



# Fluid mixing leads to main-stage cassiterite precipitation at the Xiling Sn polymetallic deposit, SE China: evidence from fluid inclusions and multiple stable isotopes (H–O–S)

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## Abstract

The Xiling Sn deposit in eastern Guangdong Province comprises the Fengdishan Sn and the Saozhoudi Sn–Pb–Zn ore blocks and has long been regarded as a volcanic–subvolcanic system related to Sn polymetallic mineralization. Here, we present fluid inclusion microthermometric data from different ore stages and H–O–S isotope data of hydrothermal minerals to constrain the genesis of the Xiling deposit. Fluid inclusions from stage I have  $T_h$  values from ~ 340 to 420 °C and salinities from ~ 15 to 17 wt% NaCl equivalent, while homogenization temperatures of fluid inclusions from stages II to V range from ~ 150 to 320 °C, and salinities range between ~ 1 and 6 wt% equivalent. The oxygen and hydrogen isotopic composition of quartz and cassiterite ( $\delta D_{\text{fluid}} - 65\text{‰}$ ;  $\delta^{18}O_{\text{fluid}} 3.6$  to  $6.3\text{‰}$ ) suggest that the ore-forming fluids from stage I have a distinct magmatic signature, whereas those from stage II through stage IV ( $\delta D_{\text{fluid}}$  from  $-80$  to  $-49\text{‰}$ ;  $\delta^{18}O_{\text{fluid}}$  from  $-3.7$  to  $2.5\text{‰}$ ) show characteristics of mixing between meteoric and magmatic fluids. Moreover,  $\delta^{34}S$  values for sulfides from the Fengdishan ore block have a narrow range of  $0.6$  to  $2.5\text{‰}$  with a mean close to  $0\text{‰}$ , consistent with a magmatic sulfur source. By contrast,  $\delta^{34}S$  values for ore minerals from the Saozhoudi ore block range from  $3.4$  to  $11.5\text{‰}$ , suggesting involvement of a sedimentary sulfur source. In addition, a previous geochronological study has shown that the volcanic–subvolcanic host rocks have an age of  $160$ – $170$  Ma, while the Sn polymetallic mineralization has an age of about  $145$  Ma. Our data support a model of mixing of magmatic brine from a hidden granitic intrusion with meteoric water. The S isotope data and the observed temperature gradient of the fluid system suggest that the Sn mineralization is developed in the central part of the ore system, while the Sn–Pb–Zn and Pb–Zn mineralization occurs in the distal part. This finding might have important implications for exploration in the region.

**Keywords** Fluid inclusion · H · O · S isotopes · Fluid mixing · Xiling deposit · Southeastern Coastal Metallogenic Belt

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## Introduction

Globally, most Sn deposits are genetically related to granites (Lehmann 1990), whereas some Sn ore systems are associated with volcanic or subvolcanic rocks, such as in southern Bolivia, northwestern Mexico, and southwestern United States (Christiansen et al. 1983; Tuta et al. 1988; Cunningham et al. 1991). South China is one of the important W–Sn metallogenic provinces in the world, particularly the Nanling Range (e.g., Mao et al. 2013; Yuan et al. 2008, 2011, 2018, 2019; Hu et al. 2012; Qi et al. 2012), where W–Sn mineralization is genetically related to Mesozoic granitic rocks and mainly consists of skarn-, wolframite–quartz vein-, and minor greisen-types (Mao et al. 2007). It is worth noting that there are also several Sn ± Pb ± Zn ± Ag deposits associated with the volcanic–subvolcanic rocks in the southwestern part of the

Southeastern Coastal Metallogenic Belt (SCMB) (Huang and Qi 1991), which is adjacent to the Nanling Region in the north (Fig. 1). However, recent studies have recognized that these Sn polymetallic deposits are linked with granitic intrusions, including the Feie'shan W–Sn, the Jinkeng Sn, the Changpu Sn, and the Taoxihu Sn deposits (Liu et al. 2017, 2018c; Qiu et al. 2017; Yan et al. 2017, 2018). All these studies focus on the geochronology and petrogeochemistry of the W–Sn deposits, and indicate that they are genetically related to highly fractionated I- or A-type granitic intrusions (Liu et al. 2017, 2018c; Qiu et al. 2017; Yan et al. 2017). However, the origin and magmatic–hydrothermal evolution of these deposits have attracted much less attention, and little systematic studies on fluid dynamics exist.

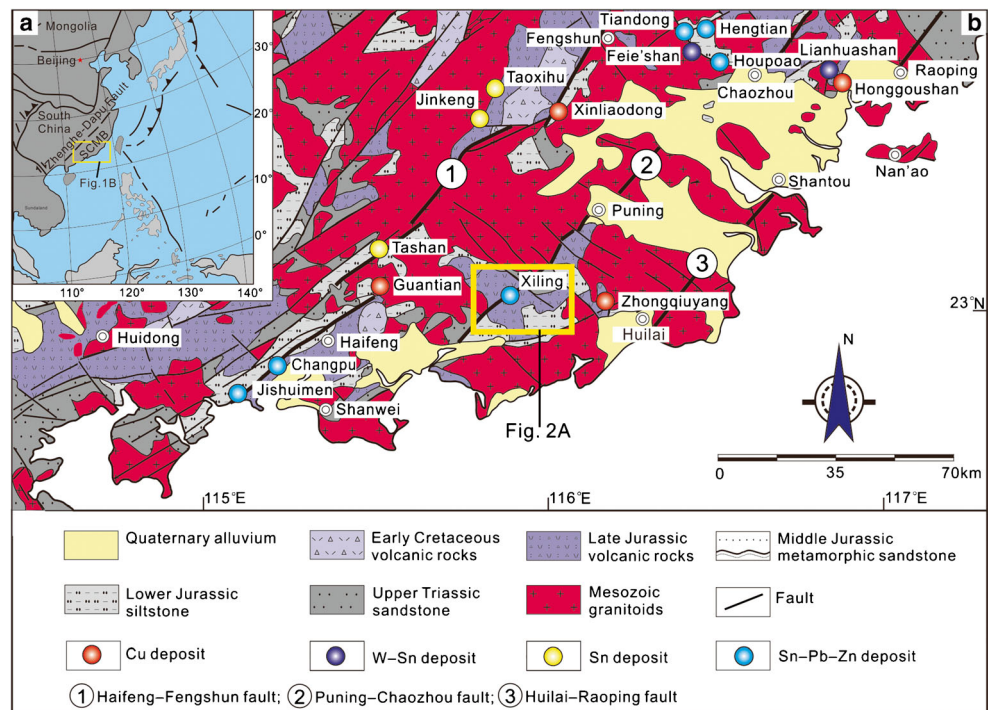
The Xiling Sn polymetallic deposit is located in eastern Guangdong Province, within the southwestern domain of the SCMB, and comprises the Fengdishan and the Saozhoudi ore blocks. The deposit has long been regarded as volcanic or subvolcanic rock-related Sn mineralization (Chen et al. 1986; Huang and Qi 1991). However, our recent geochronological data yielded a muscovite  $^{40}\text{Ar}$ – $^{39}\text{Ar}$  plateau age of  $140.6 \pm 1.0$  Ma and two cassiterite LA–ICP–MS U–Pb ages of  $146.4 \pm 1.0$  Ma and  $147.5 \pm 1.1$  Ma, and a zircon U–Pb age range from  $161.5 \pm 2.6$  to  $169.5 \pm 0.5$  Ma for volcanic–subvolcanic rocks at the Xiling deposit (Liu et al. 2018b). The ages of the volcanic–subvolcanic rocks predate the Xiling mineralization by ca. 15 to 20 m.y. There are no granitic rocks

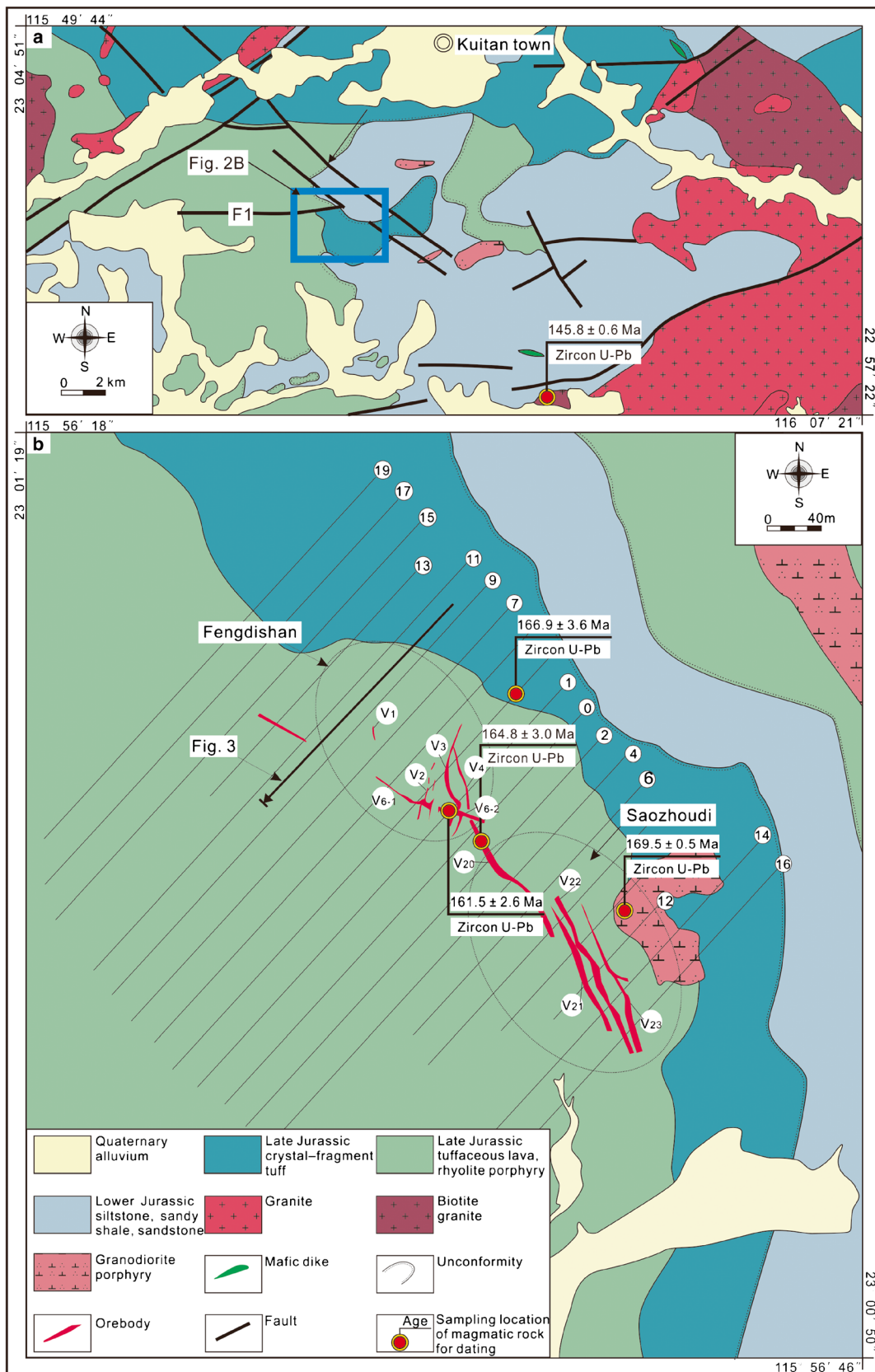
exposed in the mine area, but there is an outcrop of biotite granite about 12 km SE on which we obtained a zircon U–Pb age of  $145.8 \pm 0.6$  Ma (Fig. 2a) (Liu et al. 2018b), suggesting a new genetic model for the Xiling mineral system.

## Regional geological background

The South China Block is composed of two sub-blocks, the Yangtze Craton and the Cathaysia Block, which are separated by the Jiangshan–Shaoxing and Pingxiang–Yushan suture (e.g., Charvet et al. 1996; Zhao and Cawood 1999). The SCMB, located at the southeastern part of the Cathaysia Block, east to the Zhenghe–Dapu fault, covers the eastern Zhejiang, southeastern Fujian, and eastern Guangdong Provinces (Liu et al. 2018b) (Fig. 1a). The strata exposed include pre-Devonian metamorphic, Late Paleozoic–Middle Triassic clastic sedimentary and carbonate rocks, and Mesozoic–Cenozoic continental clastic and volcanic rocks (Shu et al. 2009). A NE–SW striking trans-lithospheric fault (Changle–Nan'ao fault) separates the SCMB into two tectonic belts from west to east: the volcanic–intrusive belt and the Pingtan–Dongshan metamorphic belt. The intrusive rocks in the region are mostly composed of high-K calc-alkaline I- and A-type granites, as well as minor syenite, quartz diorite, and gabbro (Xu et al. 2007). Previous studies have shown that three episodes of mineralization can be recognized in the SCMB:

**Fig. 1** a. Tectonic scheme of eastern Asia (after Gilder and Courtillot 1997). b. Geological map of the southwestern part of the Southeastern Coastal Metallogenic Belt (SCMB), showing the distribution of Mesozoic granitoids and polymetallic deposits (after Liu et al. 2017)





**Fig. 2** a. Geological map of the Kuitan district showing the location of the Xiling deposit. b. Geological map of the Xiling deposit (Liu et al. 2018b). Ore bodies are projected to the surface

(1) the 170–160 Ma porphyry Cu–Au and Cu–Mo mineral systems distributed mainly in the southwestern part of the SCMB, such as in the eastern Guangdong and the coastal Fujian Provinces, (2) the 145–135 Ma W–Sn mineralization and related highly fractionated I- or A-type granites occurring mostly in eastern Guangdong Province, and (3) the 120–80 Ma porphyry–epithermal Cu–Au–Ag and porphyry Cu–Mo deposits in the coastal Fujian and Zhejiang Provinces (Liu et al. 2018c).

In eastern Guangdong along the southwestern domain of the SCMB, the major stratigraphic units include Upper Triassic and Early Jurassic clastic sedimentary rocks, Late Jurassic to Early Cretaceous volcanic rocks, and Quaternary alluvial sediments (Fig. 1b). The tectonic framework is mainly controlled by the NE–SW–striking Lianhuashan fault zone, which is composed of three main faults: Haifeng–Fengshun, Puning–Chaozhou, and Huilai–Raoping fault (Xu et al. 2000) (Fig. 1b). Recent studies have recognized a phase of W–Sn polymetallic mineralization linked with granitic intrusions in the region that formed between 145 and 135 Ma (Liu et al., 2017, 2018a, b; Qiu et al. 2017; Yan et al. 2017). Liu et al. (2018c) described three prominent characteristics of the granites associated with the W–Sn mineralization in the SCMB: (1) high–K calc–alkaline, weakly peraluminous and I- or A-type affinity, (2) high degree of fractionation characterized by high Si, K, F, Na, Rb, Cs, Th, and U, and low Mg, Fe, Ca, Ba, Sr, Ti, and P, with high Rb/Sr and Rb/Ba ratios, and (3)  $\varepsilon_{\text{Hf}}(t)$  values plotting above the evolutionary trend defined by the Cathaysia basement rocks, indicating a magma source from remelting of Cathaysia basement rocks with contributions of mantle-derived components.

## Geology of the Xiling Sn deposit

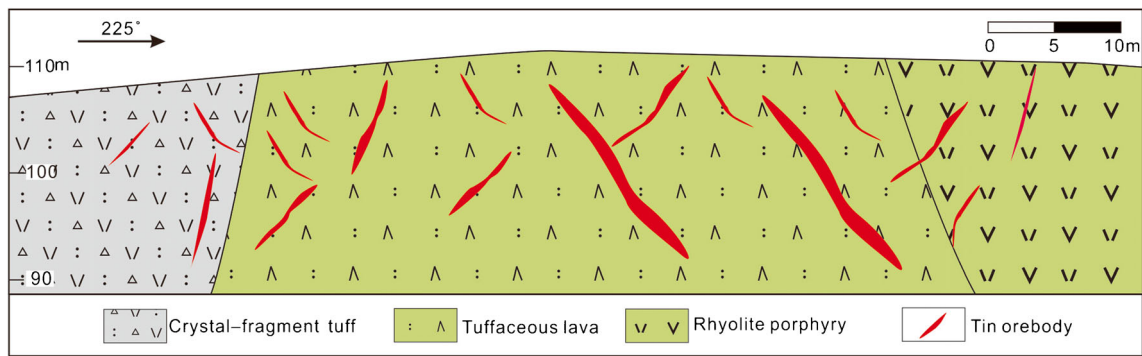
The Xiling deposit, located about 7 km southwest of Kuitan town in Huilai County in eastern Guangdong Province (Fig. 2a), was discovered in the 1940s, with underground mining in progress since the 1950s. The proven metal reserves include 15,000 t Sn with an ore grade of 2.41% Sn, and minor Pb and Zn (Huang and Qi 1991). The strata exposed in the area consist of the Lower Jurassic Jinji Formation and the Upper Jurassic Gaojiping Formation (Fig. 2b). The Jinji Formation occurs in the eastern part of the mine area and includes siltstone, sandstone, and sandy shale. The Gaojiping Formation in the central and western parts of the mine area is composed of crystal–fragment tuff (ESM Fig. 1A, B), tuffaceous lava (ESM Fig. 1C, D), and rhyolite porphyry (ESM Fig. 1). The exposed intrusive rocks are

mainly composed of biotite granite, granite, and granodiorite porphyry (Liu et al. 2018b).

The Xiling Sn mineralization is predominantly hosted in the rhyolite porphyry, tuffaceous lava, and crystal–fragment tuff (Fig. 3), and the Pb–Zn mineralization is locally developed within sandy shale. Ten orebodies have been explored in the Xiling deposit and range from 110 to –10 m in altitude. The Fengdishan ore block consists of six Sn orebodies. Orebodies V<sub>3</sub> and V<sub>4</sub> are the major Sn orebodies, with an average Sn grade of 2.73% and 0.58%, respectively. Orebody V<sub>3</sub> varies in thickness from 0.95 to 4.50 m (average 2.17 m), with a strike length of about 180 m. Orebody V<sub>4</sub> varies in thickness from 0.20 to 2.05 m (average 0.48 m) and continues along strike for about 135 m, with pinch and swell domains and branches. The Saozhoudi ore block consists of four Sn–Pb–Zn orebodies with V<sub>22</sub> and V<sub>20</sub> as the major Sn–Pb–Zn ones. V<sub>20</sub> is the largest Sn–Pb–Zn orebody ranging from 90 to –10 m in altitude, and varies in thickness from 1.5 to 8.6 m (average 5.69 m), with Sn grade of 2.99%, and a strike length of about 115 m. Orebody V<sub>22</sub> ranges from 70 to –10 m in altitude and continues along strike for about 108 m, with an average thickness of 4.95 m and an average Sn grade of 1.36%.

## Types and characters of ores

There are four ore types recognized in the Xiling deposit: (1) veins of cassiterite–feldspar–quartz, (2) disseminated ores, (3) veins of cassiterite–quartz (muscovite), and (4) veins of cassiterite–sulfide. The major orebodies occur as veins of cassiterite–feldspar–quartz, cassiterite–quartz (muscovite), and/or cassiterite–sulfide, whereas the disseminated mineralization is minor. Type 1 mineralization occurs in the central part of the Fengdishan ore block. Here, the orebodies are mostly hosted by rhyolite porphyry. The ore mineral is cassiterite, while the gangue minerals are albite and K–feldspar, with minor quartz and calcite (Figs. 4a and 5a). Type 2 mineralization is developed in the lower part of the Fengdishan ore block, and hosted by rhyolite porphyry. Ore minerals are mainly composed of cassiterite and minor arsenopyrite, while the gangue minerals are quartz and minor muscovite (Figs. 4b and 5b). Type 3 mineralization is hosted by the upper part of the Fengdishan ore block. Here, the orebodies are mostly hosted by tuffaceous lava, and more locally rhyolite porphyry and crystal–fragment tuff. Ore minerals comprise of cassiterite, and minor chalcopyrite and pyrrhotite, while the gangue minerals are dominated by quartz + muscovite ± sericite ± chlorite. Cassiterite, chalcopyrite, and pyrrhotite are euhedral to subhedral (2–10 mm) in the quartz veins, while muscovite, and minor sericite and chlorite occur as anhedral fine grains in



**Fig. 3** Geological cross section between the Fengdishan exploration lines No. 9 and No. 11 (Liu et al. 2018b)

the quartz veins (Figs. 4c, d and 5c). Type 4 mineralization is predominantly developed in the tuffaceous lava and altered sandy shale of the Saozhoudi ore block with minor occurrences in the upper part of the Fengdishan ore block. Ore minerals include galena, sphalerite, pyrite, and minor cassiterite and pyrrhotite, and the gangue minerals are quartz, muscovite, sericite, chlorite, and calcite (Figs. 4e, f and 5d, e).

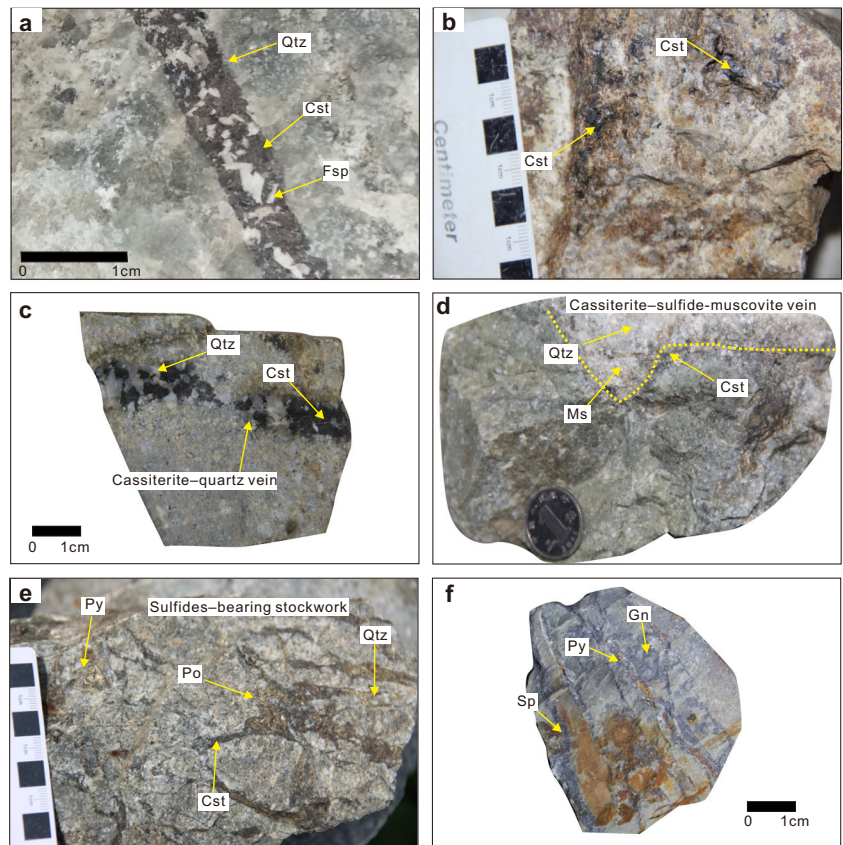
**Hydrothermal alteration and mineralization**

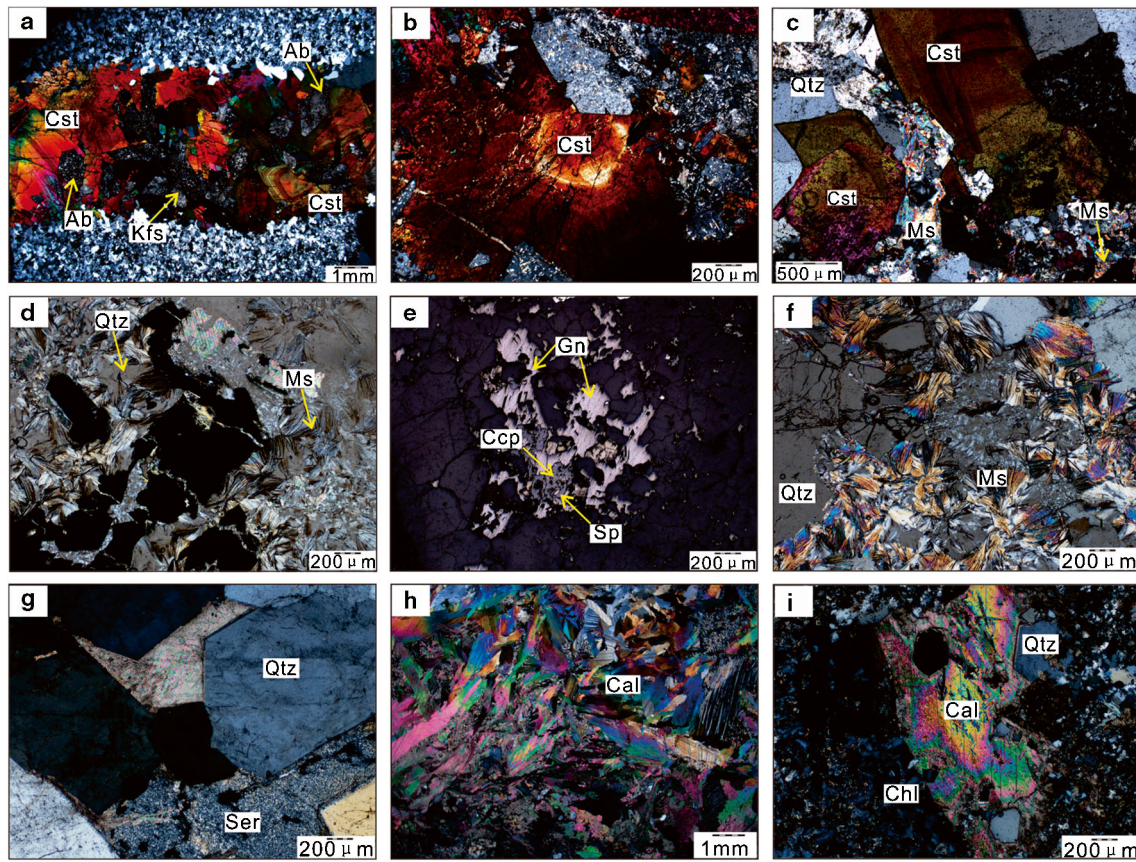
Ore-related hydrothermal alteration at the Xiling deposit is weak and includes silicification, muscovitization,

sericitization, chloritization, carbonatization, and epidotization. Silicification is developed and is represented as disseminations in all four ore types. Muscovite occurs as disseminated and on veinlets in the wall-rock, or as aggregates in cassiterite-quartz-muscovite veins, showing a close association with the Sn mineralization (Fig. 5f). Chlorite and sericite predominantly occur as clusters in sulfide-quartz-chlorite veins, and are spatially related to Pb and Zn mineralization (Fig. 5g-i).

Based on field and petrographic observations, five hydrothermal stages were identified. From early to late, these are cassiterite-feldspar-quartz (stage I), cassiterite-quartz-muscovite (stage II), sulfide-cassiterite (stage III), quartz-

**Fig. 4** Photographs of the ores of the Xiling deposit. **a** Cassiterite-feldspar-quartz vein ore. **b** Disseminated ore. **c** Cassiterite-quartz vein ore. **d** Cassiterite-sulfide-muscovite-vein ore. **e** Sulfides-bearing stockwork ore. **f** Sphalerite-galena-vein ore. Chl chlorite, Cst cassiterite, Gn galena, Ms muscovite, Po pyrrhotite, Py pyrite, Qtz quartz, Sp sphalerite





**Fig. 5** Photomicrographs of the ores and hydrothermal alteration of the Xiling deposit (under crossed polarized light). **a** Cassiterite–feldspar–quartz vein ore. **b** Disseminated ore. **c** Cassiterite–muscovite–quartz vein ore. **d** Cassiterite–sulfide–muscovite–vein ore. **e** Cassiterite–sulfide–vein ore. **f** Silicification and muscovitization. **g** Sericitization

and carbonatization. **h** Carbonatization. **i** Chloritization and carbonatization. Ab Albite, Cal calcite, Ccp chalcopyrite, Chl chlorite, Cst cassiterite, Gn galena, Ms muscovite, Kfs K–feldspar, Qtz quartz, Ser sericite, Sp sphalerite

sulfide–chlorite (stage IV), and calcite–chlorite–quartz (stage V) (Fig. 6). Stage I is characterized by cassiterite, albite, K–feldspar, and minor quartz and calcite forming veins. Stage II is the major Sn deposition stage. Ore minerals include

cassiterite, and minor arsenopyrite, pyrrhotite, and chalcopyrite. Stage III is the major Pb and Zn deposition stage with minor Sn and is characterized by cassiterite, pyrrhotite, pyrite, galena, sphalerite, and minor chalcopyrite forming veins,

**Fig. 6** Paragenetic sequence of mineralization and alteration of the Xiling deposit (Liu et al. 2018b)

Ore-forming stages	Stage I	Stage II	Stage III	Stage IV	Stage V
	Cassiterite–feldspar–quartz	Cassiterite–quartz–muscovite	Sulfide–cassiterite	Quartz–sulfide–chlorite	Calcite–chlorite–quartz
Cassiterite	■■■■■	■■■■■	■■■	■■	
Arsenopyrite		■■	■■		
Galena			■■■■■	■■■■■	
Sphalerite			■■■■■	■■■■■	
Chalcopyrite		■	■■		
Pyrrhotite		■	■■	■	
Pyrite			■■■	■■■	■
Quartz	■■■	■■■■■	■■■	■■■	■■■
Muscovite		■■■■■	■■■	■	
Sericite		■■	■■	■■■	■
Feldspar	■■■■■				
Chlorite		■	■■	■■■	■■■
Epidote					■■
Calcite	■■				■■■

Note: ■■■■■—dominant; ■■■■■—major; ■■■■■—intermediate; ■■■—minor; ■—very minor; blank—absent

veinlets, or grained in the muscovite–quartz veins. Stage IV is the major Pb and Zn deposition stage and is characterized by galena, sphalerite, and pyrite forming veins or veinlets. Finally, stage V is the latest hydrothermal and mostly barren stage characterized by the deposition of calcite, chlorite, and quartz.

## Samples and analytical methods

### Fluid inclusions

Cassiterite, quartz, and calcite samples for fluid inclusion microanalyses in this study span all ore-forming stages. More than 30 polished thick sections were examined under the microscope, of which 10 polished thin sections were selected for microthermometric measurements. Microthermometric measurements were carried out at the MNR Key Laboratory of Metallogeny and Mineral Assessment, Institute of Mineral Resources, Chinese Academy of Geological Sciences. Fluid inclusion microthermometric data were obtained using a Linkam THMSG–600 heating–freezing stage, coupled to a Leica microscope. This stage enables measurements to be taken within the range –196 to 600 °C. Freezing and heating runs were undertaken using liquid nitrogen and a thermal resistor, respectively. The estimated accuracy of temperatures is  $\pm 0.5$  °C for temperatures below 100 °C,  $\pm 1.0$  °C for temperatures in the range between 100 and 600 °C. Heating/cooling rates were restricted to  $< 10$  °C/min, and were reduced to 1 to 0.1 °C/min near phase transformations.

### Oxygen and hydrogen isotopes

Hydrogen and oxygen isotope analyses for quartz and cassiterite samples covering the main ore-forming stages were done using a MAT253EM mass spectrometer at the Isotopic Laboratory of the Institute of Mineral Resources, Chinese Academy of Geological Sciences. Prior to analysis, samples were separated using standard magnetic separation and handpicking techniques before extracting water from 40 to 80 mg of mineral separate. Oxygen isotope analyses for quartz and cassiterite were performed using the  $\text{BrF}_5$  technique of Clayton and Mayede (1963). For hydrogen isotope analyses of fluid inclusion water, quartz samples were outgassed at 105 °C for 12 h, and then fluid inclusions were decrepitated by heating to 500 to 600 °C. Water released by the fluid inclusions was converted to hydrogen gas using the zinc reduction method of Coleman et al. (1982). All  $\delta^{18}\text{O}$  and  $\delta\text{D}$  values reported here are in ‰ relative to Standard Mean Ocean Water (SMOW). The analytical uncertainty is  $\pm 0.2$ ‰ for oxygen and  $\pm 2$ ‰ for hydrogen.

### Sulfur isotopes

In situ LA–MC–ICP–MS sulfur isotope analyses were conducted on sulfide minerals from five polished thin sections covering different mineralization stages. Laser sampling was performed by a RESOLUTION–M 50 excimer ArF laser ablation system, while sulfur isotope ratios were analyzed with a Nu Plasma 1700 MC–ICP–MS at the State Key Laboratory of Continental Dynamics, Northwest University in Xi'an, China. All analyses were made using a 30- $\mu\text{m}$  laser beam with a repetition rate of 2–3 Hz and an ablation time of 50 s. Detailed analytical conditions and procedures can be found in Chen et al. (2017).

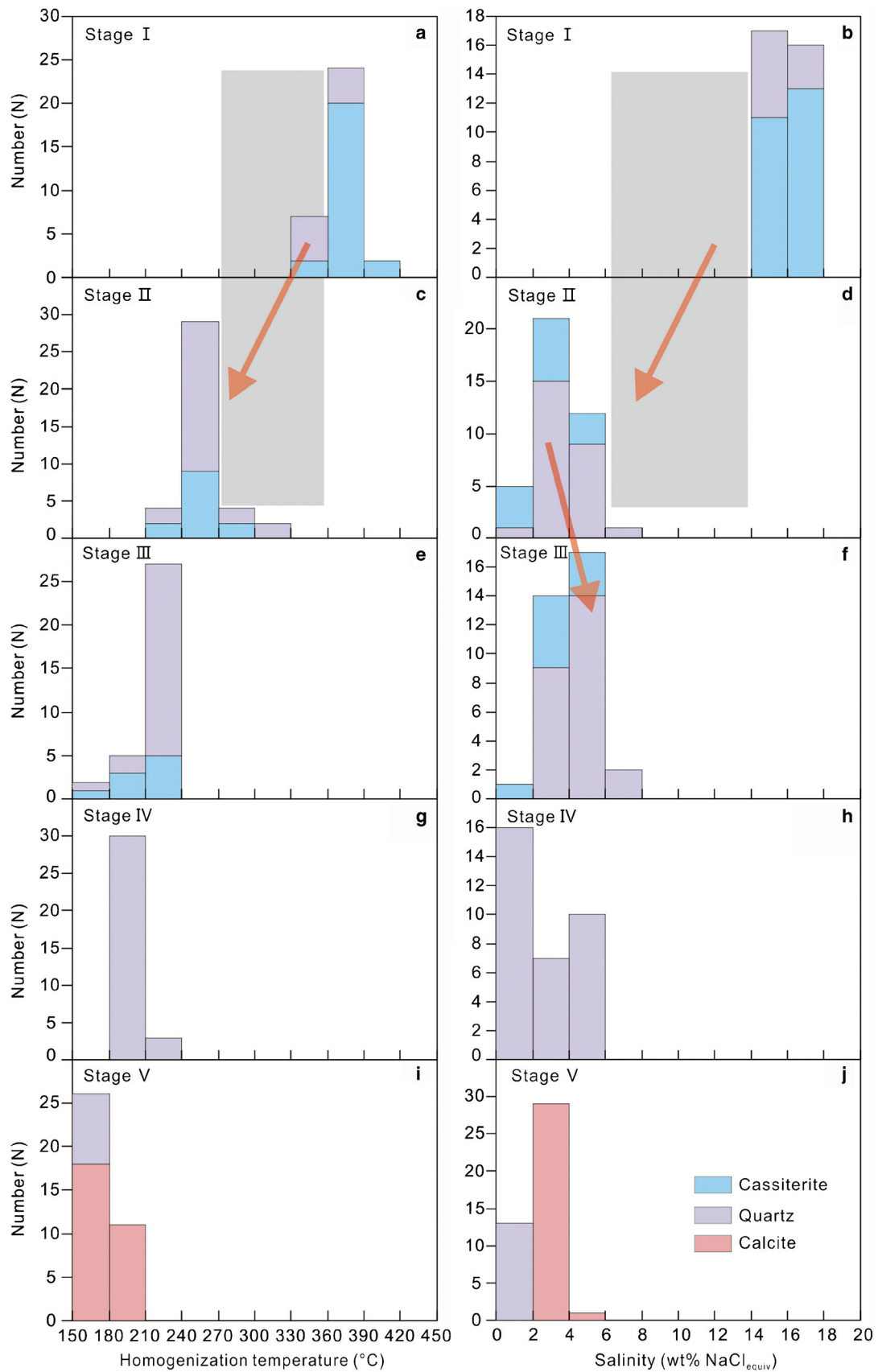
## Results

### Fluid inclusions

Representative samples of cassiterite, quartz, and calcite were selected for fluid inclusion microthermometric studies. Microscopic examination of the samples show that the type of fluid inclusions present in the samples is relatively simple, i.e., two-phase aqueous fluid inclusions. Microthermometric measurements of fluid inclusions were done if the fluid inclusion assemblages (FIA) are trapped along growth bands (Goldstein and Reynolds 1994). Features and microthermometric results for fluid inclusions from the different ore-forming stages are shown in Supplementary Table A1 and described below.

**Cassiterite–feldspar–quartz (stage I)** Fluid inclusions from stage I mineralization have been studied in cassiterite and quartz; they typically comprise a liquid  $\text{H}_2\text{O}$  phase and a vapor  $\text{H}_2\text{O}$  phase at room temperature (ESM Fig. 2A, B). Fluid inclusions in cassiterite vary from 4 to 10  $\mu\text{m}$  in size and show rounded or irregular shapes, with their vapor bubbles occupying less than 30 vol%. Their ice-melting temperatures ( $T_{\text{m-ice}}$ ) range from –11.0 to –12.8 °C, corresponding to salinities of 15.0–16.7 wt% NaCl equivalent, and homogenize into the liquid phase between 352 and 417 °C, with a peak between 350 and 380 °C. Fluid inclusions in quartz are 5 to 15  $\mu\text{m}$  in diameter and vary from rounded to highly irregular in shape, with vapor bubbles occupying 20 to 30 vol%. Their ice-melting temperatures ( $T_{\text{m-ice}}$ ) range from –10.9 to –12.8 °C, yielding salinities of 14.9–16.7 wt% NaCl equiv, while homogenization temperatures range from 336 to 373 °C (Fig. 7a, b).

**Cassiterite–quartz–muscovite (stage II)** Fluid inclusions from this stage have been measured in cassiterite and quartz, and they typically consist of liquid and vapor at room temperature (ESM Fig. 2C). Fluid inclusions in



**Fig. 7** Frequency histograms of microthermometric data for fluid inclusion from each ore-forming stage of the Xiling deposit



quartz show rounded or irregular shapes with their size varying from 4 to 16  $\mu\text{m}$ , and their vapor bubbles occupying < 40 vol%. The fluid inclusions have ice-melting temperatures ( $T_{\text{m-ice}}$ ) ranging from  $-0.9$  to  $-3.8$   $^{\circ}\text{C}$ , equivalent to salinities of 1.6–6.2 wt% NaCl equivalent, respectively, and homogenize between 231 and 312  $^{\circ}\text{C}$ , with a peak between 250 and 280  $^{\circ}\text{C}$ . Fluid inclusions in cassiterite are 3 to 10  $\mu\text{m}$  in diameter, and their vapor bubbles occupy less than 30 vol%. The ice-melting temperatures ( $T_{\text{m-ice}}$ ) range from  $-0.7$  to  $-2.8$   $^{\circ}\text{C}$ , corresponding to salinities of 1.2–4.6 wt% NaCl equivalent, respectively, with total homogenization temperatures varying from 227 to 289  $^{\circ}\text{C}$  (Fig. 7c, d).

**Sulfide–cassiterite (stage III)** Fluid inclusions in cassiterite and quartz from stage III typically consist of a liquid  $\text{H}_2\text{O}$  phase and a vapor  $\text{H}_2\text{O}$  phase at room temperature (ESM Fig. 2D). Fluid inclusions in cassiterite have sizes varying from 3 to 9  $\mu\text{m}$ , show rounded or irregular shapes, and their vapor bubbles occupy < 40 vol%. They have ice-melting temperatures ( $T_{\text{m-ice}}$ ) ranging between  $-0.7$  and  $-3.3$   $^{\circ}\text{C}$ , equivalent to salinities of 1.2–5.4 wt% NaCl equivalent, respectively, and homogenize between 170 and 227  $^{\circ}\text{C}$ . Fluid inclusions in quartz are 5 to 18  $\mu\text{m}$  in diameter with their shapes varying from rounded to highly irregular and vapor bubbles occupy between 30 and 40 vol% of the inclusions. Their ice-melting temperatures ( $T_{\text{m-ice}}$ ) are between  $-1.9$  and  $-4.2$   $^{\circ}\text{C}$ , corresponding to salinities of 3.2–6.7 wt% NaCl equivalent, respectively. Homogenization temperatures vary from 178 to 237  $^{\circ}\text{C}$ , with a peak between 210 and 237  $^{\circ}\text{C}$  (Fig. 7e, f).

**Quartz–sulfide–chlorite (stage IV)** Fluid inclusions in quartz from this stage typically comprise liquid and vapor phases at room temperature (ESM Fig. 2E). These fluid inclusions have sizes ranging from 10 to 34  $\mu\text{m}$ , and have rounded or rectangular shapes, and vapor bubbles occupying < 30 vol%. Their ice-melting temperatures ( $T_{\text{m-ice}}$ ) range from  $-0.4$  to  $-3.2$   $^{\circ}\text{C}$ , yielding salinities of 0.7–5.3 wt% NaCl equivalent, with homogenization temperatures varying from 180 to 218  $^{\circ}\text{C}$  (Fig. 7g, h).

**Calcite–chlorite–quartz (stage V)** Fluid inclusions in calcite and quartz from stage V typically comprise liquid and vapor phases at room temperature (ESM Fig. 2F). Fluid inclusions in calcite have sizes varying from 5 to 9  $\mu\text{m}$ . Their ice-melting temperatures ( $T_{\text{m-ice}}$ ) range from  $-1.5$  to  $-2.6$   $^{\circ}\text{C}$ , corresponding to salinities of 2.6 to 4.3 wt% NaCl equivalent, with homogenization temperatures ranging from 167 to 205  $^{\circ}\text{C}$  with a peak between 160 and 190  $^{\circ}\text{C}$ . Fluid inclusions in quartz are 5 to 15  $\mu\text{m}$  in diameter and they vary in shape from rounded to highly irregular, with vapor bubbles occupying < 20 vol%. They have ice-

melting temperatures ( $T_{\text{m-ice}}$ ) ranging from  $-0.7$  to  $-1$   $^{\circ}\text{C}$ , equivalent to salinities of 1.2–1.7 wt% NaCl equivalent. Homogenization temperatures are between 156 and 168  $^{\circ}\text{C}$  (Fig. 7i, j).

## Hydrogen and oxygen isotopes

The hydrogen and oxygen isotopic compositions of selected minerals (i.e., quartz and cassiterite) from the four ore-forming STAGES I–IV are given in Supplementary Table A2. The oxygen isotopic compositions of ore-forming fluids ( $\delta^{18}\text{O}_{\text{fluid}}$ ) have been calculated using the temperatures obtained by microthermometry of fluid inclusions of each ore-forming stage (i.e., stage I, 365  $^{\circ}\text{C}$ ; stage II, 258  $^{\circ}\text{C}$ ; stage III, 216  $^{\circ}\text{C}$ ; stage IV, 195  $^{\circ}\text{C}$ ), and the oxygen isotopic fractionation equation of the quartz–water system (Fekete et al. 2017) and cassiterite–water system (Zheng 1991), respectively. In stage I, the  $\delta\text{D}_{\text{fluid}}$  value for quartz is  $-65\text{‰}$ , while  $\delta^{18}\text{O}_{\text{fluid}}$  values range from 3.6 to 6.3 $\text{‰}$  (6.3 $\text{‰}$  for quartz and 3.6 to 4.7 $\text{‰}$  for cassiterite). For stage II,  $\delta\text{D}_{\text{fluid}}$  values for quartz range from  $-80$  to  $-55\text{‰}$ , while  $\delta^{18}\text{O}_{\text{fluid}}$  values range from  $-0.5$  to 2.5 $\text{‰}$  ( $-0.5$  to 0.4 $\text{‰}$  for quartz and 2.5 $\text{‰}$  for cassiterite). In stage III,  $\delta\text{D}_{\text{fluid}}$  values for quartz range from  $-63$  to  $-49\text{‰}$ , while  $\delta^{18}\text{O}_{\text{fluid}}$  values range from  $-2.0$  to  $-1.3\text{‰}$ . Finally, for stage IV, the  $\delta\text{D}_{\text{fluid}}$  value is  $-73\text{‰}$  and the  $\delta^{18}\text{O}_{\text{fluid}}$  value is  $-3.7\text{‰}$  for quartz.

## Sulfur isotope composition of sulfides

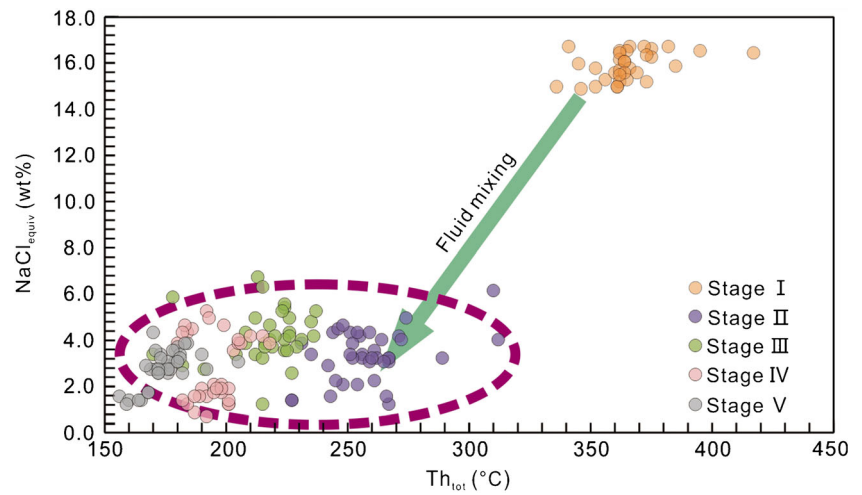
Isotopic compositions for sulfides from the Fengdishan and Saozhoudi ore blocks are given in Supplementary Table A3.  $\delta^{34}\text{S}$  values at Fengdishan range from 0.6 to 2.5 $\text{‰}$  and include values for chalcopyrite (0.6 to 2.1 $\text{‰}$ ;  $n = 9$ , average = 1.4 $\text{‰}$ ), arsenopyrite (1.4 to 2.1 $\text{‰}$ ;  $n = 5$ , average = 1.6 $\text{‰}$ ), and pyrite (1.3 to 2.5 $\text{‰}$ ;  $n = 2$ , average = 1.4 $\text{‰}$ ). At the Saozhoudi ore block,  $\delta^{34}\text{S}$  values range from 3.4 to 11.5 $\text{‰}$  and were obtained from pyrite (4.9 to 11.5 $\text{‰}$ ;  $n = 10$ , average = 7.3 $\text{‰}$ ) and galena (3.4 to 4.0 $\text{‰}$ ;  $n = 5$ , average = 3.6 $\text{‰}$ ).

## Discussion

### Fluid evolution

The fluids from stages I and II represent the main stage for Sn deposition, while those from stages III and IV represent the Sn–Pb–Zn and Pb–Zn deposition stages. Fluid inclusions in stage I have the highest homogenization temperatures between 336 and 417  $^{\circ}\text{C}$  and the highest salinities of 14.9 to 16.7 wt% NaCl equiv. By contrast, homogenization temperatures and salinities of fluid inclusions from stage II are

**Fig. 8** Plot of salinity versus homogenization temperature for fluid inclusions from different mineral assemblage stages of the Xiling deposit

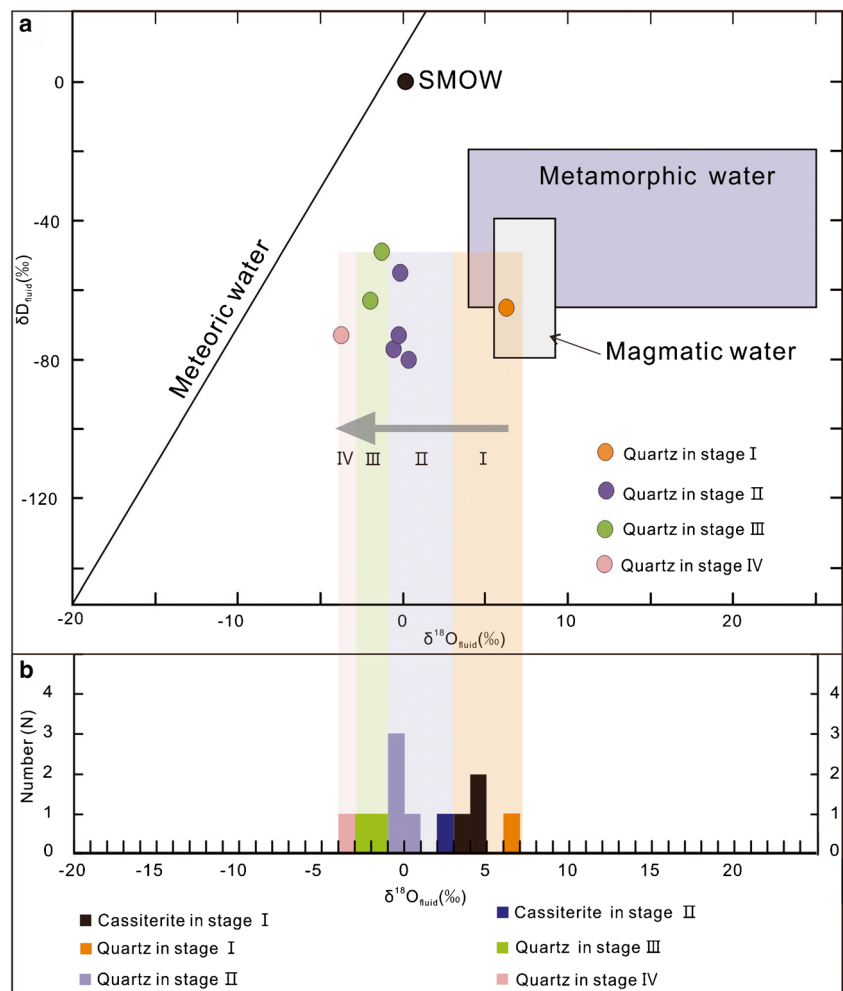


notably less, with homogenization temperatures between 227 to 312 °C and salinities of 1.2 to 6.2 wt% (Fig. 7c, d). In addition, there is a trend of progressive decrease in homogenization temperatures of fluid inclusions from stages II to V (i.e., the major Sn, Sn–Pb–Zn, and Pb–Zn deposition stages)

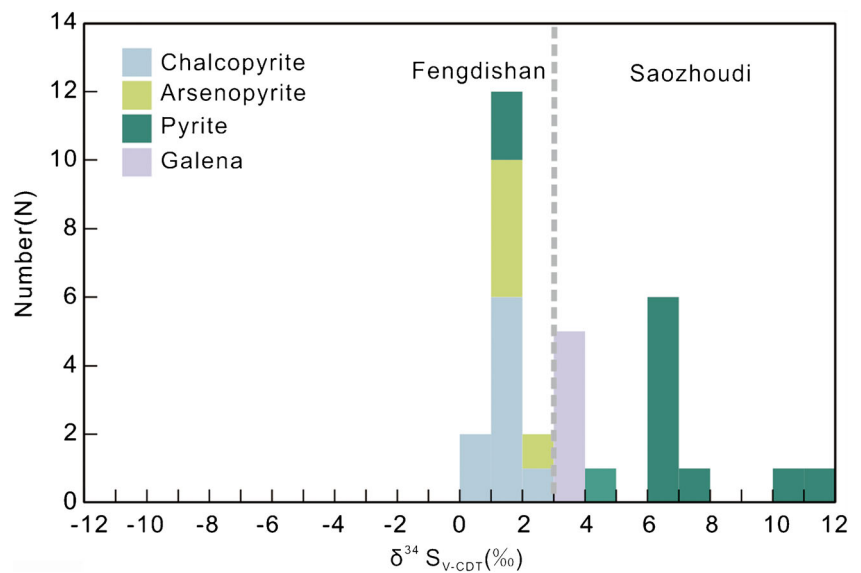
(Figs. 7 and 8). By contrast, salinities for stages II to V cover the same general range, from ~6 wt% NaCl equiv. to ~1 wt% NaCl equiv. (Fig. 8).

Oxygen and hydrogen isotopes show that the cassiterite–feldspar–quartz assemblage (stage I) has  $\delta^{18}\text{O}_{\text{fluid}}$

**Fig. 9** a. Plot of  $\delta^{18}\text{O}$  versus  $\delta\text{D}$ , showing the calculated composition for the ore-forming fluids of the Xiling deposit. The metamorphic water field, primary magmatic water field, and meteoric water line are from Taylor (1974). SMOW Standard Mean Ocean Water. b.  $\delta^{18}\text{O}_{\text{fluid}}$  isotopic composition of cassiterite and quartz of the Xiling deposit



**Fig. 10** Histogram of  $\delta^{34}\text{S}$  values for sulfide minerals from the Xiling deposit



values of 1.8 to 6.3‰ and  $\delta D_{\text{fluid}}$  values of  $-65\text{‰}$ , consistent with magmatic–hydrothermal fluids. By contrast,  $\delta^{18}\text{O}_{\text{fluid}}$  and  $\delta D_{\text{fluid}}$  values for the cassiterite–quartz–muscovite assemblage (stage II) range from  $-0.5$  to  $2.5\text{‰}$  and  $-80$  to  $-55\text{‰}$ , respectively, suggesting that meteoric water mixed with the initial magmatic brines (Fig. 9a). Moreover, there is a clear trend that increasing amounts of meteoric water mixed with the magmatic brines from stages II through IV (Fig. 9a). Fluid inclusion salinities are relatively uniform between stages II and V. Fluid inclusion salinities increase slightly between stages II and III, suggesting that multiple pulses of brine with high salinities mixed with meteoric water during ore formation.

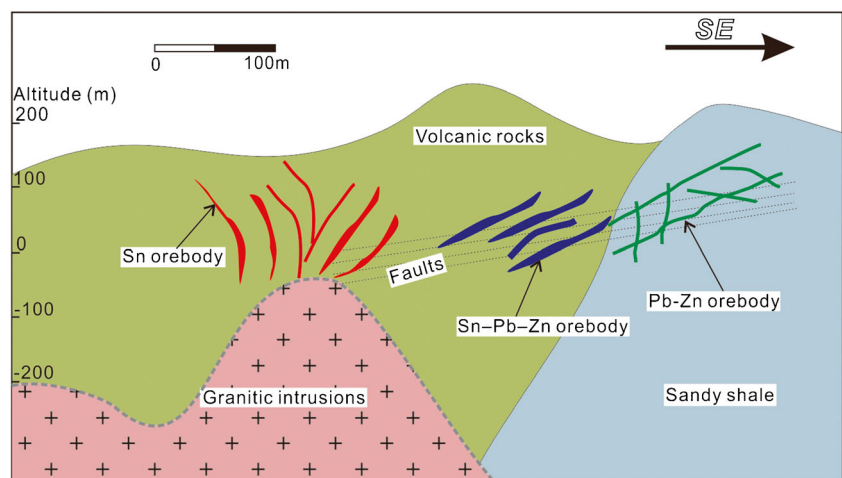
Recent geochronological data on the Xiling deposit are consistent with our interpretation. The two cassiterite samples used for U–Pb dating were taken from the disseminated and cassiterite–quartz vein ores, while the Ar–Ar muscovite sample was taken from the cassiterite–sulfide–vein ore, which is predominantly composed of galena,

sphalerite, pyrite, and minor cassiterite. In combination with the different paragenetic minerals and the apparent age gap of at least 4 myrs between the cassiterite U–Pb and muscovite  $^{40}\text{Ar}$ – $^{39}\text{Ar}$  plateau ages, all the observations above show that the Xiling mineralization is likely related to multiple magmatic brine fluid pulses. In summary, the ore-forming fluids from stage I would appear to have a distinctly magmatic signature, whereas those from the major Sn (stage II), Sn–Pb–Zn (stage III), and Pb–Zn (stage IV) depositional stages show evidence of mixing between meteoric and magmatic–hydrothermal fluids, with possibly multiple magmatic fluid pulses involved (Fig. 9b).

**Ore precipitation**

Several mechanisms for Sn ore precipitation have been proposed, including fluid–rock interaction (Korges et al. 2017), mixing of hot magmatic brine with cooler meteoric water (Witt 1988; Heinrich 1990; Audétat et al. 1998),

**Fig. 11** A proposed model explaining the formation of the Xiling Sn polymetallic deposit (modified after Liu et al. 2018b)



and phase separation (Drummond and Ohmoto 1985; Polyá 1989). At the Xiling Sn deposit, there is no indication of co-existence of low-salinity vapor fluid inclusions with high-salinity fluid inclusions, indicating that phase separation is not responsible for ore deposition. Moreover, fluid inclusion homogenization temperatures and salinities do not show a progressive decrease but a rapid decrease during the major Sn depositional stage (i.e., stage I and stage II), inconsistent with fluid–rock interactions, which generally produce widely a strong alteration, whereas the Xiling deposit has weak wall–rock alteration (Liu et al., 2018b). For the case of Xiling, a dramatic decrease in homogenization temperatures and salinities of fluid inclusions is consistent with mixing between hot magmatic brine and cool meteoric water. Additionally, hydrogen and oxygen isotopic data also show that there is a progressive ingress of meteoric water involved in the ore-forming fluids from the early to late ore-forming stages.

Tin is transported as chloride complexes in hydrothermal fluids, either as  $\text{Sn}^{2+}$  (Eugster 1986; Wilson and Eugster 1990) or  $\text{Sn}^{4+}$  (Schmidt 2018). Hence,  $\text{SnO}_2$  can be precipitated through a decrease in the  $\text{Cl}^-$  concentration of the hydrothermal system. The drastic change in salinity from 16 to 2–4 wt% NaCl equiv., combined with a temperature drop from about 350 to about 250 °C, is best explained by mixing of a hot (magmatic) brine with cooler meteoric water.

### Source of sulfur

$\delta^{34}\text{S}$  values for both the Fengdishan and Saozhoudi ore blocks relate to sulfides only and, as no sulfate minerals are present, the sulfur species in the hydrothermal fluids are probably dominated by  $\text{H}_2\text{S}$  rather than  $\text{SO}_4^{2-}$ . Hence, the sulfur isotopic compositions of sulfides are likely to represent those of the bulk sulfur in the hydrothermal system (Ohmoto and Rye 1979).  $\delta^{34}\text{S}$  values for sulfides in the Fengdishan ore block have a narrow range from 0.6 to 2.5‰, with a mean close to 0‰, consistent with a possible magmatic sulfur source (Fig. 10) (Ohmoto and Rye 1979; Chaussidon and Lorand 1990). The sulfur isotopic compositions are similar to those of granite-related Sn deposits in eastern Guangdong Province, such as the Taoxihu Sn ( $\delta^{34}\text{S}$  values for pyrite and chalcopyrite ranging between 0.1 and 2.1‰) (Yan et al. 2017) and Sanjiaowo Sn deposits ( $\delta^{34}\text{S}$  values for pyrite and pyrrothite ranging between – 1.6 and 1.0‰) (Yan et al. 2018).

By contrast,  $\delta^{34}\text{S}$  values for sulfides from the Saozhoudi ore block range from 3.4 to 11.5‰, and are distinctly different from the Fengdishan ore block (Fig.

10). Higher  $\delta^{34}\text{S}$  values can be explained by a reduction in  $f_{\text{O}_2}$ , or the incorporation of some sedimentary sulfur through the interaction with sedimentary rocks (Ohmoto 1972). Moreover, sedimentary rocks commonly have a wide range of  $\delta^{34}\text{S}$  values (Hoefs 2009). This can explain the large variation of  $\delta^{34}\text{S}$  values seen in sulfides of the Saozhoudi ore block. Finally, the Saozhoudi ore block is at the distal end of the Xiling ore system (Liu et al., 2018b). As a result, we propose that the source of sulfur at the Saozhoudi ore block is dominantly sedimentary sulfur.

### Conclusions

Fluid inclusion homogenization temperatures and salinities show a dramatic decrease between stages I and II of the Xiling Sn deposit. Homogenization temperatures for fluid inclusions from stages II to V then show a progressive decrease, whereas the salinities are relatively uniform, even increasing slightly between stages II and III. Oxygen and hydrogen isotopes show that ore-forming fluids from stage I have a distinctly magmatic signature, whereas those from stages II to IV have characteristics of mixing between meteoric water and magmatic fluids. The results obtained in this study suggest that the ore-forming mechanism of the Xiling deposit is predominantly mixing of magmatic brine from a hidden granitic intrusion with meteoric water. In combination with S isotopes, all these observations likely suggest that Sn mineralization developed in the central part of the ore system, while the Sn–Pb–Zn and Pb–Zn orebodies occur in the distal part (Fig. 11), respectively. This finding has important implications for the exploration strategy in the region.

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