

Synthesis and antimicrobial activities of some isoxazolyl thiazolyl pyrazoles

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Abstract A series of isoxazolyl thiazolyl pyrazoles **5a–d** was synthesized by multi-step process, starting from 3-acetyl-4-hydroxy-6-methyl-2H-pyran-2-one (dehydroacetic acid, DHAA) **1**. DHAA **1** was easily converted to thiosemicarbazone **2** which on reaction with α -bromoketones yielded thiazolyl hydrazones **3**. Refluxing **3** in ethanol-acetic acid furnished 1-(5-hydroxy-3-methyl-1-substituted pyrazol-4-yl)-1,3-butanediones **4**. Finally, the title compounds **5a–d** were synthesized from **4** on treatment with hydroxylamine. The in vitro antimicrobial activity of compounds **3a–d**, **4a–d** and **5a–d** were tested. Most of the synthesized compounds exhibited significant antibacterial and antifungal activities.

Keywords Isoxazole · Thiazole · Pyrazole ·
Antibacterial · Antifungal

Introduction

The designing of new types of polyheterocyclic compounds along with the refining of procedures for synthesis is an urgent target of the modern heterocyclic chemistry. Amongst the five-membered heterocyclic compounds, much interest has been focused on the pyrazole nucleus which is known to possess a broad spectrum of biological

properties such as antipyretic, analgesic and anti-inflammatory (Menozzi *et al.*, 2000; Turan-Zitouni *et al.*, 2001; Palomer *et al.*, 2002; Bekhit *et al.*, 2008), antiparasitic (Rathelot *et al.*, 2002; Kumar *et al.*, 2006), antitubercular (Kaymakci oglu and Rollas 2002; Dixit *et al.*, 2006), anticonvulsant (Kucukguzel *et al.*, 2000), antimicrobial (El-Gaby *et al.*, 2000; Baraldi *et al.*, 2002; Akbas *et al.*, 2005), antiviral (Larsen *et al.*, 1999), anticancer (Poreba *et al.*, 2001; Ishida *et al.*, 2002; Witherington *et al.*, 2003; Rostom *et al.*, 2003; Park *et al.*, 2005), herbicidal (Liu *et al.*, 2007), etc. Likewise, thiazole nucleus present in a number of different types of compounds has been found to be associated with different types of pharmacological properties like anti-inflammatory (Holla *et al.*, 2003; Venugopala and Jayashree, 2003; Pattan *et al.*, 2007; Rostom *et al.*, 2009), antitumor (El-Subbagh *et al.*, 1999; Ramla *et al.*, 2006), antifungal (Chimenti *et al.*, 2011), antihypertensive (William *et al.*, 1992), anti-HIV (Frank *et al.*, 1995), etc. Antimicrobial activities of some substituted thiazoles are well established because it possess (S–C=N) toxophoric unit in the ring. Further, isoxazoles possess analgesic, anti-inflammatory, antimicrobial and other pharmacological activities (Habeeb *et al.*, 2001; Velikorodov and Sukhenkol, 2003; Panda *et al.*, 2009; Madhavi *et al.*, 2010), etc. Numerous biological properties are possessed by the molecules in which both pyrazole and thiazole moieties are present together (Bekhit *et al.*, 2003; Rostom, 2006). Similarly, the compounds containing pyrazole and isoxazole moieties have been shown to exhibit antihyperglycemic, analgesic, anti-inflammatory, antipyretic antibacterial, antiviral, antitumor, antifungal and antidepressant activities (Sugiura *et al.*, 1977; Jungheim, 1989; Xue *et al.*, 1998; Kang *et al.*, 2000). Furthermore, various types of pharmacological properties are exhibited by the molecules containing isoxazole and thiazole

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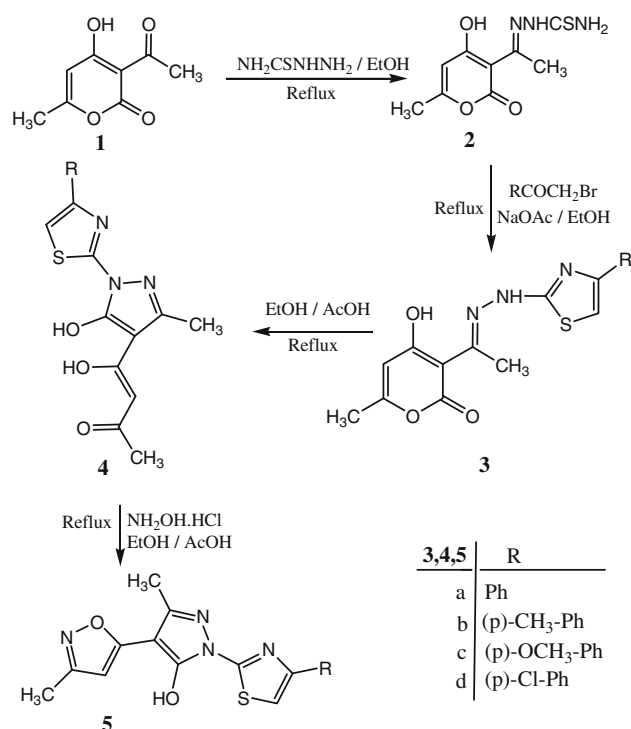
pharmacores existing in the same molecule (Rajanarendar *et al.*, 2010). It was, therefore, thought worthwhile to undertake the synthesis of some pyrazole derivatives possessing a thiazole moiety at position-1 and isoxazole moiety at position-4 to frame potential biological molecules. Keeping in view of growing interest in the reactions of significance available in literature and numerous biological properties possessed by them, we herein report a detailed account of results of our investigations on the multistep synthesis, characterization and antimicrobial activities of **3a–d**, **4a–d** and the title compounds, 3-methyl-4-(3-methylisoxazol-5-yl)-1-(4-substituted phenylthiazol-2-yl)-1*H*-pyrazol-5-ol derivatives **5a–d**.

Results and discussion

Chemistry

The title compounds 1-(4-(phenyl/4-substituted phenyl)thiazol-2-yl)-3-methyl-4-(3-methylisoxazol-5-yl)-1*H*-pyrazol-5-ol **5a–d** were synthesized as illustrated in Scheme 1. The thiosemicarbazones **2** were prepared in high yields upon treatment of 2-acetyl-4-hydroxy-6-methyl-2*H*-pyran-2-one (dehydroacetic acid, DHAA) **1** with thiosemicarbazide in equimolar quantities which were smoothly converted into the corresponding hydrazones **3** upon reaction with different 4-substituted phenacyl bromides in the presence of sodium acetate/ethanol. The hydrazones **3** thus obtained underwent rearrangement in ethanol-acetic acid mixture to afford the key intermediate 1-(5-hydroxy-3-methyl-1-substituted pyrazol-4-yl)-1,3-butanediones **4** which on subsequent treatment with hydroxylamine hydrochloride furnished 1-(4-(phenyl/4-substituted phenyl)thiazol-2-yl)-3-methyl-4-(3-methylisoxazol-5-yl)-1*H*-pyrazol-5-ol **5a–d**.

The structures of all the products were established on the basis of elemental analyses, IR, NMR (^1H and ^{13}C) and mass spectral data. The hydrazones **3** exhibited medium intensity NH, OH broad stretching bands in the region $3,600\text{--}3,100\text{ cm}^{-1}$ and strong C=O stretching bands in the region $1,699\text{--}1,684\text{ cm}^{-1}$ in their IR spectra. The ^1H NMR (400 MHz) spectra of these hydrazones displayed singlet in the region at δ 5.89–6.05 due to pyrone ring-H and singlet in the region at δ 6.62–6.85 due to thiazole ring-H which are in agreement with the results as reported in the literature [Singh *et al.*, 1993]. The IR spectra of **4** exhibited absorption bands in the regions $3,600\text{--}3,100$ and $1,718\text{--}1,706$ and $1,646\text{--}1,633\text{ cm}^{-1}$ assigned to OH and CO functionalities, respectively. The ^1H NMR (400 MHz) spectra of **4** displayed four signals due to each of the CH_3 groups present in pyrazolyl moiety in enol and keto forms, in the regions at δ 2.04–2.07 and 2.54–2.66 and δ



Scheme 1 Synthesis of 1-(4-(4-substituted-phenyl)thiazol-2-yl)-3-methyl-4-(3-methylisoxazol-5-yl)-1*H*-pyrazol-5-ols **5**

2.20–2.29 and δ 2.50–2.62, respectively, and one signal in each case due to CH_2 and olefinic-H in the region at δ 4.04–4.06 and 6.67–6.69 (exchangeable with D_2O), respectively. The singlet observed in the region at δ 7.29–7.62 was assigned to thiazole ring-H. The ratio of aromatic to aliphatic protons was found satisfactory. In ^{13}C NMR spectra of **4** displayed peaks in the regions at δ 188.48–188.87, 186.94–187.44 and 183.02–183.24 due to carbonyl carbons. All these characteristic features corroborate existence of keto-enol tautomerism in **4** (Fig. 1). Further, the % ratio of keto/enol forms of **4** (**4A**:**4B**) were calculated from ^1H NMR (400 MHz) spectral data, and were found as **4a** (35:65), **4b** (25:75), **4c** (50:50) and **4d**

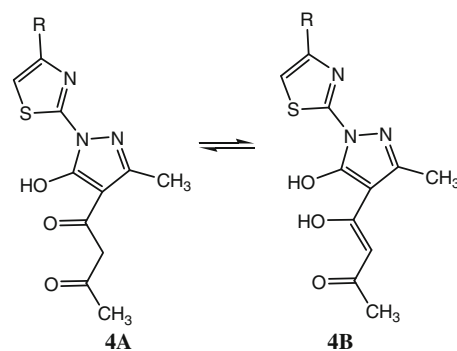


Fig. 1 Keto-enol tautomeric forms of 1-(1-(4-(4-substituted-phenyl)thiazol-2-yl)-5-hydroxy-3-methyl-1*H*-pyrazol-4-yl)butane-1,3-diones **4**

(37:63). A careful analysis of spectral data shows an interesting trend that **4a**, **4b** and **4d** exist predominantly in enolic form but in **4c** keto-enol forms are in equal amounts.

The title compounds 1-(4-(phenyl/4-substituted phenyl)thiazol-2-yl)-3-methyl-4-(3-methylisoxazol-5-yl)-1*H*-pyrazol-5-ol **5a–d** exhibited the characteristics broad absorption bands of medium intensity in the region 3,432–3,204 cm^{-1} due to OH of pyrazole moiety in their IR spectra. The ^1H NMR (400 MHz) spectra of **5a–d** showed a singlet in the region δ 6.51–6.55 due to isoxazole-H and another singlet in the region δ 7.25–7.71 assigned to thiazole-H. The mass spectral studies and elemental analysis results were found satisfactory. Physical and chemical data of the compounds **3a–d**, **4a–d** and **5a–d** are detailed in Table 1.

Antimicrobial activity

All the newly synthesized compounds has been assayed for their in vitro antibacterial activity against Gram-positive *Bacillus subtilis*, *Staphylococcus aureus* and Gram-negative *Escherichia coli* and in vitro antifungal activity against *Candida albicans* and *Aspergillus niger*. The bacterial and fungal strains selected for this study are most common and

easily available. *B. subtilis* may contaminate food and produces the proteolytic enzyme subtilisin and is responsible for causing ropiness. *S. aureus* is a common cause of boils, sties and skin infections and may cause serious infections in debilitated persons. *E. coli* can cause gastroenteritis, urinary tract infections, and neonatal meningitis. The fungus *C. albicans* is often a benign member of the mucosal flora; however, it commonly causes mucosal disease with substantial morbidity and in vulnerable patients it causes life-threatening bloodstream infections. *A. niger* is not only a xerophilic fungi, but is also a thermo-tolerant organism. Double strength nutrient broth and Sabouraud dextrose broth were employed for bacterial and fungal growth, respectively. The pMICs (negative logarithm of minimum inhibitory concentrations) were recorded as the minimum concentration of a compound that inhibits the growth of tested microorganisms. pMICs were determined by means of standard serial dilution method. The pMIC values are presented in Table 2. Most of the synthesized compounds showed good antibacterial activity with pMICs ranging from 1.435 to 1.778. The new title compounds isoxazolyl thiazolyl pyrazoles **5a–d** and their precursors **3a–d** and **4a–d** were successfully synthesized. The synthesized compounds were found to exhibit moderate

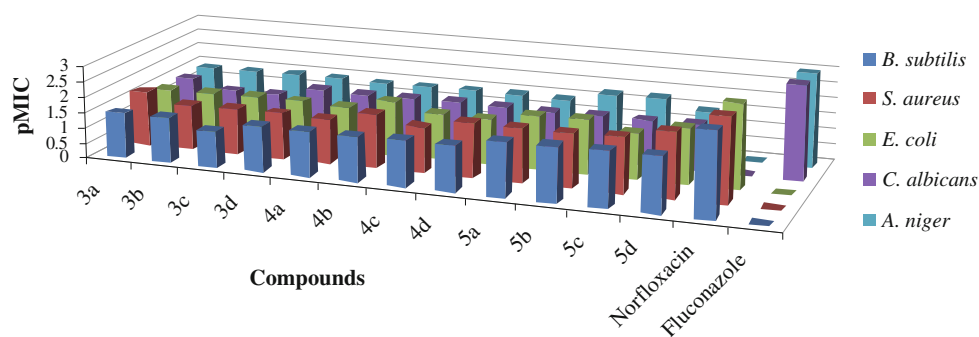
Table 1 Physical and analytical data of compounds **3a–d**, **4a–d** and **5a–d**

Compds	R	Reaction time (min)	Yield %	Melting point °C	Molecular formula	Found % (Calculated) C H N		
3a	H	30	76	188–190 Lit. mp 190–191	$\text{C}_{17}\text{H}_{15}\text{N}_3\text{O}_3\text{S}$	60.20 (59.81)	4.22 (4.43)	12.69 (12.31)
3b	CH_3	25	71	204–206	$\text{C}_{18}\text{H}_{17}\text{N}_3\text{O}_3\text{S}$	60.34 (60.83)	4.71 (4.82)	12.08 (11.82)
3c	OCH_3	30	73	189–190	$\text{C}_{18}\text{H}_{17}\text{N}_3\text{O}_4\text{S}$	57.94 (58.21)	4.96 (4.61)	11.59 (11.31)
3d	Cl	35	67	178–180	$\text{C}_{17}\text{H}_{14}\text{ClN}_3\text{O}_3\text{S}$	54.72 (54.33)	3.83 (3.75)	11.26 (11.18)
4a	H	120	57	142–144 Lit. mp 145	$\text{C}_{17}\text{H}_{15}\text{N}_3\text{O}_3\text{S}$	60.15 (59.81)	4.27 (4.43)	12.43 (12.31)
4b	CH_3	120	59	110–112	$\text{C}_{18}\text{H}_{17}\text{N}_3\text{O}_3\text{S}$	61.14 (60.83)	5.08 (4.82)	12.23 (11.82)
4c	OCH_3	125	64	158–160	$\text{C}_{18}\text{H}_{17}\text{N}_3\text{O}_4\text{S}$	58.53 (58.21)	4.93 (4.61)	10.96 (11.31)
4d	Cl	125	61	120–122	$\text{C}_{17}\text{H}_{14}\text{ClN}_3\text{O}_3\text{S}$	54.56 (54.33)	3.61 (3.75)	11.47 (11.18)
5a	H	115	68	208–210	$\text{C}_{17}\text{H}_{14}\text{N}_4\text{O}_2\text{S}$	60.28 (60.34)	4.37 (4.17)	16.58 (16.56)
5b	CH_3	125	63	220–222	$\text{C}_{18}\text{H}_{16}\text{N}_4\text{O}_2\text{S}$	61.72 (61.35)	4.81 (4.58)	15.59 (15.90)
5c	OCH_3	110	74	232–234	$\text{C}_{18}\text{H}_{16}\text{N}_4\text{O}_3\text{S}$	59.01 (58.68)	4.10 (4.38)	15.07 (15.21)
5d	Cl	125	69	258–260	$\text{C}_{17}\text{H}_{13}\text{ClN}_4\text{O}_2\text{S}$	55.13 (54.77)	3.16 (3.51)	15.16 (15.03)

Table 2 In vitro antimicrobial activity (pMIC) of compounds **3a–d**, **4a–d** and **5a–d**

Compounds	Antibacterial activity			Antifungal activity	
	<i>B. subtilis</i>	<i>S. aureus</i>	<i>E. coli</i>	<i>C. albicans</i>	<i>A. niger</i>
3a	1.435	1.737	1.435	1.435	1.435
3b	1.453	1.453	1.453	1.152	1.435
3c	1.171	1.472	1.472	1.171	1.472
3d	1.478	1.478	1.478	1.478	1.478
4a	1.435	1.435	1.435	1.435	1.435
4b	1.453	1.754	1.754	1.453	1.453
4c	1.472	1.472	1.472	1.472	1.472
4d	1.477	1.778	1.477	1.477	1.477
5a	1.733	1.733	1.733	1.432	1.432
5b	1.750	1.750	1.750	1.449	1.750
5c	1.771	1.771	1.470	1.470	1.771
5d	1.775	2.077	1.775	1.477	1.474
Norfloxacin	2.698	2.698	2.698	–	–
Fluconazole	–	–	–	3.000	3.000

antibacterial activity, e.g. **5a–d** against *B. subtilis*; **3a**, **4b**, **4d**, **5a–c** against *S. aureus* and **4b**, **5a**, **5b**, **5d** against *E. coli*, whereas the compounds **5b**, **5c** were found to be active fungicidal agent against *A. niger*. However, amongst the newly synthesized title compounds, 1-(4-(4-chlorophenyl)thiazol-2-yl)-3-methyl-4-(3-methylisoxazol-5-yl)-1*H*-pyrazol-5-ol **5d** was found to exhibit the promising antibacterial activity against *S. aureus* (pMIC = 2.077, in comparison to the standard reference norfloxacin, pMIC = 2.698). Almost all the compounds showed moderate level of antifungal activity with pMICs ranging from 1.435 to 1.478 in comparison to the standard reference fluconazole (pMIC = 3.000). However, **5b** (pMIC = 1.750) and **5c** (pMIC = 1.771) exhibited good fungicidal activity. The results of antimicrobial activity illustrate that the presence of electron donating groups, CH₃ and OCH₃ on the phenyl ring, in compounds **3–5** increases the antibacterial as well as antifungal activities. Further, the presence of electron withdrawing group, Cl on the phenyl ring, in compounds **5d** enhances the antibacterial activity to a greater extent.

Fig. 2 In vitro antimicrobial activity (pMIC) of compounds **3a–d**, **4a–d** and **5a–d**

The graphical data of antimicrobial activities of all the synthesized compounds is depicted in Fig. 2.

Conclusion

The synthesized compounds were found to exhibit moderate antibacterial activity, e.g. **5a–d** against *B. subtilis*; **3a**, **4b**, **4d**, **5a–c** against *S. aureus* and **4b**, **5a**, **5b**, **5d** against *E. coli*, whereas the compounds **5b**, **5c** were found to be active fungicidal agent against *A. niger*. However, amongst the newly synthesized title compounds, 1-(4-(4-chlorophenyl)thiazol-2-yl)-3-methyl-4-(3-methylisoxazol-5-yl)-1*H*-pyrazol-5-ol **5d** was found to exhibit the promising antibacterial activity against *S. aureus*.

Experimental

Melting points were determined in open capillaries and are uncorrected. The IR absorption spectra were scanned on Perkin Elmer Spectrum, BX II FTIR spectrometer using potassium bromide (KBr) pellets and the wave numbers were given in cm⁻¹. The ¹H NMR and ¹³C NMR spectra were recorded on Bruker Avance II 400 spectrometer at 400 and 100 MHz, respectively, in deuteriochloroform or deuteriochloroform + dimethyl sulfoxide-d₆ or dimethyl sulfoxide-d₆. The chemical shifts are reported in parts per million (δ ppm) using tetramethylsilane (TMS) as an internal standard. Coupling constants J are valued in Hertz (Hz). Mass spectra were recorded on Waters Micromass Q-ToF Micro (ESI) spectrometer. Elemental analysis was carried out using Vario Micro Cube Elementar CHNS analyser. Analytical results for C, H and N were within $\pm 0.4\%$ of the theoretical values. The purities of the compounds were checked by thin layer chromatography (TLC) using readymade silica gel (SIL G/UV₂₅₄, ALUGRAM) plates. The spots were visualized under ultraviolet (UV) lamp. Solvents were dried using standard literature procedures. The thiosemicarbazones **2** were prepared by the condensation of equimolar quantities of 2-acetyl-4-

hydroxy-6-methyl-2*H*-pyran-2-one and thiosemicarbazide in absolute alcohol (Singh *et al.*, 1993).

General procedure for the synthesis of 4-hydroxy-6-methyl-3-(1-(2-(4-substituted-phenylthiazol-2-yl)hydrazono)ethyl)-2*H*-pyran-2-ones (**3a–3d**)

To a solution of **2** (2.41 g, 0.01 mol) and anhydrous sodium acetate (0.82 g, 0.01 mol) in absolute ethanol, the appropriate *p*-substituted phenacyl bromide (0.01 mol) was added slowly with stirring. The reaction mixture was refluxed gently for 25–35 min. The yellowish crystalline solid thus obtained was filtered, washed with cold ethanol, and recrystallized from dimethyl formamide and water. Physical and analytical data are recorded in Table 1.

4-Hydroxy-6-methyl-3-(1-(2-(4-phenylthiazol-2-yl)hydrazono)ethyl)-2*H*-pyran-2-one (**3a**)

Yield 76%; mp 188–190°C (Lit. mp 190–191°C; Singh *et al.*, 1993); IR (KBr): 3,400–3,100 (NH, OH), 1,698 cm⁻¹(CO); MS: (TOF MS ES⁺) *m/z* 342.2 [M + H]⁺.

4-Hydroxy-6-methyl-3-(1-(2-(4-tolylthiazol-2-yl)hydrazono)ethyl)-2*H*-pyran-2-one (**3b**)

Yield 71%; mp 204–206°C; IR (KBr) 3,550–3,200 (NH, OH), 1,699 cm⁻¹(CO); ¹H NMR (DMSO-d₆) δ 2.16 (s, 3H, CH₃), 2.33 (s, 3H, CH₃), 2.70 (s, 3H, CH₃), 5.89 (s, 1H, pyrone-H), 6.62 (s, 1H, thiazole-H), 7.23 (d, 2H, *J* = 8.0 Hz, H-3'',5''), 7.61 (d, 2H, *J* = 8.08 Hz, H-2'',6''); MS: (TOF MS ES⁺) *m/z* 356.1 [M + H]⁺; C₁₈H₁₇N₃O₃S; Calc. C 60.83, H 4.82, N 11.82; Found C 60.34, H 4.71, N 12.08.

4-Hydroxy-3-(1-(2-(4-(4-methoxyphenyl)thiazol-2-yl)hydrazono)ethyl)-6-methyl-2*H*-pyran-2-one (**3c**)

Yield 73%; mp 189–190°C; IR (KBr): 3,600–3,220 (NH, OH), 1,687 cm⁻¹ (CO); ¹H NMR (DMSO-d₆): δ 2.08 (s, 3H, CH₃), 2.69 (s, 3H, CH₃), 3.84 (s, 3H, OCH₃), 6.05 (s, 1H, pyrone-H), 6.69 (s, 1H, thiazole-H), 6.94 (d, 2H, *J* = 8.72 Hz, H-3'',5''), 7.88 (d, 2H, *J* = 8.72 Hz, H-2'',6''); MS: (TOF MS ES⁺) *m/z* 372.2 [M + H]⁺; C₁₈H₁₇N₃O₄S; Calc. C 58.21, H 4.61, N 11.31; Found C 57.94, H 4.96, N 11.59.

3-(1-(2-(4-(4-Chlorophenyl)thiazol-2-yl)hydrazono)ethyl)-4-hydroxy-6-methyl-2*H*-pyran-2-one (**3d**)

Yield 67%; mp 178–180°C; IR (cm⁻¹): 3,500–3,200 (NH, OH), 1,684 cm⁻¹ (CO); ¹H NMR (DMSO-d₆) δ 2.20 (s, 3H, CH₃), 2.63 (s, 3H, CH₃), 5.89 (s, 1H, pyrone-H), 6.85 (s, 1H, thiazole-H), 7.38–7.76 (m, 4H, Ar-H) MS: (TOF

MS ES⁺) *m/z* 376.2 [M + H]⁺; C₁₇H₁₄ClN₃O₃S; Calc. C 54.33, H 3.75, N 11.18; Found C 54.72, H 3.83, N 11.26.

General procedure for the synthesis of 1-(1-(4-(4-substituted-phenyl)thiazol-2-yl)-5-hydroxy -3-methyl-1*H*-pyrazol-4-yl)butane-1,3-diones (**4a–4d**)

The hydrazone **3** (0.01 mol) was added in one lot to a hot solution of acetic acid and refluxed for approximately 2 h. The solid thus obtained, on cooling the reaction mixture overnight, was filtered, washed with little cold ethanol and crystallised from acetonitrile. Physical and analytical data are recorded in Table 1.

1-(5-Hydroxy-3-methyl-1-(4-phenylthiazol-2-yl)-1*H*-pyrazol-4-yl)butane-1,3-dione (**4a**)

Yield 57%; mp 142–144°C (Lit. mp 145°C; Singh *et al.*, 1993); IR (KBr) 3,550–3,200 (OH), 1,706, 1,633 cm⁻¹ (CO); ¹H NMR (DMSO-d₆) δ 2.04, 2.27 (two s, 3H, CH₃), 2.59, 2.62 (two s, 3H, CH₃), 4.04 (s, 2H, CH₂), 6.68 (s, 1H, olefinic-H), 7.29–7.84 (m, 6H, ArH and thiazole-H); ¹³C NMR (DMSO-d₆) (enol + keto forms) δ 13.49, 13.71, 24.17, 30.97, 56.01, 97.76, 100.22, 103.84, 108.77, 126.41, 128.31, 128.68, 133.96, 134.05, 149.80, 149.84, 152.19, 152.70, 153.52, 159.05, 160.08, 183.13, 187.28, 188.5; MS: (TOF MS ES⁺) *m/z* 342.2 [M + H]⁺.

Hydroxy-3-methyl-1-(4-tolylthiazol-2-yl)-1*H*-pyrazol-4-yl)butane-1,3-dione (**4b**)

Yield 59%; mp 110–112°C; IR (KBr) 3,600–3,200 (OH), 1,707, 1,646 cm⁻¹(CO); ¹H NMR (DMSO-d₆) δ 2.06, 2.27 (two s, 3H, CH₃), 2.37, (s, 3H, CH₃), 2.60, 2.64 (two s, 3H, CH₃), 4.04 (s, 2H, CH₂), 6.69 (s, 1H, olefinic-H), 7.21 (d, 2H, *J* = 7.9 Hz, H-3'', 5''), 7.39 (s, 1H, thiazole-H), 7.84 (d, 2H, *J* = 7.9 Hz, H-2'',6''); ¹³C NMR (DMSO-d₆) (enol + keto forms) δ 13.49, 13.79, 21.34, 24.15, 30.95, 56.00, 97.73, 100.19, 103.81, 108.01, 126.33, 129.22, 129.36, 131.60, 137.49, 149.84, 149.88, 152.12, 152.71, 153.52, 159.08, 160.12, 183.24, 187.17, 188.48; MS: (TOF MS ES⁺) *m/z* 356.1 [M + H]⁺; C₁₈H₁₇N₃O₃S; Calc. C 60.83, H 4.82, N 11.82; Found C 61.14, H 5.08, N 12.23.

1-(5-Hydroxy -1-(4-(4-methoxyphenyl)thiazol-2-yl) -3-methyl -1*H*-pyrazol-4-yl)butane-1,3-dione (**4c**)

Yield 64%; mp 158–160°C; IR (KBr) 3,600–3,300 (OH), 1,718, 1,646 cm⁻¹(CO); ¹H NMR (DMSO-d₆) δ 2.05, 2.20 (two s, 3H, CH₃), 2.50, 2.54 (two s, 3H, CH₃), 3.81 (s, 3H, OCH₃), 4.04 (s, 2H, CH₂), 6.67 (s, 1H, olefinic-H), 7.02 (d, *J* = 8.7 Hz, H-2'', 6''), 7.62 (s, 1H, thiazole-H), 7.93 (d, *J* = 8.7 Hz, H-3'',5''); ¹³C NMR (DMSO-d₆) (enol + keto

forms) δ 13.39, 13.76, 24.14, 30.91, 55.40, 56.41, 98.61, 100.10, 103.78, 108.88, 114.28, 123.90, 126.89, 136.71, 149.81, 149.93, 152.17, 152.87, 153.18, 159.47, 159.70, 161.51, 183.22, 186.94, 188.87; MS: (TOF MS ES⁺) m/z 372.04 [M + H]⁺; C₁₈H₁₇N₃O₄S; Calc. C 58.21, H 4.61, N 11.31; Found C 58.53, H 4.93, N 10.96.

1-(1-(4-(4-Chlorophenyl)thiazol-2-yl)-5-hydroxy-3-methyl-1H-pyrazol-4-yl)butane-1,3-dione (4d)

Yield 61%; mp 120–122°C; IR (KBr) 3,550–3,100 (OH), 1,717, 1,636 cm⁻¹(CO); ¹H NMR (DMSO-d₆) δ 2.07, 2.29 (two s, 3H, CH₃), 2.62, 2.66 (two s, 3H, CH₃), 4.06 (s, 2H, CH₂), 6.68 (s, 1H, olefinic-H), 7.40 (d, 2H, $J = 8.4$ Hz, H-2'',6''), 7.45 (s, 1H, thiazole-H), 7.933 (d, $J = 8.4$ Hz, H-3'',5''); ¹³C NMR (DMSO-d₆) (enol + keto forms) δ 13.49, 13.69, 24.17, 30.94, 55.94, 97.75, 100.25, 103.86, 109.28, 127.80, 128.76, 132.64, 132.70, 133.58, 148.66, 152.32, 152.79, 153.64, 159.13, 160.15, 183.02, 187.44, 188.51; MS: (TOF MS ES⁺) m/z 376 [M + H]⁺; C₁₇H₁₄ClN₃O₃S; Calc. C 54.33, H 3.75, N 11.18; Found C 54.56, H 3.61, N 11.47.

General procedure for the synthesis of 1-(4-(4-substituted-phenyl)thiazol-2-yl)-3-methyl-4-(3-methylisoxazol-5-yl)-1H-pyrazol-5-ols (**5a–d**)

The solution of **4** (0.01 mol) and hydroxylamine hydrochloride (0.69 g, 0.01 mol) in 50 ml of ethanol and 50 ml of acetic acid was heated at reflux for approximately 2 h. The reaction mixture was cooled and then allowed to stand overnight. The solid thus obtained, was collected and recrystallised from acetonitrile. Physical and analytical data are recorded in Table 1.

3-Methyl-4-(3-methylisoxazol-5-yl)-1-(4-phenylthiazol-2-yl)-1H-pyrazol-5-ol (5a)

Yield 68%; mp 208–210°C; IR (KBr) 3,204 cm⁻¹ (OH); ¹H NMR (DMSO-d₆) δ 2.27 (s, 3H, CH₃), 2.61 (s, 3H, CH₃), 6.51 (s, 1H, isoxazole-H), 7.33–7.46 (m, 3H, Ar-H), 7.71 (s, 1H, thiazole-H), 8.01 (d, 2H, $J = 7.36$ Hz, H-2'',6''); ¹³C NMR (DMSO-d₆) δ 11.37, 12.92, 93.91, 99.32, 109.4, 126.45, 128.44, 128.96, 134.18, 148.09, 149.63, 152.51, 158.07, 159.42, 163.49; MS: (TOF MS ES⁺) m/z 339.3 [M + H]⁺; C₁₇H₁₄N₄O₂S; Calc. C 60.34, H 4.17, N 16.56; Found C 60.28, H 4.37, N 16.58.

3-Methyl-4-(3-methylisoxazol-5-yl)-1-(4-p-tolylthiazol-2-yl)-1H-pyrazol-5-ol (5b)

Yield 63%; mp 220–222°C; IR (KBr) 3,431 cm⁻¹ (OH); ¹H NMR (DMSO-d₆) δ 2.23 (s, 3H, CH₃), 2.36 (s, 3H,

CH₃), 2.62 (s, 3H, CH₃), 6.52 (s, 1H, isoxazole-H), 6.91–7.84 (m, 5H, Ar-H, thiazole-H); ¹³C NMR (DMSO-d₆) δ 12.84, 11.41, 12.84, 21.36, 94.26, 99.31, 108.79, 126.30, 129.25, 129.39, 137.90, 147.72, 150.0, 151.37, 158.11, 159.41, 163.49; MS (TOF MS ES⁺) m/z 353.4 [M + H]⁺; C₁₈H₁₆N₄O₂S; Calc. C 61.35, H 4.58, N 15.90; Found C 61.72, H 4.81, N 15.59.

1-(4-(4-Methoxyphenyl)thiazol-2-yl)-3-methyl-4-(3-methylisoxazol-5-yl)-1H-pyrazol-5-ol (5c)

Yield 74%; mp 232–234°C; IR (cm⁻¹) 3,432 cm⁻¹ (OH); ¹H NMR (DMSO-d₆) δ 2.30 (s, 3H, CH₃), 2.64 (s, 3H, CH₃), 3.84 (s, 3H, CH₃), 6.54 (s, 1H, isoxazole-H), 6.93 (d, 2H, $J = 2.84$ Hz, H-3'',5''), 7.25 (s, 1H, thiazole-H), 7.86 (d, 2H, $J = 2.84$ Hz, H-2'',6''); ¹³C NMR (DMSO-d₆) δ 11.39, 12.83, 55.35, 94.37, 99.32, 106.71, 114.04, 126.94, 127.22, 127.62, 127.70, 147.69, 149.87, 159.46, 159.63, 163.44; MS (TOF MS ES⁺) m/z 369.2 [M + H]⁺; C₁₈H₁₆N₄O₃S; Calc. C 58.68, H 4.38, N 15.21; Found C 59.01, H 4.10, N 15.07.

1-(4-(4-Chlorophenyl)thiazol-2-yl)-3-methyl-4-(3-methylisoxazol-5-yl)-1H-pyrazol-5-ol (5d)

Yield 69%; mp 258–260°C; IR (KBr) 3,431 cm⁻¹ (OH); ¹H NMR (DMSO-d₆) δ 2.31 (s, 3H, CH₃), 2.66 (s, 3H, CH₃), 6.55 (s, 1H, isoxazole-H), 7.41 (d, 2H, $J = 8.4$ Hz, H-2'',6''), 7.44 (s, 1H, thiazole-H), 7.94 (d, 2H, $J = 8.4$ Hz, H-3'', 5''); ¹³C NMR (DMSO-d₆) δ 11.38, 12.84, 94.43, 99.40, 109.1, 127.76, 128.76, 132.75, 133.63, 138.9, 147.92, 148.81, 155.4, 159.51, 163.34; MS (TOF MS ES⁺) m/z 373.3 [M + H]⁺; C₁₇H₁₃ClN₄O₂S; Calc. C 54.77, H 3.51, N 15.03; Found C 55.13, H 3.16, N 15.16.

Antimicrobial activity

All the twelve newly synthesized compounds **3a–d**, **4a–d** and **5a–d** were screened for their in vitro antimicrobial activity against five microorganisms, two Gram-positive bacteria *B. subtilis* (MTCC 441) and *S. aureus* (MTCC 7443) and one Gram-negative bacteria *E. coli* (MTCC 42) and two fungi *C. albicans* (MTCC 183) and *A. niger* (MTCC 282) by serial tube dilution technique using two solid media Double strength nutrient broth and Sabouraud dextrose broth for bacterial and fungal growth, respectively. The stock solutions (100 μ g/ml) of all the test compounds were prepared by dissolving 1 mg of the test compound in 10 ml of dimethylsulphoxide. Norfloxacin and fluconazole were used as reference against bacteria and fungi, respectively. The fresh cultures were obtained by inoculation of respective microorganism in suitable medium (Double strength nutrient broth in case of bacteria and

Sabouraud dextrose broth in case of fungi) followed by incubation at $37 \pm 1^\circ\text{C}$. The stock solutions of the test compounds were serially diluted in test tubes containing 1 ml of sterile medium to get the concentration of 50–3.12 $\mu\text{g/ml}$ and then inoculated with 100 μl of suspension of respective microorganism in sterile saline. The inoculated test tubes were incubated at $37 \pm 1^\circ\text{C}$ for 24 h in case of *B. subtilis*, *S. aureus* and *E. coli*, at $37 \pm 1^\circ\text{C}$ for 48 h in case of *Candida albicans* and at $37 \pm 1^\circ\text{C}$ for 120 h in case of *A. niger* and their pMICs(-log of Minimum inhibitory concentrations) were determined. Minimum inhibitory concentration (MIC), in microbiology, is the lowest concentration of an antimicrobial agent that will inhibit the visible growth of a microorganism after incubation.

The reference compounds Norfloxacin and fluconazole were also tested under similar conditions to compare the results of tested compounds. The data for the antibacterial and antifungal studies are listed in Table 2 and expressed in Fig. 2.

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