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Chemical analysis and cellulose crystallinity of thermally modifed *Eucalyptus nitens* **wood from open and closed reactor systems using FTIR and X‑ray crystallography**

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Abstract

Currently there is a growing market for high quality solid wood products in Chile made from *Eucalyptus nitens*. Thermal modifcations have been used to obtain such products out of fast growing species. The chemical and crystallinity changes in the modifed wood were investigated using difuse refectance FTIR spectroscopy and crystalline analysis by X-ray difraction to analyze the diference between thermal modifcations processes using pressure under wet conditions (closed system) and processes without pressure under drier conditions (open system). In general, the FTIR spectra showed diferences in the degradation of the hemicelluloses in the peaks of the $C=O$ linkages, but almost no differences in the peaks that identify the lignin structure of the wood, as it was difcult to separate the diferent chemical reactions due to the depolymerization of lignin only observing the bands. The degree of crystallinity showed a tendency to increase at high pressure in the closed system modifications and at temperatures above 200 °C in the open system modifications, but no significant differences at low modifcation pressure and temperatures. Nonetheless, there were diferences in FTIR spectra and cellulose crystallinity when directly comparing modifcations with the same corrected mass loss under diferent conditions.

1 Introduction

There is interest in Chile to extend the use of solid wood products, especially from fast growing plantation species such as *Eucalyptus nitens* (Shining Gum). The plantation of this species has gained in importance in recent years due to its excellent growth rates, positioning it into the second fastest growing species in Chile, after *Pinus radiata*, and better cold and frost resistance than *Eucalyptus globulus* (Blue Gum), the most used hardwood plantation species (INFOR [2014](#page-7-0), [2015\)](#page-7-1). There has been an interest in extending the use to solid wood products to increase the economic returns (Muñoz et al. [2005](#page-7-2)). Thermal modifcations have

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Laboratorio de Cristalografía Aplicada, Facultad de Ingeniería, Universidad Católica de Temuco, Temuco, Chile been used to produce high quality material from plantation eucalypt species (Esteves et al. [2007](#page-7-3); de Cademartori et al. [2015;](#page-7-4) Batista et al. [2016](#page-7-5); Wentzel et al. [2019\)](#page-8-0) that can be potentially used for windows, claddings, fooring, decking or garden products (Militz and Altgen [2014\)](#page-7-6).

Thermal modifcation processes have been developed in the past decades, where the common features are temperatures between 160 and 240 °C in an atmosphere with limited oxygen content (Hill [2006](#page-7-7); Esteves and Pereira [2009;](#page-7-8) Militz and Altgen [2014](#page-7-6)). The main diferences between the processes are the kind of shielding gas used (nitrogen or steam to name a few) and whether the process is open or closed. Usually, open reactor systems work at atmospheric pressure using superheated steam, like the Thermowood[®] process (Mayes and Oksanen [2002\)](#page-7-9), which has a high temperature drying step at the beginning of the process to avoid excessive drying rates during the actual modifcation temperature, while the relative humidity (RH) of the wood reaches almost 0% at that step. As for the closed reactor systems, an example is the FirmoLin process (Willems [2009](#page-8-1)) where the pressure is generated using steam, which enables to control the RH inside the chamber thus preserving the wood moisture content during the process (Willems [2009;](#page-8-1) Altgen et al. [2016\)](#page-6-0) and preventing excessive vaporization of the

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degradation products generated during the modifcation. The diferences in the wood moisture content and the duration of the exposition to the peak modifcation temperature afect the chemical reactions and alter the temperature limit that induces the depolymerization of the cell wall constituents (Stamm and Hansen [1937;](#page-8-2) Seborg et al. [1953](#page-8-3); Borrega and Kärenlampi [2008\)](#page-7-10). The frst component to degrade are the hemicelluloses, where there is a hydrolysis of the polymer chains into oligomers and monomers, while also loosing hydrogen bonds (Tjeerdsma et al. [1998;](#page-8-4) Garrote et al. [1999](#page-7-11)). Cellulose is the least transformed wood component, as it is not as strongly afected by thermal degradation as the hemicelluloses, even if many chemical reactions are similar between them (Fengel [1967;](#page-7-12) Alen et al. [2002](#page-6-1)). However, in the amorphous region of the cellulose microfbril, hydrolytic cleavage takes place, which results in an increase in the relative crystallinity (Bhuiyan et al. [2000;](#page-7-13) Sivonen et al. [2002\)](#page-8-5). Lignin is the component that is most thermally stable, as its relative percentage after thermal modifcation usually increases (Alen et al. [2002;](#page-6-1) Esteves et al. [2008\)](#page-7-14). However, there are still chemical changes happening in the lignin during the modifcation, such as cleavage of the methoxyl groups and depolymerization of the lignin macromolecule to lower molecular weight compounds (Tjeerdsma et al. [1998](#page-8-4); Sivonen et al. [2002](#page-8-5)) as well as higher concentration of phenolic groups (Runkel [1951](#page-8-6); Kollmann and Fengel [1965](#page-7-15)). These chemical changes infuence the loss in mechanical strength of the thermally modifed wood (Tjeerdsma et al. [1998;](#page-8-4) Kubojima et al. [2000](#page-7-16)), while crystallinity can particularly infuence the brittleness (Phuong et al. [2007\)](#page-8-7) and the decrease in fexibility of the material (Kocaefe et al. [2008\)](#page-7-17).

A quick method to measure the chemical changes is the use of Fourier transform infrared spectroscopy (FTIR). The absorption observed in the FTIR spectra can be assigned to a combination of the functional groups C, H, and O that are present in celluloses, hemicelluloses, extractives, lignin and water in woody materials (Rodrigues et al. [1998\)](#page-8-8). In thermally modifed wood, FTIR has been used by Tjeerdsma and Militz ([2005](#page-8-9)) analyzing beech (*Fagus sylvatica*) and Scots pine (*Pinus sylvestris*), and in the case of species similar to *E. nitens*, Esteves et al. ([2013](#page-7-18)) used this method when analyzing thermally modifed *E. globulus* wood. They used the KBr embedding method, as described by Faix and Böttcher [\(1992\)](#page-7-19). Solid wood samples and thin slices have also been used, but to obtain reproducible spectra, the roughness and direction of the cut (radial, tangential or transversal) have to be identical (Pandey and Theagarajan [1997](#page-8-10)). To minimize the problems of the wood surface during the analysis, diffuse refectance is a technique that allows the use of samples without previous preparation to obtain the spectra, as it is an in situ measurement that gives fast results in a reliable way.

X-ray difraction has been used to measure and characterize the crystallinity of the cellulose (Segal et al. [1959](#page-8-11); Isogai and Usuda [1990](#page-7-20); Thygesen et al. [2005\)](#page-8-12). It has also been used to measure the crystallinity in thermally modifed wood (Dwianto et al. [1996;](#page-7-21) Bhuiyan et al. [2000](#page-7-13), [2001\)](#page-7-22) and even in thermally modifed eucalypt species (Cheng et al. [2017\)](#page-7-23). It has been shown that the crystallinity increases after modifcation (Bhuiyan et al. [2000](#page-7-13); Cheng et al. [2017](#page-7-23)), but it is not known if diferences in the process conditions have an infuence on the crystallinity. There also are no reports on measurements done directly in thermally modifed solid wood samples.

The aim of the present study was to use a combination of FTIR and X-ray difraction to gain deeper insight into the chemical variation and measure the degree of crystallinity to further analyze the changes that occur during the diferent thermal modifcation processes.

2 Materials and methods

2.1 Material

Wood from 19-year old *Eucalyptus nitens* plantations was obtained from the Región del Bío-Bío in Chile. Slats of $20 \times 50 \times 650$ mm³ (radial \times tangential \times longitudinal) size were prepared from kiln-dried wood, avoiding major defects and large knots. The slats had an average moisture content of 13% prior to the modifcation process. Ten slats per modifcation process were used.

2.2 Thermal modifcation

Thermal modifcation processes were performed in a laboratory-scale treatment reactor. The samples were placed in a stainless steel vessel with a volume of 65 l that could be heated up to a maximum of 260 °C with electric heating cables and cooled down to room temperature using water cooling coils. Water vapor was produced by heating an external water reservoir connected to the vessel. Exhaust valves were used for the release of excess pressure. The treatment reactor can be used as either an open (without pressure, dry conditions) or a closed (under pressure, wet conditions) system.

The open system modifications were similar to the ThermoWood® process (Mayes and Oksanen [2002\)](#page-7-9). The frst step is to rise the temperature 12 °C/h until it reaches 100 °C, followed by increasing the temperature 2 °C/h until 130 °C to emulate the high-temperature drying step of the ThermoWood® process. The third step was to increase the temperature 12 °C/h until it reaches the peak temperatures shown in Table [1.](#page-2-0) The peak temperature was then held for 3 h. Afterwards the temperature was decreased 20 °C/h until it reached 65 °C, at that point the vessel was opened and the samples were taken out of it.

Table 1 Peak temperature, maximum pressure, relative humidity (RH), fnal moisture content (MC) and corrected mass loss (CML) determined for each modifcation. The hold time at peak temperature was 3 h for all processes

Peak tempera- ture $(^{\circ}C)$	Max pressure RH $(\%)$ (MPa)		MC (%)	$CML(\%)$
Untreated				$\overline{0}$
Closed system				
150	0.14	30	4.5	2.6
160	0.18	30	3.9	5.0
170	0.23	30	3.2	6.3
150	0.47	100	5.3	11.4
160	0.61	100	4.9	18.6
170	0.77	100	4.6	20.5
Open system				
160	Atm. ^a		3.3	5.4
180	Atm. ^a		2.0	7.9
200	Atm. ^a		2.0	14.6
210	Atm. ^a		2.2	18.7
220	Atm. ^a		2.3	16.8
230	Atm. ^a		2.4	19.3

a Atmospheric pressure

As for the modifcations in the closed system, a similar schedule to the one used in the open system was used, but there was no high-temperature drying step applied, so the temperature was risen 12 °C/h until the peak temperature. It was also held for 3 h and afterwards decreased 20 °C/h up to 65 °C. Pre-vacuum (ca. 13 kPa) was applied at the beginning of each thermal modifcation process. Peak temperatures, RH and maximum pressure applied in the closed system are shown in Table [1](#page-2-0).

The fnal moisture content (MC) was measured by difference of dry weight using small pieces taken from each slat. The dry mass from the small sections and the mass loss (ML) were calculated using the formulas described by Metsä-Kortelainen et al. [\(2006\)](#page-7-24), while the corrected mass loss (CML) was measured as described by Altgen et al. ([2016](#page-6-0)), where frst the extractives are deducted from the dry weight of the selected wood slats to calculate their corrected weight.

2.3 FTIR analysis

A FTIR chemical imaging system (PerkinElmer) was used to obtain the spectra from each modification and from untreated specimens. The system consists of a spectrophotometer Frontier that has two detectors, type DTGS NIR and MIR, both covering a range between $(14,700 \text{ and } 350 \text{ cm}^{-1})$ with a resolution of 4 cm^{-1} . The imager Spotlight 400, with a detector type MCT MIR (7800–720 cm⁻¹) that has a resolution > 2 cm^{-1} , was used. The system can generate chemical spectra directly on the surface of the wood through chemical images. In this work, difuse refectance was used to obtain the spectra with a resolution of 8 cm−1 and 32 scans. The spectra were baseline corrected using an interactive baseline correction and then normalized considering maximum ordinate value in the spectrum.

The samples were cut to blocks of $10 \times 10 \times 10$ mm³ $(radial \times tangential \times longitudinal)$ and conditioned at 20 °C/65% RH for a month prior to the FTIR analysis. The radial surface was chosen for each analysis.

2.4 X‑ray difraction

The late wood from the radial surface of solid wood *Eucalyptus nitens* samples was positioned on the sample holder of a multifunctional Smartlab diffractometer (Rigaku Corporation, Japan). The equipment employs Ni-fltered Cu radiation (30 kV and 40 mA), 1 mm divergence slit, 0.3 mm anti-scatter slit, 5° Soller slits and a Rigaku D/teX 250 detector. The alignment was regularly checked against the NIST SRM660c LaB6 powder standard (NISTS [2015](#page-8-13)). Patterns were collected in the 10–30° range, counting 0.5°/s per step of 0.01°. PDXL 2 v.2.7.3.0 software was used for pattern intensity analysis. The crystallinity index (CI) was calculated according to the method by Segal et al. ([1959](#page-8-11)). Three measurements per sample were taken, as it is a proof of concept for the use of solid wood in the measurement of crystallinity, and due to the difficulty to align the late wood to the beam.

3 Results and discussion

The bands at 1748 cm^{-1} represent the ketones, which are present in free aldehyde in lignin and hemicelluloses (Rodrigues et al. [1998](#page-8-8); Michell and Higgins [2002\)](#page-7-25). The C=O linkage presents strong absorptions in FTIR spectra between 1750 and 1700 cm^{-1} , and the exact wavenumber depends on the functional group (carboxylic acid: at about 1725–1700 cm−1; ester, ketone: 1725–1705 cm−1, aldehyde: $1740-1720$ cm⁻¹) and on its structural location (Esteves et al. [2013\)](#page-7-18). As can be seen in Figs. [1,](#page-3-0) [2](#page-3-1) and [3,](#page-3-2) all thermal modifcations show a decrease in absorbance in this band as the temperature increases. The lower reactivity can be related to the decrease in free reactive hydroxyl groups in holocellulose due to the thermal degradation of the hemicelluloses, as it is more reactive than cellulose (Nguila Inari et al. [2007](#page-8-14)). There is a more noticeable decrease of this band in the modifcations at 100% RH (Fig. [2](#page-3-1)). This is mostly due to the autohydrolysis during the modifcation, which results in the formation of acetic acid that accelerates the degradation (Garrote et al. [1999](#page-7-11), [2001\)](#page-7-26). This decrease is also related to the variations in lignin due to esterifcation **Fig. 1** FTIR spectra of thermally modifed *E. nitens* under pressure at 30% RH and untreated sample. From top to bottom: Untreated, 150, 160 and 170 °C

Fig. 2 FTIR spectra of thermally modifed *E. nitens* under pressure at 100% RH and untreated sample. From top to bottom: Untreated, 150, 160 and 170 °C

Fig. 3 FTIR spectra of thermally modifed *E. nitens* at atmospheric pressure and untreated sample. From top to bottom: Untreated, 160, 180, 200, 210, 220, 230 °C

caused by the reaction of the produced acids with the wood cell wall (Tjeerdsma and Militz [2005\)](#page-8-9). Wentzel et al. [\(2019\)](#page-8-0) showed that the acetic acid concentration was higher in the modifcations at 100% RH, causing an increase in the degradation of hemicelluloses, particularly the xylans. This can be seen in the band around 1730 cm^{-1} , which is the band that represents the acetoxy groups in xylan, and in the band around 1468 cm^{-1} , which corresponds to the asymmetric deformation of C-H bond of xylan (Michell and Higgins [2002\)](#page-7-25). There is also a decrease around the peak at 1600 cm^{-1} , which is related to C=C unsaturated linkages and aromatic rings present in lignin (Rodrigues et al. [1998](#page-8-8); Michell and Higgins [2002](#page-7-25)). The changes in this band were related to lignin condensation at the expense of conjugated carbonyl groups and to the carboxylation of polysaccharides (Chow [1971;](#page-7-27) Gonzalez-Peña et al. [2009\)](#page-7-28).

Lignin in eucalypts can be seen at the band around 1505–1510 cm^{-1} (Rodrigues et al. [1998\)](#page-8-8), where the aromatic rings of hardwood lignin (Guaiacyl and Syringyl) are represented (Faix [1991\)](#page-7-29), although it can be difficult to differentiate properly its changes, as it can increase or decrease on whether there is an increase in the relative lignin (Kotilainen et al. [2000;](#page-7-30) Windeisen et al. [2007\)](#page-8-15) or the aromatic ring opening in the lignin is the one dominating (González-Peña and Hale [2009\)](#page-7-31). According to Pandey ([1999\)](#page-8-16), guaiacyl and syringyl can also be seen in the peaks around 1250 cm^{-1} , which represent the C-O stretching and bending (Rodrigues et al. [1998;](#page-8-8) Michell and Higgins [2002\)](#page-7-25). These bands shifted in the modifed wood, similar to what happened to *E. globulus* (Esteves et al. [2013\)](#page-7-18). The peaks show an increase in the band in all the thermal modifcations (Figs. [1,](#page-3-0) [2](#page-3-1) and [3](#page-3-2)), which is related to carbohydrate degradation. Something similar can be seen in the band 1385 cm⁻¹, which represents the O–H bending vibrations in phenols (Rodrigues et al. [1998\)](#page-8-8), where the OH stretching frequencies change and also increase for the thermal modifcations. This variation could be infuenced by dehydration efects of the modifcations (Moharram and Mahmoud [2008;](#page-7-32) Spiridon et al. [2011](#page-8-17)). The open system modifcations tend to have a lower MC after the process (Table [1](#page-2-0)). The bands show diferent peaks when compared directly between samples with same CML from

a drier modifcation at atmospheric pressure (210 °C) and a modifcation under pressure and 100% RH (160 °C and 0.61 bar) (Fig. [4](#page-4-0)). There are other vibrations that arise from the aromatic skeletal in lignin (around 1140 cm^{-1}) (Faix [1991](#page-7-29); Rodrigues et al. [1998](#page-8-8); Michell and Higgins [2002\)](#page-7-25) that decline as the temperature of the treatment increases, which probably contribute towards the increase in the CO band at 1708 cm−1 (González-Peña and Hale [2009](#page-7-31)). The changes in lignin can be related to the loss of syringyl units, which tend to be liberated easier than the guaiacyl monomers by a thermal degradation (Faix et al. [1990\)](#page-7-33). The decrease in content of this group could be related to the decrease in the methoxyl groups in lignin, which leads to the loss of a monomer. Previous research indicates that a shift occurs in this group, which can be attributed to the breaking of aliphatic sidechains in lignin and/or condensation reactions (Windeisen et al. [2007\)](#page-8-15). Diferences between wet and dry modifcations, which could be related to the cross-linking reactions from the degradation of hemicelluloses and lignin, are difficult to make out due to the absorbance of the aromatic rings of furan-type intermediates and lignin overlap (Altgen et al. [2018](#page-7-34)).

The bands at 1335 cm⁻¹ and 1323 cm⁻¹ represent the C–H bending of polysaccharides and the condensed units of syringyl and guaiacyl, respectively, thus combining most of the structural components of wood (Faix [1991](#page-7-29)). As these bands represent many structures, it is difficult to conclude which is the structure that causes the changes to be able to diferentiate between the open and closed system. There were distortions in the intensities of the bands in the region close to 1150–950 cm⁻¹, when DRIFT spectra were measured directly from wood surfaces (Pandey [1999\)](#page-8-16). These represent the C–O stretch in polysaccharides. Similar distortions can be seen in both open and closed system modifcations, but no diference between the modifcations can be seen, as shown in Fig. [4](#page-4-0).

Fig. 4 Comparison of FTIR spectra of thermally modifed *E. nitens* in the closed system $(160 °C$ and 0.61 bar) (top) and in the open system (210 °C and atmospheric pressure) (bottom)

The crystallinity index measured directly on solid wood samples shows a tendency to increase at elevated pressure under wet conditions and at elevated temperatures and drier conditions under atmospheric pressure, but it shows little to no change at the lower pressures and at lower temperatures (Table [2](#page-5-0)). In the closed system under wet conditions, there was an increase at 150 °C and 0.14 bar and a decrease at 230 °C under drier conditions. It is suspected that this increase is due to the sample surface, which could have had a mixture of early/late wood, and as the measurements are done in situ, it could have caused some issues at the moment of the crystallinity measurement. According to Segal et al. [\(1959](#page-8-11)), the diference between the intensities or the height of the angles between 18 and 22° (2θ) of the difraction pattern in relation to the intensity of the crystalline peak at 22° is the indicator of the degree of crystallinity. A higher diference between the intensity values is an indicator of increasing crystallinity. This could be explained by considering that cellulose chains swell when exposed to high temperature and humidity, thus increasing the distance between the crystal planes, as stated by Cheng et al. [\(2017](#page-7-23)).

At higher temperatures, there are more hemicellulose cleavage products due to depolymerisation and the production of more reactive intermediates, such as furfurals, which cause more cross linking reactions that increase the lignin polymer network (Boonstra and Tjeerdsma [2006\)](#page-7-35), causing probably an increase in the relative proportion of crystalline cellulose due to a rearrangement of the cellulose molecules (Bhuiyan et al. [2000](#page-7-13)). The acetic acid generated by the modifcation degrades the hemicelluloses

Table 2 Degree of crystallinity (CI) by X-ray difraction in open and closed thermal modifcations of *E. nitens*

Peak tempera- ture $(^{\circ}C)$	Max pressure (MPa)	I_{002}	I_{amorph}	CI(%)
Untreated		988	308	69
Closed system				
150	0.14	1363	364	73
160	0.18	881	302	66
170	0.23	1087	352	68
150	0.47	919	292	68
160	0.61	1401	298	79
170	0.77	1398	315	78
Open system				
160	Atm. ^a	1169	366	69
180	Atm. ^a	1300	385	70
200	Atm. ^a	1301	359	72
210	Atm. ^a	1510	399	74
220	Atm. ^a	1706	426	75
230	Atm. ^a	1200	392	67

a Atmospheric pressure

and could also degrade the microfbrils of the amorphous region of the cellulose, and even attack the crystalline region (Sivonen et al. [2002\)](#page-8-5). Both chemical changes occur in wet and dry modifcations, but the acidity is higher in the wet modifcation in a closed system (Wentzel et al. [2019\)](#page-8-0). Omitting the crystallinity index at 150 °C and 0.14 bar, the modifcations at lower pressures in the wet process showed no signifcant diference to the untreated samples (Fig. [5](#page-5-1)), but at higher pressure, there was a noticeable increase in the degree of crystallinity index for the modifcations at higher temperatures and pressure (Fig. [6](#page-5-2)). Bhuiyan et al. [\(2000\)](#page-7-13) suggested that under moist conditions there is a stimulating efect on the crystallization of amorphous cellulose. As already established by the FTIR spectra and Wentzel et al. ([2019](#page-8-0)), the degradation

Fig. 5 X-ray difraction of thermally modifed *E. nitens* at 30% RH in the closed system and an untreated sample. From top to bottom: 170, 160, 150 °C and untreated specimen

Fig. 6 X-ray difraction of thermally modifed *E. nitens* at 100% RH in the closed system and an untreated sample. From top to bottom: 170, 160, 150 °C and untreated specimen

Fig. 7 X-ray difraction of thermally modifed *E. nitens* at atmospheric pressure and untreated sample. From top to bottom: 230, 220, 210, 200, 180, 160 °C and untreated specimen

of hemicelluloses is closely related to the xylans, which crystallize due to the controlled pyrolysis during the modifcation, thus they could also infuence the change in the degree of crystallinity (Akgül et al. [2007](#page-6-2)). As for the open system modifcation under drier conditions, there is a steady increase in the crystallinity index as the temperature rises, excluding the modification at 230 $^{\circ}$ C (Fig. [7](#page-6-3)), but not a signifcant diference to the reference with the exception of the modifcations at 210 and 220 °C respectively. Besides the increase in acetic acid, the increase in the crystallinity index in the dry modifcation can be related to the preferred degradation of less ordered molecules (Sivonen et al. [2002](#page-8-5)). The results obtained in the open system were in contrast to what was shown by Cheng et al. ([2017\)](#page-7-23), where there was a decrease in the crystallinity when the temperature rises and an increase at 220 °C. The diference could be due to the diferent species used and the types of modifcation.

When directly comparing wet and dry modifcations with processes with the same CML (210 °C and atmospheric pressure and 160 $^{\circ}$ C and 0.61 bar) the wet modifcation presents a slightly higher crystallinity index than the dry modification, but overall the difference is not significant.

There are diferences when directly comparing modifcations with the same CML for both FTIR and X-ray difraction, but in the case of cellulose crystallinity, it is difficult to be certain that there are significant differences between modifcations under pressure in wet conditions and without pressure under drier conditions.

4 Conclusion

The biggest chemical diferences between modifcations under pressure and without pressure were observed in the variations in the peaks of the C=O linkages (1750 and 1700 cm−1). These variations can be related to the formation of acetic acid, that accelerates the degradation of the hemicelluloses, and esterification, which influences the changes in the lignin. The OH bending also showed diferent peaks between open and closed system (band 1385 cm⁻¹), but it was harder to diferentiate the lignin peaks between both processes, as it was difficult to separate the different chemical reactions due to the depolymerization of lignin by only observing the bands. On the other hand, the degree of crystallinity did not show signifcant diferences between both modifcations, although the modifcations under wet condition had slightly higher values. Further research on the measurement of the crystallinity in solid wood samples should be done, focusing on diferences between late and early wood and the direction of the sample (longitudinal, transversal, radial). There were diferences in FTIR spectra and cellulose crystallinity when directly comparing modifcations with the same CML under diferent conditions. This should be taken into consideration in further research on the properties of thermally modifed wood.

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Compliance with ethical standards

Conflict of interest On behalf of all authors, the corresponding author states that there is no confict of interest.

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