrocarbons (PAHs) are a large group of widespread organic compounds of high environmental concern. Even though PAHs occur naturally, the highest concentrations are mainly due to human activities that cause a continuous increase in PAH levels of estuarine and marine waters (Kennish 1992; Walker et al. 2001).

Direct discharges into the marine environment from point sources such as wastewater treatment plants range from <1 µg/L to over 625 µg/L, whilst concentrations of PAHs in industrial effluents range from undetectable to 4.4 mg/L (Latimer and Zheng 2003). Major sources of PAHs to the marine environment are combustion products and petroleum principally from atmospheric deposition (5×10^4 tones/year) and oil spillage (1.7×10^5 tones/year) (Kennish 1992; Meador 2003). Therefore, it is necessary to study the impact of those pollutants to marine organisms of ecological and commercial relevance.

In general, PAHs do not show extremely high acute toxicity to aquatic life, with the majority of intermediate to high molecular weight PAHs not exhibiting acute toxicity within their water solubility limits (NRCC 1983). However, there is a growing body of evidence to suggest that certain PAHs may pose a greater hazard to aquatic organisms than previously demonstrated, due to their potential to cause photo-induced toxicity (photo-toxicity) when exposed to ultraviolet (UV) light.

PAH phototoxicity has been demonstrated in the laboratory, generally for standard aquatic test species, including algae (Gala and Giesy 1992), duckweed (Huang et al. 1993), oligochaetes (Ankley et al. 1995), chironomids (Hatch and Burton 1999), amphipods (Duan et al. 2000; Hatch and Burton 1999), daphnids (Sasson-Brickson and Burton 1991; Wernersson and Dave 1998), brine shrimp (Diamond et al. 2000), mysid shrimp (Pelletier et al. 1997), mollusk larvae (Pelletier et al. 1997; Saco-Álvarez et al. 2008), fish (Diamond et al. 1995; Saco-Álvarez et al. 2008) and amphibians (Hatch and Burton 1998; Monson et al. 1999).

Here we present data on laboratory studies investigating the acute toxicity and potential photo-toxicity of fluoranthène, anthracène, pyrene and benzo[a]pyrene, under natural sunlight (environmentally realistic UV light levels)

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and in dark conditions, towards embryo-larval stages of the European clam *Ruditapes decussatus*. Historically, marine invertebrates have been widely used to monitor the biological effects of contaminants. Furthermore, the Joint Assessment Monitoring Programme (JAMP) has recommended the embryo-larval stages of bivalve as suitable bioassays for assessing water quality in Mediterranean waters (Stagg 1998). To our knowledge, the present work is the first using embryo-larval life-stages of *R. decussatus* to assess photo-toxicity of PAHs. *Ruditapes decussatus* is widely distributed along the Tunisian coast. This species is economically important, being exported mostly to Europe (Hamza-Chaffai et al. 2003). Its spawning season, in our region, extends from March to September, but we can obtain gametes throughout the year by conditioning adults in the laboratory (Medhioub et al. 2006). This species is a sedentary filter-feeding marine bivalve, thus satisfying criteria in biomonitoring programs for a good bioindicator of pollution.

Materials and Methods

Adults of *R. decussatus* collected in pristine sites from natural local populations (Oued Maltine; SE of Tunisia) were transferred to the hatchery of the National Institute of Marine Sciences and Technologies, where they were maintained in aquaria with running natural seawater and fed with the microalgae *Isochrysis galbana* and *Chaetoceros calcitrans* for at least 1 week until the experiments began. Handling conditions of the adult stock were $22.5 \pm 0.5^\circ\text{C}$ temperature, 36.5 ± 1.5 ppt salinity, 7.2 ± 1.2 mg l⁻¹ O₂ and 7.69 ± 0.13 pH (mean \pm SD).

Adult clams were induced to spawn by thermal stimulation (temperature cycles at 18°C and 28°C) in separated beakers with 0.2 μm filtered seawater. Viable gametes from healthy males and females were selected and filtered at 32 μm (sperm) and 100 μm (eggs) to remove impurities.

Experimental concentrations were chosen on the basis of preliminary trials and on data from the literature (Pelletier et al. 1997). Experimental solutions were made by dissolving analytical grade anthracene, fluoranthene, pyrene and benzo[a]pyrene (Sigma–Aldrich, Steinheim, Germany) in HPLC-grade acetone, due to the low solubility of PAHs in seawater (Kennish 1992). Two separate dilution series (five concentrations) were prepared for each PAH, one for testing under sunlight and the second for testing in the dark. Nominal and actual PAH concentrations are shown in Table 1. One control was used in each exposure consisting of artificial seawater (ASW) prepared as in Zarogian et al. (1969). The experimental concentrations were obtained by diluting the stock solutions in ASW. During this dilution, equal amounts of acetone (less than 200 $\mu\text{l l}^{-1}$), found not

Table 1 Nominal and measured ($t = 0$) concentrations ($\mu\text{g/l}$) of the tested PAH used in the embryotoxicity and larval mortality tests both in dark and under sunlight conditions

PAH	Concentrations ($\mu\text{g/l}$)			
	Dark		Sunlight	
	Nominal	Measured	Nominal	Measured
Anthracène	25	27.8	0.5	0.54
	50	48.4	2.5	2.38
	100	107	5	6.3
	250	238	25	26.6
	500	516	50	55
Fluoranthène	12.5	11.9	1	1.4
	25	22.9	2.5	2.7
	50	53.7	5	4.8
	125	124	10	9.2
	250	247.9	25	22.6
Pyrène	8	9.8	0.5	0.47
	32	35.5	2.5	2.4
	64	61	10	11.2
	128	132	25	24.7
	256	251	50	52.4
Benzo[a]pyrene	0.5	0.59	0.1	0.13
	2.5	2.7	0.25	0.28
	5	4.7	0.5	0.48
	25	28.8	1	1.08
	50	52.6	5	5.15

to be toxic in preliminary tests, were added to each experimental beaker with PAHs solutions. All glassware was acid-washed (HNO₃ 10 % vol.) and rinsed with acetone and distilled water before the experiments.

Test solutions intended for chemical analysis were collected from the experimental vials at the beginning of the tests. The test solutions were poured into a separatory funnel and PAHs were extracted with dichloromethane (USEPA 1980). After substitution of the solvent by acetonitrile, the concentration of the PAHs was determined by HPLC (C 20 AT, provided by Shimadzu Co. Tokyo, Japan) coupled with fluorimetric detection. The samples or standard solutions were directly injected onto a PAH–C18 column (20 cm–4.0 mm) at 30°C. Isocratic elution was performed with a mobile phase of acetonitrile/water (60/40, v/v) at a flow rate of 1.0 mL/min. The recoveries in the extraction method were about 94 % for the measured compounds.

For the embryotoxicity test, the fertilized eggs (1 ml ca. 400 eggs/ml), obtained by in vitro fertilization, were transferred to 15 ml vials containing PAH solutions that were incubated under two light conditions (natural sunlight and dark) at ambient temperature (22°C) for 24 h, until the second larval stage (D-veliger) was attained. After incubation, a few drops of 40 % buffered formalin were added,

and the percentage of normal D-hatched larvae was recorded. A larva was considered malformed when some developmental defect was observed (embryos that had not reached the D-veliger stage or D-veliger larvae with convex hinge, incomplete shell or protruding mantle). The EC_{50} was defined as the PAH concentration that resulted in a 50 % reduction in the number of normal embryos. The physico-chemical conditions of the experiments were 32 ± 0.15 ppt salinity, 6.92 ± 0.40 mg l⁻¹ O₂ and 7.75 ± 0.15 pH (mean \pm std).

In the larval mortality test, fertilized eggs were filtered through a 20 μ m mesh nylon net and then washed gently three times with filtered seawater (1 μ m). They were then resuspended in ASW at 23°C. After 24 h, the swimming veliger larvae (D-larvae stage) obtained were re-suspended in a 2 l glass beakers (approximately 2×10^4 larvae), containing 1.8 l of PAH solutions or ASW (control). Larvae were also incubated under two light conditions (natural sunlight and dark) at ambient temperature 24°C and during 96 h with 12/12 h photoperiod. Veliger larvae were fed daily with phytoplankton (*Isochrysis taiti* and *C. calcitrans*). The number of dead larvae was then assessed under a microscope at the end of the experiment, and the percentage of larval mortality was calculated using the following formula (% Mortality = $100 \times \text{dead larvae} / \text{total number of larvae}$). All experiments were conducted in five replicates.

Ultraviolet light levels in natural sunlight were measured at different times of the day. UVA levels ranged from 970 to 2,300 μ W/cm² with an average level of $1,655 \pm 101$ μ W/cm². UVB levels ranged from 140 to 512 μ W/cm² with an average level of 345 ± 87 μ W/cm². Thus, the ultraviolet light levels in the natural sunlight exposures were three to four times greater than those in previous studies using artificial UV light exposures (Pelletier et al. 1997).

Statistical analyses were conducted using the SPSS version 17.0 statistical software (IBM Corporation, USA). Differences between treatments were tested for significance by means of one-way analysis of variance (ANOVA). When differences among groups were significant the Dunnett's test was employed to compare the control group and each of the experimental groups. The EC_{10} , EC_{50} and LC_{50} and their 95 % confidence intervals (95CI) were calculated according to the Probit method after normalizing data to the mean control response using Abbot's formula (Emmens 1948). For analysis, data were first arcsine-transformed to achieve normality (Hayes 1991).

Results and Discussion

Results from the analytical chemistry applied for checking the nominal PAH concentrations in seawater at the

beginning of the experiment are shown in Table 1. Measured concentrations were within 11–18 % of the nominal concentrations. In general, no significant modifications of nominal values of the PAHs concentrations were shown.

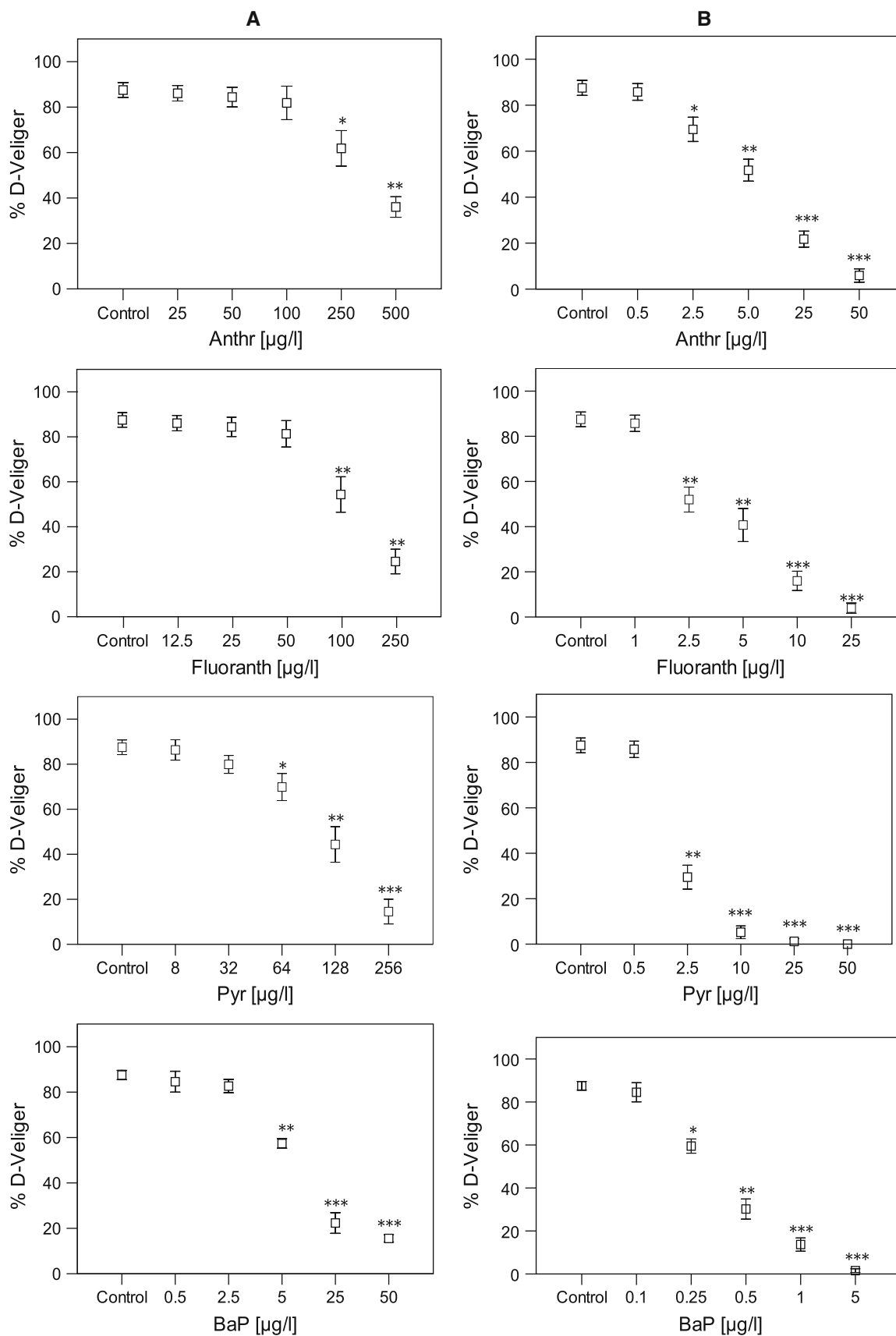
Following 24 h of exposure of *R. decussatus* embryos to increased concentrations of anthracene, significant ($p < 0.05$) decrease of the D-veliger larvae percentages were observed in the concentrations of 250 and 2.5 μ g l⁻¹ in the dark and under sunlight conditions, respectively (Fig. 1). Under sunlight conditions, the inhibition of the embryonic development exceeded 95 % at 25 μ g l⁻¹ of anthracene. The EC_{10} and EC_{50} values of this PAH were 27.9 and 361.1 μ g l⁻¹ (156.5 and 2,026 nM) in the dark, and 0.48 and 4.97 μ g l⁻¹ (2.7 and 27.9 nM) under sunlight conditions, respectively (Table 2).

Fluoranthene also showed toxicity to clam embryos by significantly ($p < 0.05$) reducing the percentages of D-veliger larvae up to 40 %, compared to the control, at concentrations as high as 100 and 2.5 μ g l⁻¹ in dark and under sunlight conditions, respectively (Fig. 1).

The 24 h exposure of clam embryos to increasing concentrations of pyrene resulted in a significant ($p < 0.05$) reductions of D-veliger larvae percentages of 30 and 70 % both at dark and under sunlight conditions from a concentration of 64 and 2.5 μ g l⁻¹, respectively. Clam embryonic development was totally inhibited at the highest tested concentration (50 μ g l⁻¹) under sunlight conditions (Fig. 1). The calculated EC_{10} and EC_{50} values were shown in Table 2. The fourth PAH tested, the benzo[a]pyrene (BaP), was toxic to clam embryos from 5 and 0.25 μ g l⁻¹ in dark and under sunlight conditions, respectively, as shown in Fig. 1. Compared to control, the reduction of normal D-veliger larvae percentages, at these concentrations, was up to 40 %. Under sunlight conditions, BaP induced a total inhibition (99 %) of the clam embryonic development at a concentration of 5 μ g l⁻¹, yielding EC_{10} and EC_{50} values of 0.07 and 0.31 μ g l⁻¹ (0.28 and 1.23 nM), respectively.

Based on the EC_{50} values, the toxicity of the tested PAHs to clam embryos was up to 72, 35, 60 and 23 times higher under sunlight than dark conditions, respectively, for anthracene, fluoranthene, pyrene and benzo[a]pyrene. This indicates a definite phototoxic response of those compounds.

In dark conditions, anthracene, pyrene and benzo[a]pyrene were not ecologically toxic to *R. decussatus* embryos since their EC_{50} values (361.1, 113.8, 7.15 μ g l⁻¹, respectively) were above their respective seawater solubility limits (45, 130 and 3.8 μ g l⁻¹, respectively) (Mackay et al. 1992; Callahan et al. 1979; U.S. Environmental Protection Agency 1982). Conversely, the fluoranthene EC_{50} in dark condition was lower than its seawater solubility limit (206 μ g l⁻¹, U.S. Environmental Protection Agency 1982) which indicates the toxicity of this PAH in the absence of UV light.



◀ **Fig. 1** Percentage of D-veliger larvae after 24 h exposure, in dark (a) and under sunlight (b), of *R. decussatus* fertilized eggs to different concentrations ($\mu\text{g/l}$) of anthracene, fluoranthene, pyrene and benzo[a]pyrene. Error bars represent standard deviations. ($n = 5$, * $p < 0.05$; ** $p < 0.01$; *** $p < 0.001$)

After a 96 h exposure period of *R. decussatus* veliger larvae to increasing concentrations of four PAHs in dark and under sunlight conditions, mortality percentages were recorded and LC_{50} values were calculated. Anthracene induced significant ($p < 0.05$) larval mortality at concentrations of 50 (10 %) and $5 \mu\text{g l}^{-1}$ (28 %) in dark and under sunlight conditions, respectively. The presence of UV light (sunlight) induced anthracene toxicity, with 100 % mortality occurring at $50 \mu\text{g l}^{-1}$ (Fig. 2). The calculated LC_{50} of anthracene in both conditions are given in Table 2.

In darkness, exposure of larvae to fluoranthene resulted in 18.6 ± 3.2 % of mortality (compared to control) at a concentration of $50 \mu\text{g l}^{-1}$, with an LC_{50} of $285.8 \mu\text{g l}^{-1}$. Under sunlight conditions, the lowest anthracene concentration that induced significant mortality (36.5 ± 9.7 %) was $2.5 \mu\text{g l}^{-1}$, with an LC_{50} of $8.75 \mu\text{g l}^{-1}$ (Table 2). Pyrene resulted in toxicity to clam larvae at a concentration as low as $64 \mu\text{g l}^{-1}$ in dark conditions, reducing the percentage of live veliger larvae by 15 % with respect to controls and yielding an LC_{50} value of $134.7 \mu\text{g l}^{-1}$ (666 nM). The calculated LC_{50} under sunlight was $10.7 \mu\text{g l}^{-1}$ (53 nM). Total mortality occurred at 256 and $50 \mu\text{g l}^{-1}$ of pyrene in darkness and under sunlight conditions, respectively (Fig. 2).

On the other hand, benzo[a]pyrene was toxic to larvae only at the two highest concentrations in dark conditions (25 and $50 \mu\text{g l}^{-1}$), with an LC_{50} value of $34.53 \mu\text{g l}^{-1}$ (137 nM). Toxicity of BaP was greatly increased under sunlight, and mortality exceeded 80 and 90 % at 1 and $5 \mu\text{g l}^{-1}$, respectively, with an LC_{50} value of $0.56 \mu\text{g l}^{-1}$ (2.2 nM).

Significant differences between experiments carried out in the dark and under sunlight were observed for all tested PAHs. According to LC_{50} values, the toxicity under sunlight was 32, 31, 12 and 61 times higher, respectively for anthracene, fluoranthene, pyrene and benzo[a]pyrene. In this experiment the tested PAHs, excepting fluoranthene, were considered not ecologically toxic in darkness, since calculated LC_{50} s were above their respective seawater solubility limits. Based on EC_{50} and LC_{50} values (molarities basis), the tested PAHs were ranked in order of higher toxicity as following: benzo[a]pyrene > pyrene > fluoranthene > anthracene.

The values of EC_{50} (24 h embryonic development) obtained under sunlight (with UV) were about 72, 35, 60 and 23 times higher than in the dark condition, respectively for anthracene, fluoranthene, pyrene and benzo[a]pyrene. For larvae, toxicity increased 32, 31, 12 and 61 times, respectively, for these same PAHs with sunlight exposure. In both bioassays, results indicated that the toxicity of the four PAHs to the clam *R. decussatus* was enhanced in the presence of UV light.

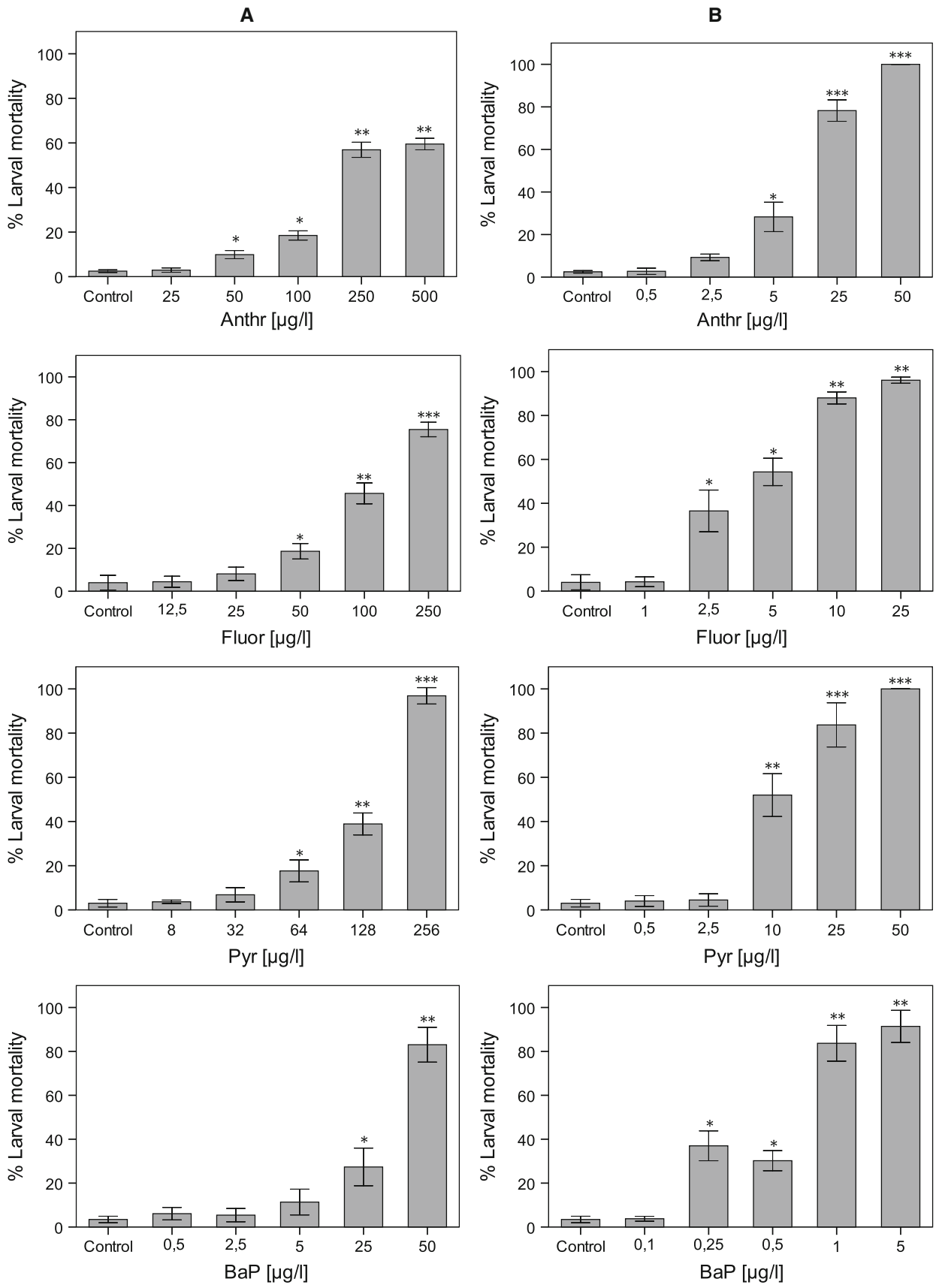
Toxicity enhancement seems to be time-dependent. In the present study, anthracene seems to be the most sensitive of the four PAHs after 24 h of exposure. Contrarily, when exposure time was 96 h, benzo[a]pyrene showed the higher sensitivity to photo-induced enhancement. The order of toxic strength, based on EC_{50} values (nM), both in dark and under sunlight conditions, was benzo[a]pyrene > pyrene > fluoranthene > anthracene.

Comparison of our results with previous research is not straightforward since studies on the effects of PAHs to aquatic organisms carried out in dark and under sunlight conditions, as those in the natural habitats of early embryonic stages of marine invertebrates, are scarce.

A comprehensive literature search has returned only a few studies where exposure to PAHs was conducted in the

Table 2 EC_{50} , EC_{10} (embryotoxicity) and LC_{50} (larval mortality) calculated for anthracene, fluoranthene, pyrene, and benzo[a]pyrene ($\mu\text{g/l}$) in dark and under sunlight conditions and their 95 % confidence intervals (in parenthesis)

PAH	Embryotoxicity test		Larval mortality test
	EC_{50}	EC_{10}	LC_{50}
Anthracene			
Dark	361.1 (278.9–509)	27.9 (17.4–38.8)	285.82 (247.7–337)
Light	4.97 (4.43–5.58)	0.48 (0.38–0.59)	8.75 (7–10.97)
Fluoranthene			
Dark	113.8 (92.65–146.5)	15.28 (10.2–20.4)	125.63 (114.5–137.4)
Light	3.21 (2.96–3.47)	0.71 (0.59–0.82)	4 (3.17–4.81)
Pyrene			
Dark	89.8 (70.13–119.17)	11.36 (6.14–17.1)	134.7 (104.63–152.7)
Light	1.52 (1.38–1.68)	0.37 (0.3–0.43)	10.7 (8.79–12.53)
B[a]P			
Dark	7.15 (5.54–9.27)	0.56 (8.32–0.86)	34.53 (31.13–37.95)
Light	0.31 (0.28–0.33)	0.07 (0.06–0.08)	0.56 (0.26–0.95)



◀ **Fig. 2** Percentage of mortality of *R. decussatus* veliger larvae after 96 h exposure to different concentrations ($\mu\text{g/l}$) of anthracene, fluoranthene, pyrene and benzo[a]pyrene in the dark (a) and under natural sunlight conditions (b). ($n = 5$, * $p < 0.05$; ** $p < 0.01$; *** $p < 0.001$)

dark. For instance, Bellas et al. (2008) investigated the photo-induced toxicity of 6 PAHs to sea-urchin (*Paracentrotus lividus*) larvae and mussel (*Mytillus galloprovincialis*) embryos. In their mussel embryo tests, they reported EC_{50} values of >640 and 657 nM for pyrene and $>1,250$ and 263 nM for fluoranthene under dark and fluorescent light conditions, respectively. Bellas and Thor (2007) also evaluated the toxicity of fluoranthene and pyrene on survival, egg production and recruitment of the copepod *Acartia tonsa*. EC_{50} values ranged between 385 and 824 nM for fluoranthene and between 295 and 306 nM for pyrene under fluorescent light (no UV). In the present work, *R. decussatus* yielded a lower degree of toxicity in the dark (EC_{50} values of 562.7 and 444 nM for fluoranthene and pyrene, respectively). In general, the clam species that we tested seems to be more sensitive to PAHs than mussels and less sensitive than copepods. In contrast, Kagan et al. (1985) did not find toxicity of fluoranthene and pyrene to embryos and larvae of freshwater organisms (brine shrimp *Artemia salina*, water flea *Daphnia magna*, mosquito *Aedes aegypti*, leopard frog *Rana pipiens*, and fish *Pimephales promelas*) when incubations were conducted in the dark. This is in agreement with previous studies suggesting higher sensitivities of marine than freshwater invertebrates to PAHs (Hutchinson et al. 1998; Leung et al. 2001; Robinson 1999).

In the present study, only fluoranthene was ecologically toxic in the dark since EC_{50} and LC_{50} values were below its maximum aqueous solubility. On the other hand pyrene, fluoranthene and benzo[a]pyrene did not cause a 50 % decrease in the biological responses below the maximum aqueous solubility. Pelletier et al. (1997) reported 48-h EC_{50} s for embryos and 96-h LC_{50} s for juveniles of the dwarf surf clam, *Mulinia lateralis* of $4,260$ and $>13,300$ $\mu\text{g l}^{-1}$ (anthracene), 58.8 and $3,310$ $\mu\text{g l}^{-1}$ (fluoranthene) and $>9,454$ and $>11,900$ $\mu\text{g l}^{-1}$ (pyrene) respectively. These EC_{50} and LC_{50} values suggest that for both anthracene and pyrene there is no toxicity to embryonic *M. lateralis*, as the EC_{50} values are above the seawater solubility limits of these PAHs. However, fluoranthene does exhibit toxicity to developing *M. lateralis* embryos in a dissolved state. The low aqueous solubility of most PAHs is an important factor to take into account in an assessment of the risk of these compounds in the marine environment, since their bioavailability and therefore the maximum lipid concentration attained in the organisms is constrained by their aqueous solubility (Di Toro and McGrath 2000).

A growing body of evidence suggests that toxicity of some intermediate molecular weight PAHs may be enhanced

in the presence of ultraviolet (UV) light. Although photo-induced toxicity of PAHs has been demonstrated in many studies with freshwater organisms since the early 1980s (reviewed by Arfsten et al. 1996), the first report of phototoxicity to marine species is relatively recent (Pelletier et al. 1997). In that study, toxicity of anthracene, fluoranthene and pyrene to larvae and juveniles of the bivalve *M. lateralis* and the mysid *Mysidopsis bahia* was 12 to $>50,000$ times higher under UV light than under fluorescent light. Likewise, Spehar et al. (1999) attained a 21–1880 times increase in fluoranthene toxicity under UV light to several freshwater and marine species compared to incubation under fluorescent light. Lyons et al. (2002) also found that, under fluorescent light, benzo[a]pyrene and pyrene provoked 50 % impairment on the embryonic development of the pacific oyster *Crassostrea gigas* at approximately 2.5 and 100 $\mu\text{g l}^{-1}$ respectively, whilst under UV light 1 and 5 $\mu\text{g l}^{-1}$ caused a 100 % inhibition of the embryonic development. Peachey (2005) reported significantly higher toxicity of fluoranthene and pyrene to larvae of three crustaceans (*Libinia dubia*, *Menippe adina* and *Panopeus herbstii*) under UV light; however, this study used very high UV light regimes (UVA: 1455 – $4,914$ μWcm^{-2} , UVB: 196 – 581 μWcm^{-2}) in comparison with realistic UV light intensities measured in aquatic habitats (Barron et al. 2000), and therefore its ecological relevance is questionable.

The forementioned studies and previous evaluations of PAH photo-toxicity with aquatic organisms only compared toxicity under fluorescent and artificial UV light exposure, and did not consider PAH toxicity in the dark. Indeed, care must be taken regarding the interpretation of photo-enhanced toxicity of PAHs under UV light in laboratory toxicity tests, since UV light itself may cause damage to marine organisms, and particularly to the sensitive early life-stages (Browman et al. 2000; Kuhn et al. 2000).

The present work reports acute effects of PAHs on the early life-stages of a marine invertebrate of ecological and commercial relevance. Although these effects were detected at levels above the reported typical concentrations in coastal areas and therefore little risk is predicted to occur, a higher level of risk may occur in PAH-polluted estuaries. Moreover, photo-activation of PAHs at naturally relevant levels may increase the risk of those compounds in the marine environment.

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